Grafting of Porphyrin Oligomers on Single-Walled Carbon Nanotubes by

Hay Coupling

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NMR and mass spectra of porphyrin 3 and 4.



Fig. S1 MALDI-TOF mass spectra of a) Zn-porphyrin 3 and b) Zn-porphyrin 4.



Fig. S2 ¹H NMR spectrum of free-base porphyrin $\bf{6}$ in CDCl₃.



Fig. S3 ¹³C NMR spectrum of free-base porphyrin 6 in CDCl₃.



Fig. S4 ¹H NMR spectrum of free-base porphyrin 7 in CDCl₃.



Fig. S5 $^{\rm 13}{\rm C}$ NMR spectrum of free-base porphyrin 7 in CDCl₃.



Fig. S6 ¹H NMR spectrum of zinc porphyrin 8 in CDCl₃.



Fig. S7 $^{\rm 13}C$ NMR spectrum of zinc porphyrin 8 in CDCl_3.



Fig. S8 ¹H NMR spectrum of zinc porphyrin 9 in CDCl₃.



Fig. S9 ¹³C NMR spectrum of zinc porphyrin 9 in CDCl₃.



Fig. S10 ^1H NMR spectrum of zinc porphyrin 3 in CDCl_3.



Fig. S11 ¹³C NMR spectrum of zinc porphyrin 3 in CDCl₃.



Fig. S12 ¹H NMR spectrum of zinc porphyrin 4 in pyridine-d5.



Fig. S13 Enlargement of the Raman spectra of *f*-SWNT **5**, **SWNT-ZnP 1**, **SWNT-ZnP 2**, ZnP **10** and ZnP **11**. On the left part, the spectra show that the D-bands of SWNT-ZnP **1** and SWNT-ZnP **2** are hidden by the signal of the porphyrin oligomers. On the right part, the spectra show no significant doping effect of the porphyrin oligomer on the G-band for **SWNT-ZnP 2** compared to *f*-SWNT **5** (shift of 1 cm⁻¹). For **SWNT-ZnP 1** (top right), an apparent shift of 3 cm⁻¹ is observed; this shift is likely due to the presence of a peak arising from the porphyrin oligomer **SWNT-ZnP 1**.