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Supporting Information

Catalyst-Controlled Switchable Reactions of β-Keto Acids to Silyl Glyoxylates

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1. Asymmetric decarboxylative addition reaction of β-keto acids to silyl glyoxylate

1.1 General procedure for the asymmetric decarboxylative addition reaction of β-keto acids to silyl glyoxylates

The mixture of β -keto acid (1, 0.20 mmol) and benzyl 2-(*tert*-butyldimethylsilyl)-2-oxoacetate (2a, 0.10 mmol) in solvent (1.0 mL) were added catalyst (20 mol%). After stirring for the indicated time, the mixture was concentrated in vacuo and the residue was purified via flash chromatography to give the desired chiral product 4a.

1.2 Investigation of silicon substituent effect (20 mol%) Et₂O, -20 °C OBn OBn 1a 4 HO TMS HOTBS HO TIPS OBn ∠OBn OBn OBn Pł 51% ee NR 42% ee 78% ee

1.3 Investigation of the catalyst effect and solvent effect^a

	0 − − − +	TBS OBn -	cataly solver	st → nt	O C	HO TBS OBn 4a
Entry	Catalyst	Solvent	T (°C)	Time (h)	Yield (%) ^b	ee (%) ^c
1	VI	THF	rt	17	27	11
2	VII	THF	rt	17	35	8
3	VIII	THF	rt	17	50	-40
4	XI	THF	rt	17	46	0
5	XII	THF	rt	17	43	2



^{*a*} Reactions were performed with 0.2 mmol of **1**, 0.10 mmol of **2a**, and 20 mol% of the catalyst in 1.0 mL of solvent. ^{*b*} Isolated yield after purification by column chromatography. ^{*c*} Determined by chiral HPLC analysis.

2. ¹H and ¹³C NMR spectra of the products















































racemic



Peak	Retention Time (min)	Area	Height	% Area
1	10.181	38613.4	1141.2	49.696
2	19.202	39085.5	479.5	50.304

chiral



Peak	Retention Time (min)	Area	Height	% Area	
1	9.679	56017	1681.9	88.866	
2	19.504	7018.3	112.4	11.134	

































