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Supporting Information

Metal-free, Base Catalysed sp² C-H Functionalization in Sulfonamidation of 1,4-Naphthoquinones

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Experimental Section

General methods: All reactions were carried out under a nitrogen atmosphere, with dry, distilled solvents in anhydrous conditions. All chemicals (1,4-naphthoquinone, freshlv Menadione, sulfonyl chlorides, metal salts) were used without further purification as commercially available unless otherwise noted. NMR spectra were registered on Bruker DRX spectrometer operating at 300 & 400 MHz for 1H and 75 &100 MHz for ¹³C. All ¹H- and ¹³C-NMR spectra were measured in DMSO-d₆ and CDCl₃ with TMS as the internal standard. HRMS-ESI was obtained by using Orbitrap mass spectrometers and ESI-MS was recorded in Thermo Scientific mass spectrometer. All the compounds were purified by column chromatography using silica gel (60-120 mesh). Sulfonyl azides,¹⁷ were prepared according to the published methods.

General procedure for the base-mediated sulfonamidation reaction of 1,4-naphthoquinone (3): A mixture of 1,4-naphthoquinone 1 (1.0 mmol), 4-methylbenzenesulfonyl azide 2a (1.0 mmol), K_2CO_3 (1.0 mmol) DMSO (3.0 mL) were taken in a 15 mL reaction tube under an open-air atmosphere and the reaction tube was placed in a magnetic stirrer. The reaction mixture was stirred at room temperature for 15 minutes. After completion of the reaction, the solvent was removed under vacuum and the reaction mixture was quenched with water (25 mL) and extracted with dichloromethane (15 mL \times 3). The combined organic layers were dried with anhydrous Na₂SO₄ and the crude product was purified over a column of silica gel (eluent: petroleum ether/ethyl acetate = 10:5) to afford the desired product N-(1,4-Dioxo-1,4-dihydron aphthalen-2-yl)-4-methylbenzenesulfonamide **3a**.

General procedure for the base mediated sulfonamidation reaction of 2-methyl-1,4naphthoquinone (Menadione) 5: A mixture of 2-Methyl-1,4-naphthoquinone 4 (1.0 mmol), 4-methylbenzenesulfonyl azide 2a (1.0 mmol), K_2CO_3 (1.0 mmol) DMSO (3.0 mL) were taken in a 15 mL reaction tube under an open air atmosphere and the reaction tube was placed in a magnetic stirrer. The reaction mixture was stirred at room temperature for 30 minutes. After completion of the reaction, the solvent was removed under vacuum and the reaction mixture was quenched with water (25 mL) and extracted with dichloromethane (15 mL \times 3). The combined organic layers were dried with anhydrous Na₂SO₄ and the crude product was purified over a column of silica gel (eluent: petroleum ether/ethyl acetate = 10:5) to afford the desired product 4-Methyl-N-(3-methyl-1,4-dioxo-1,4-dihydronaphthalen-2-yl)benzenesulfonamide 5a.



Fig.S2 Absorbance and PL spectrum of (3a-3n)



Fig.S2Absorbance and PL spectra of (5a-5f)

| | | Abcomption movimum | | Emission Maximum |
|--|----------------|--------------------|--------------------------|------------------------|
| S. No | Compound | (nm) | Excitation (nm) | (nm) |
| | | (IIII) | $(\lambda_{excitation})$ | $(\lambda_{emission})$ |
| | Naphthoquinone | 331 | 340 | |
| 1. | 3a | 477 | 485 | 540, 573 |
| 2. | 3b | 479 | 490 | 545,582 |
| 3. | 3c | 479 | 490 | 542,575 |
| 4. | 3d | 479 | 490 | 541, 566 |
| 5. | 3e | 479 | 490 | 539, 569 |
| 6. | 3g | 334, 478 | 490 | 541,562,583 |
| 7. | 3h | 468 | 480 | 540, 572 |
| 8. | 3i | 473 | 480 | 542, 577 |
| 9. | 3ј | 476 | 485 | 545, 578 |
| 10. | 3k | 460 | 470 | 535, 603 |
| 11. | 31 | 455 | 465 | 542, 575 |
| 12. | 3m | 488 | 500 | 540, 569 |
| 13. | 3n | 488 | 500 | 540, 573 |
| 14. | Menadione | 332 | 340 | |
| 15. | 5a | 506 | 515 | 566, 591 |
| 16. | 5b | 503 | 510 | 560, 598 |
| 17. | 5c | 502 | 510 | 561, 587 |
| 18. | 5d | 501 | 510 | 581 |
| 19. | 5e | 386, 496 | 505 | 564 |
| 20. | 5f | 485 | 495 | 561, 585 |
| Absorbance and Pl spectrum are studied in 1 X 10 ⁻⁴ M concentration and solutions are | | | | |
| made in DMSO as solvent. | | | | |

Table S1. Absorbance and fluorescence data of sulfonamidation derivatives ofmenadione and 1,4-naphthoquinone.

| Compound First redox wave | | Second redox wave | | Third redox wave | | |
|---|---------------------|----------------------|---------------------|---------------------|---------------------|---------------------|
| | E _{pc/Epa} | $E_{1/2} (\Delta E)$ | E _{pc/Epa} | $E_{1/2}(\Delta E)$ | E _{pc/Epa} | $E_{1/2}(\Delta E)$ |
| Naphthoquinone | -0.75/-0.41 | -0.58 (343) | -1.40/-1.04 | -1.22 (365) | - | - |
| 3a | -0.82/-0.64 | -0.73 (182) | -1.17/-0.93 | -1.05 (241) | -1.60/-1.25 | -1.42 (354) |
| 3ј | -0.81/-0.57 | -0.69 (241) | -1.14/-0.93 | -1.03 (210) | -1.75/-1.29 | -1.52 (458) |
| 3e | -0.80/-0.62 | -0.71 (187) | -1.14/-0.97 | -1.05 (168) | -1.76/-1.36 | -1.56 (401) |
| Menadione | -0.77/-0.47 | -0.62 (303) | -1.24/-0.88 | -1.06 (356) | -1.44/-1.07 | -1.25 (369) |
| 5a | -0.58/-0.76 | -0.67 (177) | -0.75/-1.15 | -0.94 (405) | -1.00/-1.58 | -1.29 (579) |
| 5c | -0.80/-0.59 | -0.69 (212) | -1.17/-0.96 | -1.06 (206) | -1.53/-1.22 | -1.37 (311) |
| 5f | -0.62/-0.80 | -0.71 (178) | -0.89/-1.05 | -0.97 (164) | -1.30/-1.25 | -1.27 (56) |
| ^[a] DMSO; I= 0.1m n-Bu ₄ NClO ₄ . (Epc and Epa (V), E1/2= (Epc+ Epa)/2 (V). v= 200 mV s ⁻¹ ; $\Delta E = E_{pa}-E_{pc}$ (mV); | | | | | | |
| reference electrode= KCl(3m)/Ag/AgCl; working electrode= glassy carbon disk of 0.07 cm ² area; auxiliary | | | | | | |
| electrode= Pt wire. | | | | | | |

Table S2. Electrochemical data measured using cyclic voltammetry (CV) for 1,4-naphthoquinone and menadione sulfonamidation derivatives in selective.







| $ \begin{array}{c} $ | <i>N</i> -(<i>1</i> , <i>4</i> - <i>Dioxo</i> - <i>1</i> , <i>4</i> - <i>dihydronaphthalen</i> - <i>2</i> - <i>yl</i>) <i>methanesulfonamide</i> (table 2, 3m); Yellow colored solid; m.p 178 °C; yield 58%; ¹ H-NMR (300 MHz, DMSO- <i>d</i> ₆) δ 8.04 (d, <i>J</i> = 7.1 Hz, 1H), 7.98 (d, <i>J</i> = 7.3 Hz, 1H), 7.81 (dd, <i>J</i> = 15.2, 7.4 Hz, 2H), 6.80 (s, 1H), 3.26 (s, 3H). ¹³ C- NMR (75 MHz, DMSO- <i>d</i> ₆) δ 182.03, 178.10, 140.52, 133.15, 132.92, 131.81, 130.11, 128.77, 125.60, 124.89, 124.24, 112.10, 77.44, 77.00, 76.56, 39.17, 38.89, 38.81, 38.61, 38.33, 38.05, 37.78, 37.49; HRMS (ESI): <i>m</i> / <i>z</i> calculated for C ₁₁ H ₁₀ O ₄ NS [M + H] ⁺ 252.0325, found: 252.0324. |
|--|---|
| $ \begin{array}{c} $ | <i>N</i> -(<i>1</i> , <i>4</i> - <i>Dioxo</i> - <i>1</i> , <i>4</i> - <i>dihydronaphthalen</i> - <i>2</i> - <i>yl</i>) <i>cyclopropanesulfonamide</i> (table 2, 3n); Yellow colored solid; m.p 172 °C; yield 59%; ¹ H-NMR (300 MHz, CDCl ₃) δ 9.96 (s, 1H), 8.08 (d, <i>J</i> = 7.2 Hz, 1H), 8.01 (d, <i>J</i> = 6.7 Hz, 1H), 7.80 (dd, <i>J</i> = 14.5, 7.5 Hz, 2H), 6.85 (s, 1H), 2.94 (dd, <i>J</i> = 8.4, 3.9 Hz, 1H), 1.27 (dd, <i>J</i> = 6.5, 4.1 Hz, 2H), 1.09 (dd, <i>J</i> = 12.2, 6.6 Hz, 2H). ¹³ C-NMR (75 MHz, DMSO- <i>d</i> ₆) δ 184.55, 180.49, 143.33, 135.71, 134.55, 132.19, 131.24, 127.24, 126.41, 114.44, 31.36, 6.37; HRMS (ESI): <i>m</i> / <i>z</i> calculated for C ₁₃ H ₁₂ O ₄ NS [M+H] ⁺ 278.0482, found: 278.0494. |
| $ \begin{array}{c} $ | 4-Methyl-N-(3-methyl-1,4-dioxo-1,4-dihydronaphthalen-2- yl)benzenesulfonamide (table 3, 5a); Yellow colored solid; m.p 207- 210 °C; yield 76%; ¹ H-NMR (300 MHz, CDCl ₃) δ 9.19 (s, 1H), 8.05 (d, $J = 5.0$ Hz, 1H), 7.90 (d, $J = 4.3$ Hz, 2H), 7.71 (dd, $J = 14.7$, 5.1 Hz, 5H), 2.41 (d, $J = 4.5$ Hz, 3H), 2.24 (d, $J = 4.8$ Hz, 3H). ¹³ C NMR (75 MHz, DMSO- d_6) δ 185.48, 180.98, 143.99, 142.34, 139.76, 139.21, 135.15, 134.93, 132.38, 131.36, 130.28, 127.51, 126.91, 21.91, 14.27; HRMS (ESI): m/z calculated for C ₁₈ H ₁₅ O ₄ NNaS [M+ Na] ⁺ 364.0614, found: 364.0614. |
| O CH ₃ O N S O 5b | <i>N-(3-Methyl-1,4-dioxo-1,4-dihydronaphthalen-2-yl)benzenesulfonamide</i> (table 3, 5b); Yellow colored solid; m.p 186 °C; yield 74%; ¹ H NMR (300 MHz, DMSO- <i>d</i> ₆) δ 9.88 (s, 1H), 8.00 (d, <i>J</i> = 6.9 Hz, 1H), 7.86 (d, <i>J</i> = 7.8 Hz, 3H), 7.81 – 7.73 (m, 2H), 7.61 (d, <i>J</i> = 7.0 Hz, 1H), 7.53 (t, <i>J</i> = 7.3 Hz, 2H), 2.13 (s, 3H). ¹³ C-NMR (75 MHz, DMSO- <i>d</i> ₆) δ 182.68, 178.21, 139.70, 139.26, 137.11, 132.25, 132.00, 130.83, 129.74, 128.67, 126.97, 124.78, 124.20, 11.59; HRMS (ESI): <i>m</i> / <i>z</i> calculated for C ₁₇ H ₁₃ NNaO ₄ S [M+Na] ⁺ 350.0463, found: 350.0469. |



| O O H O H O H O H O H O H O H O H O H O | <i>N</i> -(<i>8</i> - <i>Hydroxy</i> - <i>1</i> , <i>4</i> - <i>dioxo</i> - <i>1</i> , <i>4</i> - <i>dihydronaphthalen</i> - <i>2</i> - <i>yl</i>)- <i>4</i> - <i>methylbenzenesulfonamide</i> (table 2, 7): Yellow solid; yield 76%; ¹ H NMR (300 MHz, CDCl ₃): δ 11.81 (s, 1H), 7.75 (d, J = 12.0 Hz, 2H), 7.54 (d, J = 8.0 Hz, 2H), 7.32 (d, J = 12 Hz, 2H), 7.19 (d, J = 8.0 Hz, 1H), 6.86 (s, 2H), 2.39 (s, 3H). ¹³ C NMR (75 MHz, CDCl ₃) δ 190.19, 184.13, 161.32, 146.12, 139.48, 138.54, 136.46, 135.42, 131.66, 130.19, 127.43, 124.37, 119.03, 114.06, 21.15; ESI-MS: <i>m</i> / <i>z</i> calculated for C ₁₇ H ₁₄ NO ₅ S [M+H] ⁺ 344.05, found: 344.15. |
|--|---|
| NH ₂ 8 | 2-Aminonaphthalene-1,4-dione (Scheme 1, 8): Orange colored solid; yield $(56\%)^{1}$ H NMR $(500 \text{ MHz}, \text{CDCl}_{3}) \delta 8.06 (d, J = 11.5, 2H), 7.72 (d, J = 7.6, 1H), 7.63 (d, J = 7.6, 1H), 6.01 (s, 1H), 5.32 (s, 2H);^{13}C NMR (126 \text{ MHz}, \text{CDCl}_{3}) \delta 183.49, 182.01, 149.11, 134.40, 133.45, 131.95, 130.53, 126.03, 125.86, 104.21; (ESI): m/z calculated for C_{10}H_7NO_2 [M+H]^+ 173.04, found:174.10.$ |
| O CH ₃ NH ₂ 9 | 2-Amino-3-methylnaphthalene-1,4-dione (Scheme 1, 9): Orange colored solid; yield (60%) ¹ H NMR (400 MHz, CDCl ₃) δ 8.08 (d, $J =$ 7.6, 1H), 8.01 (d, $J =$ 7.6, Hz, 1H), 7.68 (t, $J =$ 7.6, 1H), 7.59 (t, $J =$ 7.5, 1H), 5.03 (s, 2H), 2.01 (s, 3H). ¹³ C NMR (101 MHz, CDCl ₃) δ 182.83, 181.13, 145.32, 134.17, 131.97, 126.15, 125.72, 113.12, 9.17; HRMS (ESI): m/z calculated for C ₁₇ H ₁₃ N ₂ O ₆ S[M+H] ⁺ 373.0489, found: 373.0495. |







Fig. 6. ¹³C-NMR spectrum of N-(1,4-dioxo-1,4-dihydronaphthalen-2-yl)biphenyl-4-sulfonamide (Table 2, **3c**).





























Fig. 33. ¹H-NMR spectrum of 4-Bromo-N-(3-methyl-1,4-dioxo-1,4-dihydronaphthalen-2-yl)benzenesulfonamide (Table 3, **5d**).





















References

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