

Supporting Information

to

Experimental and Computational Study of Isomerically Pure Soluble Azaphthalocyanines and their Complexes and Boron Azasubphthalocyanines of Varying Number of Aza Units

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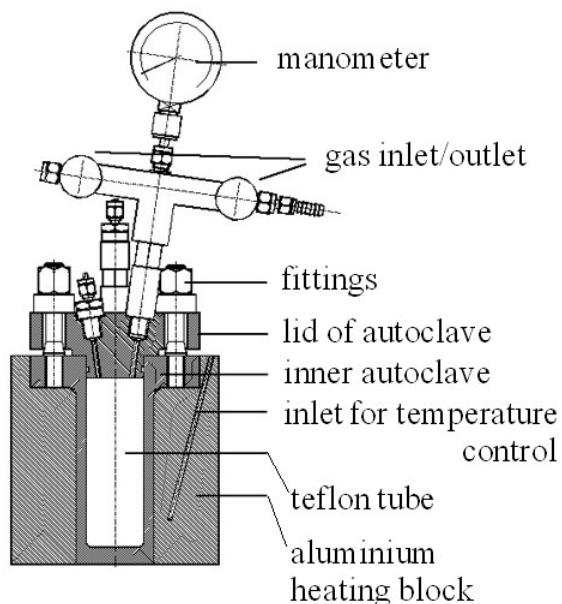
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1.1 Experimental Section

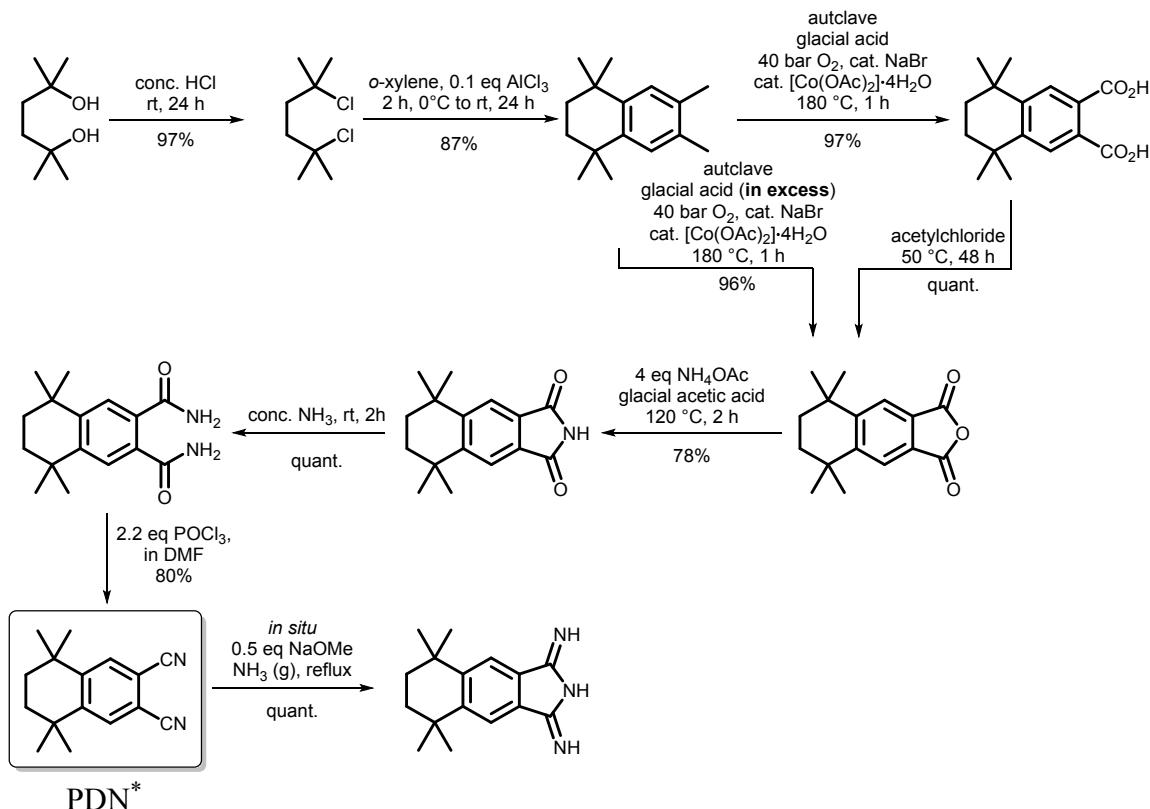
All reactions in which high pressure was needed, were carried out in an autoclave. Therefore, a glass (pressure <10 bar) or a steel autoclave (<100 bar) was used. The 100 mL stainless steel autoclaves *V4A-Edelstahlautoklaven* were constructed by the precision engineering department of the Philipps-Universität Marburg (Department of Chemistry). Solids and liquids were weighed out in a 100 mL teflon flask or in the corresponding glass flask, a magnetic stirrer was added and the autoclave was closed and sealed using EPDM-O-rings of CLEFF DICHTUNG company. The actual pressure was regulated using a manometer. The temperature was adjusted using a corresponding aluminium block in which the thermometer of the magnetic stirrer MR 3003 HEIDOLPH could be added (Pt-100-thermal element). If necessary, the heating block might be preheated (for *Pc* reactions). If necessary, the glass autoclaves were heated in an oil bath. A sketch of the steel autoclave is displayed below.



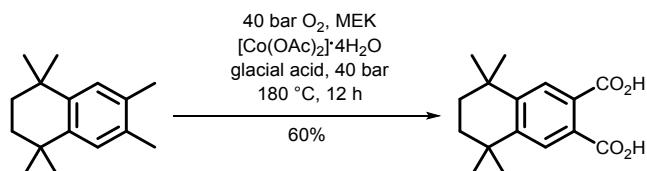
Sketch of the used steel autoclave: *V4A-stainless steel autoclave*.

1.1.1 Synthesis of PDN*

The synthesis of the PDN* **1** was carried out according to MIKHALENKO.^[1] The synthesis of 1,1,4,4-tetramethyltetralin-6,7-dicarboxylic acid (VI) was modified by using an overpressure of O₂ in a [Co(OAc)₂]·4H₂O catalytic autoclave reaction.

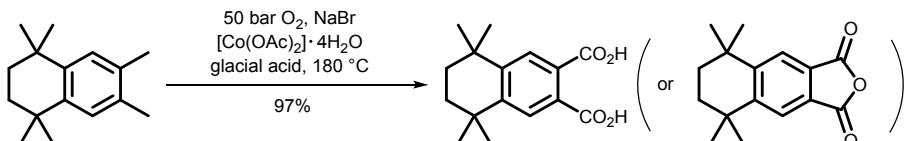


1.1.2 Synthesis of 1,1,4,4-Tetramethyltetralin-6,7-dicarboxylic acid (TTDA)



Method a) The reaction was carried out in an autoclave reaction as described in method b), using 50 bar O₂ and 0.1 eq [Co(OAc)₂]·4H₂O as catalyst, 5 eq MEK as cocatalyst instead of NaBr, and 5 mL glacial acetic acid /g tetraline as solvent. The mixture was stirred for 12 h while the autoclave was recharged several times with O₂, until the pressure stayed constant. Workup was carried out according to method b).

Yield: 60% (only carboxylic acid was observed). - **¹H NMR:** according to analysis below.

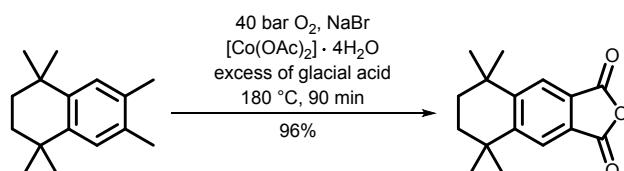


Method b) 15 g tetraline (69.3 mmol, 1 eq) were dissolved in 50 mL glacial acid in a 100 mL teflon tube and added into an autoclave. The teflon tube was additionally filled-up with 750 mg $[\text{Co}(\text{OAc})_2] \cdot 4\text{H}_2\text{O}$ (3 mmol, 4 mol%) and 300 mg NaBr (3 mmol, 4 mol%). The mixture was briefly stirred until everything was dissolved. The autoclave was charged with O_2 to a pressure of 20 bar. With stirring, the solution was heated to 180 °C. After a short period of initiation (<30 min) the reaction started and the pressure decreased. The autoclave was recharged with O_2 to 50 bar as often as necessary, until the pressure stayed constant at about 50 bar. After 2 h, the reaction was complete. The reaction may also be stirred overnight. The resulting solution was concentrated under reduced pressure and was washed with water to remove Co-salts and NaBr traces. The product can be additionally precipitated in 20% aq. HCl (500 mL/20 g acid), followed by filtration and washing with water.

Yield: 18.5 g, 67 mmol, 97%. - **$^1\text{H NMR}$** ($\text{DMSO}-d_6$, 300 MHz): $\delta = 13.07$ (s, 2 H, COOH), 7.57 (s, 2 H, Ar- CH), 1.66 (s, 4 H, $-\text{CH}_2$), 1.25 (s, 12 H, $-\text{CH}_3$) ppm. - **MS** (ESI(+), MeOH): $m/z = 277.1436$, cal. for $\text{C}_{16}\text{H}_{20}\text{O}_4 + \text{H}_1$: 277.1434. - **Elemental analysis** ($\text{C}_{16}\text{H}_{20}\text{O}_4$, M = 276.33): fnd. (cal.): C 69.83% (69.55%), H 7.18% (7.30%), N 7.53% (0.00%).

Additional information: In some cases, the corresponding anhydride was detected by $^1\text{H NMR}$ spectroscopy. The amount of anhydride is dependent on the amount of glacial acid used in the reaction in ratio to the tetraline amount. No other by-products could be observed. A complete turnover was determined. The high nitrogen value in elemental analysis is caused by the measurement in closed setup in the glovebox, neither in the synthesis nor in precursor steps were nitrogen containing chemicals used.

1.1.3 Synthesis of the Anhydride of 1,1,4,4-Tetramethyltetralin-6,7-dicarboxylic acid



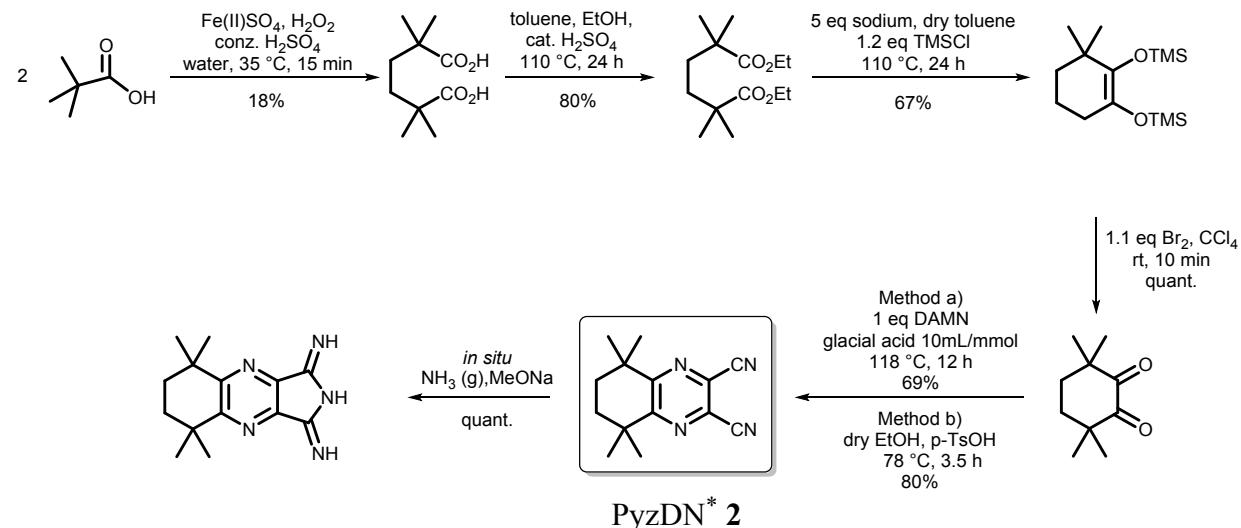
50 mg NaBr (486 μmol , ~3.5 mol%), 125 mg $[\text{Co}(\text{OAc})_2] \cdot 4\text{H}_2\text{O}$ (502 μmol , ~3.5 mol%), 3.00 g tetraline (13.9 mmol, 1 eq) were added into an autoclave and dissolved in 10 mL glacial acid. The autoclave was charged with 20 bar O_2 and heated to 180 °C for 90 min. The pressure increased to 30 bar, dropped down to 20 bar and was recharged to 30-40 bar two

more times. The autoclave was recharged as often as needed, until the pressure stayed constant. Carboxylic acid anhydride was precipitated out of the solution and washed with water. A second/third portion was yielded overnight, filtered, washed with water and finally dried in vacuum.

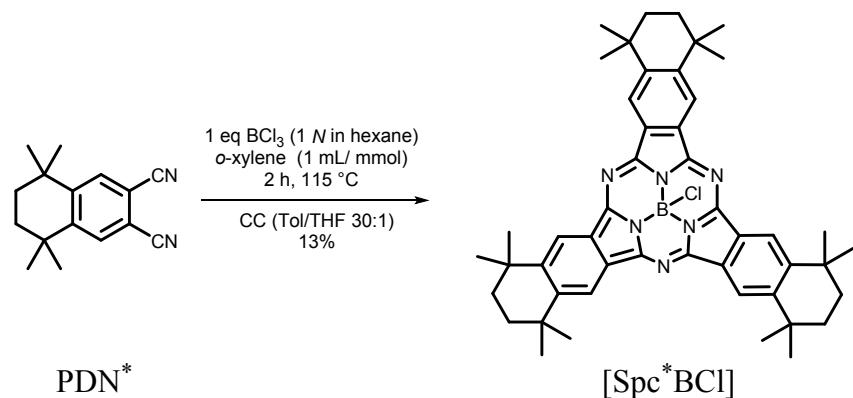
Yield: 3.43 g, 13.3 mmol, 96%. - **¹H NMR** (CDCl_3 , 300 MHz): $\delta = 7.95$ (s, 2 H, Ar-CH), 1.76 (s, 4 H, - CH_2), 1.36 (s, 12 H, - CH_3) ppm.

1.1.4 Synthesis of PyzDN^{*}

The synthesis of 3,3,6,6-tetramethylcyclohexane-1,2-dione (**XIV**) was carried out following the procedure of JONES.^[2] The synthesis of PyzDN^{*} **2** was carried out according to the procedure of SEIKEL.^[3]



1.1.5 Synthesis of [Spc^{*}BCl] ^[4,5]



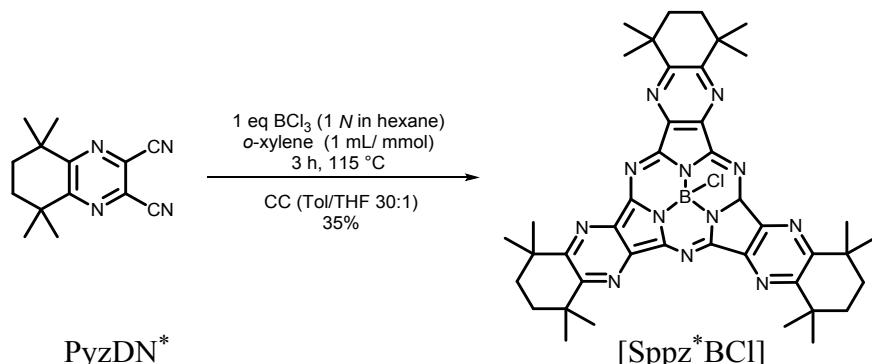
1 eq PDN was dissolved in 2 mL toluene /5 mmol PDN. Fresh BCl_3 solution (1 M in heptane or in *o*-xylene, 3 eq BCl_3 per intended $[\text{SpcBCl}]$) was added at once and then heated to 150 °C. The yellowish solution turned deep pink. After heating for about 5 h, the solution was loaded onto a short silica pluck, purified with DCM/EE to remove the boron compounds and then purified by CC (PE/EA 10:1).

Yield: ~5%. - R_f (PE/EE 10:1) = 0.4. - **$^1\text{H NMR}$** (C_6D_6 , 300 MHz): δ = 8.91 (s, 6 H, Ar-CH), 1.65-1.52 (m, 12 H, - CH_2), 1.42 (s, 18 H, - CH_3), 1.10 (s, 18 H, - CH_3) ppm. - **$^1\text{H NMR}$** (CD_2Cl_2 , 300 MHz): δ = 8.84 (s, 6 H, Ar-CH), 1.94-1.80 (m, 12 H, - CH_2), 1.66 (s, 18 H, - CH_3), 1.44 (s, 18 H, - CH_3) ppm. - **$^{13}\text{C NMR}$** (CD_2Cl_2 , 75 MHz): δ = 149.6, 149.2, 129.4, 120.4, 36.0, 35.3, 32.9, 32.3 ppm. - **UV-Vis** (DCM): λ = 582 (s), 528 (sh), 318 (s), 267 (s) nm. - **Fluorescence** (DCM, λ_{ex} = 350 nm): λ = 592 nm. - Φ_F (λ_{ex} = 490 nm) = 0.18. - Φ_Δ = 0.64. - **MS** (APCI-HRMS(+)): m/z = 761.4247 [$\text{M}+\text{H}]^+$, cal. for $\text{C}_{48}\text{H}_{54}\text{B}_1\text{Cl}_1\text{N}_6\text{H}_1$: 761.4272.

X-ray: Crystals could be obtained out of CD_2Cl_2 in the NMR tube at rt.

Additional information: In some cases, degradation of $[\text{Spc}^*\text{BCl}]$ in the form of a fast decolouration was observed in acidic solvents.

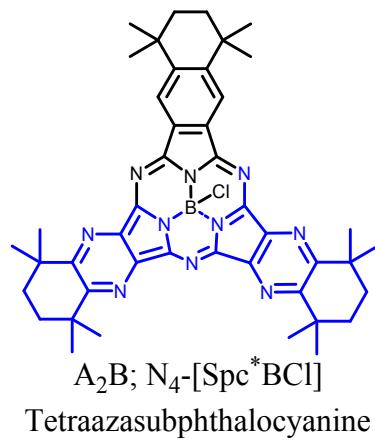
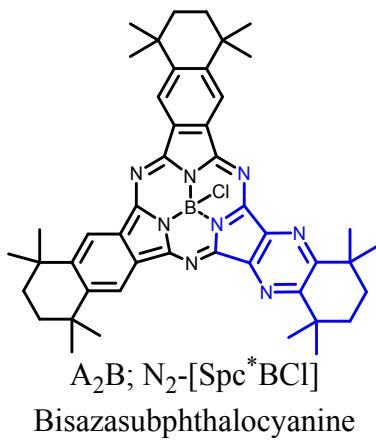
1.1.6 Synthesis of $[\text{Sppz}^*\text{BCl}]$ [4,5]



According to procedure described in 1.1.5. The pink solution was purified by CC (Tol/THF 30:1). The pink solid was dried in vacuum.

Yield: 35%. - R_f (Tol/THF 30:1) = 0.35. - R_f (DCM) = 0.6. - **$^1\text{H NMR}$** (CDCl_3 , 300 MHz): δ = 1.94–2.07 (m, 12 H, - CH_2), 1.75 (s, 18 H, - CH_3), 1.52 (s, 18 H, - CH_3) ppm. - **$^1\text{H NMR}$** (CD_2Cl_2 , 300 MHz): δ = 1.53 (s, 18 H, - CH_3), 1.78 (s, 18 H, - CH_3), 1.96–2.16 (m, 12 H, - CH_2) ppm. - **UV-Vis** (DCM): λ = 534 (s), 489 (sh), 334 (m), 303 (s) nm. - **Fluorescence** (DCM, λ_{ex} = 350 nm): λ = 546 nm. - Φ_F (λ_{ex} = 490 nm) = 0.19. - Φ_Δ = 0.88. - **MS** (APCI-HRMS(+)): m/z = 767.3964 [$\text{M}+\text{H}]^+$, cal. for $\text{C}_{42}\text{H}_{48}\text{B}_1\text{Cl}_1\text{N}_{12}\text{H}_1$: 767.3980.

1.1.7 Synthesis of Azasubphthalocyanines N_x-[Spc^{*}BCl]



According to procedure described in 1.1.5 using PDN^{*}/PyzDN^{*} in a 1:1 ratio. After evaporation of the solvent, the solid residue was loaded onto silica and the crude product was purified via gradient CC (Tol → Tol/THF).

[Spc^{*}BCl]: Yield: <1%. - The product was identified by using UV-Vis and MS. The analysis is in accordance to the one described in section 1.1.5.

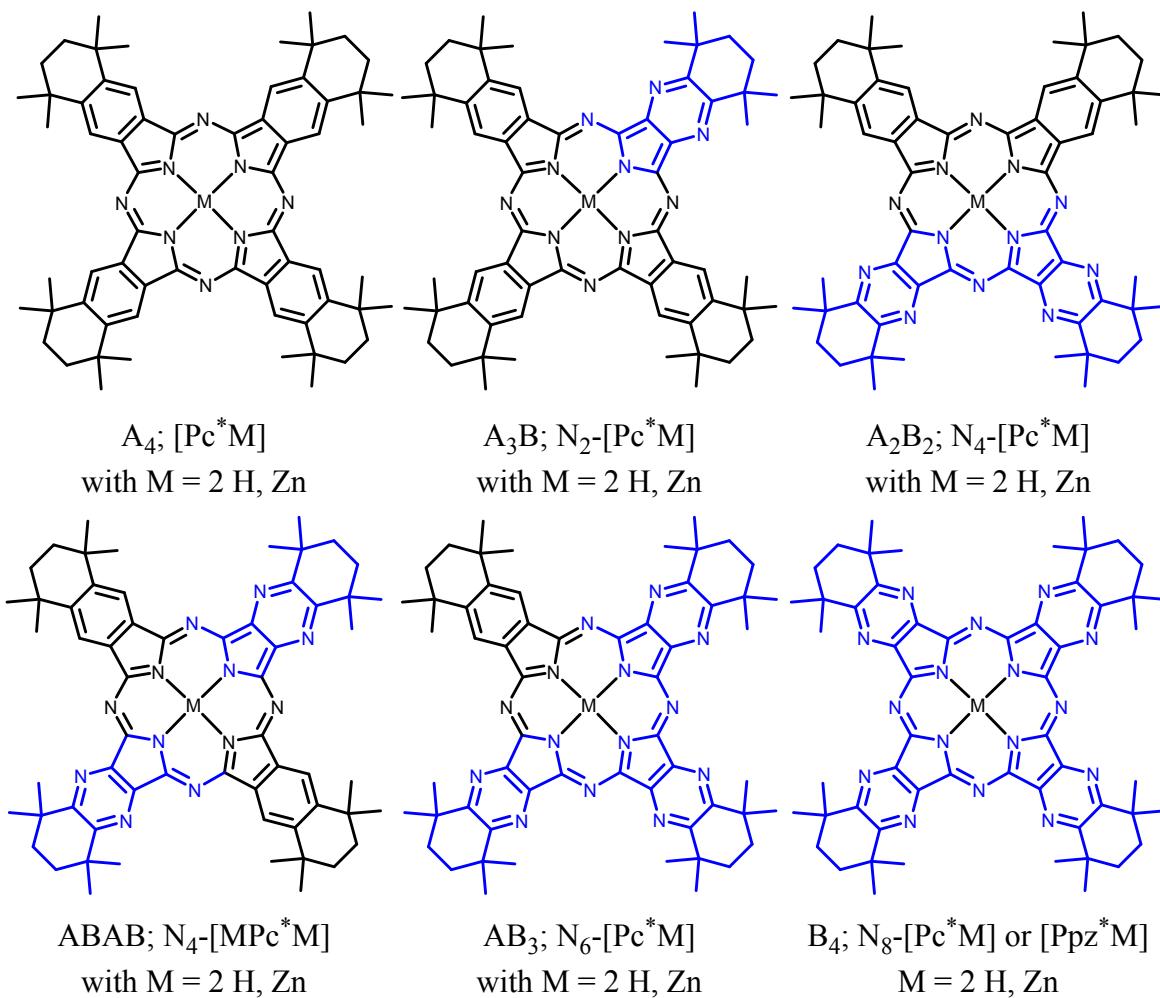
N₂-[Spc^{*}BCl]: Yield: 8 mg, 10.5 μmol, 1%. - **¹H NMR** (CDCl₃, 300 MHz): δ = 8.87 (s, 2 H, Ar-CH), 8.78 (s, 2 H, Ar-CH), 1.77-1.55 (m, 12 H, -CH₂), 1.38 (s, 9 H, -CH₃), 1.35 (s, 9 H, -CH₃), 1.05 (s, 9 H, -CH₃), 1.03 (s, 9 H, -CH₃) ppm. - **UV-Vis** (DCM): λ = 572 (s), 519 (sh), 327 (m), 303 (s) nm. - **Fluorescence** (DCM, λ_{ex} = 350 nm): λ = 577 nm. - Φ_F (λ_{ex} = 490 nm) = 0.24. - Φ_Δ = 0.67. - **MS** (APCI-HRMS(+)): m/z = 763.4162 [M+H]⁺, cal. for C₄₆H₅₃B₁Cl₁N₈+H₁: 763.4177.

N₄-[Spc^{*}BCl]: Yield: 12 mg, 15.7 μmol, 2%. - **¹H NMR** (CDCl₃, 300 MHz): δ = 8.78 (s, 2 H, Ar-CH), 1.66-1.51 (m, 12 H, -CH₂), 1.38 (s, 9 H, -CH₃), 1.38 (s, 9 H, -CH₃), 1.35 (s, 9 H, -CH₃), 1.35 (s, 9 H, -CH₃) ppm. - **UV-Vis** (DCM): λ = 557 (s), 506 (sh), 330 (s), 296 (s) nm. - **Fluorescence** (DCM, λ_{ex} = 350 nm): λ = 567 nm. - Φ_F (λ_{ex} = 490 nm) = 0.22. - Φ_Δ = 0.65. - **MS** (APCI-HRMS(+)): m/z = 765.4069 [M+H]⁺, cal. for C₄₄H₅₁B₁Cl₁N₁₀+H₁: 765.4082.

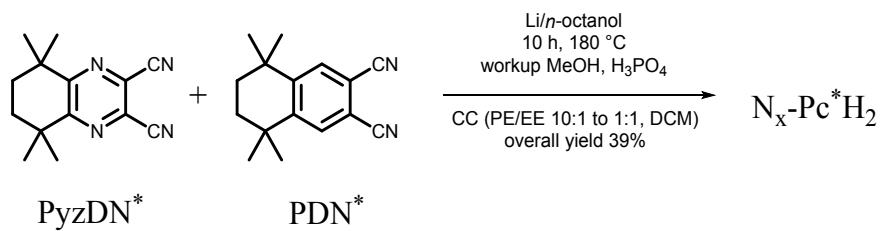
[Sppz^{*}BCl]: Yield: <1%. - The analysis is in accordance to the one described in section 1.1.6.

Additional information: For the synthesis of N_x-[Spc^{*}BCl], depending on the desired product, the corresponding ratios of dinitriles was varied. In the ¹H NMR spectra, dinitriles traces were observed.

1.1.8 Synthesis of Asymmetrical Phthalocyanines: $N_x\text{-Pc}^*\text{H}_2$



1.1.9 Synthesis of $N_x\text{-Pc}^*\text{H}_2$ [4]



According to the procedure described in the main paper, using PyzDN^{*} **2**/PDN^{*} **1** in a 1:1 ratio. The solution turned dark green. The isomers were separated by CC (PE(40/60)/EE gradient). At first, Pc^*H_2 and $N_2\text{-Pc}^*\text{H}_2$ were eluted, then ABAB and the symmetrical A₂B₂ $N_4\text{-Pc}^*\text{H}_2$ were eluted. After changing the solvent to PE/EE 1:1 a mixture of A₂B₂ $N_4\text{-Pc}^*\text{H}_2$ and AB₃ $N_6\text{-Pc}^*\text{H}_2$ was eluted. Finally, Ppz^{*}H₂ was eluted with pure DCM. In a second CC (DCM), A₂B₂ $N_4\text{-Pc}^*\text{H}_2$ and AB₃ $N_6\text{-Pc}^*\text{H}_2$ were separated.

Pc^{*}H₂: Yield: 2%. - The analysis is in accordance with values in the main paper. - Φ_F ($\lambda_{ex} = 350 \text{ nm}$) = 0.44; - ($\lambda_{ex} = 598 \text{ nm}$) = 0.45. - Φ_Δ = 0.16.

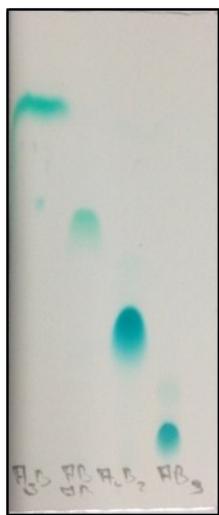
N₂-Pc^{*}H₂: **Yield:** 4%. - **¹H NMR** (CDCl₃, 300 MHz): δ = 9.69 (s, 2 H, Ar-CH), 9.65 (s, 2 H, Ar-CH), 9.42 (s, 2 H, Ar-CH), 2.19 (s, 4 H, -CH₂), 2.07 (s, 8 H, -CH₂), 2.04 (s, 4 H, -CH₂), 1.90 (s, 12 H, -CH₃), 1.83 (s, 12 H, -CH₃), 1.81 (s, 12 H, -CH₃), 1.81 (s, 12 H, -CH₃), -0.07 (s, 2 H, -NH) ppm. - **¹³C NMR** (C₆D₆, 75 MHz): δ = 160.6, 149.8, 149.2, 148.6, 137.0, 133.0, 132.2, 122.2, 121.6, 39.1, 36.1, 36.1, 35.9, 35.4, 35.3, 34.7, 32.9, 32.8, 32.7, 30.9 ppm. - not all quartary atoms could be detected. - **IR** (ATR, 400-4000 cm⁻¹): $\tilde{\nu}$ = 3475 (vw), 3297 (vw), 2956 (m), 2921 (m), 2855 (m), 2555 (vw), 2348 (vw), 1711 (m), 1688 (m), 1621 (m), 1539 (m), 1498 (s), 1460 (m), 1382 (m), 1358 (m), 1328 (m), 1301 (s), 1258 (m), 1242 (m), 1188 (m), 1159 (m), 1140 (m), 1119 (m), 1089 (m), 1071 (m), 1022 (s), 999 (m), 980 (m), 891 (m), 848 (m), 804 (m), 755 (s), 722 (m), 681 (s), 622 (m), 541 (m) cm⁻¹. - **CV** (DCM, [TBA]PF₆, Fc): E_{ox1} = 0.43, E_{red1} = -1.23, E_{red2} = -1.65 V. - **UV-Vis** (DCM): λ = 687 (s), 619 (sh), 340 (s), 306 (s), 232 (s) nm. - Φ_F (λ_{ex} = 350 nm) = 0.31; - (λ_{ex} = 598 nm) = 0.33. - Φ_Δ = 0.14. - **MS** (APCI-HRMS(+)): m/z = 957.5992 [M+H]⁺, cal. for C₆₂H₇₂N₁₀+H₁: 957.6014. - **Elemental analysis** (C₆₂H₇₂N₁₀, M = 957.30 g/mol): fnd. (cal.): C: 75.29% (77.79%), H: 8.92% (7.58%), N: 10.14% (14.63%).

A₂B₂ N₄-Pc^{*}H₂: **Yield:** 17%. - **¹H NMR** (CDCl₃, 300 MHz): δ = 9.56 (s, 2 H, Ar-CH), 9.50 (s, 2 H, Ar-CH), 2.20 (s, 8 H, -CH₂), 2.05 (s, 8 H, -CH₂), 1.91 (s, 24 H, -CH₃), 1.81 (s, 12 H, -CH₃), 1.80 (s, 12 H, -CH₃), -0.04 (s, 2 H, -NH) ppm. - **¹H NMR** (Pyridine-d₅, 300 MHz): δ = 9.87 (s, 2 H, Ar-CH), 9.68 (s, 2 H, Ar-CH), 1.94 (s, 8 H, -CH₂), 1.76 (s, 8 H, -CH₂), 1.72 (s, 12 H, -CH₃), 1.67 (s, 12 H, -CH₃), 1.46 (s, 12 H, -CH₃), 1.37 (s, 12 H, -CH₃), -0.26 (s, 2 H, -NH) ppm. - **¹H NMR** (C₆D₆, 300 MHz): δ = 9.83 (s, 2 H, Ar-CH), 9.82 (s, 2 H, Ar-CH), 1.86 (s, 8 H, -CH₂), 1.79 (s, 8 H, -CH₂), 1.75 (s, 24 H, -CH₃), 1.48 (s, 12 H, -CH₃), 1.38 (s, 12 H, -CH₃), -0.31 (s, 2 H, -NH) ppm. - **¹³C NMR** (C₆D₆, 75 MHz): δ = 162.4, 150.0, 149.6, 145.1, 144.4, 135.2, 134.8, 122.7, 122.0, 39.4, 39.3, 36.1, 36.0, 35.2, 34.6, 34.5, 32.7, 32.7, 21.1, 30.8 ppm. - **IR** (ATR, 400-4000 cm⁻¹): $\tilde{\nu}$ = 3526 (w), 3284 (w), 2915 (s), 2857 (s), 1736 (w), 1638 (w), 1553 (w), 1496 (w), 1455 (s), 1381 (w), 1358 (m), 1328 (m), 1300 (s), 1254 (m), 1242 (m), 1190 (m), 1162 (m), 1148 (m), 1128 (m), 1083 (w), 1048 (w), 1019 (s), 1001 (s), 983 (s), 949 (w), 936 (w), 893 (m), 851 (m), 828 (m), 761 (s), 752 (s), 717 (m), 700 (m), 679 (m), 623 (w), 560 (w), 543 (m), 507 (m), 443 (w), 429 (w) cm⁻¹. - **CV** (DCM, [TBA]PF₆, Fc): E_{ox1} = 0.28, E_{red1} = -1.47, E_{red2} = -1.79 V. - **UV-Vis** (DCM): λ = 680 (s), 656 (s), 625 (sh), 348 (s), 232 (s) nm. - Φ_F (λ_{ex} = 350 nm) = 0.13; - (λ_{ex} = 598 nm) = 0.14. - Φ_Δ = 0.07. - **MS** (APCI-HRMS(+)): m/z = 959.5921 [M+H]⁺, cal. for C₆₀H₇₀N₁₂+H₁: 959.5919. - **Elemental**

analysis ($C_{60}H_{70}N_{12}$, M = 959.28 g/mol): fnd. (cal.): C: 72.03% (75.12%), H: 7.36% (7.36%), N: 16.36% (17.52%).

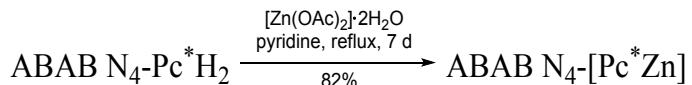
N₆-Pc*H₂: **Yield:** 9%. - **¹H NMR** (CDCl₃, 300 MHz): δ = 9.75 (s, 2 H, Ar-CH), 2.22 (s, 4 H, -CH₂), 2.19 (s, 12 H, -CH₂), 1.93 (s, 12 H, -CH₃), 1.91 (s, 12 H, -CH₃), 1.90 (s, 12 H, -CH₃), 1.83 (s, 12 H, -CH₃), -0.53 (s, 2 H, -NH) ppm. - **¹³C NMR** (C₆D₆, 75 MHz): δ = 162.0, 139.1, 131.3, 134.1 (w), 132.9 (w), 132.9 (w), 131.3 (w), 127.1 (w), 123.1, 46.0, 41.3 (w), 39.7, 39.3, 36.3, 34.5, 32.8, 31.8, 31.7, 31.1, 30.9, 30.8, 29.8, 29.5, 29.3, 29.2, 28.6, 28.1, 27.5, 27.5 ppm. - 6 weak signal, marked with (w); not all quartary atoms could be detected. - **IR** (ATR, 400-4000 cm⁻¹): $\tilde{\nu}$ = 3291 (w), 2957 (s), 2924 (s), 2856 (s), 1727 (s), 1693 (s), 1637 (m), 1540 (m), 1503 (m), 1457 (s), 1411 (m), 1381 (m), 1360 (m), 1330 (m), 1305 (m), 1281 (m), 1257 (s), 1192 (s), 1157 (w), 1139 (m), 1127 (s), 1086 (m), 1072 (m), 1021 (s), 995 (s), 952 (m), 935 (w), 893 (w), 852 (m), 828 (m), 804 (m), 756 (m), 744 (s), 719 (s), 700 (w), 677 (s), 631 (m), 542 (w), 506 (w), 466 (w), 429 (w) cm⁻¹. - **CV** (DCM, [TBA]PF₆, Fc): E_{ox1} = 0.35, E_{red1} = -1.47, E_{red2} = -1.82 V. - **UV-Vis** (DCM): λ = 660 (s), 597 (sh), 343 (s), 234 (s) nm. - **Φ_F** (λ_{ex} = 350 nm) = 0.05; - (λ_{ex} = 598 nm) = 0.03. - **Φ_Δ** = 0.04. - **MS** (APCI-HRMS(+)): *m/z* = 961.5825 [M+H]⁺, cal. for C₅₈H₆₈N₁₄+H₁: 961.5824. - **Elemental analysis** (C₅₈H₆₈N₁₄, M = 961.25 g/mol): fnd. (cal.): C: 71.99% (72.47%), H: 9.49% (7.13%), N: 12.38% (20.40%).

Ppz*H₂: **Yield:** 9%. - The analysis is in accordance with values in the main paper. - **Φ_F** (λ_{ex} = 350 nm) = 0.03; - (λ_{ex} = 598 nm) = 0.03. - **Φ_Δ** = 0.05.



N_x-Pc*H₂ in DCM

1.1.10 Synthesis of N₄-[Pc^{*}Zn] using [Zn(OAc)₂]



General Procedure for the Synthesis of N_x-[Pc^{*}Zn] using [Zn(OAc)₂]:

1 eq ABAB or A₂B₂ N₄-Pc^{*}H₂ was dissolved in pyridine and 1.2 eq [Zn(OAc)₂][·]2H₂O were added. After refluxing for 1 h, the reaction was monitored by TLC (DCM), and was stirred for another 1 h. The resulting blue solutions were concentrated in vacuum, filtered, and washed with an excess of water. After washing the product off the filter paper using CHCl₃/MeOH 3:1, the blue product was dried in vacuum and purified by preparative TLC or CC (Tol/THF 20:1).

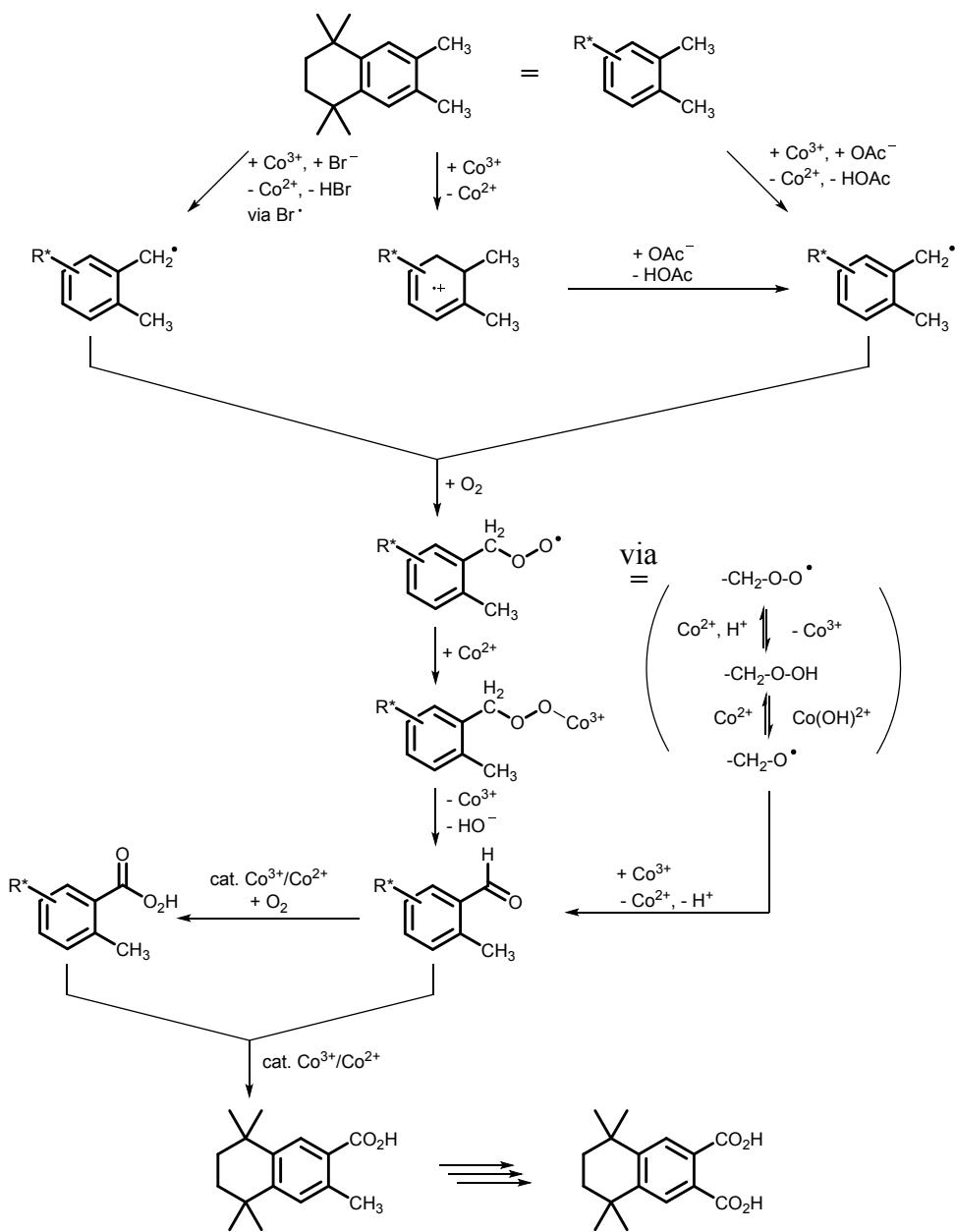
ABAB N₄-[Pc^{*}Zn]: Yield: 82%. - **R_f** (Tol/THF 20:1) = 0.63. - **¹H NMR** (C₆D₆, 300 MHz): δ = 9.96 (s, 4 H, Ar-CH), 1.91 (s, 8 H, -CH₂), 1.85 (s, 24 H, -CH₃), 1.78 (s, 8 H, -CH₂), 1.45 (s, 24 H, -CH₃) ppm. - **UV-Vis** (DCM): λ = 687 (s), 657 (s), 631 (sh), 596 (sh), 357 (s) nm. - **Φ_F** (λ_{ex} = 350 nm) = 0.16; - (λ_{ex} = 598 nm) = 0.18. - **Φ_Δ** = 0.73. - **MS** (APCI-HRMS(+)): *m/z* = 1021.5073 [M+H]⁺, cal. for C₆₀H₆₈N₁₂Zn₁+H₁: 1021.5054.

A₂B₂ N₄-[Pc^{*}Zn]: Yield: 76%. - The analysis is in accordance with the data described above.

All other N_x-[Pc^{*}Zn] are described in the Paper.

1.2 Suggested Radical Autoxidation Mechanism of Tetraline

In literature, such Co(III) or Mn(III) assisted aerobic oxidations are discussed as radical chain reactions. Rate determining step is the selective benzylic C-H activation, H abstraction by Co(III)-coordinated acetate (quasi acetyl radical ligands) or by bromine radicals generated in the presence of bromide promotor. Instead of H atom abstraction, a sequence of electron transfer followed by deprotonation of the formed radical cation is discussed. The benzylic radicals $\text{Ar}-\text{CH}_2\cdot$ are trapped by oxygen, the peroxy radical is reduced by Co(II). Hock cleavage leads to a carbaldehyde, which undergoes further metal catalysed autoxidation steps to the carboxylic acid:



1.3 Absorbance and Emission Spectra

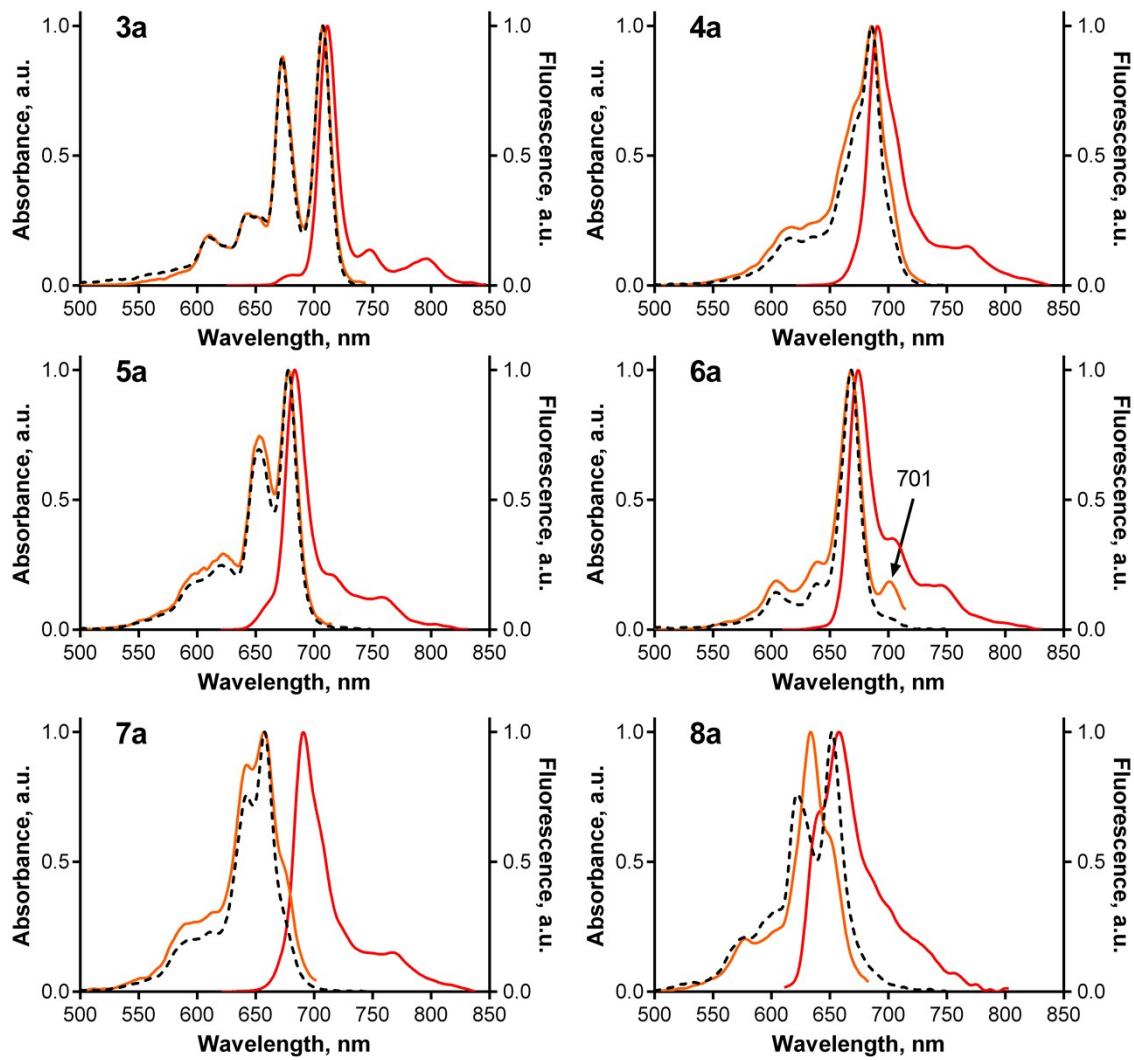


Figure SI-1. Normalized absorption (black, dashed), emission (red) and excitation (orange) spectra of $N_x\text{-Pc}^*\text{H}_2$ in THF.

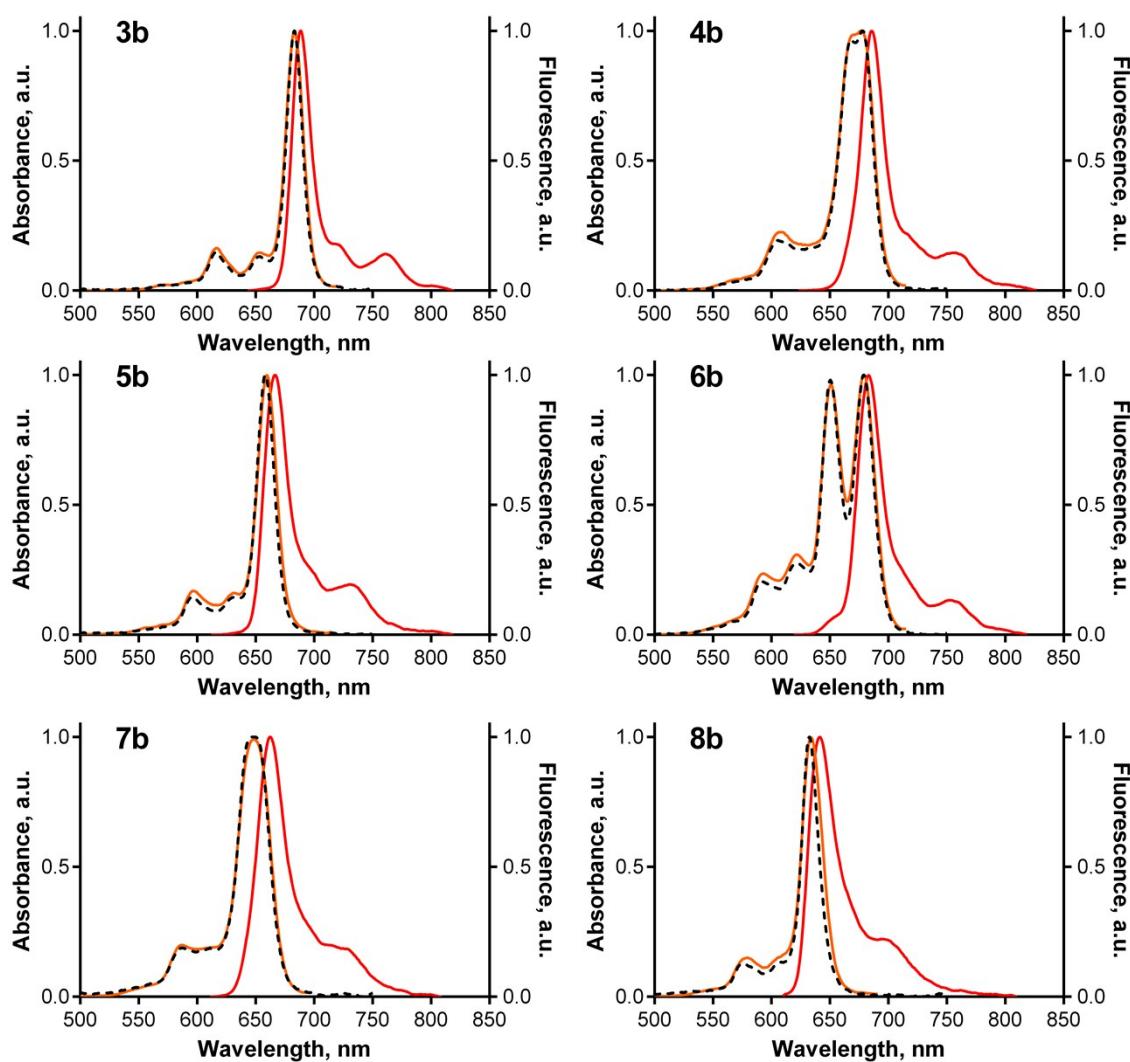


Figure SI-2. Normalized absorption (black, dashed), emission (red) and excitation (orange) spectra of $N_x\text{-Pc}^*\text{Zn}$ in THF.

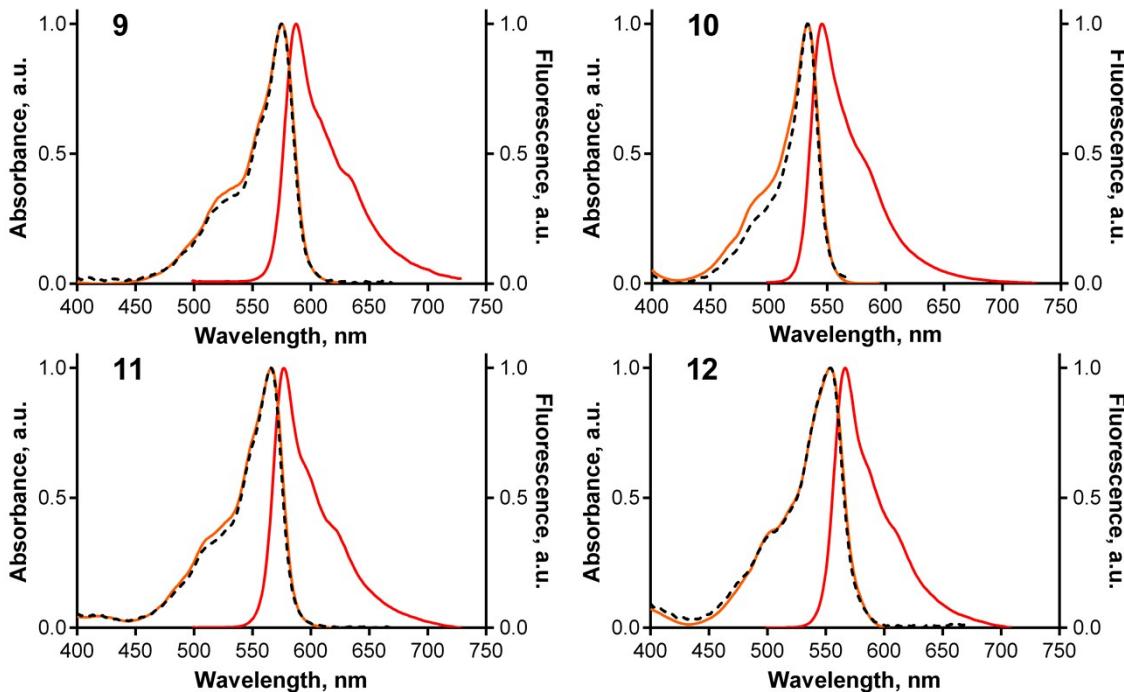
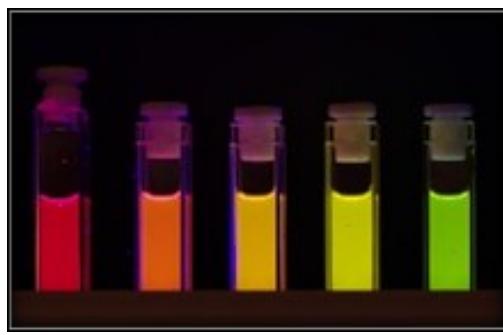


Figure SI-3. Bottom: Normalized absorption (black, dashed), emission (red) and excitation (orange) spectra of $N_x\text{-Spc}^*\text{BCl}$ in MeOH. Top: Cuvettes of a [PpzZn] (left) in THF and $N_x\text{-[Spc}^*\text{BCl]}$ in MeOH, with increasing number of [-N=] in $N_x\text{-[Spc}^*\text{BCl}]$ from $x = 0, 2, 4$, to 6 (left to right). Samples are illuminated with light at $\lambda = 365$ nm.

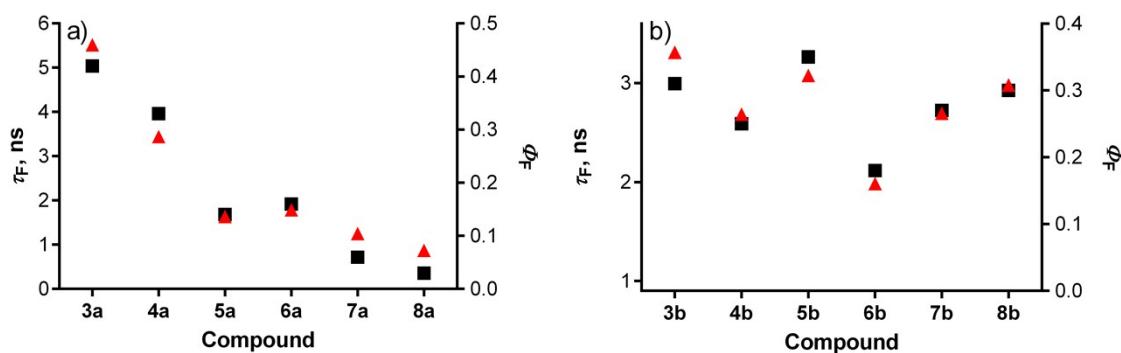


Figure SI-4. Correlation between Φ_F (black squares) and τ_F (red triangles) values for $N_x\text{-Pc}^*\text{H}_2$ in THF (a) and $N_x\text{-Pc}^*\text{Zn}$ in THF (b). For purposes of this presentation, the relative contribution of each process to the average lifetime for compounds (7a and 8a) was calculated.

1.4 Computational Studies

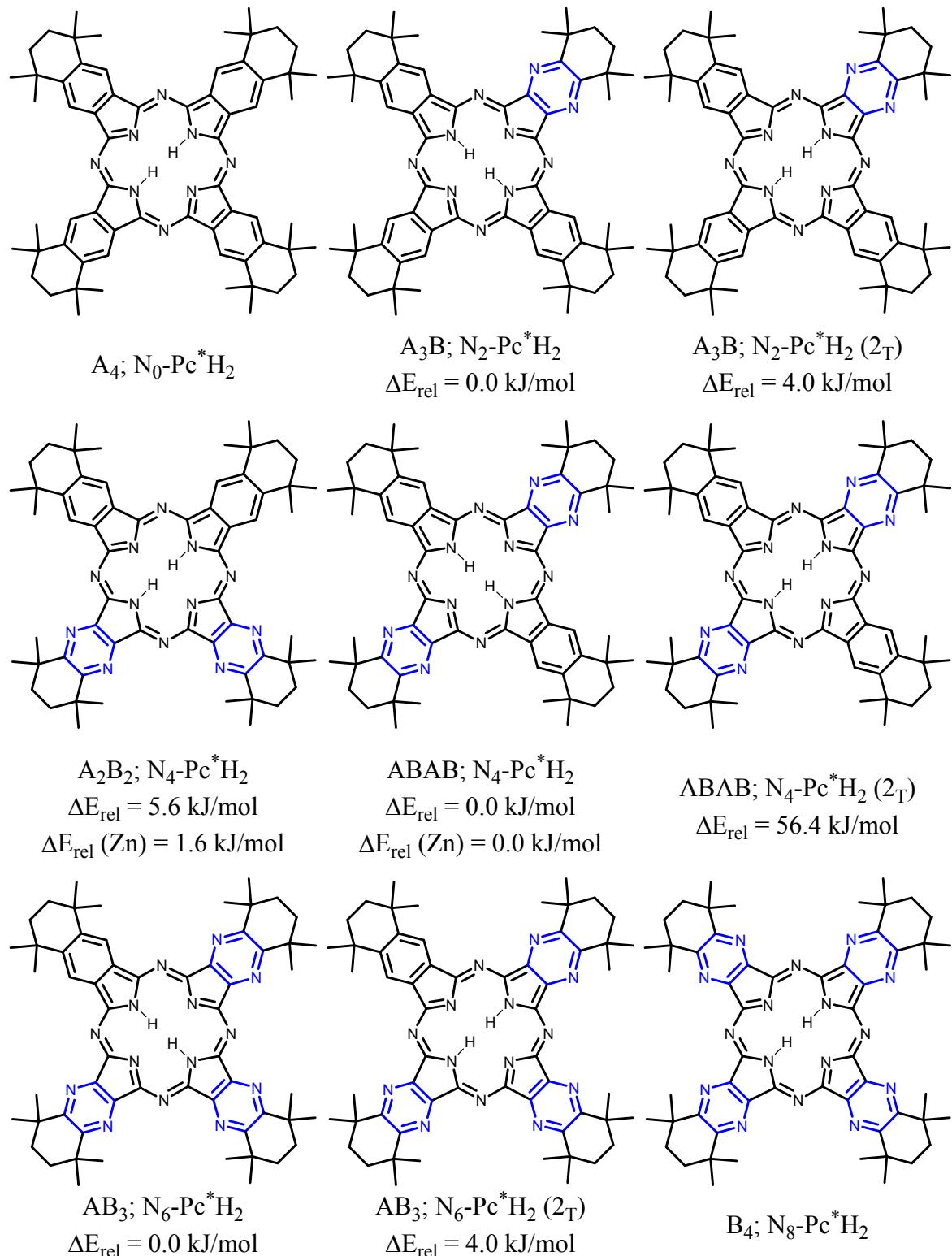


Figure SI-5. Overview of the phthalocyanines $N_x\text{-Pc}^*\text{H}_2$ computationally investigated. Regioisomers are numbered and tautomers marked (subscript T). Energies are given relative to the most stable isomer (ΔE_{rel}). Zn-substituted phthalocyanines [Pc^*Zn] are derived by substituting the two central H atoms by Zn. Relative isomer energies are given as $\Delta E_{\text{rel}}(\text{Zn})$, no tautomers are found for [Pc^*Zn].

Table SI-1. Computed excitation energies (Δλ in nm), oscillator strengths (f), frontier orbital gaps ($\Delta_{H-L/L+1}$ in nm) and molecular orbital character of the transitions for the Q-bands. Dipole moment (μ) and isotropic polarizability ($\Delta\alpha$) in a.u.^a

Molecule	λ / nm	$\Delta\lambda$ / nm	f	Character ^[b]	$\Delta_{H-L/L+1}$ ^[c]	μ ^[d]	$\Delta\alpha$ ^[e]
$N_8\text{-Pc}^*\text{H}_2$	541	0	0.42	$H \rightarrow L+1$	781	0.00	0.0
	536	0	0.53	$H \rightarrow L$	818		
$N_6\text{-Pc}^*\text{H}_2$	550	9	0.56	$H \rightarrow L$	861	0.87	23.6
	549	13	0.49	$H \rightarrow L+1$	797		
$N_6\text{-Pc}^*\text{H}_2 (2_T)$	559	18	0.47	$H \rightarrow L+1$	839	0.92	26.6
	552	16	0.60	$H \rightarrow L$	848		
$N_4\text{-Pc}^*\text{H}_2 (A_2B_2)$	563	22	0.55	$H \rightarrow L+1$ (39%)	845	1.25	28.3
				$H \rightarrow L$ (27%)			
				$H \rightarrow L$ (41%)			
	559	23	0.63	$H \rightarrow L+1$ (18%)			
				$H-2 \rightarrow L+1$ (15%)			
$N_4\text{-Pc}^*\text{H}_2 (\text{ABAB})$	561	20	0.62	$H \rightarrow L$	910	0.00	47.8
	560	24	0.55	$H \rightarrow L+1$	815		
$N_4\text{-Pc}^*\text{H}_2 (\text{ABAB}, 2_T)$	583	42	0.56	$H \rightarrow L+1$	892	0.06	61.8
	580	44	0.54	$H \rightarrow L$	911		
$N_2\text{-Pc}^*\text{H}_2$	569	28	0.62	$H \rightarrow L+1$	855	0.90	71.1
	564	28	0.72	$H \rightarrow L$	915		
$N_2\text{-Pc}^*\text{H}_2 (2_T)$	577	36	0.61	$H \rightarrow L+1$	897	0.87	73.3
	571	35	0.69	$H \rightarrow L$	900		
$N_0\text{-Pc}^*\text{H}_2$	580	39	0.69	$H \rightarrow L+1$	903	0.00	94.9
	568	32	0.81	$H \rightarrow L$	923		
$N_8\text{-[Pc}^*\text{Zn]}$	532	0	0.51	$H \rightarrow L$	788	0.00	0.0
	532	0	0.51	$H \rightarrow L+1$	788		
$N_6\text{-[Pc}^*\text{Zn]}$	549	17	0.55	$H \rightarrow L$	840	0.89	24.6
	544	12	0.59	$H \rightarrow L+1$	810		
$N_4\text{-[Pc}^*\text{Zn}] (A_2B_2)$	556	26	0.62	$H \rightarrow L$	852	1.25	47.6
	554	22	0.63	$H \rightarrow L+1$	848		
$N_4\text{-[Pc}^*\text{Zn}] (\text{ABAB})$	565	33	0.60	$H \rightarrow L$	898	0.00	49.5
	557	26	0.63	$H \rightarrow L+1$	831		
$N_2\text{-[Pc}^*\text{Zn}]$	566	34	0.69	$H \rightarrow L$	897	0.88	71.1
	564	32	0.69	$H \rightarrow L+1$	863		
$N_0\text{-[Pc}^*\text{Zn}]$	570	38	0.77	$H \rightarrow L$	902	0.00	93.7
	570	38	0.77	$H \rightarrow L+1$	901		

^[a] Shift relative to Q1/Q2 band of $N_8\text{-Pc}^*\text{H}_2$. ^[b] dominant MO transitions for the Q-bands (H: HOMO, H-2: HOMO-2, L: LUMO, L+1: LUMO+1). For $N_4\text{-Pc}^*\text{H}_2$, relative contributions are given in percent. ^[c] Frontier MO gap of the orbitals given in the “character” column HOMO/LUMO and HOMO/LUMO+1 in nm. ^[d] All compounds exhibit an in-plane dipole moment only pointing toward the nitrogen-rich parts of the molecule. ^[e] Polarizabilities are given relative to $B_4\text{H}_2$ ($\alpha = 1091.3$ a.u.) and $B_4\text{Zn}$ ($\alpha = 1097.1$ a.u.), respectively.

1.5 ^1H NMR Spectra

1.5.1 Subphthalocyanines

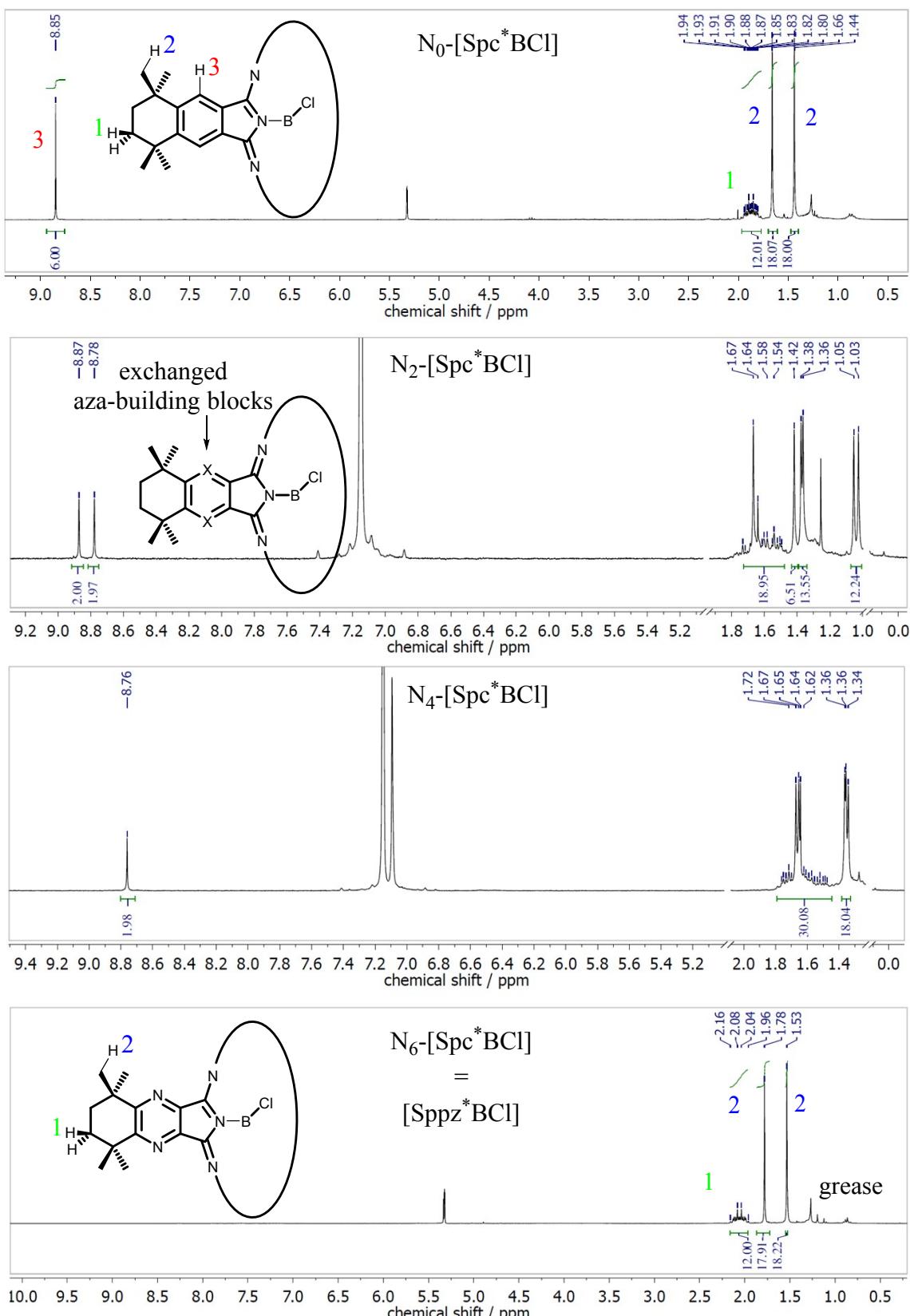
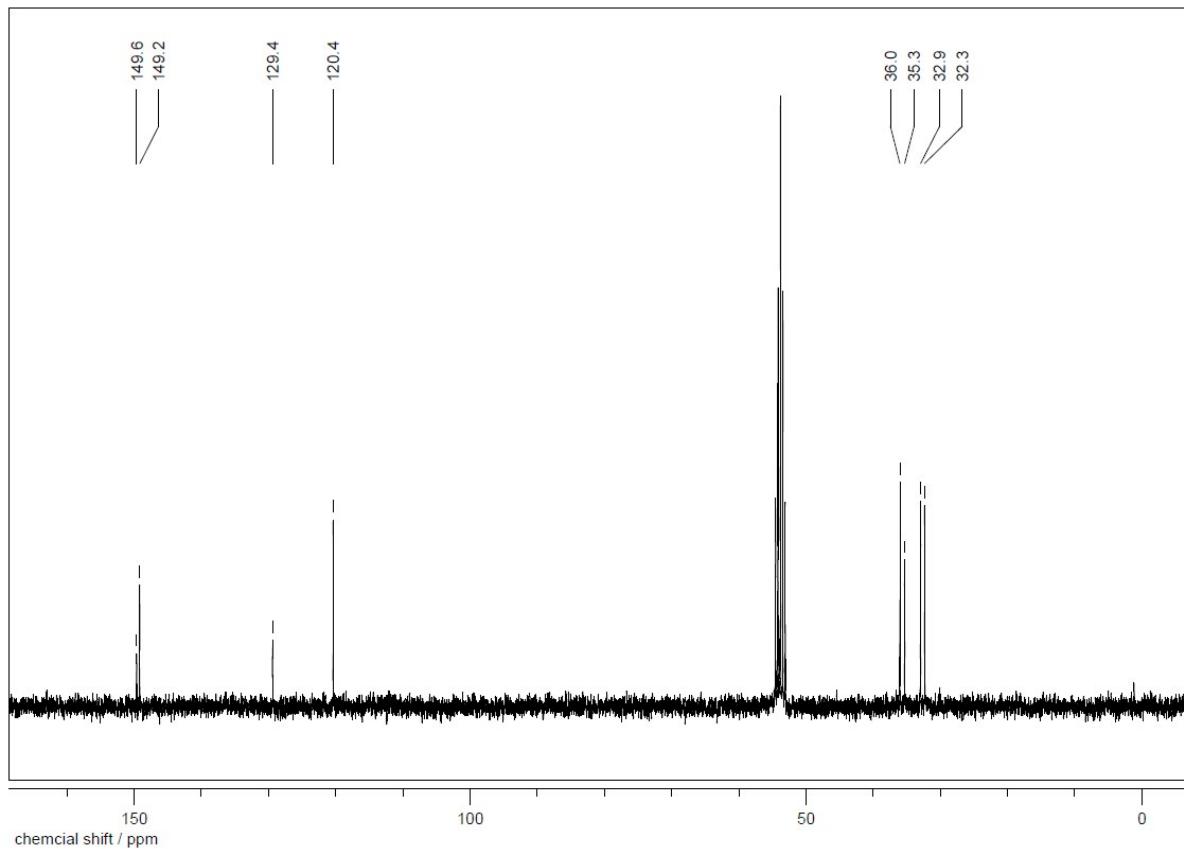


Figure SI-6. ^1H NMR spectrum of $\text{N}_x[\text{Spc}^*\text{BCl}]$ in DCM-d_2 , 300 MHz, or C_6D_6 , respectively, with increasing number of $x = \text{N}$.

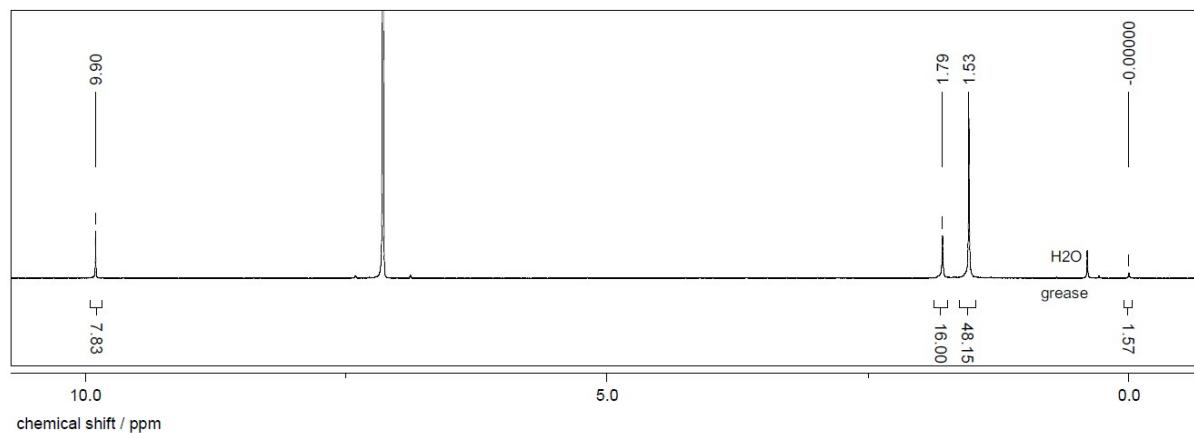
¹³C NMR N₀-[Spc^{*}BCl]



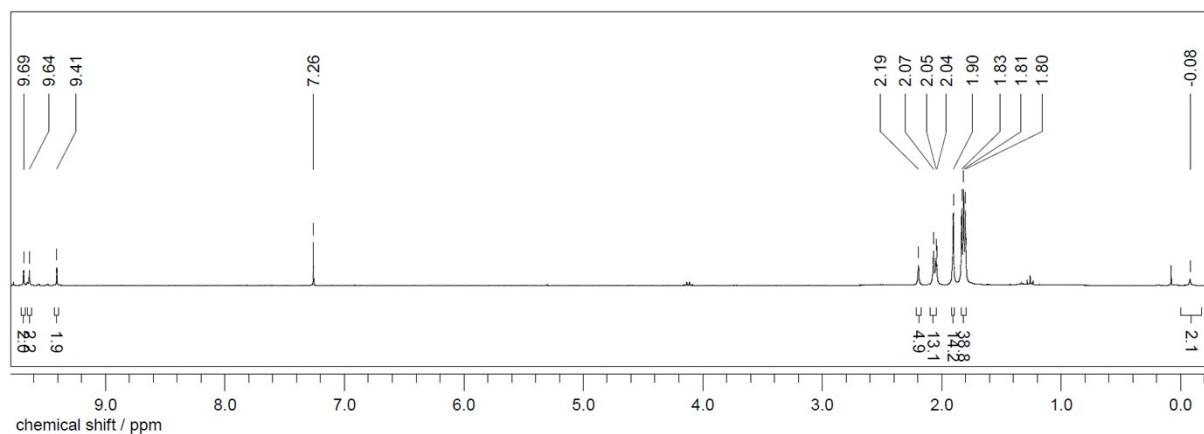
Because of the decreasing stability and solubility no ¹³C NMR spectra in sufficient purity could be obtained of the other N_x-[Spc^{*}BCl].

1.5.2 Azaphthalocyanines

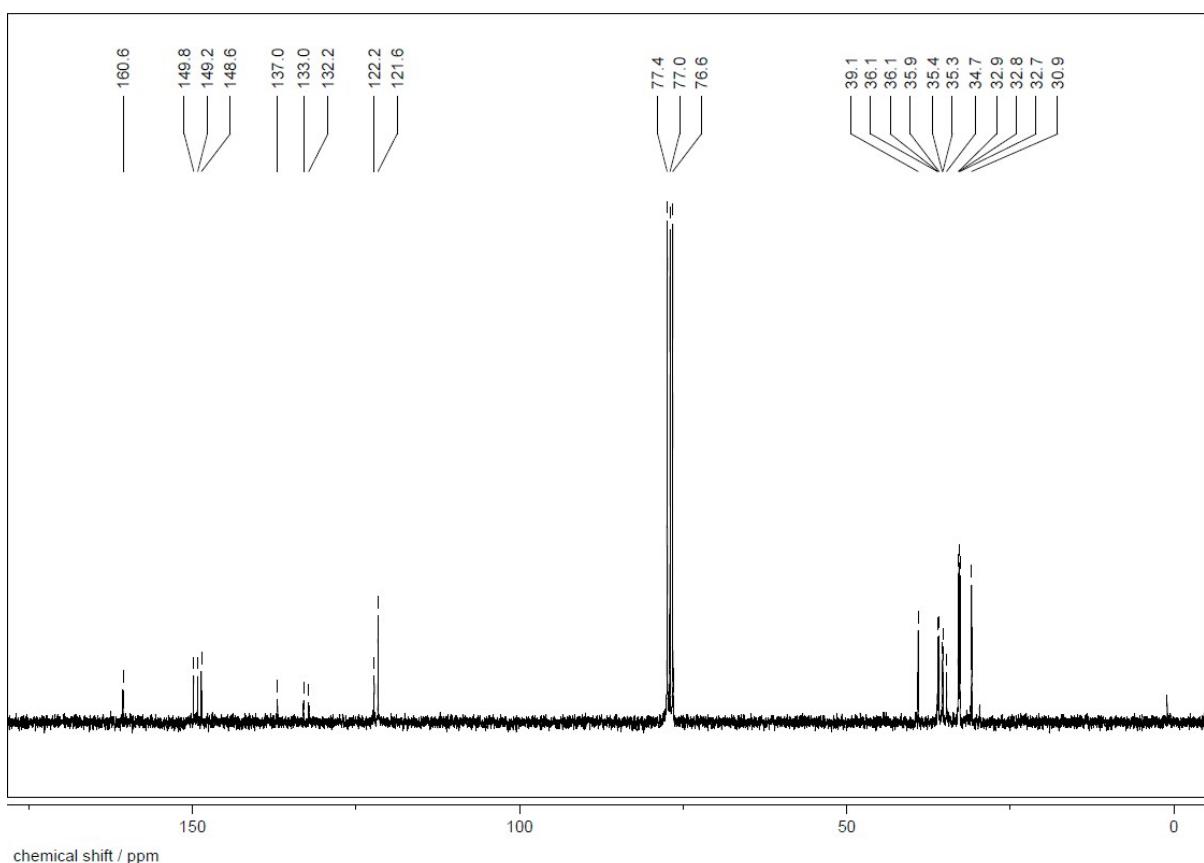
N₀-Pc*H₂ [3-5]



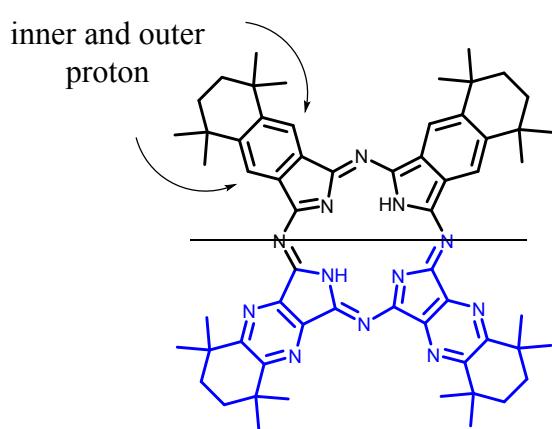
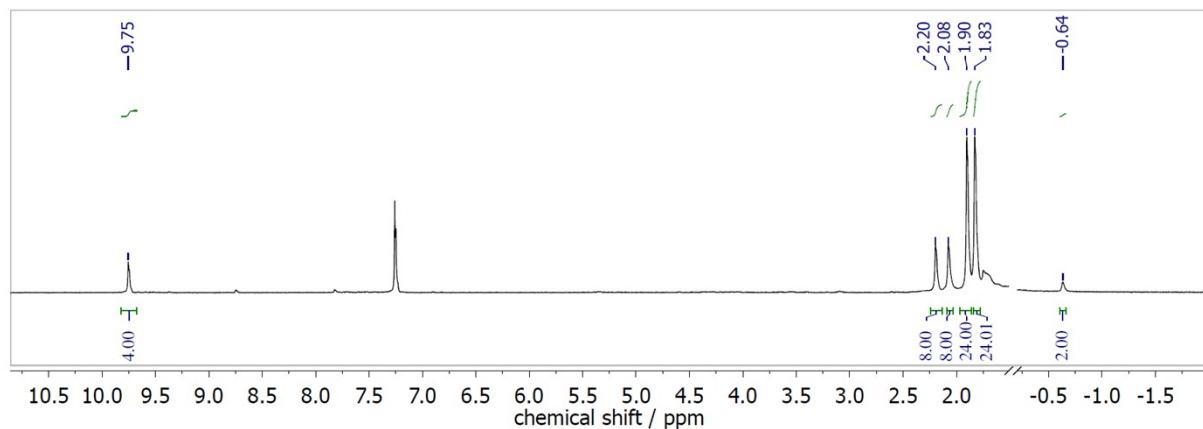
¹H NMR N₂-Pc*H₂



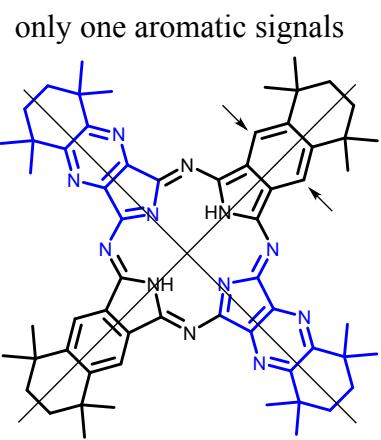
¹³C NMR N₂-Pc*H₂



N₄-Pc*H₂



A₂B₂
Tetraazaphthalocyanine
two aromatic signals



ABAB
Tetraazaphthalocyanine

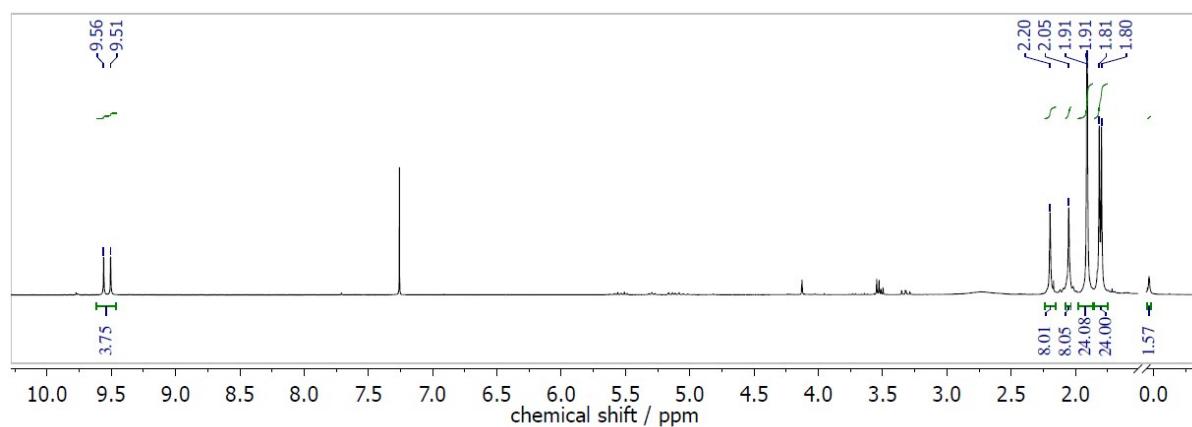
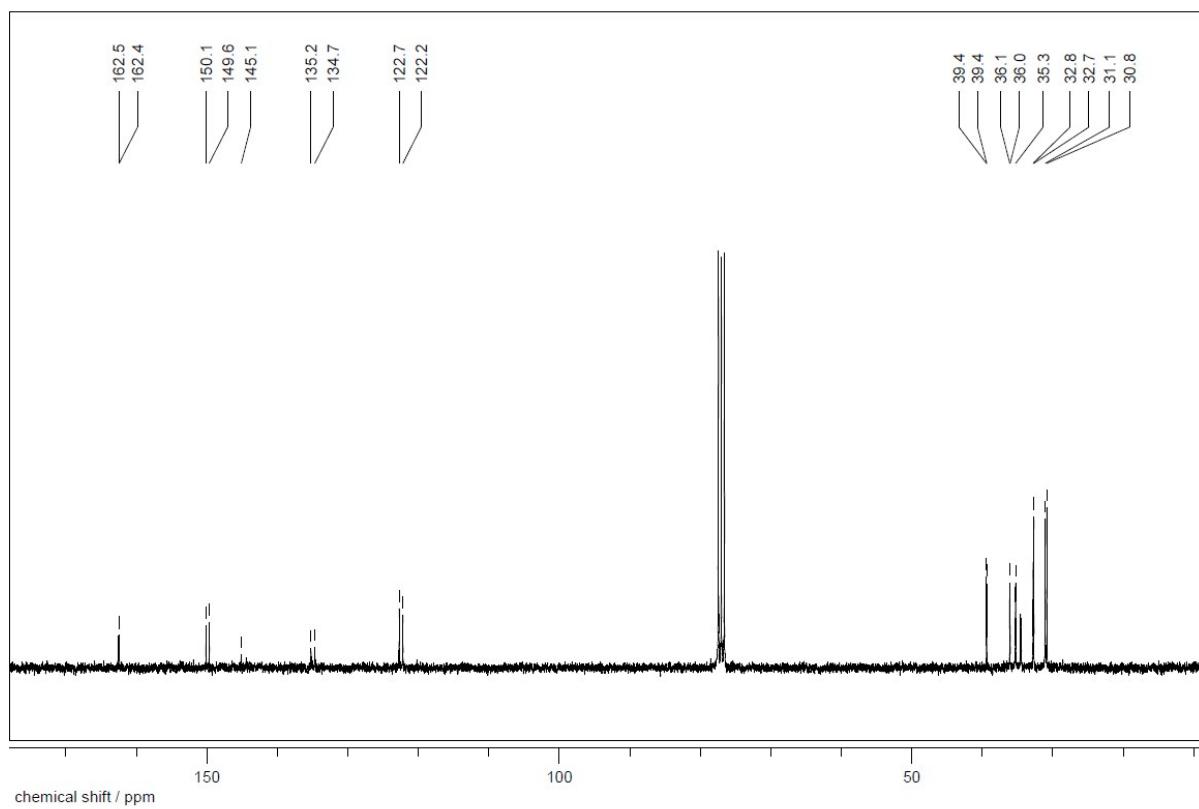


Figure SI-7b. ¹H NMR spectrum of ABAB (above) / A₂B₂ (below) N₄-Pc*H₂ in CDCl₃, 300 MHz.

^{13}C NMR $\text{A}_2\text{B}_2 \text{N}_4\text{-Pc}^*\text{H}_2$



^{13}C NMR ABAB $\text{N}_4\text{-Pc}^*\text{H}_2$

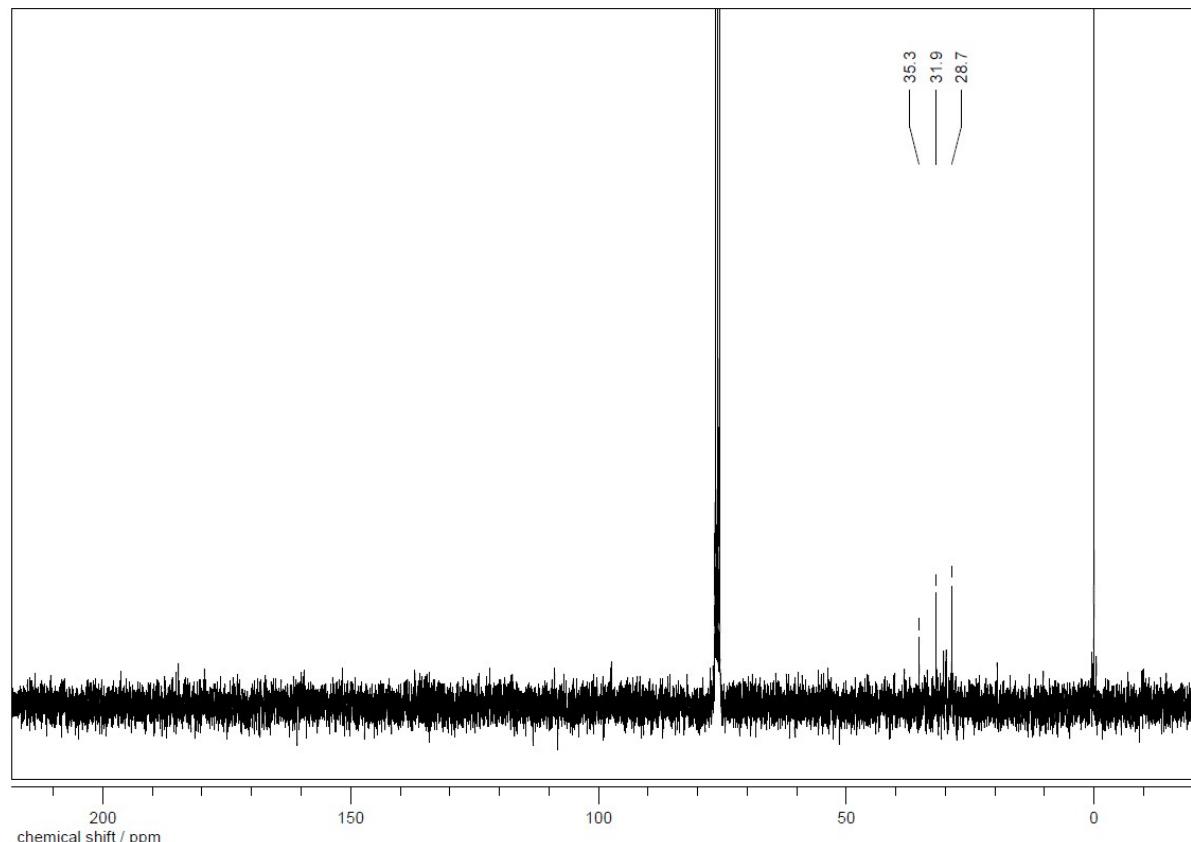
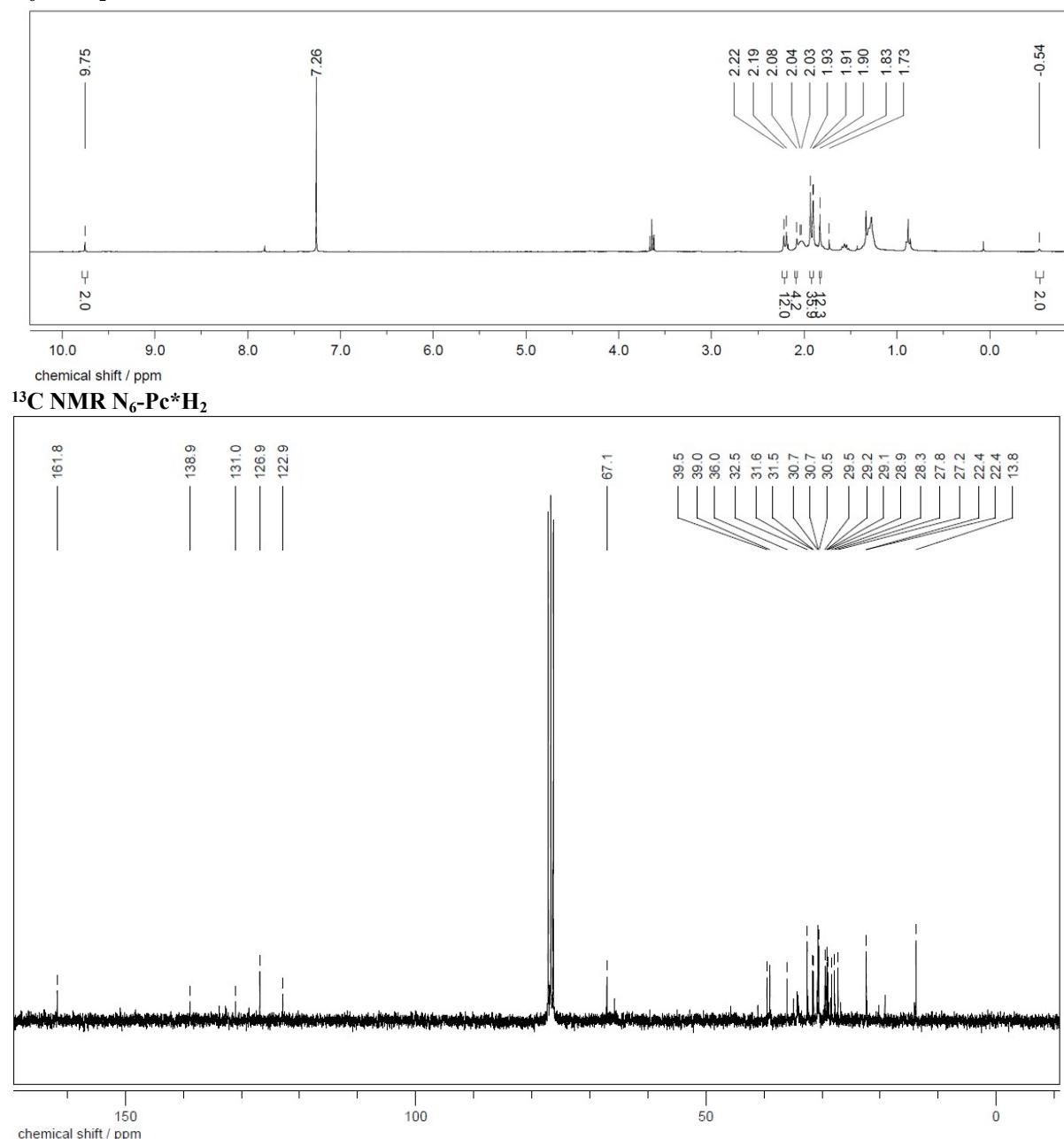


Figure SI-7c. ^{13}C NMR spectra of ABAB (above) / A_2B_2 (below) $\text{N}_4\text{-Pc}^*\text{H}_2$ in CDCl_3 , 75 MHz.

N₆-Pc*H₂



N₈-Pc*H₂ or Ppz*H₂^[3-5]

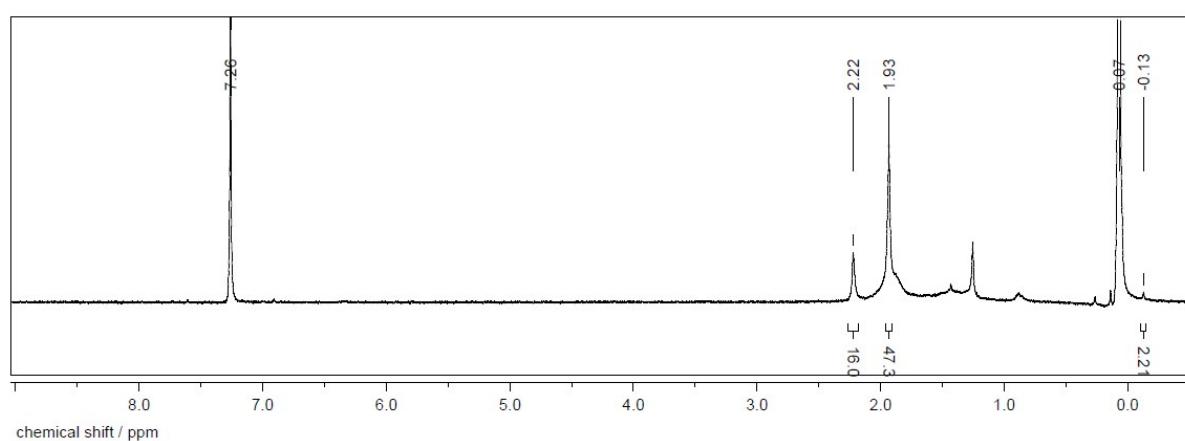
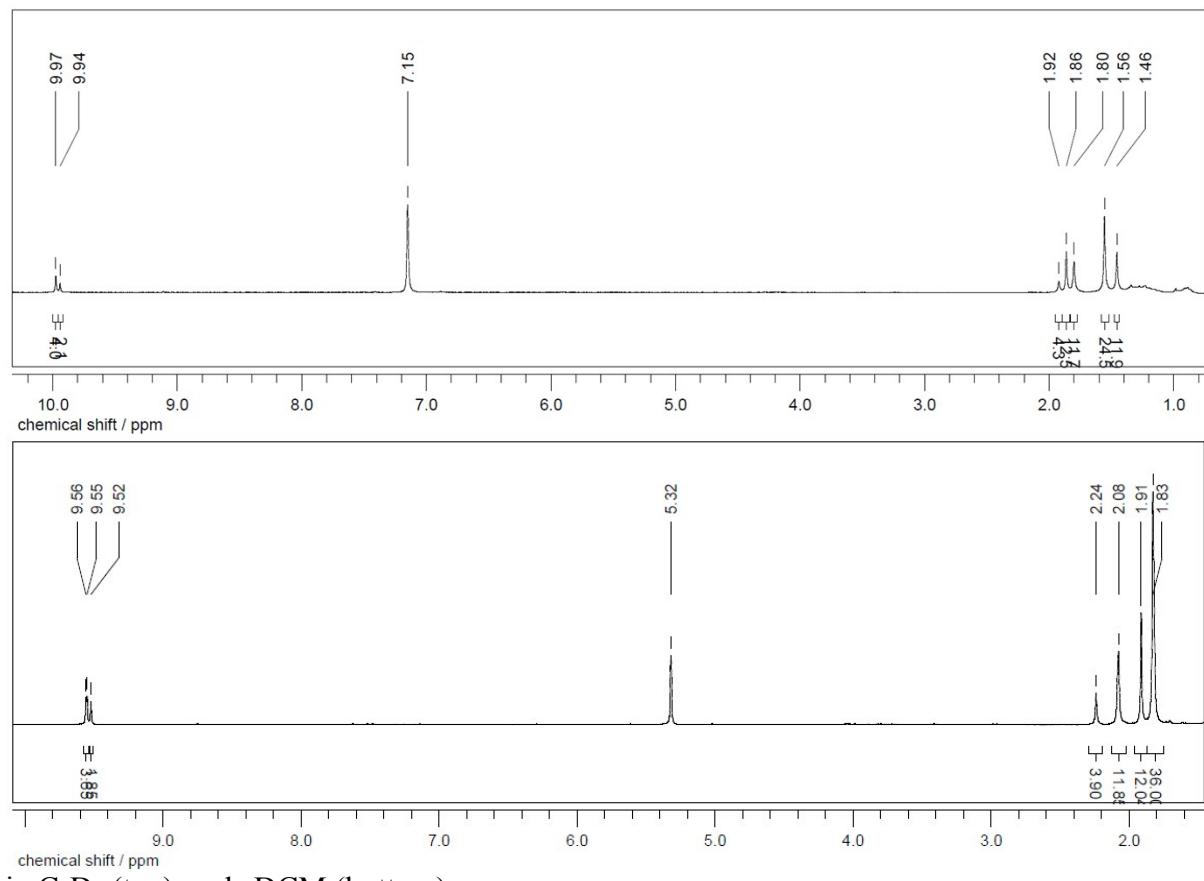


Figure SI-7a. ¹H/¹³C NMR spectrum of N_x-Pc'H₂ in CDCl₃, 300 MHz.

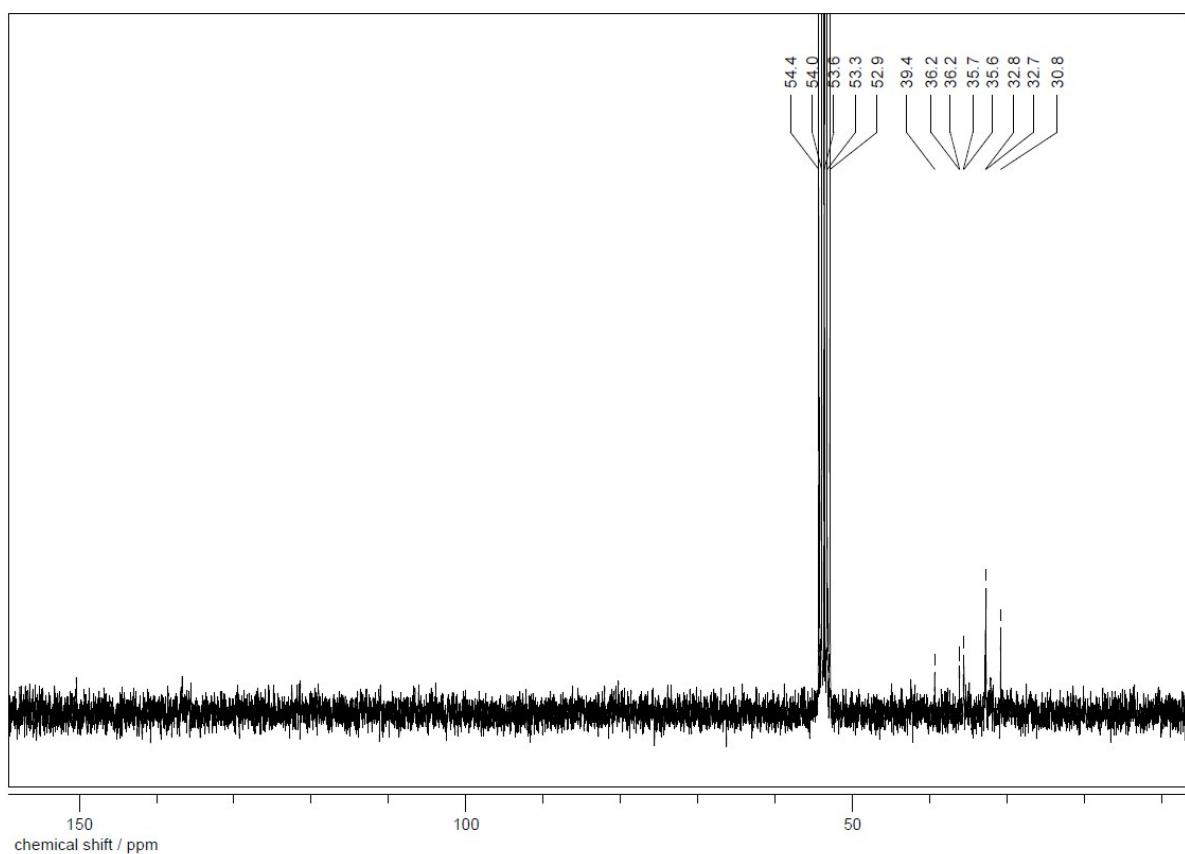
N₀-[Pc^{*}Zn] [1, 3-5]

¹H NMR N₂-[Pc^{*}Zn]

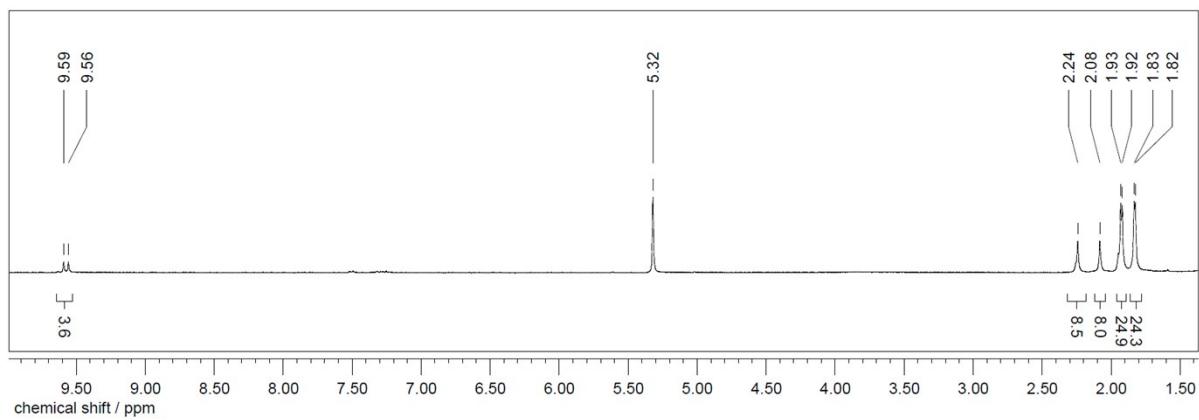


in C₆D₆ (top) or d₂-DCM (bottom).

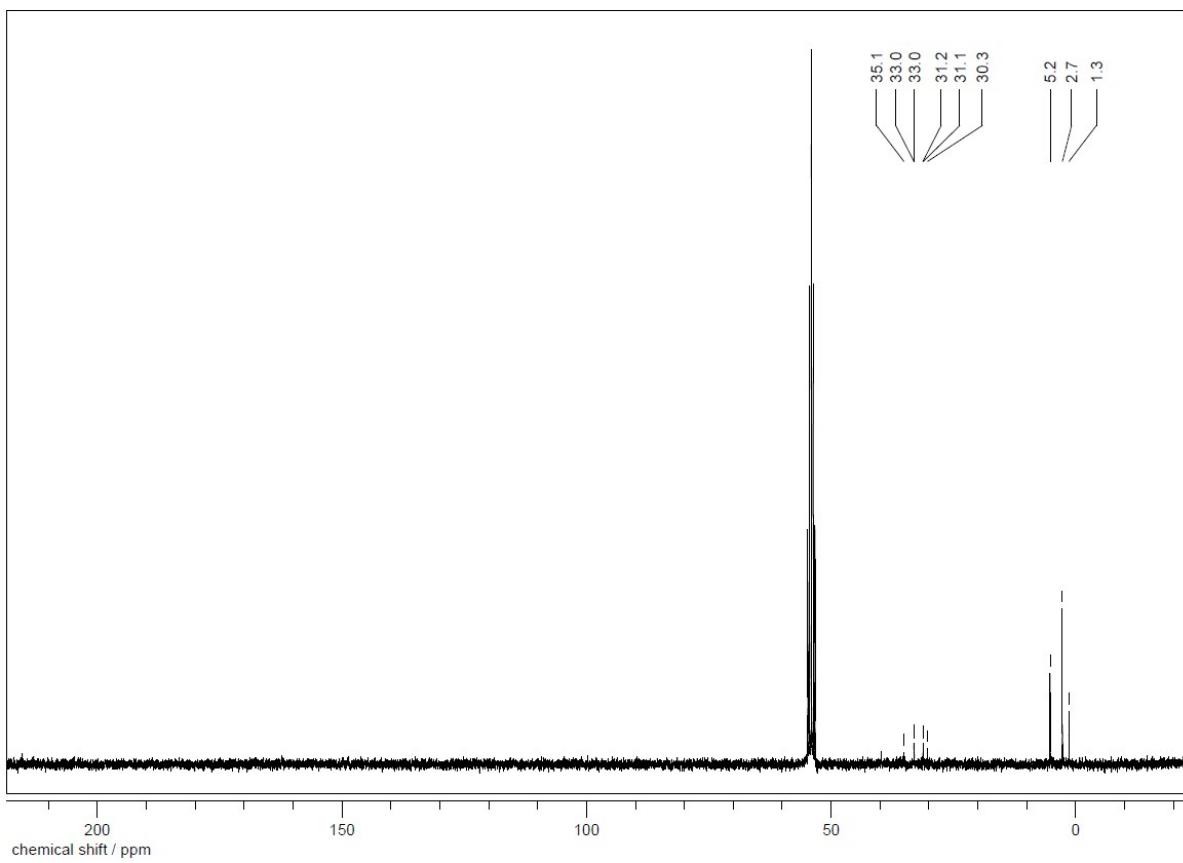
¹³C NMR N₂-[Pc*Zn]



¹H NMR N₄-[Pc*Zn]



¹³C NMR N₄-[Pc*Zn]



N₆-[Pc*Zn]

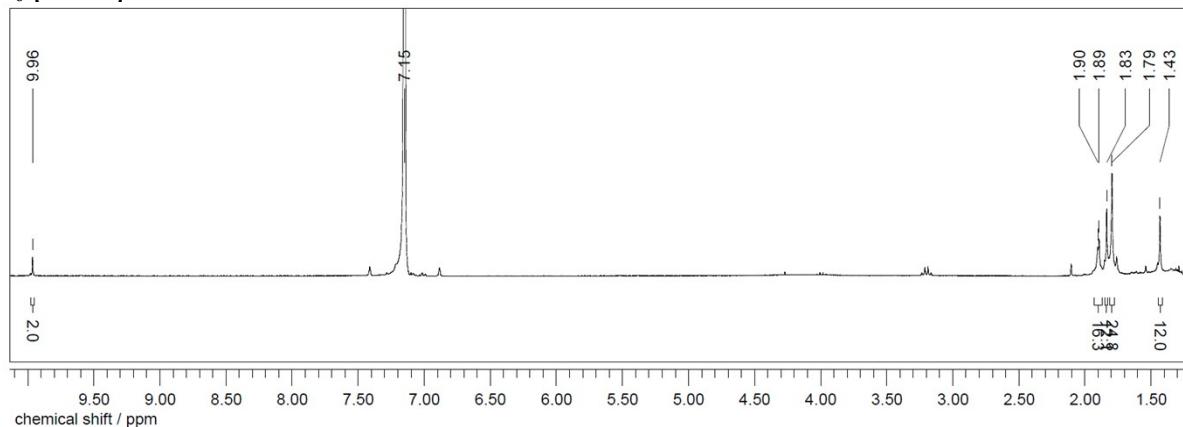
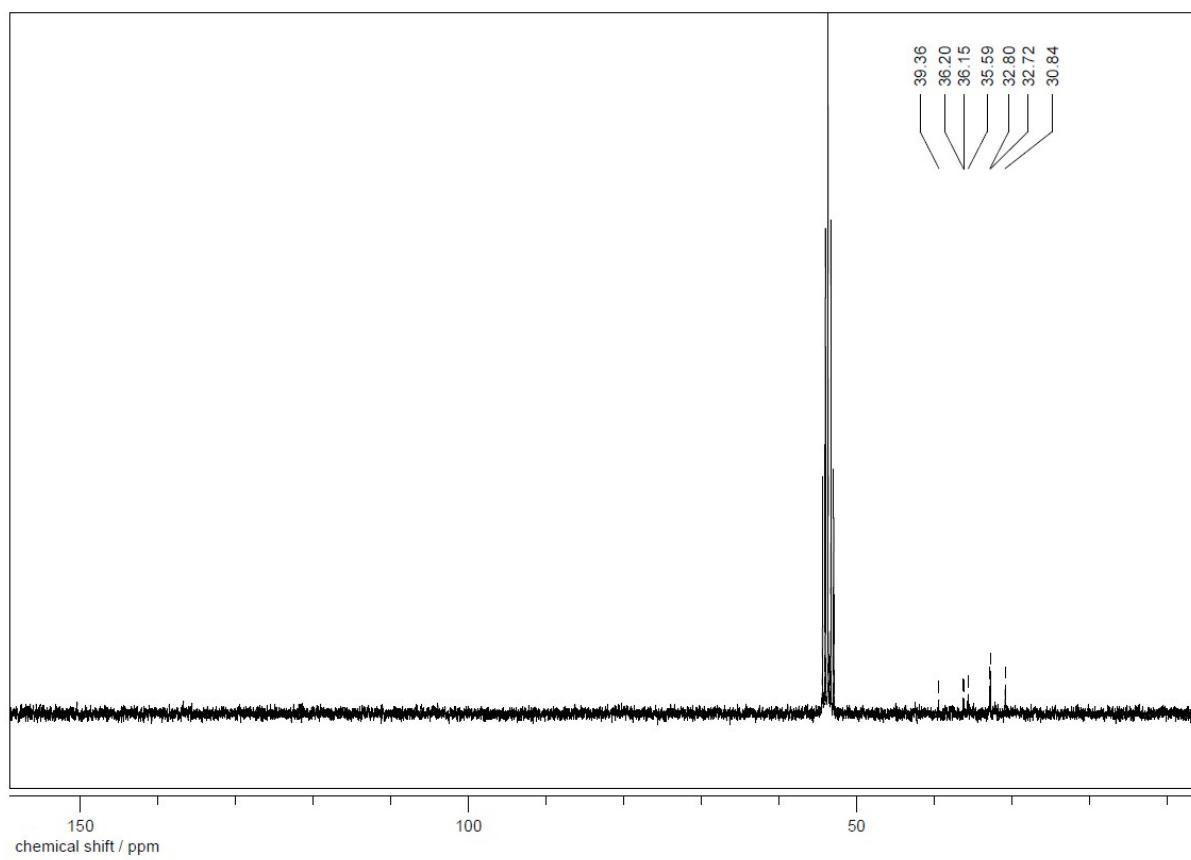


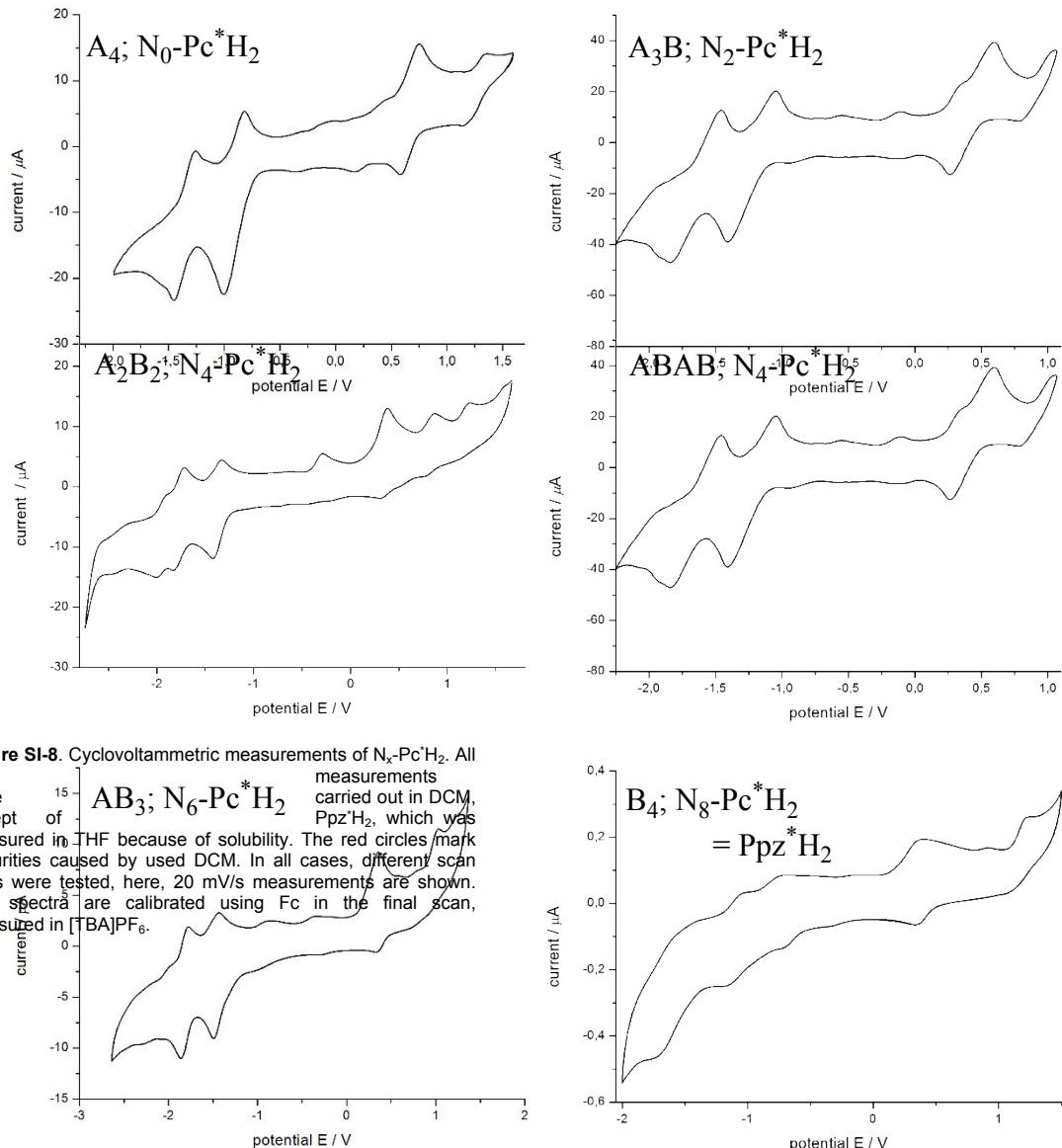
Figure SI-7c. ¹H/¹³C NMR spectrum of N_x-[Pc*Zn] with X = 2, 4, and 6 in C₆D₆, and CD₂Cl₂, respectively, 300 MHz.

¹³C NMR N₆-[Pc*Zn]



N₈-[Pc*Zn]^[3-5]

1.6 Cyclovoltammetric Measurements

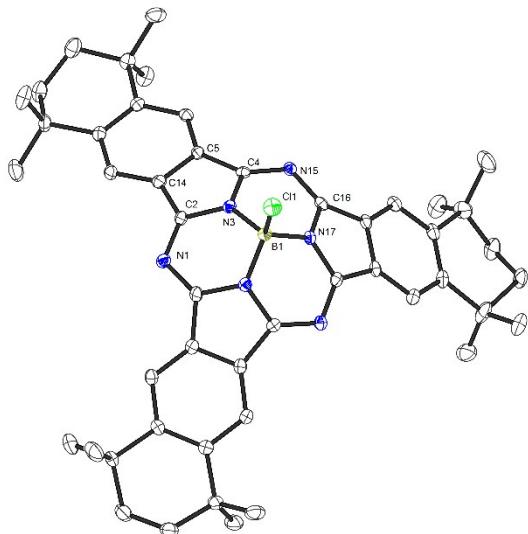


1.7 Crystal Structures

Crystals were measured with a Bruker D8 QUEST area detector. Crystal structures were resolved and refined by Dr. K. Harms of the service Department of Chemistry of the Philipps-Universität Marburg.

Diffractometer type	Bruker D8 QUEST area detector
Wavelength	0.71073 Å
Temperature	100(2) K
Index ranges	-14<=h<=14, -28<=k<=28, -15<=l<=15
Data collection software	BRUKER APEX2 2014.9-0 ^[6]
Cell refinement software	BRUKER SAINT ^[7]
Data reduction software	SAINT V8.34A (Bruker AXS Inc., 2013) ^[7]
Programs used	XT V2014/1 (Bruker AXS Inc., 2014) ^[8,9] SHELXL-2014/7 (Sheldrick, 2014) ^[8,10] DIAMOND (Crystal Impact) ^[11]

1.7.1 Crystal Structure of [Spc^{*}BCl]



Habitus, colour	prism, colourless
Crystal size	0.22 x 0.22 x 0.09 mm ³
Crystal system	Orthorhombic
Space group	Cmc2 ₁
Unit cell dimensions	Z = 4 a = 24.8236(10) Å α = 90°. b = 16.3090(6) Å β = 90°. c = 10.9728(4) Å γ = 90°. 4442.3(3) Å ³
Volume	4442.3(3) Å ³
Cell determination	9874 peaks with Theta 2.4 to 25.4°.
Empirical formula	C ₄₉ H ₅₆ B Cl ₃ N ₆
Moiety formula	C ₄₈ H ₅₄ B Cl N ₆ , CH ₂ Cl ₂
Formula weight	846.15
Density (calculated)	1.265 Mg/m ³
Absorption coefficient	0.248 mm ⁻¹
F(000)	1792
Solution and refinement:	
Reflections collected	28577
Independent reflections	4177 [R(int) = 0.0342]
Completeness to theta = 25.242°	99.9 %
Observed reflections	4031[I>2sigma(I)]
Reflections used for refinement	4177
Absorption correction	Semi-empirical from equivalents ^[12]
Max. and min. transmission	0.98 and 0.86
Flack parameter (absolute struct.)	0.001(14)
Largest diff. peak and hole	0.283 and -0.192 e.Å ⁻³
Solution	Direct methods ^[8]
Refinement	Full-matrix least-squares on F ² ^[8]
Treatment of hydrogen atoms	Calculated positions, constr. ref.
Data / restraints / parameters	4177 / 44 / 311
Goodness-of-fit on F ²	1.049
R index (all data)	wR2 = 0.0754
R index conventional [I>2sigma(I)]	R1 = 0.0293

Table SI-2. Atomic coordinates and equivalent isotropic displacement parameters (\AA^2) for ML153F_0m. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)	Occupancy
N1	0.5000	0.08718(16)	0.7070(3)	0.0163(6)	1
N3	0.54732(7)	0.15242(11)	0.54643(17)	0.0130(4)	1
N15	0.59330(7)	0.26327(11)	0.45407(16)	0.0136(4)	1
N17	0.5000	0.24139(16)	0.4157(2)	0.0127(5)	1
C2	0.54616(9)	0.11154(13)	0.6547(2)	0.0149(4)	1
C4	0.59270(8)	0.19816(13)	0.5299(2)	0.0128(4)	1
C5	0.62944(9)	0.17210(13)	0.6260(2)	0.0138(4)	1
C6	0.68271(9)	0.19023(13)	0.6541(2)	0.0141(4)	1
C7	0.70788(9)	0.15358(13)	0.7545(2)	0.0144(4)	1
C8	0.76727(10)	0.17460(14)	0.7780(2)	0.0186(5)	1
C9	0.79202(10)	0.11748(16)	0.8738(2)	0.0253(6)	1
C10	0.75505(11)	0.10051(17)	0.9799(2)	0.0257(6)	1
C11	0.70338(10)	0.05628(15)	0.9411(2)	0.0197(5)	1
C12	0.67783(9)	0.10049(13)	0.8318(2)	0.0150(4)	1
C13	0.62411(9)	0.08414(14)	0.8058(2)	0.0162(5)	1
C14	0.60011(9)	0.11829(13)	0.7033(2)	0.0148(5)	1
C16	0.54605(9)	0.28670(13)	0.40512(19)	0.0137(4)	1
C18	0.52856(9)	0.36689(14)	0.3620(2)	0.0145(4)	1
C19	0.55637(9)	0.43836(13)	0.3359(2)	0.0174(5)	1
C20	0.52846(9)	0.51054(14)	0.3088(2)	0.0180(5)	1
C21	0.56208(11)	0.58749(16)	0.2815(3)	0.0287(6)	1
C22	0.5284(2)	0.6650(3)	0.2921(7)	0.0259(11)	0.5
C23	0.4778(2)	0.6530(3)	0.2168(6)	0.0242(11)	0.5
C24	0.66584(12)	0.05663(19)	1.0520(2)	0.0328(6)	1
C25	0.71454(12)	-0.03249(16)	0.9045(3)	0.0306(6)	1
C26	0.80131(9)	0.16592(17)	0.6619(2)	0.0267(6)	1
C27	0.77020(10)	0.26457(15)	0.8206(2)	0.0242(5)	1
C28	0.59861(12)	0.57227(19)	0.1717(3)	0.0376(7)	1
C29	0.5886(4)	0.6179(6)	0.3816(10)	0.033(2)	0.5
C29A	0.6058(3)	0.6001(6)	0.3970(9)	0.0251(17)	0.5
B1	0.5000	0.1569(2)	0.4646(3)	0.0130(7)	1
Cl1	0.5000	0.07386(5)	0.34930(7)	0.02119(19)	1
C1S	0.5136(3)	0.1520(5)	1.0477(6)	0.054(3)	0.5
C11S	0.56140(14)	0.22218(19)	1.1014(2)	0.0507(6)	0.5
Cl2S	0.44951(15)	0.1926(3)	1.0606(5)	0.1026(15)	0.5

Table SI-3. Bond lengths [Å] and angles [°] for ML153F_0m.

N1-C2	1.342(3)	C20-C21	1.537(3)
N1-C2#1	1.342(3)	C21-C29	1.374(10)
N3-C2	1.363(3)	C21-C22	1.521(6)
N3-C4	1.363(3)	C21-C28	1.528(4)
N3-B1	1.480(3)	C21-C29A	1.682(8)
N15-C16	1.345(3)	C22-C23	1.514(7)
N15-C4	1.349(3)	C22-C29	1.947(10)
N17-C16	1.366(3)	C22-H22A	0.9700
N17-C16#1	1.366(3)	C22-H22B	0.9700
N17-B1	1.479(4)	C23-H23A	0.9700
C2-C14	1.446(3)	C23-H23B	0.9701
C4-C5	1.457(3)	C24-H24A	0.9800
C5-C6	1.390(3)	C24-H24B	0.9800
C5-C14	1.421(3)	C24-H24C	0.9800
C6-C7	1.401(3)	C25-H25A	0.9800
C6-H6	0.9500	C25-H25B	0.9800
C7-C12	1.423(3)	C25-H25C	0.9800
C7-C8	1.536(3)	C26-H26A	0.9800
C8-C9	1.533(3)	C26-H26B	0.9800
C8-C26	1.535(3)	C26-H26C	0.9800
C8-C27	1.542(3)	C27-H27A	0.9800
C9-C10	1.508(4)	C27-H27B	0.9800
C9-H9A	0.9900	C27-H27C	0.9800
C9-H9B	0.9900	C28-H28A	0.9800
C10-C11	1.532(3)	C28-H28B	0.9800
C10-H10A	0.9900	C28-H28C	0.9800
C10-H10B	0.9900	C29-H29A	0.9800
C11-C25	1.528(4)	C29-H29B	0.9800
C11-C24	1.533(4)	C29-H29C	0.9800
C11-C12	1.537(3)	C29A-H29D	0.9800
C12-C13	1.389(3)	C29A-H29E	0.9800
C13-C14	1.390(3)	C29A-H29F	0.9800
C13-H13	0.9500	B1-N3#1	1.480(3)
C16-C18	1.457(3)	B1-C11	1.853(4)
C18-C19	1.385(3)	C1S-Cl2S	1.730(9)
C18-C18#1	1.418(4)	C1S-Cl1S	1.750(7)
C19-C20	1.398(3)	C1S-H1S1	0.9900
C19-H19	0.9500	C1S-H1S2	0.9900
C20-C20#1	1.413(5)		
C2-N1-C2#1	117.3(3)	C7-C6-H6	119.7
C2-N3-C4	113.63(19)	C6-C7-C12	119.6(2)
C2-N3-B1	122.4(2)	C6-C7-C8	117.7(2)
C4-N3-B1	123.3(2)	C12-C7-C8	122.6(2)
C16-N15-C4	117.41(18)	C9-C8-C26	107.0(2)
C16-N17-C16#1	113.6(3)	C9-C8-C7	111.37(19)
C16-N17-B1	122.34(13)	C26-C8-C7	111.61(19)
C16#1-N17-B1	122.34(13)	C9-C8-C27	110.59(19)
N1-C2-N3	122.4(2)	C26-C8-C27	108.3(2)
N1-C2-C14	131.0(2)	C7-C8-C27	107.97(19)
N3-C2-C14	105.34(19)	C10-C9-C8	113.4(2)
N15-C4-N3	121.43(19)	C10-C9-H9A	108.9
N15-C4-C5	131.98(19)	C8-C9-H9A	108.9
N3-C4-C5	105.18(18)	C10-C9-H9B	108.9
C6-C5-C14	119.1(2)	C8-C9-H9B	108.9
C6-C5-C4	133.9(2)	H9A-C9-H9B	107.7
C14-C5-C4	106.92(19)	C9-C10-C11	112.4(2)
C5-C6-C7	120.5(2)	C9-C10-H10A	109.1
C5-C6-H6	119.7	C11-C10-H10A	109.1

C9-C10-H10B	109.1	H24A-C24-H24C	109.5
C11-C10-H10B	109.1	H24B-C24-H24C	109.5
H10A-C10-H10B	107.9	C11-C25-H25A	109.5
C25-C11-C10	111.6(2)	C11-C25-H25B	109.5
C25-C11-C24	108.8(2)	H25A-C25-H25B	109.5
C10-C11-C24	106.7(2)	C11-C25-H25C	109.5
C25-C11-C12	108.3(2)	H25A-C25-H25C	109.5
C10-C11-C12	110.0(2)	H25B-C25-H25C	109.5
C24-C11-C12	111.6(2)	C8-C26-H26A	109.5
C13-C12-C7	119.9(2)	C8-C26-H26B	109.5
C13-C12-C11	117.8(2)	H26A-C26-H26B	109.5
C7-C12-C11	122.30(19)	C8-C26-H26C	109.5
C12-C13-C14	120.0(2)	H26A-C26-H26C	109.5
C12-C13-H13	120.0	H26B-C26-H26C	109.5
C14-C13-H13	120.0	C8-C27-H27A	109.5
C13-C14-C5	120.7(2)	C8-C27-H27B	109.5
C13-C14-C2	131.7(2)	H27A-C27-H27B	109.5
C5-C14-C2	107.56(19)	C8-C27-H27C	109.5
N15-C16-N17	122.8(2)	H27A-C27-H27C	109.5
N15-C16-C18	130.1(2)	H27B-C27-H27C	109.5
N17-C16-C18	105.30(19)	C21-C28-H28A	109.5
C19-C18-C18#1	119.90(13)	C21-C28-H28B	109.5
C19-C18-C16	132.4(2)	H28A-C28-H28B	109.5
C18#1-C18-C16	107.34(12)	C21-C28-H28C	109.5
C18-C19-C20	120.4(2)	H28A-C28-H28C	109.5
C18-C19-H19	119.8	H28B-C28-H28C	109.5
C20-C19-H19	119.8	C21-C29-C22	51.0(3)
C19-C20-C20#1	119.71(13)	C21-C29-H29A	109.5
C19-C20-C21	117.4(2)	C22-C29-H29A	159.2
C20#1-C20-C21	122.90(13)	C21-C29-H29B	109.5
C29-C21-C22	84.4(5)	C22-C29-H29B	75.8
C29-C21-C28	113.9(5)	H29A-C29-H29B	109.5
C22-C21-C28	121.5(3)	C21-C29-H29C	109.5
C29-C21-C20	113.5(5)	C22-C29-H29C	86.5
C22-C21-C20	111.4(3)	H29A-C29-H29C	109.5
C28-C21-C20	110.1(2)	H29B-C29-H29C	109.5
C28-C21-C29A	103.4(4)	C21-C29A-H29D	109.5
C20-C21-C29A	107.6(4)	C21-C29A-H29E	109.5
C23-C22-C21	107.8(4)	H29D-C29A-H29E	109.5
C23-C22-C29	149.3(5)	C21-C29A-H29F	109.5
C21-C22-C29	44.6(4)	H29D-C29A-H29F	109.5
C23-C22-H22A	110.5	H29E-C29A-H29F	109.5
C21-C22-H22A	110.3	N17-B1-N3#1	105.42(18)
C29-C22-H22A	94.7	N17-B1-N3	105.42(18)
C23-C22-H22B	109.7	N3#1-B1-N3	105.0(3)
C21-C22-H22B	109.9	N17-B1-Cl1	115.7(2)
C29-C22-H22B	76.7	N3#1-B1-Cl1	112.22(16)
H22A-C22-H22B	108.6	N3-B1-Cl1	112.22(16)
C22-C23-H23A	109.0	Cl2S-C1S-Cl1S	110.2(4)
C22-C23-H23B	109.8	Cl2S-C1S-H1S1	109.6
H23A-C23-H23B	108.2	Cl1S-C1S-H1S1	109.6
C11-C24-H24A	109.5	Cl2S-C1S-H1S2	109.6
C11-C24-H24B	109.5	Cl1S-C1S-H1S2	109.6
H24A-C24-H24B	109.5	H1S1-C1S-H1S2	108.1
C11-C24-H24C	109.5		

Symmetry transformations used to generate equivalent atoms:
#1 -x+1,y,z

Table SI-4. Anisotropic displacement parameters (\AA^2) for ML153F_0m. The anisotropic displacement factor exponent takes the form: $-2\alpha^2[\ h^2 a^*{}^2 U^{11} + \dots + 2 h k a^* b^* U^{12}]$

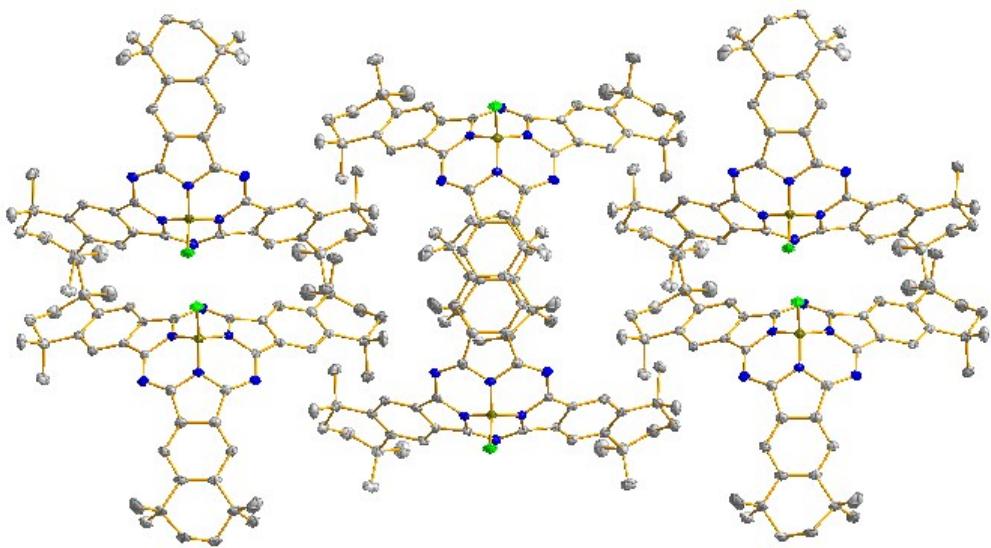
	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
N1	0.0136(13)	0.0147(13)	0.0207(14)	0.0030(11)	0.000	0.000
N3	0.0135(9)	0.0100(9)	0.0155(9)	0.0009(7)	0.0006(7)	0.0013(7)
N15	0.0156(9)	0.0129(9)	0.0123(9)	-0.0002(7)	0.0014(7)	0.0011(7)
N17	0.0153(13)	0.0113(12)	0.0114(12)	0.0010(10)	0.000	0.000
C2	0.0173(11)	0.0102(10)	0.0173(11)	0.0029(9)	0.0010(9)	0.0025(9)
C4	0.0122(10)	0.0117(10)	0.0147(10)	-0.0022(9)	0.0022(9)	0.0010(8)
C5	0.0167(11)	0.0100(10)	0.0148(10)	0.0005(8)	0.0019(8)	0.0025(9)
C6	0.0153(10)	0.0122(10)	0.0148(10)	-0.0007(9)	0.0028(9)	0.0001(9)
C7	0.0164(11)	0.0109(10)	0.0161(10)	-0.0032(9)	0.0000(9)	0.0017(8)
C8	0.0185(11)	0.0173(11)	0.0199(12)	0.0001(9)	-0.0031(9)	-0.0025(9)
C9	0.0220(12)	0.0222(12)	0.0318(15)	0.0045(11)	-0.0094(11)	-0.0037(10)
C10	0.0293(13)	0.0249(13)	0.0228(13)	0.0037(10)	-0.0088(11)	-0.0007(11)
C11	0.0214(12)	0.0212(12)	0.0165(12)	0.0062(9)	-0.0038(9)	-0.0004(10)
C12	0.0191(10)	0.0128(10)	0.0132(11)	-0.0013(8)	-0.0010(9)	0.0024(8)
C13	0.0188(11)	0.0141(10)	0.0159(10)	0.0025(9)	0.0013(9)	0.0019(9)
C14	0.0157(11)	0.0118(10)	0.0168(11)	0.0001(8)	0.0017(9)	0.0014(8)
C16	0.0168(11)	0.0143(11)	0.0100(10)	-0.0016(9)	0.0027(9)	-0.0014(9)
C18	0.0201(11)	0.0139(10)	0.0094(9)	0.0008(8)	0.0016(9)	0.0012(9)
C19	0.0190(10)	0.0170(10)	0.0163(11)	0.0021(9)	0.0024(10)	-0.0019(9)
C20	0.0263(12)	0.0137(11)	0.0139(10)	0.0023(9)	0.0025(9)	-0.0004(10)
C21	0.0269(14)	0.0163(12)	0.0429(16)	0.0090(11)	0.0101(12)	-0.0015(11)
C22	0.032(3)	0.013(2)	0.033(3)	0.004(2)	-0.007(2)	-0.004(2)
C23	0.028(2)	0.015(2)	0.030(3)	0.010(2)	-0.003(2)	-0.0007(18)
C24	0.0364(15)	0.0440(17)	0.0179(13)	0.0104(12)	-0.0024(11)	-0.0013(13)
C25	0.0389(15)	0.0209(13)	0.0318(13)	0.0056(11)	-0.0133(13)	0.0037(11)
C26	0.0145(11)	0.0363(14)	0.0293(14)	-0.0023(12)	-0.0007(11)	-0.0002(10)
C27	0.0251(12)	0.0206(12)	0.0268(13)	-0.0015(10)	-0.0076(10)	-0.0041(10)
C28	0.0354(16)	0.0410(16)	0.0362(17)	0.0160(14)	0.0075(13)	-0.0115(13)
C29	0.030(5)	0.024(4)	0.046(5)	0.008(3)	-0.010(4)	-0.006(3)
C29A	0.022(4)	0.018(4)	0.035(4)	0.002(3)	-0.016(3)	-0.005(3)
B1	0.0137(17)	0.0111(16)	0.0141(17)	-0.0016(14)	0.000	0.000
Cl1	0.0270(4)	0.0151(4)	0.0215(4)	-0.0066(3)	0.000	0.000
C1S	0.081(8)	0.047(3)	0.035(3)	0.002(3)	-0.009(3)	-0.023(4)
C11S	0.0548(13)	0.0534(15)	0.0440(11)	0.0083(9)	-0.0010(9)	-0.0125(11)
Cl2S	0.062(2)	0.095(3)	0.151(4)	0.058(3)	-0.045(2)	-0.032(2)

Table SI-5. Hydrogen coordinates and isotropic displacement parameters (\AA^2) for ML153F_0m.

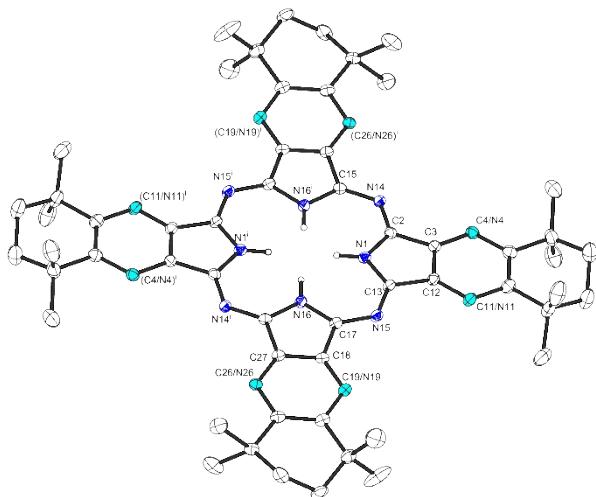
	x	y	z	U(eq)	Occupancy
H6	0.7022	0.2278	0.6048	0.017	1
H9A	0.8016	0.0648	0.8344	0.030	1
H9B	0.8257	0.1425	0.9048	0.030	1
H10A	0.7454	0.1530	1.0196	0.031	1
H10B	0.7743	0.0665	1.0405	0.031	1
H13	0.6038	0.0496	0.8582	0.019	1
H19	0.5946	0.4383	0.3364	0.021	1
H22A	0.5486	0.7120	0.2629	0.031	0.5
H22B	0.5187	0.6744	0.3766	0.031	0.5
H23A	0.4878	0.6332	0.1366	0.029	0.5
H23B	0.4593	0.7050	0.2071	0.029	0.5
H24A	0.6331	0.0256	1.0330	0.049	1
H24B	0.6562	0.1133	1.0725	0.049	1
H24C	0.6842	0.0313	1.1215	0.049	1
H25A	0.6806	-0.0596	0.8834	0.046	1
H25B	0.7315	-0.0614	0.9727	0.046	1
H25C	0.7387	-0.0334	0.8339	0.046	1
H26A	0.8392	0.1772	0.6810	0.040	1
H26B	0.7886	0.2050	0.6005	0.040	1
H26C	0.7979	0.1100	0.6302	0.040	1
H27A	0.8079	0.2797	0.8350	0.036	1
H27B	0.7497	0.2709	0.8963	0.036	1
H27C	0.7549	0.3003	0.7577	0.036	1
H28A	0.6183	0.6227	0.1519	0.056	1
H28B	0.5767	0.5558	0.1016	0.056	1
H28C	0.6243	0.5286	0.1911	0.056	1
H29A	0.6135	0.5764	0.4130	0.050	0.5
H29B	0.5623	0.6320	0.4449	0.050	0.5
H29C	0.6088	0.6671	0.3584	0.050	0.5
H29D	0.6335	0.5572	0.3933	0.038	0.5
H29E	0.5866	0.5963	0.4748	0.038	0.5
H29F	0.6230	0.6540	0.3904	0.038	0.5
H1S1	0.5212	0.1388	0.9613	0.065	0.5
H1S2	0.5160	0.1006	1.0955	0.065	0.5

Table SI-6. Torsion angles [°] for ML153F_0m.

C2#1-N1-C2-N3	-6.4(4)		
C2#1-N1-C2-C14	158.84(17)	C16#1-N17-C16-C18	-10.8(3)
C4-N3-C2-N1	156.6(2)	B1-N17-C16-C18	-176.2(2)
B1-N3-C2-N1	-14.2(3)	N15-C16-C18-C19	14.2(4)
C4-N3-C2-C14	-11.9(2)	N17-C16-C18-C19	179.0(3)
B1-N3-C2-C14	177.3(2)	N15-C16-C18-C18#1	-158.6(2)
C16-N15-C4-N3	6.9(3)	N17-C16-C18-C18#1	6.19(19)
C16-N15-C4-C5	-157.4(2)	C18#1-C18-C19-C20	-0.4(3)
C2-N3-C4-N15	-156.0(2)	C16-C18-C19-C20	-172.5(2)
B1-N3-C4-N15	14.7(3)	C18-C19-C20-C20#1	0.4(3)
C2-N3-C4-C5	12.0(2)	C18-C19-C20-C21	179.7(2)
B1-N3-C4-C5	-177.4(2)	C19-C20-C21-C29	-69.4(5)
N15-C4-C5-C6	-19.3(4)	C20#1-C20-C21-C29	109.9(4)
N3-C4-C5-C6	174.6(2)	C19-C20-C21-C22	-162.6(4)
N15-C4-C5-C14	159.3(2)	C20#1-C20-C21-C22	16.7(4)
N3-C4-C5-C14	-6.9(2)	C19-C20-C21-C28	59.7(3)
C14-C5-C6-C7	2.4(3)	C20#1-C20-C21-C28	-121.05(18)
C4-C5-C6-C7	-179.2(2)	C19-C20-C21-C29A	-52.3(4)
C5-C6-C7-C12	-3.1(3)	C20#1-C20-C21-C29A	127.0(4)
C5-C6-C7-C8	178.3(2)	C29-C21-C22-C23	-163.8(6)
C6-C7-C8-C9	-168.6(2)	C28-C21-C22-C23	81.5(5)
C12-C7-C8-C9	12.9(3)	C20-C21-C22-C23	-50.7(5)
C6-C7-C8-C26	-49.0(3)	C28-C21-C22-C29	-114.6(6)
C12-C7-C8-C26	132.4(2)	C20-C21-C22-C29	113.1(5)
C6-C7-C8-C27	69.8(3)	C28-C21-C29-C22	122.0(4)
C12-C7-C8-C27	-108.7(2)	C20-C21-C29-C22	-110.9(4)
C26-C8-C9-C10	-163.7(2)	C16-N17-B1-N3#1	137.5(2)
C7-C8-C9-C10	-41.5(3)	C16#1-N17-B1-N3#1	-26.6(3)
C27-C8-C9-C10	78.6(3)	C16-N17-B1-N3	26.6(3)
C8-C9-C10-C11	62.5(3)	C16#1-N17-B1-N3	-137.5(2)
C9-C10-C11-C25	71.0(3)	C16-N17-B1-Cl1	-98.0(2)
C9-C10-C11-C24	-170.3(2)	C16#1-N17-B1-Cl1	98.0(2)
C9-C10-C11-C12	-49.2(3)	C2-N3-B1-N17	140.7(2)
C6-C7-C12-C13	1.2(3)	C4-N3-B1-N17	-29.2(3)
C8-C7-C12-C13	179.7(2)	C2-N3-B1-N3#1	29.6(3)
C6-C7-C12-C11	178.1(2)	C4-N3-B1-N3#1	-140.25(18)
C8-C7-C12-C11	-3.4(3)	C2-N3-B1-Cl1	-92.6(2)
C25-C11-C12-C13	75.7(3)	C4-N3-B1-Cl1	97.6(2)
C10-C11-C12-C13	-162.2(2)		
C24-C11-C12-C13	-44.1(3)		
C25-C11-C12-C7	-101.3(3)		
C10-C11-C12-C7	20.8(3)		
C24-C11-C12-C7	139.0(2)		
C7-C12-C13-C14	1.3(3)		
C11-C12-C13-C14	-175.7(2)		
C12-C13-C14-C5	-2.0(3)		
C12-C13-C14-C2	179.6(2)		
C6-C5-C14-C13	0.1(3)		
C4-C5-C14-C13	-178.7(2)		
C6-C5-C14-C2	178.88(19)		
C4-C5-C14-C2	0.1(2)		
N1-C2-C14-C13	18.2(4)		
N3-C2-C14-C13	-174.7(2)		
N1-C2-C14-C5	-160.4(3)		
N3-C2-C14-C5	6.7(2)		
C4-N15-C16-N17	-9.1(3)		
C4-N15-C16-C18	153.4(2)		
C16#1-N17-C16-N15	155.38(15)		
B1-N17-C16-N15	-10.0(4)		



1.7.2 Crystal Structure of N₂-Pc*H₂



Habitus, colour	plate, dark
Crystal size	0.29 x 0.22 x 0.05 mm ³
Crystal system	Monoclinic
Space group	P2 ₁ /c
Unit cell dimensions	Z = 2 a = 11.7078(4) Å b = 23.2759(7) Å c = 12.9884(5) Å 3539.3(2) Å ³
Volume	9908 peaks with Theta 2.3 to 25.2°.
Cell determination	C ₆₆ H ₇₆ Cl ₁₂ N ₁₀
Empirical formula	C ₆₂ H ₇₂ N ₁₀ , 4(CHCl ₃)
Moiety formula	1434.76
Formula weight	1.346 Mg/m ³
Density (calculated)	0.516 mm ⁻¹
Absorption coefficient	1492
F(000)	
Solution and refinement:	
Reflections collected	35003
Independent reflections	6438 [R(int) = 0.0640]
Completeness to theta = 25.242°	99.9 %
Observed reflections	4815[I>2sigma(I)]
Reflections used for refinement	6438
Absorption correction	Semi-empirical from equivalents [12]
Max. and min. transmission	0.97 and 0.84
Largest diff. peak and hole	0.491 and -0.364 e.Å ⁻³
Solution	Direct methods [8]
Refinement	Full-matrix least-squares on F ² [8]
Treatment of hydrogen atoms	Mixture of constr. and indep. refinement
Data / restraints / parameters	6438 / 203 / 602
Goodness-of-fit on F ²	1.023
R index (all data)	wR2 = 0.1178
R index conventional [I>2sigma(I)]	R1 = 0.0489

Table SI-7. Atomic coordinates and equivalent isotropic displacement parameters (\AA^2) for ML026F1_0m. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)	Occupancy
N1	0.46545(16)	0.57666(8)	0.44525(14)	0.0153(4)	1
C2	0.3647(2)	0.60621(10)	0.45428(17)	0.0157(5)	1
C3	0.3755(2)	0.66129(10)	0.40227(17)	0.0156(5)	1
C4	0.30198(19)	0.70643(10)	0.38898(17)	0.0184(5)	0.75
N4	0.30198(19)	0.70643(10)	0.38898(17)	0.0184(5)	0.25
C5	0.3388(2)	0.75406(11)	0.33488(19)	0.0204(5)	1
C6	0.2532(2)	0.80284(11)	0.3170(2)	0.0268(6)	1
C7	0.3150(2)	0.85787(12)	0.2849(2)	0.0333(7)	1
C8	0.4002(2)	0.84750(12)	0.2011(2)	0.0338(7)	1
C9	0.4967(2)	0.80731(11)	0.2343(2)	0.0268(6)	1
C10	0.4506(2)	0.75531(11)	0.29354(18)	0.0202(5)	1
C11	0.5243(2)	0.70961(10)	0.30888(17)	0.0193(5)	0.75
N11	0.5243(2)	0.70961(10)	0.30888(17)	0.0193(5)	0.25
C12	0.4863(2)	0.66329(10)	0.36324(17)	0.0160(5)	1
C13	0.54049(19)	0.60941(10)	0.39127(17)	0.0154(5)	1
N14	0.27091(16)	0.59019(8)	0.50347(14)	0.0148(4)	1
N15	0.64678(16)	0.59652(8)	0.36288(15)	0.0171(4)	1
C15	0.26385(19)	0.54088(10)	0.55360(17)	0.0151(5)	1
N16	0.65576(16)	0.50203(8)	0.44055(14)	0.0152(4)	1
C17	0.6968(2)	0.54665(10)	0.38416(17)	0.0161(5)	1
C18	0.81001(19)	0.53126(10)	0.34750(17)	0.0154(5)	1
C19	0.88411(19)	0.55965(10)	0.28424(17)	0.0177(5)	0.75
N19	0.88411(19)	0.55965(10)	0.28424(17)	0.0177(5)	0.25
C20	0.9858(2)	0.53270(10)	0.25919(18)	0.0178(5)	1
C21	1.0653(2)	0.56474(11)	0.18526(19)	0.0230(6)	1
C22	1.1584(4)	0.5251(2)	0.1448(4)	0.0275(11)	0.8
C23	1.2118(3)	0.48884(18)	0.2312(3)	0.0250(9)	0.8
C22A	1.1861(17)	0.5330(11)	0.1886(15)	0.039(6)	0.2
C23A	1.1811(17)	0.4745(8)	0.1838(17)	0.040(4)	0.2
C24	1.1240(2)	0.44669(11)	0.27591(18)	0.0190(5)	1
C25	1.0129(2)	0.47795(10)	0.30069(17)	0.0170(5)	1
C26	0.93612(18)	0.45022(10)	0.36436(17)	0.0162(5)	0.75
N26	0.93612(18)	0.45022(10)	0.36436(17)	0.0162(5)	0.25
C27	0.83524(19)	0.47700(10)	0.38659(17)	0.0158(5)	1
C28	0.9988(3)	0.58457(17)	0.0908(2)	0.0527(10)	1
C29	1.1156(3)	0.61711(13)	0.2400(3)	0.0416(8)	1
C30	1.1785(5)	0.4210(2)	0.3759(4)	0.0294(12)	0.8
C31	1.1000(5)	0.3983(2)	0.2006(4)	0.0241(11)	0.8
C30A	1.193(2)	0.4464(8)	0.3646(18)	0.035(6)	0.2
C31A	1.0955(19)	0.3834(11)	0.2407(18)	0.031(5)	0.2
C33	0.1689(2)	0.78408(13)	0.2317(2)	0.0380(7)	1
C34	0.1873(3)	0.81629(13)	0.4147(3)	0.0427(8)	1
C35	0.5593(3)	0.78816(14)	0.1376(2)	0.0475(9)	1
C36	0.5807(3)	0.83935(13)	0.3052(3)	0.0411(8)	1
C1S	0.4189(4)	1.07467(18)	0.3759(3)	0.0589(10)	1
Cl1	0.3082(13)	0.9987(7)	0.4049(13)	0.087(4)	0.089(3)
Cl2	0.384(2)	0.9986(8)	0.3634(16)	0.080(4)	0.079(3)
Cl3	0.4567(5)	1.00427(14)	0.3302(4)	0.0668(13)	0.635(3)
Cl4	0.4849(10)	1.0172(5)	0.3321(10)	0.069(3)	0.234(3)
Cl5	0.5264(15)	1.0658(9)	0.3633(14)	0.080(4)	0.070(2)
Cl6	0.54339(12)	1.11200(7)	0.41534(12)	0.0615(4)	0.843(3)
Cl7	0.488(3)	1.1245(12)	0.411(2)	0.077(6)	0.054(3)
Cl8	0.345(2)	1.1101(9)	0.4684(17)	0.137(8)	0.103(3)
Cl9	0.31974(12)	1.07437(15)	0.47132(11)	0.0788(8)	0.794(3)
Cl10	0.3195(16)	1.0394(10)	0.4564(17)	0.098(5)	0.099(3)
C2S	0.8490(3)	0.70760(18)	0.4147(3)	0.0404(10)	0.9

C2SA	0.962(4)	0.733(2)	0.560(4)	0.078(6)	0.1
Cl11	0.844(2)	0.7483(13)	0.650(2)	0.092(6)	0.049(2)
Cl12	0.788(2)	0.8059(13)	0.5125(17)	0.090(5)	0.058(3)
Cl13	0.80543(14)	0.76779(9)	0.49037(10)	0.0609(5)	0.785(3)
Cl14	0.7662(10)	0.7308(7)	0.4884(9)	0.088(3)	0.123(3)
Cl15	0.887(2)	0.7915(12)	0.4584(19)	0.097(5)	0.058(2)
Cl16	0.7871(13)	0.6961(7)	0.4299(14)	0.090(5)	0.095(3)
Cl17	0.85946(7)	0.72932(4)	0.28604(6)	0.0343(3)	0.884(2)
Cl18	0.9371(18)	0.6712(10)	0.4653(18)	0.068(5)	0.079(3)
Cl19	0.98215(13)	0.68209(7)	0.46156(13)	0.0437(4)	0.809(3)
Cl20	0.9905(18)	0.7094(11)	0.524(2)	0.081(5)	0.061(3)

Table SI-8. Bond lengths [Å] and angles [°] for ML026F1_0m.

N1-C13	1.363(3)	C22A-H22C	0.9900
N1-C2	1.371(3)	C22A-H22D	0.9900
N1-H1	0.8800	C23A-C24	1.521(19)
C2-N14	1.330(3)	C23A-H23C	0.9900
C2-C3	1.455(3)	C23A-H23D	0.9900
C3-N4	1.368(3)	C24-C30A	1.40(3)
C3-C4	1.368(3)	C24-C31	1.516(7)
C3-C12	1.398(3)	C24-C25	1.528(3)
C4-C5	1.384(3)	C24-C30	1.561(6)
C4-H4	0.9500	C24-C31A	1.58(3)
N4-C5	1.384(3)	C25-N26	1.387(3)
C5-C10	1.421(3)	C25-C26	1.387(3)
C5-C6	1.531(3)	C26-C27	1.369(3)
C6-C34	1.524(4)	C26-H26	0.9500
C6-C7	1.532(4)	N26-C27	1.369(3)
C6-C33	1.539(4)	C27-C15#1	1.464(3)
C7-C8	1.503(4)	C28-H28A	0.9800
C7-H7A	0.9900	C28-H28B	0.9800
C7-H7B	0.9900	C28-H28C	0.9800
C8-C9	1.525(4)	C29-H29A	0.9800
C8-H8A	0.9900	C29-H29B	0.9800
C8-H8B	0.9900	C29-H29C	0.9800
C9-C35	1.527(4)	C30-H30A	0.9800
C9-C36	1.534(4)	C30-H30B	0.9800
C9-C10	1.535(3)	C30-H30C	0.9800
C10-N11	1.382(3)	C31-H31A	0.9800
C10-C11	1.382(3)	C31-H31B	0.9800
C11-C12	1.366(3)	C31-H31C	0.9800
C11-H11	0.9500	C30A-H30D	0.9800
N11-C12	1.366(3)	C30A-H30E	0.9800
C12-C13	1.450(3)	C30A-H30F	0.9800
C13-N15	1.336(3)	C31A-H31D	0.9800
N14-C15	1.323(3)	C31A-H31E	0.9800
N15-C17	1.328(3)	C31A-H31F	0.9800
C15-N16#1	1.374(3)	C33-H33A	0.9800
C15-C27#1	1.464(3)	C33-H33B	0.9800
N16-C17	1.362(3)	C33-H33C	0.9800
N16-C15#1	1.374(3)	C34-H34A	0.9800
N16-H16	0.8800	C34-H34B	0.9800
C17-C18	1.458(3)	C34-H34C	0.9800
C18-N19	1.371(3)	C35-H35A	0.9800
C18-C19	1.371(3)	C35-H35B	0.9800
C18-C27	1.392(3)	C35-H35C	0.9800
C19-C20	1.388(3)	C36-H36A	0.9800
C19-H19	0.9500	C36-H36B	0.9800
N19-C20	1.388(3)	C36-H36C	0.9800
C20-C25	1.418(3)	C1S-Cl5	1.287(17)
C20-C21	1.537(3)	C1S-Cl7	1.48(3)
C21-C28	1.518(4)	C1S-Cl4	1.648(12)
C21-C29	1.527(4)	C1S-Cl8	1.701(18)
C21-C22	1.527(6)	C1S-Cl9	1.708(4)
C21-C22A	1.60(2)	C1S-Cl6	1.768(5)
C22-C23	1.531(6)	C1S-Cl10	1.774(18)
C22-H22A	0.9900	C1S-Cl3	1.799(6)
C22-H22B	0.9900	C1S-Cl2	1.82(2)
C23-C24	1.539(5)	C1S-Cl1	2.227(18)
C23-H23A	0.9900	C1S-H1S	1.09(5)
C23-H23B	0.9900	Cl1-Cl2	1.04(3)
C22A-C23A	1.36(3)	Cl1-Cl10	1.17(3)

C11-C14	2.32(2)	C2SA-C118	1.92(6)
Cl2-Cl4	1.33(3)	C2SA-Cl15	2.08(6)
Cl2-Cl10	1.72(3)	C2SA-H2SA	1.0000
Cl2-Cl5	2.29(3)	Cl11-Cl14	2.32(3)
Cl4-Cl5	1.29(2)	Cl11-Cl12	2.32(4)
Cl5-Cl7	1.57(4)	Cl12-Cl15	1.40(3)
Cl7-Cl8	1.87(4)	Cl12-Cl14	1.79(3)
Cl8-Cl10	1.68(3)	Cl14-Cl16	1.138(18)
C2S-Cl17	1.751(4)	Cl14-Cl15	2.04(3)
C2S-Cl19	1.770(4)	Cl14-Cl18	2.46(2)
C2S-Cl13	1.789(5)	Cl15-Cl20	2.41(3)
C2S-H2S	1.0000	Cl16-Cl18	1.90(2)
C2SA-Cl11	1.85(6)	Cl18-Cl20	1.32(3)
C13-N1-C2	108.86(19)	C11-C10-C9	117.2(2)
C13-N1-H1	125.6	C5-C10-C9	122.4(2)
C2-N1-H1	125.6	C12-C11-C10	118.4(2)
N14-C2-N1	128.0(2)	C12-C11-H11	120.8
N14-C2-C3	123.1(2)	C10-C11-H11	120.8
N1-C2-C3	108.85(19)	C12-N11-C10	118.4(2)
N4-C3-C12	120.9(2)	N11-C12-C3	121.5(2)
C4-C3-C12	120.9(2)	C11-C12-C3	121.5(2)
N4-C3-C2	132.7(2)	N11-C12-C13	131.9(2)
C4-C3-C2	132.7(2)	C11-C12-C13	131.9(2)
C12-C3-C2	106.4(2)	C3-C12-C13	106.58(19)
C3-C4-C5	118.7(2)	N15-C13-N1	128.6(2)
C3-C4-H4	120.6	N15-C13-C12	122.1(2)
C5-C4-H4	120.6	N1-C13-C12	109.27(19)
C3-N4-C5	118.7(2)	C15-N14-C2	122.4(2)
N4-C5-C10	120.1(2)	C17-N15-C13	123.3(2)
C4-C5-C10	120.1(2)	N14-C15-N16#1	127.7(2)
N4-C5-C6	117.7(2)	N14-C15-C27#1	124.2(2)
C4-C5-C6	117.7(2)	N16#1-C15-C27#1	108.09(19)
C10-C5-C6	122.2(2)	C17-N16-C15#1	109.80(19)
C34-C6-C5	111.3(2)	C17-N16-H16	125.1
C34-C6-C7	107.4(2)	C15#1-N16-H16	125.1
C5-C6-C7	110.5(2)	N15-C17-N16	128.4(2)
C34-C6-C33	109.4(2)	N15-C17-C18	123.2(2)
C5-C6-C33	108.2(2)	N16-C17-C18	108.4(2)
C7-C6-C33	110.1(2)	N19-C18-C27	121.5(2)
C8-C7-C6	112.5(2)	C19-C18-C27	121.5(2)
C8-C7-H7A	109.1	N19-C18-C17	131.3(2)
C6-C7-H7A	109.1	C19-C18-C17	131.3(2)
C8-C7-H7B	109.1	C27-C18-C17	107.1(2)
C6-C7-H7B	109.1	C18-C19-C20	118.2(2)
H7A-C7-H7B	107.8	C18-C19-H19	120.9
C7-C8-C9	112.9(2)	C20-C19-H19	120.9
C7-C8-H8A	109.0	C18-N19-C20	118.2(2)
C9-C8-H8A	109.0	N19-C20-C25	120.4(2)
C7-C8-H8B	109.0	C19-C20-C25	120.4(2)
C9-C8-H8B	109.0	N19-C20-C21	117.0(2)
H8A-C8-H8B	107.8	C19-C20-C21	117.0(2)
C8-C9-C35	107.9(2)	C25-C20-C21	122.6(2)
C8-C9-C36	109.8(2)	C28-C21-C29	109.0(3)
C35-C9-C36	109.0(3)	C28-C21-C22	105.5(3)
C8-C9-C10	111.2(2)	C29-C21-C22	111.7(3)
C35-C9-C10	110.9(2)	C28-C21-C20	110.1(2)
C36-C9-C10	108.0(2)	C29-C21-C20	109.2(2)
N11-C10-C5	120.4(2)	C22-C21-C20	111.1(3)
C11-C10-C5	120.4(2)	C28-C21-C22A	127.5(7)
N11-C10-C9	117.2(2)	C29-C21-C22A	91.1(10)

C20-C21-C22A	107.5(8)	H29A-C29-H29C	109.5
C21-C22-C23	111.6(3)	H29B-C29-H29C	109.5
C21-C22-H22A	109.3	C24-C30-H30A	109.5
C23-C22-H22A	109.3	C24-C30-H30B	109.5
C21-C22-H22B	109.3	H30A-C30-H30B	109.5
C23-C22-H22B	109.3	C24-C30-H30C	109.5
H22A-C22-H22B	108.0	H30A-C30-H30C	109.5
C22-C23-C24	111.0(3)	H30B-C30-H30C	109.5
C22-C23-H23A	109.4	C24-C31-H31A	109.5
C24-C23-H23A	109.4	C24-C31-H31B	109.5
C22-C23-H23B	109.4	H31A-C31-H31B	109.5
C24-C23-H23B	109.4	C24-C31-H31C	109.5
H23A-C23-H23B	108.0	H31A-C31-H31C	109.5
C23A-C22A-C21	115.0(18)	H31B-C31-H31C	109.5
C23A-C22A-H22C	108.5	C24-C30A-H30D	109.5
C21-C22A-H22C	108.5	C24-C30A-H30E	109.5
C23A-C22A-H22D	108.5	H30D-C30A-H30E	109.5
C21-C22A-H22D	108.5	C24-C30A-H30F	109.5
H22C-C22A-H22D	107.5	H30D-C30A-H30F	109.5
C22A-C23A-C24	114.1(16)	H30E-C30A-H30F	109.5
C22A-C23A-H23C	108.7	C24-C31A-H31D	109.5
C24-C23A-H23C	108.7	C24-C31A-H31E	109.5
C22A-C23A-H23D	108.7	H31D-C31A-H31E	109.5
C24-C23A-H23D	108.7	C24-C31A-H31F	109.5
H23C-C23A-H23D	107.6	H31D-C31A-H31F	109.5
C30A-C24-C23A	113.4(13)	H31E-C31A-H31F	109.5
C30A-C24-C25	108.2(9)	C6-C33-H33A	109.5
C31-C24-C25	109.7(3)	C6-C33-H33B	109.5
C23A-C24-C25	110.2(8)	H33A-C33-H33B	109.5
C31-C24-C23	110.5(3)	C6-C33-H33C	109.5
C25-C24-C23	110.5(2)	H33A-C33-H33C	109.5
C31-C24-C30	108.9(3)	H33B-C33-H33C	109.5
C25-C24-C30	110.4(3)	C6-C34-H34A	109.5
C23-C24-C30	106.9(3)	C6-C34-H34B	109.5
C30A-C24-C31A	110.6(12)	H34A-C34-H34B	109.5
C23A-C24-C31A	105.2(11)	C6-C34-H34C	109.5
C25-C24-C31A	109.1(8)	H34A-C34-H34C	109.5
N26-C25-C20	120.1(2)	H34B-C34-H34C	109.5
C26-C25-C20	120.1(2)	C9-C35-H35A	109.5
N26-C25-C24	117.6(2)	C9-C35-H35B	109.5
C26-C25-C24	117.6(2)	H35A-C35-H35B	109.5
C20-C25-C24	122.4(2)	C9-C35-H35C	109.5
C27-C26-C25	118.7(2)	H35A-C35-H35C	109.5
C27-C26-H26	120.7	H35B-C35-H35C	109.5
C25-C26-H26	120.7	C9-C36-H36A	109.5
C27-N26-C25	118.7(2)	C9-C36-H36B	109.5
N26-C27-C18	121.1(2)	H36A-C36-H36B	109.5
C26-C27-C18	121.1(2)	C9-C36-H36C	109.5
N26-C27-C15#1	132.3(2)	H36A-C36-H36C	109.5
C26-C27-C15#1	132.3(2)	H36B-C36-H36C	109.5
C18-C27-C15#1	106.6(2)	Cl5-C1S-Cl7	68.5(16)
C21-C28-H28A	109.5	Cl5-C1S-Cl4	50.5(11)
C21-C28-H28B	109.5	Cl7-C1S-Cl4	118.9(14)
H28A-C28-H28B	109.5	Cl5-C1S-Cl8	132.2(13)
C21-C28-H28C	109.5	Cl7-C1S-Cl8	71.5(15)
H28A-C28-H28C	109.5	Cl4-C1S-Cl8	151.7(9)
H28B-C28-H28C	109.5	Cl9-C1S-Cl6	110.9(2)
C21-C29-H29A	109.5	Cl5-C1S-Cl10	130.5(11)
C21-C29-H29B	109.5	Cl7-C1S-Cl10	122.5(14)
H29A-C29-H29B	109.5	Cl4-C1S-Cl10	98.1(9)
C21-C29-H29C	109.5	Cl8-C1S-Cl10	57.7(9)

C19-C1S-C13	114.0(3)	Cl5-Cl7-Cl8	105.2(19)
Cl6-C1S-C13	109.8(3)	Cl10-Cl8-C1S	63.3(10)
Cl5-C1S-C12	92.9(13)	Cl10-Cl8-Cl7	107.3(15)
Cl7-C1S-C12	155.0(14)	C1S-Cl8-Cl7	48.8(10)
Cl4-C1S-C12	44.5(9)	Cl1-Cl10-Cl8	150(2)
Cl8-C1S-C12	114.7(11)	Cl1-Cl10-Cl2	36.5(12)
Cl10-C1S-C12	57.1(11)	Cl8-Cl10-Cl2	121.9(15)
Cl5-C1S-C11	117.8(11)	Cl1-Cl10-C1S	96.2(15)
Cl7-C1S-C11	152.2(11)	Cl8-Cl10-C1S	59.0(9)
Cl4-C1S-C11	72.0(6)	Cl2-Cl10-C1S	62.9(10)
Cl8-C1S-C11	87.9(9)	Cl17-C2S-Cl19	110.8(2)
Cl10-C1S-C11	31.4(8)	Cl17-C2S-Cl13	108.8(2)
Cl2-C1S-C11	27.7(8)	Cl19-C2S-Cl13	109.2(2)
Cl5-C1S-H1S	116(3)	Cl17-C2S-H2S	109.4
Cl7-C1S-H1S	99(3)	Cl19-C2S-H2S	109.4
Cl4-C1S-H1S	109(2)	Cl13-C2S-H2S	109.4
Cl8-C1S-H1S	94(3)	Cl11-C2SA-Cl18	116(3)
Cl9-C1S-H1S	103(2)	Cl11-C2SA-Cl15	88(3)
Cl6-C1S-H1S	114(2)	Cl18-C2SA-Cl15	91(2)
Cl10-C1S-H1S	109(3)	Cl11-C2SA-H2SA	118.1
Cl3-C1S-H1S	106(2)	Cl18-C2SA-H2SA	118.1
Cl2-C1S-H1S	105(2)	Cl15-C2SA-H2SA	118.1
Cl1-C1S-H1S	101(2)	C2SA-Cl11-Cl14	71.6(19)
Cl2-Cl1-Cl10	102(2)	C2SA-Cl11-Cl12	80(2)
Cl2-Cl1-C1S	54.2(13)	Cl14-Cl11-Cl12	45.5(9)
Cl10-Cl1-C1S	52.4(11)	Cl15-Cl12-Cl14	78.4(18)
Cl2-Cl1-Cl4	12.7(13)	Cl15-Cl12-Cl11	91.2(18)
Cl10-Cl1-Cl4	89.2(13)	Cl14-Cl12-Cl11	67.0(12)
C1S-C11-C14	42.4(4)	Cl16-Cl14-Cl12	140.9(17)
Cl1-Cl2-Cl4	157(2)	Cl16-Cl14-Cl15	102.1(14)
Cl1-Cl2-Cl10	41.6(14)	Cl12-Cl14-Cl15	42.2(11)
Cl4-Cl2-Cl10	115.7(15)	Cl16-Cl14-Cl11	130.2(13)
Cl1-Cl2-C1S	98.2(16)	Cl12-Cl14-Cl11	67.4(12)
Cl4-Cl2-C1S	60.7(9)	Cl15-Cl14-Cl11	77.6(11)
Cl10-Cl2-C1S	60.0(9)	Cl16-Cl14-Cl18	48.4(10)
Cl1-Cl2-Cl5	128.5(18)	Cl12-Cl14-Cl18	117.0(11)
Cl4-Cl2-Cl5	28.8(8)	Cl15-Cl14-Cl18	78.4(9)
Cl10-Cl2-Cl5	87.0(12)	Cl11-Cl14-Cl18	84.1(10)
C1S-C12-C15	34.2(6)	Cl12-Cl15-Cl14	59.4(17)
Cl5-C14-C12	121.5(13)	Cl12-Cl15-C2SA	101(2)
Cl5-C14-C1S	50.1(8)	Cl14-Cl15-C2SA	73.6(18)
Cl2-Cl4-C1S	74.8(9)	Cl12-Cl15-Cl20	115(2)
Cl5-C14-Cl1	111.6(11)	Cl14-Cl15-Cl20	74.4(11)
Cl2-Cl4-Cl1	10.0(10)	C2SA-Cl15-Cl20	18.9(14)
C1S-C14-Cl1	65.6(6)	Cl14-Cl16-Cl18	105.0(13)
C1S-C15-C14	79.4(13)	Cl20-Cl18-Cl16	111.4(17)
C1S-C15-C17	61.7(14)	Cl20-Cl18-C2SA	19.4(18)
Cl4-Cl5-Cl7	141.0(19)	Cl16-Cl18-C2SA	93.5(19)
C1S-C15-C12	52.8(9)	Cl20-Cl18-Cl14	86.0(15)
Cl4-Cl5-Cl2	29.6(9)	Cl16-Cl18-Cl14	26.6(6)
Cl7-Cl5-Cl2	112.6(16)	C2SA-Cl18-Cl14	67.4(17)
C1S-C17-Cl5	49.8(11)	Cl18-Cl20-Cl15	95.7(17)
C1S-C17-Cl8	59.6(13)		

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y+1,-z+1

Table SI-9. Anisotropic displacement parameters (\AA^2) for ML026F1_0m. The anisotropic displacement factor exponent takes the form:
 $-2\alpha^2[h^2 a^*{}^2 U^{11} + \dots + 2hka^*b^*U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
N1	0.0131(10)	0.0183(10)	0.0147(10)	0.0016(8)	0.0015(8)	0.0013(8)
C2	0.0179(12)	0.0154(12)	0.0139(12)	-0.0022(9)	-0.0003(9)	0.0006(10)
C3	0.0167(12)	0.0150(12)	0.0149(12)	-0.0002(9)	0.0002(9)	0.0008(10)
C4	0.0167(12)	0.0184(12)	0.0202(12)	-0.0007(10)	0.0023(9)	0.0010(10)
N4	0.0167(12)	0.0184(12)	0.0202(12)	-0.0007(10)	0.0023(9)	0.0010(10)
C5	0.0208(13)	0.0205(13)	0.0199(13)	-0.0022(10)	0.0001(10)	-0.0006(11)
C6	0.0256(14)	0.0201(14)	0.0348(15)	0.0019(12)	0.0036(12)	0.0048(11)
C7	0.0317(16)	0.0253(15)	0.0430(17)	0.0067(13)	-0.0007(13)	0.0037(12)
C8	0.0312(16)	0.0294(16)	0.0407(17)	0.0159(13)	0.0001(13)	0.0002(13)
C9	0.0256(14)	0.0273(15)	0.0277(14)	0.0121(12)	0.0007(11)	0.0007(12)
C10	0.0206(13)	0.0229(13)	0.0170(12)	0.0025(10)	-0.0023(10)	-0.0023(11)
C11	0.0176(12)	0.0224(13)	0.0181(12)	0.0059(10)	0.0013(9)	0.0007(10)
N11	0.0176(12)	0.0224(13)	0.0181(12)	0.0059(10)	0.0013(9)	0.0007(10)
C12	0.0161(12)	0.0179(12)	0.0140(12)	0.0021(9)	-0.0001(9)	0.0012(10)
C13	0.0151(12)	0.0184(12)	0.0128(11)	0.0030(9)	0.0009(9)	-0.0005(10)
N14	0.0148(10)	0.0149(10)	0.0146(10)	0.0012(8)	0.0010(8)	0.0003(8)
N15	0.0151(10)	0.0193(11)	0.0171(10)	0.0047(8)	0.0033(8)	0.0016(8)
C15	0.0126(12)	0.0183(12)	0.0145(12)	-0.0012(10)	0.0004(9)	-0.0008(10)
N16	0.0138(10)	0.0161(10)	0.0156(10)	0.0023(8)	0.0006(8)	0.0005(8)
C17	0.0156(12)	0.0182(13)	0.0145(12)	0.0014(9)	0.0011(9)	-0.0004(10)
C18	0.0141(12)	0.0197(13)	0.0124(11)	-0.0010(9)	0.0011(9)	-0.0004(10)
C19	0.0185(12)	0.0186(12)	0.0160(12)	0.0022(9)	-0.0004(9)	-0.0016(10)
N19	0.0185(12)	0.0186(12)	0.0160(12)	0.0022(9)	-0.0004(9)	-0.0016(10)
C20	0.0158(12)	0.0227(13)	0.0147(12)	-0.0005(10)	-0.0006(10)	-0.0052(10)
C21	0.0219(13)	0.0247(14)	0.0225(13)	0.0026(11)	0.0043(11)	-0.0045(11)
C22	0.030(3)	0.031(2)	0.022(3)	0.005(2)	0.012(2)	-0.0058(18)
C23	0.013(2)	0.024(2)	0.038(3)	-0.0029(19)	0.0085(17)	-0.0014(16)
C22A	0.017(10)	0.084(18)	0.018(11)	0.024(11)	0.009(8)	0.016(10)
C23A	0.029(11)	0.038(12)	0.053(13)	0.002(10)	0.007(9)	-0.002(8)
C24	0.0164(12)	0.0230(13)	0.0178(13)	-0.0017(10)	0.0024(10)	0.0000(10)
C25	0.0148(12)	0.0231(13)	0.0130(11)	-0.0027(10)	-0.0014(9)	-0.0030(10)
C26	0.0145(11)	0.0177(12)	0.0164(11)	-0.0011(9)	0.0010(9)	-0.0007(9)
N26	0.0145(11)	0.0177(12)	0.0164(11)	-0.0011(9)	0.0010(9)	-0.0007(9)
C27	0.0151(12)	0.0192(12)	0.0130(11)	-0.0003(9)	0.0005(9)	-0.0003(10)
C28	0.0424(19)	0.086(3)	0.0302(17)	0.0246(17)	0.0032(14)	-0.0227(19)
C29	0.0400(18)	0.0344(17)	0.051(2)	-0.0028(15)	0.0123(15)	-0.0174(14)
C30	0.020(2)	0.043(3)	0.026(2)	0.007(3)	-0.0024(17)	0.009(3)
C31	0.022(2)	0.023(3)	0.027(3)	-0.004(2)	0.004(2)	-0.0021(19)
C30A	0.051(13)	0.031(12)	0.023(9)	0.015(10)	0.017(8)	0.032(12)
C31A	0.009(8)	0.038(14)	0.047(16)	-0.009(10)	0.007(10)	0.002(8)
C33	0.0272(16)	0.0336(17)	0.053(2)	0.0007(14)	-0.0085(14)	0.0079(13)
C34	0.0494(19)	0.0290(16)	0.050(2)	0.0006(14)	0.0182(16)	0.0109(15)
C35	0.056(2)	0.047(2)	0.0390(19)	0.0223(15)	0.0205(16)	0.0119(17)
C36	0.0308(16)	0.0363(17)	0.056(2)	0.0178(15)	-0.0056(15)	-0.0115(14)
C1S	0.064(2)	0.076(3)	0.0364(19)	-0.0010(17)	-0.0009(17)	0.0133(19)
Cl1	0.076(6)	0.081(7)	0.103(9)	0.023(5)	-0.001(5)	0.029(5)
Cl2	0.067(6)	0.080(6)	0.094(9)	0.007(5)	-0.005(6)	0.020(5)
Cl3	0.084(3)	0.0396(11)	0.0765(17)	0.0016(10)	-0.0378(18)	0.0096(13)
Cl4	0.049(4)	0.098(6)	0.058(4)	-0.012(4)	-0.008(3)	0.023(4)
Cl5	0.064(5)	0.101(7)	0.075(10)	-0.004(6)	-0.002(5)	0.015(5)
Cl6	0.0392(7)	0.0820(10)	0.0634(8)	0.0098(7)	0.0074(6)	-0.0053(7)
Cl7	0.086(10)	0.088(7)	0.057(11)	0.009(7)	0.005(9)	0.005(6)
Cl8	0.136(12)	0.112(8)	0.164(15)	-0.018(8)	0.074(11)	-0.011(7)
Cl9	0.0354(7)	0.154(2)	0.0474(8)	-0.0111(11)	0.0083(5)	-0.0282(11)
Cl10	0.084(8)	0.099(8)	0.112(9)	0.004(6)	0.031(7)	0.008(6)
C2S	0.027(2)	0.052(2)	0.043(2)	0.0160(17)	0.0086(17)	-0.0024(18)

C2SA	0.067(9)	0.091(12)	0.075(11)	-0.003(8)	0.003(7)	0.003(8)
Cl11	0.072(11)	0.119(14)	0.084(8)	-0.015(8)	0.004(7)	0.000(10)
Cl12	0.094(10)	0.103(9)	0.072(10)	-0.015(7)	0.009(8)	0.022(7)
Cl13	0.0700(10)	0.0714(11)	0.0415(7)	-0.0011(7)	0.0193(6)	0.0237(9)
Cl14	0.069(6)	0.107(8)	0.088(7)	-0.021(6)	0.004(5)	0.028(5)
Cl15	0.093(9)	0.112(9)	0.087(10)	-0.009(7)	0.010(7)	0.022(7)
Cl16	0.068(7)	0.091(8)	0.110(9)	-0.026(7)	-0.021(6)	0.030(6)
Cl17	0.0291(5)	0.0373(5)	0.0367(5)	0.0076(4)	0.0024(3)	0.0030(3)
Cl18	0.055(8)	0.083(9)	0.065(9)	0.003(7)	0.005(7)	0.006(7)
Cl19	0.0368(8)	0.0535(8)	0.0407(7)	0.0124(6)	-0.0020(7)	0.0067(7)
Cl20	0.064(8)	0.096(10)	0.084(11)	-0.005(8)	0.004(7)	0.000(7)

Table SI-10. Hydrogen coordinates and isotropic displacement parameters (\AA^2) for ML026F1_0m.

	x	y	z	U(eq)	Occupancy
H1	0.4794	0.5422	0.4702	0.018	0.5
H4	0.2271	0.7051	0.4164	0.022	0.75
H7A	0.2580	0.8864	0.2607	0.040	1
H7B	0.3550	0.8743	0.3457	0.040	1
H1S	0.373(4)	1.0953(19)	0.312(4)	0.102(15)	1
H8A	0.4332	0.8848	0.1799	0.041	1
H8B	0.3602	0.8308	0.1407	0.041	1
H11	0.5995	0.7104	0.2823	0.023	0.75
H16	0.5879	0.5008	0.4690	0.018	0.5
H19	0.8663	0.5968	0.2582	0.021	0.75
H22A	1.2186	0.5484	0.1120	0.033	0.8
H22B	1.1253	0.4993	0.0917	0.033	0.8
H23A	1.2405	0.5145	0.2865	0.030	0.8
H23B	1.2775	0.4670	0.2040	0.030	0.8
H22C	1.2323	0.5470	0.1303	0.047	0.2
H22D	1.2265	0.5440	0.2529	0.047	0.2
H23C	1.2597	0.4592	0.1783	0.048	0.2
H23D	1.1388	0.4633	0.1205	0.048	0.2
H26	0.9532	0.4134	0.3920	0.019	0.75
H28A	1.0514	0.6018	0.0414	0.079	1
H28B	0.9417	0.6131	0.1111	0.079	1
H28C	0.9603	0.5516	0.0589	0.079	1
H29A	1.1697	0.6364	0.1945	0.062	1
H29B	1.1553	0.6048	0.3031	0.062	1
H29C	1.0540	0.6438	0.2577	0.062	1
H30A	1.1319	0.3888	0.3999	0.044	0.8
H30B	1.1820	0.4506	0.4294	0.044	0.8
H30C	1.2559	0.4074	0.3613	0.044	0.8
H31A	1.0472	0.3708	0.2317	0.036	0.8
H31B	1.1716	0.3788	0.1839	0.036	0.8
H31C	1.0655	0.4141	0.1375	0.036	0.8
H30D	1.1548	0.4244	0.4188	0.052	0.2
H30E	1.2052	0.4859	0.3880	0.052	0.2
H30F	1.2664	0.4285	0.3491	0.052	0.2
H31D	1.0643	0.3619	0.2988	0.047	0.2
H31E	1.1654	0.3646	0.2169	0.047	0.2
H31F	1.0392	0.3844	0.1844	0.047	0.2
H33A	0.1170	0.8159	0.2154	0.057	1
H33B	0.1247	0.7510	0.2555	0.057	1
H33C	0.2114	0.7734	0.1700	0.057	1
H34A	0.1391	0.8502	0.4031	0.064	1
H34B	0.2413	0.8240	0.4712	0.064	1
H34C	0.1392	0.7834	0.4325	0.064	1
H35A	0.6271	0.7659	0.1574	0.071	1
H35B	0.5826	0.8220	0.0982	0.071	1
H35C	0.5084	0.7643	0.0952	0.071	1
H36A	0.6457	0.8143	0.3217	0.062	1
H36B	0.5421	0.8502	0.3689	0.062	1
H36C	0.6081	0.8740	0.2704	0.062	1
H2S	0.7907	0.6764	0.4201	0.048	0.9
H2SA	1.0422	0.7411	0.5841	0.093	0.1

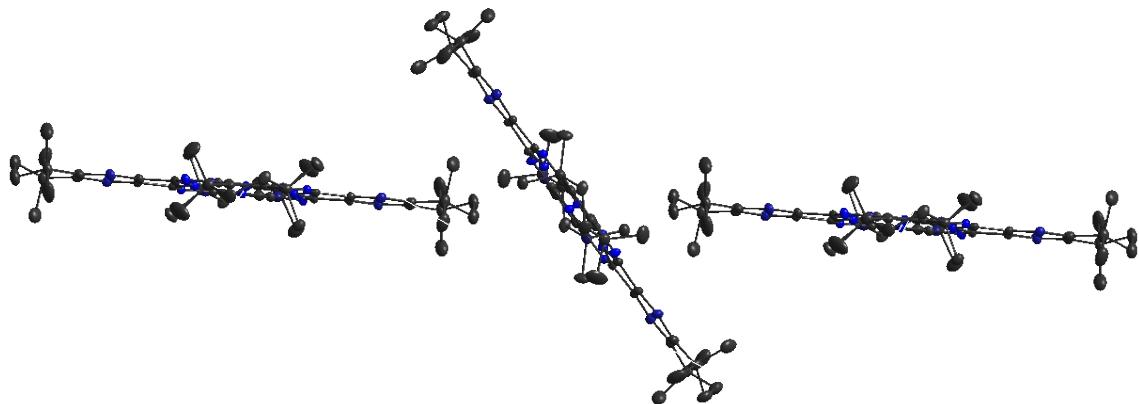
Table SI-11. Torsion angles [°] for ML026F1_0m.

C13-N1-C2-N14	-178.1(2)	C2-C3-C12-C11	179.5(2)
C13-N1-C2-C3	0.4(3)	N4-C3-C12-C13	179.6(2)
N14-C2-C3-N4	-1.0(4)	C4-C3-C12-C13	179.6(2)
N1-C2-C3-N4	-179.5(2)	C2-C3-C12-C13	0.5(2)
N14-C2-C3-C4	-1.0(4)	C2-N1-C13-N15	-178.1(2)
N1-C2-C3-C4	-179.5(2)	C2-N1-C13-C12	-0.1(3)
N14-C2-C3-C12	178.0(2)	N11-C12-C13-N15	-1.0(4)
N1-C2-C3-C12	-0.5(3)	C11-C12-C13-N15	-1.0(4)
C12-C3-C4-C5	0.8(3)	C3-C12-C13-N15	177.9(2)
C2-C3-C4-C5	179.7(2)	N11-C12-C13-N1	-179.2(2)
C12-C3-N4-C5	0.8(3)	C11-C12-C13-N1	-179.2(2)
C2-C3-N4-C5	179.7(2)	C3-C12-C13-N1	-0.3(3)
C3-N4-C5-C10	0.5(3)	N1-C2-N14-C15	0.8(4)
C3-N4-C5-C6	177.4(2)	C3-C2-N14-C15	-177.5(2)
C3-C4-C5-C10	0.5(3)	N1-C13-N15-C17	0.5(4)
C3-C4-C5-C6	177.4(2)	C12-C13-N15-C17	-177.4(2)
N4-C5-C6-C34	44.4(3)	C2-N14-C15-N16#1	-3.8(4)
C4-C5-C6-C34	44.4(3)	C2-N14-C15-C27#1	176.5(2)
C10-C5-C6-C34	-138.8(3)	C13-N15-C17-N16	-3.7(4)
N4-C5-C6-C7	163.7(2)	C13-N15-C17-C18	176.0(2)
C4-C5-C6-C7	163.7(2)	C15#1-N16-C17-N15	-178.2(2)
C10-C5-C6-C7	-19.5(3)	C15#1-N16-C17-C18	2.1(3)
N4-C5-C6-C33	-75.7(3)	N15-C17-C18-N19	-3.4(4)
C4-C5-C6-C33	-75.7(3)	N16-C17-C18-N19	176.3(2)
C10-C5-C6-C33	101.0(3)	N15-C17-C18-C19	-3.4(4)
C34-C6-C7-C8	169.3(3)	N16-C17-C18-C19	176.3(2)
C5-C6-C7-C8	47.8(3)	N15-C17-C18-C27	178.6(2)
C33-C6-C7-C8	-71.7(3)	N16-C17-C18-C27	-1.7(3)
C6-C7-C8-C9	-62.9(3)	C27-C18-C19-C20	0.3(3)
C7-C8-C9-C35	166.0(3)	C17-C18-C19-C20	-177.4(2)
C7-C8-C9-C36	-75.4(3)	C27-C18-N19-C20	0.3(3)
C7-C8-C9-C10	44.1(3)	C17-C18-N19-C20	-177.4(2)
N4-C5-C10-N11	-1.4(4)	C18-N19-C20-C25	-1.7(3)
C6-C5-C10-N11	-178.1(2)	C18-N19-C20-C21	178.0(2)
C4-C5-C10-C11	-1.4(4)	C18-C19-C20-C25	-1.7(3)
C6-C5-C10-C11	-178.1(2)	C18-C19-C20-C21	178.0(2)
N4-C5-C10-C9	-179.1(2)	N19-C20-C21-C28	-50.0(3)
C4-C5-C10-C9	-179.1(2)	C19-C20-C21-C28	-50.0(3)
C6-C5-C10-C9	4.2(4)	C25-C20-C21-C28	129.7(3)
C8-C9-C10-N11	166.5(2)	N19-C20-C21-C29	69.7(3)
C35-C9-C10-N11	46.4(3)	C19-C20-C21-C29	69.7(3)
C36-C9-C10-N11	-72.9(3)	C25-C20-C21-C29	-110.6(3)
C8-C9-C10-C11	166.5(2)	C19-C20-C21-C22	-166.6(3)
C35-C9-C10-C11	46.4(3)	C25-C20-C21-C22	13.1(4)
C36-C9-C10-C11	-72.9(3)	N19-C20-C21-C22A	167.3(9)
C8-C9-C10-C5	-15.7(3)	C25-C20-C21-C22A	-13.0(10)
C35-C9-C10-C5	-135.9(3)	C28-C21-C22-C23	-164.4(3)
C36-C9-C10-C5	104.8(3)	C29-C21-C22-C23	77.3(4)
C5-C10-C11-C12	0.9(4)	C20-C21-C22-C23	-45.0(4)
C9-C10-C11-C12	178.7(2)	C21-C22-C23-C24	65.6(4)
C5-C10-N11-C12	0.9(4)	C28-C21-C22A-C23A	-89.0(17)
C9-C10-N11-C12	178.7(2)	C29-C21-C22A-C23A	155.8(16)
C10-N11-C12-C3	0.5(4)	C20-C21-C22A-C23A	45.2(18)
C10-N11-C12-C13	179.3(2)	C21-C22A-C23A-C24	-65(2)
C10-C11-C12-C3	0.5(4)	C22A-C23A-C24-C30A	-75(2)
C10-C11-C12-C13	179.3(2)	C22A-C23A-C24-C25	46.6(19)
N4-C3-C12-N11	-1.4(4)	C22A-C23A-C24-C31A	164.1(18)
C2-C3-C12-N11	179.5(2)	C22-C23-C24-C31	72.6(4)
C4-C3-C12-C11	-1.4(4)	C22-C23-C24-C25	-48.9(4)

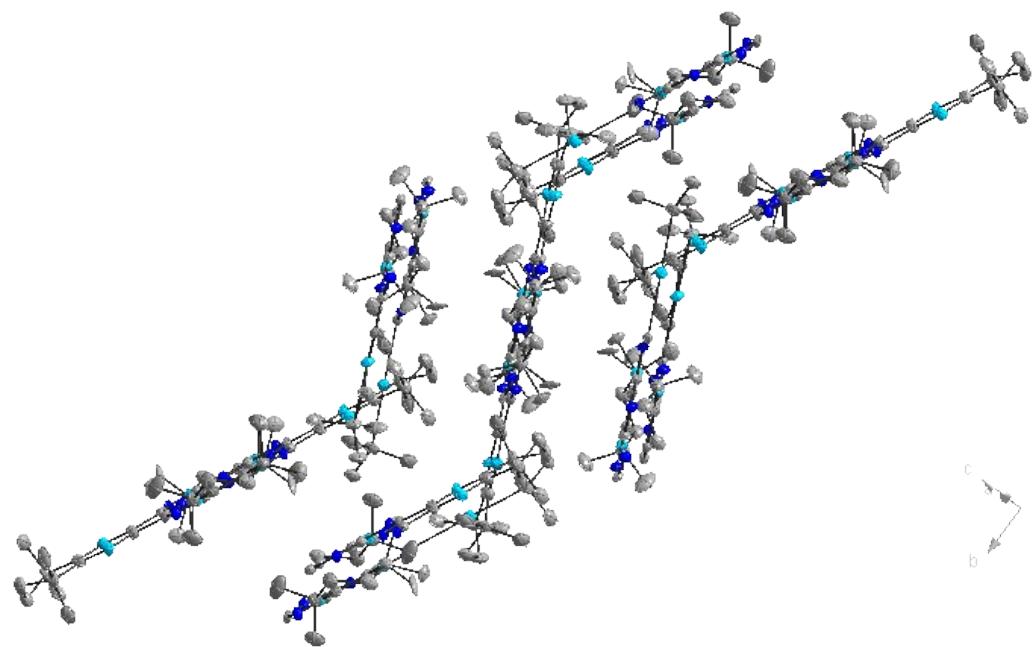
C22-C23-C24-C30	-169.0(4)	Cl8-C1S-Cl2-Cl5	-139.9(13)
N19-C20-C25-N26	1.9(3)	Cl10-C1S-Cl2-Cl5	-137.8(12)
C21-C20-C25-N26	-177.8(2)	Cl1-C1S-Cl2-Cl5	-155.2(16)
C19-C20-C25-C26	1.9(3)	Cl1-Cl2-Cl4-Cl5	-4(6)
C21-C20-C25-C26	-177.8(2)	Cl10-Cl2-Cl4-Cl5	-6(2)
N19-C20-C25-C24	-179.7(2)	C1S-Cl2-Cl4-Cl5	18.9(14)
C19-C20-C25-C24	-179.7(2)	Cl1-Cl2-Cl4-C1S	-23(5)
C21-C20-C25-C24	0.6(3)	Cl10-Cl2-Cl4-C1S	-24.9(13)
C30A-C24-C25-N26	-71.5(9)	Cl5-Cl2-Cl4-C1S	-18.9(14)
C23A-C24-C25-N26	164.0(9)	Cl10-Cl2-Cl4-Cl1	-2(4)
C31A-C24-C25-N26	49.0(9)	C1S-Cl2-Cl4-Cl1	23(5)
C31-C24-C25-C26	73.7(3)	Cl5-Cl2-Cl4-Cl1	4(6)
C23-C24-C25-C26	-164.2(3)	Cl7-C1S-Cl4-Cl5	-2.7(17)
C30-C24-C25-C26	-46.3(3)	Cl8-C1S-Cl4-Cl5	-108(2)
C30A-C24-C25-C20	110.0(9)	Cl10-C1S-Cl4-Cl5	-137.0(14)
C31-C24-C25-C20	-104.8(3)	Cl2-C1S-Cl4-Cl5	-158.9(16)
C23A-C24-C25-C20	-14.4(9)	Cl1-C1S-Cl4-Cl5	-154.6(12)
C23-C24-C25-C20	17.3(3)	Cl5-C1S-Cl4-Cl2	158.9(16)
C30-C24-C25-C20	135.2(3)	Cl7-C1S-Cl4-Cl2	156.2(15)
C31A-C24-C25-C20	-129.5(9)	Cl8-C1S-Cl4-Cl2	51(2)
C20-C25-C26-C27	-0.6(3)	Cl10-C1S-Cl4-Cl2	21.8(13)
C24-C25-C26-C27	-179.1(2)	Cl1-C1S-Cl4-Cl2	4.2(10)
C20-C25-N26-C27	-0.6(3)	Cl5-C1S-Cl4-Cl1	154.6(12)
C24-C25-N26-C27	-179.1(2)	Cl7-C1S-Cl4-Cl1	152.0(13)
C25-N26-C27-C18	-0.8(3)	Cl8-C1S-Cl4-Cl1	47(2)
C25-N26-C27-C15#1	177.3(2)	Cl10-C1S-Cl4-Cl1	17.6(9)
C25-C26-C27-C18	-0.8(3)	Cl2-C1S-Cl4-Cl1	-4.2(10)
C25-C26-C27-C15#1	177.3(2)	Cl7-C1S-Cl5-Cl4	177.5(16)
N19-C18-C27-N26	0.9(4)	Cl8-C1S-Cl5-Cl4	142.4(14)
C17-C18-C27-N26	179.1(2)	Cl10-C1S-Cl5-Cl4	62.4(18)
C19-C18-C27-C26	0.9(4)	Cl2-C1S-Cl5-Cl4	14.7(11)
C17-C18-C27-C26	179.1(2)	Cl1-C1S-Cl5-Cl4	27.4(13)
N19-C18-C27-C15#1	-177.6(2)	Cl4-C1S-Cl5-Cl7	-177.5(16)
C19-C18-C27-C15#1	-177.6(2)	Cl8-C1S-Cl5-Cl7	-35.0(18)
C17-C18-C27-C15#1	0.6(2)	Cl10-C1S-Cl5-Cl7	-115.0(18)
Cl10-Cl1-Cl2-Cl4	-3(6)	Cl2-C1S-Cl5-Cl7	-162.8(13)
C1S-Cl1-Cl2-Cl4	20(4)	Cl1-C1S-Cl5-Cl7	-150.1(12)
C1S-Cl1-Cl2-Cl10	22.8(17)	Cl7-C1S-Cl5-Cl2	162.8(13)
Cl4-Cl1-Cl2-Cl10	3(6)	Cl4-C1S-Cl5-Cl2	-14.7(11)
Cl10-Cl1-Cl2-C1S	-22.8(17)	Cl8-C1S-Cl5-Cl2	127.8(15)
Cl4-Cl1-Cl2-C1S	-20(4)	Cl10-C1S-Cl5-Cl2	47.8(15)
Cl10-Cl1-Cl2-Cl5	-5(2)	Cl1-C1S-Cl5-Cl2	12.7(9)
C1S-Cl1-Cl2-Cl5	17.6(10)	Cl2-Cl4-Cl5-C1S	-24.1(18)
Cl4-Cl1-Cl2-Cl5	-2(4)	Cl1-Cl4-Cl5-C1S	-24.8(11)
Cl5-C1S-Cl2-Cl1	155.2(16)	Cl2-Cl4-Cl5-Cl7	-21(4)
Cl7-C1S-Cl2-Cl1	115(3)	C1S-Cl4-Cl5-Cl7	4(2)
Cl4-C1S-Cl2-Cl1	171(2)	Cl1-Cl4-Cl5-Cl7	-21(3)
Cl8-C1S-Cl2-Cl1	15.3(19)	C1S-Cl4-Cl5-Cl2	24.1(18)
Cl10-C1S-Cl2-Cl1	17.3(14)	Cl1-Cl4-Cl5-Cl2	-0.7(11)
Cl5-C1S-Cl2-Cl4	-16.2(12)	Cl4-C1S-Cl7-Cl5	2.2(14)
Cl7-C1S-Cl2-Cl4	-57(3)	Cl8-C1S-Cl7-Cl5	153.4(14)
Cl8-C1S-Cl2-Cl4	-156.1(13)	Cl10-C1S-Cl7-Cl5	125.1(14)
Cl10-C1S-Cl2-Cl4	-154.0(15)	Cl2-C1S-Cl7-Cl5	44(3)
Cl11-C1S-Cl2-Cl4	-171(2)	Cl1-C1S-Cl7-Cl5	109(3)
Cl5-C1S-Cl2-Cl10	137.8(12)	Cl5-C1S-Cl7-Cl8	-153.4(14)
Cl7-C1S-Cl2-Cl10	97(3)	Cl4-C1S-Cl7-Cl8	-151.2(11)
Cl4-C1S-Cl2-Cl10	154.0(15)	Cl10-C1S-Cl7-Cl8	-28.2(17)
Cl8-C1S-Cl2-Cl10	-2.1(15)	Cl2-C1S-Cl7-Cl8	-109(3)
Cl11-C1S-Cl2-Cl10	-17.3(14)	Cl1-C1S-Cl7-Cl8	-44(3)
Cl7-C1S-Cl2-Cl5	-41(3)	Cl4-Cl5-Cl7-C1S	-4(2)
Cl4-C1S-Cl2-Cl5	16.2(12)	Cl2-Cl5-Cl7-C1S	-14.8(11)

C1S-C15-C17-C18	23.6(13)	Cl15-C112-C14-C118	26.1(18)
Cl4-C15-C17-C18	20(3)	Cl11-C112-C14-C118	-70.3(12)
Cl2-C15-C17-C18	9(2)	Cl11-C112-C15-C114	66.3(11)
Cl5-C1S-C18-C110	-117.6(17)	Cl14-C112-C15-C2SA	-64(2)
Cl7-C1S-C18-C110	-151.8(17)	Cl11-C112-C15-C2SA	3(2)
Cl4-C1S-C18-C110	-35(3)	Cl14-C112-C15-C120	-51.3(17)
Cl2-C1S-C18-C110	2.1(15)	Cl11-C112-C15-C120	15(2)
Cl1-C1S-C18-C110	9.1(11)	Cl12-C114-C116-C118	83(2)
Cl5-C1S-C18-C17	34.3(19)	Cl15-C114-C116-C118	62.3(15)
Cl4-C1S-C18-C17	117(2)	Cl11-C114-C116-C118	-22(3)
Cl10-C1S-C18-C17	151.8(17)	Cl16-C118-C120-C115	-27.6(18)
Cl2-C1S-C18-C17	153.9(14)	C2SA-C118-C120-C115	-52(6)
Cl1-C1S-C18-C17	160.9(13)	Cl14-C118-C120-C115	-35.9(11)
C1S-C17-C18-C110	26.2(16)		
Cl5-C17-C18-C110	5(2)		
Cl5-C17-C18-C1S	-20.8(11)		
Cl2-C11-C110-C18	55(4)		
C1S-C11-C110-C18	31(3)		
Cl4-C11-C110-C18	54(4)		
C1S-C11-C110-C12	-23.4(18)		
Cl4-C11-C110-C12	-0.6(13)		
Cl2-C11-C110-C1S	23.4(17)		
Cl4-C11-C110-C1S	22.8(9)		
C1S-C18-C110-C11	-37(4)		
Cl7-C18-C110-C11	-59(5)		
C1S-C18-C110-C12	-2.3(17)		
Cl7-C18-C110-C12	-24(3)		
Cl7-C18-C110-C1S	-21.8(13)		
Cl4-C12-C110-C11	179(2)		
C1S-C12-C110-C11	154(2)		
Cl5-C12-C110-C11	175.9(19)		
Cl1-C12-C110-C18	-151(3)		
Cl4-C12-C110-C18	27(3)		
C1S-C12-C110-C18	2.2(17)		
Cl5-C12-C110-C18	24(2)		
Cl1-C12-C110-C1S	-154(2)		
Cl4-C12-C110-C1S	25.1(14)		
Cl5-C12-C110-C1S	22.2(6)		
Cl5-C1S-C110-C11	-77(2)		
Cl7-C1S-C110-C11	-165.6(18)		
Cl4-C1S-C110-C11	-33.5(15)		
C18-C1S-C110-C11	162(2)		
Cl2-C1S-C110-C11	-15.4(12)		
Cl5-C1S-C110-C18	120.4(18)		
Cl7-C1S-C110-C18	32(2)		
Cl4-C1S-C110-C18	164.1(13)		
Cl2-C1S-C110-C18	-177.8(16)		
Cl1-C1S-C110-C18	-162(2)		
Cl5-C1S-C110-C12	-61.8(18)		
Cl7-C1S-C110-C12	-150.2(17)		
Cl4-C1S-C110-C12	-18.1(10)		
C18-C1S-C110-C12	177.8(16)		
Cl1-C1S-C110-C12	15.4(12)		
Cl18-C2SA-C111-C114	-42(3)		
Cl115-C2SA-C111-C114	48.2(15)		
Cl118-C2SA-C111-C112	-89(3)		
Cl115-C2SA-C111-C112	1.9(17)		
Cl115-C112-C114-C116	-30(3)		
Cl111-C112-C114-C116	-127(2)		
Cl111-C112-C114-C115	-96.4(16)		
Cl115-C112-C114-C111	96.4(16)		

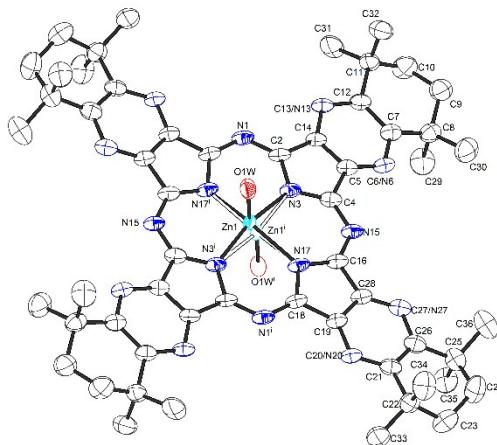
c
b



c
b



1.7.3 Crystal Structure of A₂B₂N₄-[Pc^{*}Zn·H₂O]



Habitus, colour	prism, dark
Crystal size	0.27 x 0.06 x 0.03 mm ³
Crystal system	Monoclinic
Space group	P2 ₁ /c
Unit cell dimensions	$a = 6.0419(5)$ Å $\alpha = 90^\circ$. $b = 16.2175(13)$ Å $\beta = 95.305(3)^\circ$. $c = 31.008(2)$ Å $\gamma = 90^\circ$.
Volume	3025.3(4) Å ³
Cell determination	9556 peaks with Theta 2.3 to 25.2°.
Empirical formula	C ₆₀ H ₇₀ N ₁₂ O ₂ Zn
Moiety formula	C ₆₀ H ₇₀ N ₁₂ O ₂ Zn
Formula weight	1040.65
Density (calculated)	1.142 Mg/m ³
Absorption coefficient	0.454 mm ⁻¹
F(000)	1104
Solution and refinement:	
Reflections collected	32818
Independent reflections	5598 [R(int) = 0.0711]
Completeness to theta = 25.242°	99.9 %
Observed reflections	4276[I > 2(I)]
Reflections used for refinement	5598
Absorption correction	Numerical Mu Calculated ^[12]
Max. and min. transmission	0.99 and 0.89
Largest diff. peak and hole	0.288 and -0.347 e.Å ⁻³
Solution	Direct methods
Refinement	Full-matrix least-squares on F ²
Treatment of hydrogen atoms	Calculated positions, constr. ref.
Programs used	XT V2014/1 (Bruker AXS Inc., 2014) ^[13] SHELXL-2014/7 (Sheldrick, 2014) ^[14] DIAMOND (Crystal Impact) ^[11] ShelXle (Hübschle, Sheldrick, Dittrich, 2011) ^[15]
Data / restraints / parameters	5598 / 0 / 353
Goodness-of-fit on F ²	1.021
R index (all data)	wR2 = 0.1467
R index conventional [I>2sigma(I)]	R1 = 0.0580

Table SI-12. Atomic coordinates and equivalent isotropic displacement parameters (\AA^2) for mL105c2_0m_sq. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)	Occupancy
Zn1	1.49975(18)	0.51807(4)	0.50622(3)	0.0242(2)	0.5
N1	1.5282(5)	0.39227(14)	0.59501(8)	0.0354(6)	1
C2	1.3557(6)	0.37918(17)	0.56599(9)	0.0336(7)	1
N3	1.3123(5)	0.41696(14)	0.52645(8)	0.0330(6)	1
C4	1.1271(6)	0.38191(18)	0.50470(10)	0.0357(8)	1
C5	1.0459(5)	0.31799(17)	0.53226(9)	0.0324(7)	1
C6	0.8816(5)	0.26266(17)	0.52335(9)	0.0369(7)	0.5
N6	0.8816(5)	0.26266(17)	0.52335(9)	0.0369(7)	0.5
C7	0.8475(6)	0.20533(19)	0.55472(10)	0.0378(8)	1
C8	0.6829(6)	0.1355(2)	0.54251(11)	0.0459(9)	1
C9	0.6011(7)	0.0982(3)	0.58286(13)	0.0594(11)	1
C10	0.7865(8)	0.0790(2)	0.61729(13)	0.0609(11)	1
C11	0.9179(6)	0.1542(2)	0.63326(11)	0.0446(9)	1
C12	0.9761(6)	0.20836(19)	0.59537(10)	0.0362(8)	1
C13	1.1473(5)	0.26387(16)	0.60328(9)	0.0362(7)	0.5
N13	1.1473(5)	0.26387(16)	0.60328(9)	0.0362(7)	0.5
C14	1.1822(5)	0.31757(16)	0.57052(9)	0.0306(7)	1
N15	1.0358(5)	0.39883(15)	0.46520(8)	0.0350(6)	1
C16	1.1103(6)	0.45652(18)	0.43907(9)	0.0345(7)	1
N17	1.2882(5)	0.50688(15)	0.44710(8)	0.0348(6)	1
C18	1.3091(6)	0.55471(17)	0.41116(9)	0.0335(7)	1
C19	1.1260(6)	0.53481(17)	0.37847(9)	0.0339(7)	1
C20	1.0650(5)	0.56503(17)	0.33820(9)	0.0357(7)	0.5
N20	1.0650(5)	0.56503(17)	0.33820(9)	0.0357(7)	0.5
C21	0.8899(6)	0.52815(19)	0.31471(10)	0.0392(8)	1
C22	0.8002(6)	0.5685(2)	0.27201(11)	0.0448(8)	1
C23	0.6709(8)	0.5058(3)	0.24279(13)	0.0640(11)	1
C24	0.5072(8)	0.4553(3)	0.26589(15)	0.0672(12)	1
C25	0.6171(6)	0.4051(2)	0.30292(12)	0.0492(9)	1
C26	0.7813(6)	0.4578(2)	0.33098(11)	0.0417(8)	1
C27	0.8404(5)	0.43133(18)	0.37254(9)	0.0404(7)	0.5
N27	0.8404(5)	0.43133(18)	0.37254(9)	0.0404(7)	0.5
C28	1.0081(6)	0.47124(18)	0.39565(10)	0.0347(7)	1
C29	0.8078(7)	0.0725(2)	0.51705(15)	0.0657(12)	1
C30	0.4821(7)	0.1666(3)	0.51310(14)	0.0660(12)	1
C31	1.1256(7)	0.1267(3)	0.66111(13)	0.0644(12)	1
C32	0.7765(7)	0.2076(3)	0.66089(12)	0.0589(11)	1
C33	0.9850(7)	0.6044(3)	0.24761(13)	0.0650(12)	1
C34	0.6459(7)	0.6390(2)	0.28378(15)	0.0665(12)	1
C35	0.7495(7)	0.3332(2)	0.28583(14)	0.0601(11)	1
C36	0.4342(8)	0.3688(3)	0.32869(17)	0.0783(14)	1
O1W	1.3888(7)	0.6241(3)	0.53911(14)	0.0403(11)	0.5

Table SI-13. Bond lengths [Å] and angles [°] for mL105c2_0m_sq.

Zn1-Zn1#1	0.7019(11)	C20-H20	0.9400
Zn1-N17#1	1.886(3)	C21-C26	1.431(5)
Zn1-N3#1	1.908(3)	C21-C22	1.531(5)
Zn1-N3	2.121(3)	C22-C33	1.521(5)
Zn1-O1W	2.138(4)	C22-C23	1.528(5)
Zn1-N17	2.143(3)	C22-C34	1.540(5)
N1-C2	1.330(4)	C23-C24	1.514(6)
N1-C18#1	1.333(4)	C23-H23A	0.9800
C2-N3	1.374(4)	C23-H23B	0.9800
C2-C14	1.464(4)	C24-C25	1.510(6)
N3-C4	1.375(4)	C24-H24A	0.9800
N3-Zn1#1	1.908(3)	C24-H24B	0.9800
C4-N15	1.325(4)	C25-C26	1.521(5)
C4-C5	1.457(4)	C25-C35	1.537(5)
C5-C6	1.348(4)	C25-C36	1.540(6)
C5-C14	1.380(4)	C26-C27	1.373(5)
C6-C7	1.375(4)	C27-C28	1.351(4)
C6-H6	0.9400	C27-H27	0.9400
C7-C12	1.419(5)	C29-H29A	0.9700
C7-C8	1.531(5)	C29-H29B	0.9700
C8-C9	1.513(5)	C29-H29C	0.9700
C8-C29	1.531(5)	C30-H30A	0.9700
C8-C30	1.534(5)	C30-H30B	0.9700
C9-C10	1.507(6)	C30-H30C	0.9700
C9-H9A	0.9800	C31-H31A	0.9700
C9-H9B	0.9800	C31-H31B	0.9700
C10-C11	1.514(5)	C31-H31C	0.9700
C10-H10A	0.9800	C32-H32A	0.9700
C10-H10B	0.9800	C32-H32B	0.9700
C11-C31	1.523(5)	C32-H32C	0.9700
C11-C32	1.533(5)	C33-H33A	0.9700
C11-C12	1.533(4)	C33-H33B	0.9700
C12-C13	1.376(4)	C33-H33C	0.9700
C13-C14	1.369(4)	C34-H34A	0.9700
C13-H13	0.9400	C34-H34B	0.9700
N15-C16	1.343(4)	C34-H34C	0.9700
C16-N17	1.355(4)	C35-H35A	0.9700
C16-C28	1.448(4)	C35-H35B	0.9700
N17-C18	1.373(4)	C35-H35C	0.9700
N17-Zn1#1	1.886(3)	C36-H36A	0.9700
C18-N1#1	1.333(4)	C36-H36B	0.9700
C18-C19	1.465(5)	C36-H36C	0.9700
C19-C20	1.360(4)	O1W-H1WA	0.8643
C19-C28	1.387(4)	O1W-H1WB	0.8644
C20-C21	1.366(5)		
Zn1#1-Zn1-N17#1	101.9(2)	N3-Zn1-N17	83.97(10)
Zn1#1-Zn1-N3#1	97.8(2)	O1W-Zn1-N17	106.71(14)
N17#1-Zn1-N3#1	97.51(12)	C2-N1-C18#1	123.2(2)
Zn1#1-Zn1-N3	63.01(17)	N1-C2-N3	127.3(3)
N17#1-Zn1-N3	87.16(11)	N1-C2-C14	124.6(3)
N3#1-Zn1-N3	160.86(4)	N3-C2-C14	108.1(3)
Zn1#1-Zn1-O1W	162.0(2)	C2-N3-C4	109.4(3)
N17#1-Zn1-O1W	91.56(14)	C2-N3-Zn1#1	130.3(2)
N3#1-Zn1-O1W	92.15(14)	C4-N3-Zn1#1	117.73(19)
N3-Zn1-O1W	106.33(14)	C2-N3-Zn1	123.1(2)
Zn1#1-Zn1-N17	59.45(17)	C4-N3-Zn1	127.40(19)
N17#1-Zn1-N17	161.30(4)	Zn1#1-N3-Zn1	19.14(4)
N3#1-Zn1-N17	85.97(11)	N15-C4-N3	128.2(3)

N15-C4-C5	123.8(3)	C19-C20-H20	121.2
N3-C4-C5	108.0(3)	C21-C20-H20	121.2
C6-C5-C14	122.9(3)	C20-C21-C26	121.3(3)
C6-C5-C4	129.4(3)	C20-C21-C22	118.0(3)
C14-C5-C4	107.5(3)	C26-C21-C22	120.6(3)
C5-C6-C7	117.4(3)	C33-C22-C23	108.5(3)
C5-C6-H6	121.3	C33-C22-C21	112.1(3)
C7-C6-H6	121.3	C23-C22-C21	110.4(3)
C6-C7-C12	120.0(3)	C33-C22-C34	108.8(3)
C6-C7-C8	117.5(3)	C23-C22-C34	110.3(3)
C12-C7-C8	122.3(3)	C21-C22-C34	106.6(3)
C9-C8-C29	111.9(3)	C24-C23-C22	113.6(3)
C9-C8-C7	110.2(3)	C24-C23-H23A	108.9
C29-C8-C7	106.3(3)	C22-C23-H23A	108.9
C9-C8-C30	108.8(3)	C24-C23-H23B	108.9
C29-C8-C30	108.2(3)	C22-C23-H23B	108.9
C7-C8-C30	111.3(3)	H23A-C23-H23B	107.7
C10-C9-C8	112.9(3)	C25-C24-C23	113.0(4)
C10-C9-H9A	109.0	C25-C24-H24A	109.0
C8-C9-H9A	109.0	C23-C24-H24A	109.0
C10-C9-H9B	109.0	C25-C24-H24B	109.0
C8-C9-H9B	109.0	C23-C24-H24B	109.0
H9A-C9-H9B	107.8	H24A-C24-H24B	107.8
C9-C10-C11	113.5(3)	C24-C25-C26	110.5(3)
C9-C10-H10A	108.9	C24-C25-C35	110.7(3)
C11-C10-H10A	108.9	C26-C25-C35	106.9(3)
C9-C10-H10B	108.9	C24-C25-C36	108.4(4)
C11-C10-H10B	108.9	C26-C25-C36	112.3(3)
H10A-C10-H10B	107.7	C35-C25-C36	108.1(3)
C10-C11-C31	109.2(3)	C27-C26-C21	119.5(3)
C10-C11-C32	109.6(3)	C27-C26-C25	117.4(3)
C31-C11-C32	108.7(3)	C21-C26-C25	122.8(3)
C10-C11-C12	111.2(3)	C28-C27-C26	117.7(3)
C31-C11-C12	111.5(3)	C28-C27-H27	121.1
C32-C11-C12	106.5(3)	C26-C27-H27	121.1
C13-C12-C7	121.5(3)	C27-C28-C19	122.6(3)
C13-C12-C11	117.4(3)	C27-C28-C16	130.4(3)
C7-C12-C11	121.0(3)	C19-C28-C16	106.8(3)
C14-C13-C12	116.6(3)	C8-C29-H29A	109.5
C14-C13-H13	121.7	C8-C29-H29B	109.5
C12-C13-H13	121.7	H29A-C29-H29B	109.5
C13-C14-C5	121.4(3)	C8-C29-H29C	109.5
C13-C14-C2	131.8(3)	H29A-C29-H29C	109.5
C5-C14-C2	106.8(2)	H29B-C29-H29C	109.5
C4-N15-C16	124.6(3)	C8-C30-H30A	109.5
N15-C16-N17	127.9(3)	C8-C30-H30B	109.5
N15-C16-C28	122.6(3)	H30A-C30-H30B	109.5
N17-C16-C28	109.5(3)	C8-C30-H30C	109.5
C16-N17-C18	108.8(3)	H30A-C30-H30C	109.5
C16-N17-Zn1#1	118.9(2)	H30B-C30-H30C	109.5
C18-N17-Zn1#1	129.6(2)	C11-C31-H31A	109.5
C16-N17-Zn1	127.5(2)	C11-C31-H31B	109.5
C18-N17-Zn1	123.7(2)	H31A-C31-H31B	109.5
Zn1#1-N17-Zn1	18.69(4)	C11-C31-H31C	109.5
N1#1-C18-N17	127.2(3)	H31A-C31-H31C	109.5
N1#1-C18-C19	124.2(3)	H31B-C31-H31C	109.5
N17-C18-C19	108.5(3)	C11-C32-H32A	109.5
C20-C19-C28	121.0(3)	C11-C32-H32B	109.5
C20-C19-C18	132.7(3)	H32A-C32-H32B	109.5
C28-C19-C18	106.2(3)	C11-C32-H32C	109.5
C19-C20-C21	117.5(3)	H32A-C32-H32C	109.5

H32B-C32-H32C	109.5	C25-C35-H35B	109.5
C22-C33-H33A	109.5	H35A-C35-H35B	109.5
C22-C33-H33B	109.5	C25-C35-H35C	109.5
H33A-C33-H33B	109.5	H35A-C35-H35C	109.5
C22-C33-H33C	109.5	H35B-C35-H35C	109.5
H33A-C33-H33C	109.5	C25-C36-H36A	109.5
H33B-C33-H33C	109.5	C25-C36-H36B	109.5
C22-C34-H34A	109.5	H36A-C36-H36B	109.5
C22-C34-H34B	109.5	C25-C36-H36C	109.5
H34A-C34-H34B	109.5	H36A-C36-H36C	109.5
C22-C34-H34C	109.5	H36B-C36-H36C	109.5
H34A-C34-H34C	109.5	Zn1-O1W-H1WA	109.6
H34B-C34-H34C	109.5	Zn1-O1W-H1WB	109.6
C25-C35-H35A	109.5	H1WA-O1W-H1WB	109.2

Symmetry transformations used to generate equivalent atoms:

#1 -x+3,-y+1,-z+1

Table SI-14. Anisotropic displacement parameters (\AA^2) for mL105c2_0m_sq. The anisotropic displacement factor exponent takes the form:
 $-2\alpha^2[h^2 a^*{}^2 U^{11} + \dots + 2 h k a^* b^* U^{12}]$

	U ¹¹	U ²²	U ³³	U ²³	U ¹³	U ¹²
Zn1	0.0235(3)	0.0261(6)	0.0219(6)	0.0052(4)	-0.0039(4)	-0.0076(5)
N1	0.0602(19)	0.0240(13)	0.0228(14)	0.0029(10)	0.0085(13)	0.0018(13)
C2	0.058(2)	0.0241(15)	0.0201(16)	-0.0001(12)	0.0107(15)	0.0024(15)
N3	0.0509(17)	0.0247(12)	0.0239(13)	0.0044(10)	0.0071(12)	0.0008(12)
C4	0.053(2)	0.0245(15)	0.0316(18)	0.0031(13)	0.0125(15)	0.0041(15)
C5	0.048(2)	0.0214(15)	0.0295(17)	0.0023(12)	0.0100(14)	0.0044(14)
C6	0.0522(19)	0.0354(15)	0.0234(14)	0.0011(12)	0.0041(13)	0.0099(15)
N6	0.0522(19)	0.0354(15)	0.0234(14)	0.0011(12)	0.0041(13)	0.0099(15)
C7	0.046(2)	0.0348(17)	0.0340(18)	0.0005(14)	0.0079(15)	0.0119(15)
C8	0.049(2)	0.049(2)	0.0392(19)	-0.0031(16)	0.0024(16)	0.0106(17)
C9	0.065(3)	0.060(2)	0.053(2)	-0.0006(19)	0.007(2)	-0.012(2)
C10	0.079(3)	0.048(2)	0.054(2)	0.0135(18)	0.000(2)	-0.007(2)
C11	0.045(2)	0.052(2)	0.0362(19)	0.0154(16)	0.0047(16)	-0.0002(17)
C12	0.046(2)	0.0317(16)	0.0312(18)	0.0041(13)	0.0082(15)	0.0086(15)
C13	0.0548(19)	0.0303(15)	0.0248(15)	0.0056(12)	0.0104(13)	0.0044(14)
N13	0.0548(19)	0.0303(15)	0.0248(15)	0.0056(12)	0.0104(13)	0.0044(14)
C14	0.049(2)	0.0201(14)	0.0241(16)	0.0008(12)	0.0118(14)	0.0019(13)
N15	0.0520(18)	0.0293(13)	0.0243(14)	0.0046(11)	0.0072(12)	0.0039(12)
C16	0.055(2)	0.0250(15)	0.0249(16)	0.0013(12)	0.0090(15)	0.0038(15)
N17	0.0569(18)	0.0264(13)	0.0212(13)	0.0039(10)	0.0037(12)	0.0007(13)
C18	0.056(2)	0.0241(15)	0.0205(15)	0.0010(12)	0.0042(14)	0.0065(15)
C19	0.052(2)	0.0241(15)	0.0260(16)	-0.0009(12)	0.0054(14)	0.0025(14)
C20	0.0510(19)	0.0328(15)	0.0237(15)	0.0030(12)	0.0052(13)	0.0021(14)
N20	0.0510(19)	0.0328(15)	0.0237(15)	0.0030(12)	0.0052(13)	0.0021(14)
C21	0.048(2)	0.0360(18)	0.0354(18)	-0.0011(14)	0.0137(16)	0.0086(16)
C22	0.043(2)	0.048(2)	0.042(2)	0.0088(16)	-0.0002(16)	0.0076(17)
C23	0.080(3)	0.062(3)	0.047(2)	0.0016(19)	-0.009(2)	0.003(2)
C24	0.063(3)	0.058(2)	0.077(3)	0.000(2)	-0.014(2)	-0.010(2)
C25	0.047(2)	0.0400(19)	0.062(2)	-0.0024(17)	0.0103(18)	-0.0003(17)
C26	0.049(2)	0.0364(18)	0.042(2)	-0.0015(15)	0.0149(16)	0.0072(16)
C27	0.055(2)	0.0325(15)	0.0356(17)	0.0016(13)	0.0148(15)	0.0020(14)
N27	0.055(2)	0.0325(15)	0.0356(17)	0.0016(13)	0.0148(15)	0.0020(14)
C28	0.051(2)	0.0276(15)	0.0260(16)	0.0002(13)	0.0055(14)	0.0036(15)
C29	0.063(3)	0.052(2)	0.083(3)	-0.023(2)	0.011(2)	0.011(2)
C30	0.056(3)	0.078(3)	0.063(3)	-0.005(2)	-0.002(2)	0.015(2)
C31	0.057(3)	0.076(3)	0.061(3)	0.043(2)	0.007(2)	0.000(2)
C32	0.061(3)	0.071(3)	0.047(2)	0.002(2)	0.0188(19)	0.000(2)
C33	0.058(3)	0.092(3)	0.044(2)	0.029(2)	-0.0033(19)	0.003(2)
C34	0.058(3)	0.051(2)	0.089(3)	0.003(2)	-0.005(2)	0.016(2)
C35	0.064(3)	0.046(2)	0.070(3)	-0.0147(19)	0.008(2)	0.003(2)
C36	0.055(3)	0.081(3)	0.102(4)	0.003(3)	0.024(3)	-0.007(2)
O1W	0.027(2)	0.046(3)	0.047(3)	-0.007(2)	0.003(2)	-0.001(2)

Table SI-15. Hydrogen coordinates and isotropic displacement parameters (\AA^2) for mL105c2_0m_sq.

	x	y	z	U(eq)	Occupancy
H6	0.7937	0.2632	0.4967	0.044	0.5
H9A	0.4972	0.1366	0.5947	0.071	1
H9B	0.5197	0.0473	0.5750	0.071	1
H10A	0.8878	0.0393	0.6056	0.073	1
H10B	0.7234	0.0529	0.6419	0.073	1
H13	1.2358	0.2649	0.6298	0.043	0.5
H20	1.1406	0.6097	0.3269	0.043	0.5
H23A	0.5900	0.5351	0.2186	0.077	1
H23B	0.7766	0.4684	0.2307	0.077	1
H24A	0.3981	0.4925	0.2771	0.081	1
H24B	0.4270	0.4181	0.2450	0.081	1
H27	0.7666	0.3870	0.3845	0.049	0.5
H29A	0.7083	0.0278	0.5076	0.099	1
H29B	0.9329	0.0508	0.5355	0.099	1
H29C	0.8614	0.0991	0.4920	0.099	1
H30A	0.3726	0.1229	0.5088	0.099	1
H30B	0.5300	0.1829	0.4853	0.099	1
H30C	0.4169	0.2136	0.5265	0.099	1
H31A	1.2093	0.1748	0.6717	0.097	1
H31B	1.2170	0.0931	0.6439	0.097	1
H31C	1.0829	0.0947	0.6854	0.097	1
H32A	0.8638	0.2541	0.6725	0.088	1
H32B	0.7291	0.1749	0.6846	0.088	1
H32C	0.6470	0.2276	0.6431	0.088	1
H33A	0.9209	0.6314	0.2215	0.098	1
H33B	1.0830	0.5604	0.2399	0.098	1
H33C	1.0691	0.6441	0.2658	0.098	1
H34A	0.5716	0.6623	0.2575	0.100	1
H34B	0.7330	0.6814	0.2995	0.100	1
H34C	0.5359	0.6176	0.3018	0.100	1
H35A	0.8144	0.3006	0.3100	0.090	1
H35B	0.8668	0.3547	0.2697	0.090	1
H35C	0.6510	0.2988	0.2670	0.090	1
H36A	0.5014	0.3338	0.3517	0.118	1
H36B	0.3323	0.3364	0.3095	0.118	1
H36C	0.3536	0.4132	0.3412	0.118	1
H1WA	1.4889	0.6385	0.5594	0.060	0.5
H1WB	1.2668	0.6126	0.5504	0.060	0.5

Table SI-16. Torsion angles [°] for mL105c2_0m_sq.

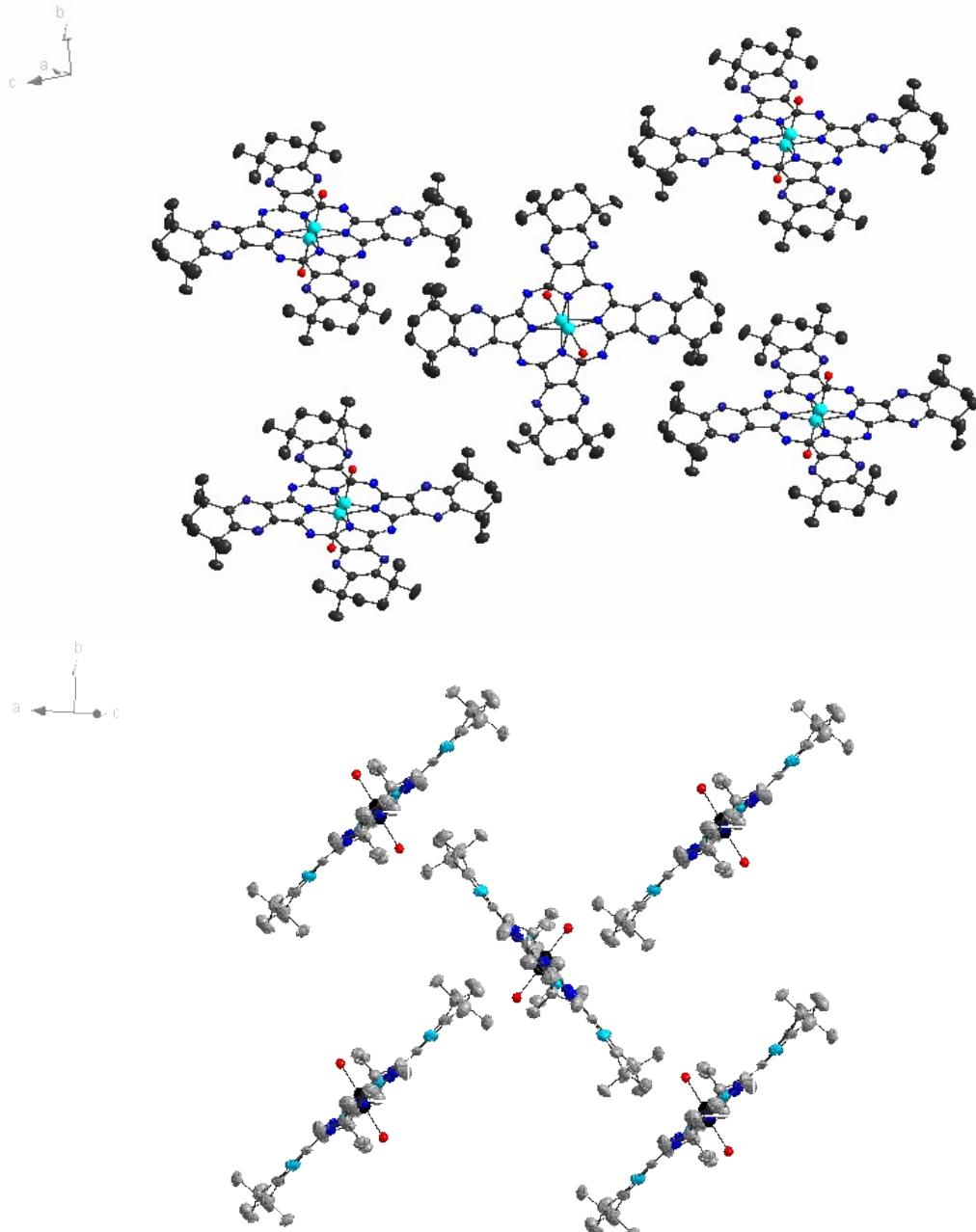
C18#1-N1-C2-N3	-1.2(5)	C4-N15-C16-N17	0.7(5)
C18#1-N1-C2-C14	175.7(3)	C4-N15-C16-C28	178.3(3)
N1-C2-N3-C4	175.4(3)	N15-C16-N17-C18	178.1(3)
C14-C2-N3-C4	-2.0(3)	C28-C16-N17-C18	0.2(3)
N1-C2-N3-Zn1#1	14.4(5)	N15-C16-N17-Zn1#1	15.1(4)
C14-C2-N3-Zn1#1	-163.0(2)	C28-C16-N17-Zn1#1	-162.8(2)
N1-C2-N3-Zn1	-7.8(4)	N15-C16-N17-Zn1	-4.8(5)
C14-C2-N3-Zn1	174.78(18)	C28-C16-N17-Zn1	177.3(2)
C2-N3-C4-N15	-177.5(3)	C16-N17-C18-N1#1	-174.9(3)
Zn1#1-N3-C4-N15	-13.8(4)	Zn1#1-N17-C18-N1#1	-14.3(4)
Zn1-N3-C4-N15	5.9(5)	Zn1-N17-C18-N1#1	7.9(4)
C2-N3-C4-C5	0.4(3)	C16-N17-C18-C19	1.9(3)
Zn1#1-N3-C4-C5	164.15(19)	Zn1#1-N17-C18-C19	162.5(2)
Zn1-N3-C4-C5	-176.16(19)	Zn1-N17-C18-C19	-175.29(19)
N15-C4-C5-C6	4.4(5)	N1#1-C18-C19-C20	-5.1(5)
N3-C4-C5-C6	-173.6(3)	N17-C18-C19-C20	177.9(3)
N15-C4-C5-C14	179.4(3)	N1#1-C18-C19-C28	173.5(3)
N3-C4-C5-C14	1.4(3)	N17-C18-C19-C28	-3.4(3)
C14-C5-C6-C7	1.9(4)	C28-C19-C20-C21	-2.8(4)
C4-C5-C6-C7	176.1(3)	C18-C19-C20-C21	175.7(3)
C5-C6-C7-C12	2.9(4)	C19-C20-C21-C26	-3.0(5)
C5-C6-C7-C8	-171.9(3)	C19-C20-C21-C22	172.1(3)
C6-C7-C8-C9	-159.1(3)	C20-C21-C22-C33	37.8(4)
C12-C7-C8-C9	26.2(4)	C26-C21-C22-C33	-147.2(3)
C6-C7-C8-C29	79.4(4)	C20-C21-C22-C23	159.0(3)
C12-C7-C8-C29	-95.3(4)	C26-C21-C22-C23	-26.0(4)
C6-C7-C8-C30	-38.3(4)	C20-C21-C22-C34	-81.2(4)
C12-C7-C8-C30	147.0(3)	C26-C21-C22-C34	93.9(4)
C29-C8-C9-C10	70.3(4)	C33-C22-C23-C24	170.3(3)
C7-C8-C9-C10	-47.8(4)	C21-C22-C23-C24	47.0(4)
C30-C8-C9-C10	-170.1(3)	C34-C22-C23-C24	-70.6(4)
C8-C9-C10-C11	61.1(5)	C22-C23-C24-C25	-60.7(5)
C9-C10-C11-C31	-169.2(3)	C23-C24-C25-C26	46.8(5)
C9-C10-C11-C32	71.8(4)	C23-C24-C25-C35	-71.4(4)
C9-C10-C11-C12	-45.7(5)	C23-C24-C25-C36	170.3(4)
C6-C7-C12-C13	-5.0(5)	C20-C21-C26-C27	6.1(5)
C8-C7-C12-C13	169.6(3)	C22-C21-C26-C27	-168.8(3)
C6-C7-C12-C11	170.3(3)	C20-C21-C26-C25	-167.9(3)
C8-C7-C12-C11	-15.1(5)	C22-C21-C26-C25	17.2(5)
C10-C11-C12-C13	-160.9(3)	C24-C25-C26-C27	159.3(3)
C31-C11-C12-C13	-38.7(4)	C35-C25-C26-C27	-80.1(4)
C32-C11-C12-C13	79.8(4)	C36-C25-C26-C27	38.2(4)
C10-C11-C12-C7	23.6(4)	C24-C25-C26-C21	-26.5(5)
C31-C11-C12-C7	145.8(3)	C35-C25-C26-C21	94.0(4)
C32-C11-C12-C7	-95.7(4)	C36-C25-C26-C21	-147.7(3)
C7-C12-C13-C14	2.1(4)	C21-C26-C27-C28	-3.2(5)
C11-C12-C13-C14	-173.4(3)	C25-C26-C27-C28	171.1(3)
C12-C13-C14-C5	2.7(4)	C26-C27-C28-C19	-2.5(5)
C12-C13-C14-C2	-174.4(3)	C26-C27-C28-C16	-178.1(3)
C6-C5-C14-C13	-4.8(5)	C20-C19-C28-C27	5.8(5)
C4-C5-C14-C13	179.8(3)	C18-C19-C28-C27	-173.1(3)
C6-C5-C14-C2	172.9(3)	C20-C19-C28-C16	-177.7(3)
C4-C5-C14-C2	-2.5(3)	C18-C19-C28-C16	3.4(3)
N1-C2-C14-C13	2.8(5)	N15-C16-C28-C27	-4.3(5)
N3-C2-C14-C13	-179.8(3)	N17-C16-C28-C27	173.8(3)
N1-C2-C14-C5	-174.6(3)	N15-C16-C28-C19	179.6(3)
N3-C2-C14-C5	2.8(3)	N17-C16-C28-C19	-2.4(3)
N3-C4-N15-C16	-1.3(5)		
C5-C4-N15-C16	-178.9(3)		

Table SI-17. Hydrogen bonds for mL105c2_0m_sq [Å and °].

D-H...A	d(D-H)	d(H...A)	d(D...A)	∠(DHA)
O1W-H1WB...N15#2	0.86	1.86	2.583(5)	140.6
O1W-H1WB...N27#2	0.86	2.63	3.307(5)	135.7

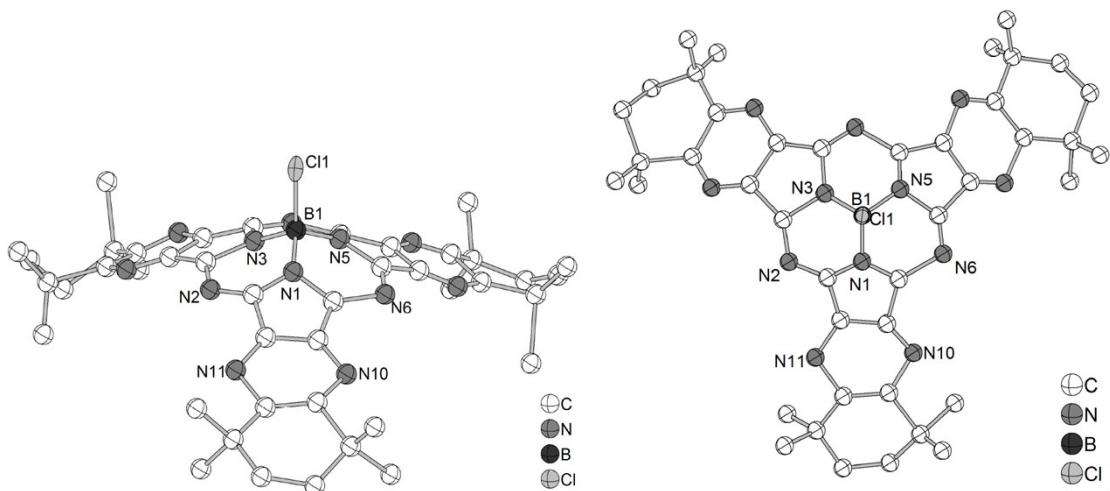
Symmetry transformations used to generate equivalent atoms:

#1 -x+3,-y+1,-z+1 #2 -x+2,-y+1,-z+1



1.7.4 Crystal Structure of [Sppz^{*}BCl]

Crystallographer: Dr. Benjamin Oelkers



Empirical formula C49 H56 B Cl N12

Formula weight 859.32

Temperature 100(2) K

Wavelength 0.71073 Å

Crystal system Monoclinic

Space group P 21/c

Unit cell dimensions $a = 7.253(3)$ Å $\alpha = 90^\circ$.

$b = 25.008(5)$ Å $\beta = 98.83(3)^\circ$.

$c = 25.698(7)$ Å $\gamma = 90^\circ$.

Volume 4606(2) Å³

Z 4

Density (calculated) 1.239 Mg/m³

Absorption coefficient 0.132 mm⁻¹

F(000) 1824

Crystal size 0.3 x 0.06 x 0.03 mm³

Theta range for data collection 1.14 to 20.00°.

Index ranges -6<=h<=6, -24<=k<=24, -24<=l<=24

Reflections collected 13040

Independent reflections 4273 [R(int) = 0.2894]

Completeness to theta = 20.00° 99.8 %

Absorption correction None

Refinement method Full-matrix least-squares on F²

Data / restraints / parameters 4273 / 0 / 197

Goodness-of-fit on F² 0.841

Final R indices [I>2sigma(I)] R1 = 0.1654, wR2 = 0.3401

R indices (all data) R1 = 0.4135, wR2 = 0.4094

Largest diff. peak and hole 0.382 and -0.431 e.Å⁻³

Note: Because of the poor diffraction of the crystal, no satisfying refinement could be obtained. To gain a preliminary crystal structure of low resolution only reflexes up to theta = 20° were used. All deflection coefficients C, N, and B were held constant isotropic and refined with common parameters. The interpretation and orientation of the solvent molecules are, in contrast to the general figure of the complex molecule, not of sufficient quality for a discussion. Because of the poor data, no detailed discussion of bond lengths is possible however a documentation of the unit cell is possible.

1.8 References

- [1] S. A. Mikhalenko, L. I. Solov'eva, E. A. Luk'yanetz, *The Journal of General Chemistry of the USSR* **1991**, *61*, 996–1003.
- [2] P. Jones, G. B. Villeneuve, C. Fei, J. DeMarte, A. J. Haggarty, K. T. Nwe, D. A. Martin, A. M. Lebuis, J. M. Finkelstein, B. J. Gour-Salin et al., *J. Med. Chem.* **1998**, *41*, 3062–3077.
- [3] E. Seikel, B. Oelkers, O. Burghaus, J. Sundermeyer, *Inorg. Chem.* **2013**, *52*, 4451–4457.
- [4] E. Seikel, *Dissertation*, Philipps-Universität Marburg, Marburg, **2012**.
- [5] M. Liebold, *Masterarbeit*, Philipps-Universität Marburg, Marburg, **2012**.
- [6] APEX2, Bruker AXS Inc., Madison, Wisconsin, USA, **2014**.
- [7] SAINT, Bruker AXS Inc., Madison, Wisconsin, USA, **2013**.
- [8] G. M. Sheldrick, *Acta crystallographica. Section A, Foundations of crystallography* **2008**, *64*, 112–122.
- [9] G. M. Sheldrick, *SHELXT*, Universität Göttingen, Göttingen, Germany, **2014**.
- [10] G. M. Sheldrick, *SHELXL*, Universität Göttingen, Göttingen, Germany, **2014**.
- [11] K. Brandenburg, *Diamond - Crystal and Molecular Structure Visualization*, Crystal Impact - Dr. H. Putz & Dr. K. Brandenburg GbR, Bonn, Germany, **2014**.
- [12] SADABS, Bruker AXS Inc., Madison, Wisconsin, USA, **2014**.
- [13] G. M. Sheldrick, *Acta Crystallogr A Found Adv* **2015**, *71*, 3–8.
- [14] G. M. Sheldrick, *Acta crystallographica. Section C, Structural chemistry* **2015**, *71*, 3–8.
- [15] C. B. Hübschle, G. M. Sheldrick, B. Dittrich, *Journal of applied crystallography* **2011**, *44*, 1281–1284.