

Chemo-enzymatic synthesis of (*E*)-2,3-diaryl-5-styryl-*trans*-2,3-dihydrobenzofuran-based scaffolds and their *in vitro* and *in silico* evaluation as a novel sub-family of potential allosteric modulators of the 90 kDa heat shock protein (Hsp90)

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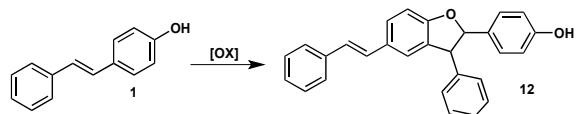
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- ^b Università degli Studi di Milano, Dipartimento di Scienze Farmaceutiche, via Mangiagalli 25, Milano, 20133, Italy;
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SUPPORTING INFORMATION

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Metal-mediated oxidation of compound 1



The metal-mediated oxidations were conducted following published procedures on analogues compounds.¹ A water solution of the metal oxidant (4 M, 5 equiv.) was added to a ethanolic solution of the un-substituted (*E*)-4-hydroxystilbene **1** (50 mM). The resulting mixture was stirred at 35 °C for 24 - 48 h, monitoring the reaction by TLC analysis (CHCl₃ : acetone = 9 : 1)

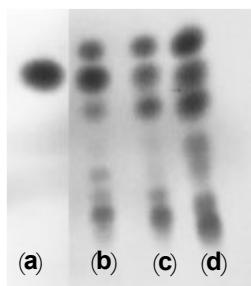


Figure S1: TLC plate of the reaction outcome of the CuCl₂ and FeCl₃-mediated oxidation of **1**: (a) standard of **1**; (b) overnight reaction with CuCl₂; (c) overnight and (d) 48 h reaction with FeCl₃.

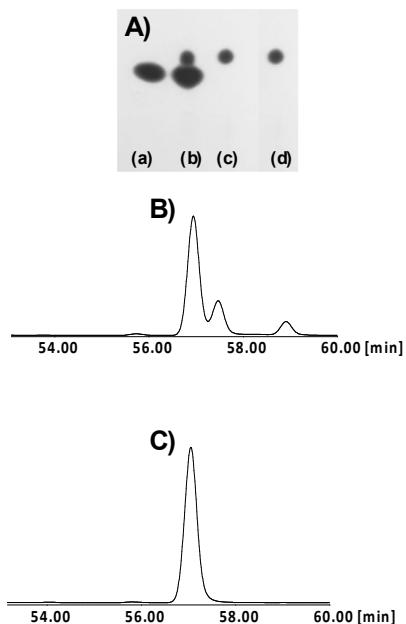
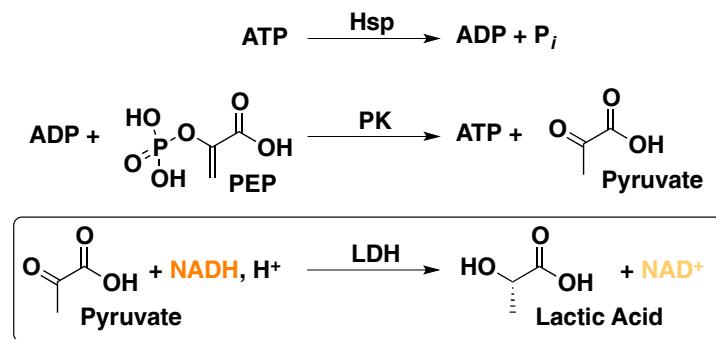


Figure S1b. A): TLC plate of: (a) standard of **1**; (b) co-chromatography of **a** and **c**; (c) MnO₂-mediated oxidation and (d) laccase-mediated oxidation after 24 h. **B)** and **C)**: RP-HPLC chromatograms of the oxidation outcomes of **1** with MnO₂ and laccase, respectively.

¹ Takaya, Y.; Terashima, K.; Ito, J.; He, Y. H.; Tateoka, M.; Yamaguchi, N.; Niwa, M. *Tetrahedron* **2005**, *61* (43), 10285–10290

The oxidation with manganese(IV) oxide, run under the same experimental conditions, gave better results. The reaction proceeded smoothly and, as shown in **Figure 1Sb**, a single spot could be detected by TLC. The product(s) was isolated and its mass spectra indicated the formation of dimer(s). Unfortunately, the ¹H-NMR spectrum showed the presence of an inseparable mixture of oxidized regioisomeric compounds, as further confirmed by reverse-phase HPLC analysis (**Figure S1b**).

In vitro assay



Scheme S1. Enzymatic cascade reactions to evaluate by spectrophotometer the ATPase activity of the Hsp90 yeast homologous, Hsc82: the ADP produced by ATP hydrolysis during the chaperone cycle reacts with PEP (phosphoenolpyruvate) to form pyruvate in a reaction catalyzed by the enzyme pyruvate kinase (PK), whereby ATP is regenerated. Pyruvate then continues the cascade process and is reduced to lactic acid by action of lactate dehydrogenase (LDH), whereby NADH is oxidized to NAD⁺.

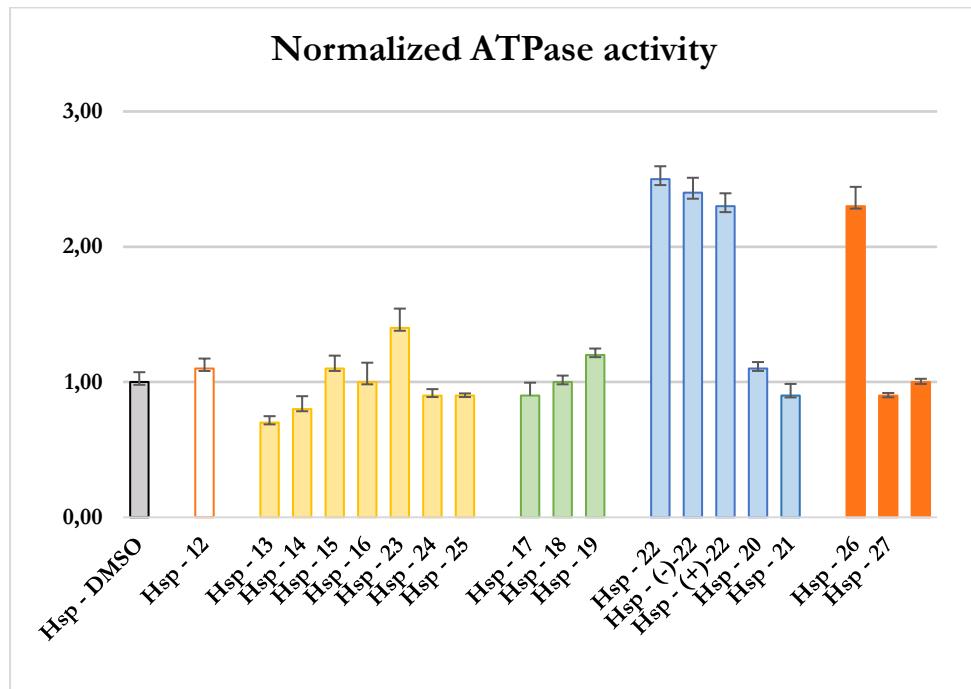
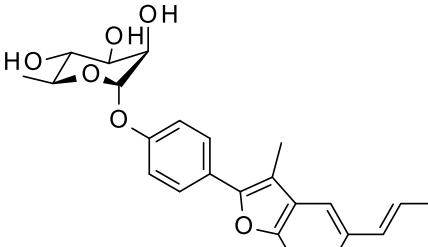
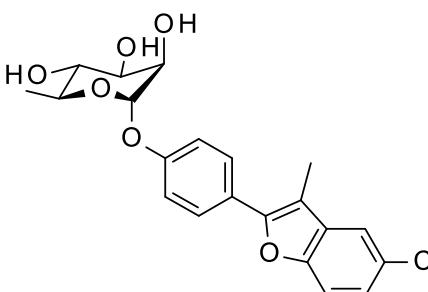
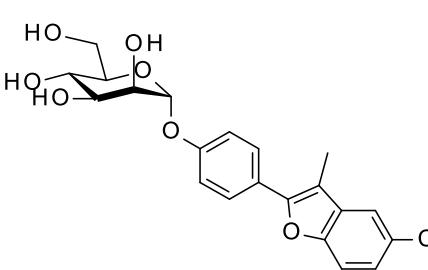
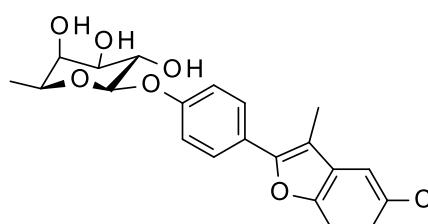
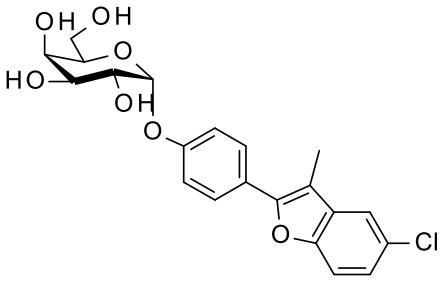
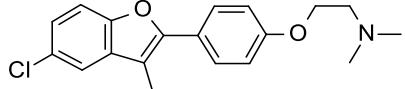
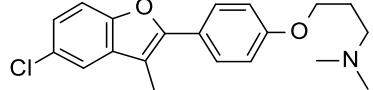
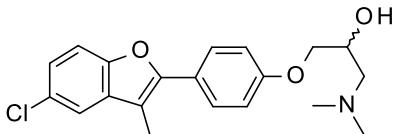
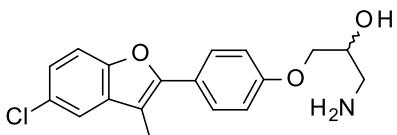
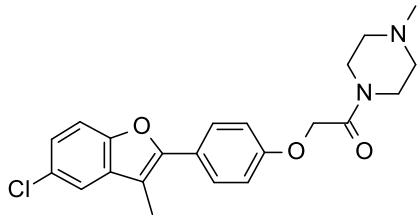
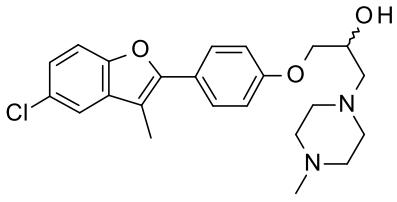
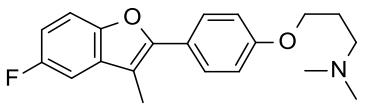
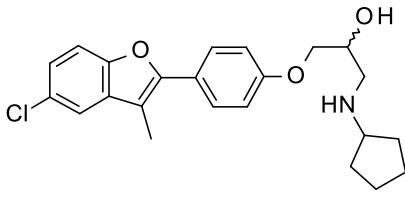
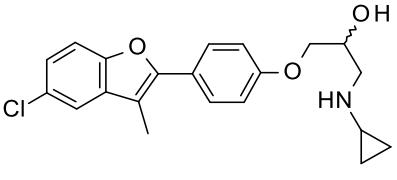
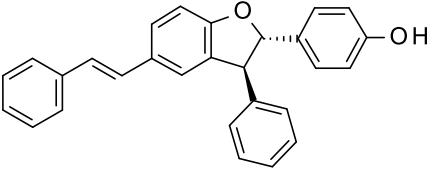
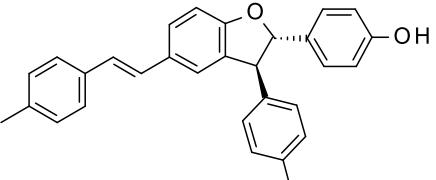


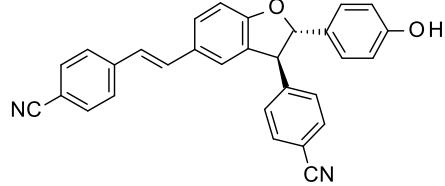
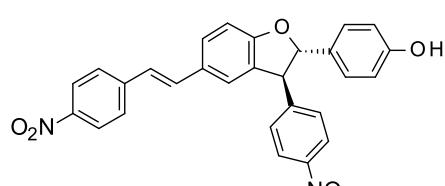
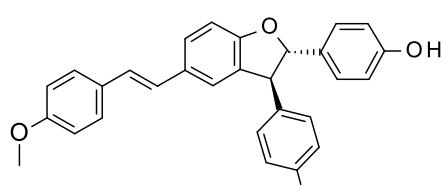
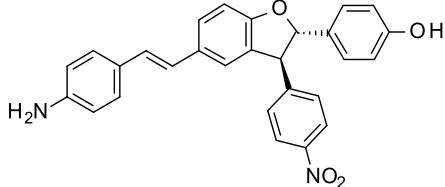
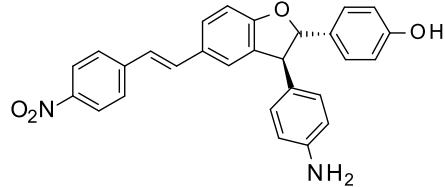
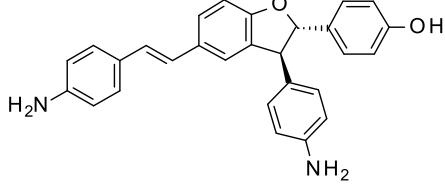
Figure S2: Normalized ATPase activities of Hsp incubated with all the different synthesized compounds.

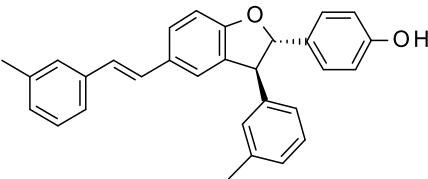
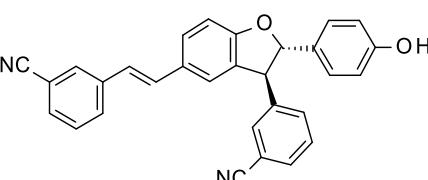
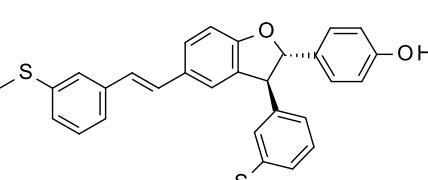
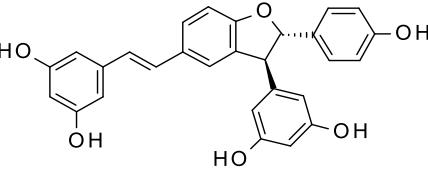
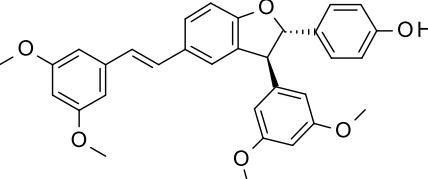
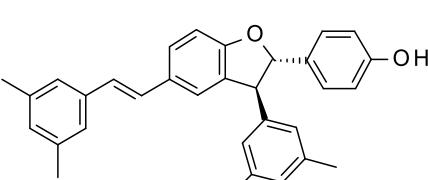
Table S1: Structures and activity data of Hsp90 modulators

Normalized Hsp90's ATPase cycle activation	Ligand Efficiency	Compound, Structure
1.24	-0.26	CheCosp-1 
2.27	-0.27	CheCosp-9 
2.74	-0.26	CheCosp-16 
1.17	-0.26	CheCosp-21 

2.35	-0.26	
4.01	-0.32	
5.3	-0.30	
4.15	-0.30	
4.52	-0.28	
1.86	-0.35	

2.39	-0.24	CheCosp-36 
4.24	-0.27	CheCosp-57 
2.23	-0.26	CheCosp-43 
1.55	-0.30	CheCosp-42 
1.1	-0.19	Compound (\pm)-12 
0.7	-0.17	Compound (\pm)-13 

0.8	-0.13	Compound (\pm)-14 
1.1	-0.23	Compound (\pm)-15 
1.0	-0.16	Compound (\pm)-16 
1.0	-0.17	Compound (\pm)-24 
0.90	-0.14	Compound (\pm)-25 
1.4	-0.18	Compound (\pm)-23 

0.9	-0.18	Compound (\pm)-17 
1.0	-0.18	Compound (\pm)-18 
1.2	-0.18	Compound (\pm)-19 
2.5	-0.25	Compound (\pm)-22 
1.1	-0.16	Compound (\pm)-20 
0.9	-0.17	Compound (\pm)-21 

2.3	-0.18	Compound (\pm)-26 <chem>O=C1[C@H](C2=CC=C(C=C2)[C@H]1c3ccccc3)C(OCCN(C)C)c4ccc(O)cc4</chem>
0.9	-0.16	Compound (\pm)-27 <chem>O=C1[C@H](C2=CC=C(C=C2)[C@H]1c3ccccc3)C(OCCN(C)C)c4ccc(O)c5ccc(O)c45</chem>
1	-0.15	Compound (\pm)-28 <chem>O=C1[C@H](C2=CC=C(C=C2)[C@H]1c3ccccc3)C(OCCN(C)C)c4ccc(C)c5ccc(C)c45</chem>

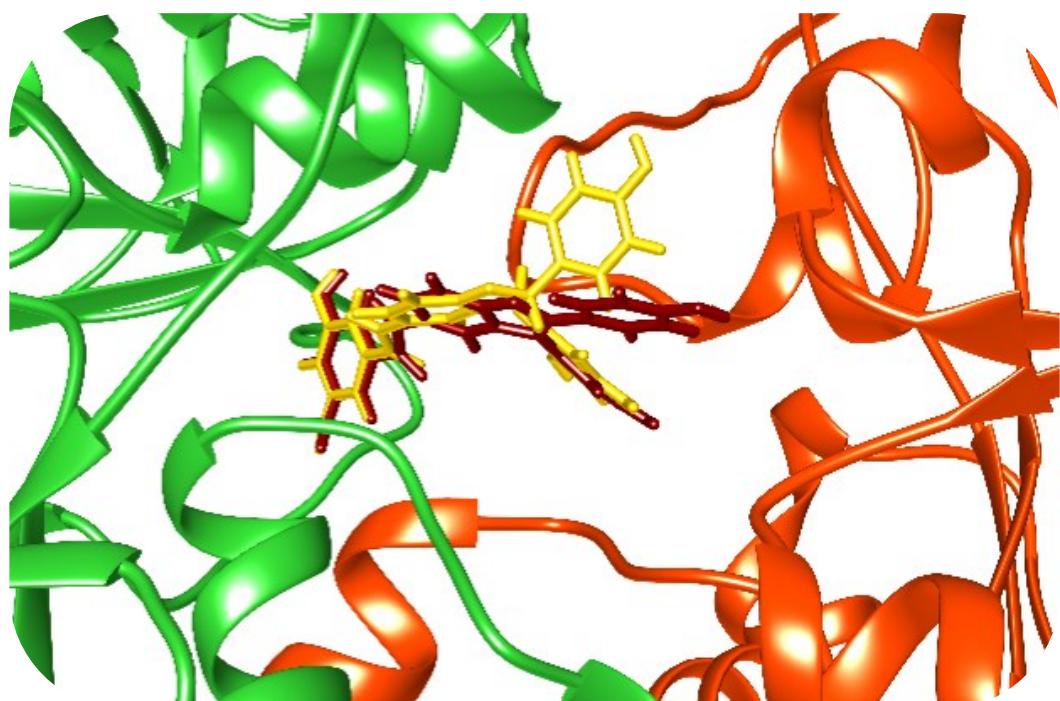
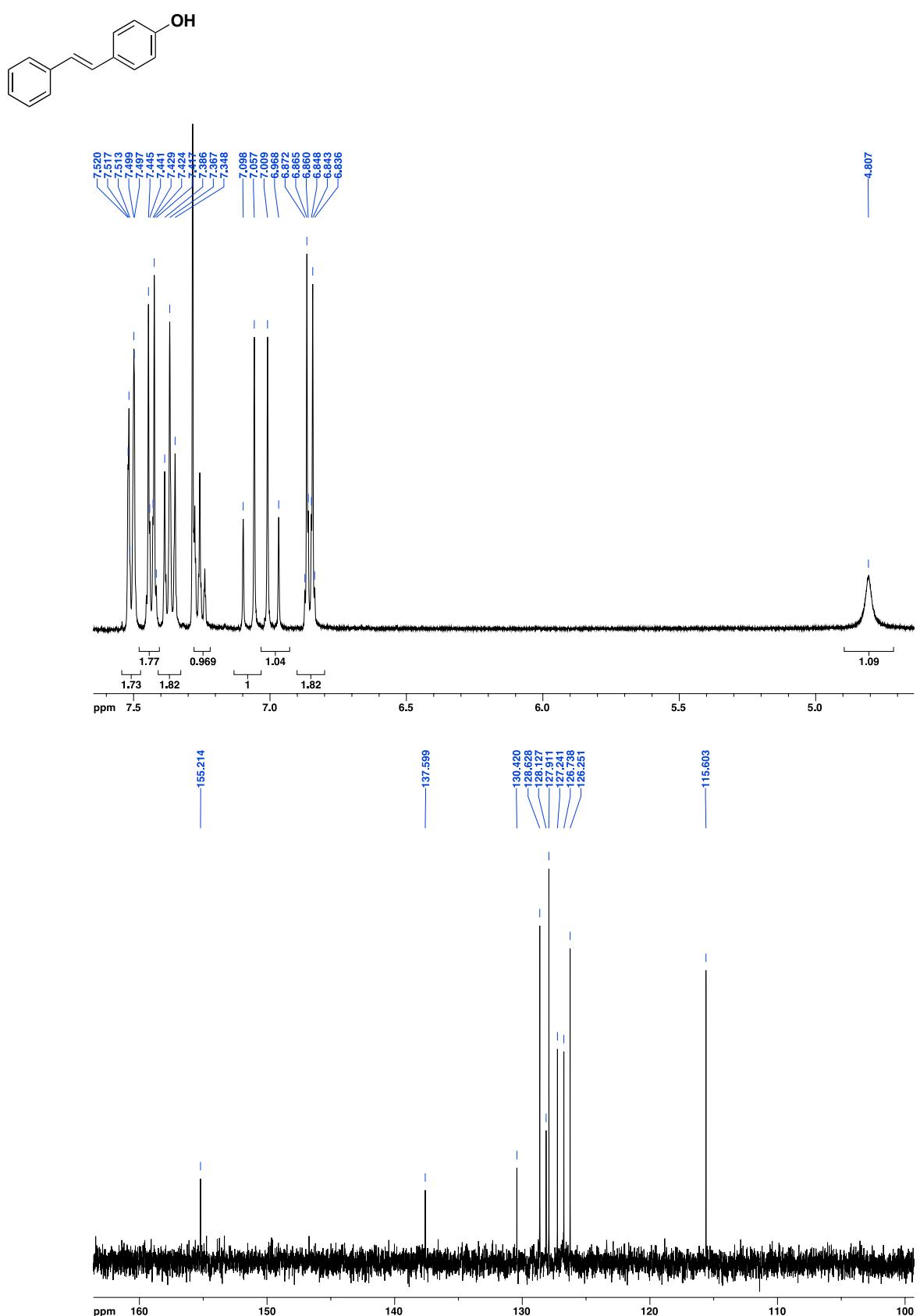


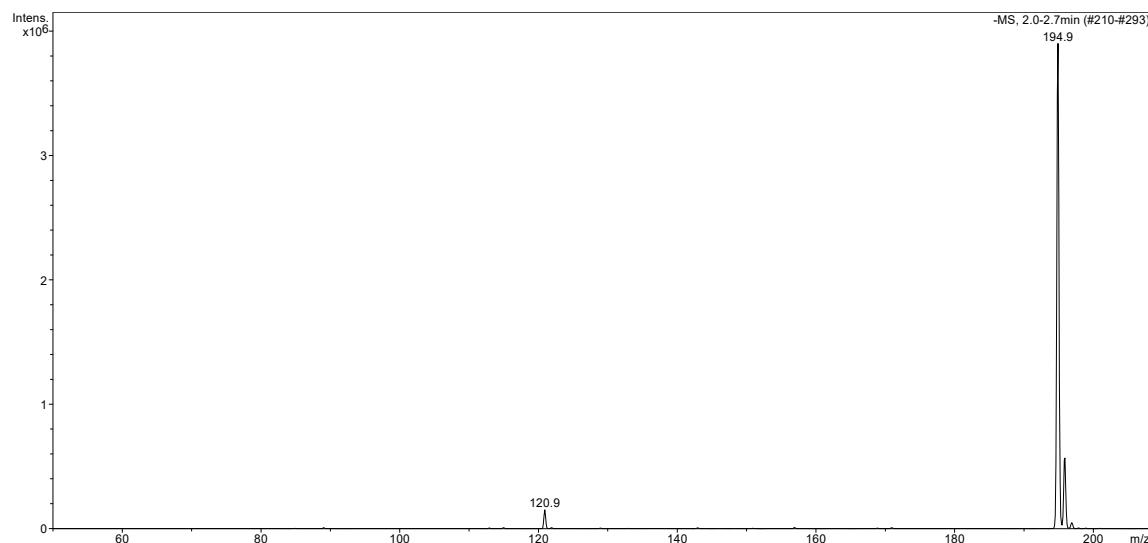
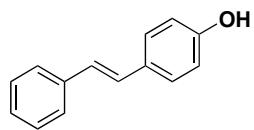
Figure S3: Superposition of the best pose for the two enantiomers (*R*) and (*S*) of compound 22 shown in red and yellow, respectively. Protomers A and B of Hsp90 are depicted in green and orange, respectively.

¹H, ¹³C-NMR and mass spectra

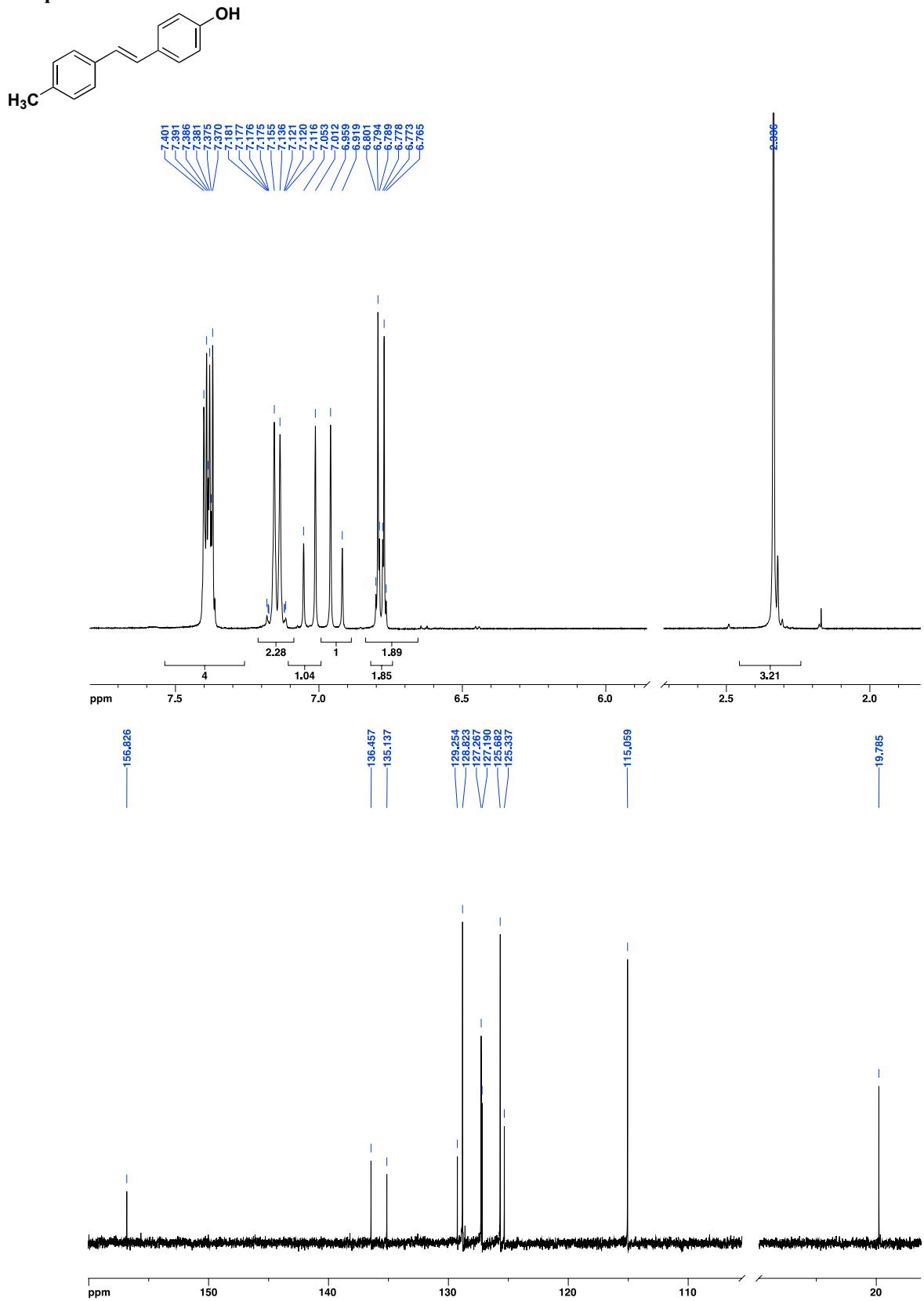
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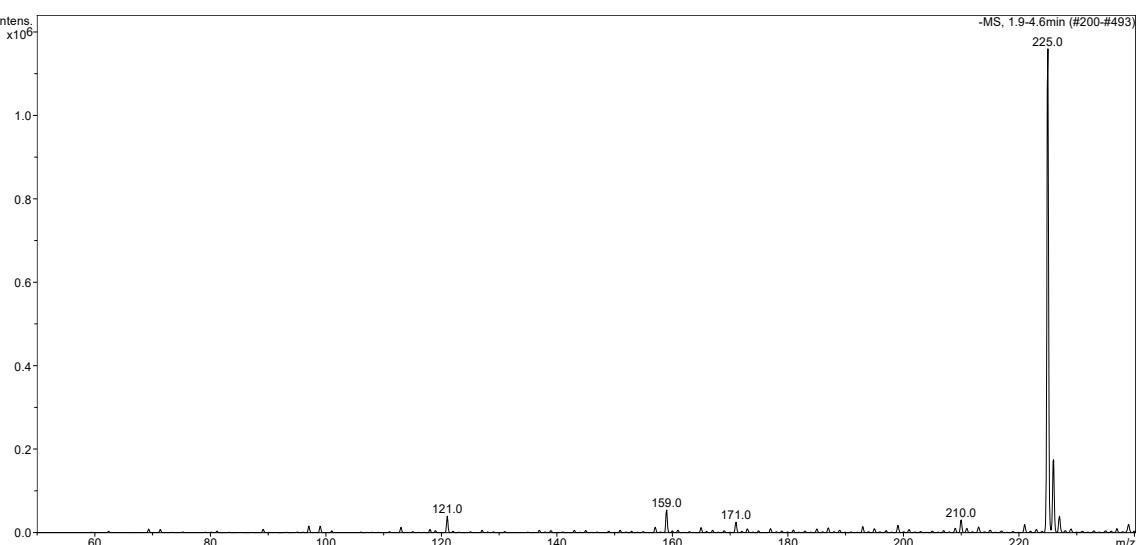
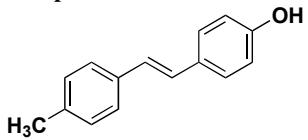
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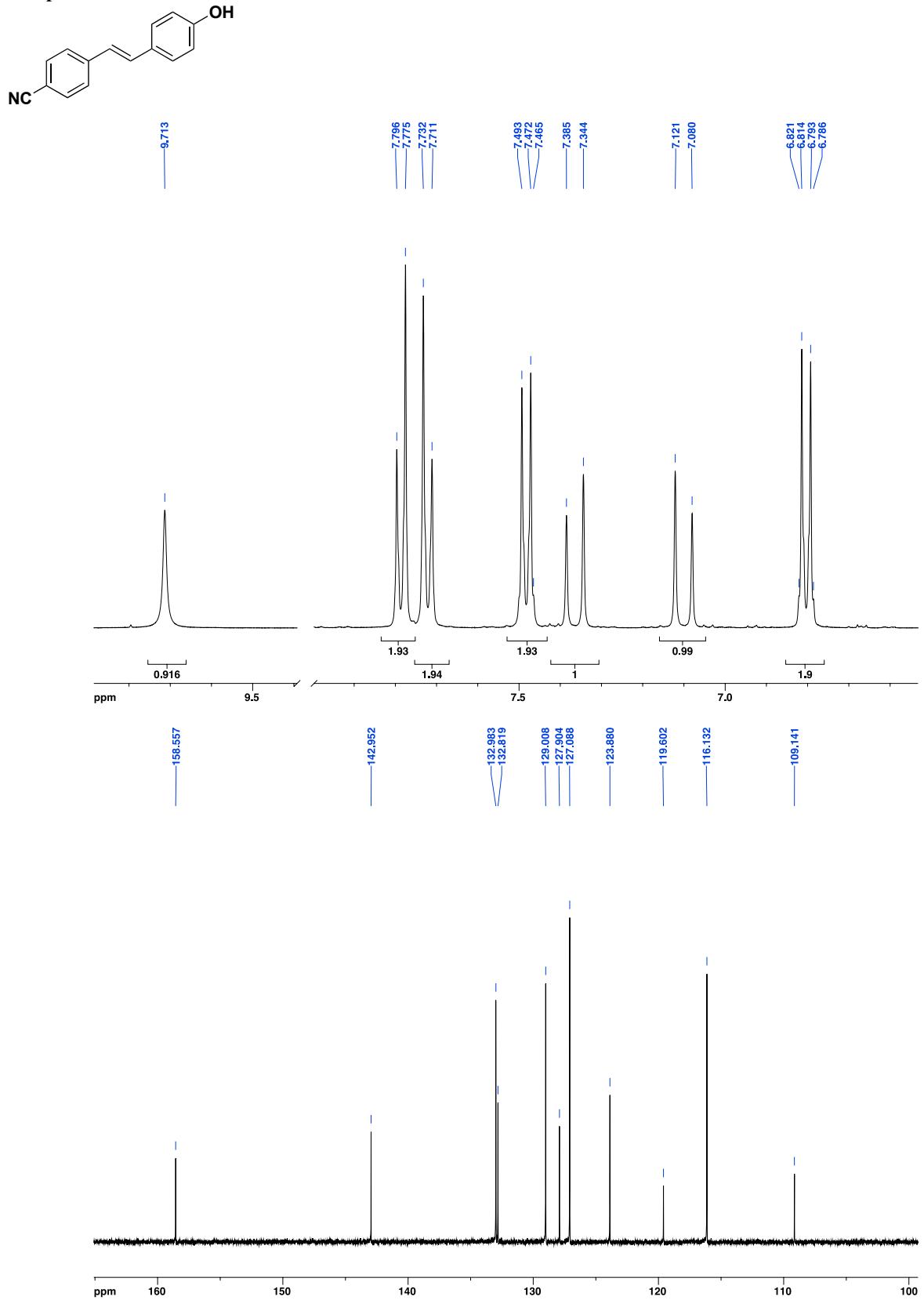
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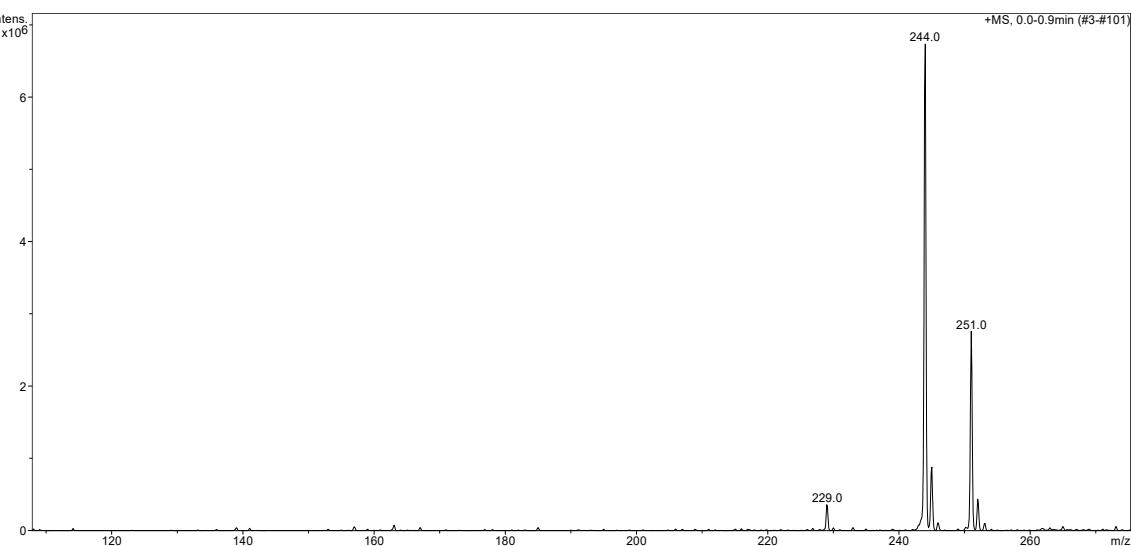
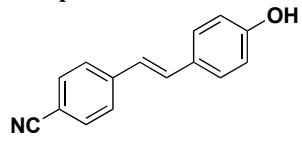
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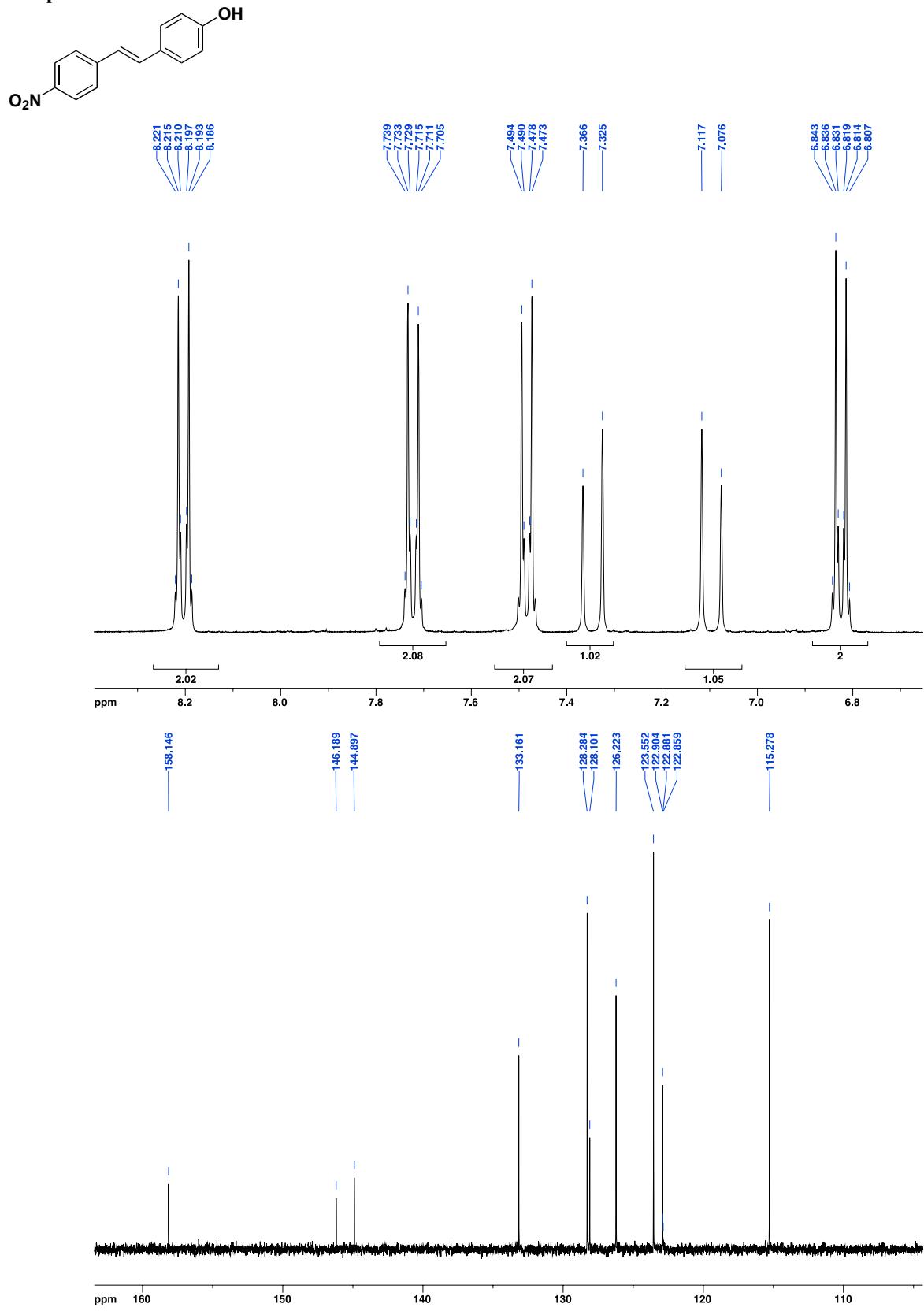
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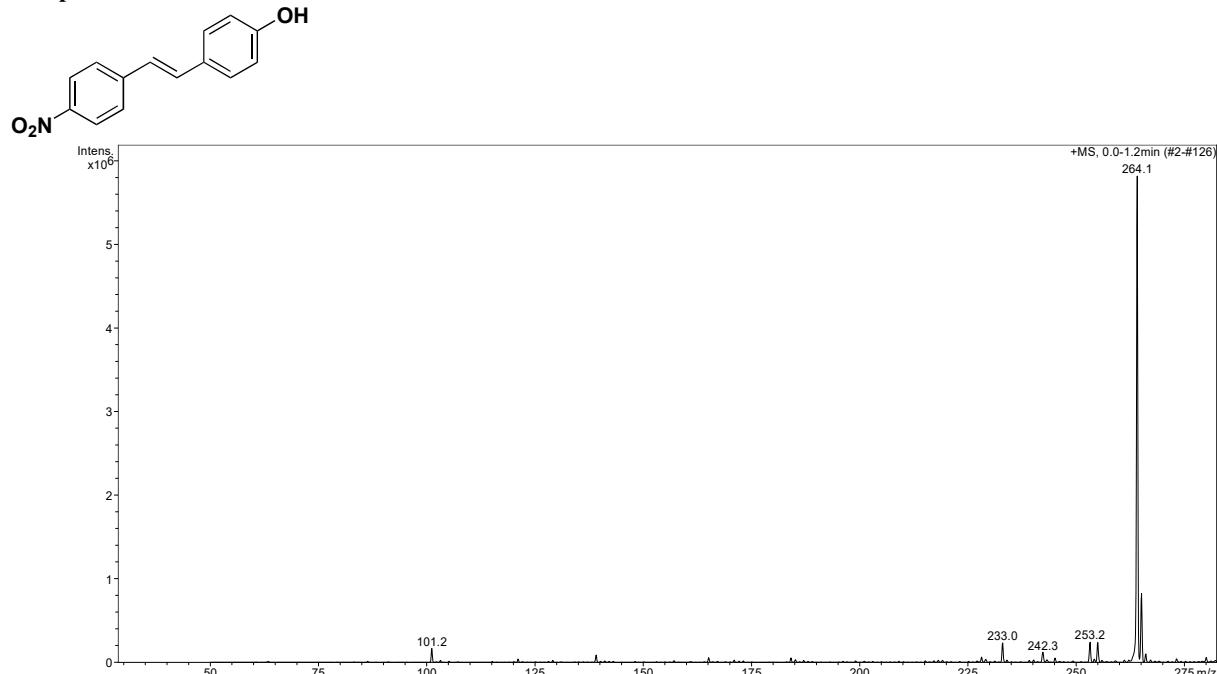
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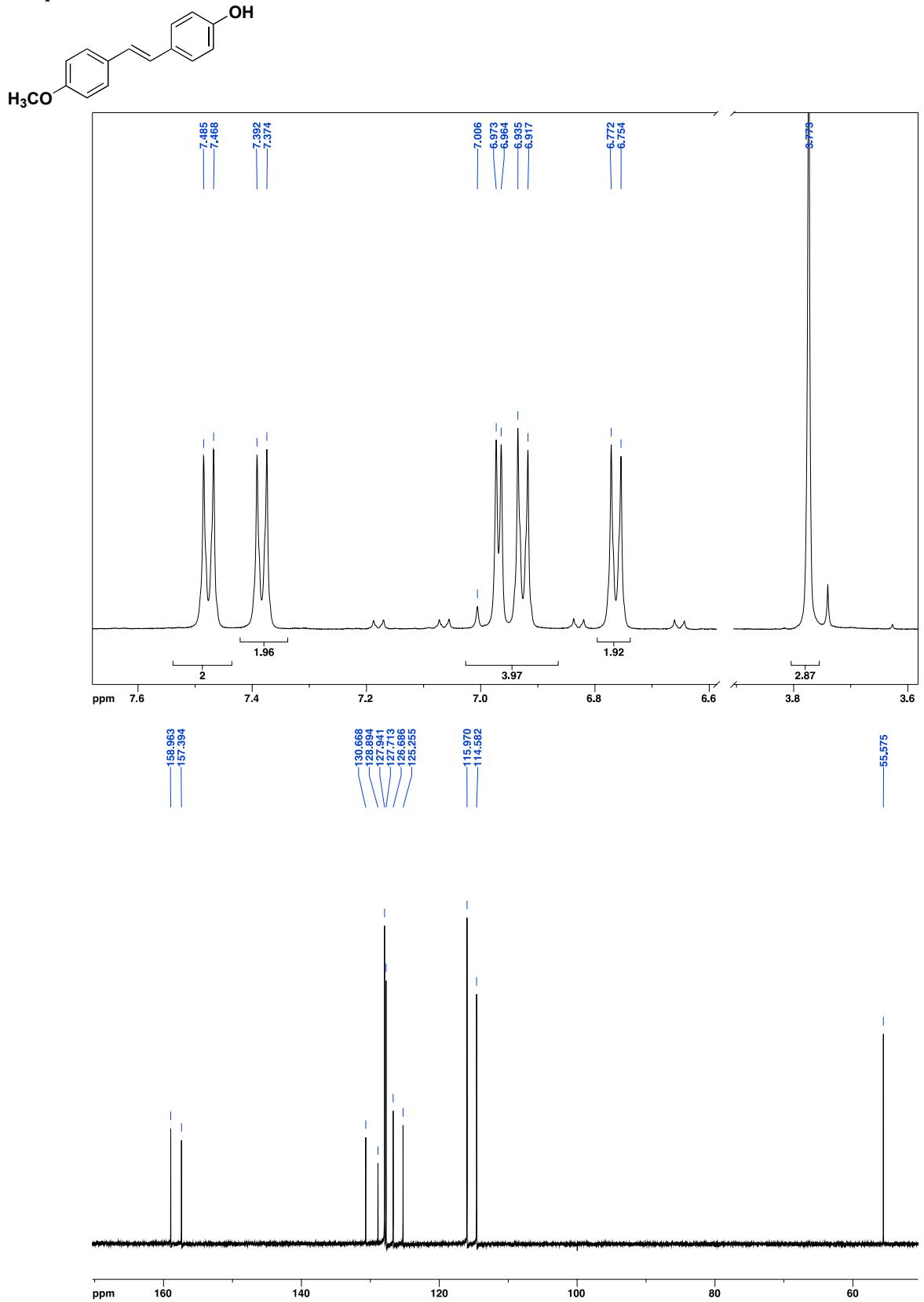
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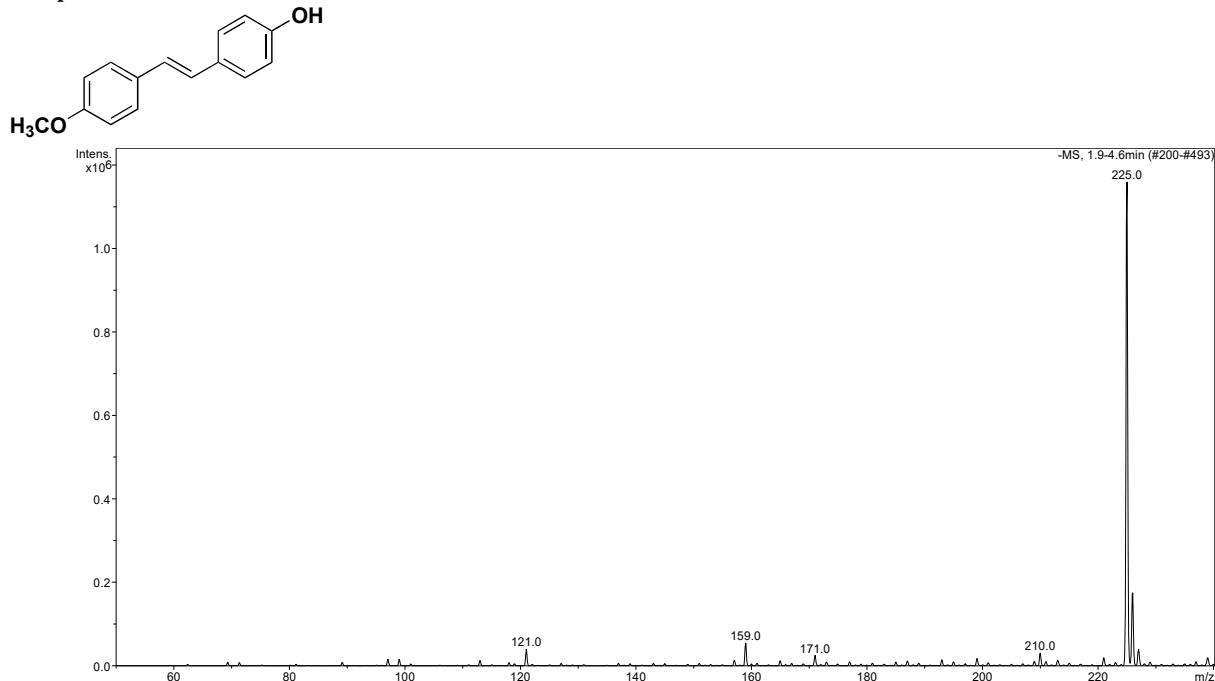
Compound 4



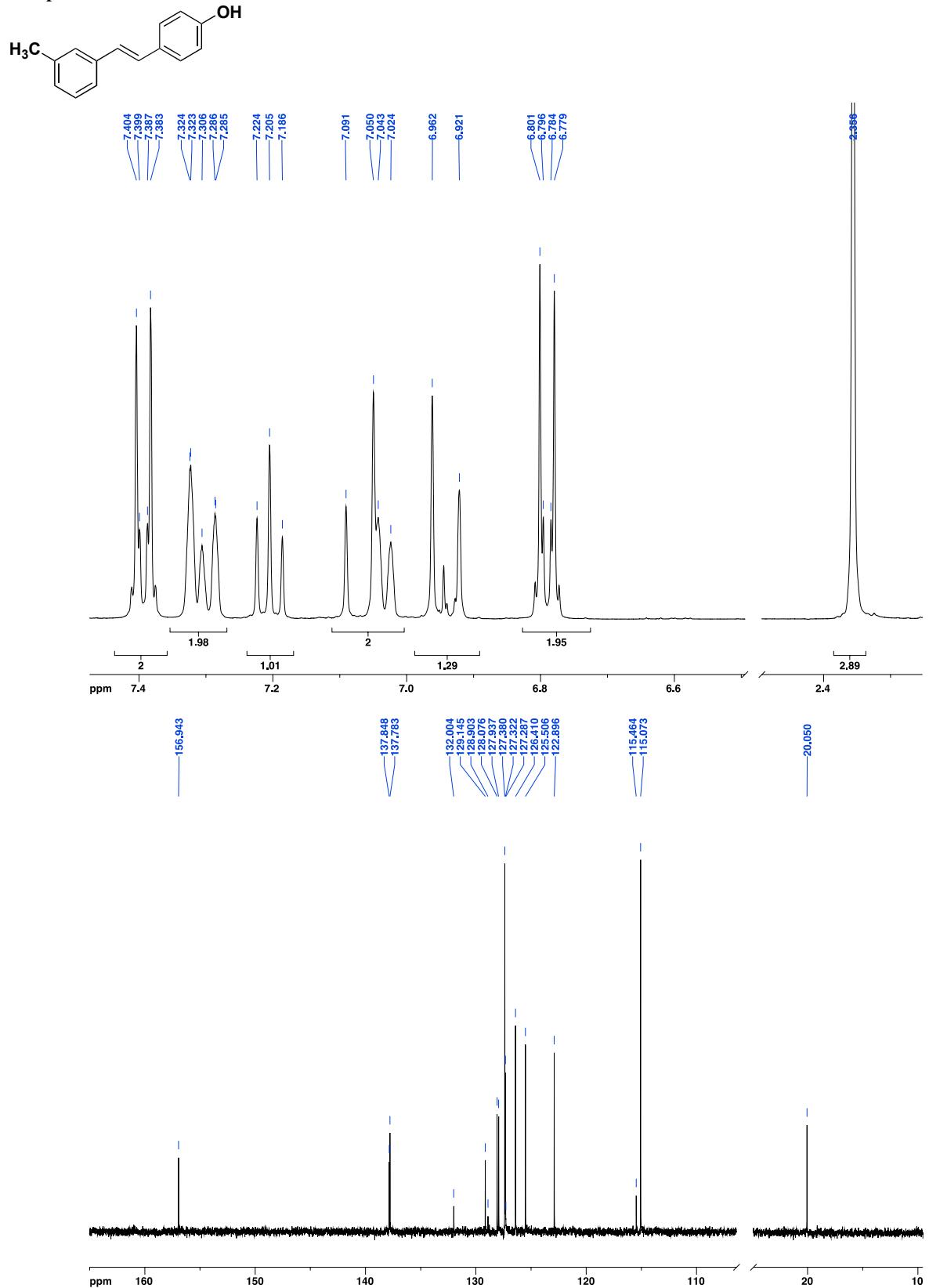
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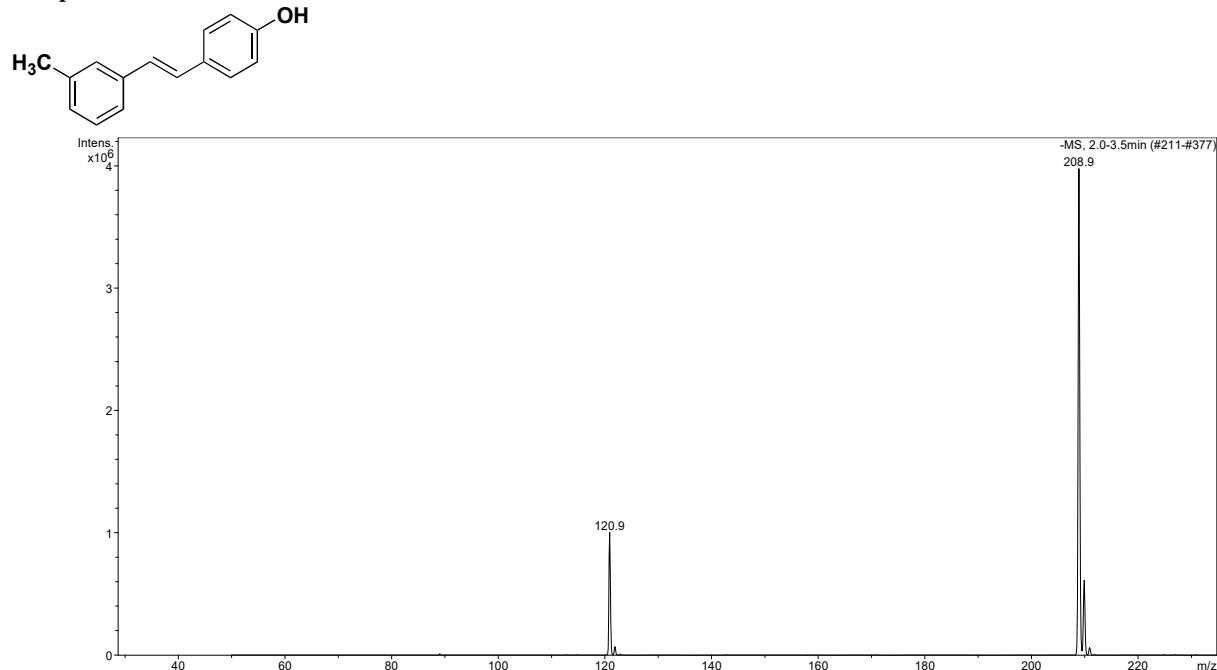
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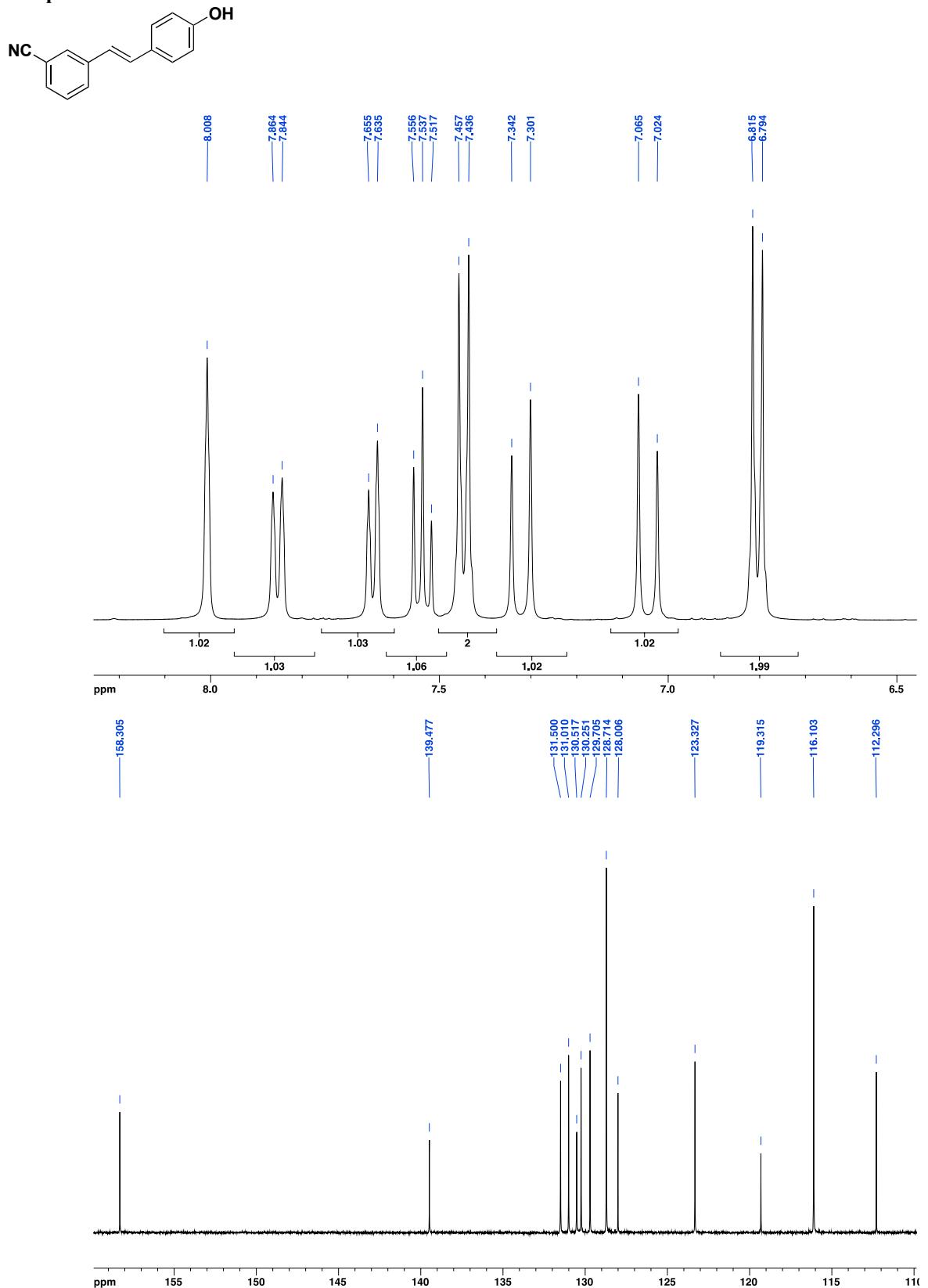
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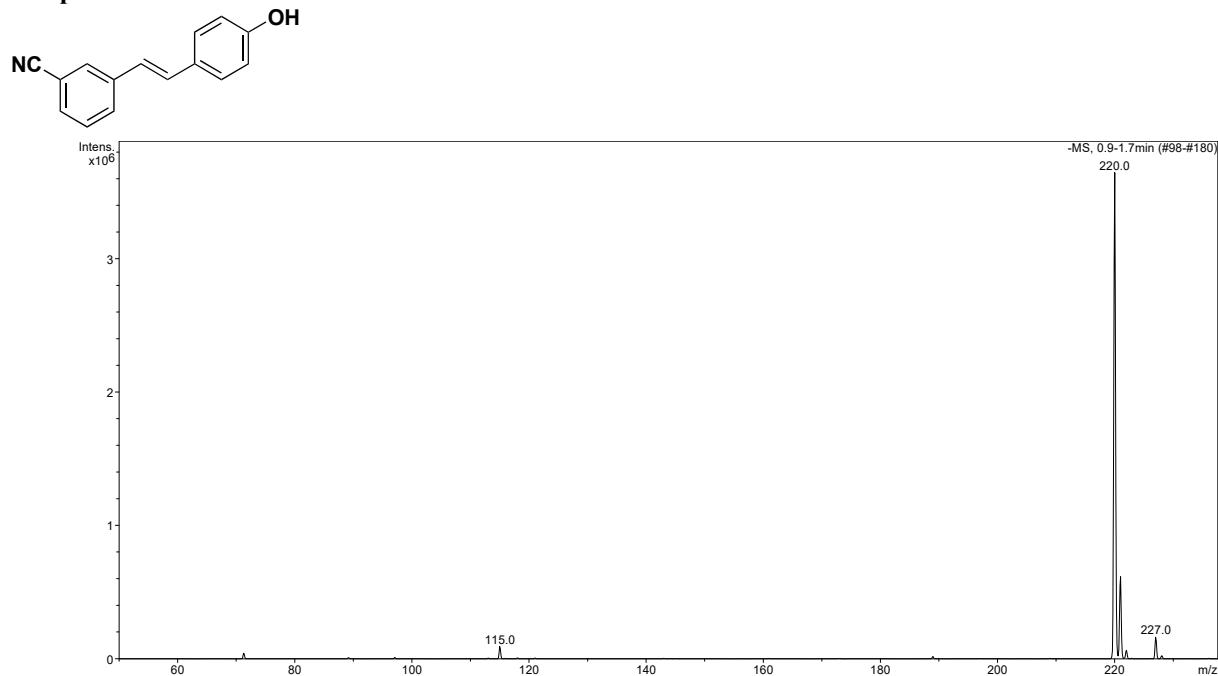
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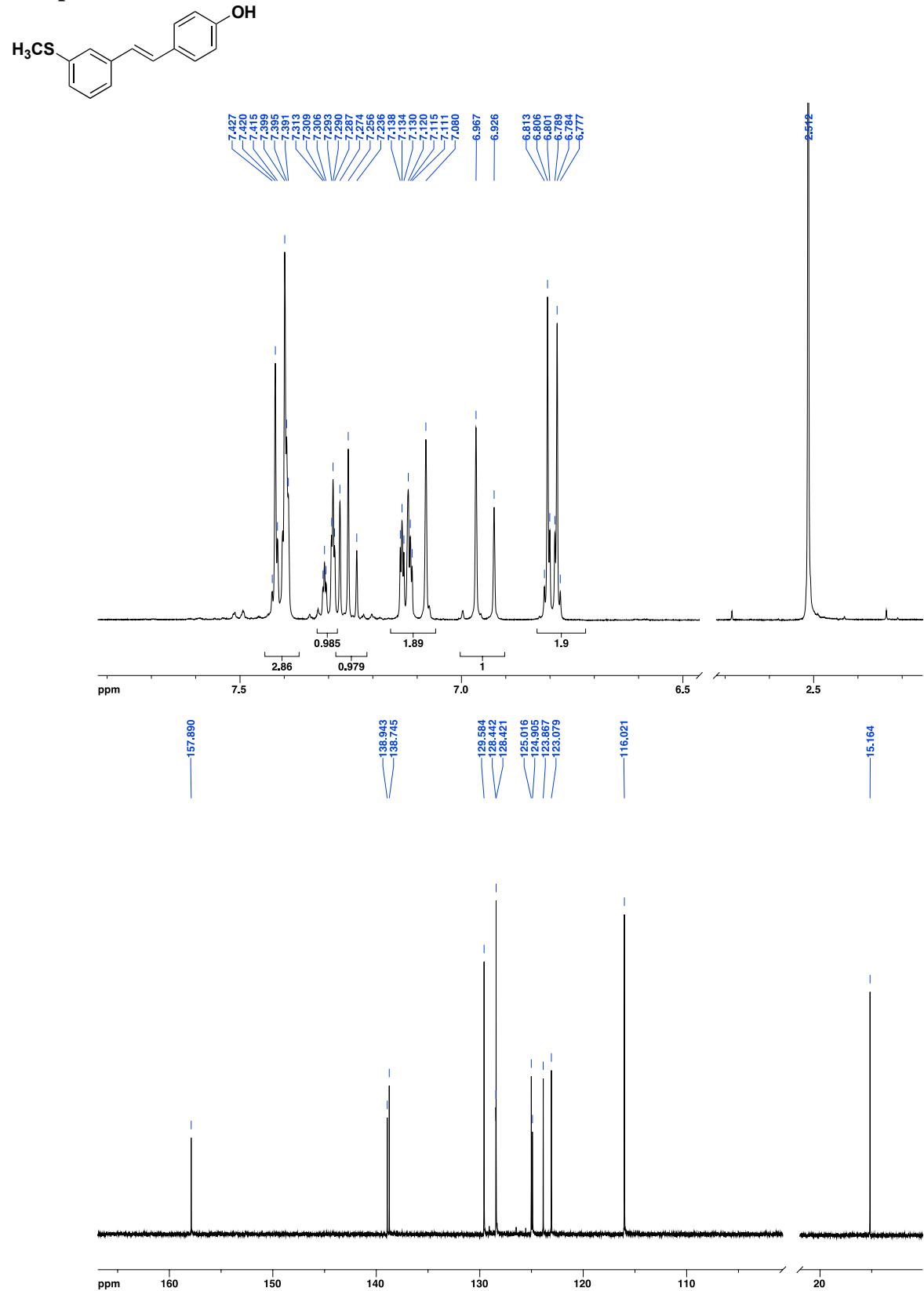
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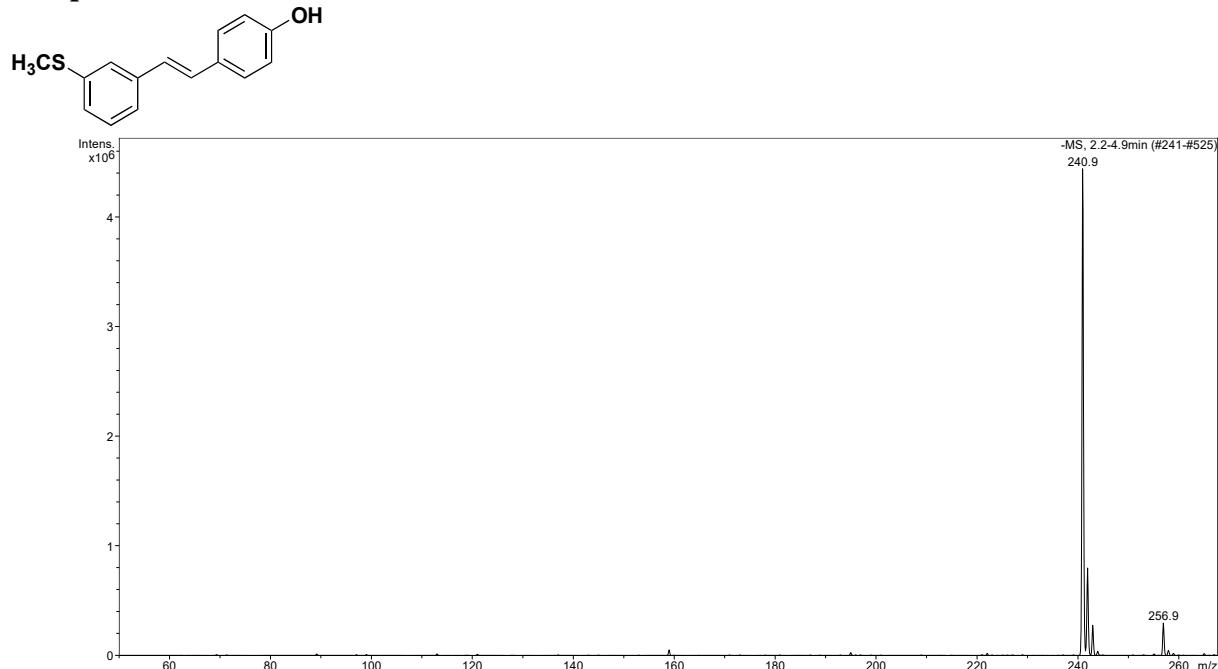
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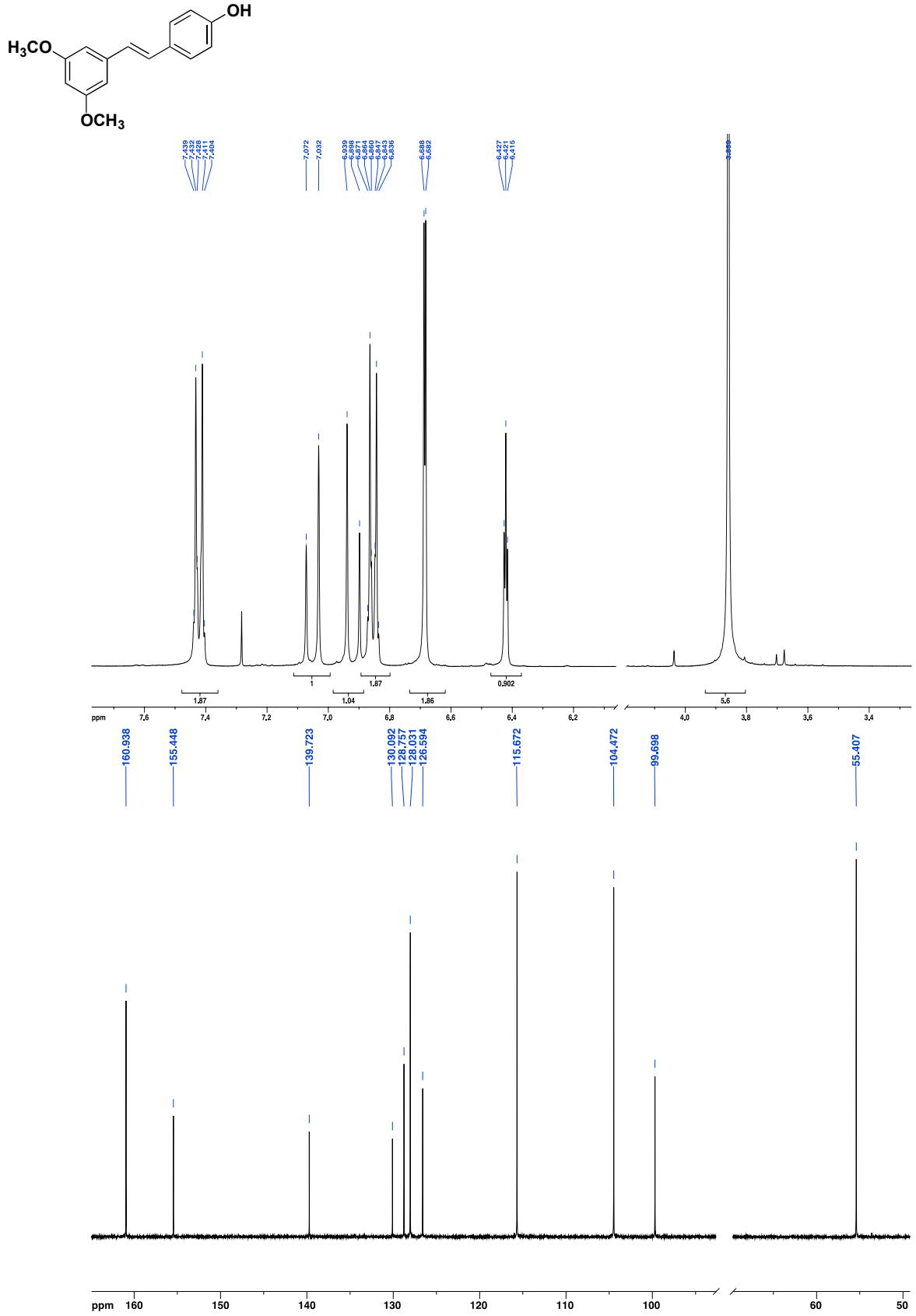
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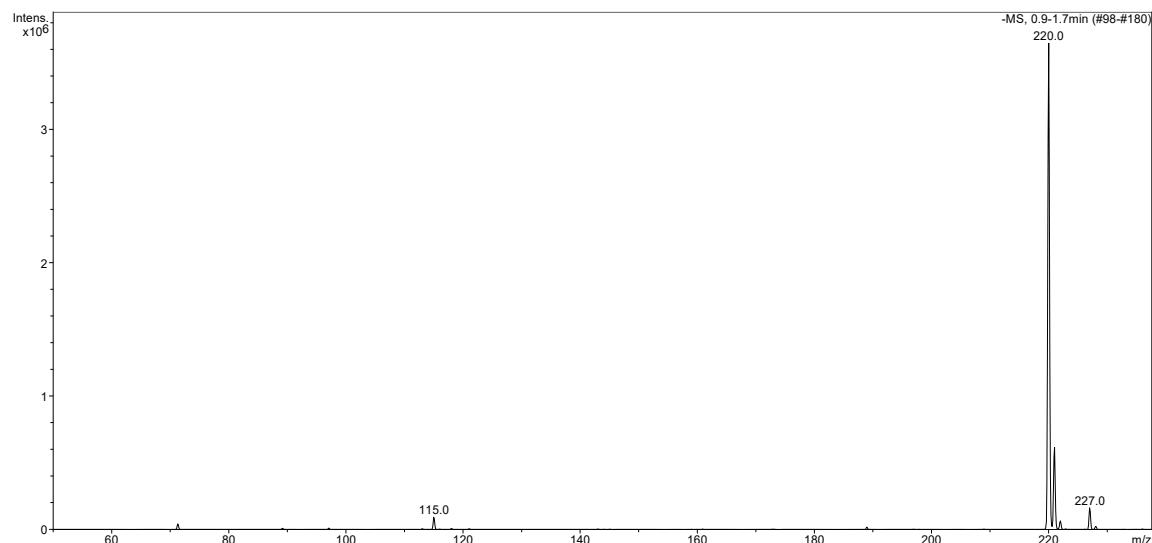
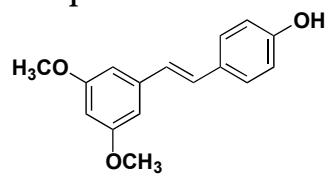
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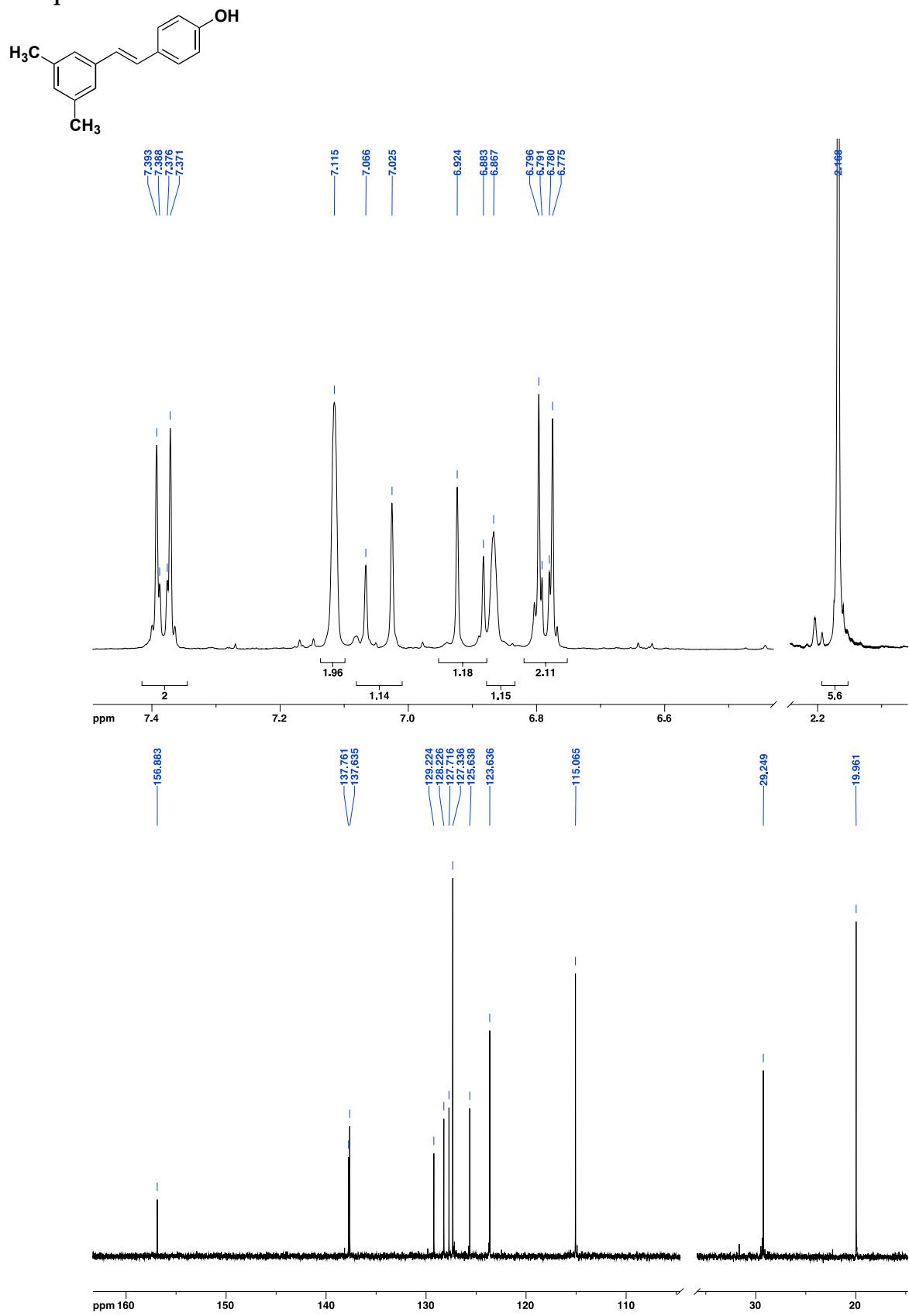
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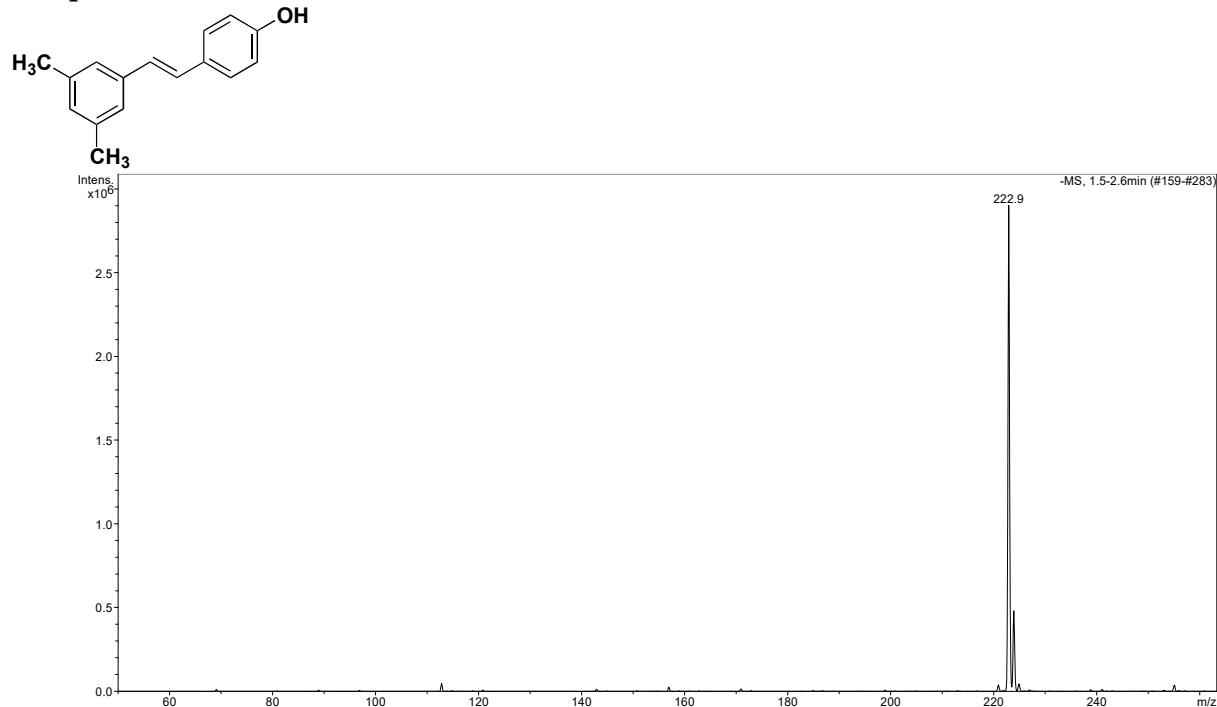
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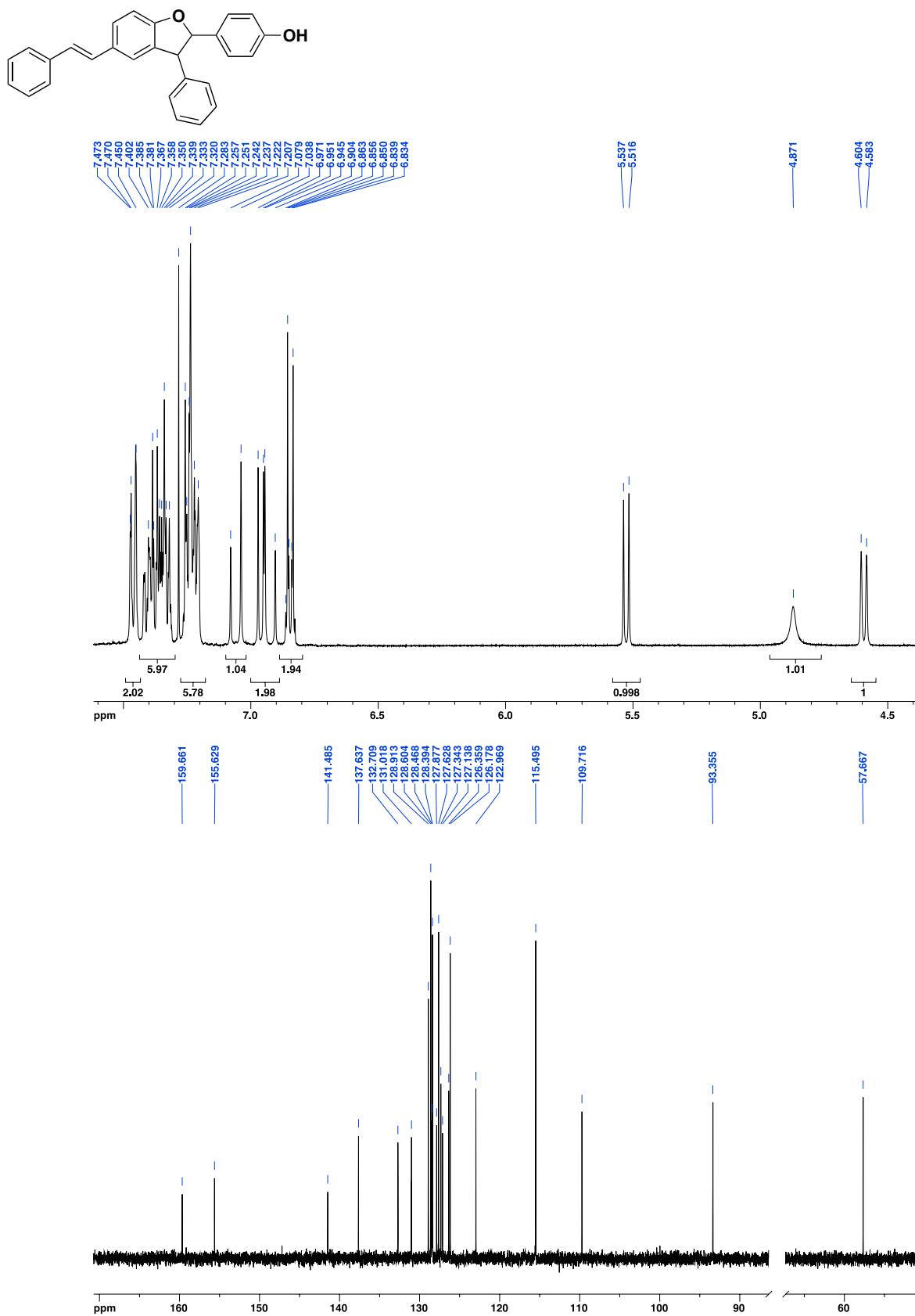
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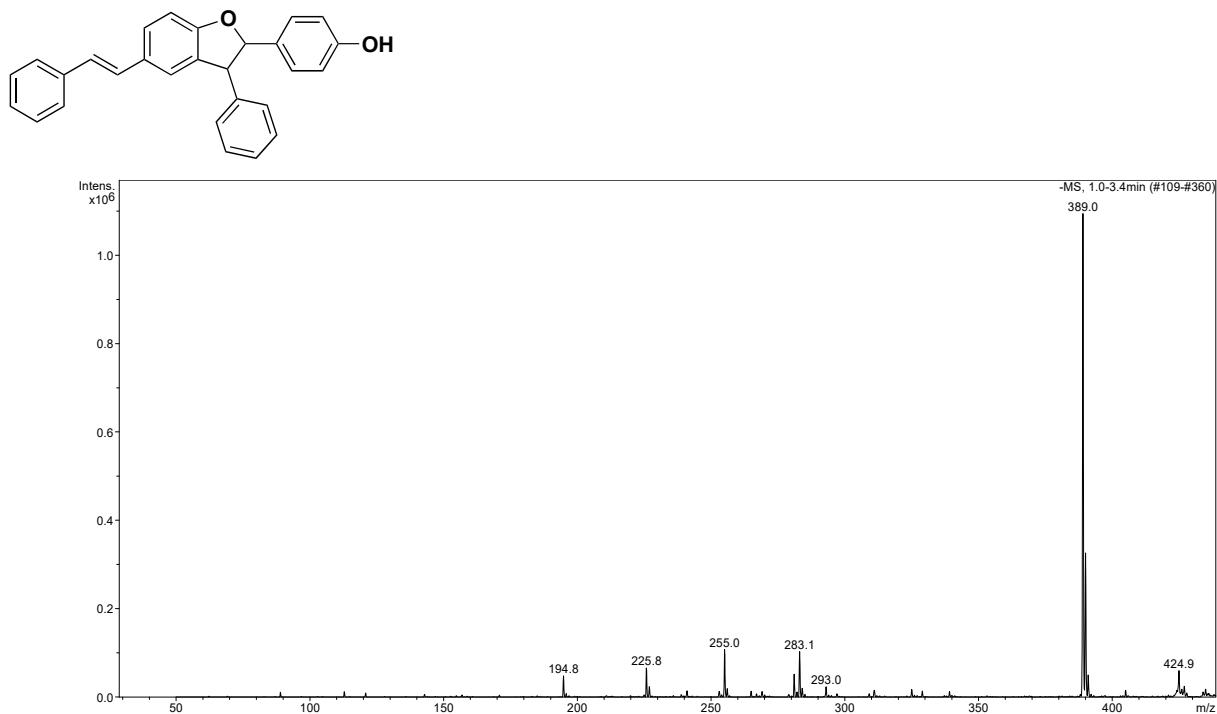
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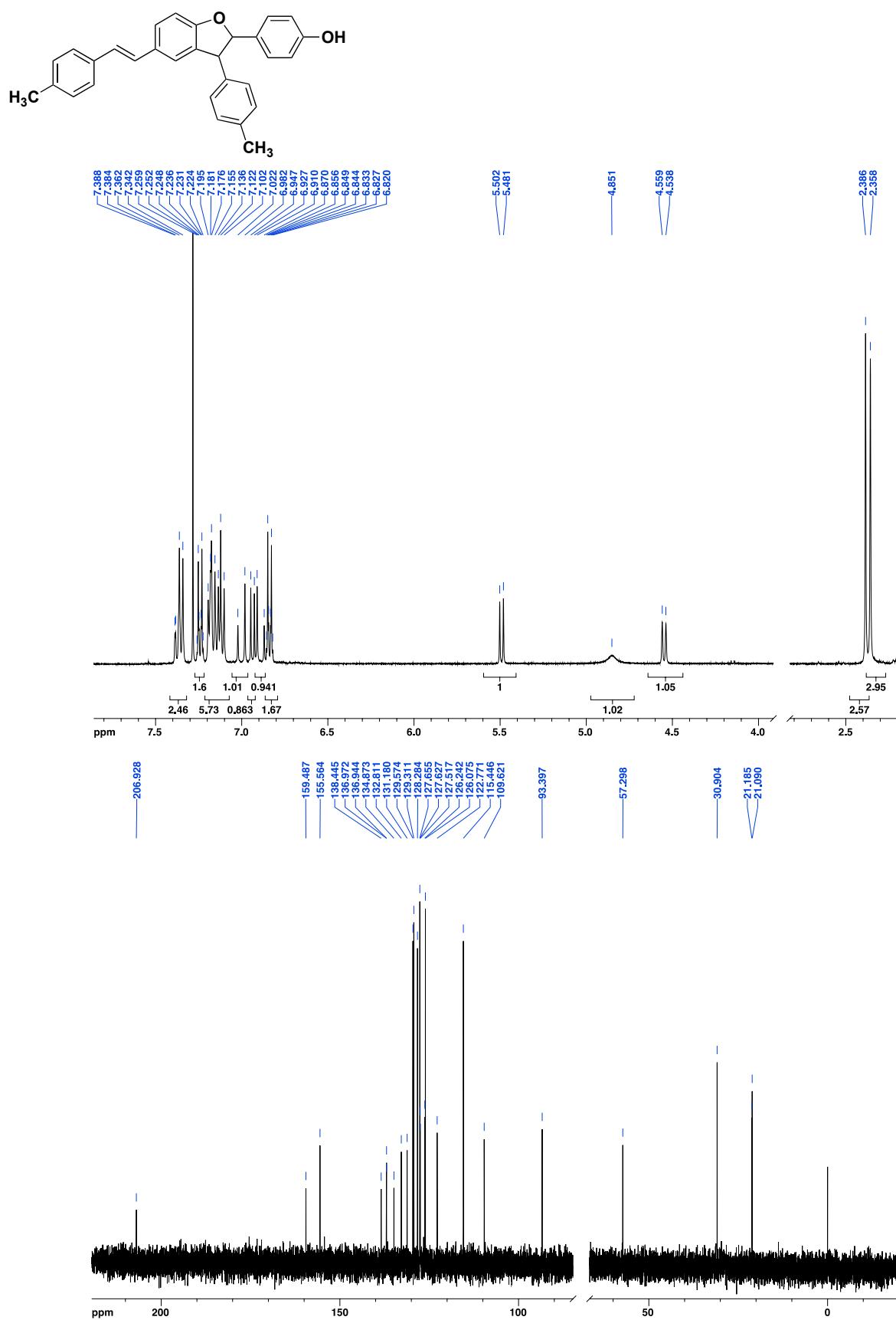
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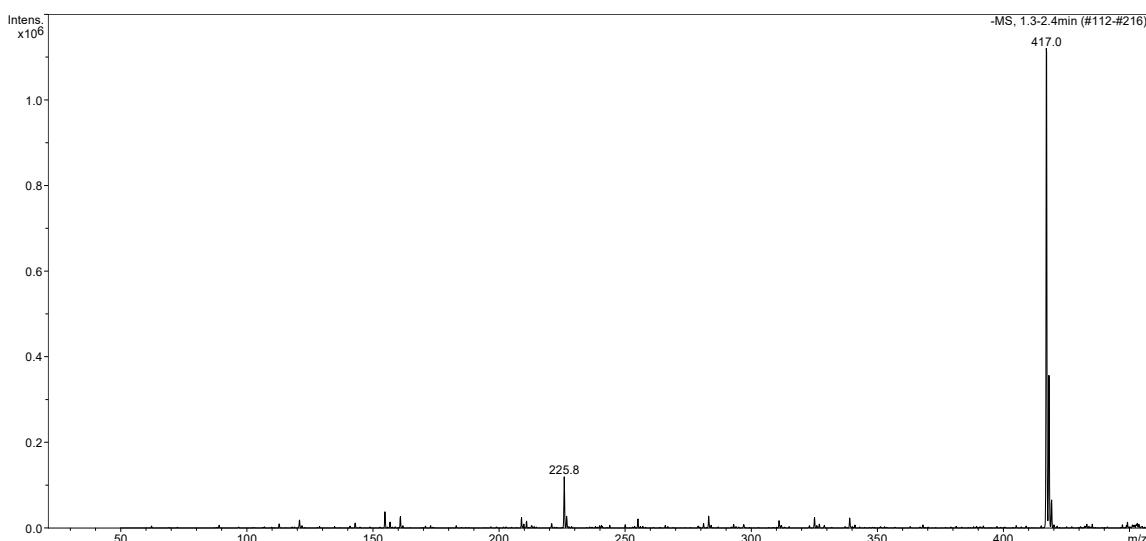
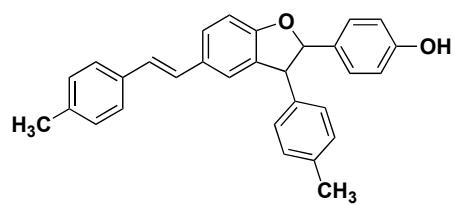
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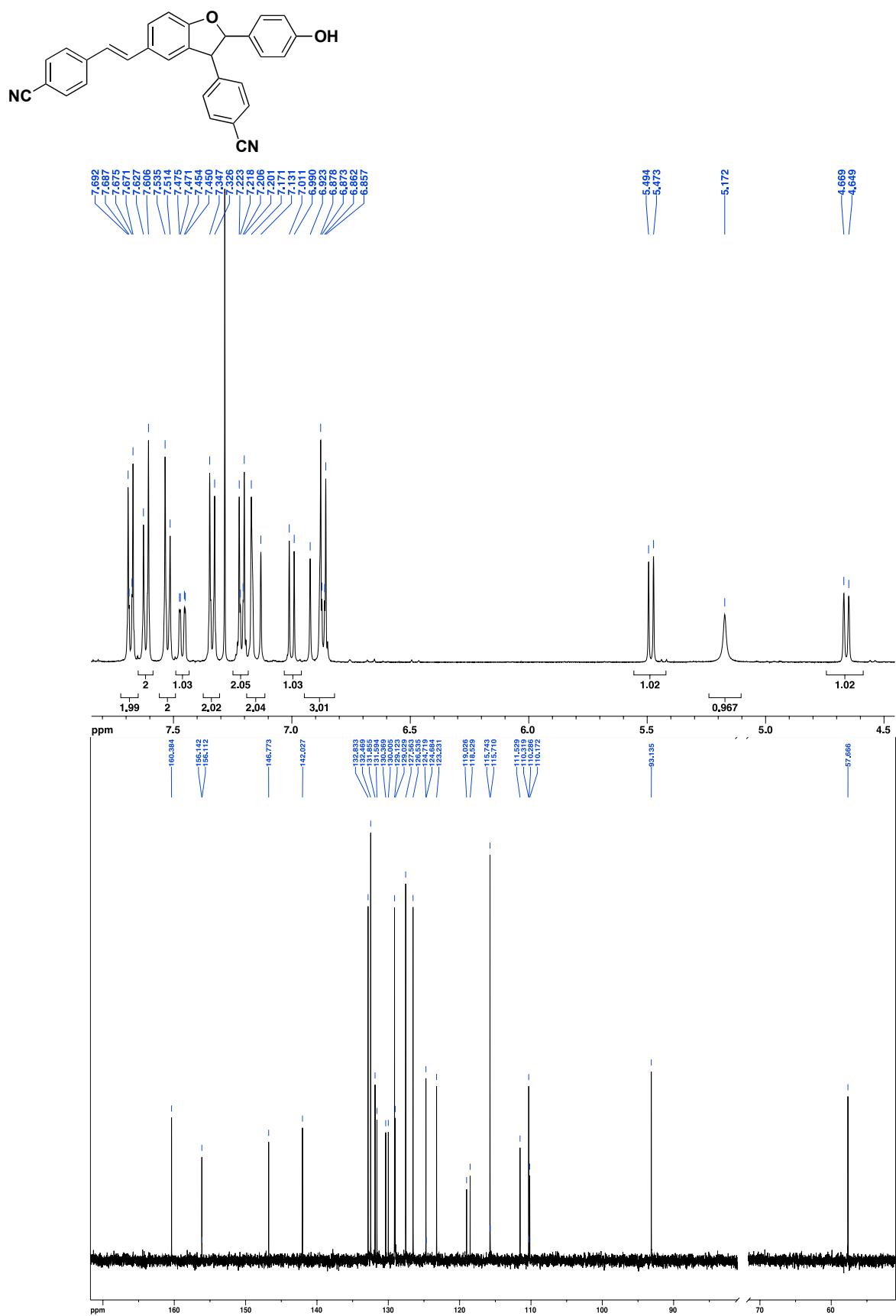
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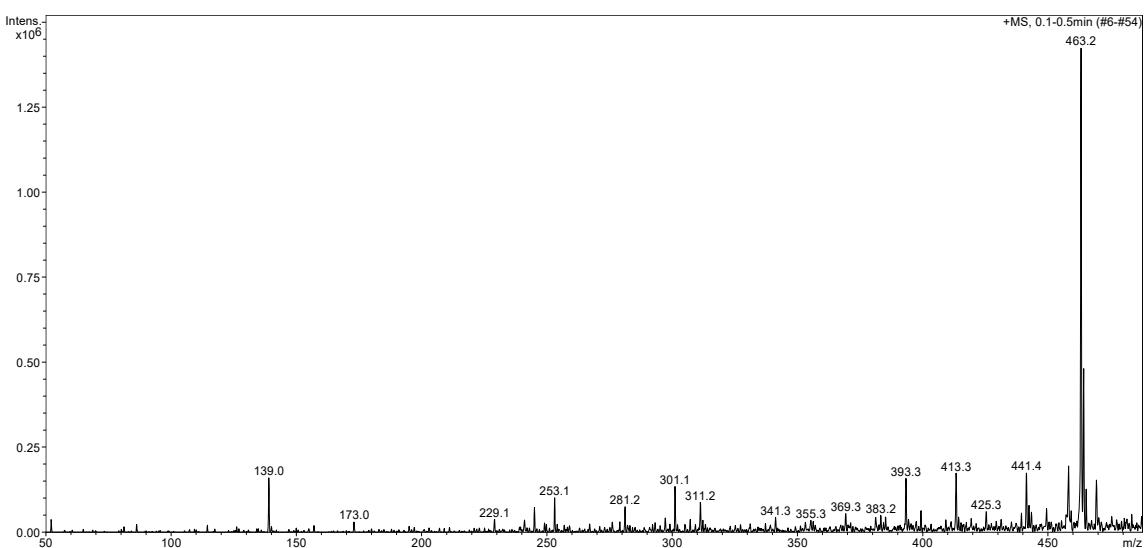
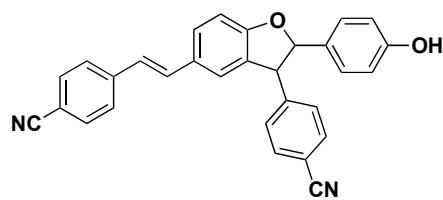
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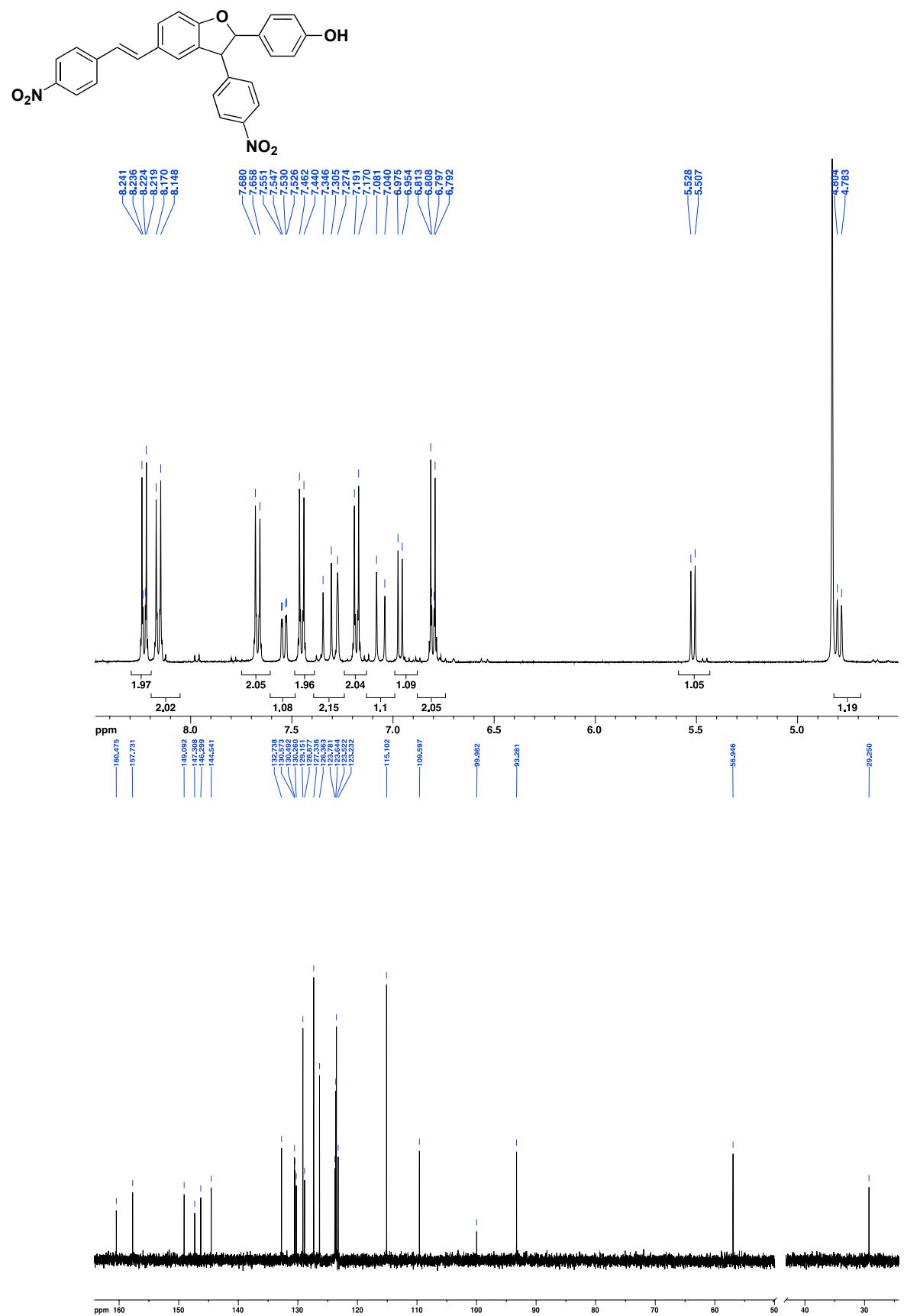
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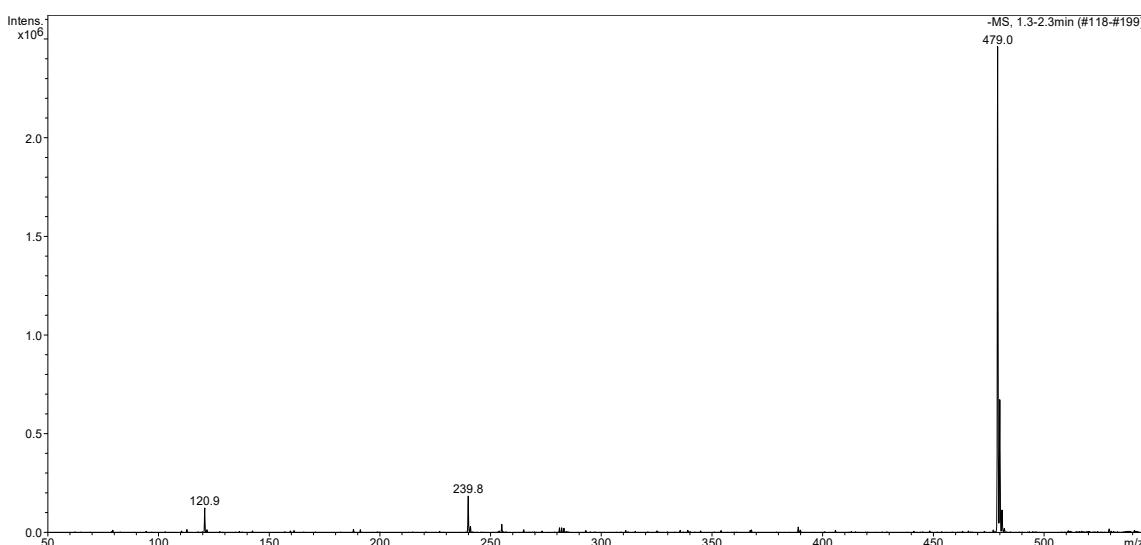
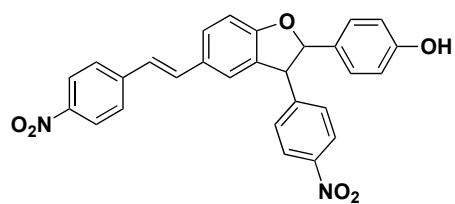
Compound 14



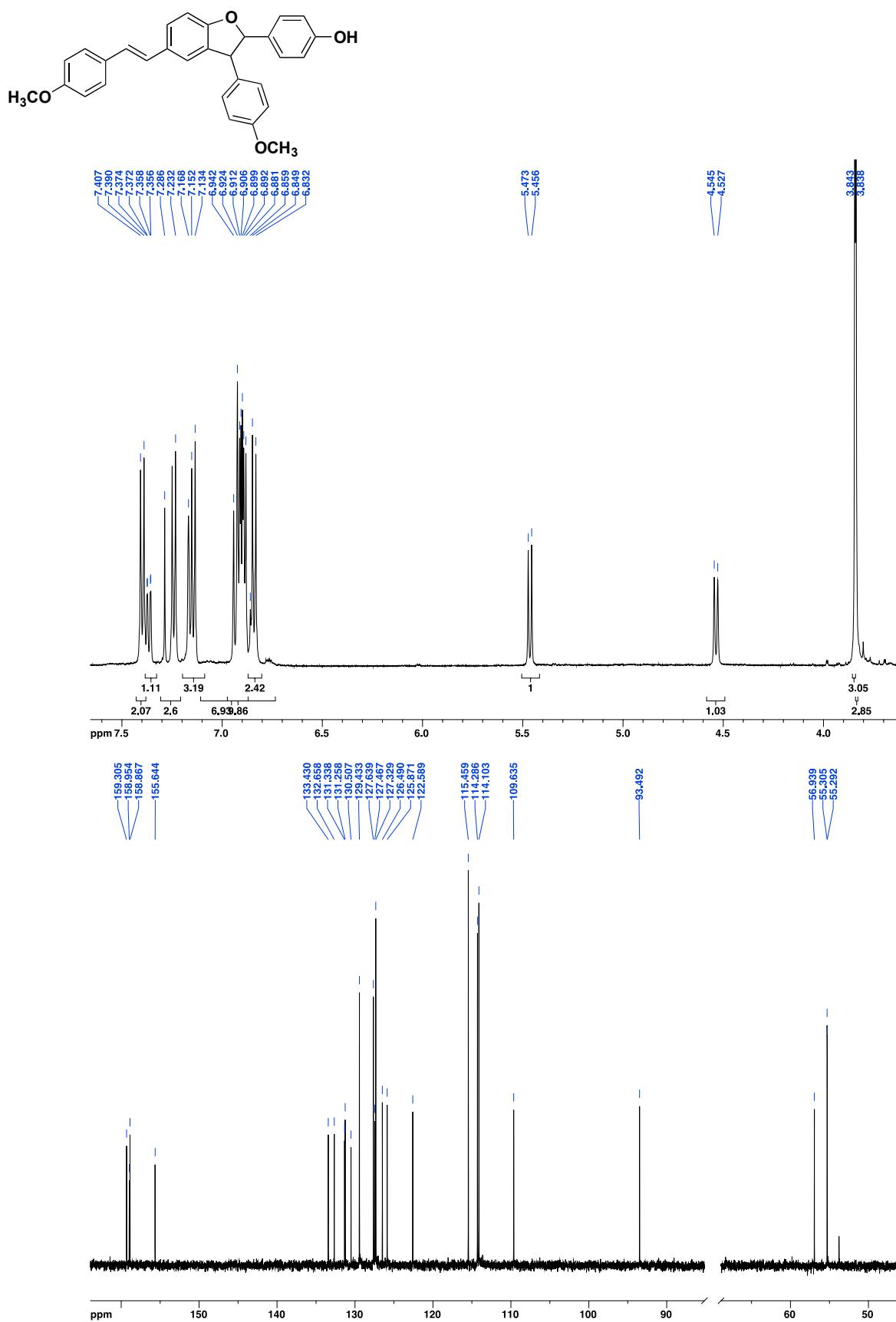
Compound 15



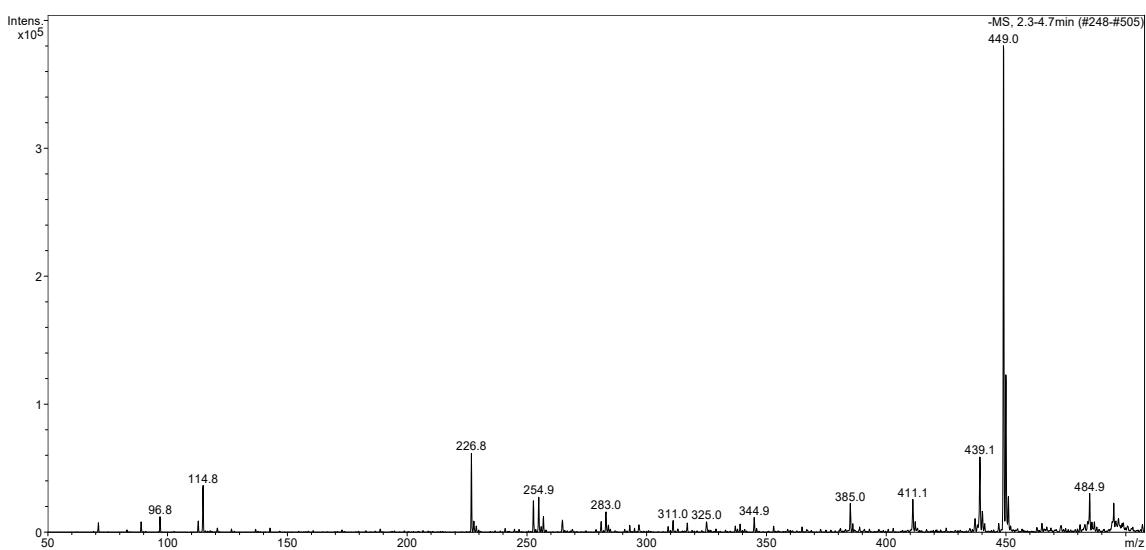
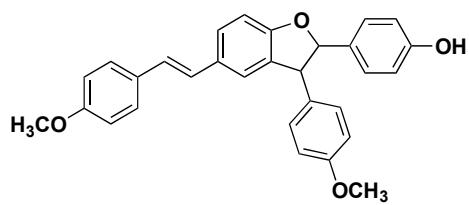
Compound 15



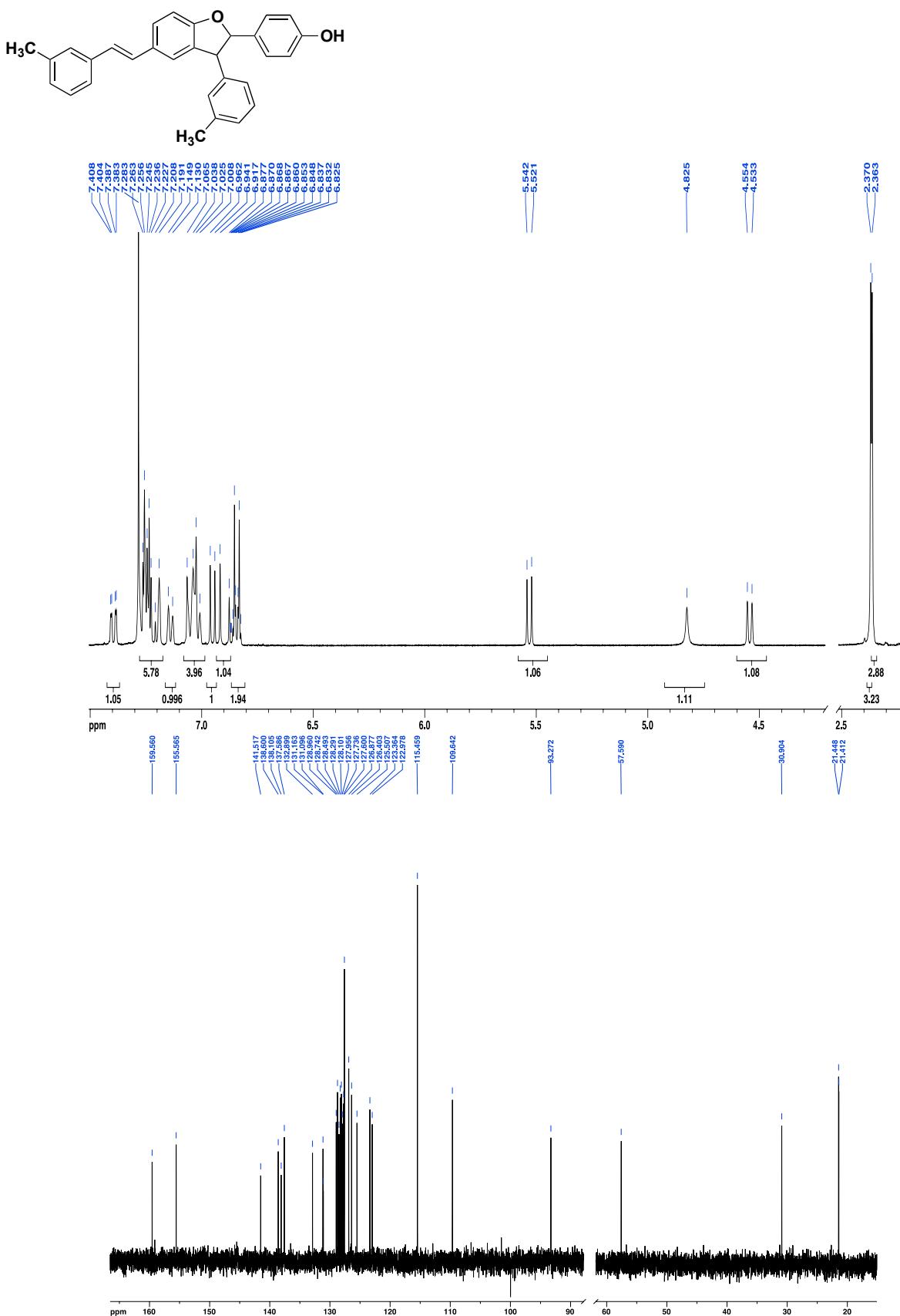
Compound 16



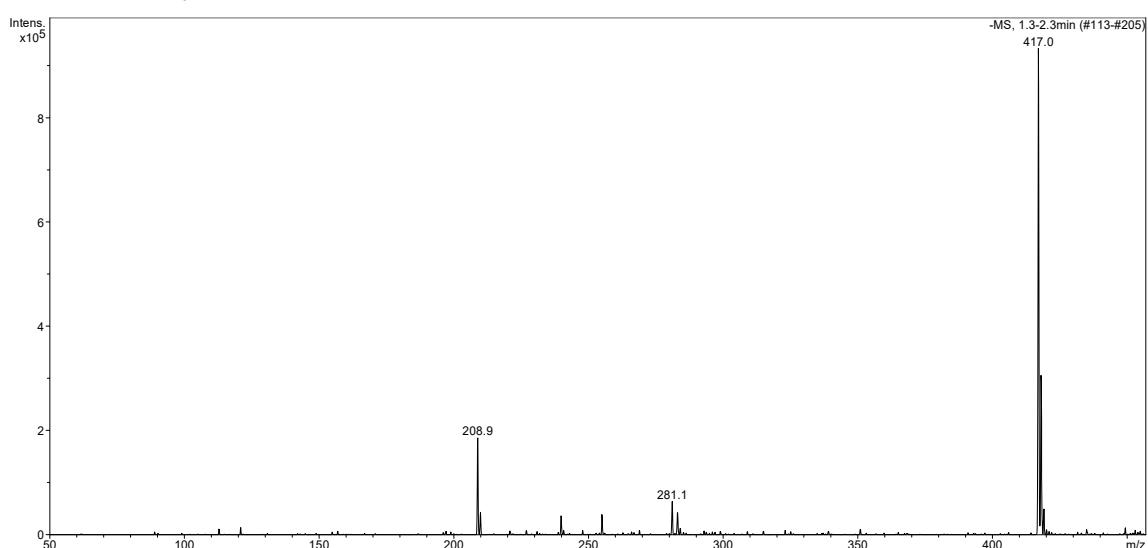
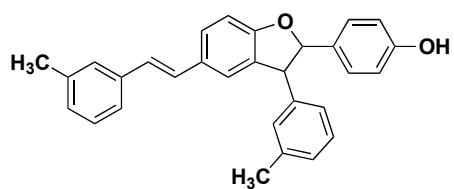
Compound 16



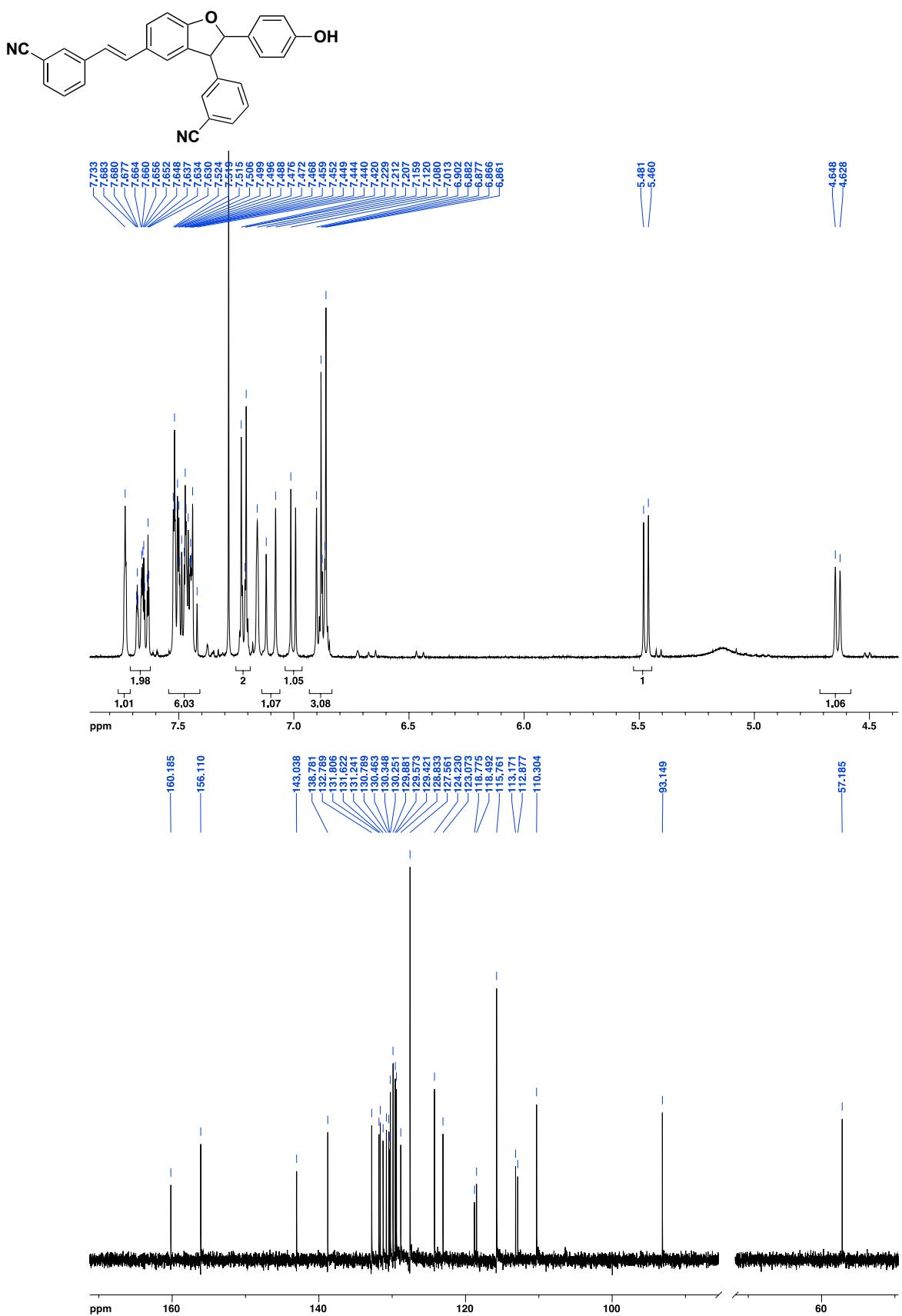
Compound 17



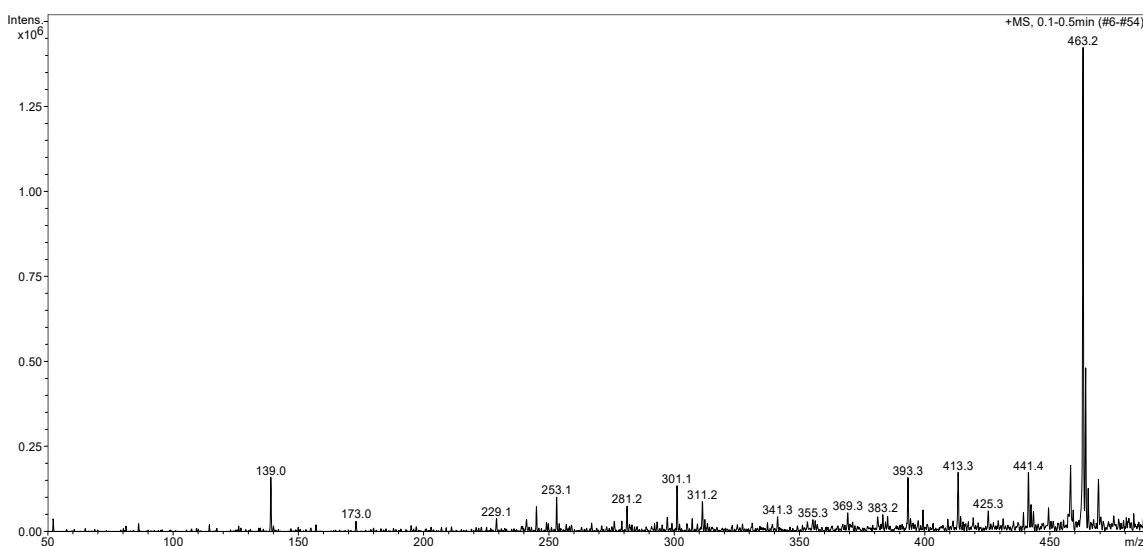
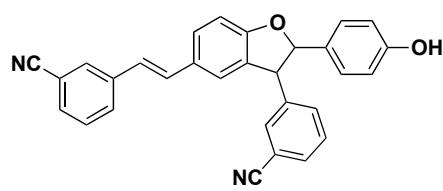
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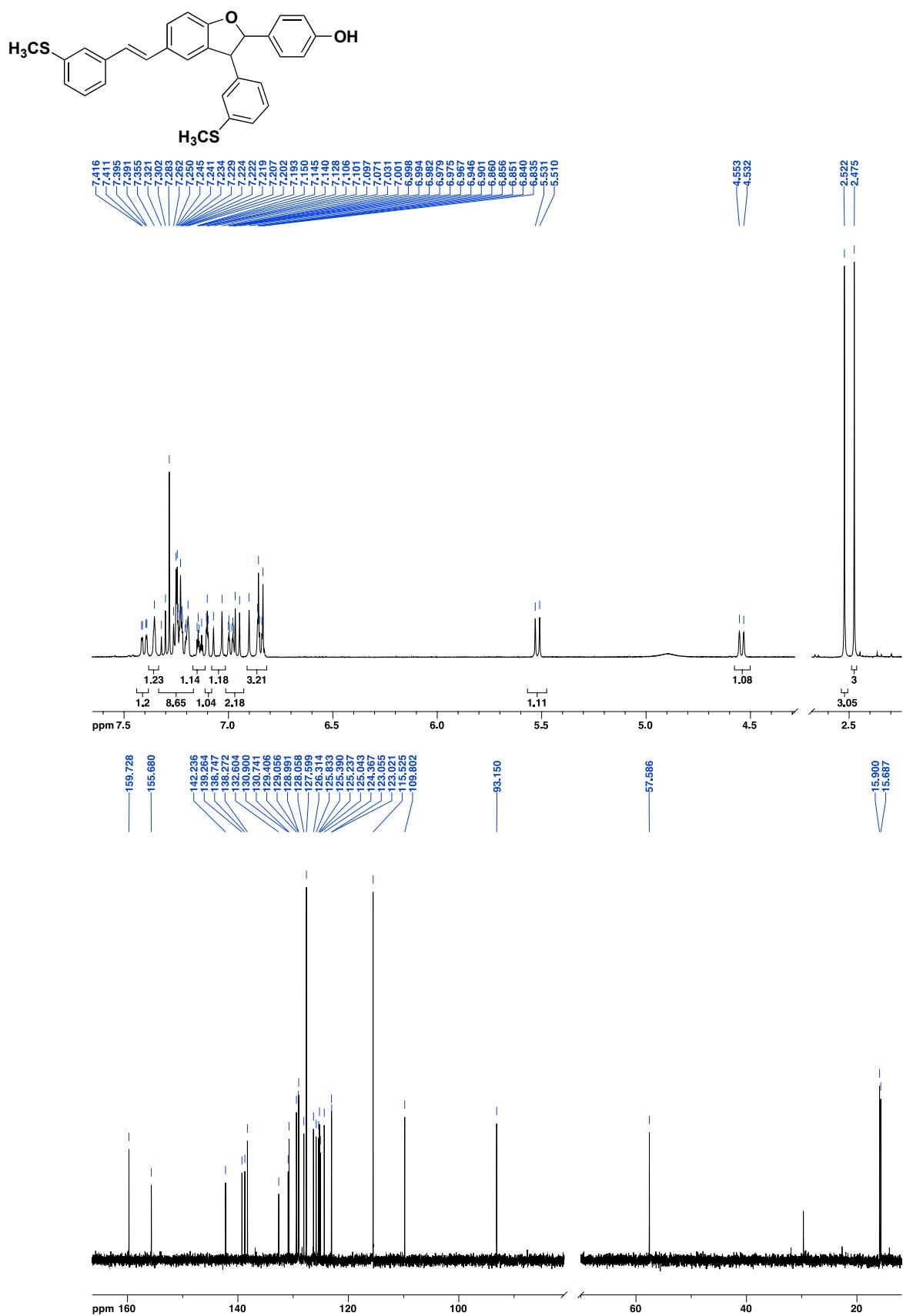
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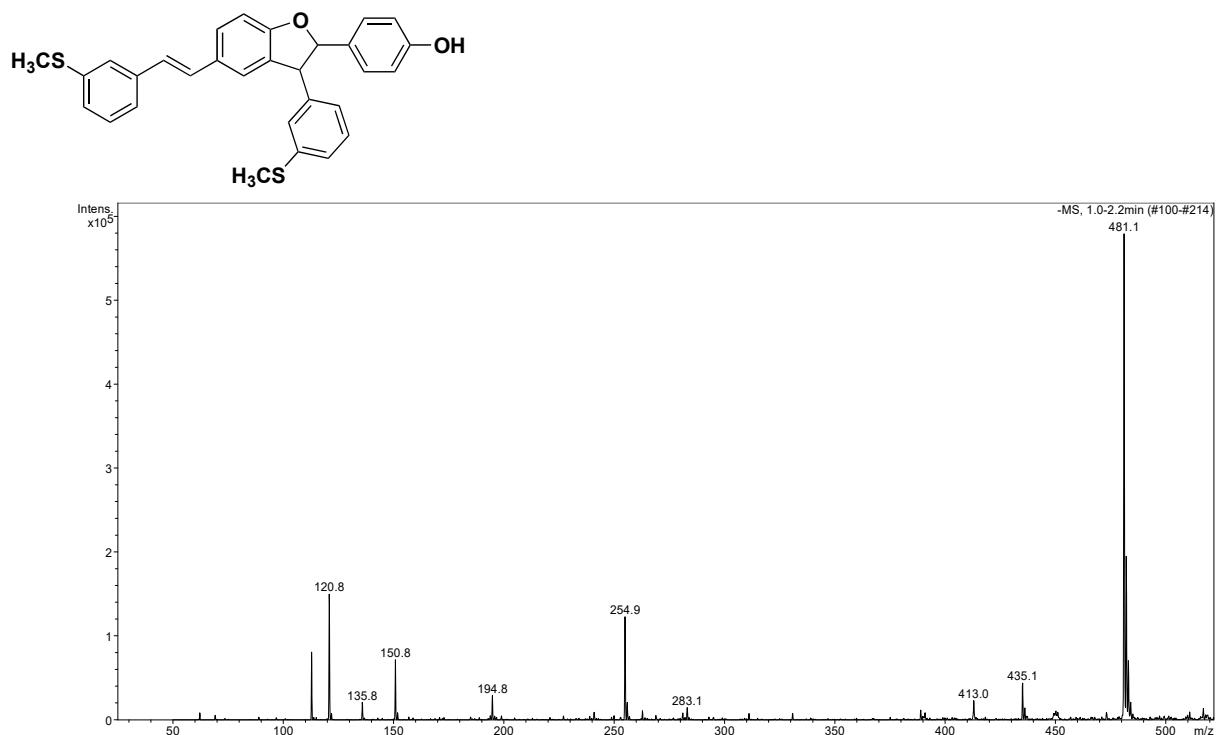
Compound 18



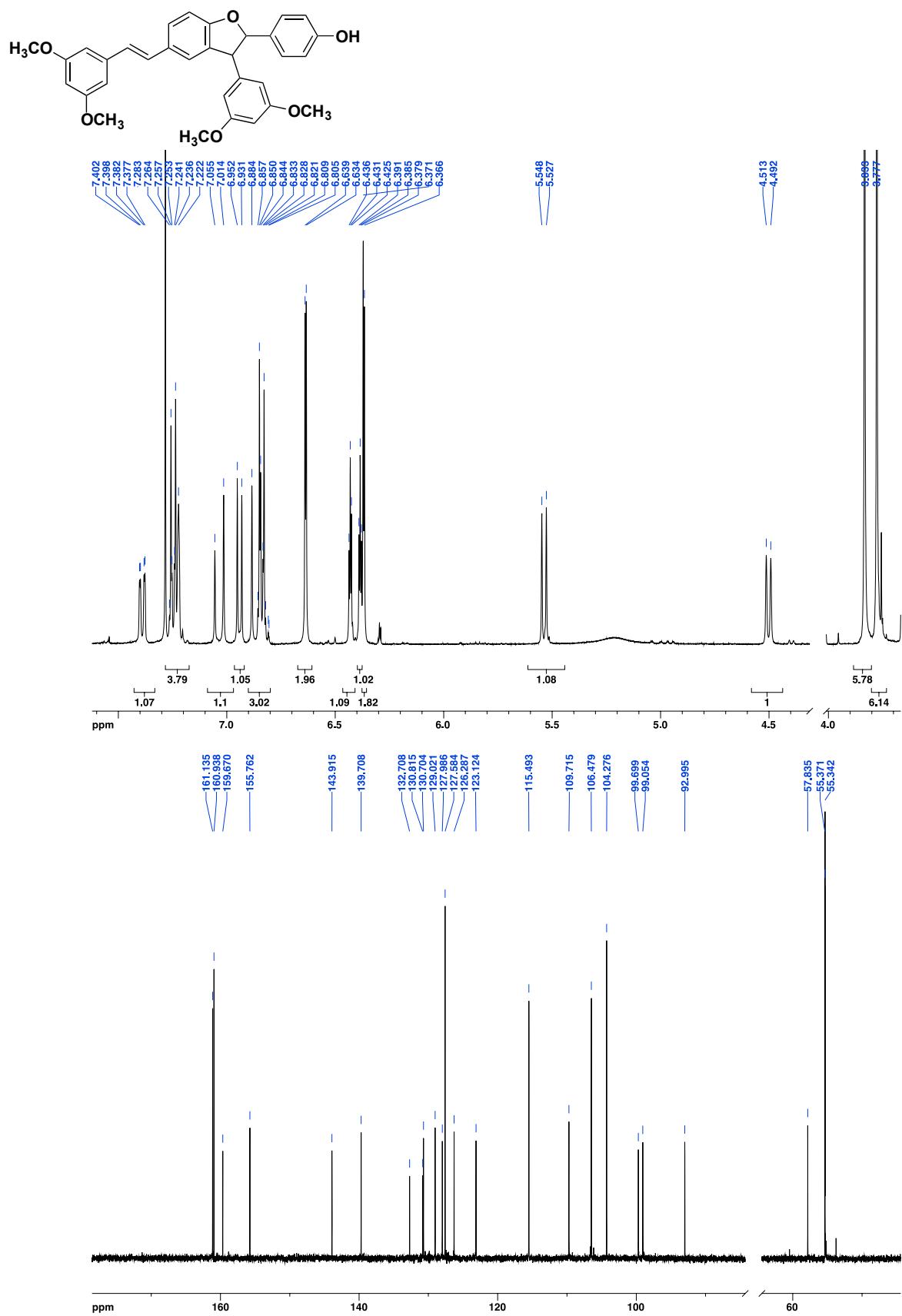
Compound 19



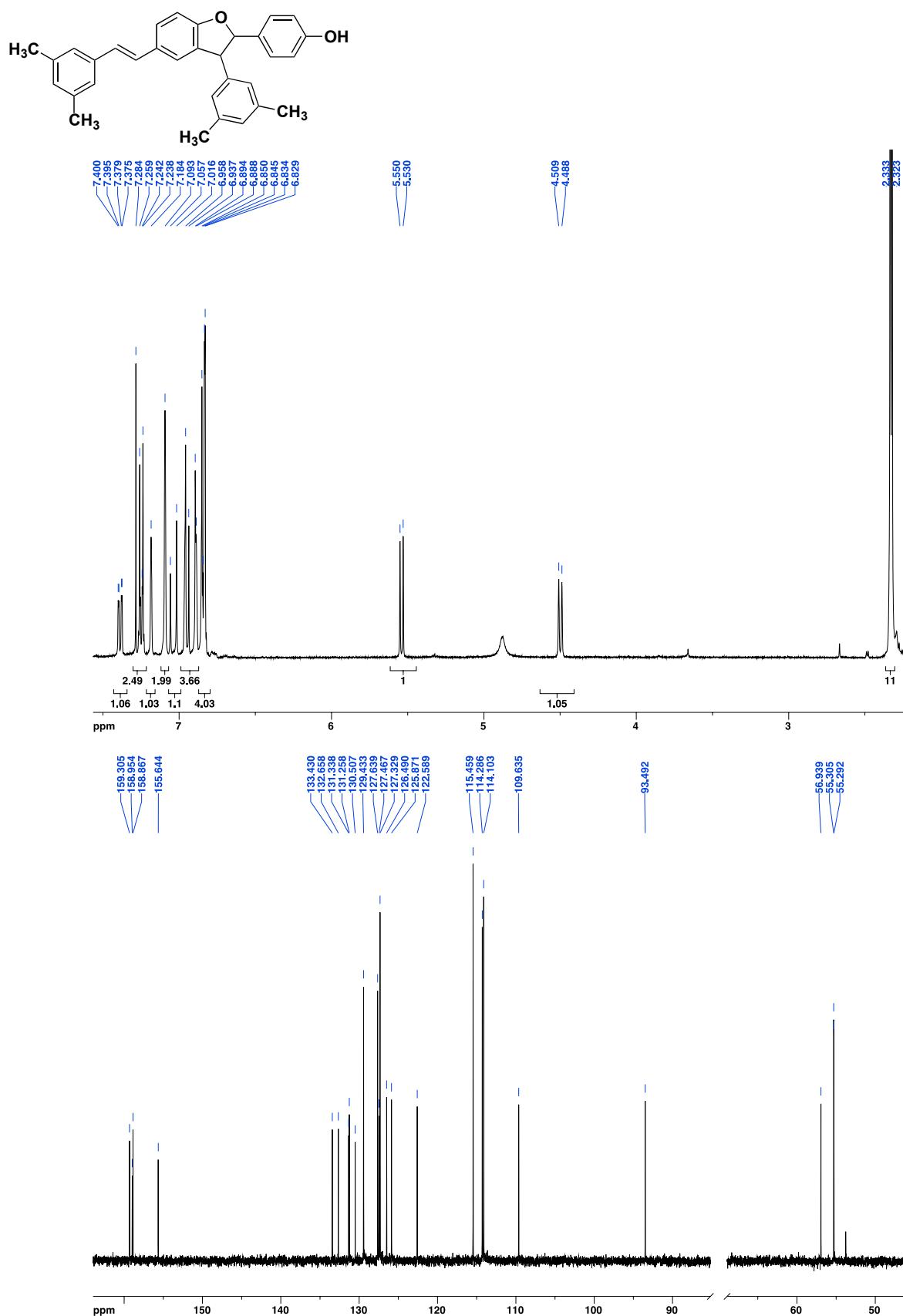
Compound 19



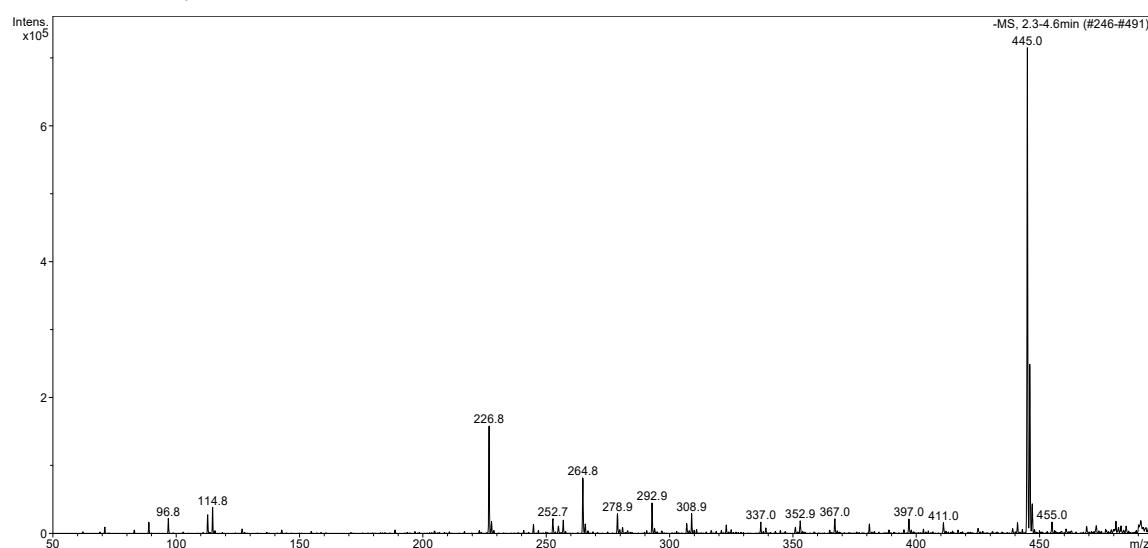
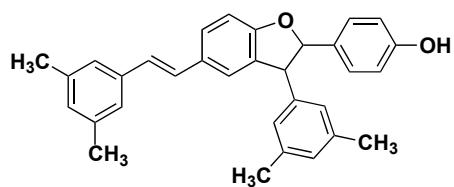
Compound 20



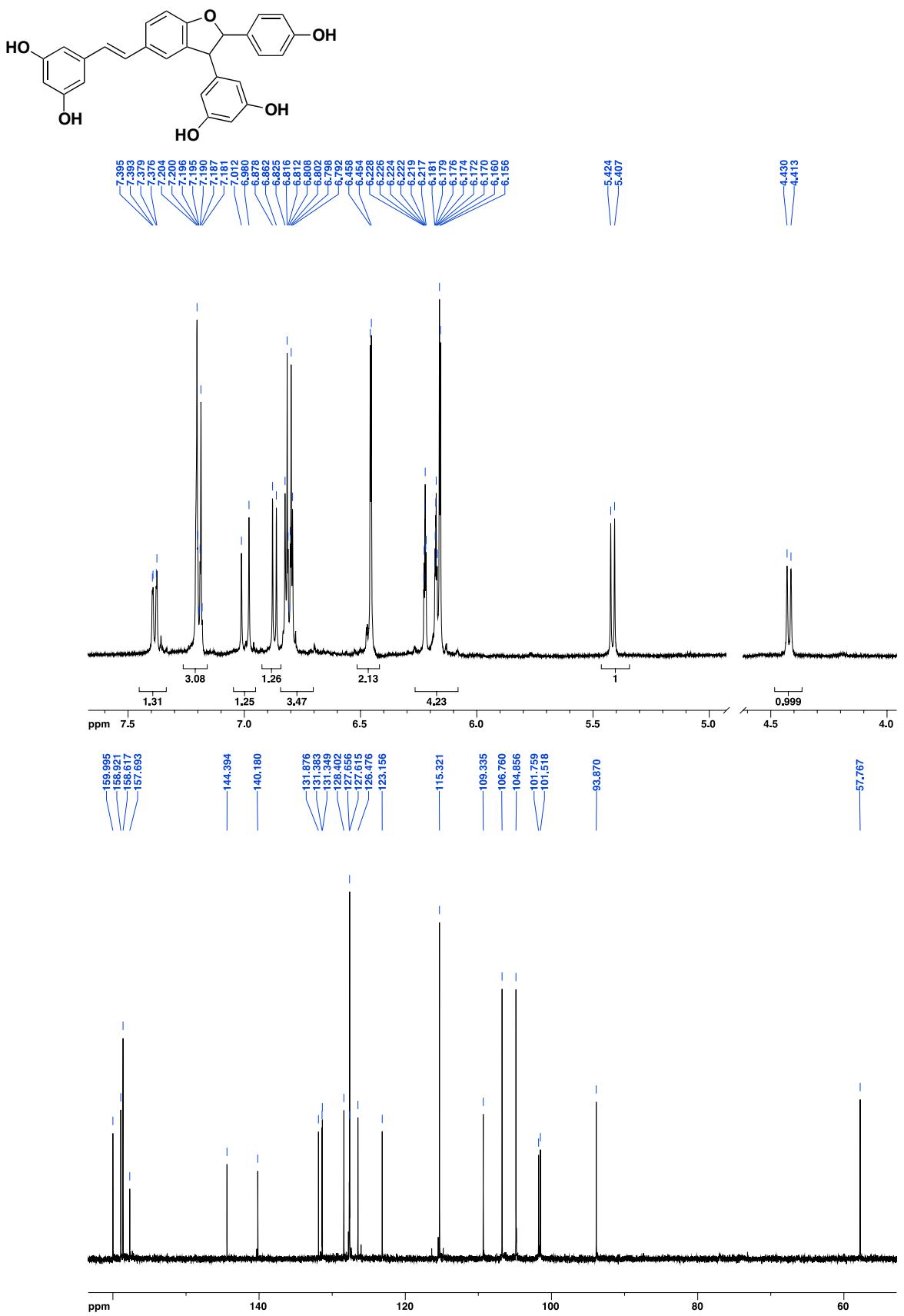
Compound 21



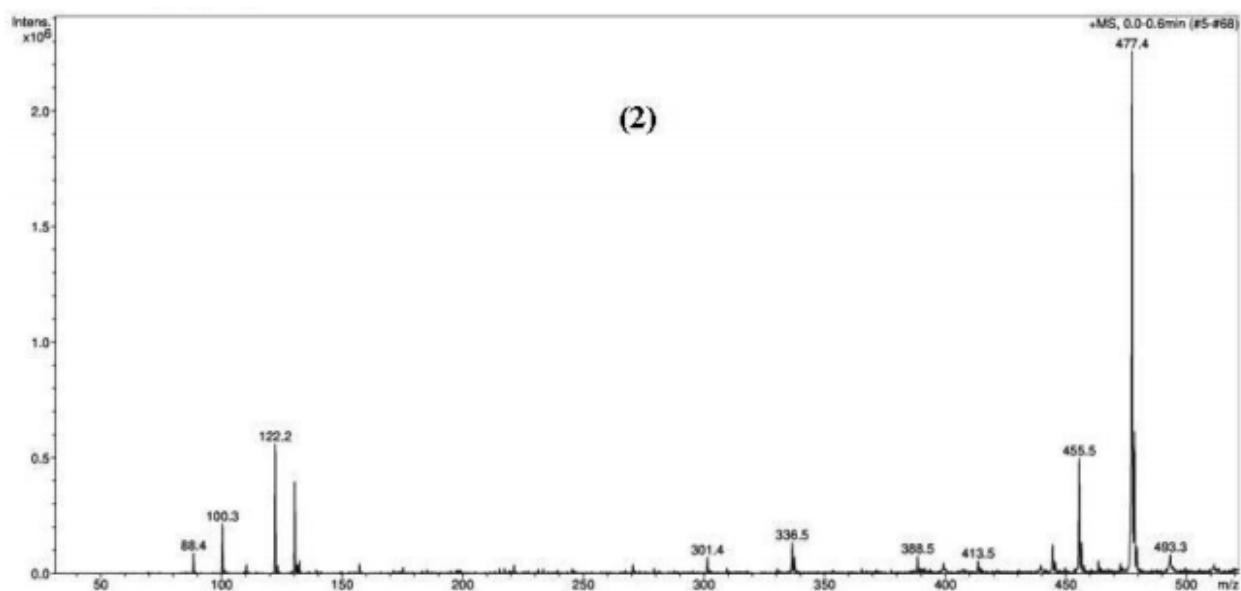
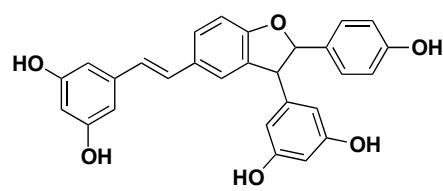
Compound 21



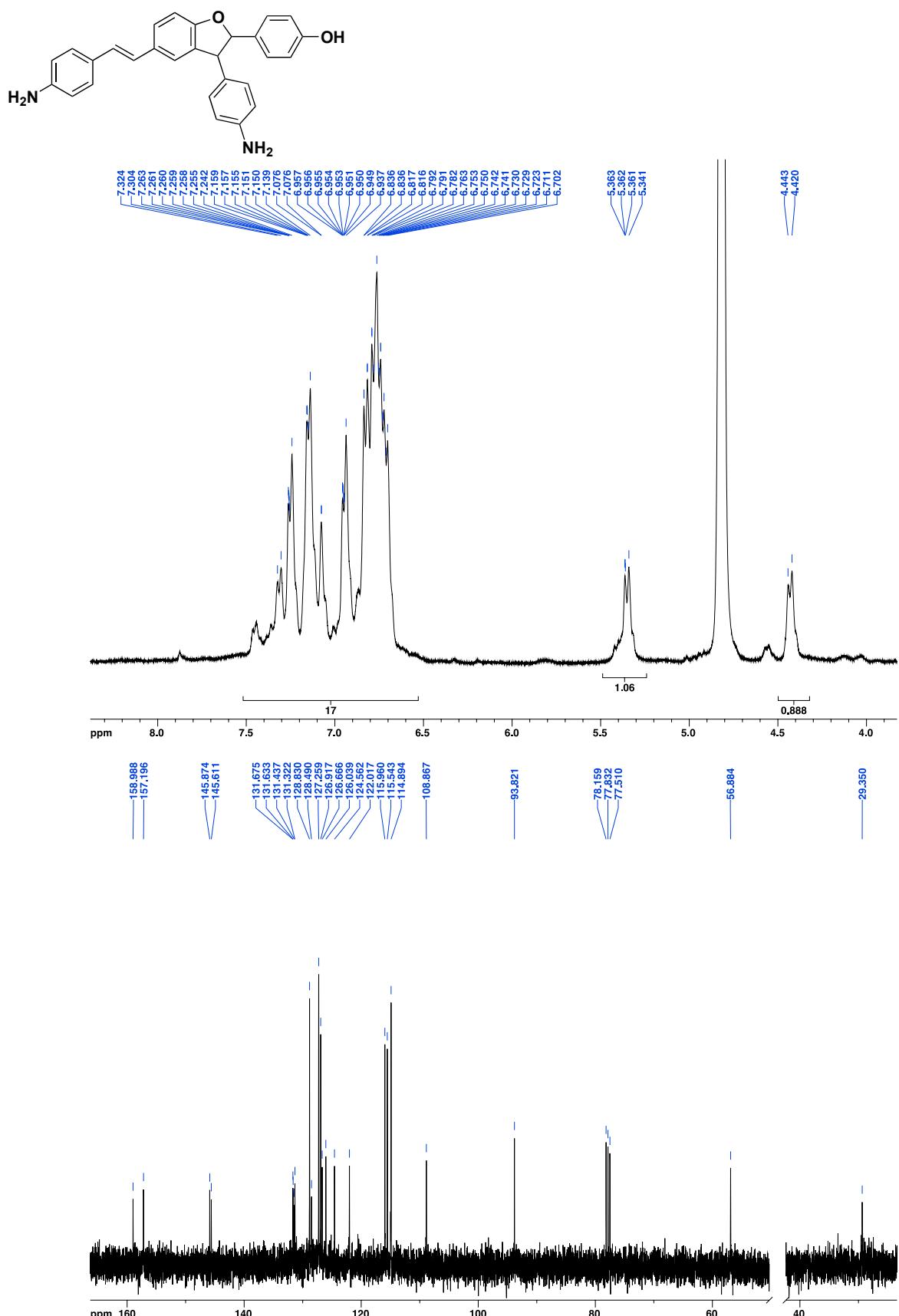
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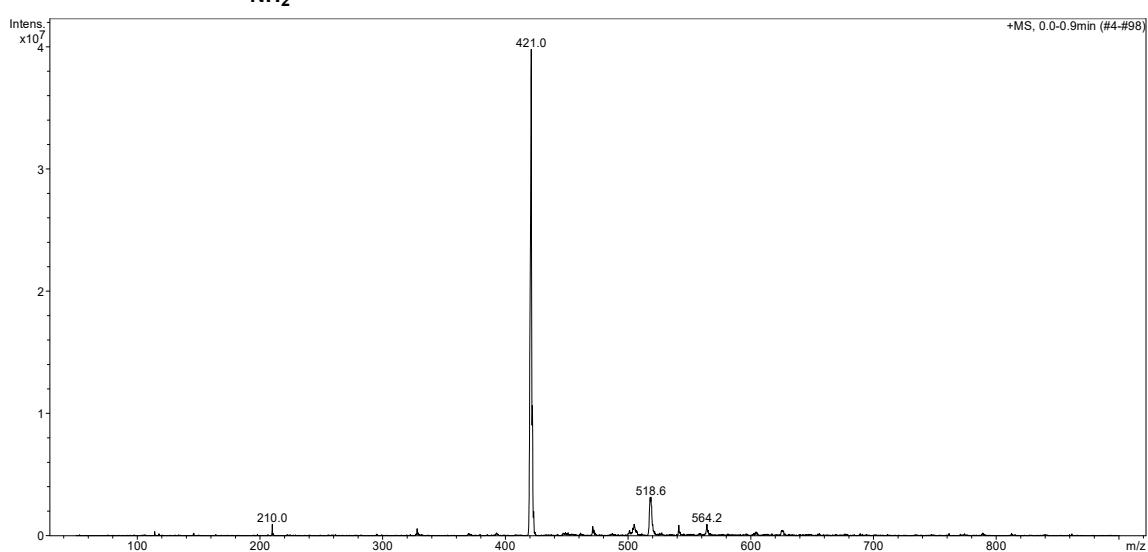
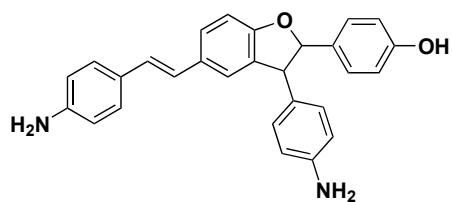
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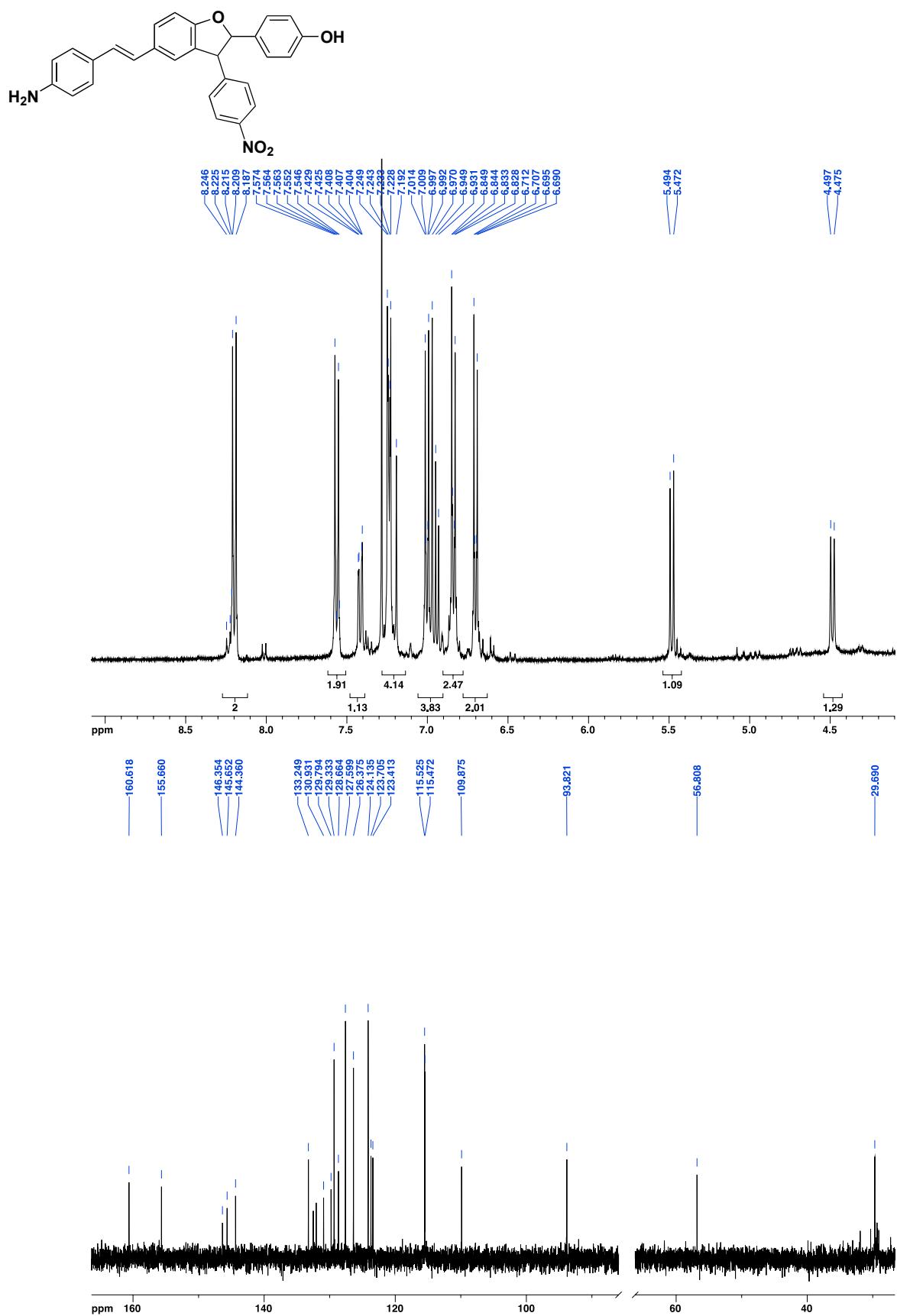
Compound 23



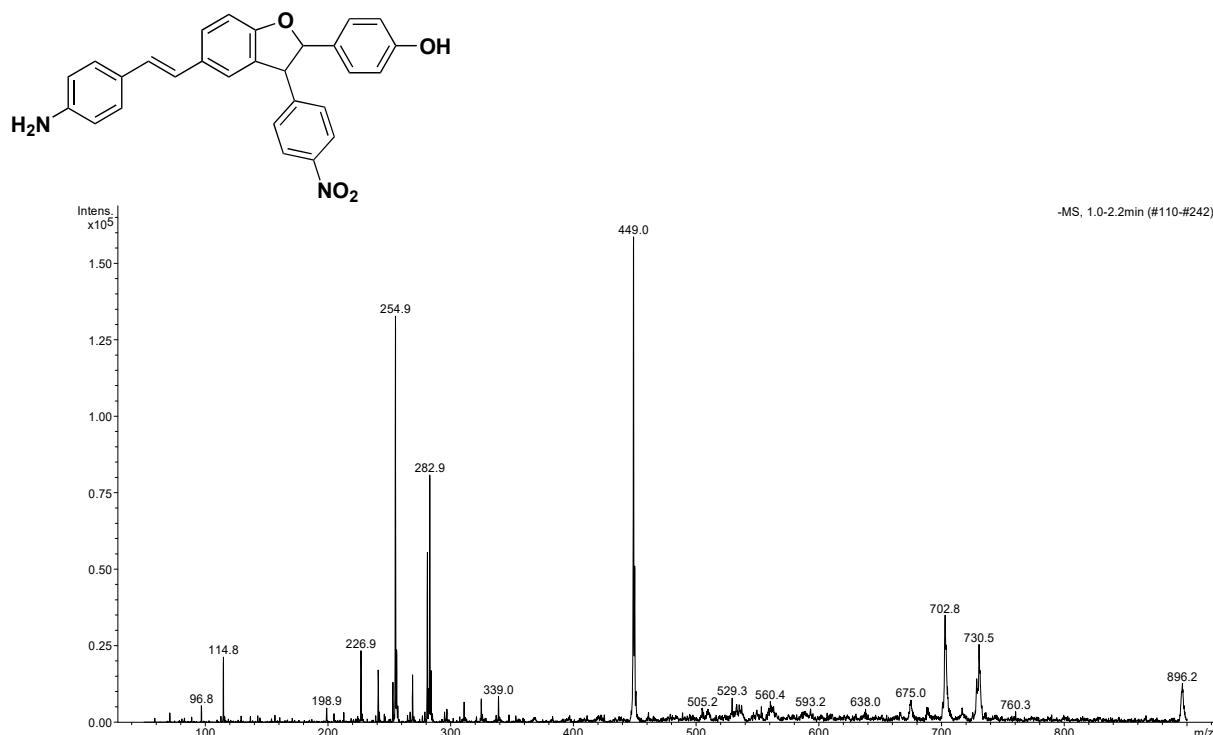
Compound 23



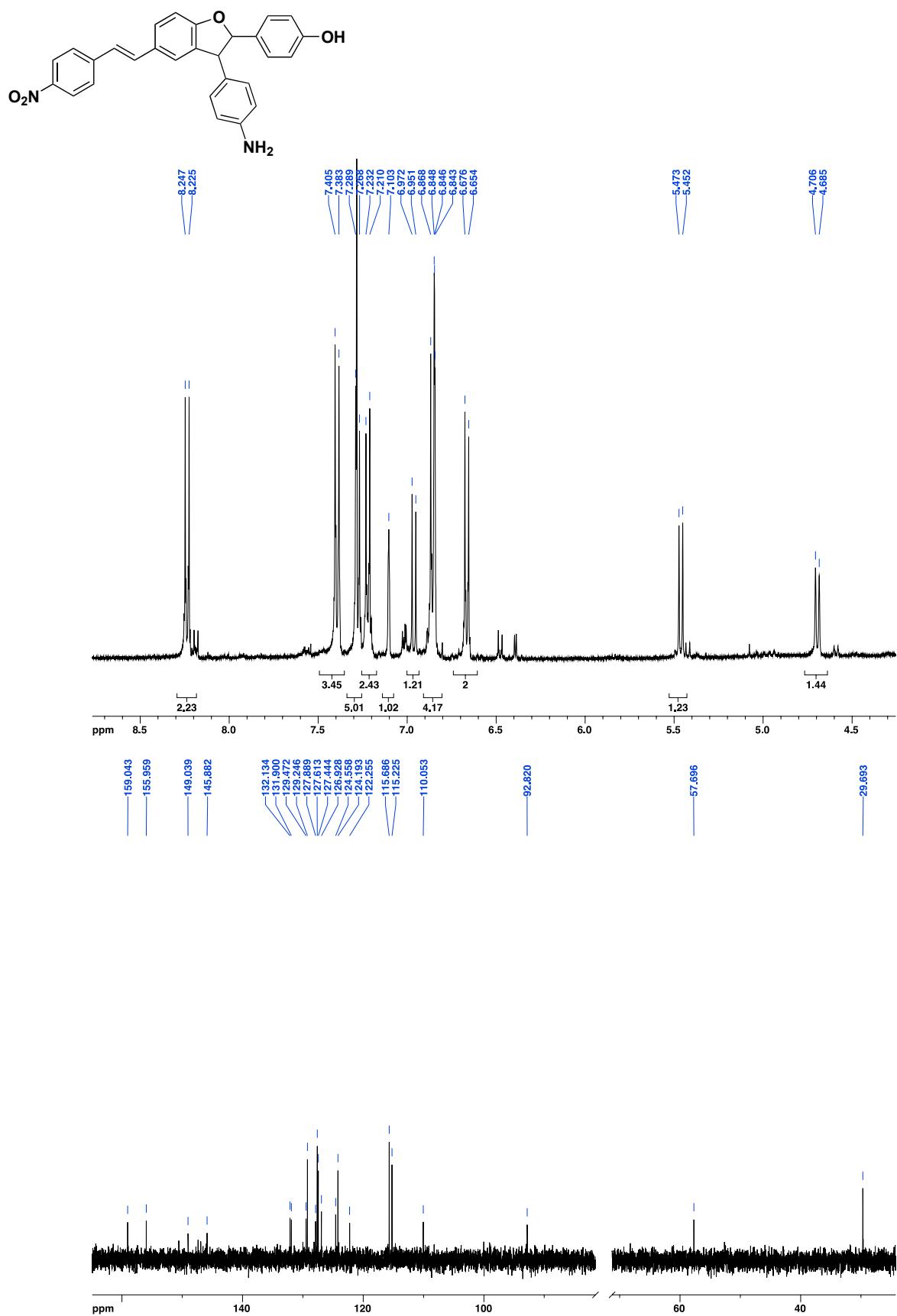
Compound 24



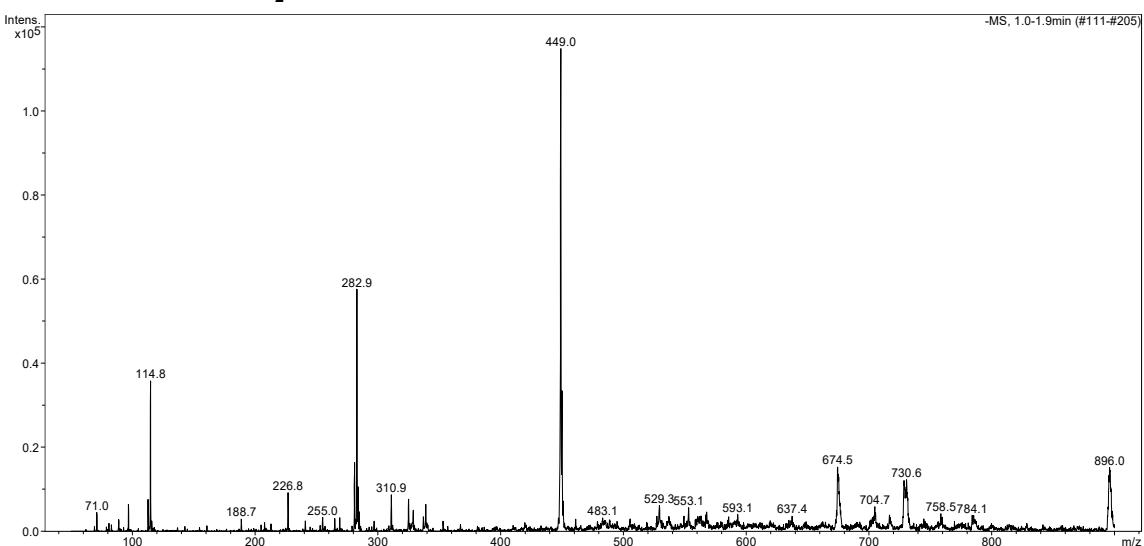
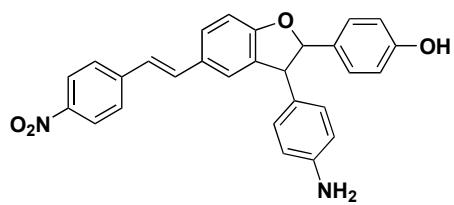
Compound 24



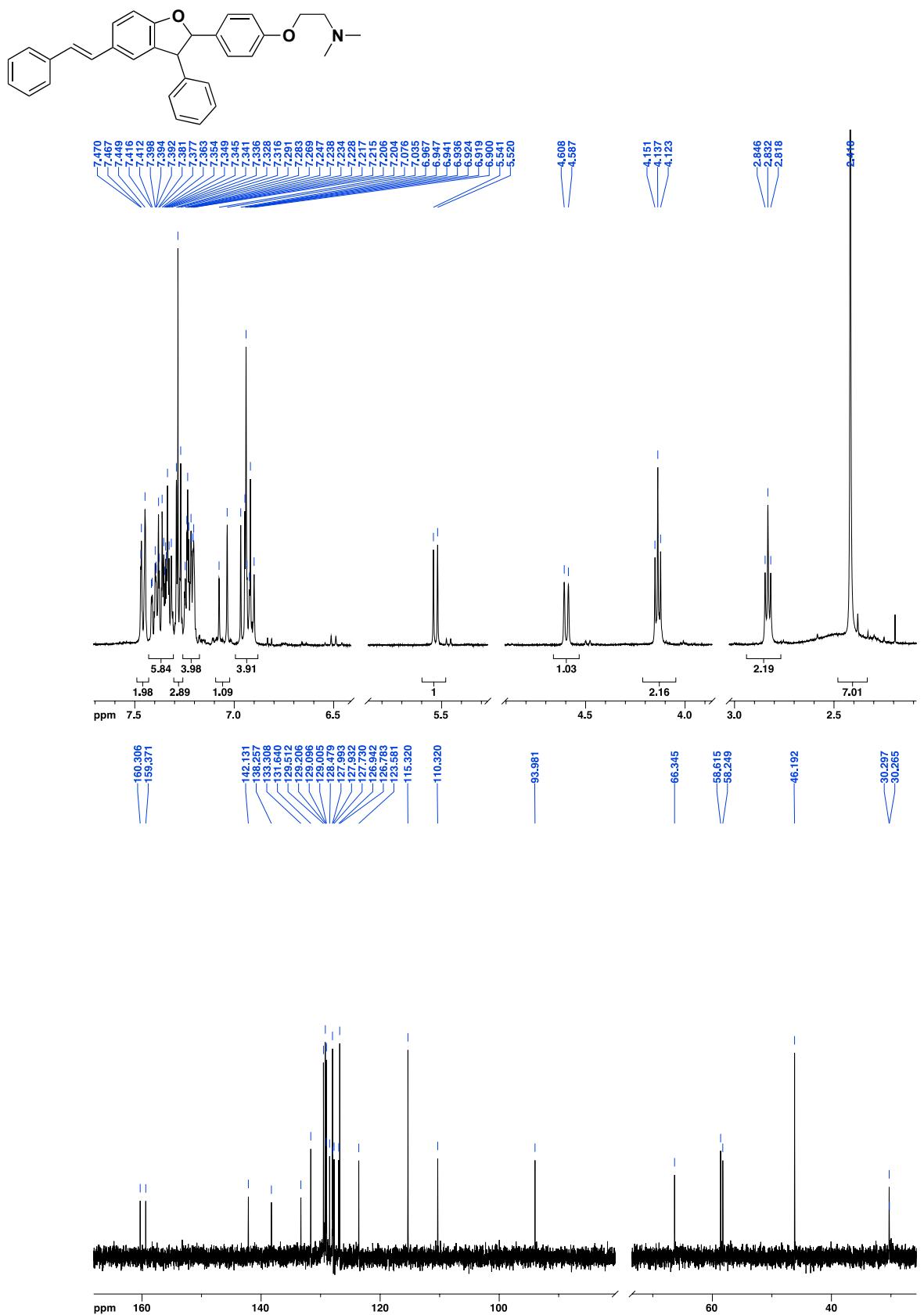
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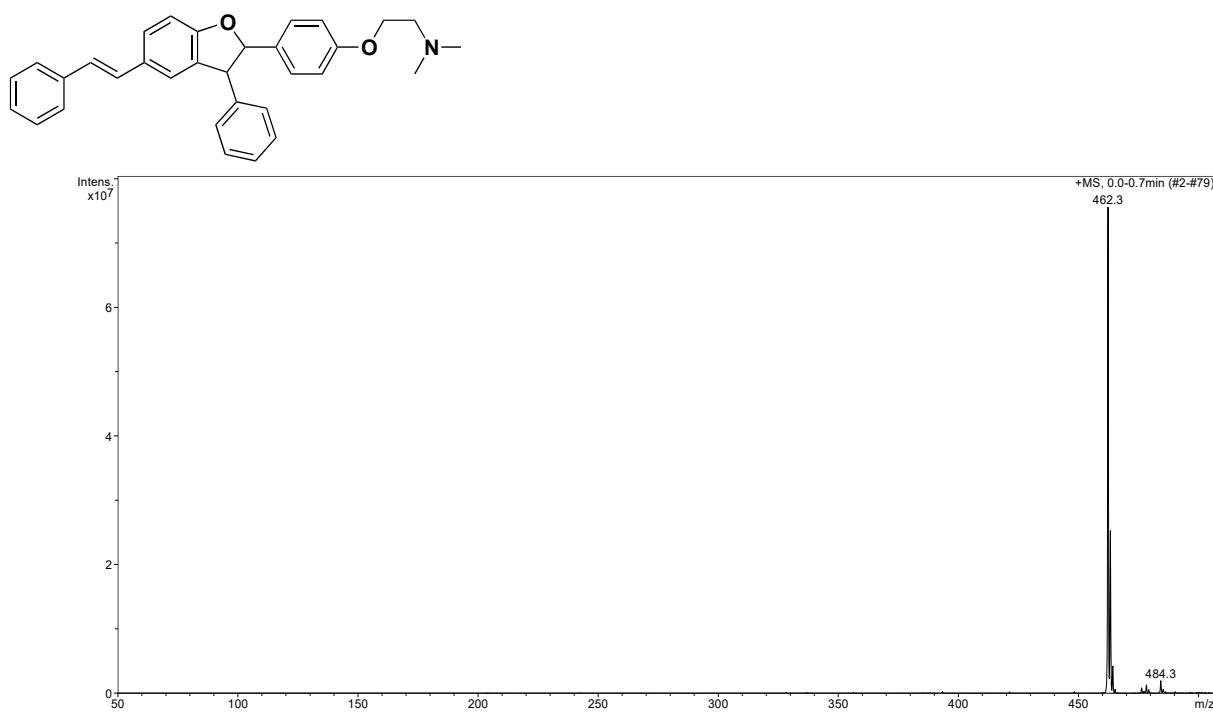
Compound 25



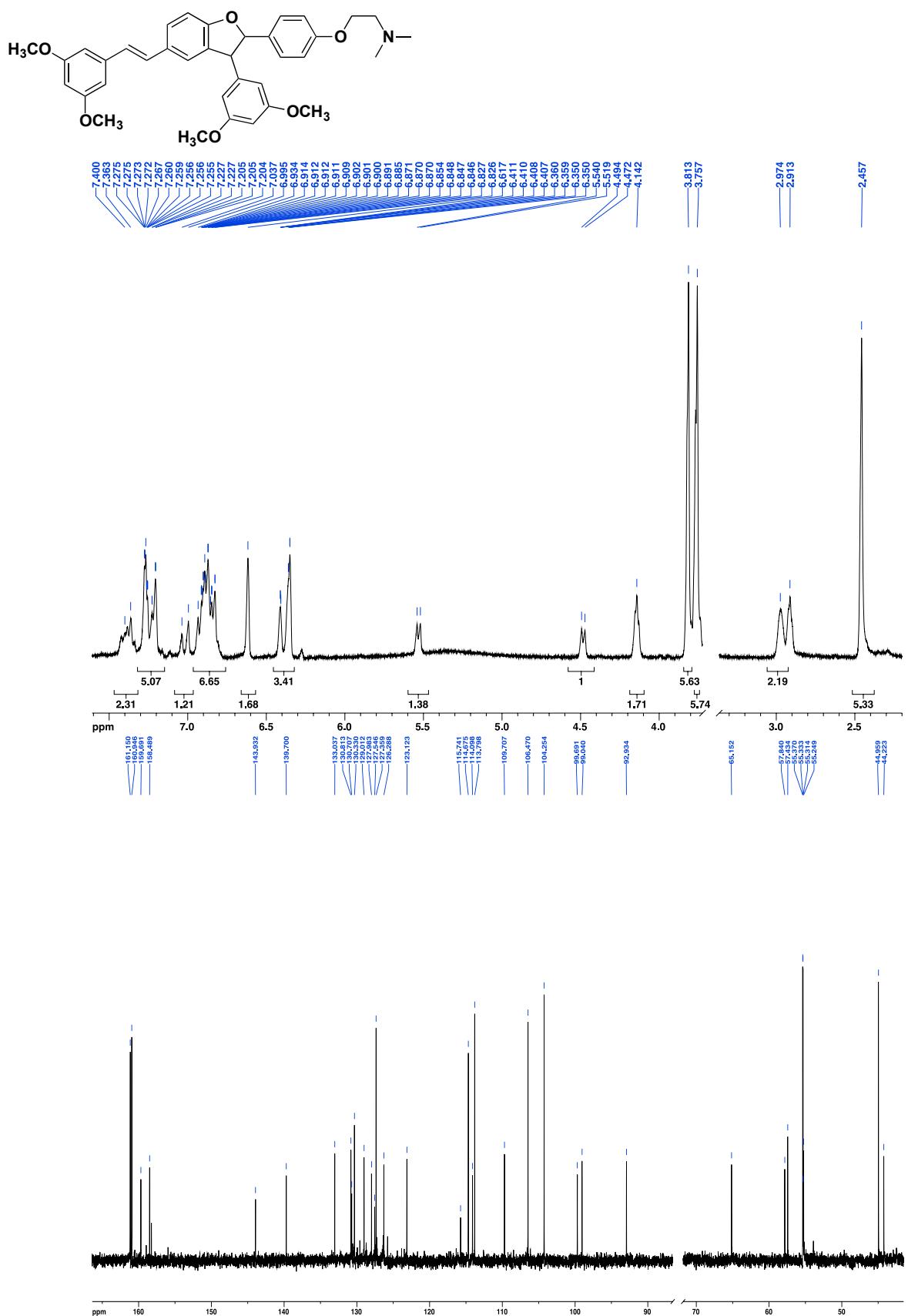
Compound 26



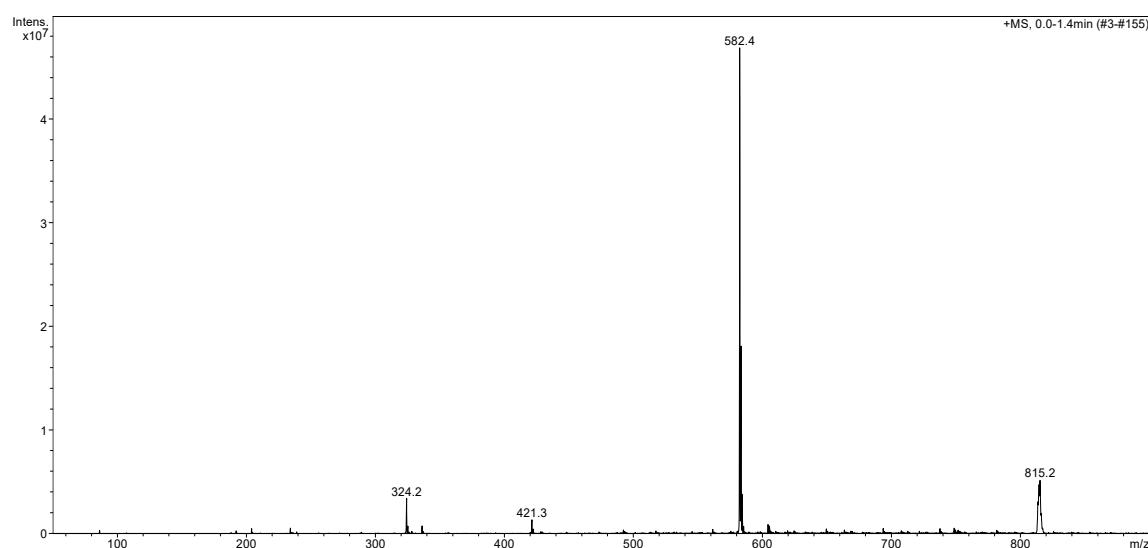
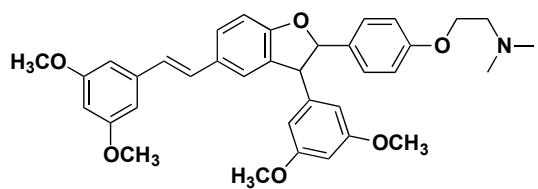
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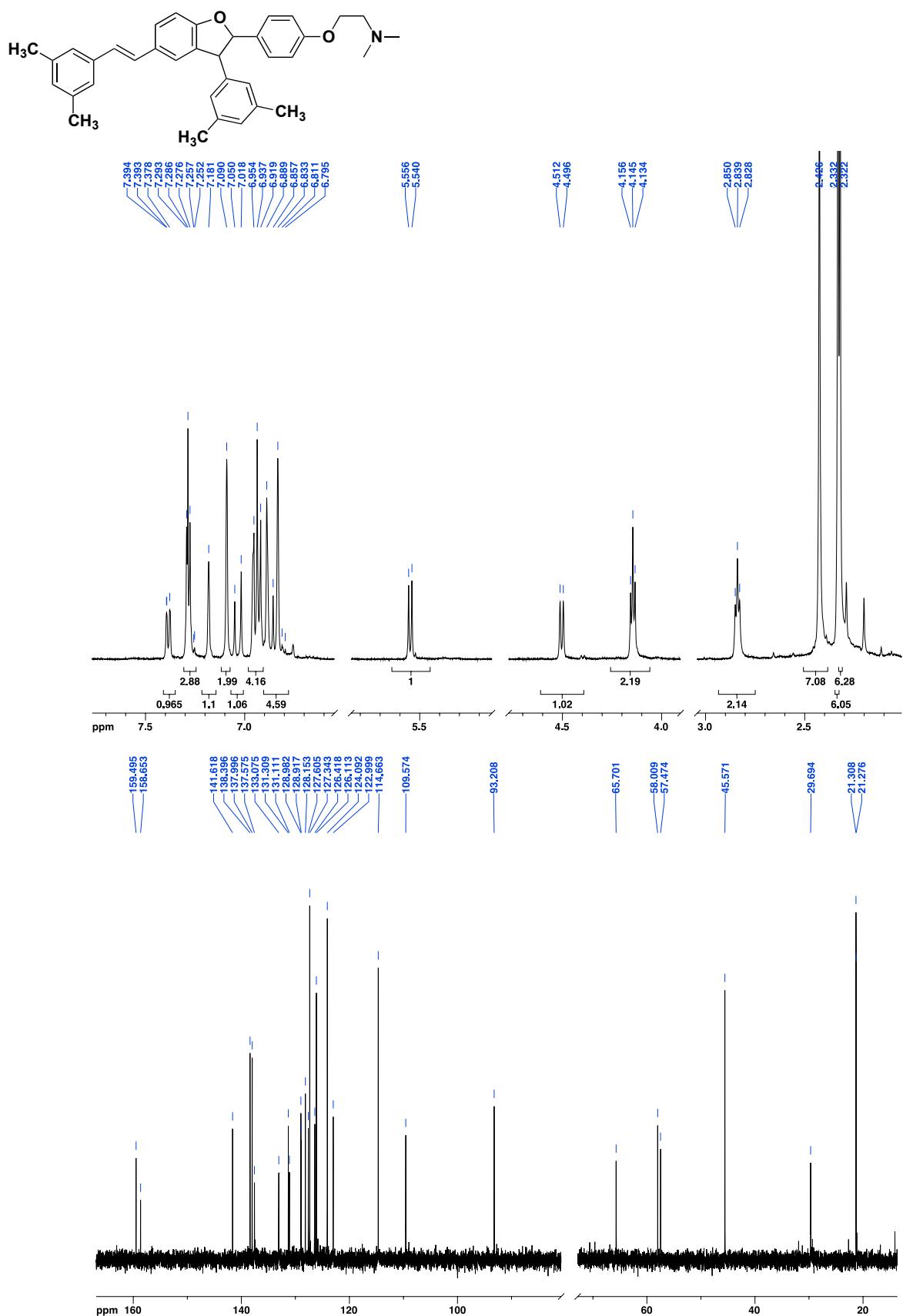
Compound 27



Compound 27



Compound 28



Compound 28

