Electronic Supplementary Material (ESI) for Organic & Biomolecular Chemistry. This journal is © The Royal Society of Chemistry 2018

# "Choose-a-Size" Control in the Synthesis of Sucrose Based Urea and Thiourea Macrocycles

Katarzyna Łęczycka-Wilk,\* Filip Ulatowski, Piotr Cmoch, Sławomir Jarosz\*

Institute of Organic Chemistry, Polish Academy of Sciences, Kasprzaka 44/52, 01-224 Warsaw, Poland. Contact e-mail: SJ: slawomir.jarosz@icho.edu.pl, KŁW: kleczycka@icho.edu.pl

# **Supporting Information**

### **Contents**

1.	SYNTHETIC PROCEDURES	2
2.	COPIES OF THE NMR SPECTRA	3
3.	TITRATION EXPERIMENTS	15
4.	SINGLE CRYSTAL X-RAY MEASUREMENT	20
5.	REFERENCES	20

1. The synthesis of monomeric macrocyclic derivative 3 in a flow mode



2. The synthesis of monomeric macrocyclic derivative 5 in a flow mode



# 2. Copies of NMR spectra



Fig. S1. <sup>1</sup>H NMR (600 MHz) and <sup>13</sup>C NMR (151 MHz) spectra of compound 2a in DMF- $d_6$ .





Fig. S2. COSY ( $^{1}H-^{1}H$ ) spectra of compound 2a in DMF- $d_{6}$ .



**Fig. S3.** HSQC ( $^{1}H^{-13}C$ ) spectra of compound **2a** in DMF-*d*<sub>6</sub>.



Fig. S4. <sup>1</sup>H NMR (600 MHz) and <sup>13</sup>C NMR (151 MHz) spectra of compound 2b in acetone-d<sub>6</sub>.



b)



**Fig. S5.** COSY ( $^{1}H^{-1}H$ ) spectra of compound **2b** in acetone- $d_{6}$ .



**Fig. S6.** HSQC ( $^{1}H^{-13}C$ ) spectra of compound **2b** in acetone- $d_{6}$ .



Fig. S7. <sup>1</sup>H NMR (600 MHz) and <sup>13</sup>C NMR (151 MHz) spectra of compound 3 in acetone- $d_6$ .





**Fig. S8.** COSY ( $^{1}H^{-1}H$ ) spectra of compound **3** in acetone- $d_{6}$ .



**Fig. S9.** HSQC ( $^{1}\text{H}-^{13}\text{C}$ ) spectra of compound **3** in acetone- $d_{6}$ .



Fig. S10. <sup>1</sup>H NMR (600 MHz) and <sup>13</sup>C NMR (151 MHz) spectra of compound 3 in acetone- $d_6$ .



**Fig. S11.** COSY ( $^{1}H^{-1}H$ ) spectra of compound **5** in acetone-*d*<sub>6</sub>.



**Fig. S12.** HSQC ( $^{1}H^{-13}C$ ) spectra of compound **5** in acetone- $d_{6}$ .

## 3. Titration experiments

All solutions were prepared in the HPLC gradient-grade acetronitrile. TBA salts (TBACl, TBAH<sub>2</sub>PO<sub>4</sub>, CH<sub>3</sub>COOTBA, and PhCOOTBA) were dried overnight at room temperature under high vacuum (1 mbar) directly before use. A 1 cm cuvette was filled with ~2.2 mL of the host solution and aliquots of the guest solution were added by a syringe pump. UV-Vis spectrum was recorded after each step once the mixture was homogenised by magnetic stirrer (30 s). The collected data were then analysed by HyperSpec. In case of hosts **3** and **5**, addition of guest had no effect on the spectra. In case of **2a** and **2b** the data were fitted with 1:1 and 1:1+1:2 models. Low values of the residuals are obtained with simple 1:1 model and no significant improvement in fitting quality is observed with 1:1+1:2 model. The residuals are most likely to arise from small errors in concentrations of the reagents. .



Fig. 13 Plots of UV-Vis titration of host 2a with TBACl.



Fig. 14 Plots of UV-Vis titration of host 2a with AcOTBA.



Fig. 15 Plots of UV-Vis titration of host 2b with TBACl.



Fig. 16 Plots of UV-Vis titration of host 2b with AcOTBA.

# 4. Single crystal X-ray measurement

Single crystal X-ray diffraction measurements were carried out on a Agilent Supernova diffractometer, at 100K with monochromated Mo K $\alpha$  radiation (0.7107Å). The data reduction was made by using CrysAlisPRO [1] software. The structures were solved by direct methods and refined on F<sup>2</sup> by full-matrix least-squares by using SHELXS97 and SHELXL97 [2]. All non-hydrogen atoms were refined as anisotropic while hydrogen atoms were placed in calculated positions, and refined in riding mode.

**Crystal of 3.** orthorhombic, P212121, a= 11.1611(7), b=18.4375(11), c= 23.2735(12)Å, V= 4789.3(5)Å <sup>3</sup>, Z=4,  $D_{calc}$ =1.385g cm<sup>-1</sup>,  $\mu$ =1.782 mm<sup>-1</sup>, R1=0.0635 for 6643 [ Fo > 4 $\sigma$ (Fo)] and 0. 1075 for all data, wR2=0. 1795, S=0. 911

CCDC 1584391 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via <a href="http://www.ccdc.cam.ac.uk/data\_request/cif">www.ccdc.cam.ac.uk/data\_request/cif</a>

#### 5. References

- [1] Agilent (2011). CrysAlis PRO. Agilent Technologies, Yarnton, England.
- [2] Sheldrick, G. M. (2008) Acta Cryst. A64 112-122.