

SUPPLEMENTARY INFORMATION

Selected anthraquinones as potential free radical scavengers and P-glycoprotein inhibitors

Svetlana Jeremić^{1,*}, Ana Amić², Marijana Stanojević-Pirković³, Zoran Marković¹

¹Department of Chemical-Technological Sciences, State University of Novi Pazar, Vuka Karadžića bb,
36300 Novi Pazar, Serbia

²Department of Chemistry, Josip Juraj Strossmayer University of Osijek, Cara Hadrijana 8a,
31000 Osijek, Croatia

³Faculty of Medical Sciences, University of Kragujevac, Svetozara Markovića 69,
34000 Kragujevac, Serbia.

* Corresponding author. Tel.: 00381 20 317 754; fax: 00381 20 337 669.

E-mail address: sjeremic@np.ac.rs (S. Jeremić)

Table S1. The PCM/SMD/M06-2X/6-311++G(d,p) reaction enthalpies and Gibbs free energies (in kJ mol⁻¹) for HAT mechanism of free radical formation from corresponding neutral molecules in polar and nonpolar environment.

Free radical	Benzene		Pentyl ethanoate		DMSO		Water	
	$\Delta_r H_{\text{BDE}}$	$\Delta_r G_{\text{BDE}}$						
·OH	489.55	458.51	490.53	459.49	491.87	460.83	503.56	472.53
(CH ₃) ₃ C–O·	437.57	403.02	437.77	403.30	438.53	404.13	451.87	417.12
·OCH ₃	429.77	395.80	429.48	395.54	429.72	395.81	443.43	409.55
CCl ₃ –O–O·	386.12	351.58	387.31	352.64	388.34	353.49	405.83	370.90
PhO·	369.19	333.80	368.88	333.46	369.01	333.56	377.19	341.78
·OOH	359.06	324.52	358.19	325.32	357.11	323.97	372.65	339.94
CH ₂ =CH–O–O·	354.33	322.13	353.25	322.38	352.51	321.43	370.09	339.05
CH ₃ –O–O·	351.38	318.95	349.40	315.67	348.26	314.66	366.53	332.68
CH ₂ =CH–CH ₂ –O–O·	350.35	317.43	349.19	316.05	348.05	314.39	366.59	332.54
O ₂ · ⁻	257.54	229.94	259.17	231.55	260.49	232.85	298.06	270.38

Figure S1. Numerical values of spin density distribution (SDD) for the most stable radicals of investigated anthraquinones, calculated in pentyl ethanoate.

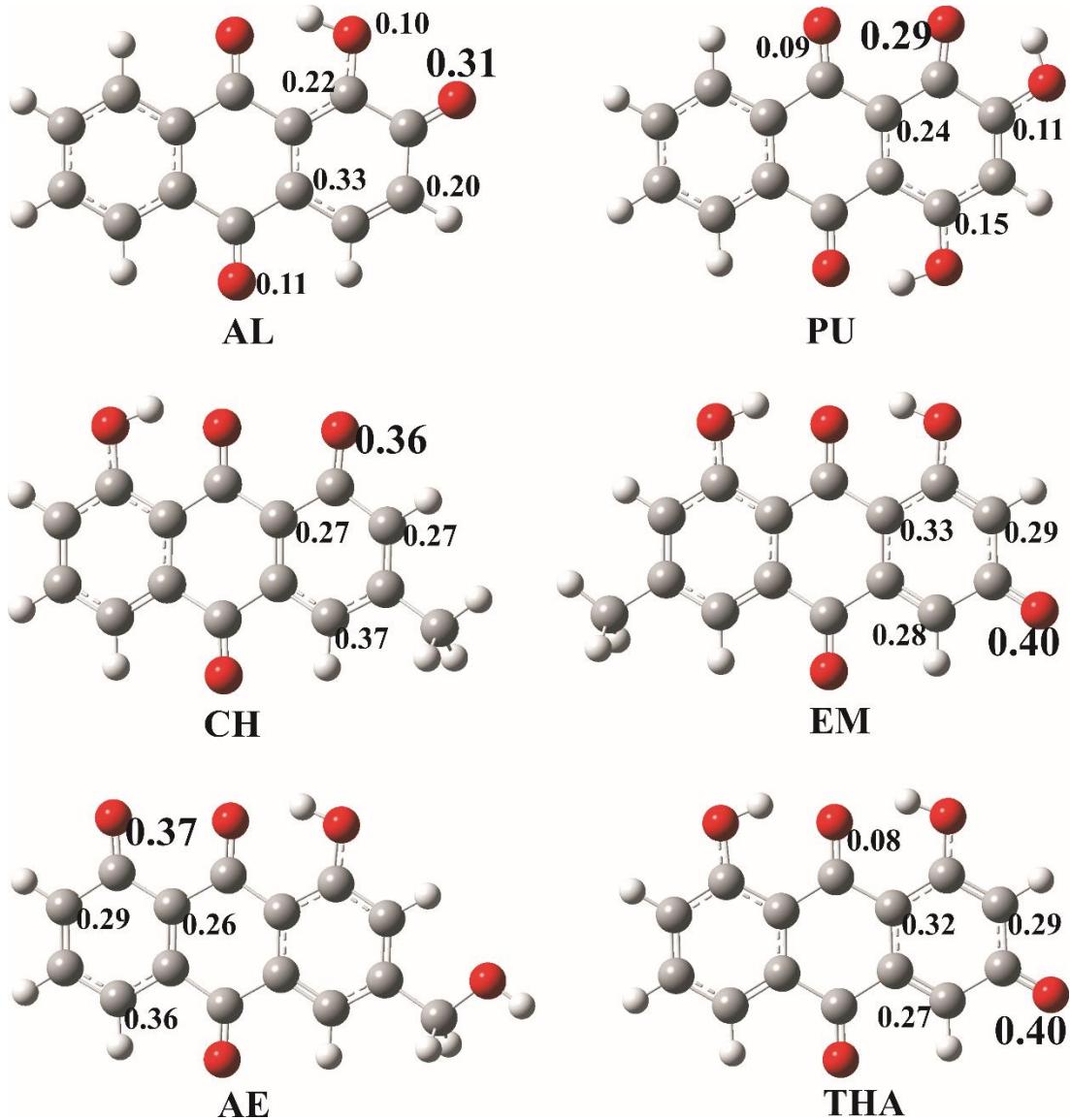


Figure S2. Optimized structures of four selected anthracycline and anthracene drugs: doxorubicin (DX), daunorubicin (DA), mitoxantrone (MX) and bisantrene (BA).

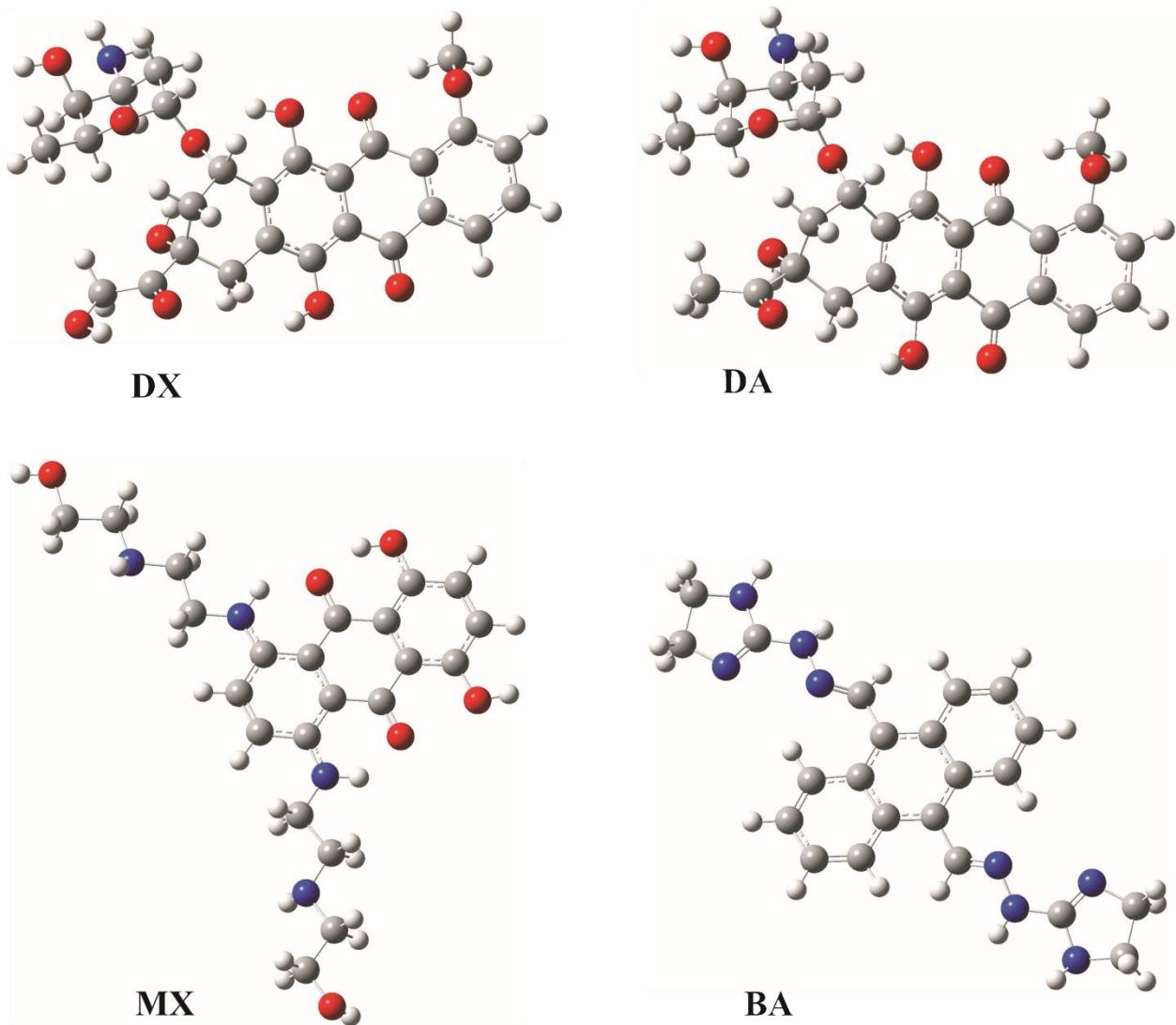


Table S2. Results of docking analysis of monoanions of AL, PU, CH, AE and TA, and selected anticancer drugs DX, DA and BA, in reaction with P-glycoprotein: type of antraquinone monoanion / anticancer drug with its role in bond formation; corresponding amino acid with its position in the primary structure of protein, and with its role in bond formation; type of interaction during bond formation; distance in Å between respective active cites of ligand and amino acids; pairwise interaction energy (E_i) in kcal mol⁻¹; estimated total free energy of binding (E_b) in kcal mol⁻¹, estimated inhibition constant (K_i) in µM.

Ligand		Amino acid	Type of interaction	Atom distance	E_i	E_b	K_i	
AL ⁻	π-orbitals electron donor	Lys 884	Alkyl group is electron acceptor	π – alkyl hydrophobic	4.474	0.02	-4.68	372.34
	H-donor	Glu 885	H-acceptor	Conventional hydrogen bond	2.134	0.60		
	π-orbitals electron acceptor	Glu 885	C-H bond is electron donor	π – σ hydrophobic	3.140	0.06		
	H-acceptor	Gly 888	H-donor	Conventional hydrogen bond	2.422	0.37		
	π-orbitals electron donor	Pro 923	Alkyl group is electron acceptor	π – alkyl hydrophobic	4.481	0.02		
PU ⁻	π-orbitals electron donor	Lys 884	Alkyl group is electron acceptor	π – alkyl hydrophobic	5.395	0.01	-4.67	376.63
	H-donor	Glu 885	H-acceptor	Conventional hydrogen bond	2.003	0.60		
	H-acceptor	Gly 888	H-donor	Conventional hydrogen bond	2.509	0.27		
	H-donor	Ser 919	H-acceptor	Conventional hydrogen bond	2.159	0.58		
	π-orbitals electron donor	Pro 923	Alkyl group is electron acceptor	π – alkyl hydrophobic	4.519	0.02		
CH ⁻	Alkyl group is electron donor	Lys 884	Alkyl group is electron acceptor	alkyl – alkyl hydrophobic	4.584	0.02	-4.89	261.11
	H-donor	Glu 885	H-acceptor	Conventional hydrogen bond	1.965	0.60		

	π -orbitals electron acceptor	Glu 885	C-H bond is electron donor	$\pi - \sigma$ hydrophobic	3.160	0.03		
	H-acceptor	Gly 888	H-donor	Conventional hydrogen bond	2.422	0.37		
	π -orbitals electron donor	Pro 923	Alkyl group is electron acceptor	$\pi - \text{alkyl}$ hydrophobic	4.353	0.02		
AE⁻	H-acceptor	Lys 884	H-donor	Carbon hydrogen bond	3.701	0.03	-4.61	414.40
	H-donor	Glu 885	H-acceptor	Conventional hydrogen bond	1.731	0.60		
	π -orbitals electron acceptor	Glu 885	C-H bond is electron donor	$\pi - \sigma$ hydrophobic	3.483	0.02		
	H-acceptor	Gly 888	H-donor	Conventional hydrogen bond	2.278	0.48		
	H-donor	Ser 919	H-acceptor	Conventional hydrogen bond	1.960	0.60		
	π -orbitals electron donor	Pro 923	Alkyl group is electron acceptor	$\pi - \text{alkyl}$ hydrophobic	4.867	0.02		
TA⁻	H-donor	Lys 884	H-acceptor	Conventional hydrogen bond	2.085	0.60	-4.81	298.24
	π -orbitals electron donor	Lys 884	Alkyl group is electron acceptor	$\pi - \text{alkyl}$ hydrophobic	4.887	0.02		
	H-donor	Glu 885	H-acceptor	Conventional hydrogen bond	2.029	0.60		
	π -orbitals electron acceptor	Glu 885	C-H bond is electron donor	$\pi - \sigma$ hydrophobic	3.175	0.03		
	H-acceptor	Gly 888	H-donor	Conventional hydrogen bond	2.325	0.48		
	H-acceptor	Pro 923	H-donor	Carbon hydrogen bond	3.124	0.02		
DX	Alkyl group is electron donor	Lys 884	Alkyl group is electron acceptor	alkyl – alkyl hydrophobic	4.528	0.02	-3.02	6.12 · 10³
	H-donor	Glu 887	H-acceptor	Conventional hydrogen bond	2.434	0.37		

	H-donor	Glu 887	H-acceptor	Conventional hydrogen bond	2.110	0.60		
	H-acceptor	Ser 919	H-donor	Carbon hydrogen bond	2.929	0.01		
	π -orbitals electron acceptor	Ile 922	C-H bond is electron donor	$\pi - \sigma$ hydrophobic	3.813	0.02		
	π -orbitals electron acceptor	Ile 922	C-H bond is electron donor	$\pi - \sigma$ hydrophobic	3.946	0.02		
	π -orbitals electron donor	Ile 922	Alkyl group is electron acceptor	$\pi - \text{alkyl}$ hydrophobic	5.367	0.01		
	π -orbitals electron donor	Pro 923	Alkyl group is electron acceptor	$\pi - \text{alkyl}$ hydrophobic	5.493	0.01		
DA	Alkyl group is electron donor	Lys 884	Alkyl group is electron acceptor	alkyl – alkyl hydrophobic	4.260	0.02	-3.95	$1.26 \cdot 10^3$
	H-donor	Glu 887	H-acceptor	Conventional hydrogen bond	2.233	0.58		
	H-donor	Glu 887	H-acceptor	Conventional hydrogen bond	1.957	0.60		
	π -orbitals electron acceptor	Ile 922	C-H bond is electron donor	$\pi - \sigma$ hydrophobic	3.452	0.02		
	π -orbitals electron acceptor	Ile 922	C-H bond is electron donor	$\pi - \sigma$ hydrophobic	3.258	0.01		
	π -orbitals electron donor	Ile 922	Alkyl group is electron acceptor	$\pi - \text{alkyl}$ hydrophobic	5.354	0.01		
BA	H-donor	Lys 884	H-acceptor	Carbon hydrogen bond	3.095	0.02	-5.08	187.80
	π -orbitals electron donor	Lys 884	Alkyl group is electron acceptor	$\pi - \text{alkyl}$ hydrophobic	5.289	0.01		
	π -orbitals electron acceptor	Glu 885	C-H bond is electron donor	$\pi - \sigma$ hydrophobic	3.176	0.03		
	H-donor	Glu 887	H-acceptor	Conventional hydrogen bond	1.750	0.60		

H-acceptor	Gly 888	H-donor	Carbon hydrogen bond	3.344	0.02
π -orbitals electron donor	Pro 923	Alkyl group is electron acceptor	π – alkyl hydrophobic	4.984	0.01

Figure S3. Docking positions of anions of alizarin (a), purpurin (b), chrysophanol (c), aloe-emodin (d) and 1,3,8- trihydroxyanthraquinone (e) with P-glycoprotein.

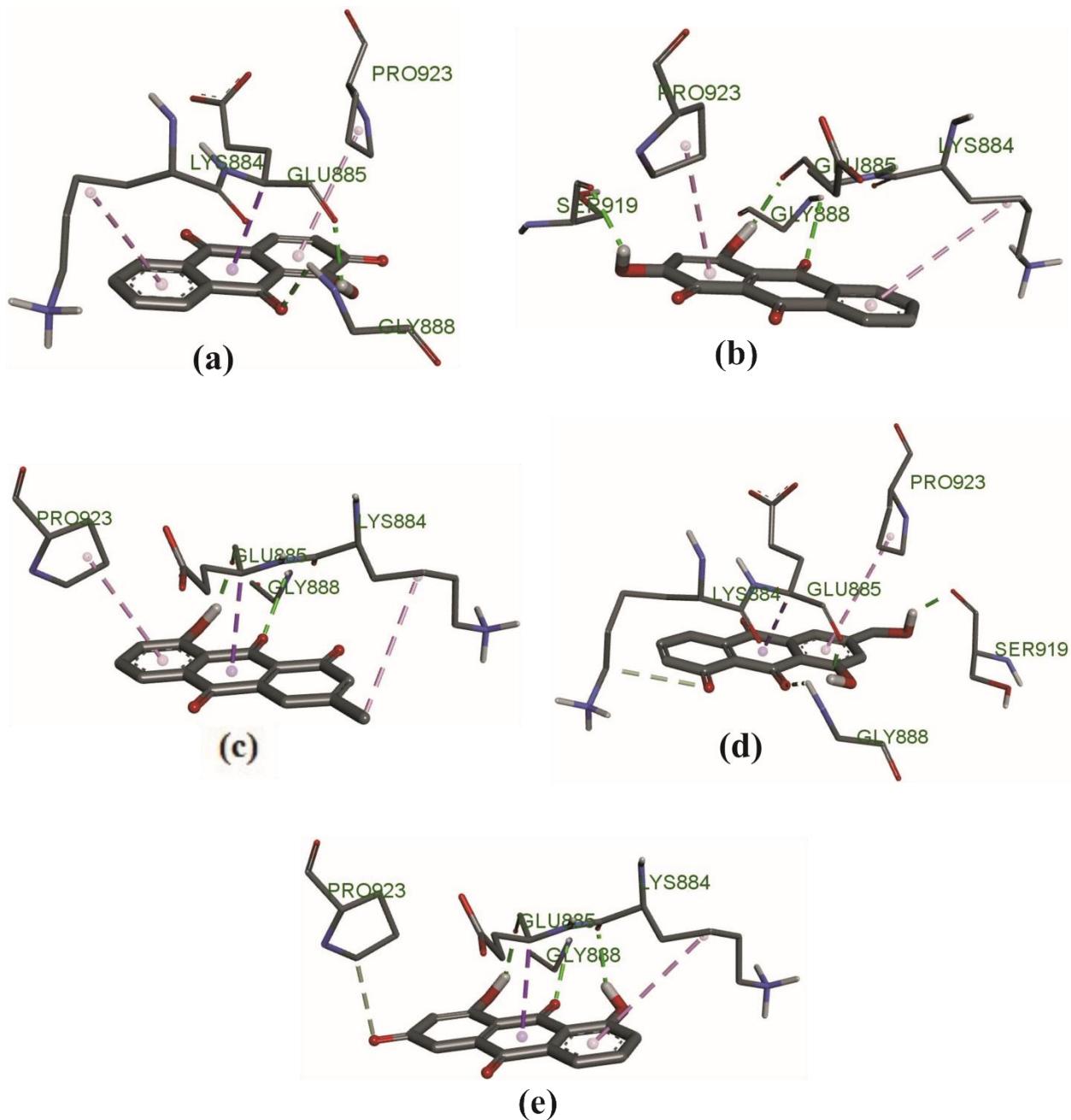


Figure S4. Docking positions of doxorubicin (a), daunorubicin (b) and bisantrene (c) with P-glycoprotein.

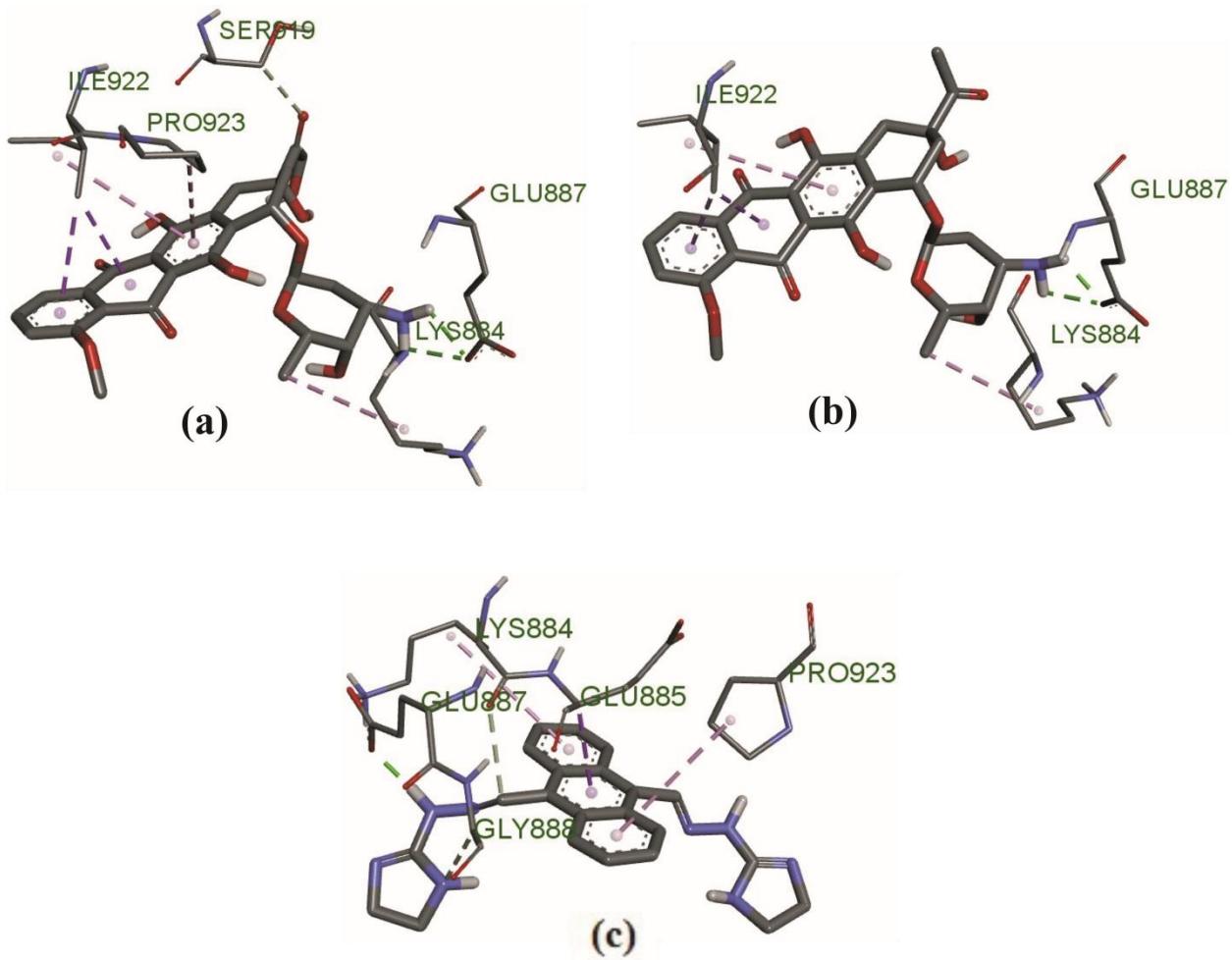


Figure S5. Position of docking of emodin-anion (rounded) with P-glycoprotein tertiary structure.

