Electronic Supplementary Material (ESI) for Photochemical & Photobiological Sciences. This journal is © The Royal Society of Chemistry and Owner Societies 2018

Supporting Information

Supporting Information

β-Diketone derivatives: influence of the chelating group on the photophysical and mechanofluorochromic properties

Marine Louis,^[a] Régis Guillot,^[b] Rémi Métivier*^[a] and Clémence Allain*^[a]

[a] PPSM, ENS Cachan, CNRS, Université Paris-Saclay, 94235 Cachan (France)

[b] ICMMO, Univ Paris-Sud, CNRS, Université Paris-Saclay, 91405 Orsay (France)

* Corresponding authors: metivier@ppsm.ens-cachan.fr, clemence.allain@ens-cachan.fr

 Table S1. Crystallographic data and structure refinement details.

Compounds	P-Ester
Empirical Formula	C ₁₈ H ₁₅ O ₅
M_r	312.31
Crystal size, mm ³	0.12 x 0.11 x 0.03
Crystal system	monoclinic
Space group	Сс
a, Å	34.531(3)
b, Å	7.3710(7)
c, Å	11.7082(9)
α, °	90
β, °	100.238(9)
γ, °	90
Cell volume, Å ³	2932.6(5)
Z ; Z'	8;2
Т, К	100(1)
Radiation type ; wavelength Å	ΜοΚα ; 0.71073
F ₀₀₀	1304
μ , mm ⁻¹	0.104
heta range, °	2.397 - 23.392
Reflection collected	41 501
Reflections unique	4 252
R _{int}	0.0721
GOF	1.071
Refl. obs. $(I \ge 2\sigma(I))$	3 592
Parameters	421
wR2 (all data)	0.1523
R value $(I > 2\sigma(I))$	0.0619
Largest diff. peak and hole (eÅ ⁻³)	0.443 ; -0.368

NMR spectra





Differential Scanning Calorimetry and Thermogravimetric Analysis

Differential scanning calorimetry and thermogravimetric analyses were performed on a Mettler Toledo TGA/DSC Star System, under air at 5 to 10°C / min.





Fluorescence decays

P-Ester in THF solution $\langle \tau \rangle$ = 0.19 ns at 430 nm



DPB-Ester in THF solution $\langle \tau \rangle = 0.92$ ns at 440 nm





P-Ester in the powder state $\langle \tau \rangle$ = 2.41 ns at 450 nm

DPB-Ester in the powder state $\langle \tau \rangle$ = 2.32 ns at 550 nm



Fluorescence excitation spectra



