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### **Supporting Information**

# Nickel-Catalyzed Regioselective Arylation of Aromatic amides with Aryl Iodides Enabled by *N,O*-Bidendate Directing Group

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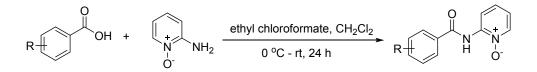
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#### **1. Experimental Section**

All of 2-aminopyridine 1-oxide was synthesized according to the known method [R. Adams, S. Miyano, *J. Am. Chem. Soc.* **1954**, *76*, 2785]. The aromatic amides were synthetized from the corresponding aromatic acids and 2-aminopyridine 1-oxide (see the following experimental procedure for detail).

#### 1.1 General Procedure for the Preparation of Amide Substrates



To a 100 mL reaction flask in ice-water bath were added  $CH_2Cl_2$  (30 mL), aromatic acid (17.5 mmol), NEt<sub>3</sub> (3.14 g, 35.0 mmol), Ethyl chloroformate (1.90 g, 17.5 mmol), 2-aminopyridine 1-oxide (1.00 g, 10.0 mmol). The reaction mixture was gradually warmed to room temperature, and stirred for 24 hours. The reaction was quenched with water (50 mL) and extracted with  $CH_2Cl_2$  (3 × 20 mL). The mixture was dried by MgSO<sub>4</sub> and filtered and concentrated under reduced pressure to yield a crude product, which was purified by flash chromatography (silica gel, ethyl acetate / ethanol = 6:1 to 1:1), affording the desired product as a white to yellow solid.

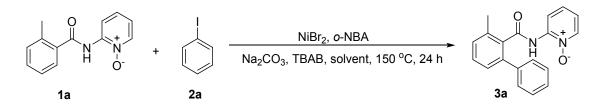
# 1.2 Optimization of reaction conditions

	+	NiBr <sub>2</sub> , <i>o</i> -NBA base, TBAB, toluene,150 °C, 24 h	
1a	2a		3a
entry		base	yield [%]b
1		Na <sub>2</sub> CO <sub>3</sub>	71
2		K <sub>2</sub> CO <sub>3</sub>	45
3		Cs <sub>2</sub> CO <sub>3</sub>	60
4		KHCO3	43
5		NaHCO <sub>3</sub>	40
6		CH <sub>3</sub> COONa	45
7		K <sub>3</sub> PO <sub>4</sub>	35
8		'BuOLi	n.r.
9		KF	n.r.
10		pyridine	n.r.

## **1.2.1** Optimization of the base on the model reaction<sup>*a*</sup>

<sup>*a*</sup>Conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), NiBr<sub>2</sub> (0.04 mmol, 20 mol%), *o*-NBA (0.08 mmol, 40 mol%), base (0.4 mmol, 2.0 equiv), TBAB (0.8 mmol, 4.0 equiv), toluene (1.5 mL), 150 °C, N<sub>2</sub> atmosphere, 24 h. <sup>*b*</sup>Isolated yields. *o*-NBA = *o*-nitrobenzoic acid. n.r. = no reaction.

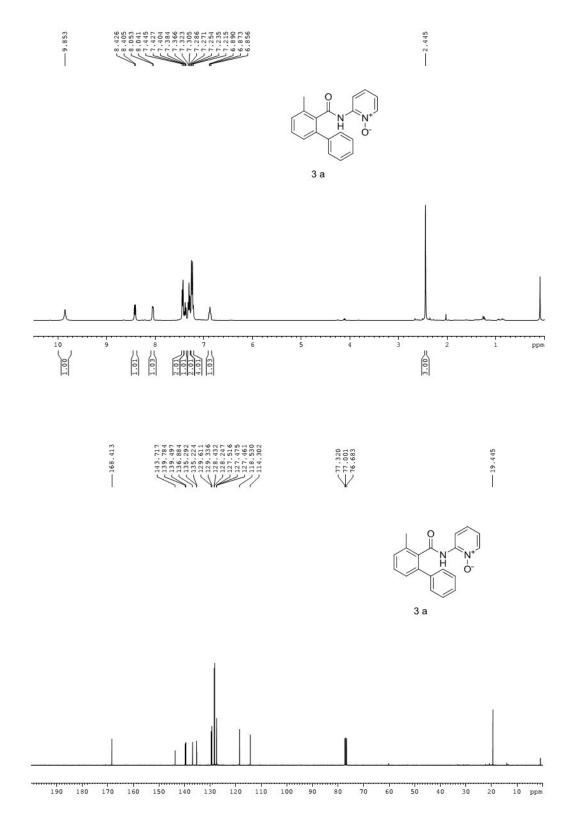
#### 1.2.2 Optimization of the solvent on the model reaction<sup>a</sup>

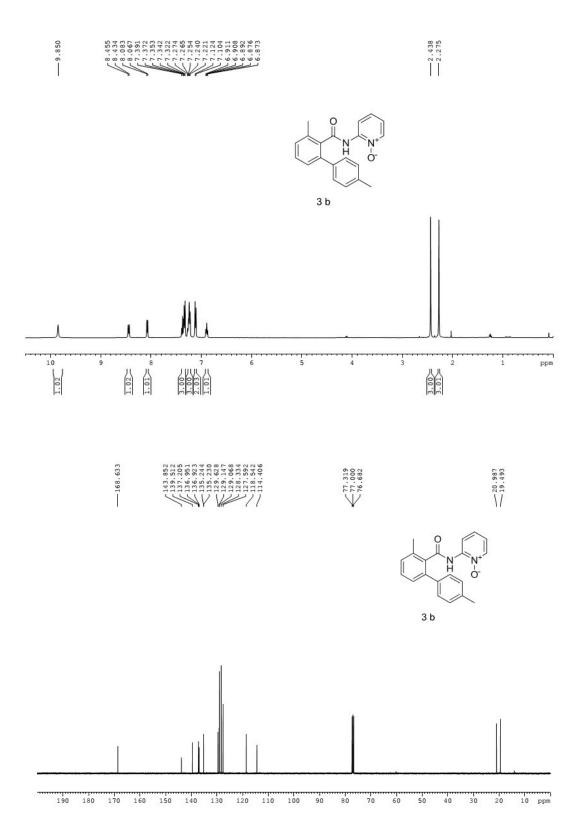


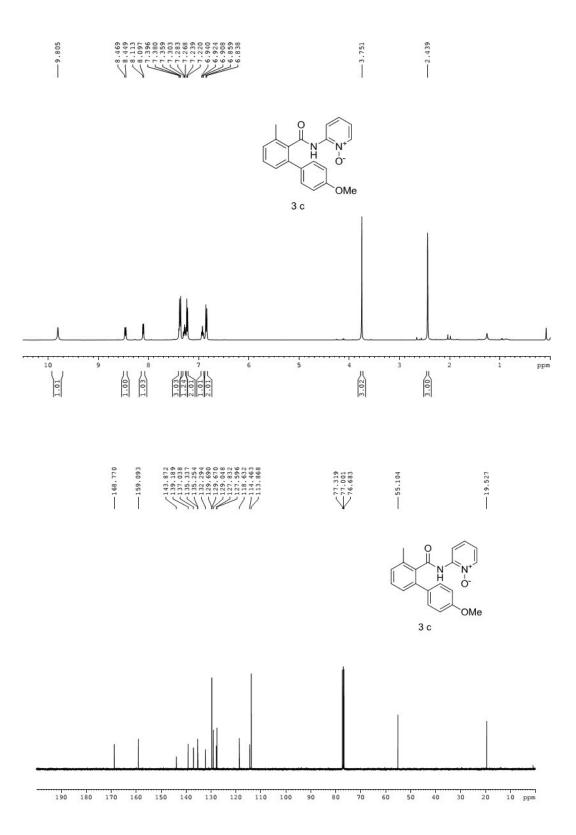
entry	solvent	yield [%] <sup>b</sup>
1	toluene	71
2	dioxane	59
3	THF	44
4	CH <sub>3</sub> CH <sub>2</sub> OH	36
5	PhCF <sub>3</sub>	55
6	DCE	40
7	DMF	42
8	DMA	37
9	DMSO	34
10	NMP	trace

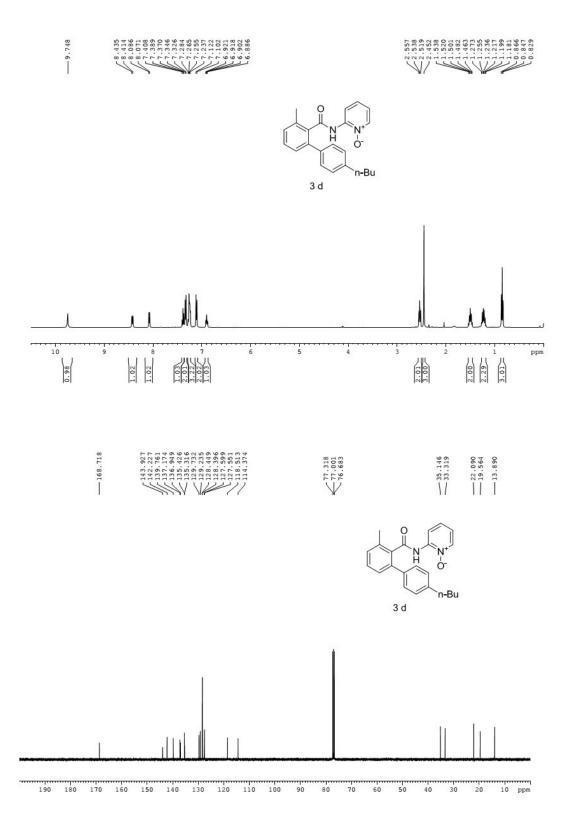
<sup>a</sup>Conditions: **1a** (0.2 mmol), **2a** (0.6 mmol), NiBr<sub>2</sub> (0.04 mmol, 20 mol%), *o*-NBA (0.08 mmol, 40 mol%), Na<sub>2</sub>CO<sub>3</sub> (0.4 mmol, 2.0 equiv), TBAB (0.8 mmol, 4.0 equiv), solvent (1.5 mL), 150 °C, N<sub>2</sub> atmosphere, 24 h. <sup>*b*</sup>Isolated yields. *o*-NBA = *o*-nitrobenzoic acid. DCE = 1,2-dichloroethane. DMA = Dimethylacetamide. NMP = 1-Methyl-2-pyrrolidinone.

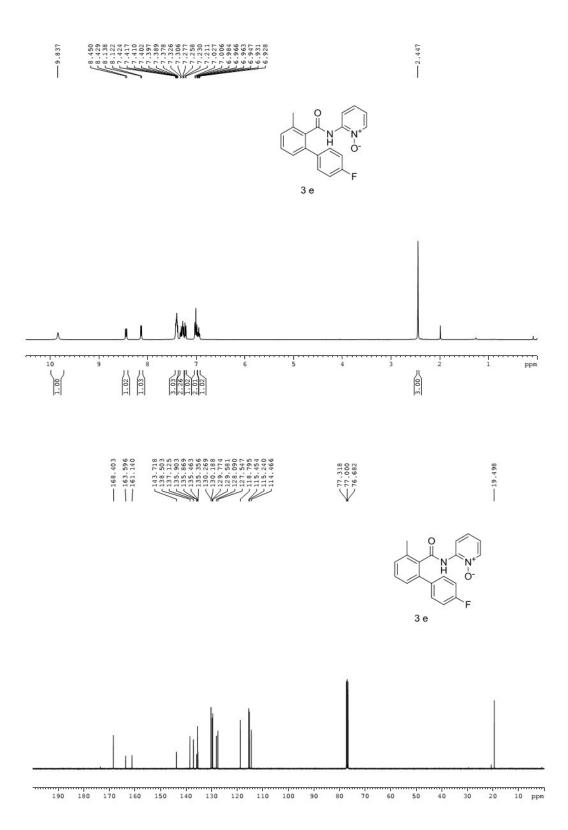
# 2. <sup>1</sup>H and <sup>13</sup>C NMR spectra

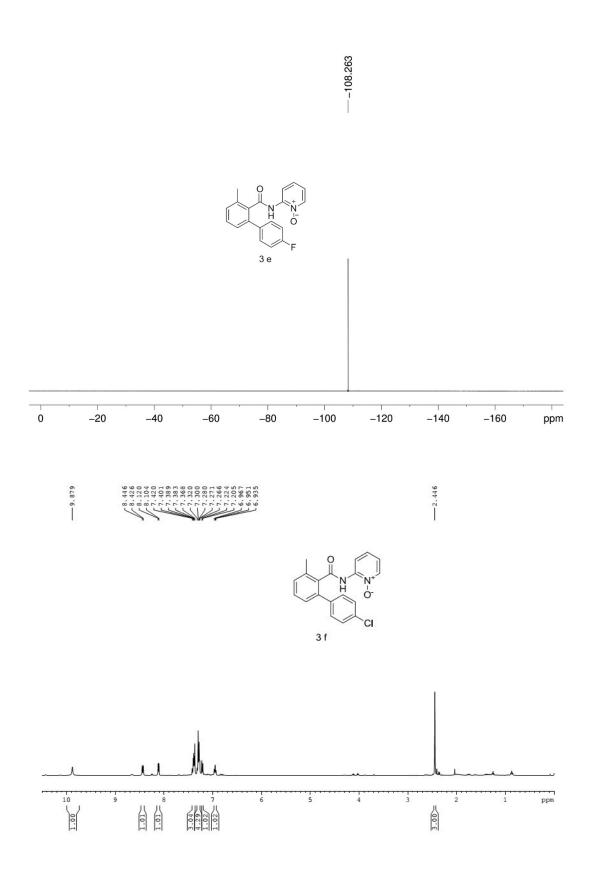


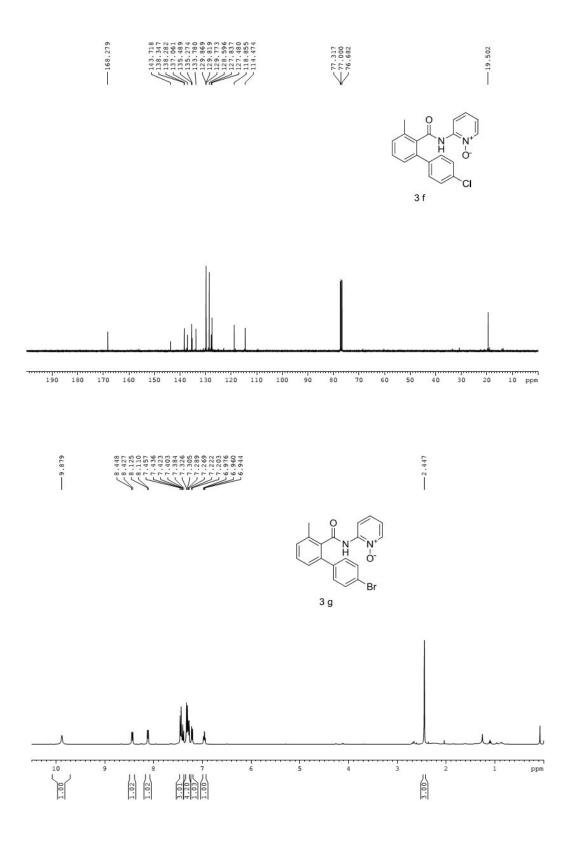


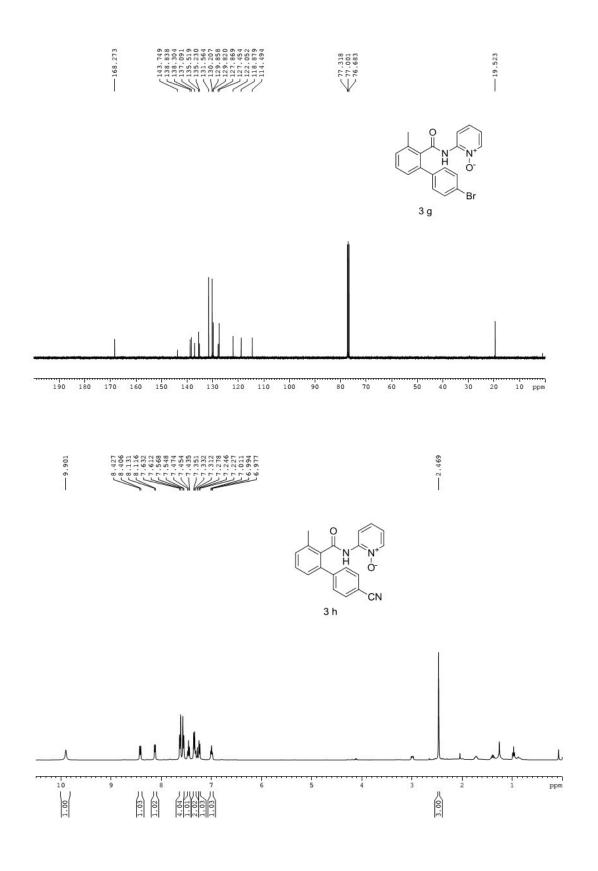


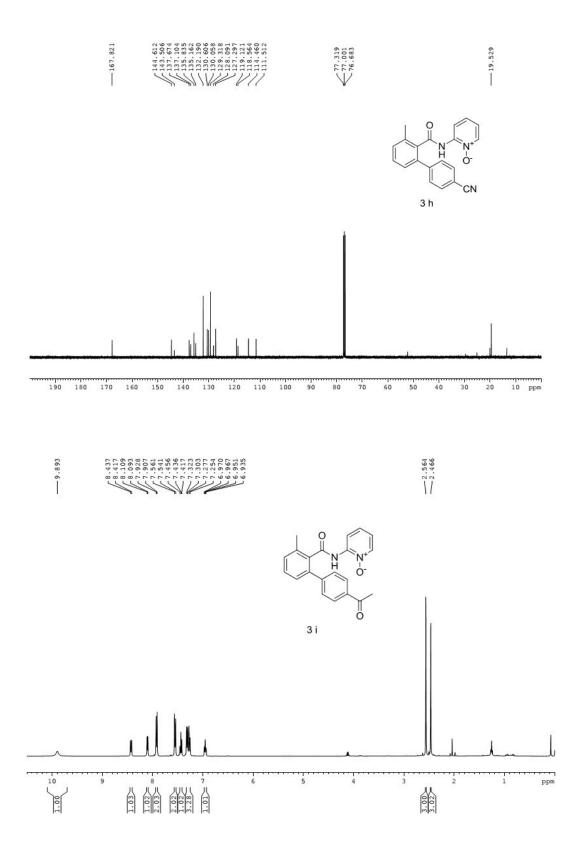


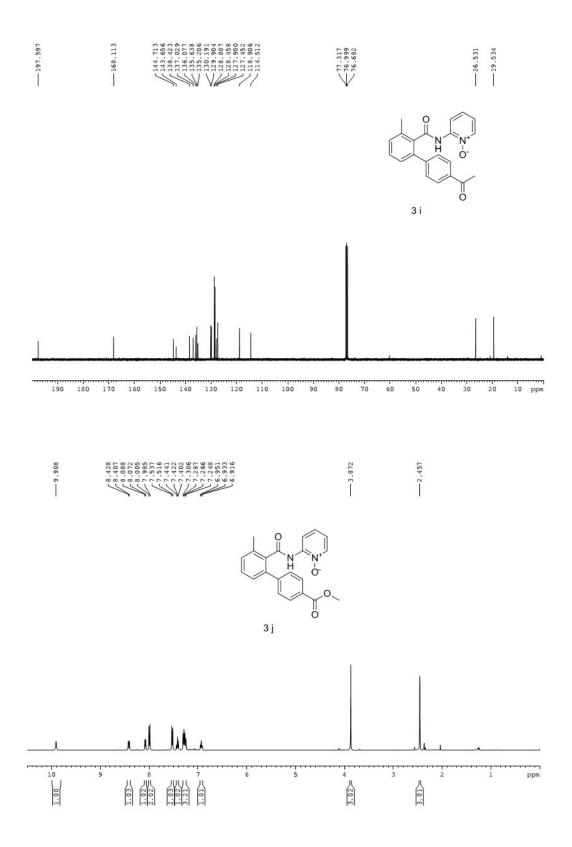


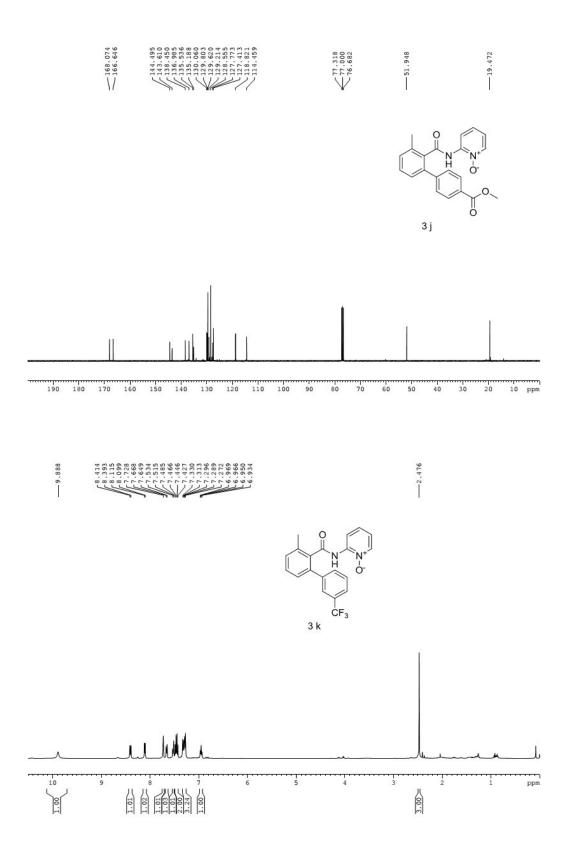


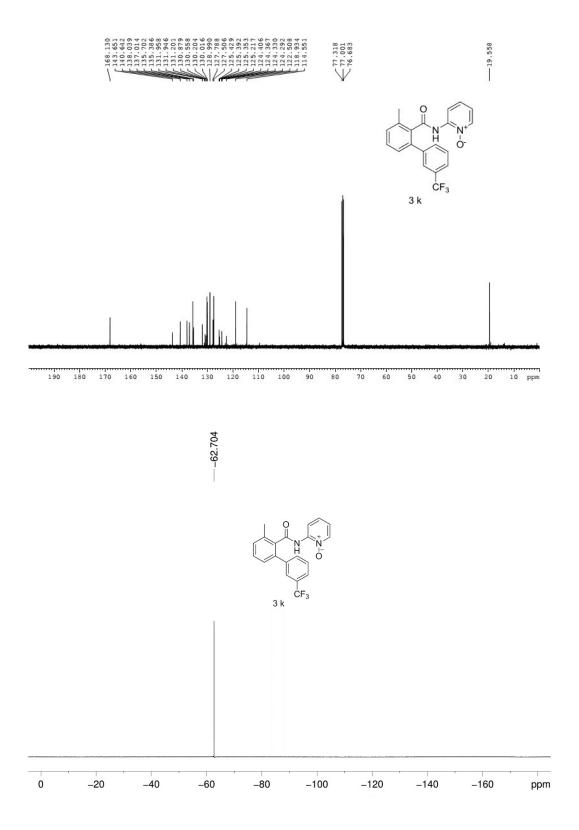


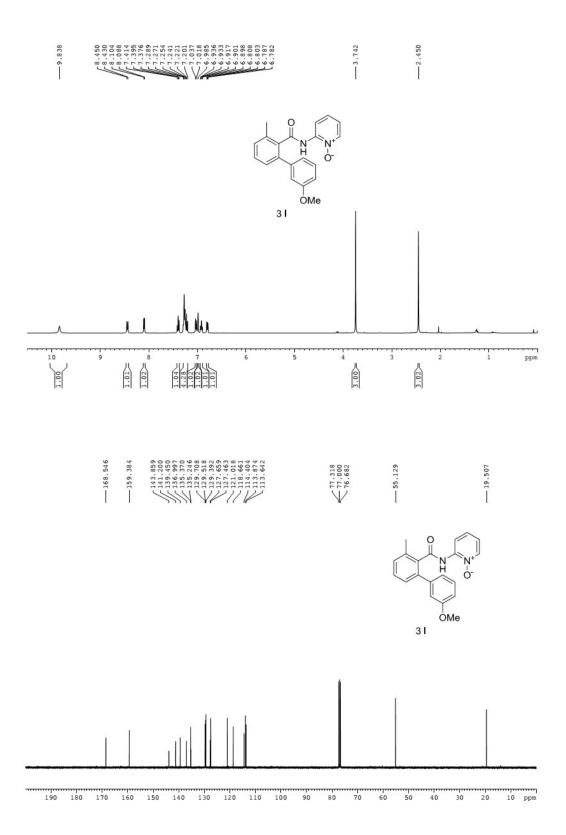


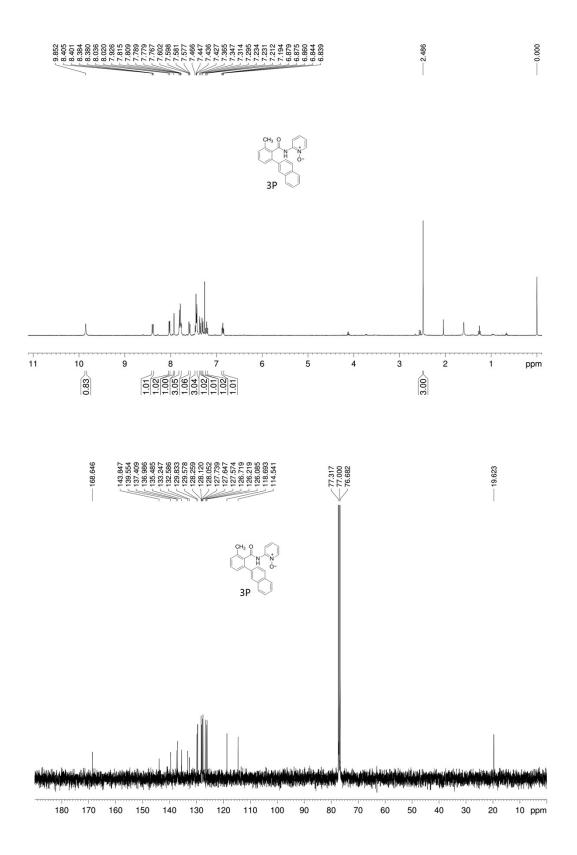


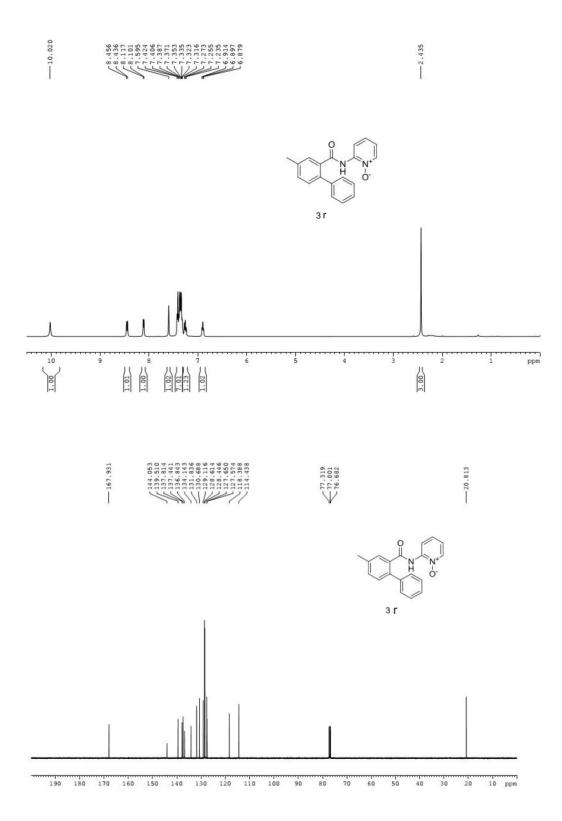


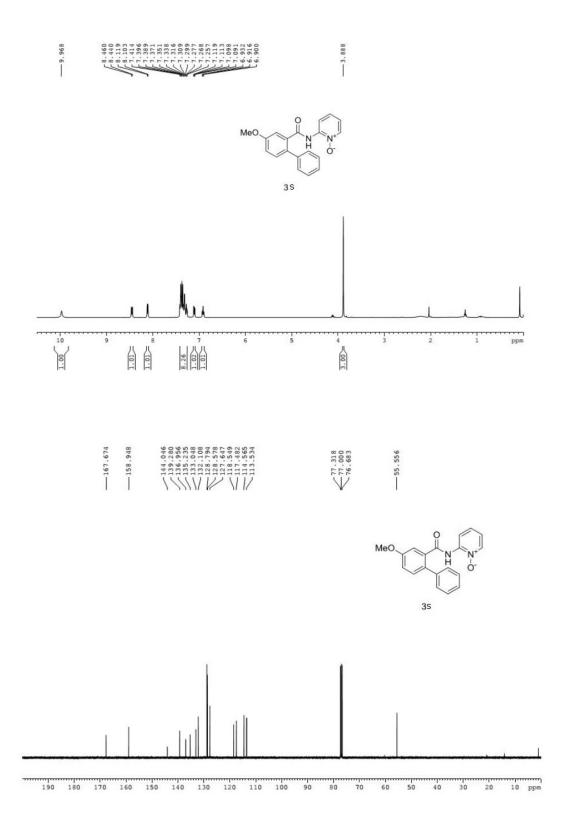


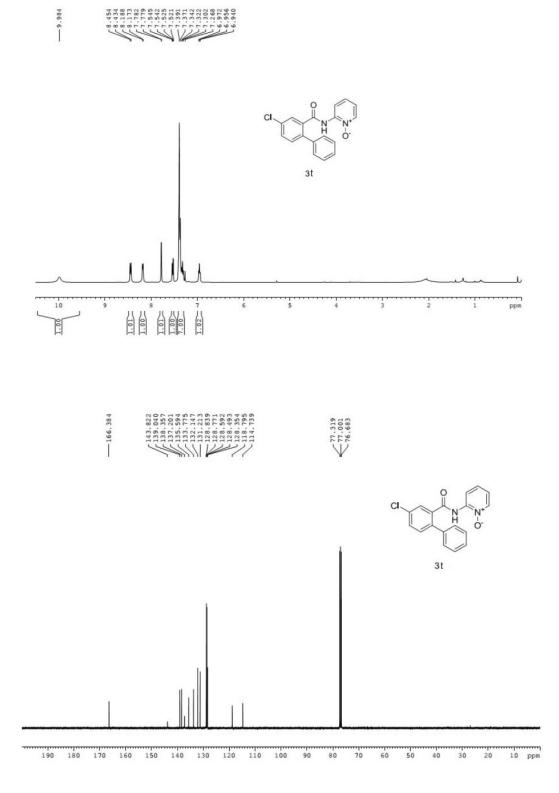


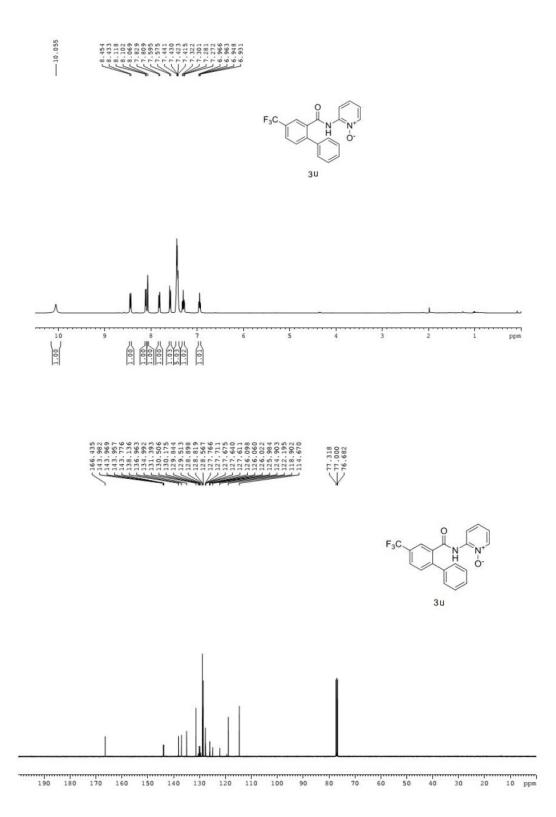












3 u 0 -20 -40 -60 -80 -100 -120 -140 -160 ppm -2.677 459 440 359 L6.953 L6.937 L6.933 3V M J.L

23

5

3

3.00

4

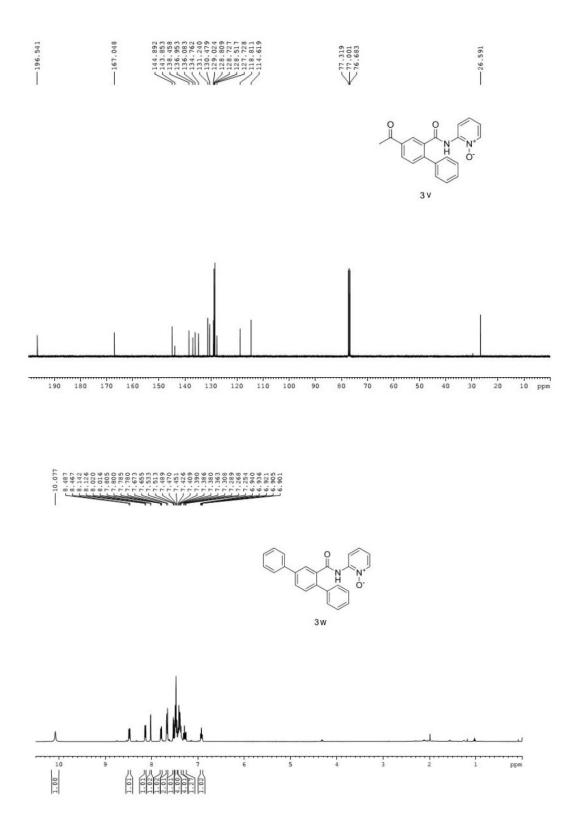
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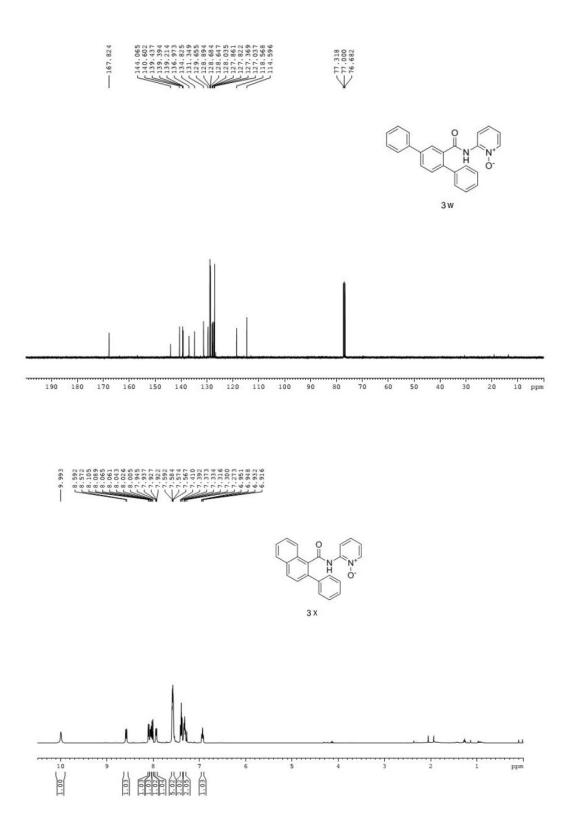
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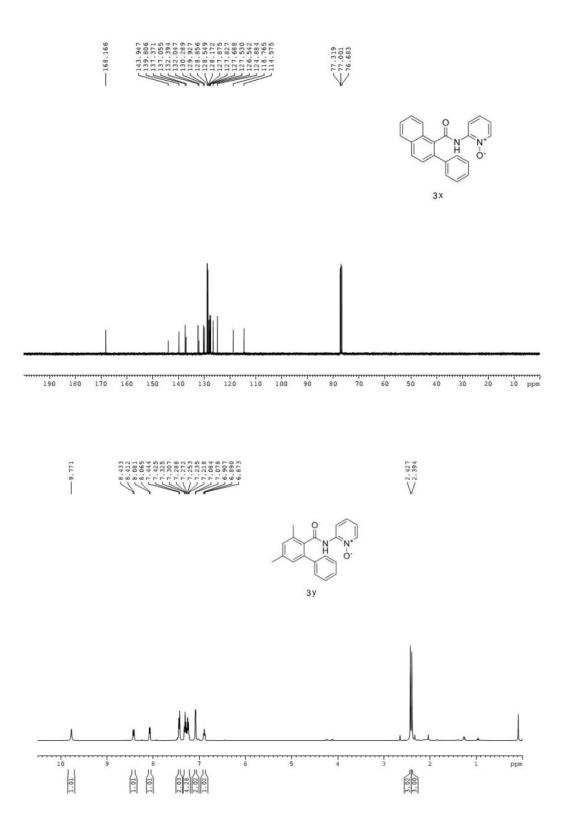
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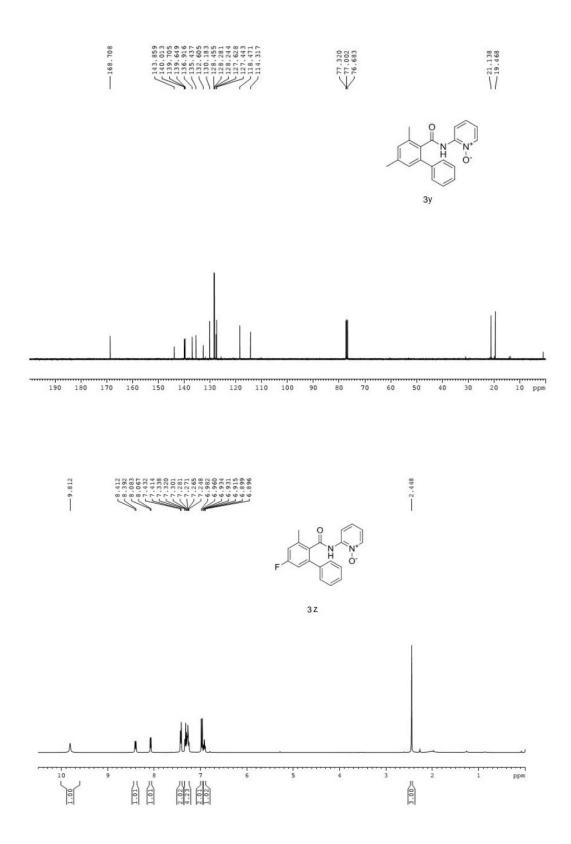
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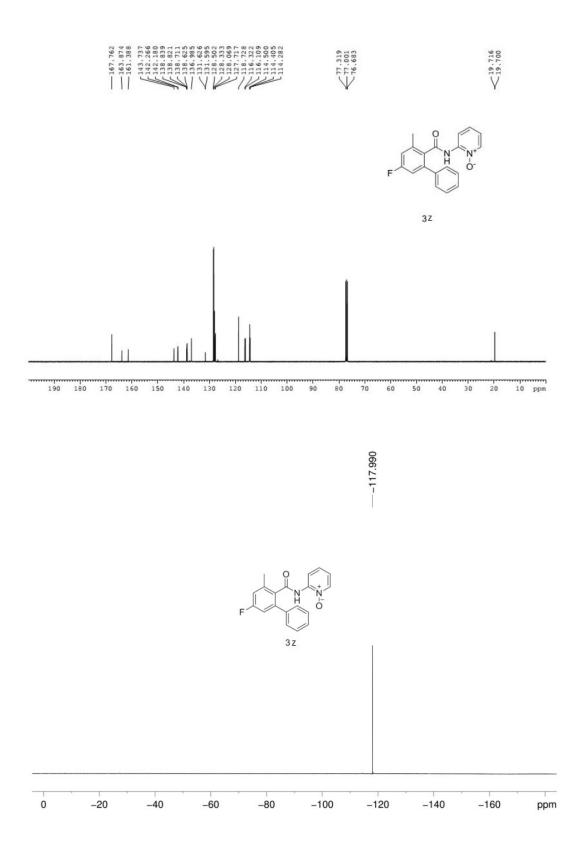
1.02 2.01 1.02 5.01 1.30 1.30 2.01



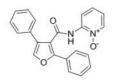


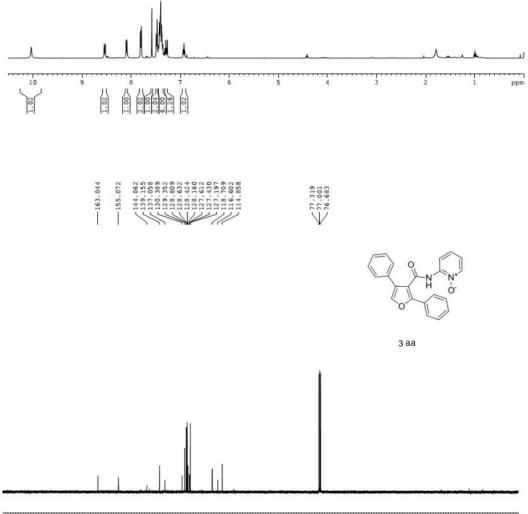












190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 ppm

