

Supporting Information

Shape-Controlled Synthesis of Gallium-Palladium (GaPd₂) Nanocrystals as High-Performance Electrocatalysts for Hydrogen Evolution Reaction

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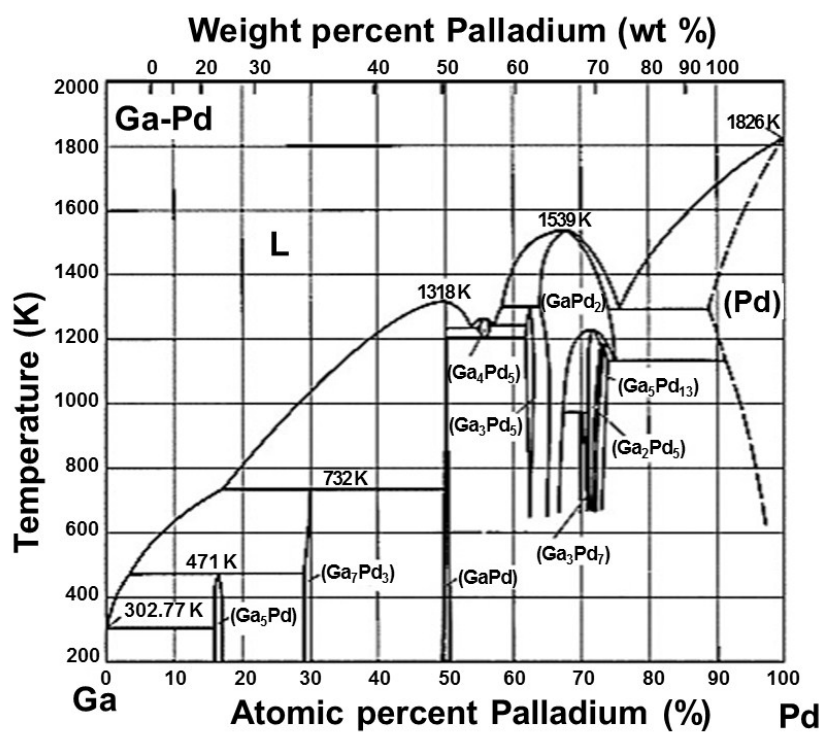


Fig. S1 Phase diagram of gallium and palladium.

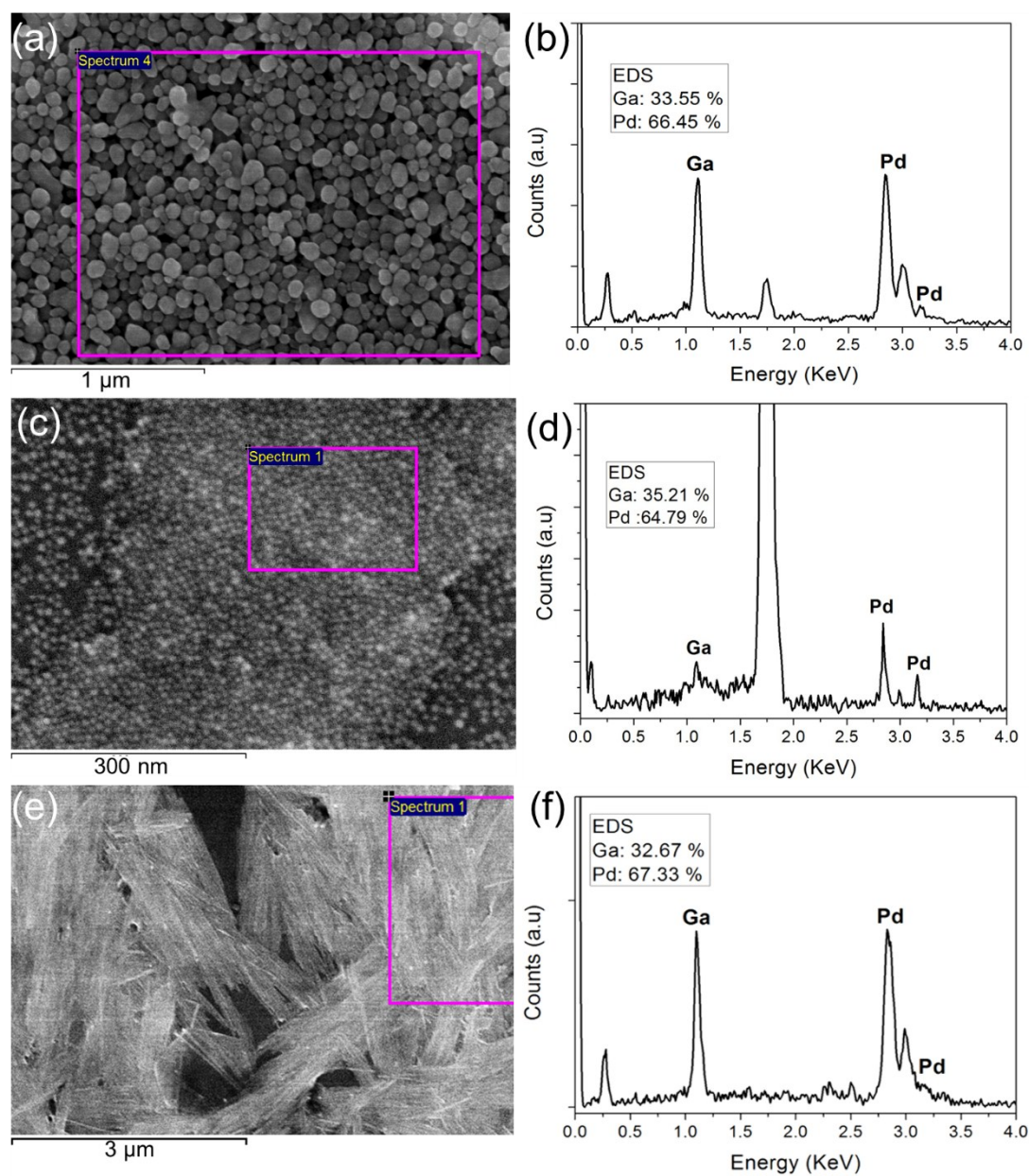


Fig. S2 Energy dispersive spectroscopy (EDS) analysis under SEM (a-b) GaPd₂ polyhedron (c-d) GaPd₂ nanoparticles (e-f) GaPd₂ nanowires

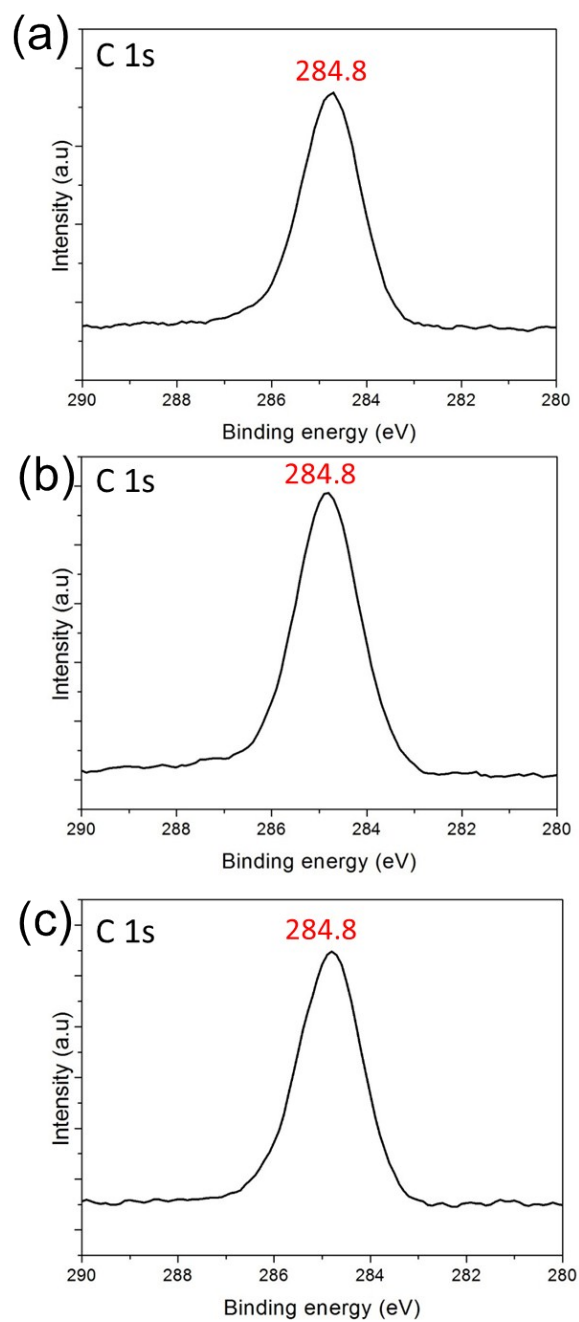


Fig. S3 XPS spectra C1s calibration of (a) GaPd₂ polyhedron (b) GaPd₂ nanoparticles (c) GaPd₂ nanowires

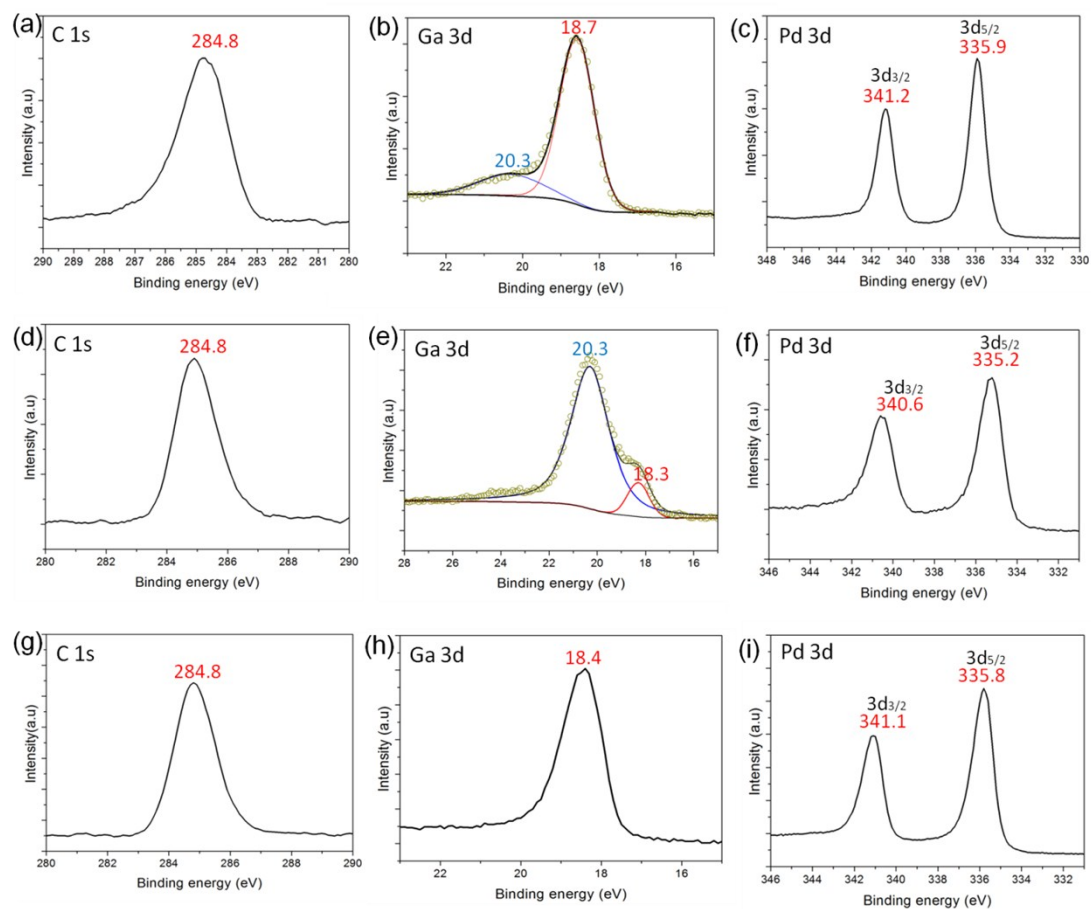


Fig. S4 XPS spectra of GaPd₂ nanomaterials (data obtained after 2 weeks of samples preparation) (a-c) GaPd₂ polyhedral nanocrystals (d-f) GaPd₂ nanoparticles (g-i) GaPd₂ nanowires

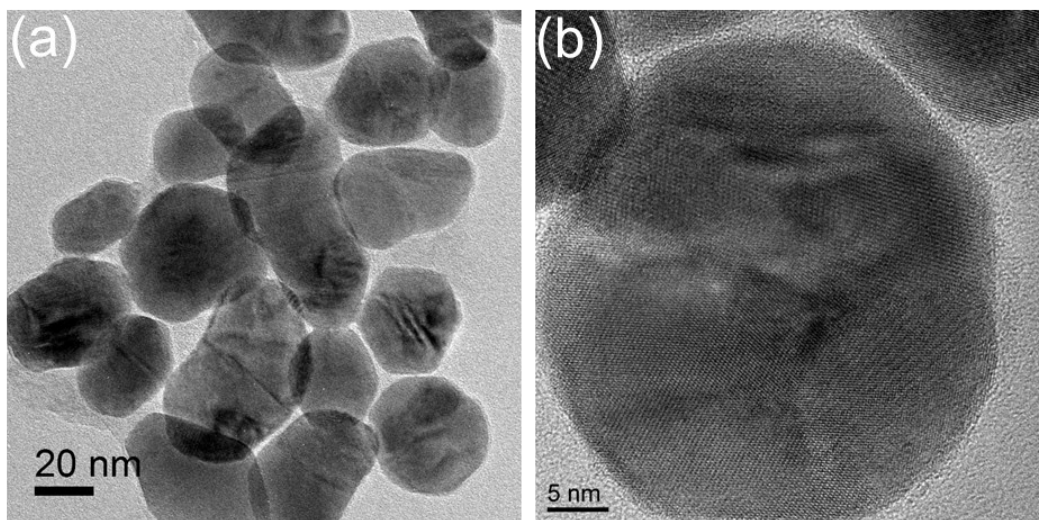


Fig. S5 (a-b) The polyhedral GaPd₂ nanocrystals TEM images

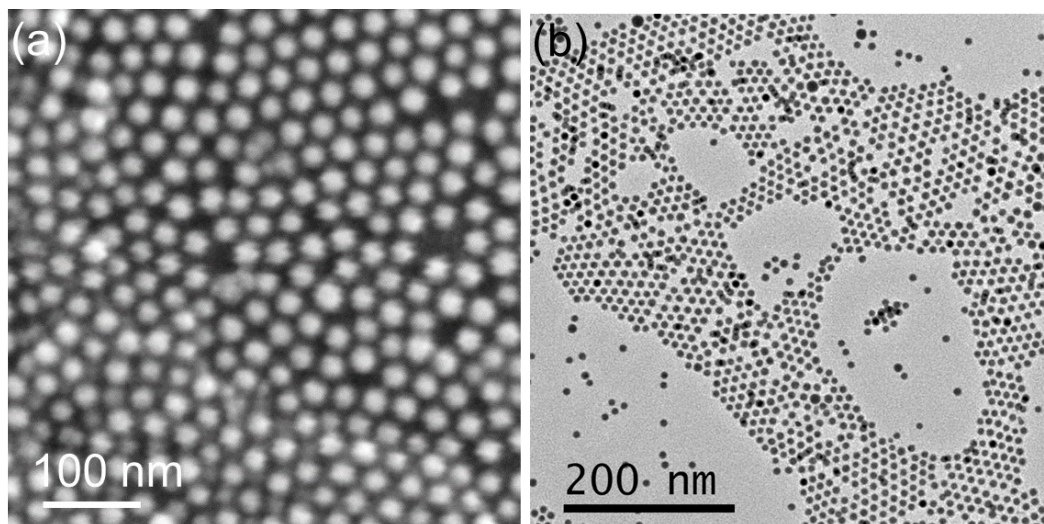


Fig. S6 (a) The GaPd₂ nanoparticles SEM image (b) TEM image

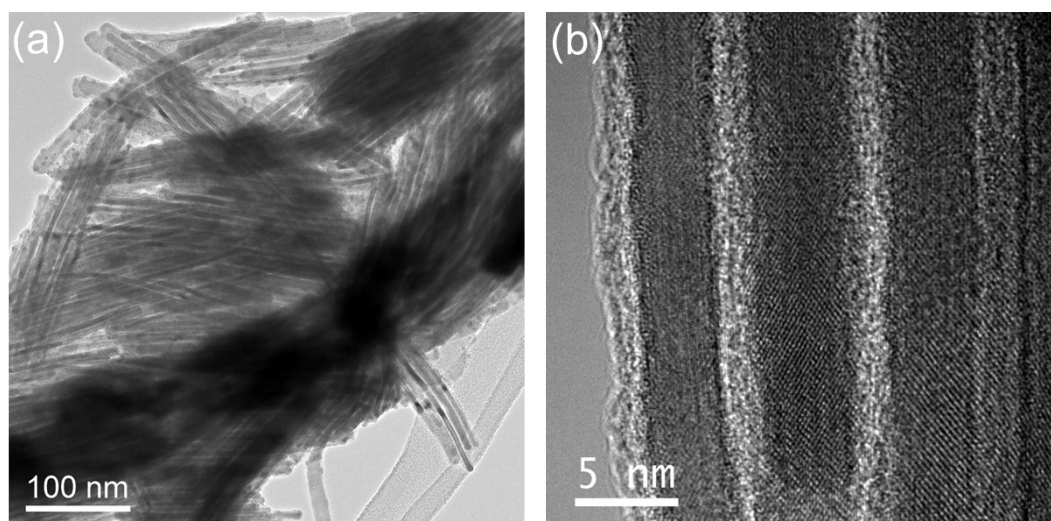


Fig. S7 (a-b) The GaPd₂ nanowires TEM images

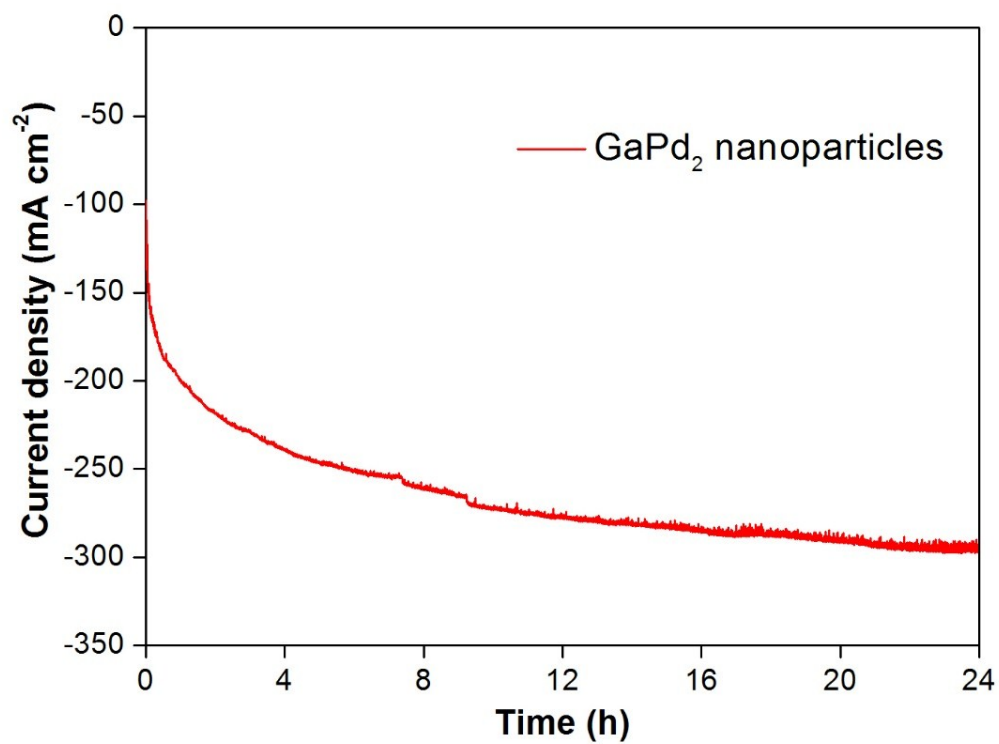


Fig. S8 The pre *i-t* testing of GaPd₂ nanoparticles with applied voltage of -0.5 V vs. RHE over 24 hours in 0.5 M H₂SO₄ electrolyte

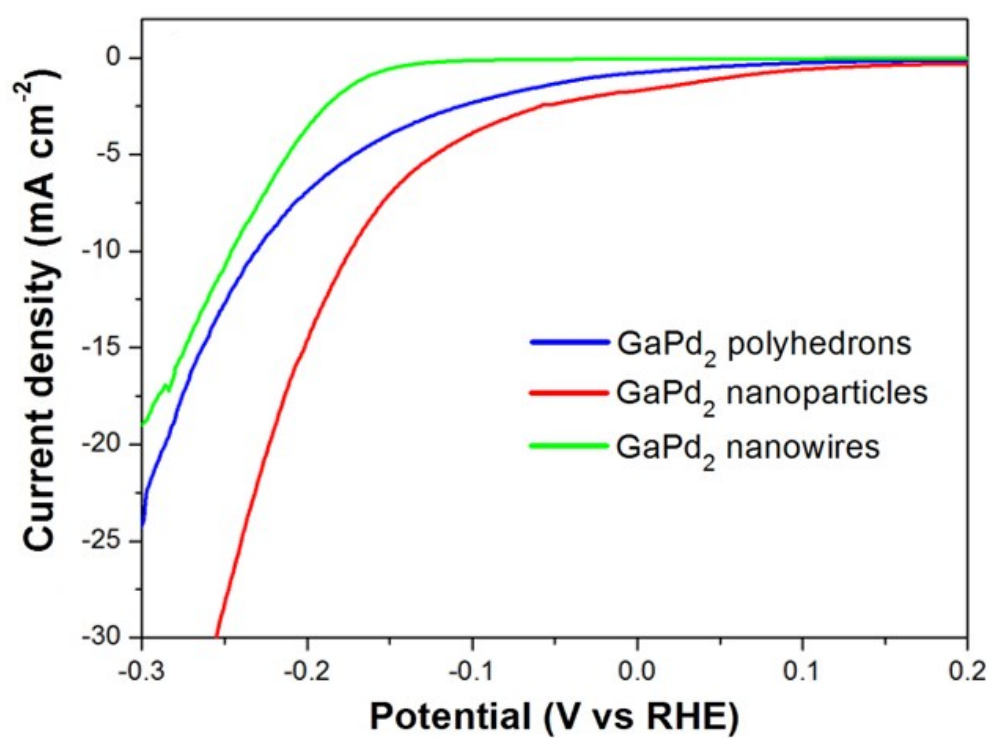


Fig. S9 Representative polarization curves of GaPd₂ polyhedrons, GaPd₂ nanoparticles and GaPd₂ nanowire working electrodes in 0.5 M H₂SO₄ solution before pre *i-t* testing for 24 hours

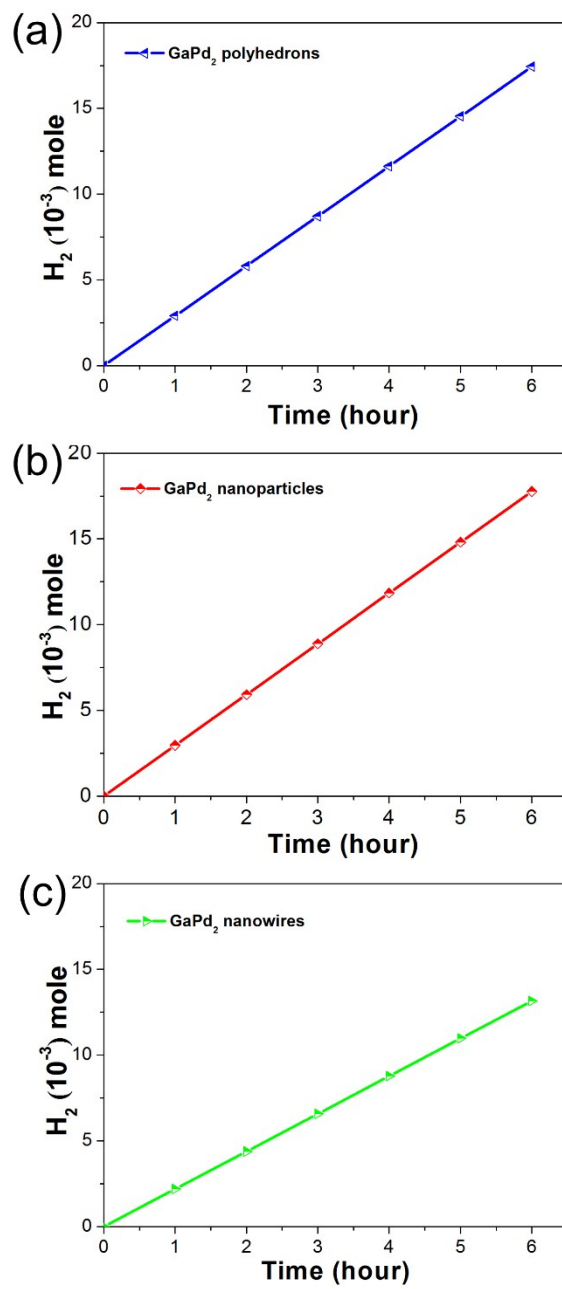


Fig. S10 Theoretical hydrogen production versus time for GaPd₂ (a) polyhedrons (b) nanoparticles (c) nanowires at a constant voltage of 800 mV for 6 hours”

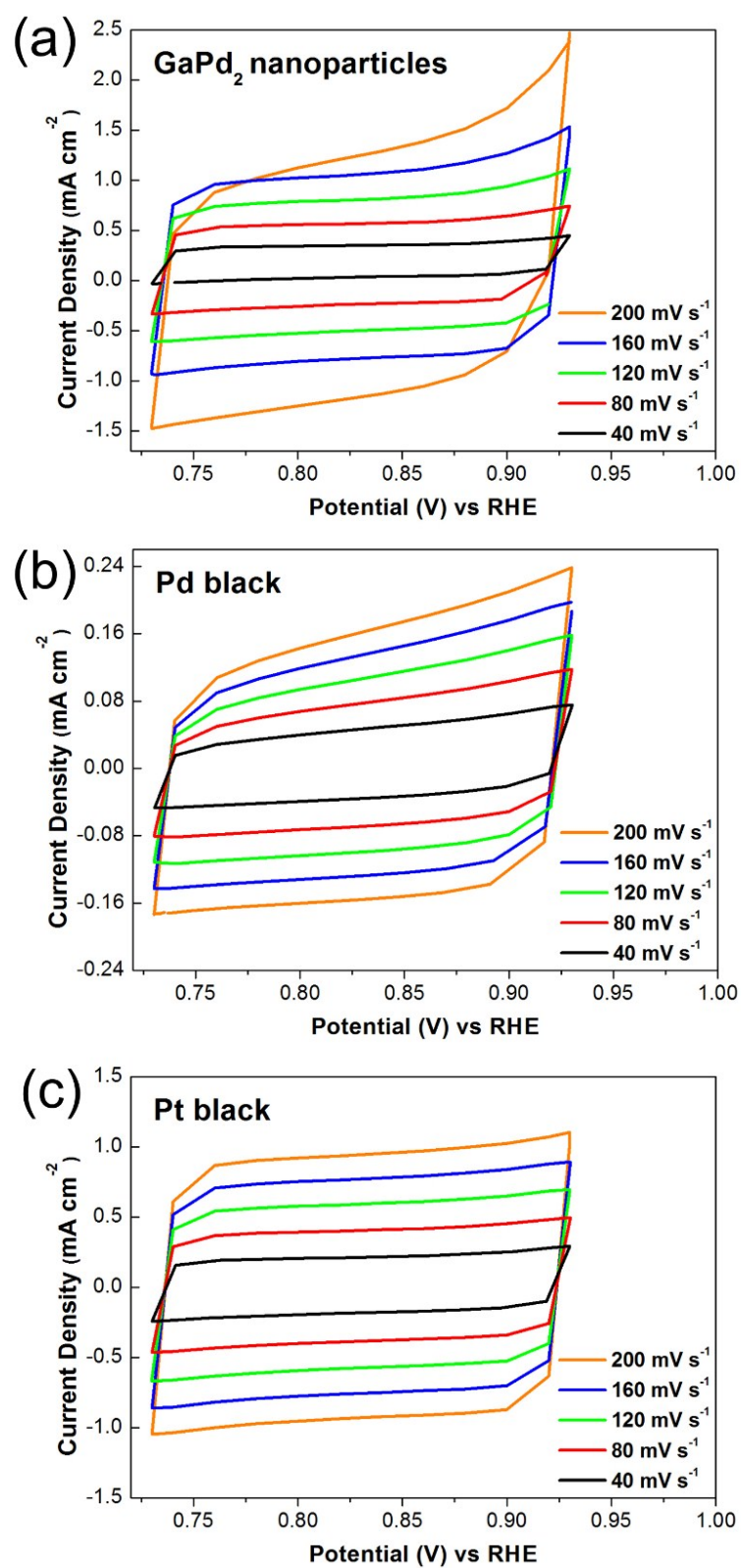


Fig. S11 (a) GaPd₂ (b) Pd black (c) Pt black at various scan rates (40, 80, 120, 160, 200 mV/s) in the region of 0.73-0.93 V vs RHE

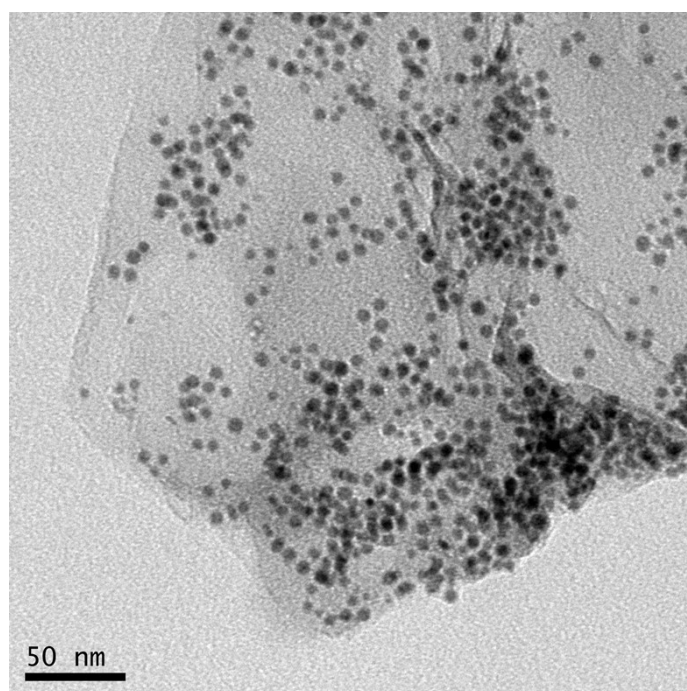


Fig. S12 Representative TEM image of Gapd₂ nanoparticles loaded on reduced graphene oxide (rGO).

Table S1 Comparison of HER activity for the GaPd₂ electrocatalyst with other recently reported Pd-based highly efficient electrocatalyst in acidic media.

Catalyst	Electrolyte	Mass loading ($\mu\text{g cm}^{-2}$)	Overpotential @10 mA/cm ² (mV)	Tafel slope (mV dec ⁻¹)	J ₀ (mA cm ⁻²)	IR correctio n	Ref.
GaPd ₂ polyhedrons			33.2	55.2	2.67		
GaPd ₂ nanoparticles	0.5 M H ₂ SO ₄	300	24.3	49	2.69	No	This work
GaPd ₂ nanowires			50.1	57.2	0.96		
GaPd ₂ /rGO			31.5	48.3	1.80		
PdNP@N-CNTs	0.5 M H ₂ SO ₄	285	46.8 ^a	33	--	--	1
Porous Pd NP assemblies	0.5 M H ₂ SO ₄	280	37.5 ^a	30	0.24	--	2
Pd/graphene	0.5 M H ₂ SO ₄	445	84	46	--	--	3
Ni@Pd ₄ :1/PEI-rGO _{50:1} (10 wt%)	0.5 M H ₂ SO ₄	140	59	54	3.91	--	4
AgPd	0.1 M H ₂ SO ₄	--	151 ^a	156	0.0126	--	5
porous Pd-CN _x	0.5 M H ₂ SO ₄	43(Pd)	55	35	0.4	--	6
Pd cube/PEI-rGO _{50:1} (10 wt. %)	0.5 M H ₂ SO ₄	140	38 ^a	34	2.29	Yes	7
PdMnCo/NC	0.5 M H ₂ SO ₄	285	39	31	--	--	8
PdCu	0.5 M H ₂ SO ₄	76.65	56	51	--	--	9
PdCu ₃	0.5 M H ₂ SO ₄	280	50 ^a	34	0.18	--	10
Pd ₇₁ Cu ₂₉ /C	0.1 M H ₂ SO ₄	100	75	48	0.38	Yes	11
Pd _{3.02} Te NWs /rGO	0.5 M H ₂ SO ₄	285	43 ^a	90	--	--	12
rGO-PdPS	0.5 M H ₂ SO ₄	250	90 ^a	46	--	--	13
Pd ₄ Se	0.5 M H ₂ SO ₄	--	94 ^a	50	0.23	Yes	14

Note: ^a The numerical value is calculated from the figure in the reference.

Reference

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