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Supporting information

Synthesis and Properties of Dithienylethene-Functionalized Switchable Antibacterial Agents

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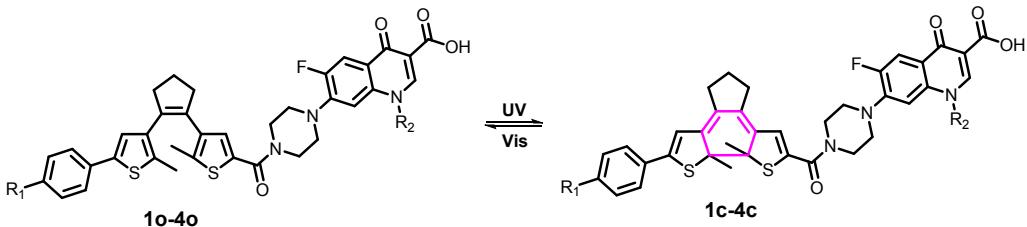
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Scheme S1 Photoisomerization of switchable antibacterial agents **1-4**.

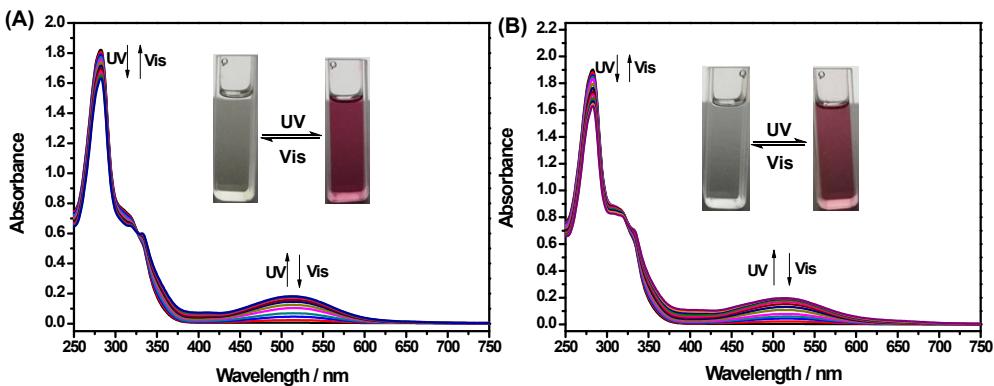


Fig. S1 Absorption spectral changes of dithienylethenes **1b** and **3b** with 254 nm UV and > 402 nm visible light irradiation in DMSO (2.0×10^{-5} mol/L) at 298 K. (Inset) Corresponding color changes of dithienylethene **1b** and **3b** in DMSO upon photoirradiation. (A) absorption spectral changes for **1b** [0 s, 5 s, 10 s, 15 s, 25 s, 35 s, 45 s, 55 s, 70 s, 85 s, 100 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 60 s, 80 s, 100 s, 130 s, 160 s, 190 s (PSS-open)]; (B) absorption spectral changes for **3b** [0 s, 5 s, 10 s, 15 s, 25 s, 35 s, 45 s, 55 s, 60 s, 75 s, 90 s, 105 s, 120 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 50 s, 60 s, 80 s, 100 s, 130 s, 160 s, 190 s, 230 s, 260 s (PSS-open)].

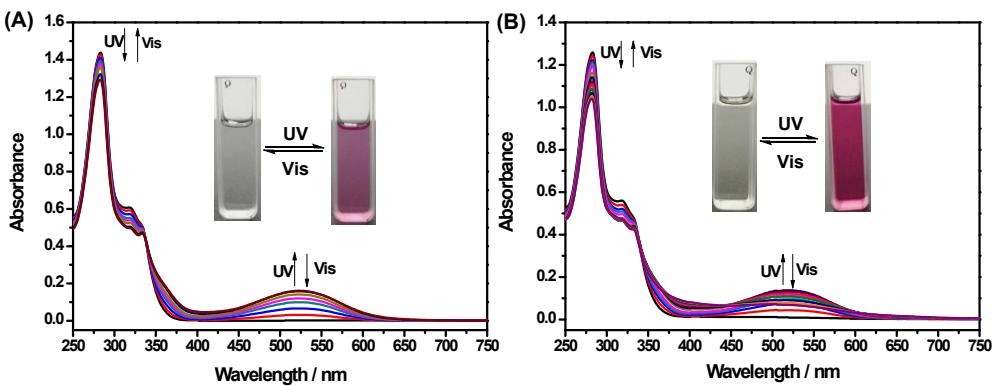


Fig. S2 Absorption spectral changes of dithienylethenes **2a** and **2b** with 254 nm UV and > 402 nm visible light irradiation in DMSO (2.0×10^{-5} mol/L) at 298 K. (Inset) Corresponding color changes of dithienylethene **2a** and **2b** in DMSO upon photoirradiation. (A) absorption spectral changes for **2a** [0 s, 5 s, 10 s, 15 s, 20 s, 30 s, 45 s, 60 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 60 s, 80 s, 110 s (PSS-open)]; (B) absorption spectral changes for **2b** [0 s, 5 s, 10 s, 15 s, 20 s, 30 s, 40 s, 50 s, 60 s, 75 s, 95 s, 115 s, 135 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 60 s, 80 s, 100 s, 130 s, 160 s, 200 s, 240 s, 280 s (PSS-open)].

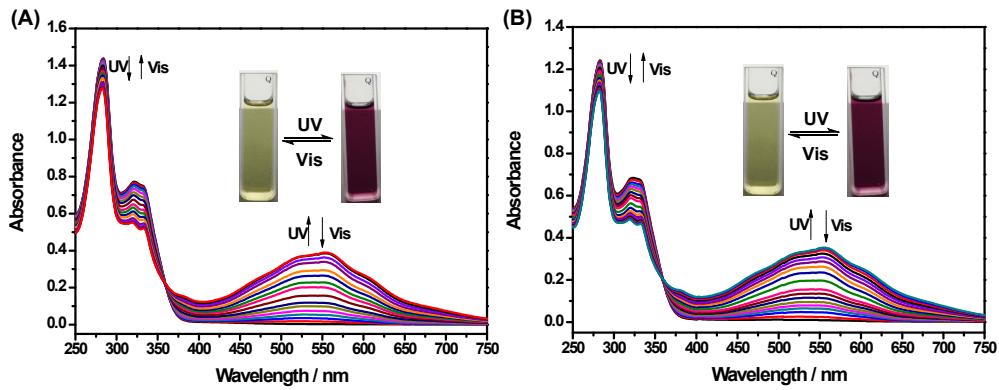


Fig. S3 Absorption spectral changes of dithienylethenes **4a** and **4b** with 254 nm UV and > 402 nm visible light irradiation in DMSO (2.0×10^{-5} mol/L) at 298 K. (Inset) Corresponding color changes of dithienylethene **4a** and **4b** in DMSO upon photoirradiation. (A) absorption spectral changes for **4a** [0 s, 3 s, 6 s, 9 s, 12 s, 15 s, 20 s, 25 s, 30 s, 40 s, 50 s, 60 s, 70 s, 80 s, 90 s, 100 s, 115 s, 125 s, 135 s (PSS-closed); 0 s, 5 s, 10 s, 15 s, 20 s, 25 s, 30 s, 40 s, 60 s, 80 s, 100 s, 200 s, 220 s, 240 s, 260 s, 280 s, 330 s (PSS-open)]; (B) absorption spectral changes for **4b** [0 s, 3 s, 6 s, 9 s, 12 s, 15 s, 20 s, 25 s, 30 s, 40 s, 50 s, 60 s, 70 s, 80 s, 90 s, 100 s, 110 s, 120 s (PSS-closed); 0 s, 5 s, 10 s, 15 s, 20 s, 25 s, 30 s, 35 s, 40 s, 50 s, 60 s, 80 s, 110 s, 140 s, 170 s, 200 s, 230 s, 270 s (PSS-open)].

Table S1 Absorption characteristics and photochromic quantum yields of **1-4** in DMSO (2.0×10^{-5} mol/L)

Compounds	$\lambda_{\max}^{\text{Abs}} / \text{nm}^{\text{a}}$	$\lambda_{\max}^{\text{Abs}} / \text{nm}^{\text{b}}$	Φ^{c}	
	($\epsilon \times 10^4$)	($\epsilon \times 10^4$)	(Open)	(PSS)
1a	283 (1.08)	510 (0.11)	0.21 (510)	0.0014 (283)
1b	281 (0.92)	510 (0.10)	0.18 (510)	0.0011 (281)
2a	283 (0.72)	522 (0.09)	0.24 (522)	0.0040 (283)
2b	283 (0.63)	520 (0.07)	0.20 (520)	0.0034 (283)
3a	282 (0.66)	511 (0.07)	0.25 (511)	0.0076 (282)
3b	283 (0.95)	510 (0.10)	0.16 (542)	0.0055 (296)
4a	283 (0.73)	542 (0.20)	0.33 (542)	0.0042 (283)
4b	282 (0.62)	543 (0.18)	0.31 (543)	0.0038 (282)

a Absorption maxima of open-ring isomers.

b Absorption maxima of closed-ring isomers.

c Quantum yields of open-ring ($\phi_{\text{o-c}}$) and closed-ring isomers ($\phi_{\text{c-o}}$), respectively.

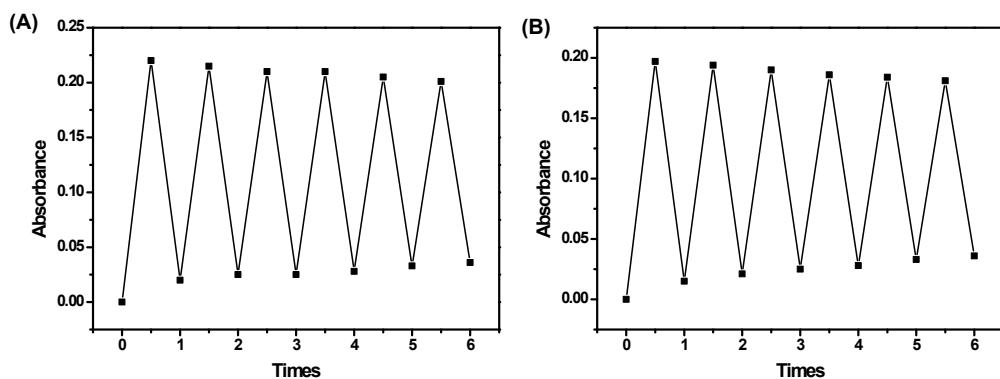


Fig. S4 Fatigue resistance of dithienylethenes **1a** (A) at 510 nm and **1b** (B) at 510 nm on alternate excitation at 254 nm and > 402 nm Vis light irradiation over six cycles in DMSO (2.0×10^{-5} mol/L) at 298 K.

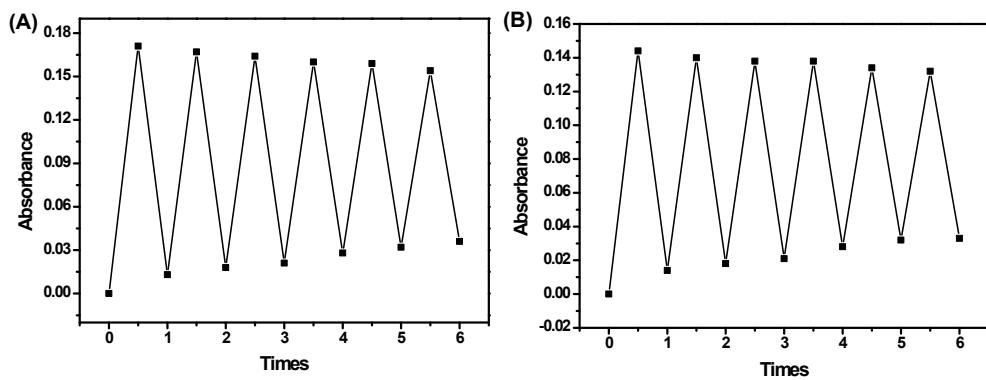


Fig. S5 Fatigue resistance of dithienylethenes **2a** (A) at 522 nm and **2b** (B) at 520 nm on alternate excitation at 254 nm and > 402 nm Vis light irradiation over six cycles in DMSO (2.0×10^{-5} mol/L) at 298 K.

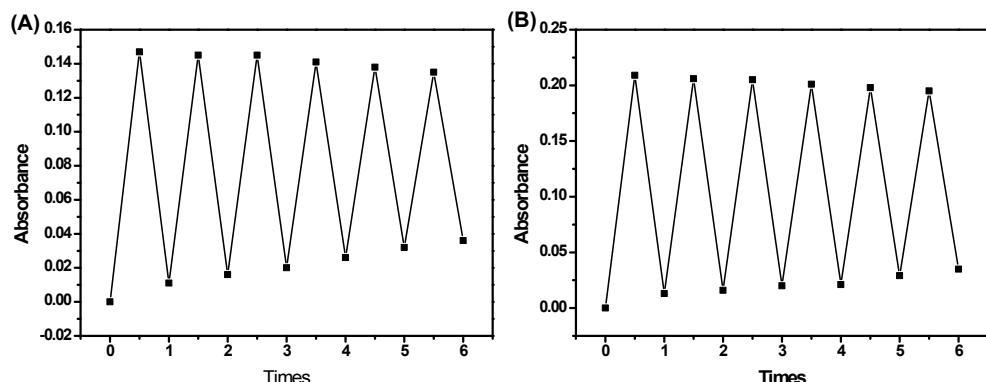


Fig. S6 Fatigue resistance of dithienylethenes **3a** (A) at 511 nm and **3b** (B) at 510 nm on alternate excitation at 254 nm and > 402 nm Vis light irradiation over six cycles in DMSO (2.0×10^{-5} mol/L) at 298 K.

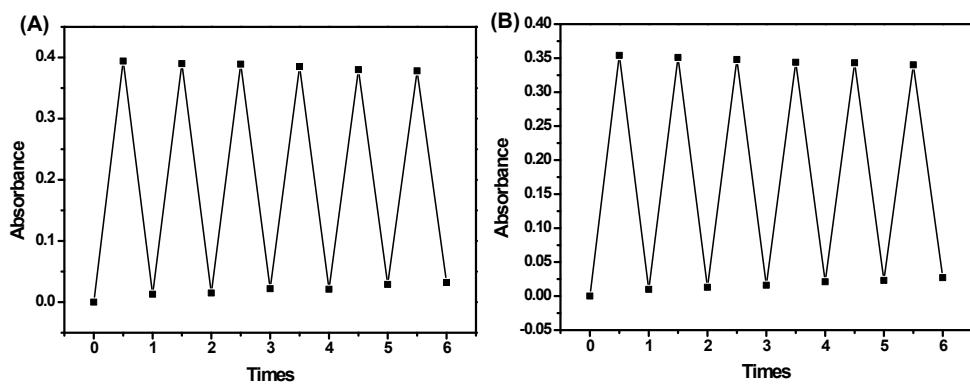


Fig. S7 Fatigue resistance of dithienylethenes **4a** (A) at 542 nm and **4b** (B) at 543 nm on alternate excitation at 254 nm and > 402 nm Vis light irradiation over six cycles in DMSO (2.0×10^{-5} mol/L) at 298 K.

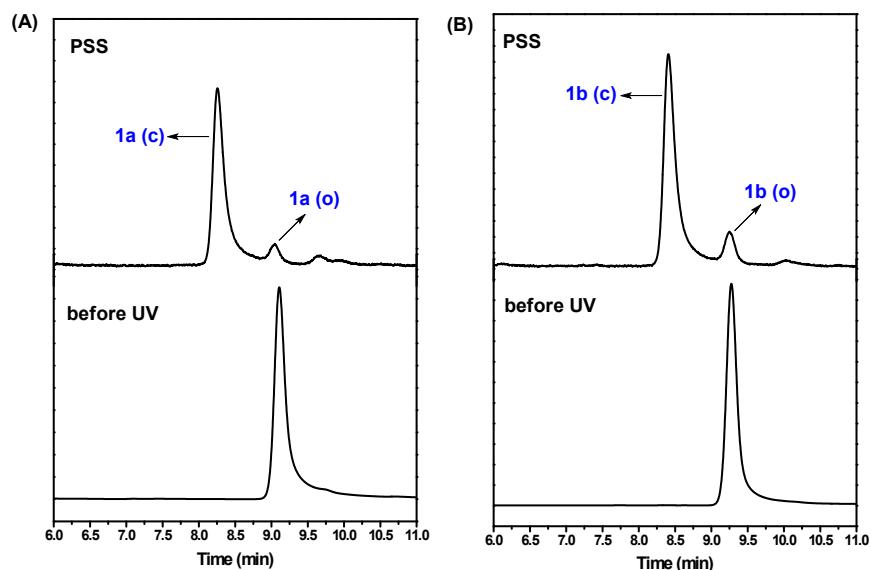


Fig. S8 HPLC traces of **1** before UV irradiation and in the PSS state detected at the isosbestic point of 326 nm ($\text{CH}_3\text{CN} / \text{H}_2\text{O} = 80 / 20$, v / v) at a flow rate of 1.0 mL min^{-1} , (A) HPLC for **1a**; (B) HPLC for **1b**.

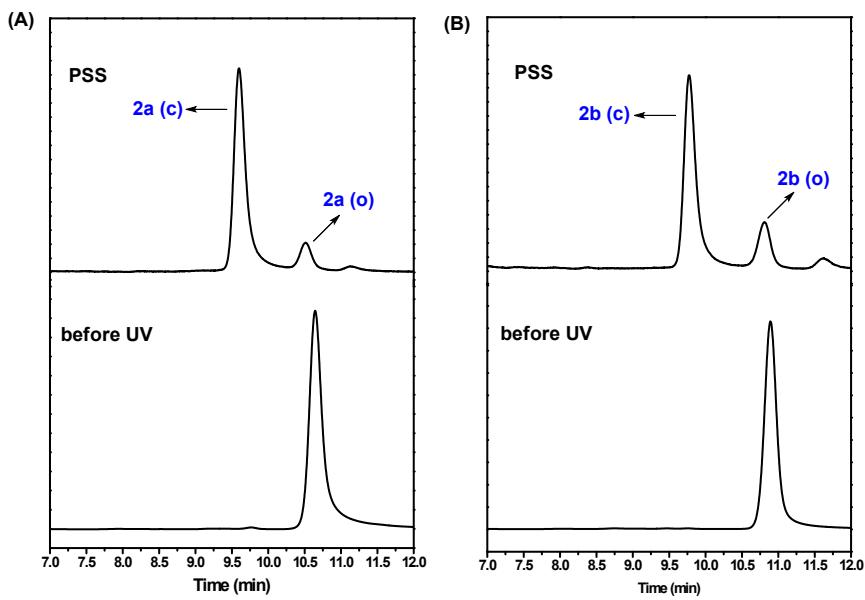


Fig. S9 HPLC traces of **2** before UV irradiation and in the PSS state detected at the isosbestic point of 338 nm ($\text{CH}_3\text{CN} / \text{H}_2\text{O} = 80 / 20, \text{v/v}$) at a flow rate of 1.0 mL min^{-1} , (A) HPLC for **2a**; (B) HPLC for **2b**.

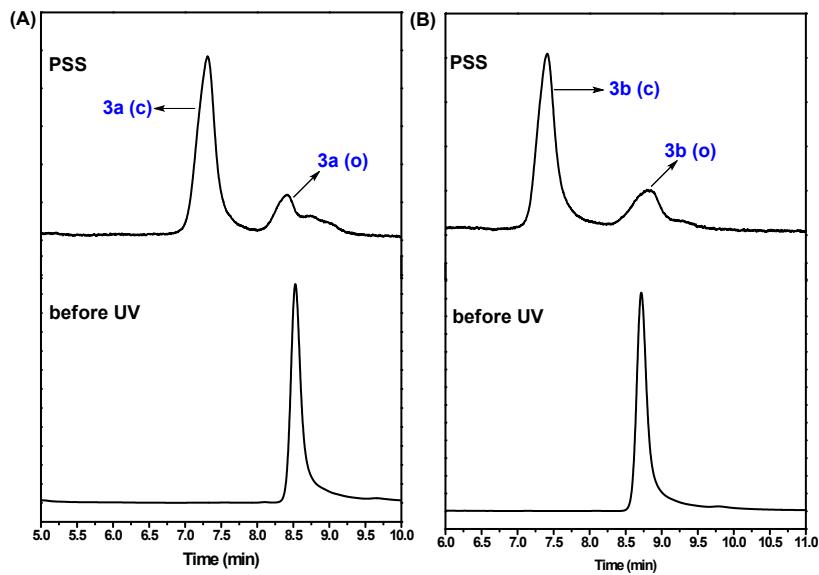


Fig. S10 HPLC traces of **3** before UV irradiation and in the PSS state detected at the isosbestic point of 324 nm ($\text{CH}_3\text{CN} / \text{H}_2\text{O} = 80 / 20, \text{v/v}$) at a flow rate of 1.0 mL min^{-1} , (A) HPLC for **3a**; (B) HPLC for **3b**.

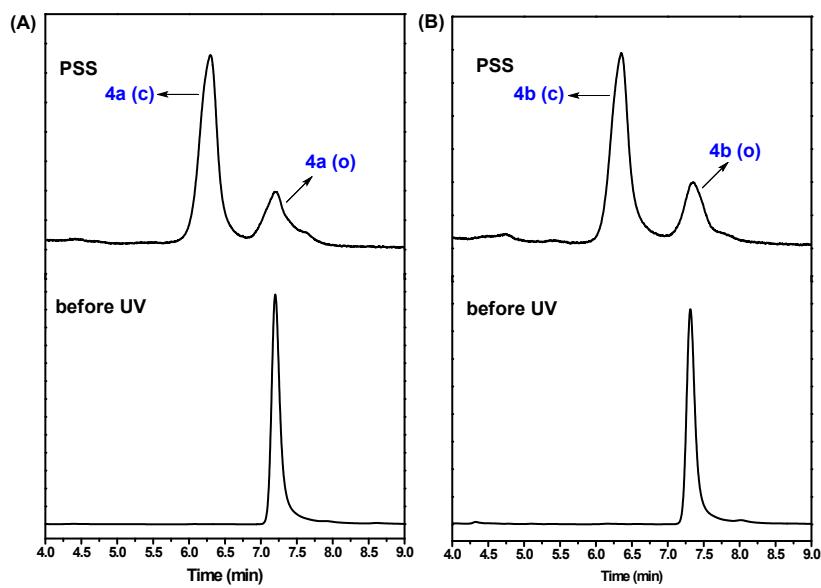


Fig. S11 HPLC traces of **4** before UV irradiation and in the PSS state detected at the isosbestic point of 359 nm ($\text{CH}_3\text{CN} / \text{H}_2\text{O} = 80 / 20, \text{v} / \text{v}$) at a flow rate of 1.0 mL min^{-1} , (A) HPLC for **4a**; (B) HPLC for **4b**.

Table S2 The conversion ratio of switchable antibacterial agents **1-4** at the photostationary state (PSS).

Compounds	Conversion ratio (%)	Compounds	Conversion ratio (%)
1a	93	1b	88
2a	89	2b	81
3a	89	3b	80
4a	83	4b	79

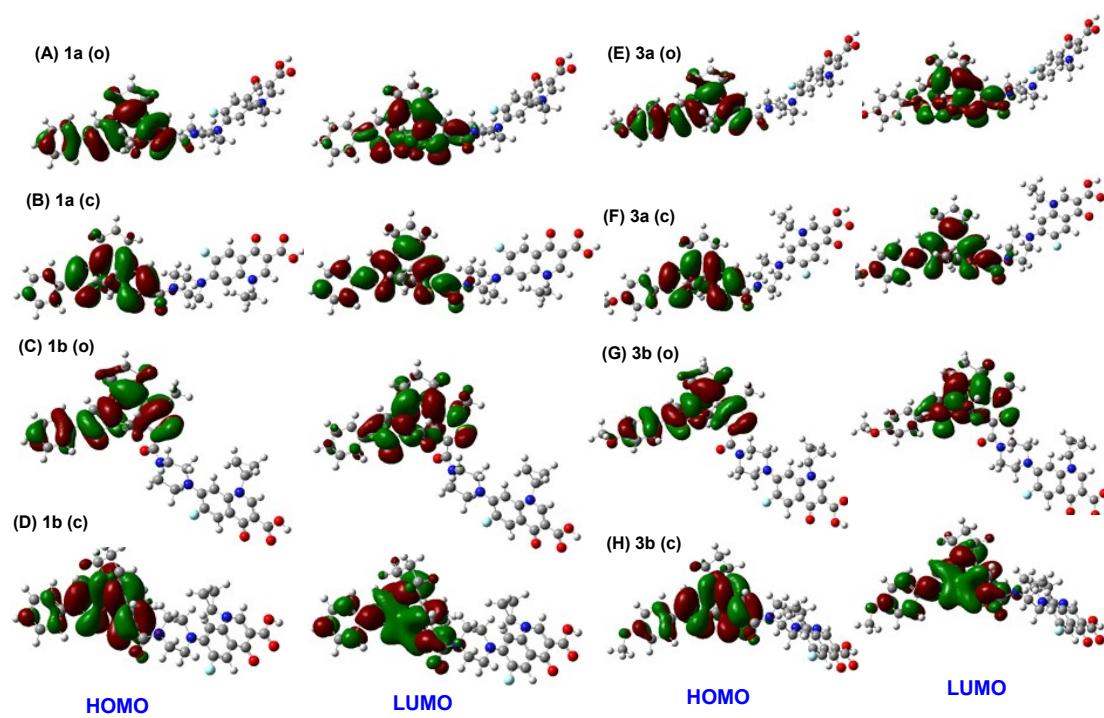


Fig. S12 Frontier molecular orbital profiles of **1** and **3** based on DFT calculations at the B3LYP/6-31G* level by using the Gaussian 09 program.

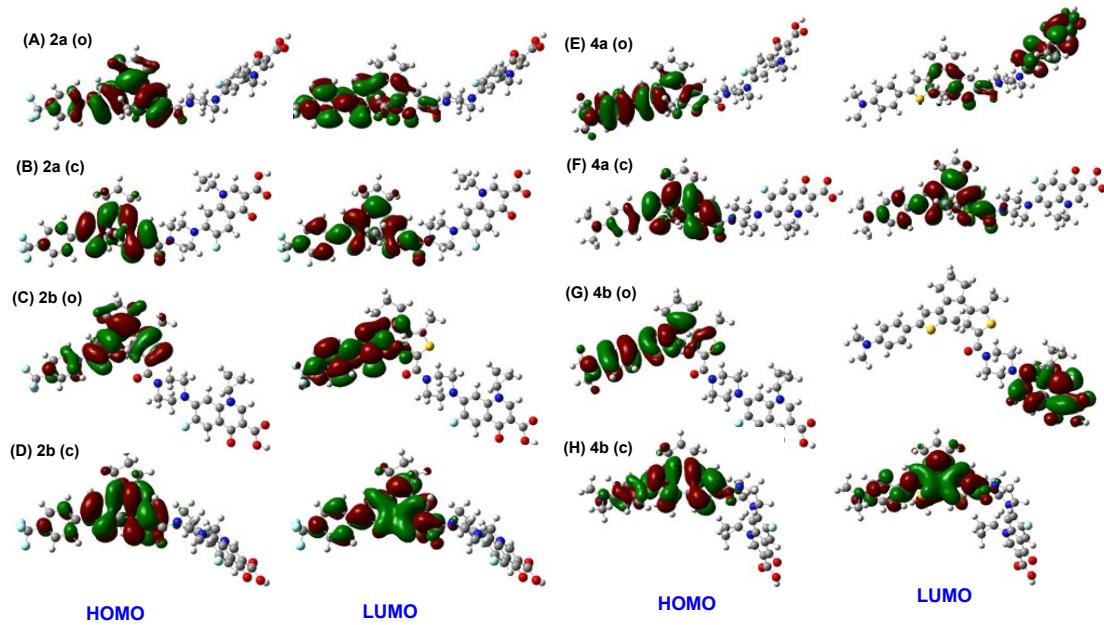


Fig. S13 Frontier molecular orbital profiles of **2** and **4** based on DFT calculations at the B3LYP/6-31G* level by using the Gaussian 09 program.

Table S3 Calculated parameters for dithienylethenes **1-4**.

Cpds	HOMO (ev)	LUMO (ev)	Eg (ev)	Cpds	HOMO (ev)	LUMO (ev)	Eg (ev)
1a (o)	-5.24	-1.16	4.08	1a (c)	-4.08	-2.40	1.68
1b (o)	-5.16	-1.12	4.04	1b (c)	-4.18	-2.49	1.69
2a (o)	-5.42	-1.33	4.09	2a (c)	-4.33	-2.63	1.70
2b (o)	-5.32	-1.23	4.09	2b (c)	-4.38	-2.74	1.64
3a (o)	-5.13	-1.15	3.98	3a (c)	-4.15	-2.44	1.71
3b (o)	-5.12	-1.09	4.03	3b (c)	-4.11	-2.42	1.69
4a (o)	-4.62	-1.04	3.58	4a (c)	-3.74	-2.03	1.71
4b (o)	-4.50	-1.01	3.49	4b (c)	-3.81	-2.13	1.68

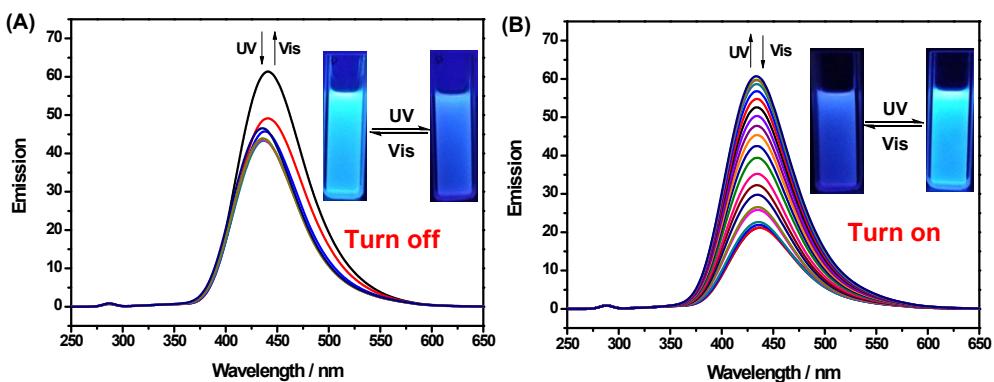


Fig. S14 Emission spectral changes of dithienylethenes **1b** and **3b** with 254 nm UV and > 402 nm visible light irradiation in DMSO (2.0×10^{-5} mol/L) at 298 K. (Inset) Corresponding fluorescence intensity changes of dithienylethene **1b** and **3b** in DMSO upon photoirradiation. (A) emission spectral changes for **1b** [0 s, 5 s, 10 s, 15 s, 20 s, 25 s, 30 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 50 s, 60 s (PSS-open)]; (B) emission spectral changes for **3b** (0 s, 5 s, 10 s, 15 s, 20 s, 25 s, 30 s, 35 s, 45 s, 55 s, 65 s, 80 s, 95 s, 110 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 50 s, 70 s, 90 s, 110 s, 130 s, 150 s, 170 s, 200 s, 230 s (PSS-open)].

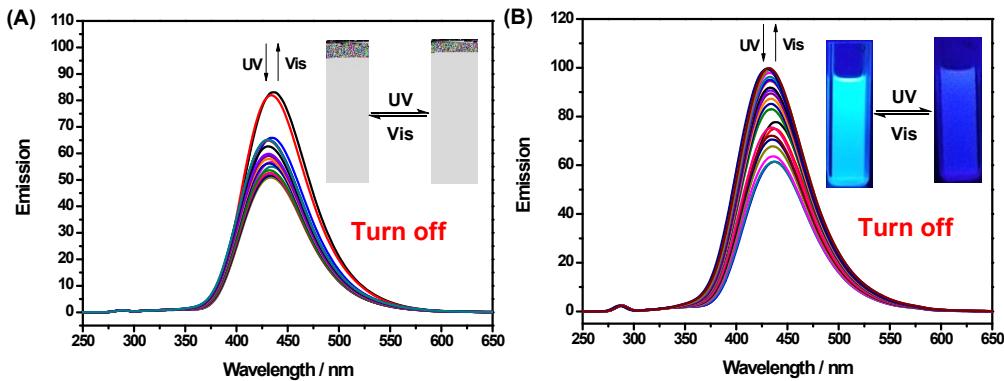


Fig. S15 Emission spectral changes of dithienylethenes **2a** and **2b** with 254 nm UV and > 402 nm visible light irradiation in DMSO (2.0×10^{-5} mol/L) at 298 K. (Inset) Corresponding fluorescence intensity changes of dithienylethene **2a** and **2b** in DMSO upon photoirradiation. (A) emission spectral changes for **2a** [0 s, 5 s, 10 s, 15 s, 25 s, 30 s, 45 s, 55 s, 60 s, 65 s, 70 s, 75 s, 80 s, 85 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 50 s, 60 s, 80 s, 100 s, 120 s, 150 s, 180 s, 200 s (PSS-open)]; (B) emission spectral changes for **2b** [0 s, 5 s, 10 s, 15 s, 25 s, 30 s, 45 s, 55 s, 60 s, 65 s, 70 s, 75 s, 80 s, 90 s, 95 s, 100 s, 110 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 60 s, 80 s, 100 s, 120 s, 140 s, 170 s, 200 s, 230 s, 260 s, 300 s (PSS-open)]

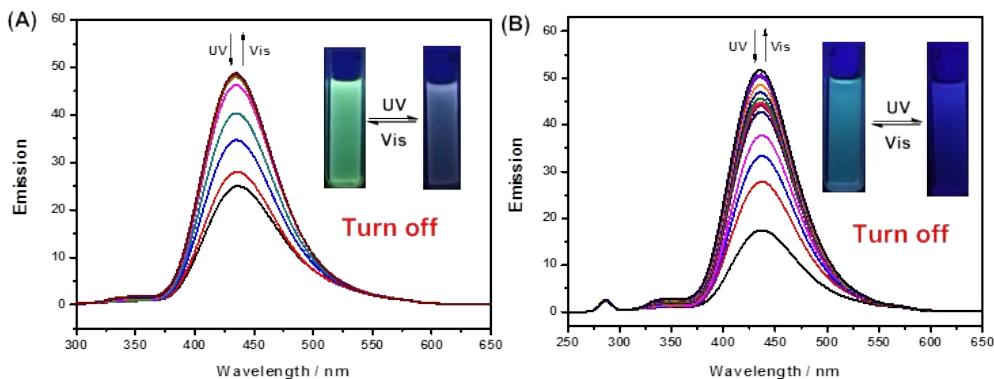


Fig. S16 Emission spectral changes of dithienylethenes **4a** and **4b** with 254 nm UV and > 402 nm visible light irradiation in DMSO (2.0×10^{-5} mol/L) at 298 K. (Inset) Corresponding fluorescence intensity changes of dithienylethene **4a** and **4b** in DMSO upon photoirradiation. (A) emission spectral changes for **4a** [0 s, 5 s, 10 s, 15 s, 20 s, 25 s, 30 s, 35 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 60 s, 80 s, 100 s (PSS-open)]; (B) emission spectral changes for **4b** [0 s, 5 s, 10 s, 15 s, 20 s, 25 s, 30 s, 35 s, 40 s, 45 s, 55 s, 65 s, 75 s, 85 s, 95 s (PSS-closed); 0 s, 10 s, 20 s, 30 s, 40 s, 50 s, 60 s, 80 s, 100 s, 120 s, 140 s, 160 s, 180 s, 200 s, 230 s (PSS-open)].

Table S4 The fluorescence data of dithienylethenes **1–4** in DMSO at 298 K (2.0×10^{-5} mol/L)

Compounds	λ_{em}^a (nm)	Φ_f (open) ^b	Φ_f (closed) ^c
1a	440	0.35	0.14
1b	440	0.29	0.12
2a	436	0.41	0.20
2b	431	0.48	0.22
3a	428	0.06	0.40
3b	432	0.11	0.33
4a	434	0.23	0.12
4b	435	0.25	0.08

^a Fluorescence emission maxima; ^b Fluorescence quantum yield of open isomers determined by a standard method with quinoline sulfate in 0.1 M aqueous H₂SO₄ ($\Phi_f = 0.55$, $\lambda_{\text{ex}} = 313$ nm) as reference; ^c Fluorescence quantum yield of closed isomers determined by a standard method with quinoline sulfate in 0.1 M aqueous H₂SO₄ ($\Phi_f = 0.55$, $\lambda_{\text{ex}} = 313$ nm) as reference.

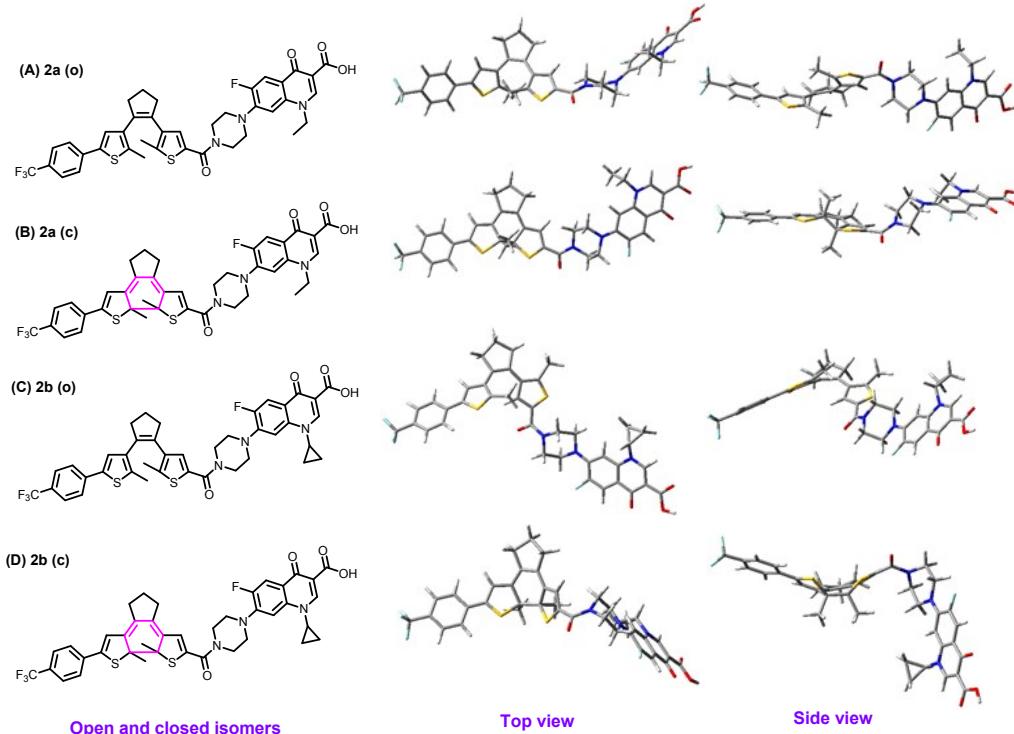


Fig. S17 The energy-minimized structure of open and closed isomers of **2a** and **2b** based on DFT calculations at the B3LYP/6-31G* level by using the Gaussian 09 program.

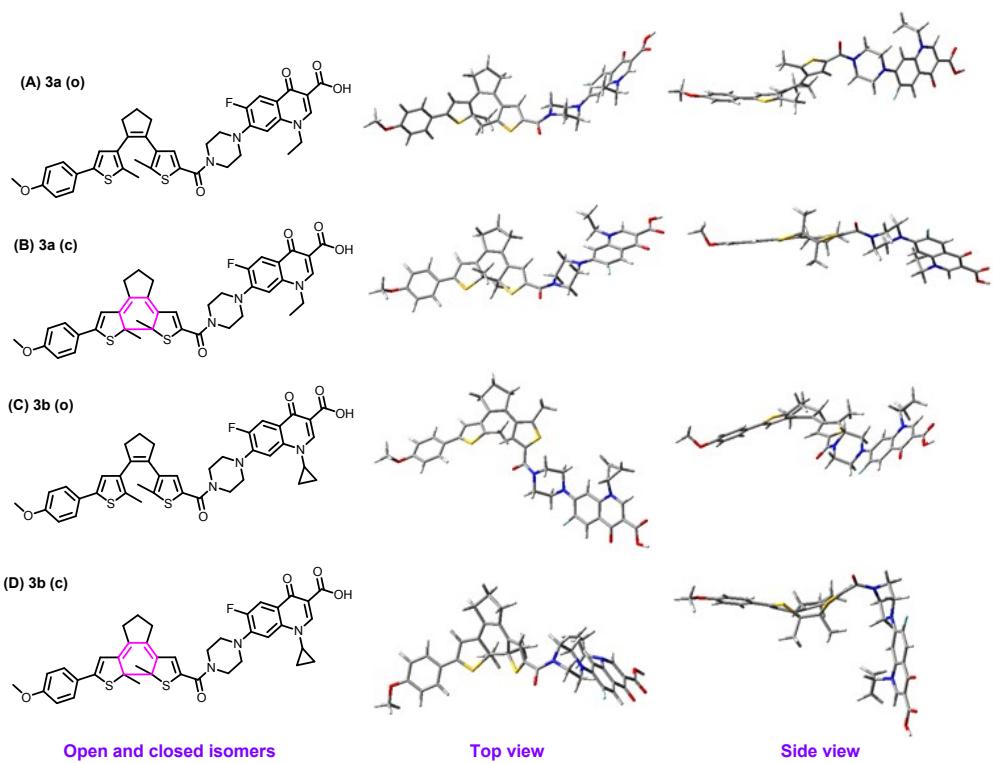


Fig. S18 The energy-minimized structure of open and closed isomers of **3a** and **3b** based on DFT calculations at the B3LYP/6-31G* level by using the Gaussian 09 program.

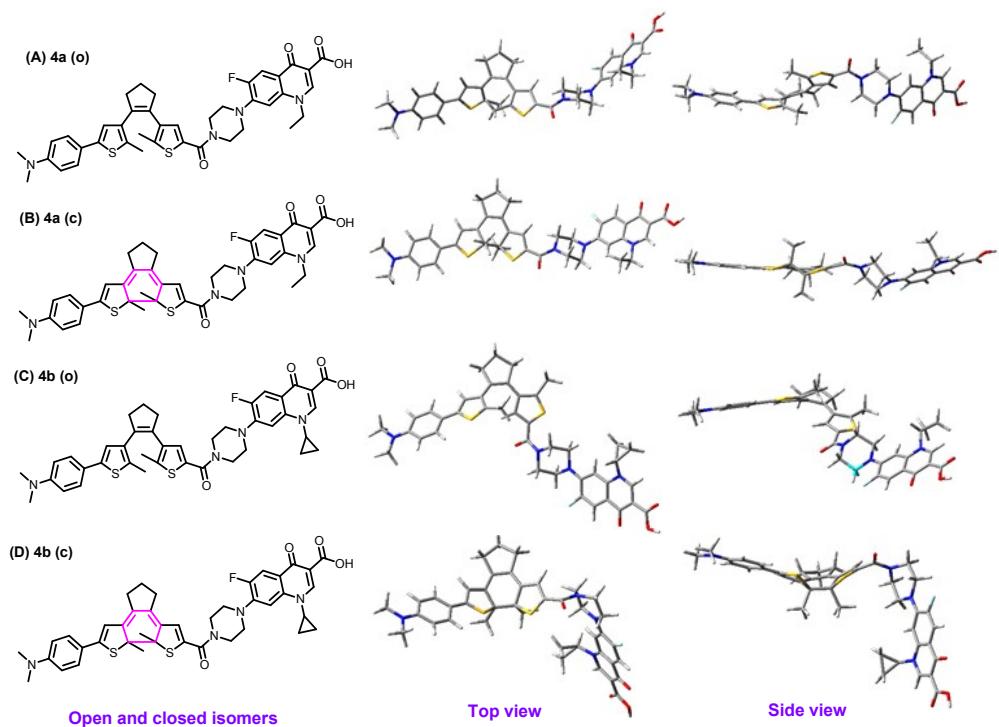


Fig. S19 The energy-minimized structure of open and closed isomers of **4a** and **4b** based on DFT calculations at the B3LYP/6-31G* level by using the Gaussian 09 program.

Appendix: NMR and Mass spectra

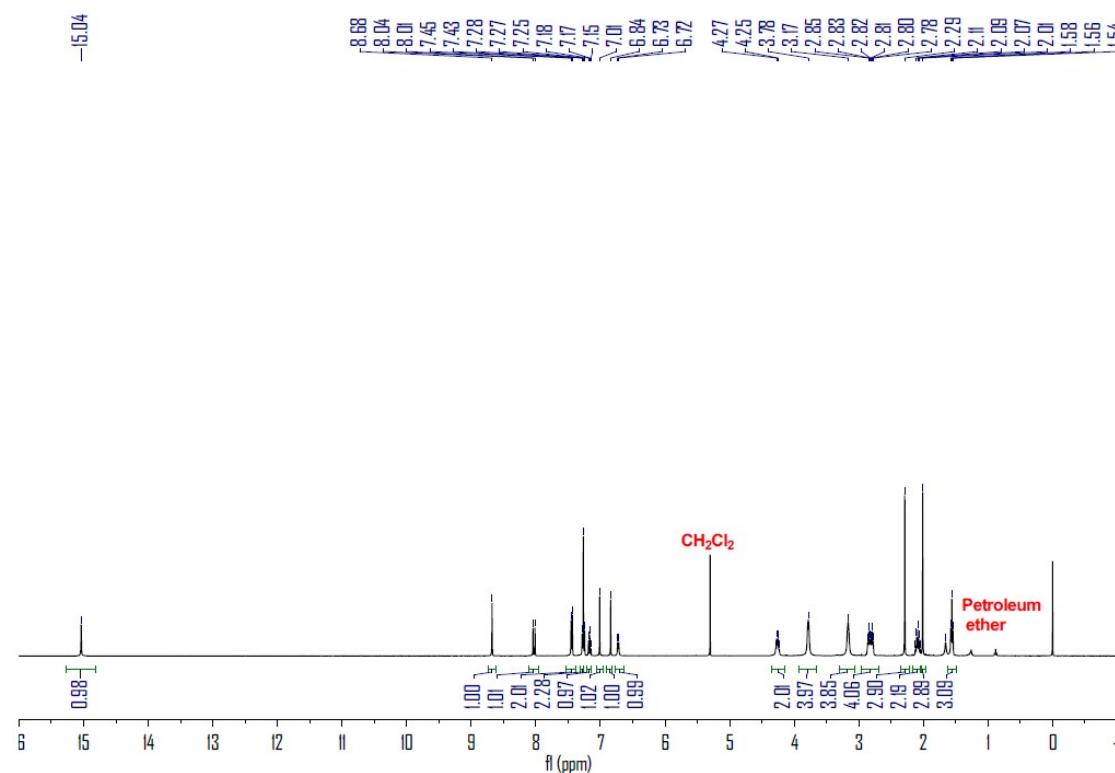


Fig. S20 400 MHz ^1H NMR spectrum of dithienylethene **1a** in CDCl_3 at room temperature.

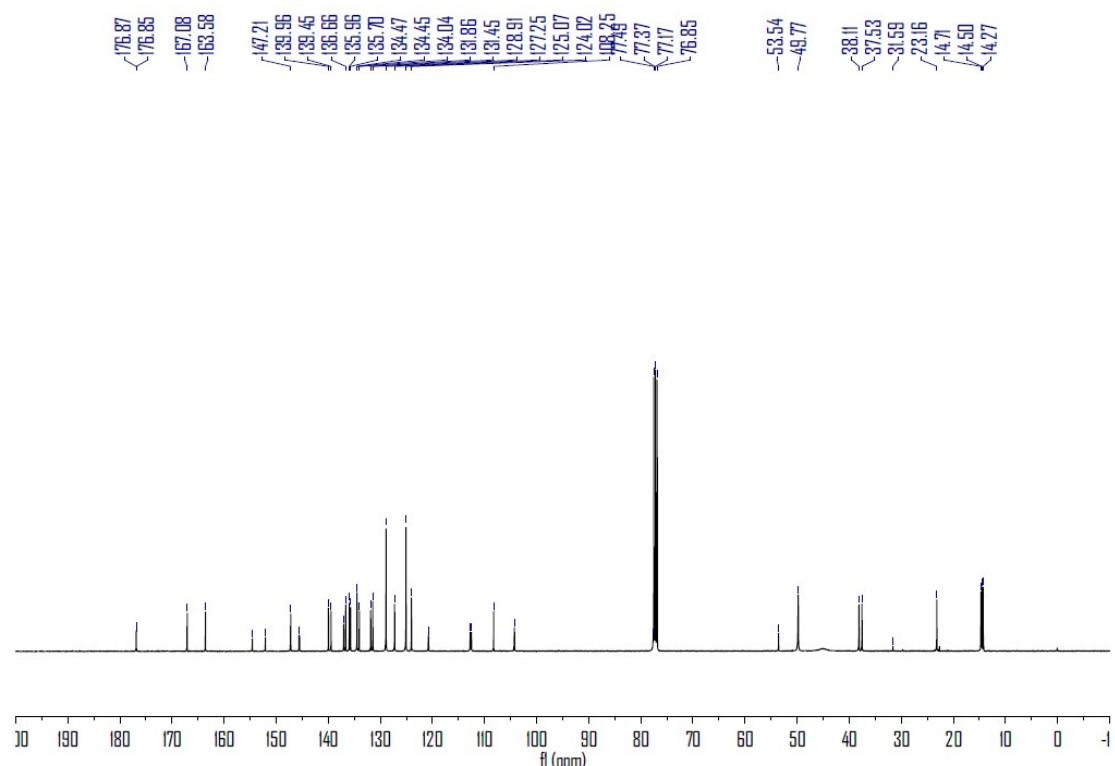


Fig. S21 100 MHz ^1H NMR spectrum of dithienylethene **1a** in CDCl_3 at room temperature.

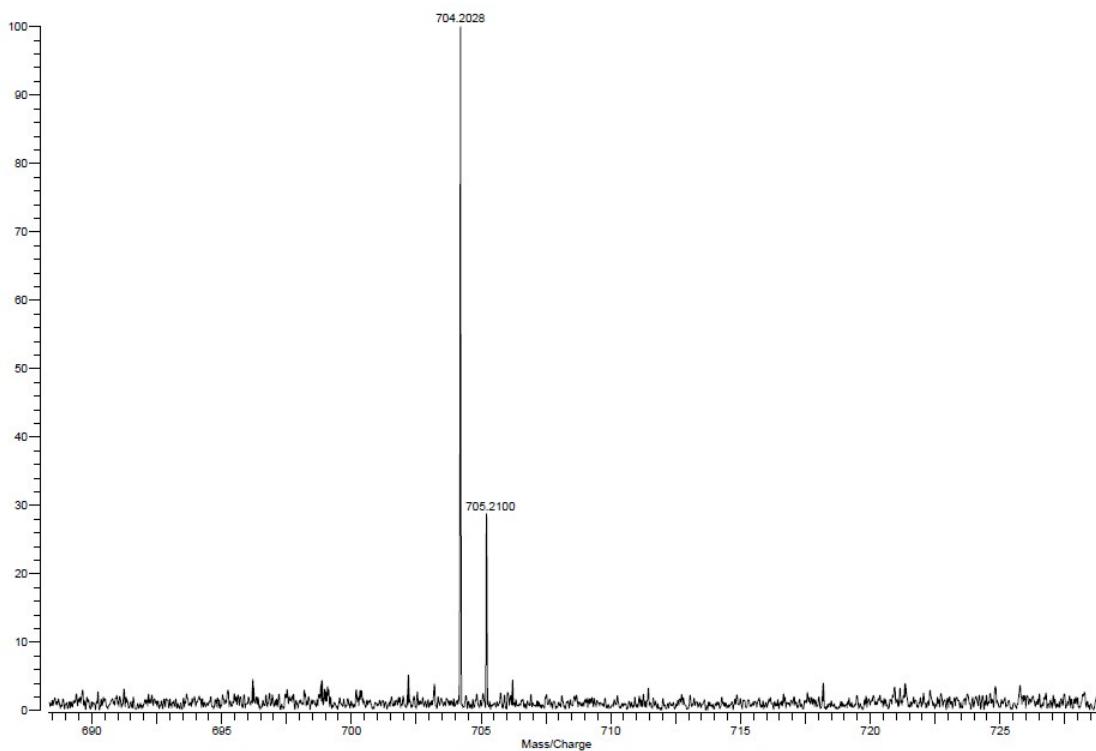


Fig. S22 HRMS of dithienylethene **1a**.

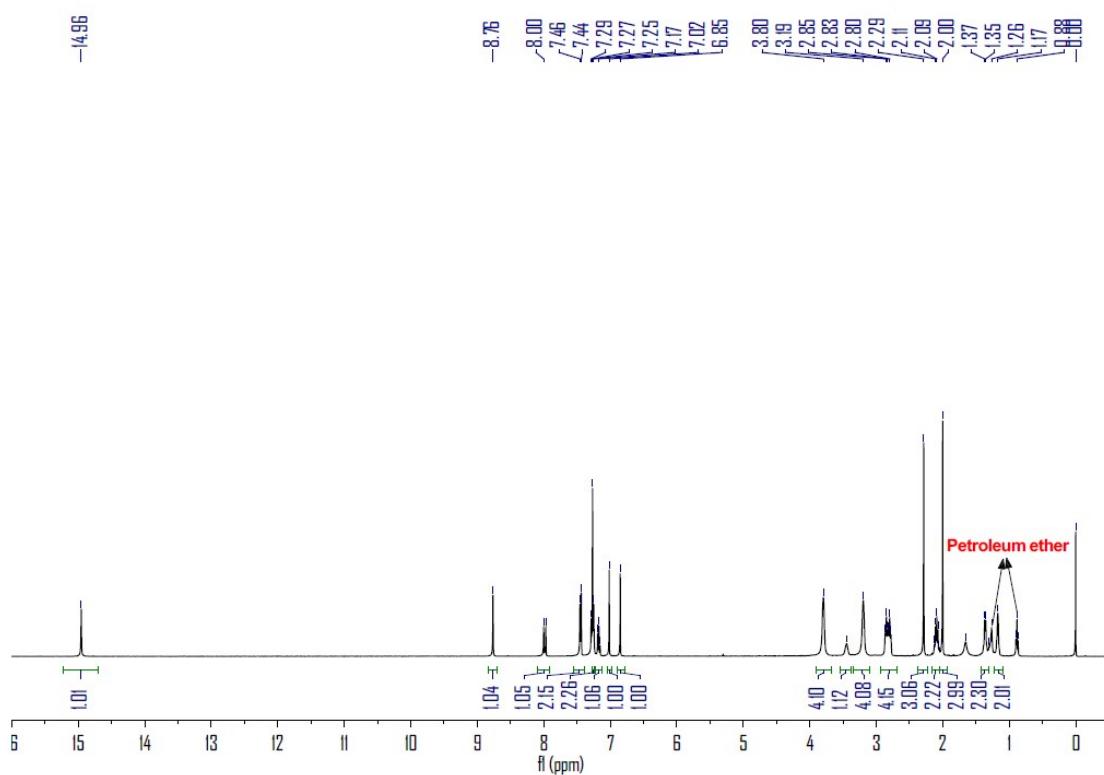


Fig. S23 400 MHz ^1H NMR spectrum of dithienylethene **1b** in CDCl_3 at room temperature.

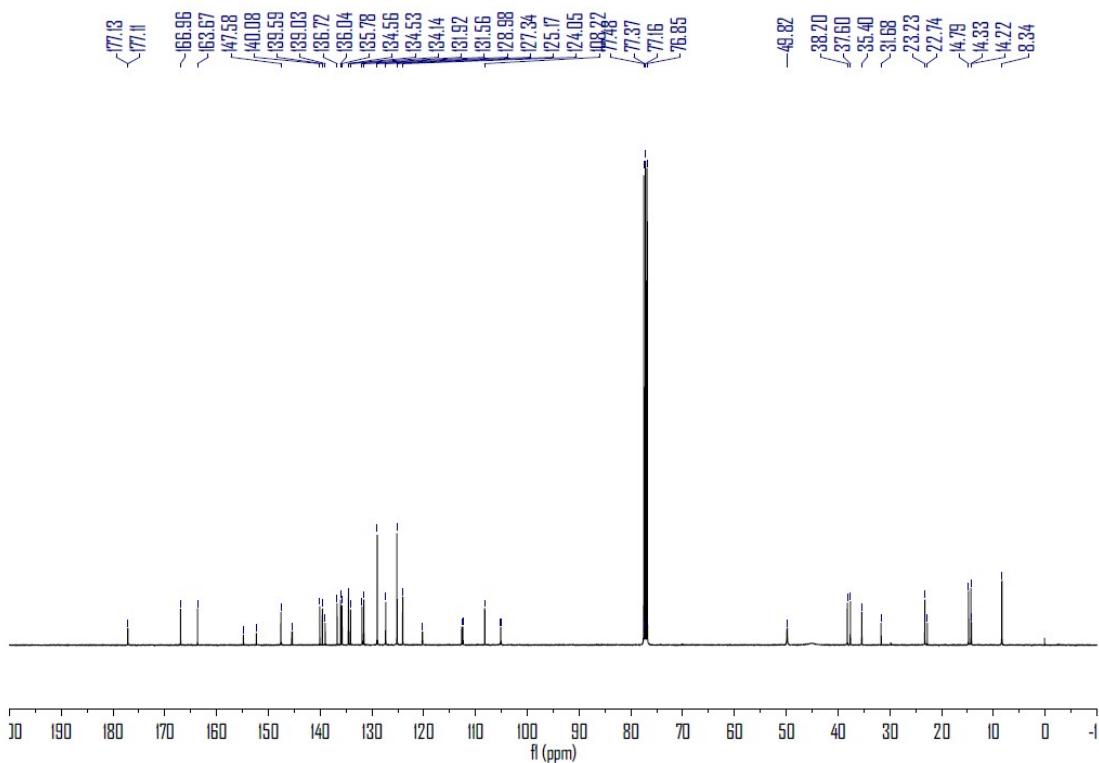


Fig. S24 100 MHz ^1H NMR spectrum of dithienylethene **1b** in CDCl_3 at room temperature.

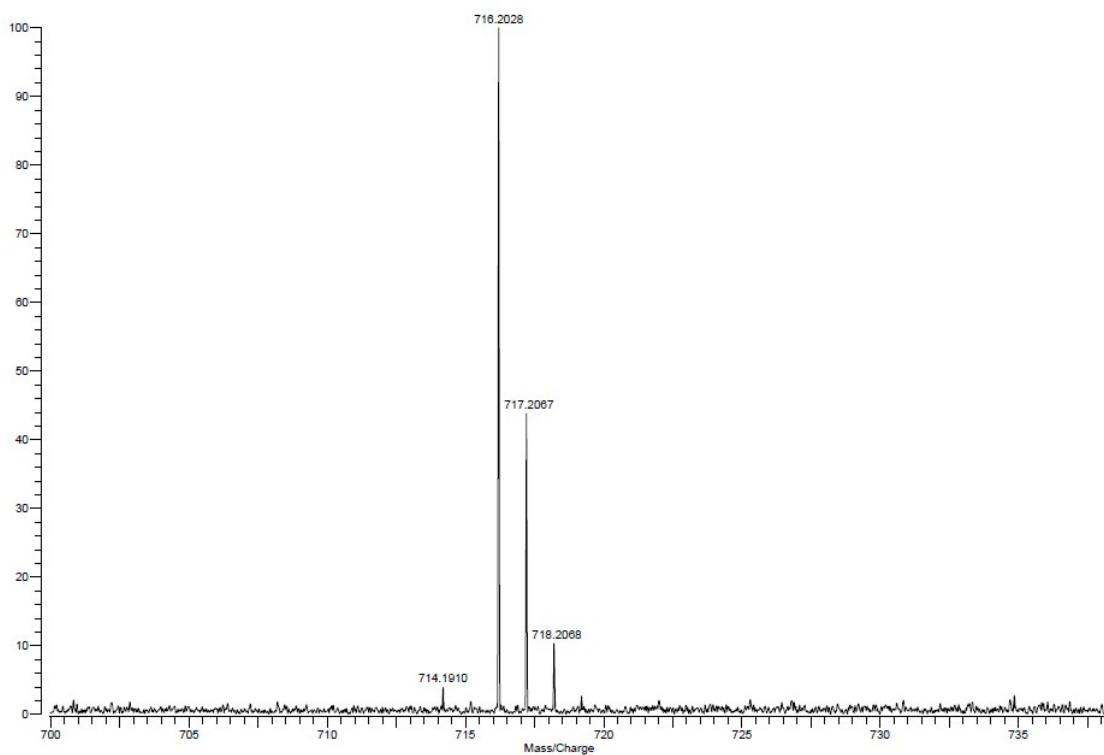


Fig. S25 HRMS of dithienylethene **1b**.

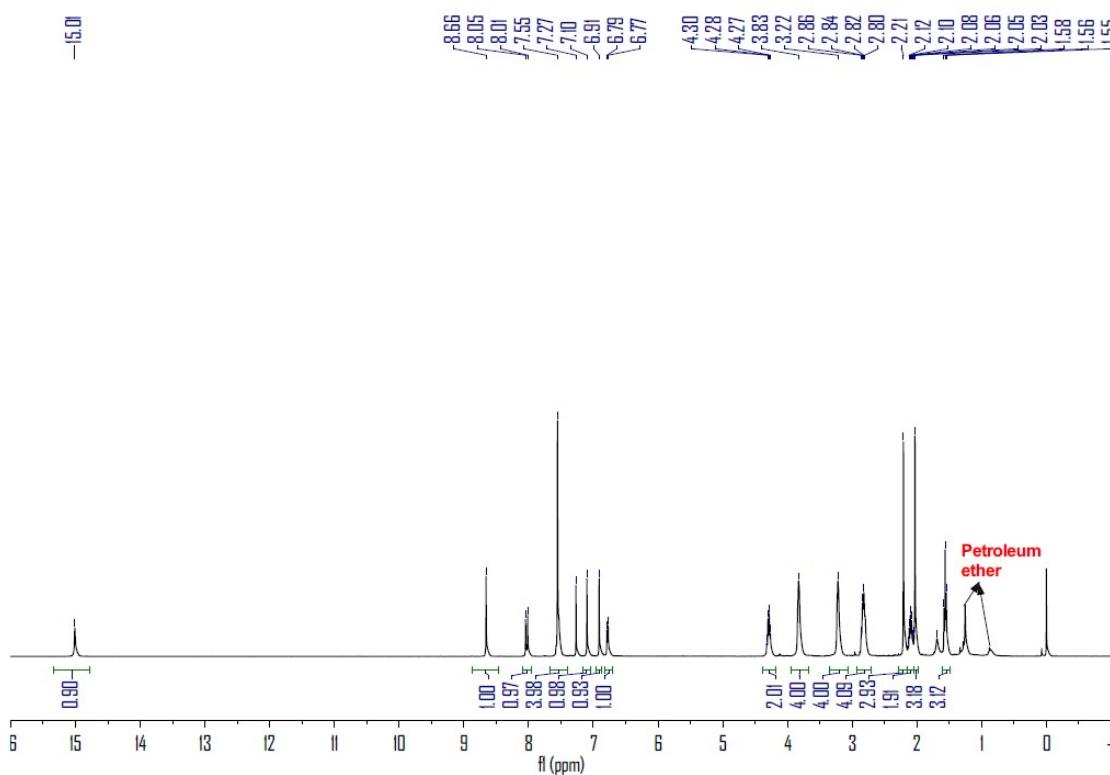


Fig. S26 400 MHz ^1H NMR spectrum of dithienylethene **2a** in CDCl_3 at room temperature.

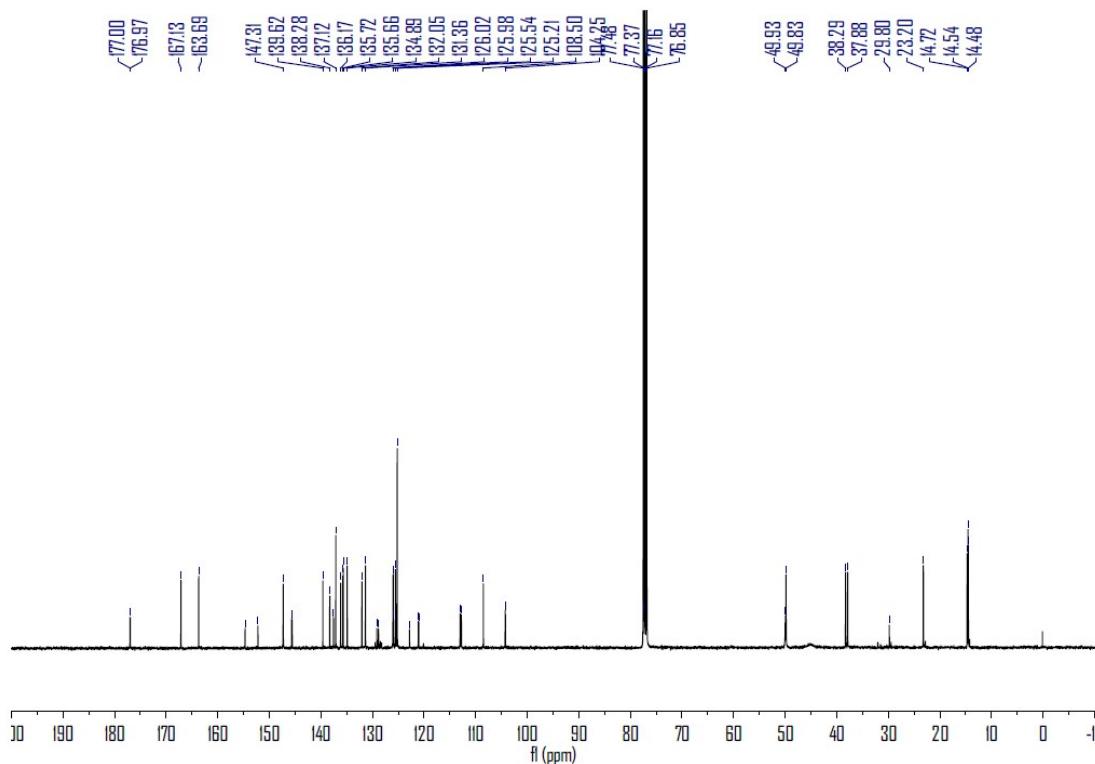


Fig. S27 100 MHz ^1H NMR spectrum of dithienylethene **2a** in CDCl_3 at room temperature.

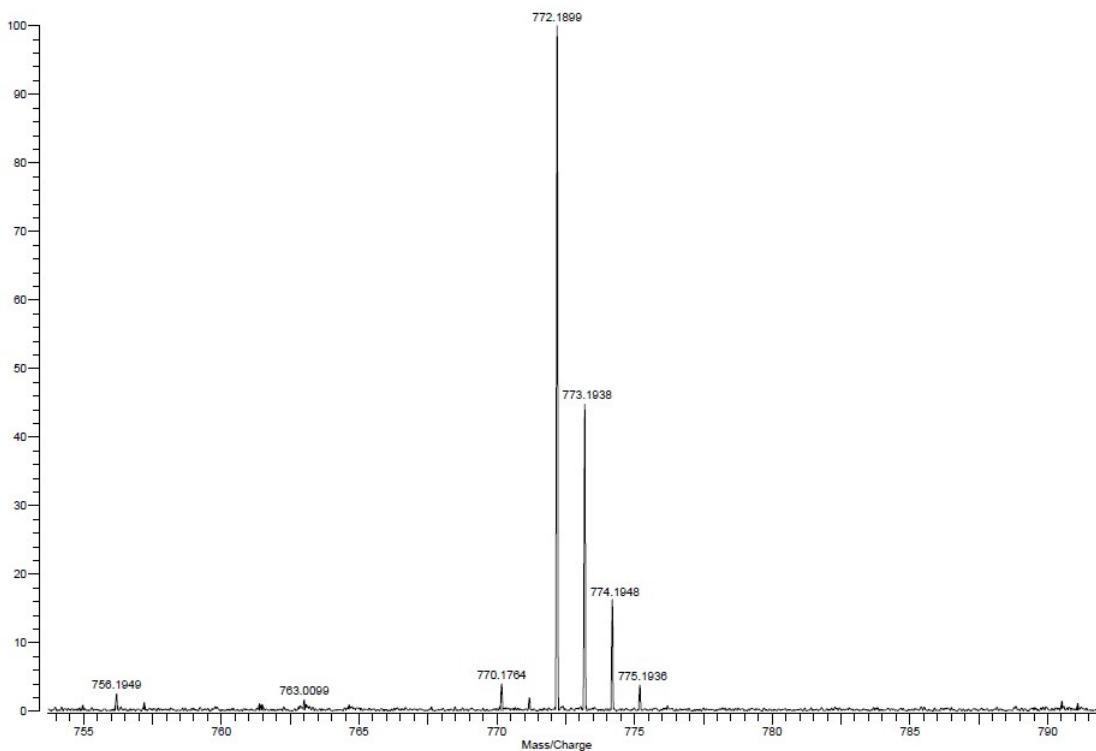


Fig. S28 HRMS of dithienylethene **2a**.

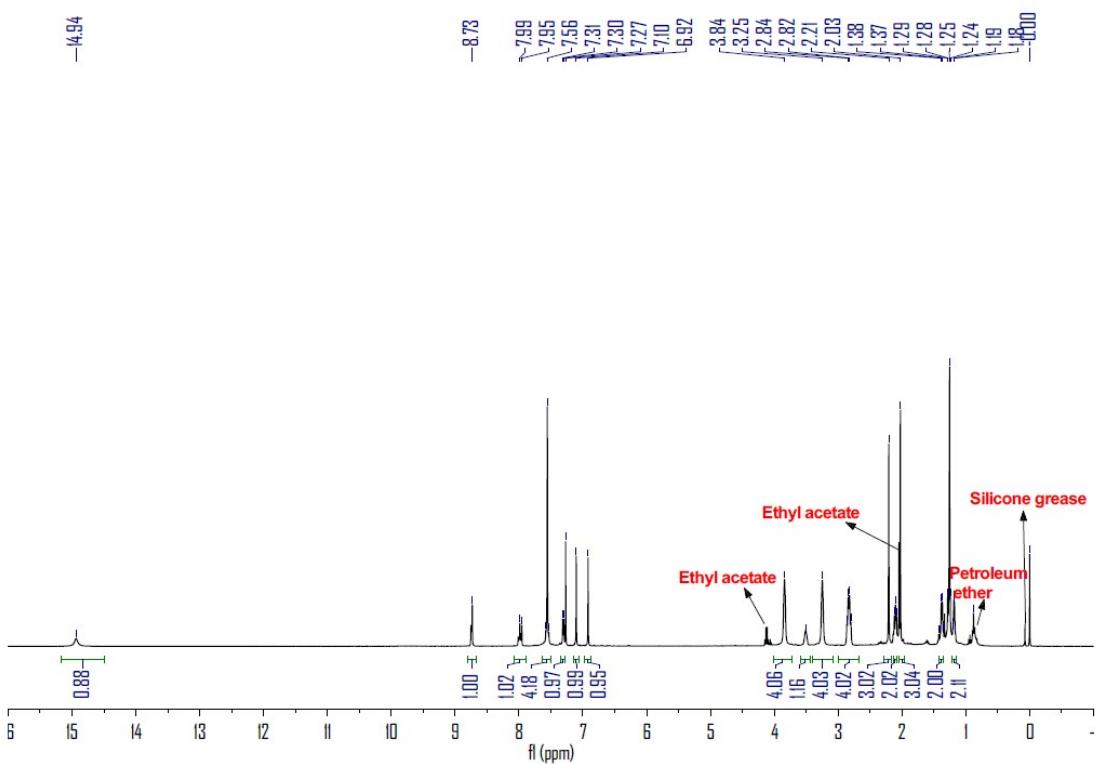


Fig. S29 400 MHz ^1H NMR spectrum of dithienylethene **2b** in CDCl_3 at room temperature.

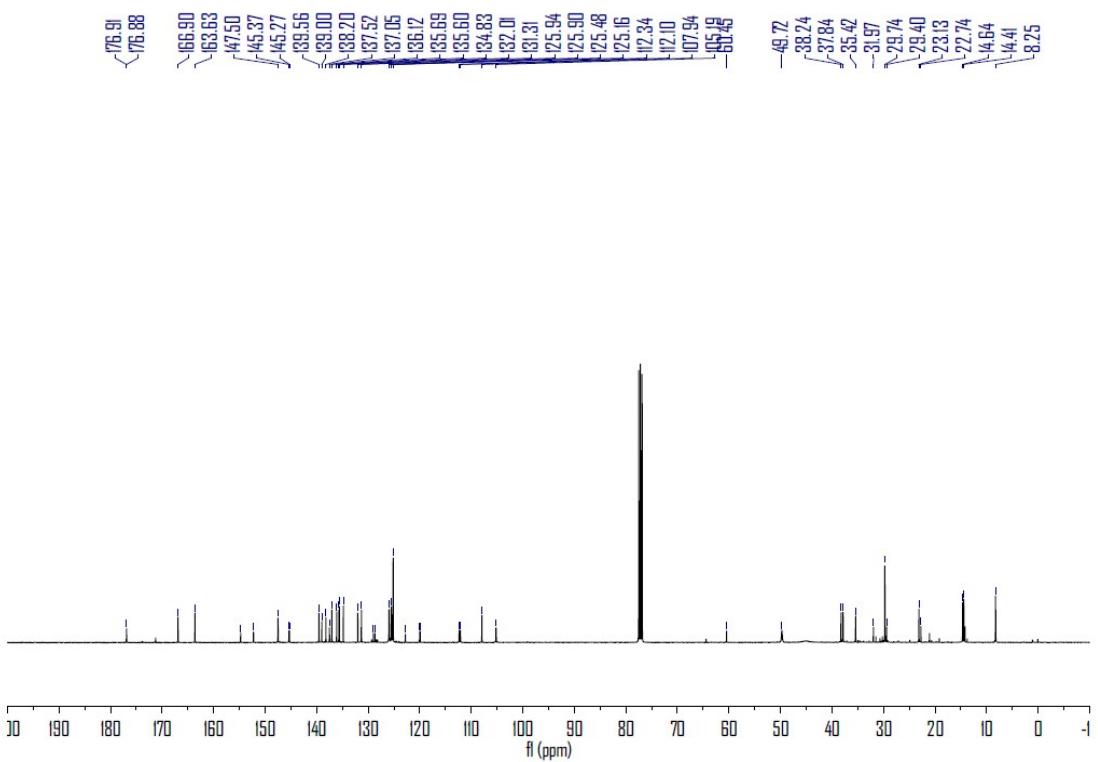


Fig. S30 100 MHz ^1H NMR spectrum of dithienylethene **2b** in CDCl_3 at room temperature.

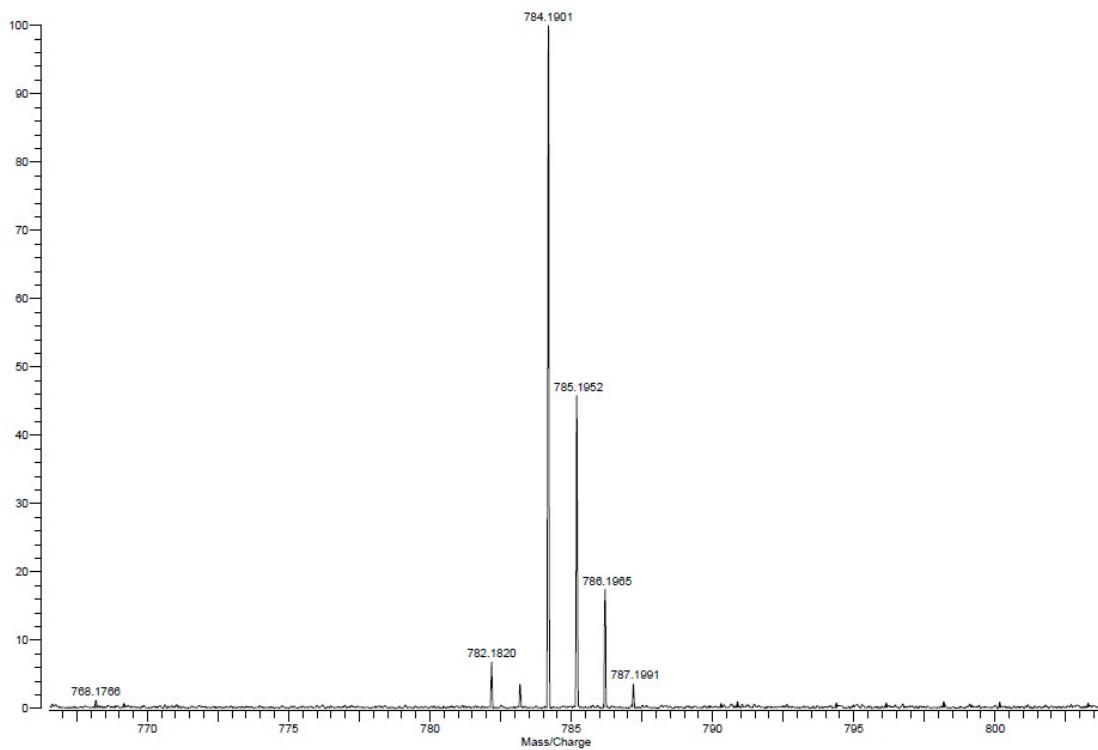


Fig. S31 HRMS of dithienylethene **2b**.

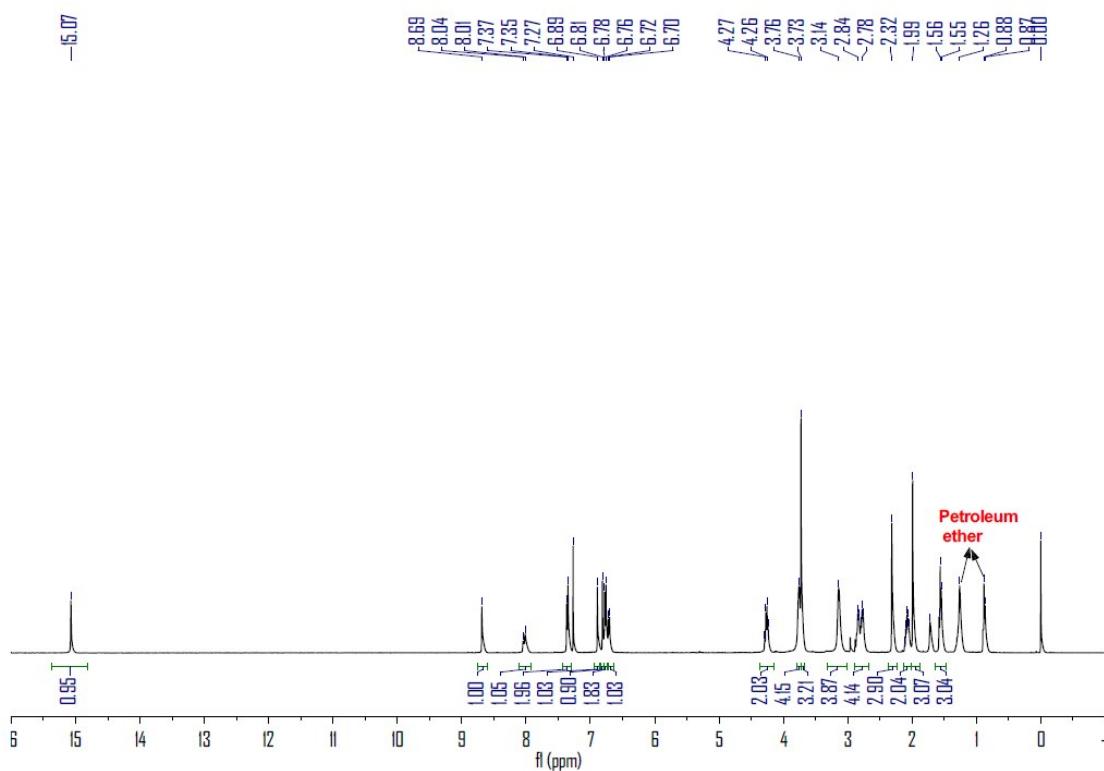


Fig. S32 400 MHz ^1H NMR spectrum of dithienylethene **3a** in CDCl_3 at room temperature.

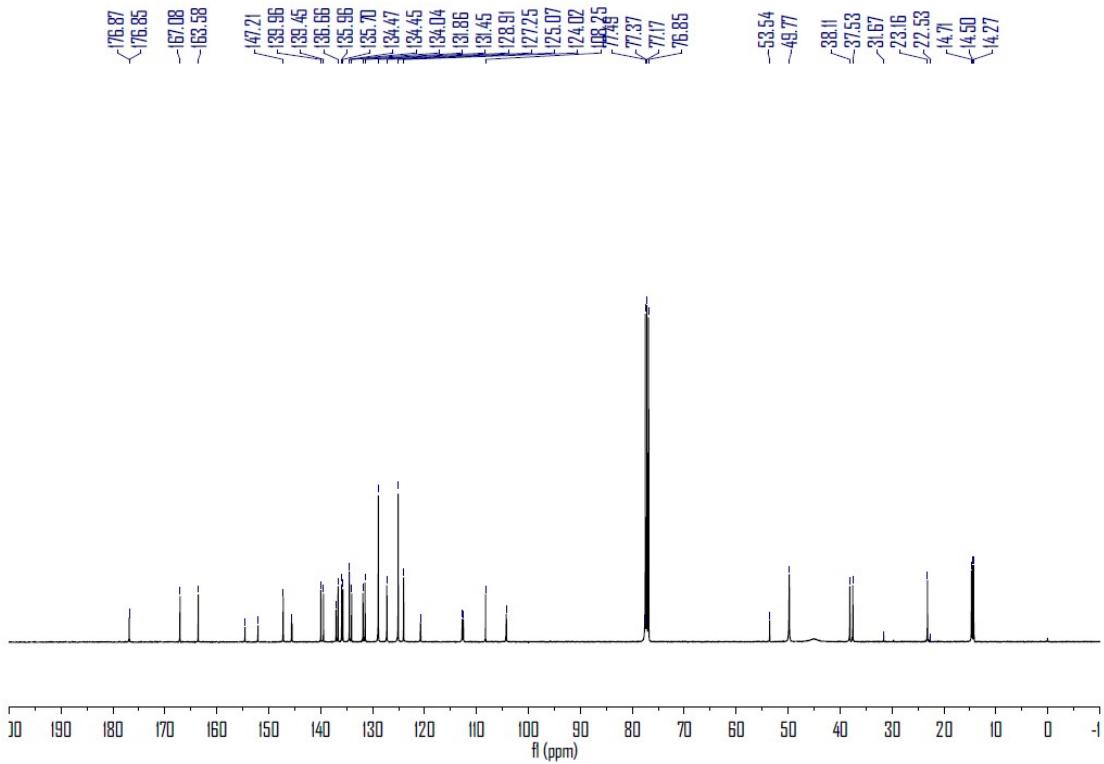


Fig. S33 100 MHz ^1H NMR spectrum of dithienylethene **3a** in CDCl_3 at room temperature.

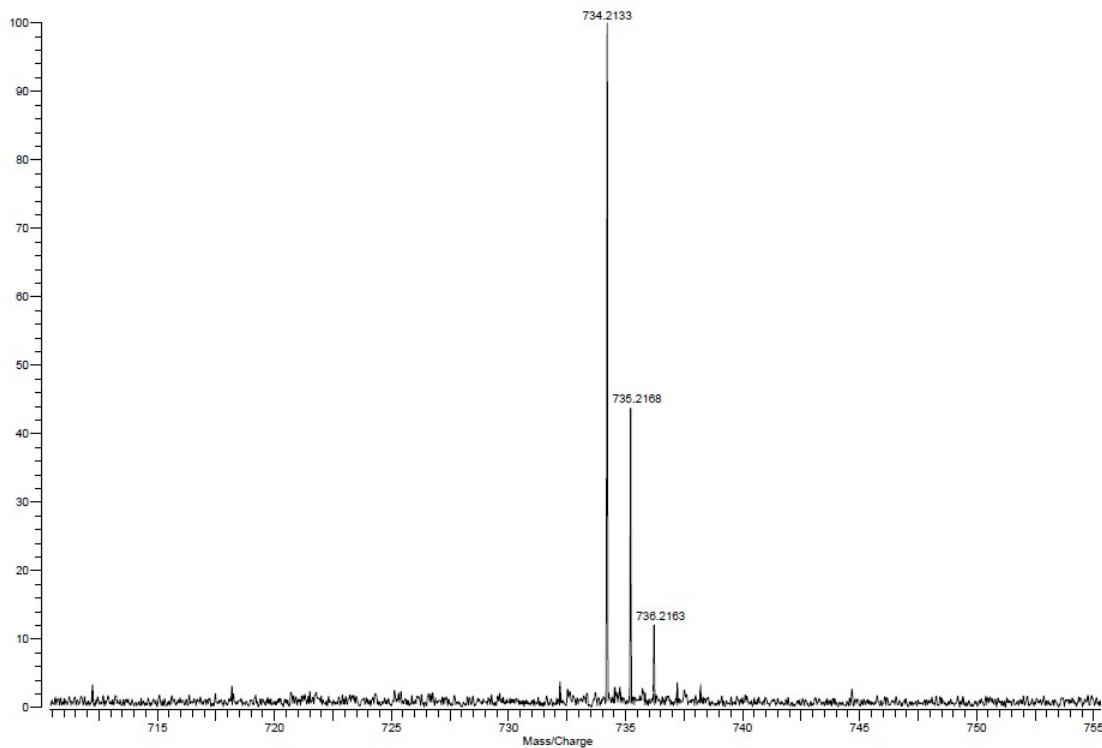


Fig. S34 HRMS of dithienylethene **3a**.

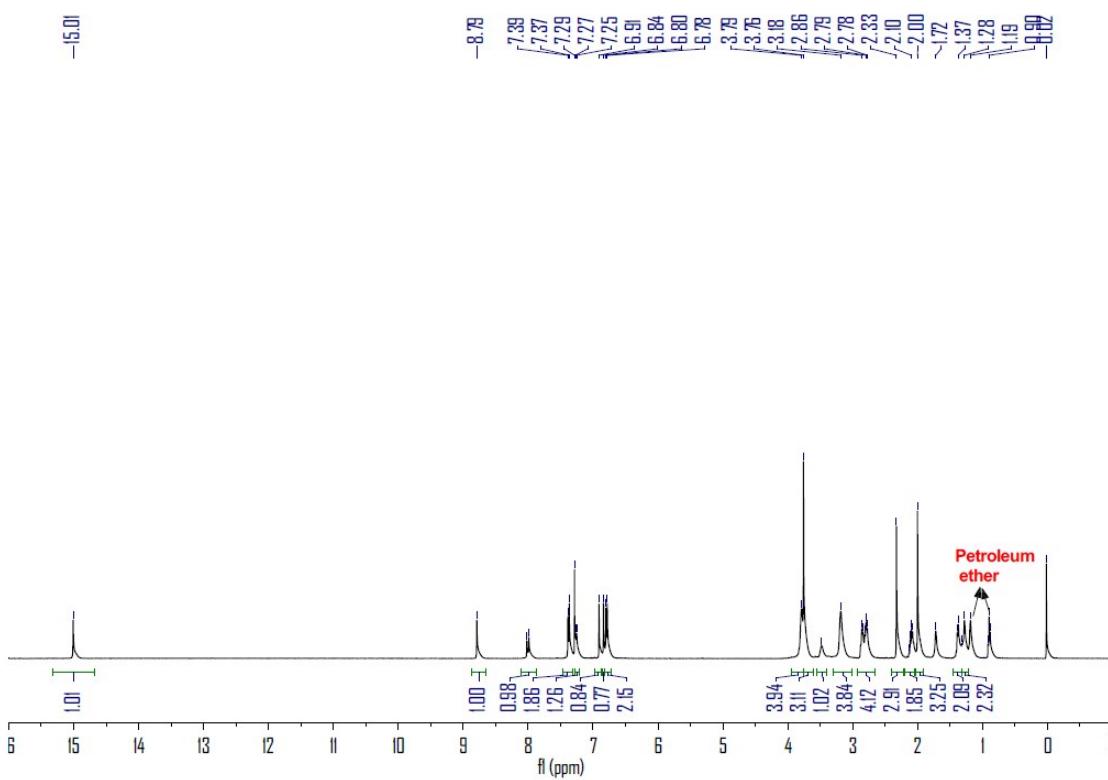


Fig. S35 400 MHz ^1H NMR spectrum of dithienylethene **3b** in CDCl_3 at room temperature.

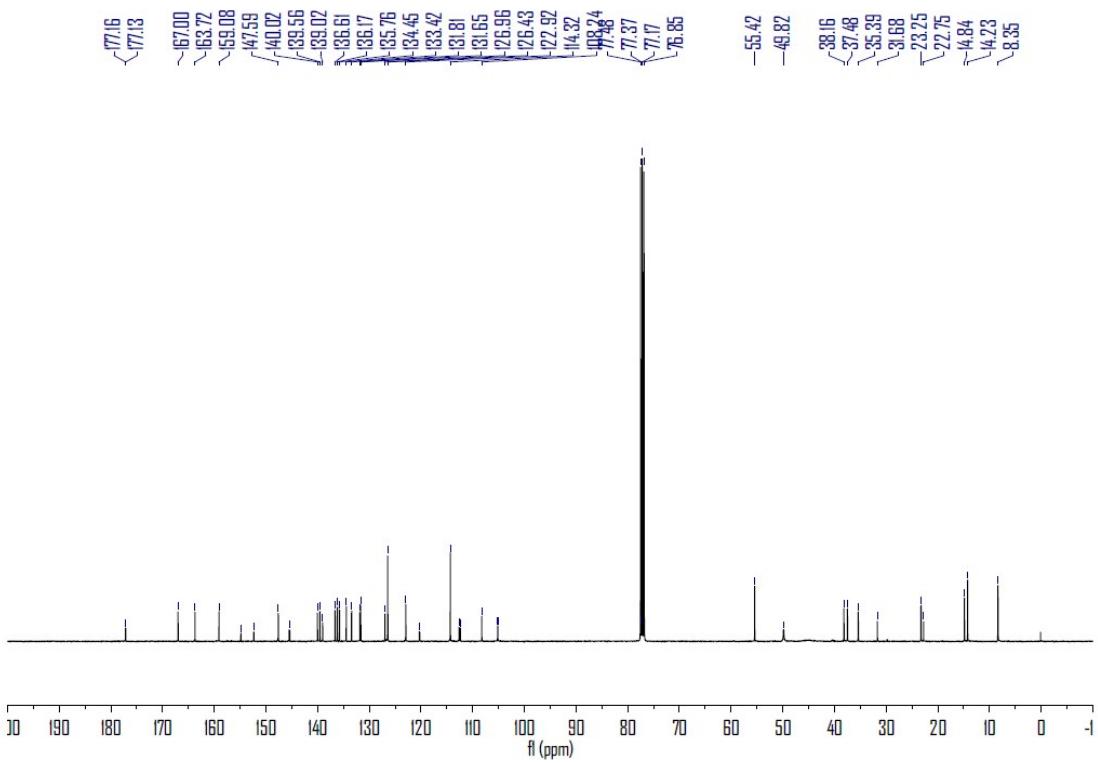


Fig. S36 100 MHz ^1H NMR spectrum of dithienylethene **3b** in CDCl_3 at room temperature.

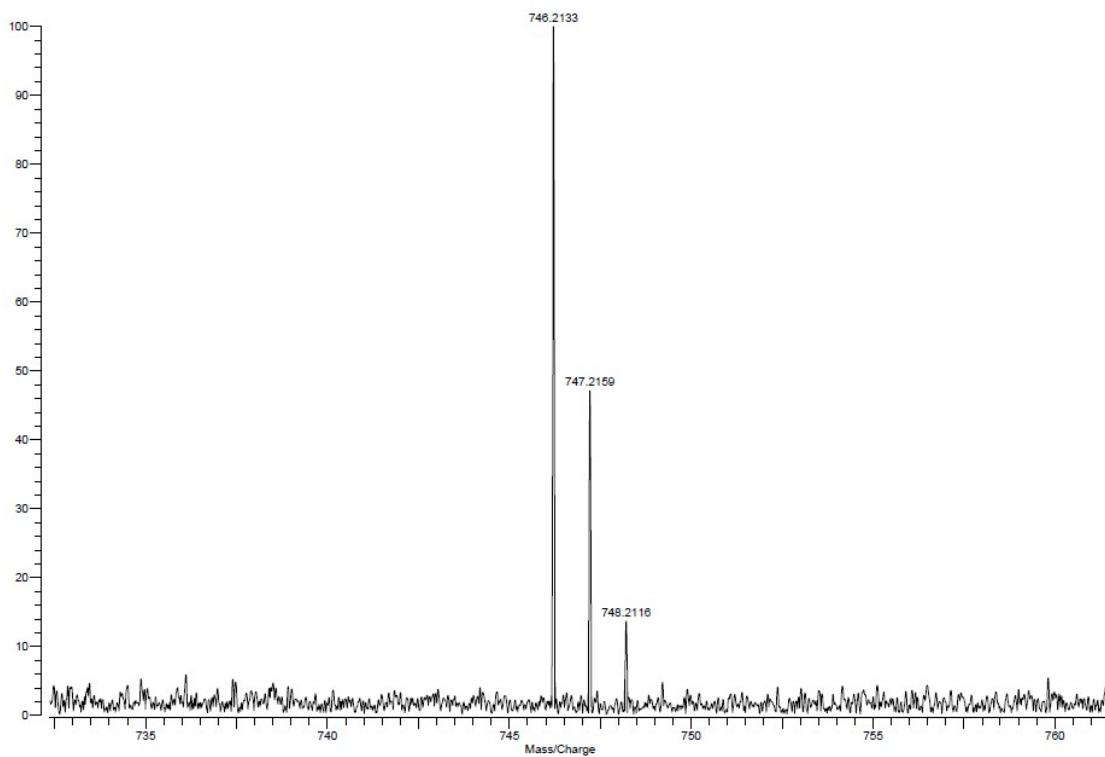


Fig. S37 HRMS of dithienylethene **3b**.

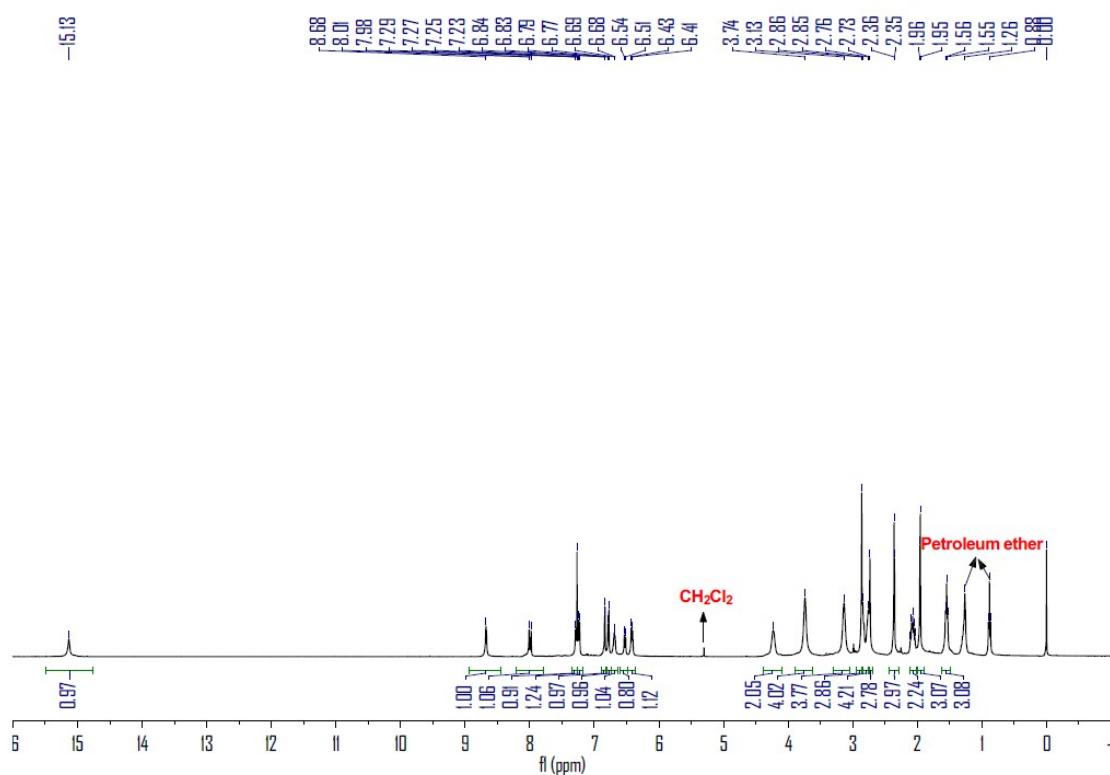


Fig. S38 400 MHz ^1H NMR spectrum of dithienylethene **4a** in CDCl_3 at room temperature.

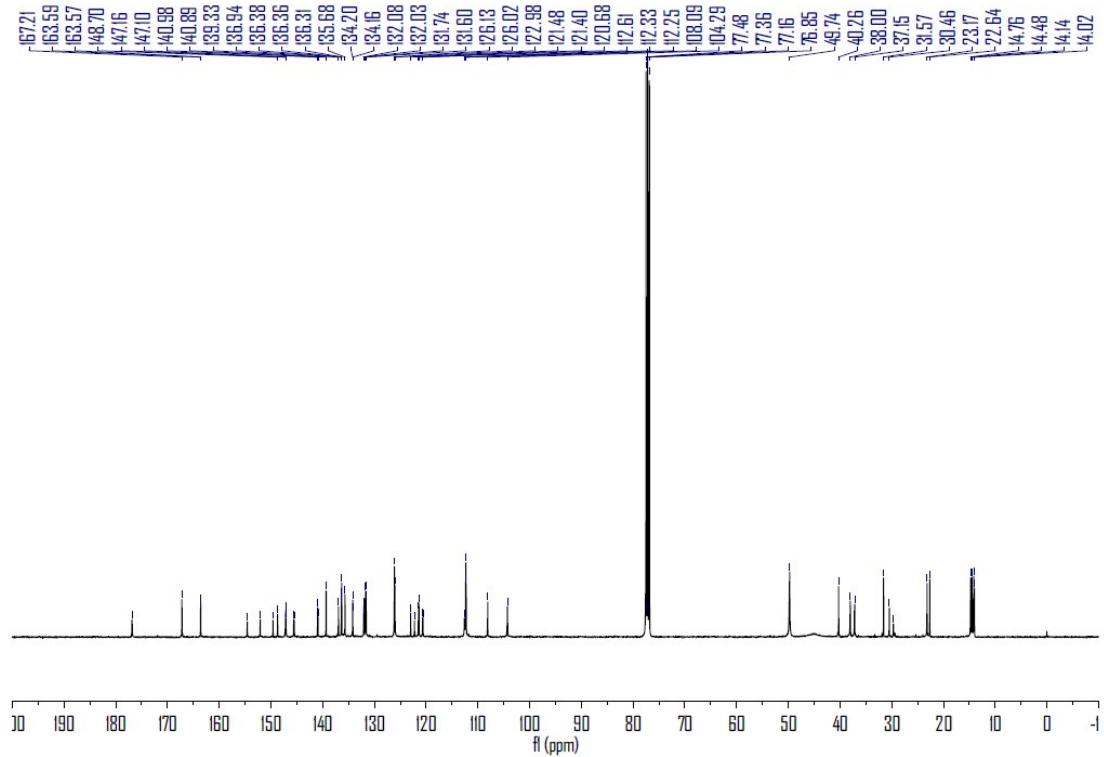


Fig. S39 100 MHz ^1H NMR spectrum of dithienylethene **4a** in CDCl_3 at room temperature.

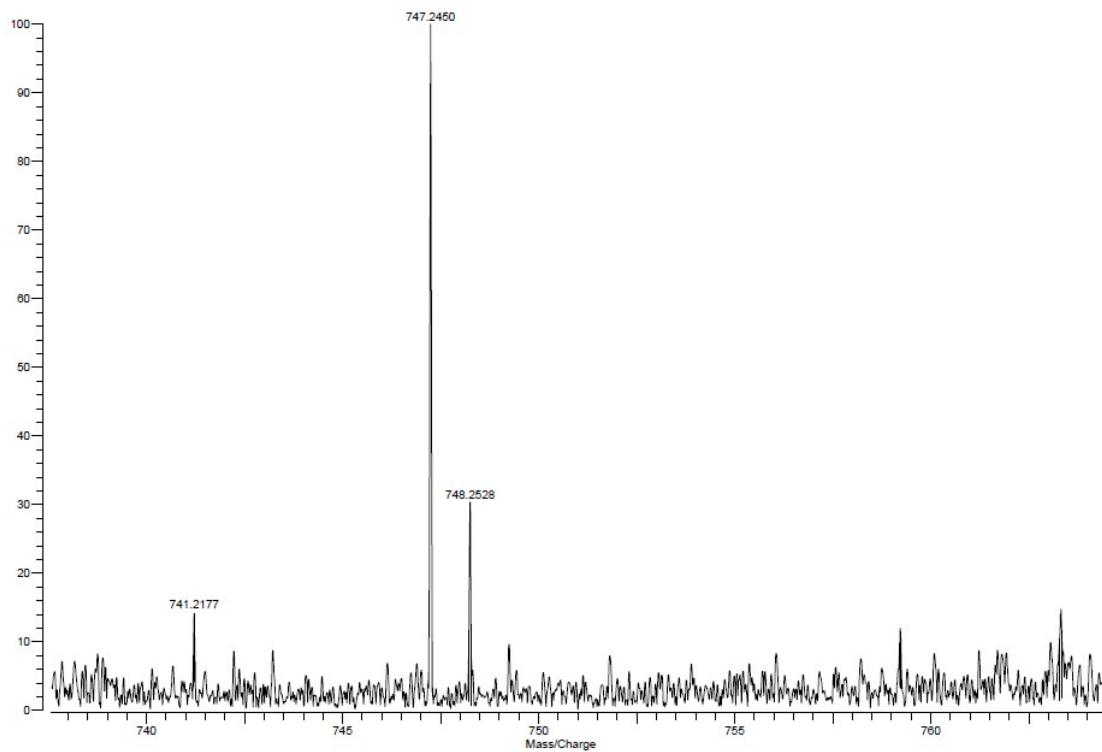


Fig. S40 HRMS of dithienylethene **4a**.

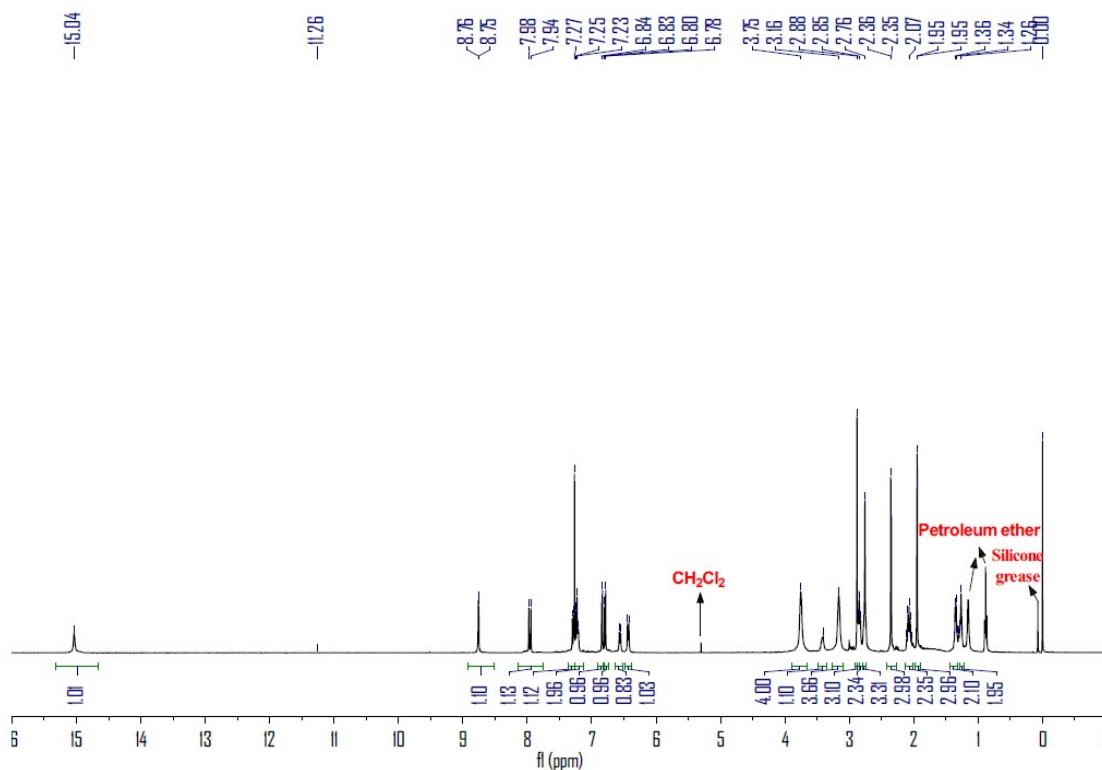


Fig. S41 400 MHz ^1H NMR spectrum of dithienylethene **4b** in CDCl_3 at room temperature.

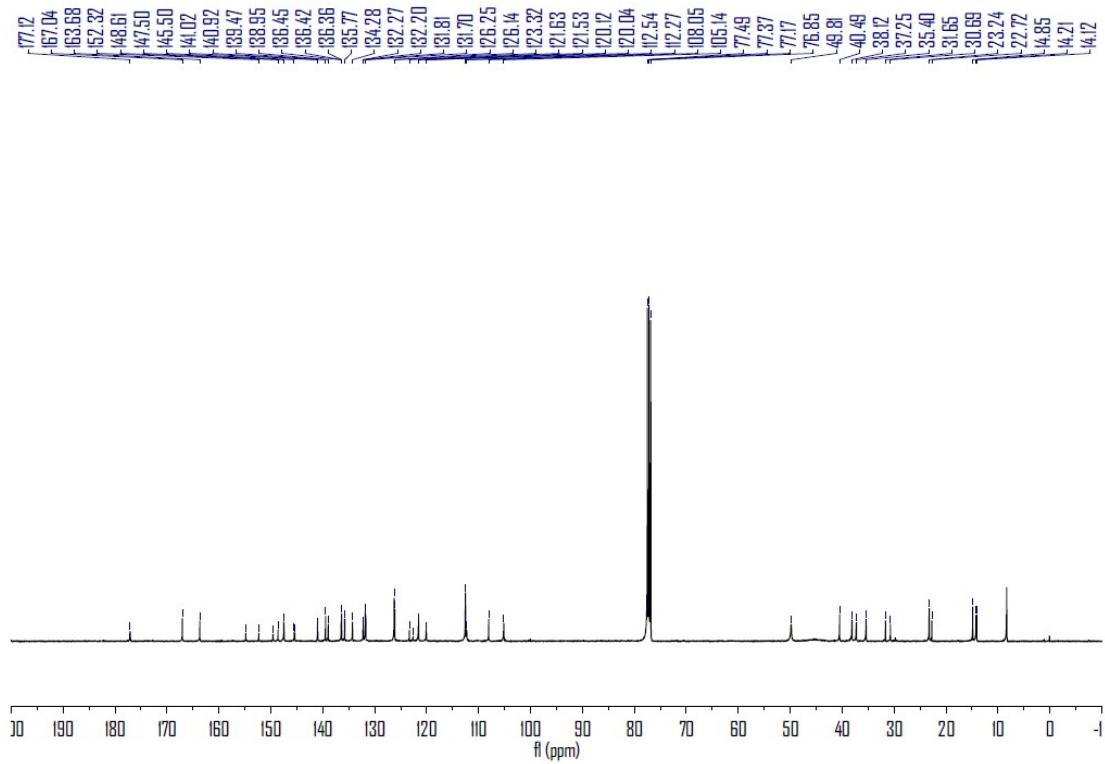


Fig. S42 100 MHz ^1H NMR spectrum of dithienylethene **4b** in CDCl_3 at room temperature.

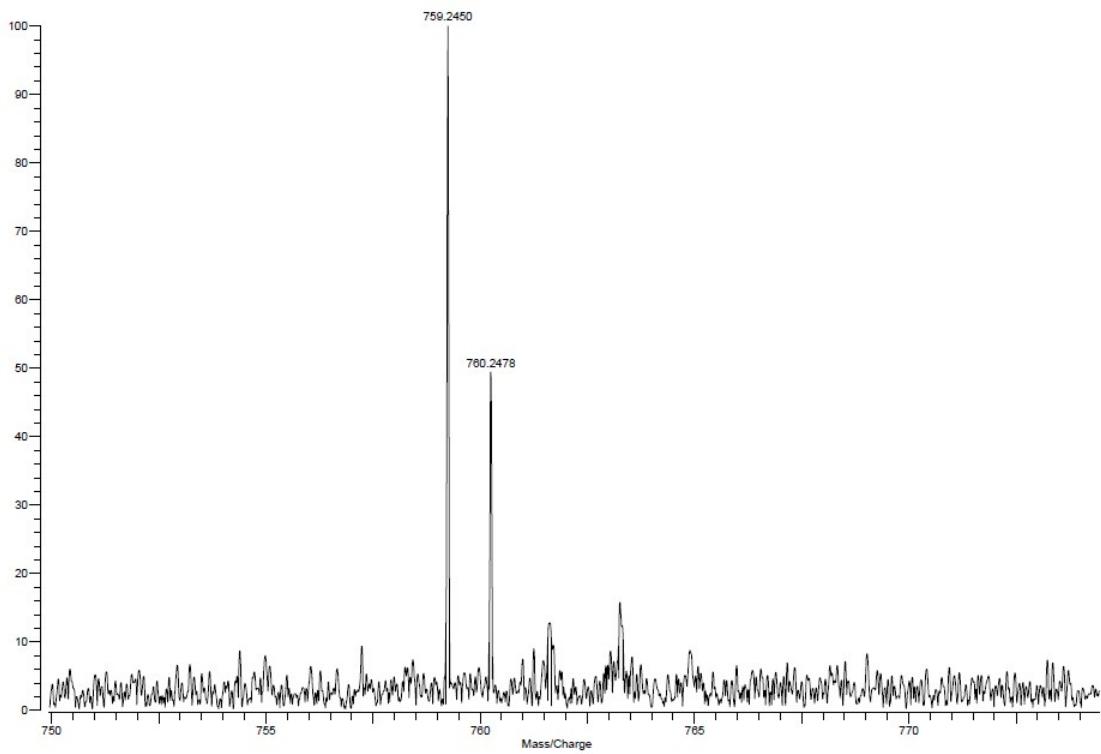


Fig. S43 HRMS of dithienylethene **4b**.

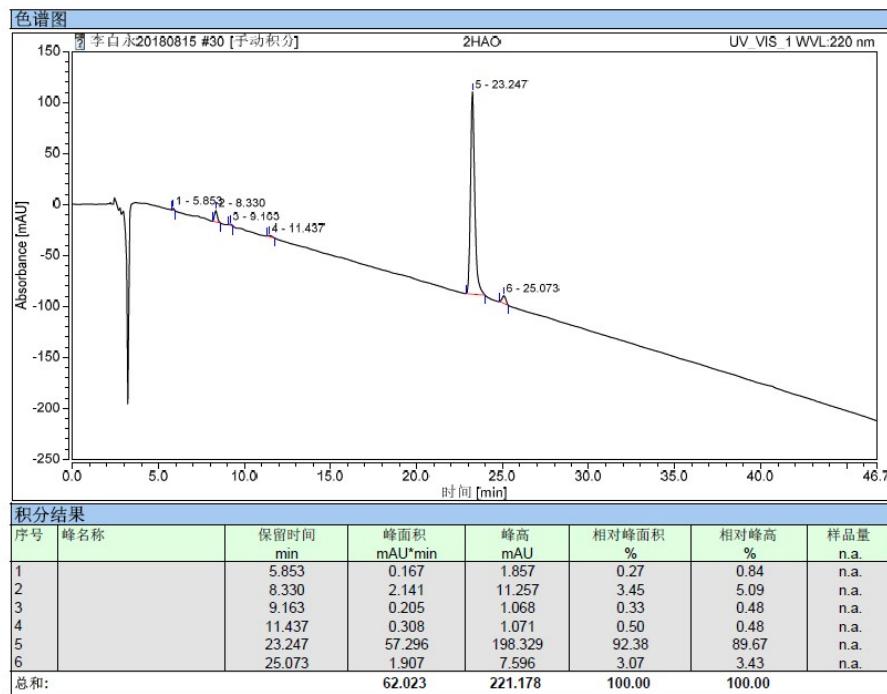


Fig. S44 HPLC of dithienylethene 1a.

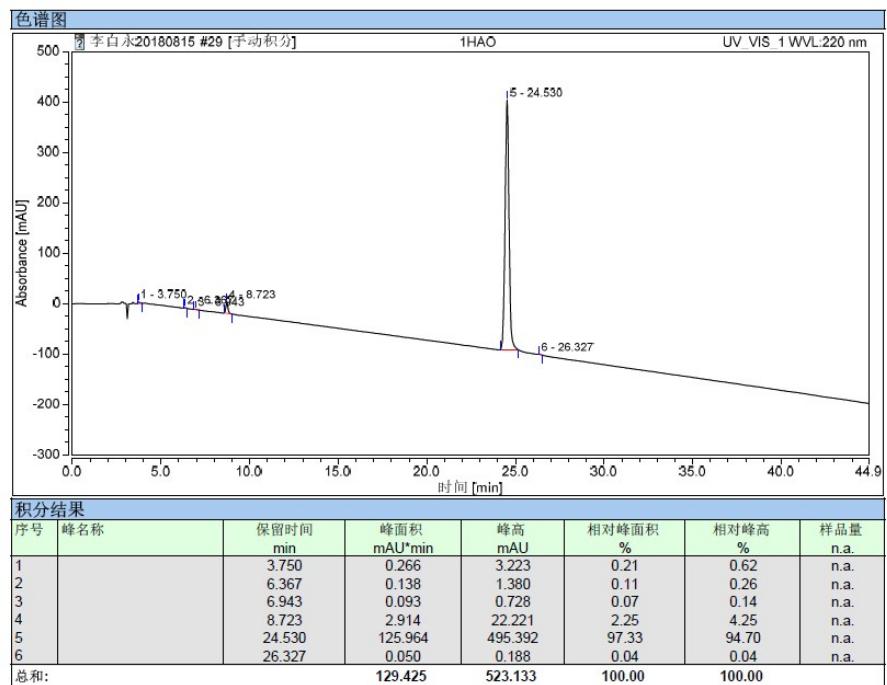


Fig. S45 HPLC of dithienylethene 1b.

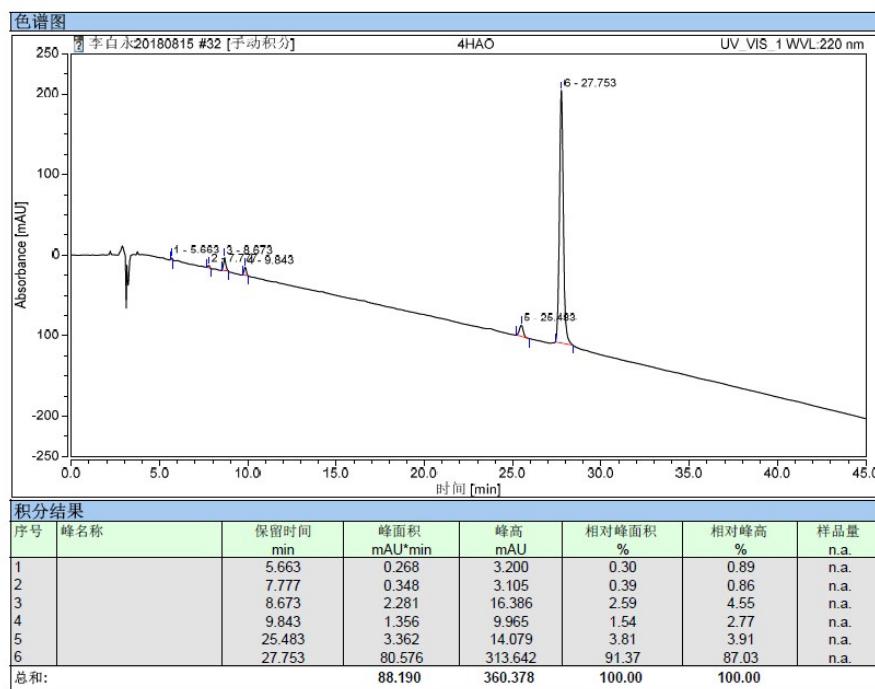


Fig. S46 HPLC of dithienylethene 2a.

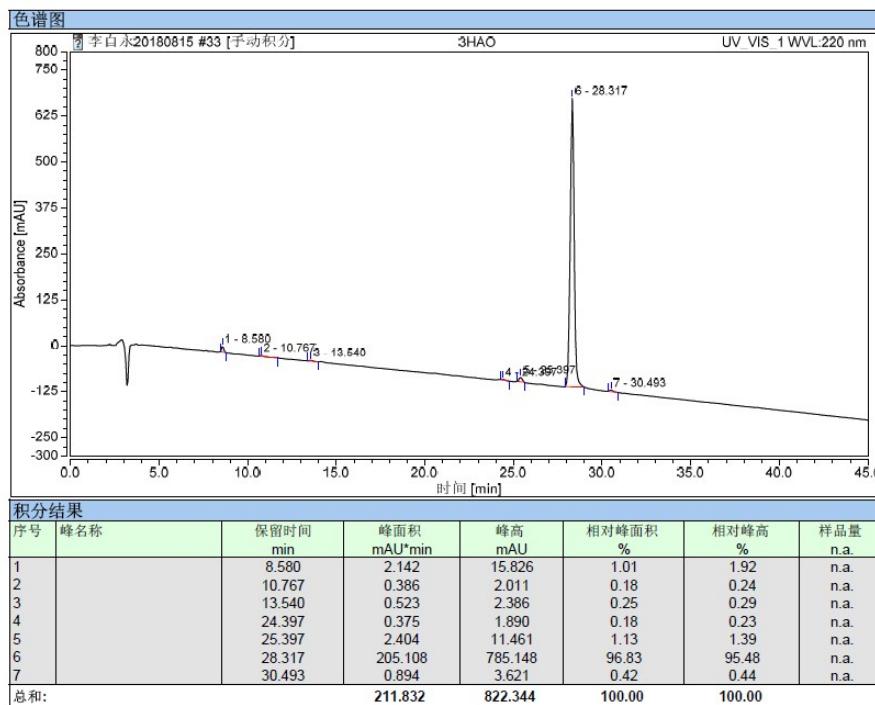


Fig. S47 HPLC of dithienylethene 2b.

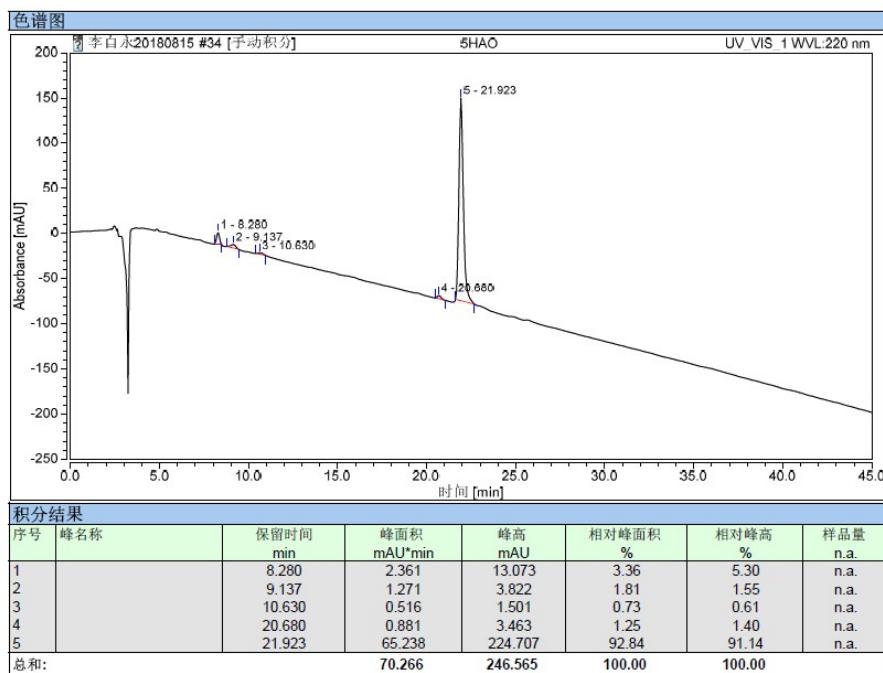


Fig. S48 HPLC of dithienylethene 3a.

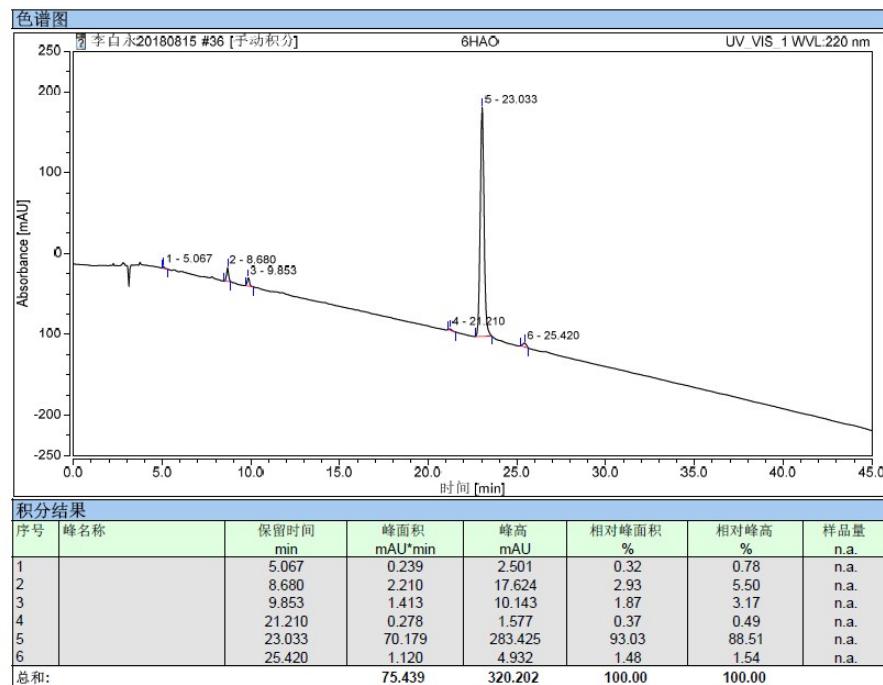


Fig. S49 HPLC of dithienylethene 3b.

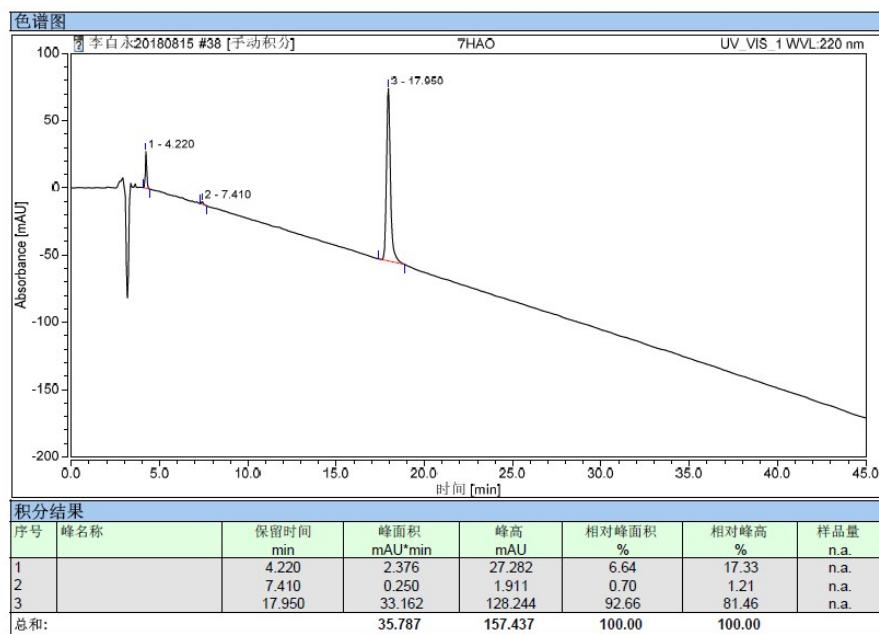


Fig. S50 HPLC of dithienylethene 4a.

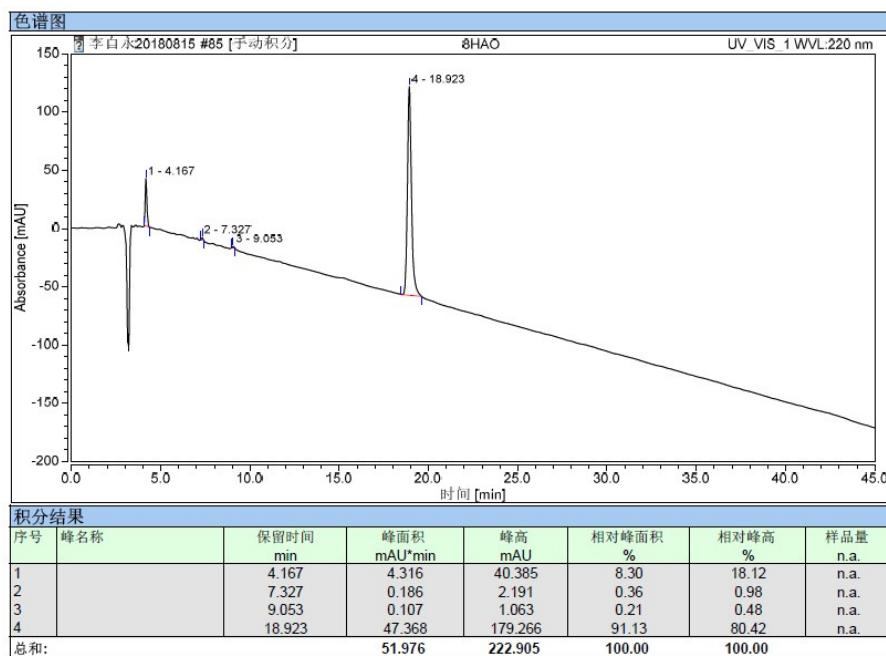


Fig. S51 HPLC of dithienylethene 4b.