### Phenanthrenequinone enantiomers with cytotoxic activities from the tubers of Pleione

### bulbocodioides

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# **Computational methods**

#### Conformational analysis

Conformational analysis was initially performed using Confab [1] with systematic search at MMFF94 force field for the configurations of compounds 1 and 4 (Figure S). Room-temperature equilibrium populations were calculated according to Boltzmann distribution law (1). Conformers with population over 1% were chosen for subsequent Quantum Mechanics (QM) calculations. The energies and populations of dominative conformers were provided in Table S1.

$$\frac{N_{i}}{N} = \frac{g_{i}e^{-\frac{E_{i}}{k_{\rm B}T}}}{\sum g_{i}e^{-\frac{E_{i}}{k_{\rm B}T}}}$$
(1)

where  $N_i$  is the number of conformer i with energy  $E_i$  and degeneracy  $g_i$  at temperature T, and  $k_{\rm B}$  is Boltzmann constant.



Figure S1. Chemical structure of configurations of compounds 1 and 4.

### ECD calculation

The theoretical calculations were carried out using Gaussian 09 [2]. At first, conformers were optimized at PM6 using semi-empirical theory method. The conformers with Boltzmann-population over 1% were chosen for further optimization at B3LYP/6-311G(d,p) in methanol using the

IEFPCM model. Vibrational frequency analysis confirmed the stable structures. Under the same condition, the ECD calculation was conducted using Time-dependent Density functional theory (TD-DFT). Rotatory strengths for a total of 30 excited states were calculated. The ECD spectrum was simulated in SpecDis [3] by overlapping Gaussian functions for each transition according to (2).

$$\Delta \varepsilon(E) = \frac{1}{2.297 \times 10^{-39}} \times \frac{1}{\sqrt{2\pi\sigma}} \sum_{i}^{A} \Delta E_{i} R_{i} e^{-\left(\frac{E-E_{i}}{2\sigma}\right)^{2}}$$
(2)

where  $\sigma$  represents the width of the band at 1/e height, and  $\Delta E_i$  and  $R_i$  are the excitation energies and rotatory strengths for transition *i*, respectively.

Parameters of $\sigma$ and UV	-shift for compoun	ds $1$ and $4$ we	ere list as follows.
	Configuration	$\sigma$ (eV)	UV-shift (nm)

O(ev) = O(ev)	u (iiiii)
1R 0.37 5	
18 0.37 5	
4R 0.40 -6	)
4S 0.40 -6	)

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# **Energies and Coordinates**

Energies at MMFF94 force field

Systematic conformational search was performed by Confab program at MMFF94 force field. Conformers for each configuration were obtained with filtration by RMSD threshold of 0.5 Å.

Configuration	Conformer	Energy (kcal/mol)	Population (%)
1R	1	75.75	40.90
1R	2	75.92	30.84
1R	3	76.25	17.47
1R	4	76.61	9.53
1S	1	76.96	69.20
1S	2	77.85	15.37
1 <b>S</b>	3	77.98	12.24
1 <b>S</b>	4	79.42	1.08
1 <b>S</b>	5	79.44	1.04
4R	1	87.32	49.33
4R	2	88.02	15.22
4R	3	88.11	13.13
4R	4	88.45	7.32
4R	5	88.57	5.98
4R	6	89.02	2.81
4R	7	89.15	2.26
4R	8	89.41	1.46
4S	1	87.32	49.33
4S	2	88.02	15.22
4S	3	88.11	13.13
4S	4	88.45	7.32
4S	5	88.57	5.98
4S	6	89.02	2.81
4S	7	89.15	2.26
4S	8	89.41	1.46

Table S1 Energies of compounds 1 and 4 at MMFF94 force field.

# Energies at B3LYP theory level

Structures for ECD calculations were optimized at B3LYP/6-311G(d,p) in methanol.

Configuration	Conformer	Structure	E (Hartree)	E (kcal/mol)	Population (%)
1R	1		-1492.87879709	-936795.58	18.34
1R	3		-1492.88020715	-936796.47	81.66
15	1		-1492.88024126	-936796.49	50.90
1S	2		-1492.88020715	-936796.47	49.10
4R	1	A A	-1571.30245880	-986007.17	23.57
4R	2	A A	-1571.30320258	-986007.64	51.81
4R	3	A.	-1571.30170837	-986006.70	10.64

# Table S2 Energies of compounds 1 and 4 at B3LYP/6-311G(d,p) in methanol.

4R	5	the second	-1571.30196594	-986006.86	13.98
4S	1		-1571.30245880	-986007.17	23.57
4S	2		-1571.30320258	-986007.64	51.81
4S	3	-	-1571.30170837	-986006.70	10.64
4S	5		-1571.30196594	-986006.86	13.98





Figure S2. Chiral HPLC analysis of compounds 1a and 1b using *n*-hexane–isopropanol (70:30).



Figure S3. Chiral HPLC analysis of compounds 2a and 2b using *n*-hexane–isopropanol (60:40).



Figure S4. Chiral HPLC analysis of compounds 3a and 3b using *n*-hexane–isopropanol (70:30).



Figure S5. Chiral HPLC analysis of compounds 4a and 4b using *n*-hexane–isopropanol (60:40).

## NMR spectra of compounds 1-4.



Figure S6. The UV spectrum of compound 1.



Figure S7. The IR spectrum of compound 1.

Abundance 1627958.8 C25 H21 ( 433.12 
 Calc Mass
 Diff (ppm)
 Abs Diff (ppm)
 Abund Match
 Spacing Mat
 Mass I

 434.1379
 2.07
 2.07
 95.9
 99.58
Match 99.86 m/z 433.129 Calc m/z 433.1306 Ion For Score Cross S Mass 434.137 ula (M tula C26 H17 N4 O C26 H18 N4 O 98.66 433.1293 97.96 434.137 434.1366 0.98 93.16 99.73 99.97 433.1296 C25 H21 07 i7 C25 H22 O7

MS Formula Results: - Scan (5.867 min) Sub (2014051403.d)

Figure S8. The HRESIMS of compound 1.



Figure S9. The <sup>1</sup>H NMR spectrum of compound 1 methanol- $d_4$  (600 MHz).



Figure S10. The <sup>13</sup>C NMR spectrum of compound 1 methanol-*d*<sub>4</sub> (150 MHz).



Figure S11. The DEPT spectrum of compound 1 methanol- $d_4$  (150 MHz).



Figure S12. The HSQC spectrum of compound 1 methanol- $d_4$  (600 MHz).



Figure S13. The HMBC spectrum of compound 1 methanol- $d_4$  (600 MHz).



Figure S14. The NOESY spectrum of compound 1 methanol- $d_4$  (600 MHz).



Figure S15. The <sup>1</sup>H-<sup>1</sup>H COSY spectrum of compound **1** methanol- $d_4$  (600 MHz).



Figure S16. The ECD spectrum of compound 1a.



Figure S17. The ECD spectrum of compound 1b.



Figure S18. The UV spectrum of compound **2**.



Figure S19 The IR spectrum of compound 2.

MS Formula Results: - Scan (6.598 min) Sub (2014051601.d)

[	m/z 2	ion	Formula	Abundance											
÷[	539.1718	(M-H)-	C32 H27 O8	2481207											
	Best	Formula (M)	Ion Formula	Calc m/z	Score V	Cross S	Mass	Calc Mass	Diff (ppm)	Abs Diff (ppm)	Abund Match	Spacing Mat	Mass Match	m/z	DBE
	Г	C33 H24 N4 O4	C33 H23 N4 04	539.1725	99.94		540.1791	540.1798	1.19	1.19	99.9	99.95	99.95	539.1718	24
	<b>P</b>	C32 H28 O8	C32 H27 08	539.1711	99.87	34.8	540.1791	540.1784	-1.26	1.26	99.64	99.97	99.94	539.1718	19

Figure S20. The HRESIMS of compound **2**.



Figure S21. The <sup>1</sup>H NMR spectrum of compound **2** in methanol- $d_4$  (600 MHz).



Figure S22. The <sup>13</sup>C NMR spectrum of compound **2** in methanol- $d_4$  (150 MHz).



Figure S23. The DEPT spectrum of compound 2 in methanol- $d_4$  (150 MHz).



Figure S24. The HSQC spectrum of compound **2** in methanol- $d_4$  (600 MHz).



Figure S25. The HMBC spectrum of compound **2** in methanol- $d_4$  (600 MHz).



Figure S26. The NOESY spectrum of compound **2** in methanol-*d*<sub>4</sub> (600 MHz).



Figure S27. The  ${}^{1}\text{H}{}^{-1}\text{H}$  COSY spectrum of compound **2** methanol- $d_{4}$  (600 MHz).



Figure S28. The online mirror-like ECD spectrum of compounds 2a (blue) and 2b (green).



Figure S29. The ECD spectrum of compound 2b.



Figure S30. The UV spectrum of compound **3**.



Figure S31. The IR spectrum of compound **3**.

MS Formula Results: - Scan (6.245 min) Sub (2014051404.d)

- 1	m/z /	lon	Formula	Abundarice												
8	483.1455	(M-H)-	C29 H23 07	213069.3												
	Best	Formula (M)	Ion Formula	Calc m/z	Score 4	Cross S	Mass	Calc Mass	Diff (ppm)	Abs Diff (ppm)	Abund Match	Spacing Mat	Mass Match	rn/z	00E	į
		C29 H24 07	C29 H23 07	483.1449	99.88	1.18	484.1527	484.1522	-1.05	1.05	99.72	99.9	99.95	483.1455	18	i
	* F	C30 H20 N4 O3	C30 H19 N4 O3	483.1463	99.55		484.1527	484.1535	1.7	1.7	98.62	99.95	99.9	483.1455	23	l
	- F	C26 H28 O7 S	C26 H27 07 S	483.1483	98.73		484.1527	484.1556	5.91	5.91	97.89	99.55	98.81	483.1455	13	i
	\$ T	C21 H28 N2 O9 S	C21 H27 N2 O9 5	483.1443	98.49		484.1527	484.1516	-2.41	2.41	95.49	\$9.47	99.8	483.1455	9	i
	* F	C30 H28 O2 S2	C30 H27 02 S2	483.1458	97.67		484.1527	484.1531	0.73	0.73	92.47	99.3	99.98	483.1455	17	l
	· / /	C17 H28 N2 O14	C17 H27 N2 014	483.1468	97.4		484.1527	484.1541	2.76	2.76	91.4	99.93	99.74	483.1455	5	i

Figure S32. The HRESIMS of compound **3**.



Figure S33. The <sup>1</sup>H NMR spectrum of compound **3** in methanol- $d_4$  (600 MHz).



Figure S34. The <sup>13</sup>C NMR spectrum of compound **3** in methanol- $d_4$  (150 MHz).



Figure S35. The DEPT spectrum of compound **3** in methanol- $d_4$  (150 MHz).



Figure S36. The HSQC spectrum of compound **3** in methanol- $d_4$  (600 MHz).



Figure S37. The HMBC spectrum of compound **3** in methanol- $d_4$  (600 MHz).



Figure S38. The NOESY spectrum of compound **3** in methanol- $d_4$  (600 MHz).



Figure S39. The  ${}^{1}\text{H}{}^{-1}\text{H}$  COSY spectrum of compound **3** methanol- $d_{4}$  (600 MHz).



Figure S40. The ECD spectrum of compound **3a**.



Figure S41. The ECD spectrum of compound **3b**.



Figure S42. The UV spectrum of compound 4.



Figure S43. The IR spectrum of compound 4.

MS Formula Results: - Scan (6.804 min) Sub (2014051406.d)

		m/z i	fon	Formula	Abundance											
3		467.1486	(M-H)-	C29 H23 O6	31105.7											
	Г	Best	Formula (M)	Ion Formula	Calc m/z	Score 5	Cross S	Mass	Calc Mass	Diff (ppm)	Abs Diff (ppm)	Abund Match	Spacing Mat.	Mass Match	m/z	300.
	÷Е	0.00	C29 H24 O6	C29 H23 O6	467.15	99.8	1.1.1.1	468.1558	458.1573	3.13	3,13	99.93	99.92	\$9.67	467.1486	18
	3E	Г	C30 H20 N4 O2	C30 H19 N4 O2	467.1513	99.33		468.1558	458.1586	5.96	5.96	99.73	99.93	8.82	467.1486	23
	10	Г	C33 H24 O S	C33 H23 O S	467.1475	99.23		468.1558	468.1548	-2.23	2.23	97.68	99.88	99.83	467.1486	22
	하는		C24 H24 N2 O8	C24 H23 N2 O8	467.146	95.87		468.1558	468.1533	-5.48	5.48	97.78	99.96	98.98	467.1486	14
	11	r -	C21 H28 N2 O8 S	C21 H27 N2 O8 S	457.1494	97.84		468.1558	468.1566	1.71	1.71	92.9	99.65	99.9	467.1486	9
	파	F	C30 H28 O S2	C30 H27 O 52	457,1509	97.52		468.1558	468.1582	4.97	4.97	93.14	99.5	99.16	467.1486	17
	۵Þ	<b></b>	C25 H28 N2 O3 52	C25 H27 N2 03 S2	457,1469	97.32		468.1558	468.1541	-3.64	3.64	91.92	99.34	99.55	457.1486	13

Figure S44. The HRESIMS of compound **4**.



Figure S45. The <sup>1</sup>H NMR spectrum of compound **4** in methanol- $d_4$  (600 MHz).



Figure S46. The  ${}^{13}$ C NMR spectrum of compound 4 in methanol- $d_4$  (150 MHz).



Figure S47. The DEPT spectrum of compound 4 in methanol- $d_4$  (150 MHz).



Figure S48. The HSQC spectrum of compound 4 in methanol- $d_4$  (600 MHz).



Figure S49. The HMBC spectrum of compound 4 in methanol- $d_4$  (600 MHz).



Figure S50. The NOESY spectrum of compound 4 in methanol-d<sub>4</sub> (600 MHz).



Figure S51. The <sup>1</sup>H-<sup>1</sup>H COSY spectrum of compound **4** methanol-*d*<sub>4</sub> (600 MHz).



Figure S52. The ECD spectrum of compound 4a.



Figure S53. The ECD spectrum of compound 4b.