

**Photocyclization of diarylethenes:
the effect of imidazole on the oxidative photodegradation process.**

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^a*N. D. Zelinsky Institute of Organic Chemistry, Russian Academy of Sciences, 47,
Leninsky prosp., 119991 Moscow, Russian Federation, e-mail: shir@ioc.ac.ru*

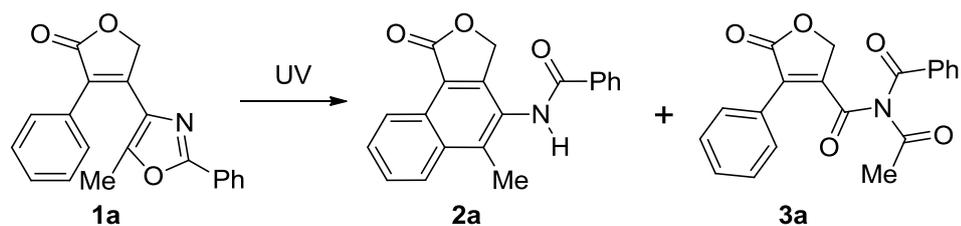
^b*Institute of Physical and Organic Chemistry, Southern Federal University, 194/2 Stachka
Avenue, Rostov on Don 344090, Russian Federation*

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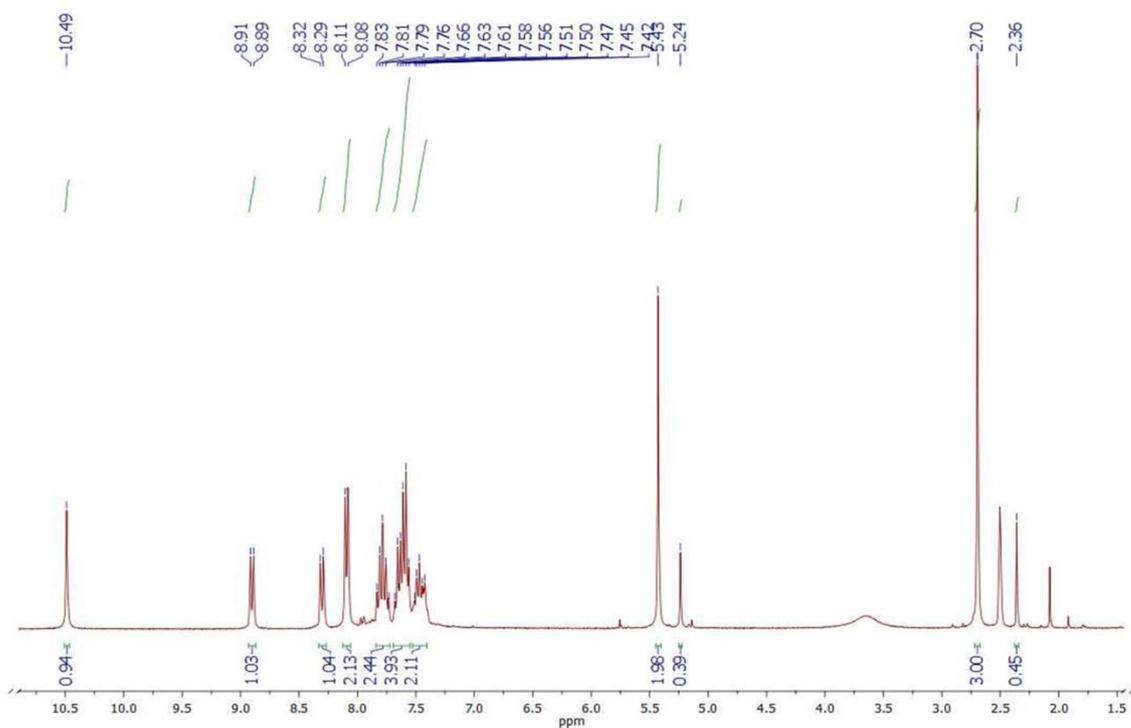
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I. Solvent effects (NMR spectra of reaction mixtures)

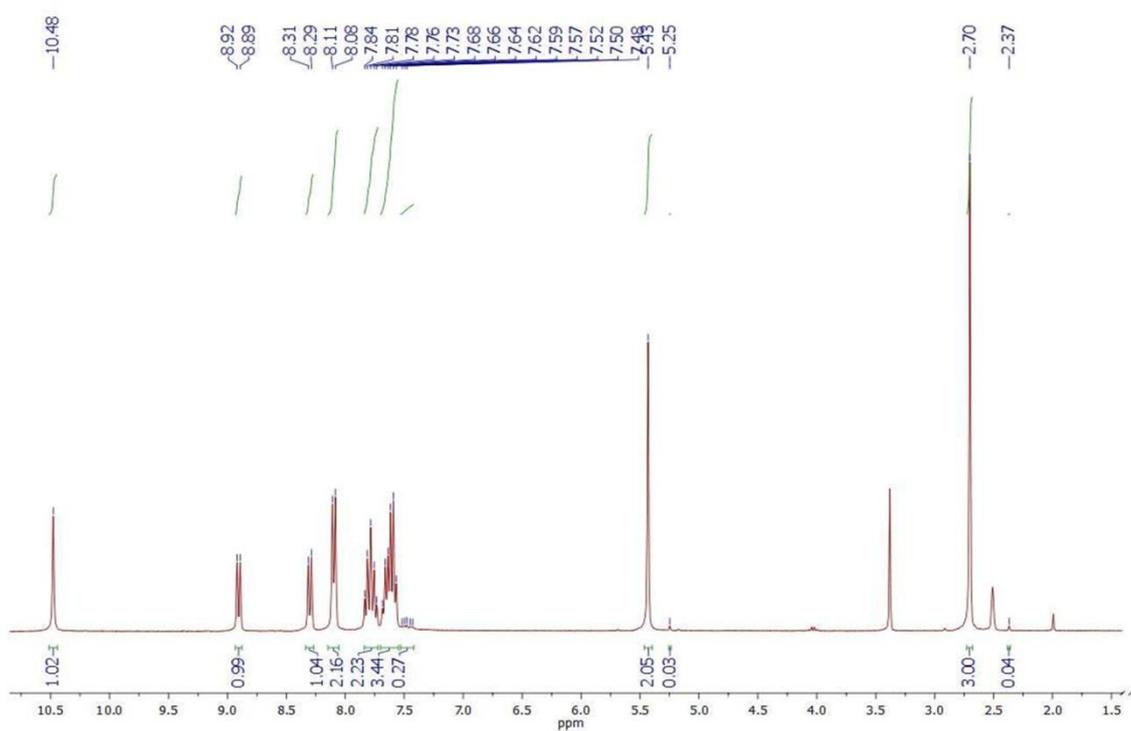
I.1. Diarylethene 1a



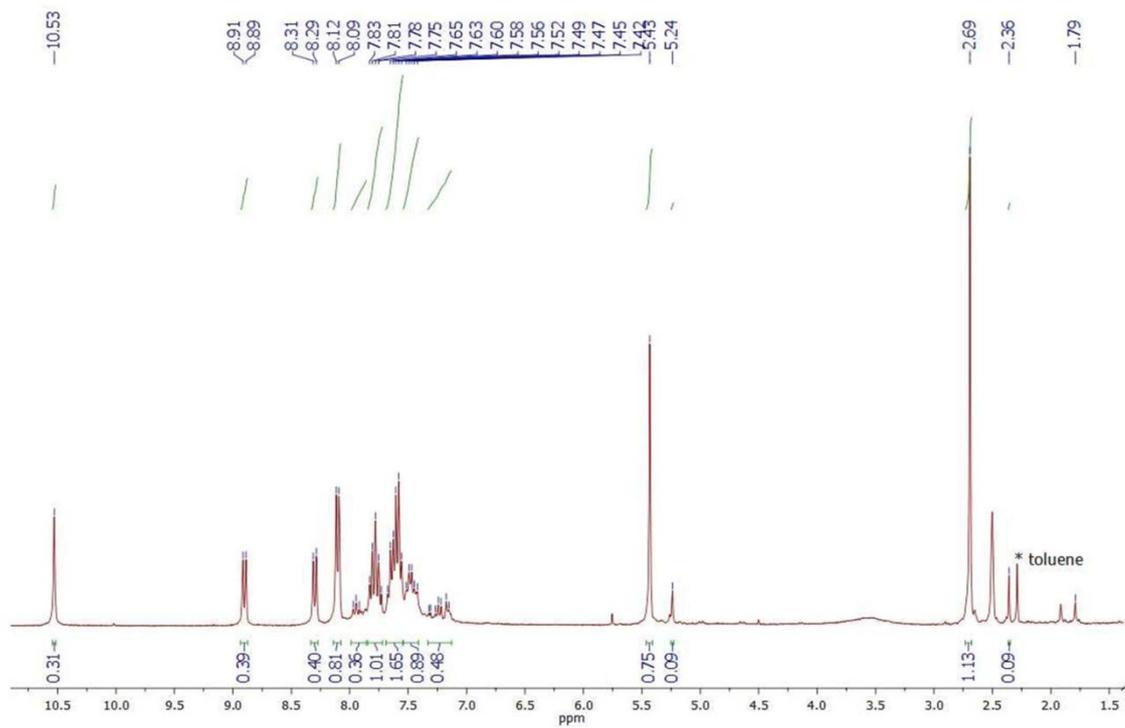
Solvent: MeCN



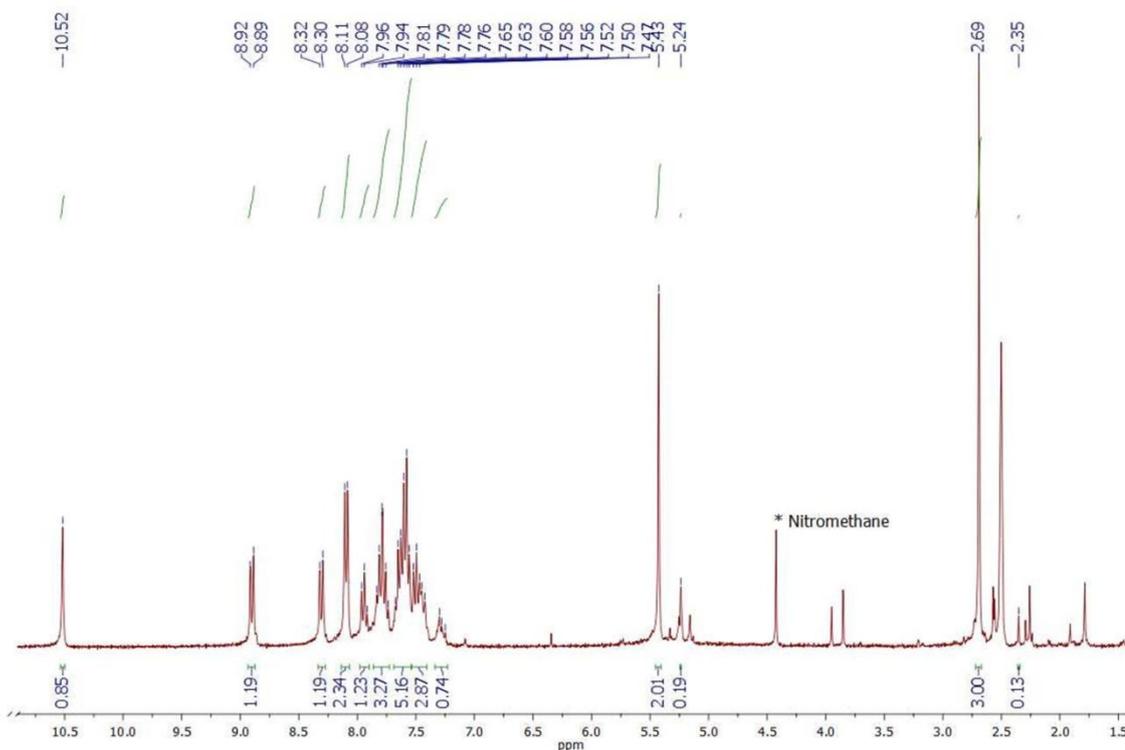
Solvent: DMF



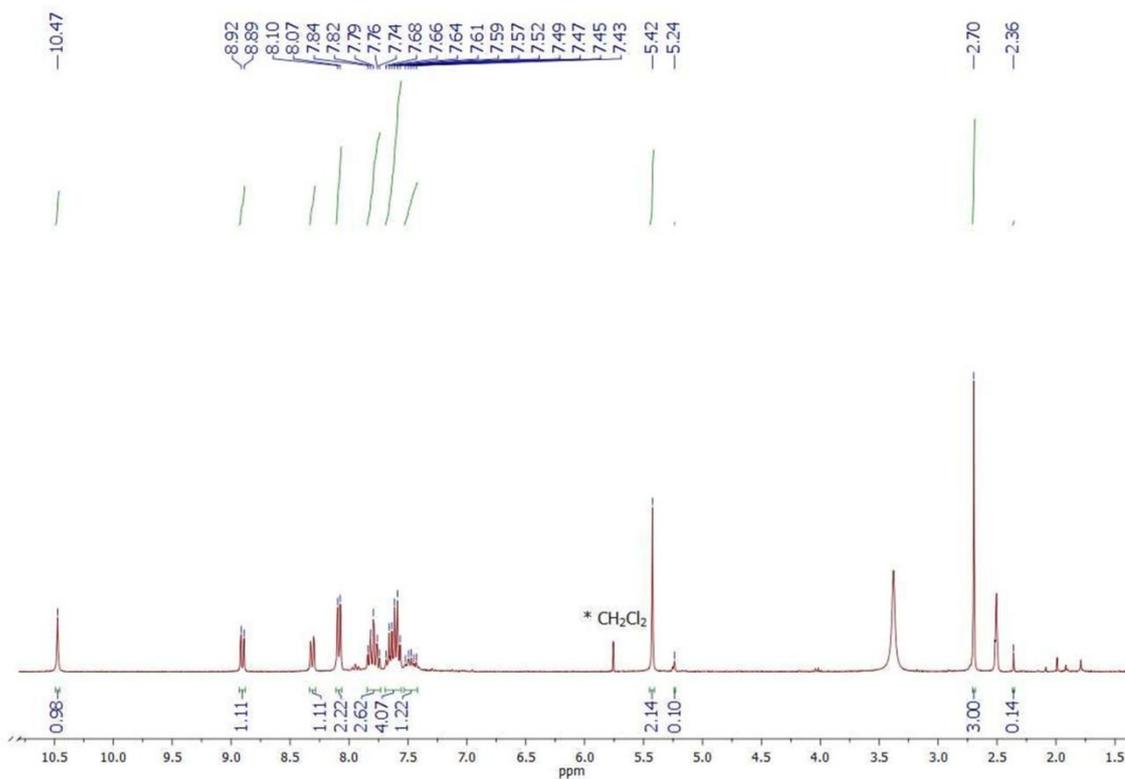
Solvent: toluene



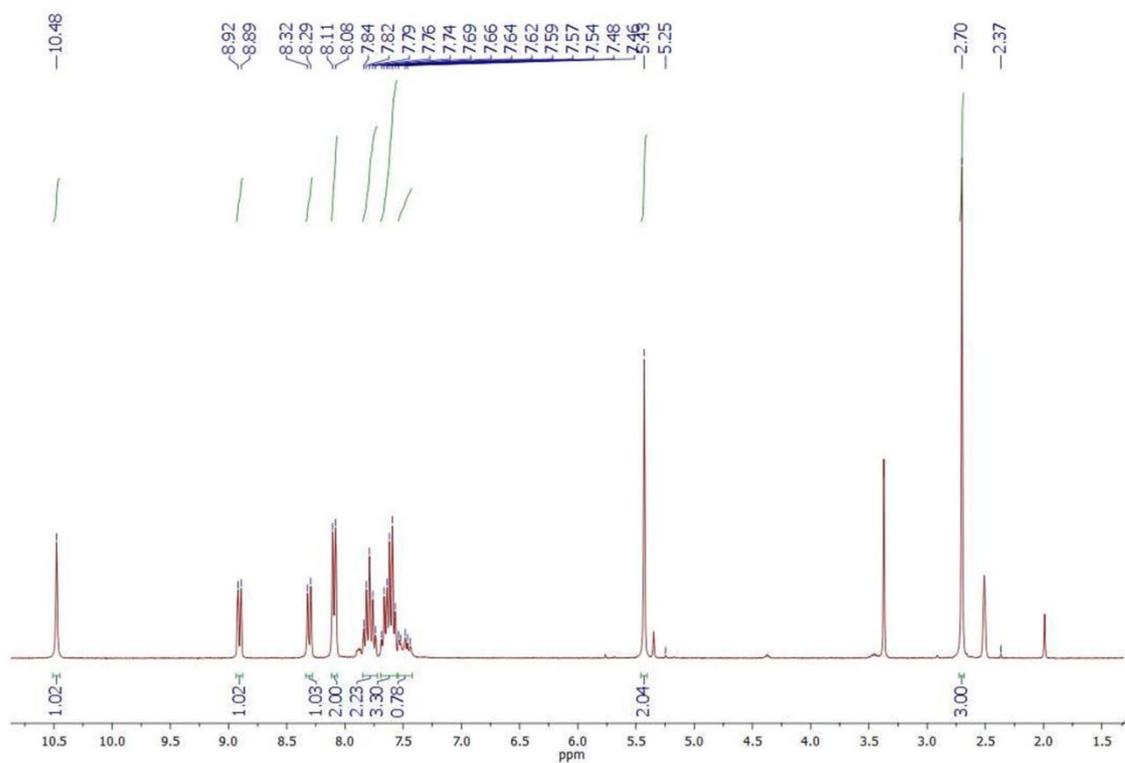
Solvent: CH₃NO₂



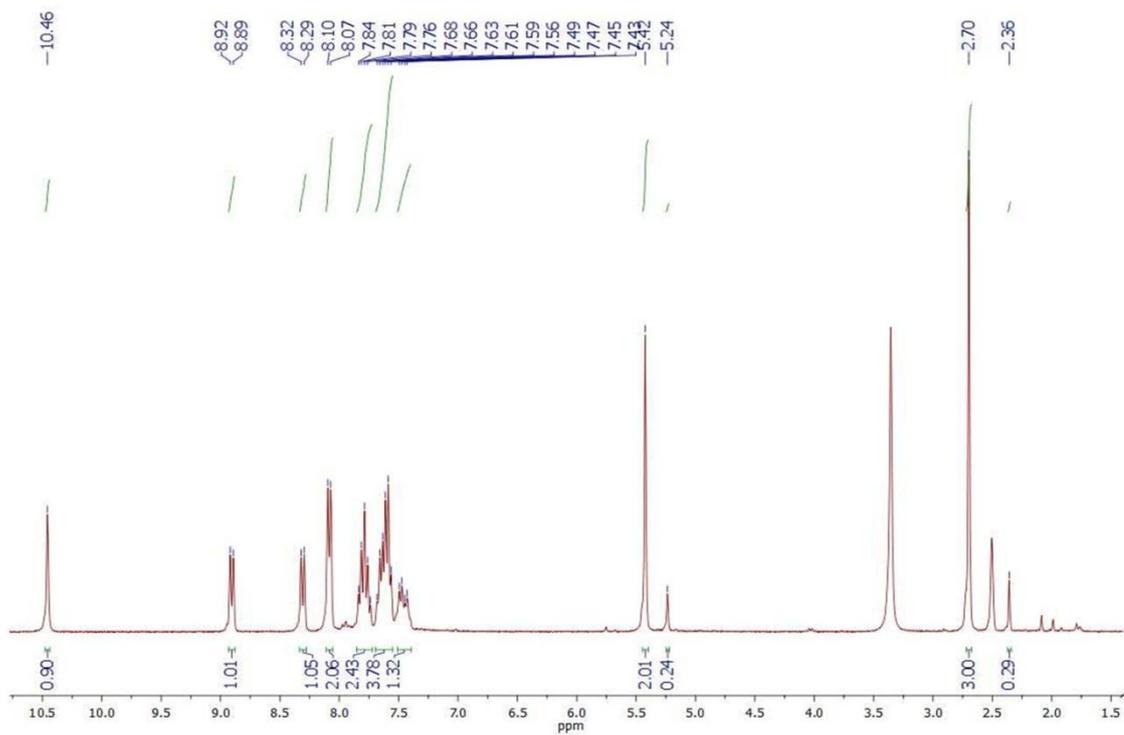
Solvent: CH₂Cl₂



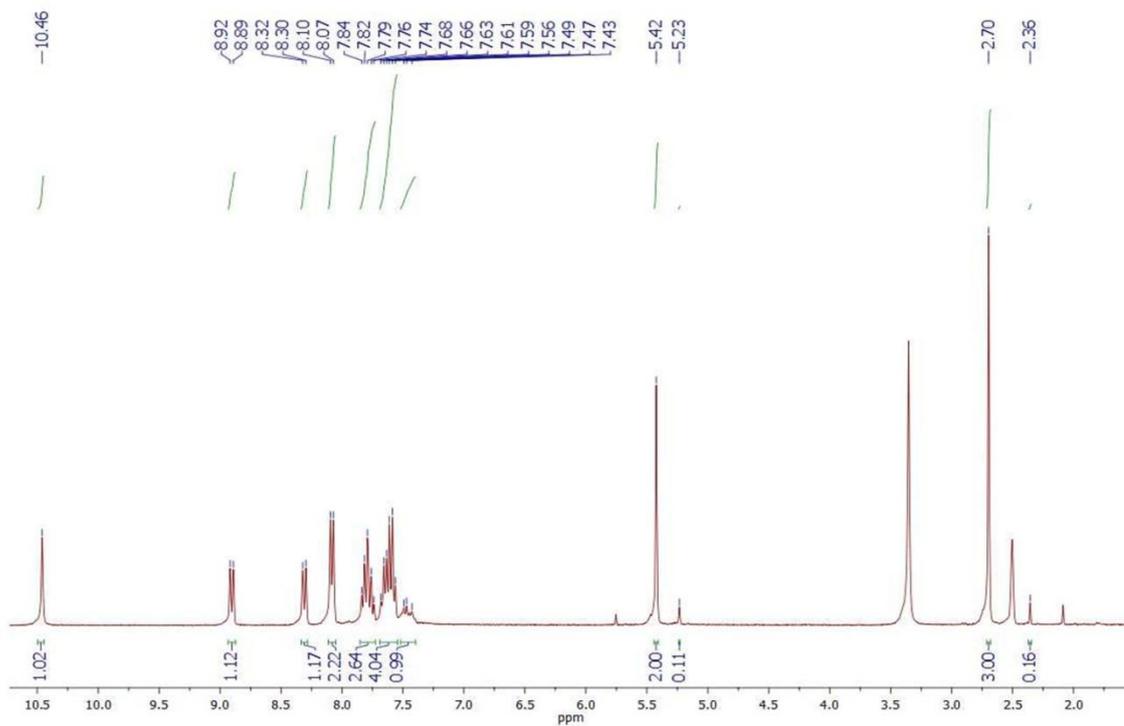
Solvent: EtOH



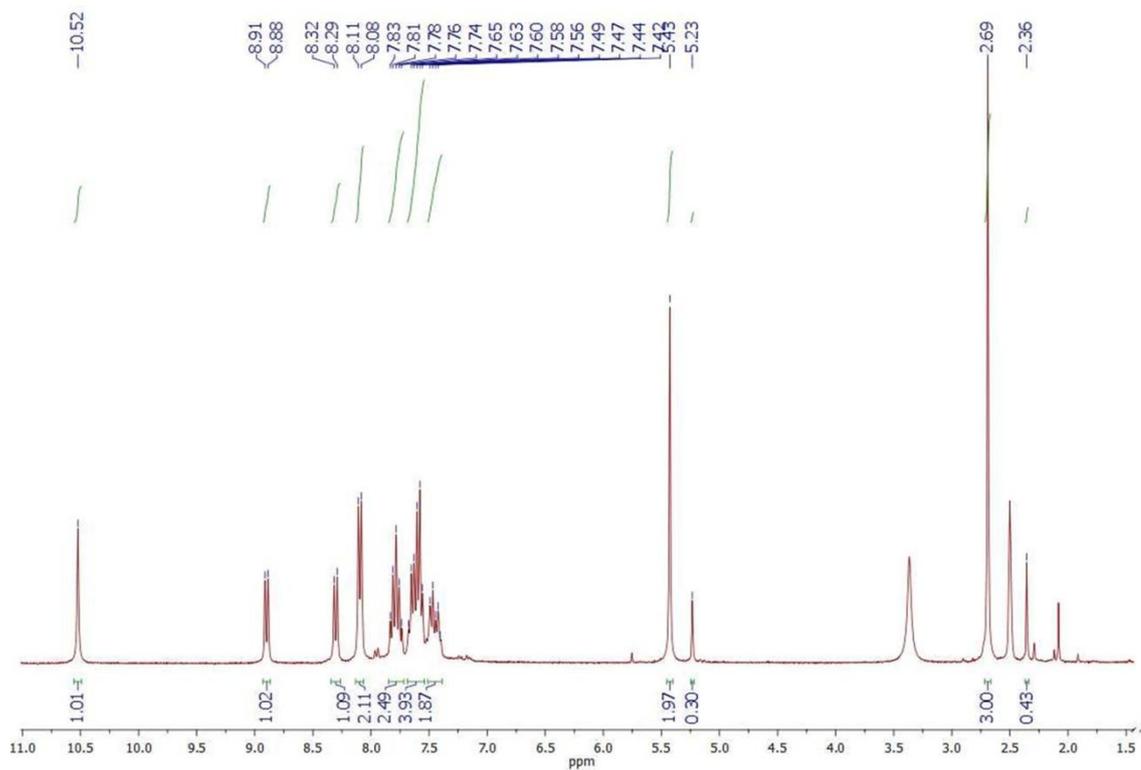
Solvent: EtOAc



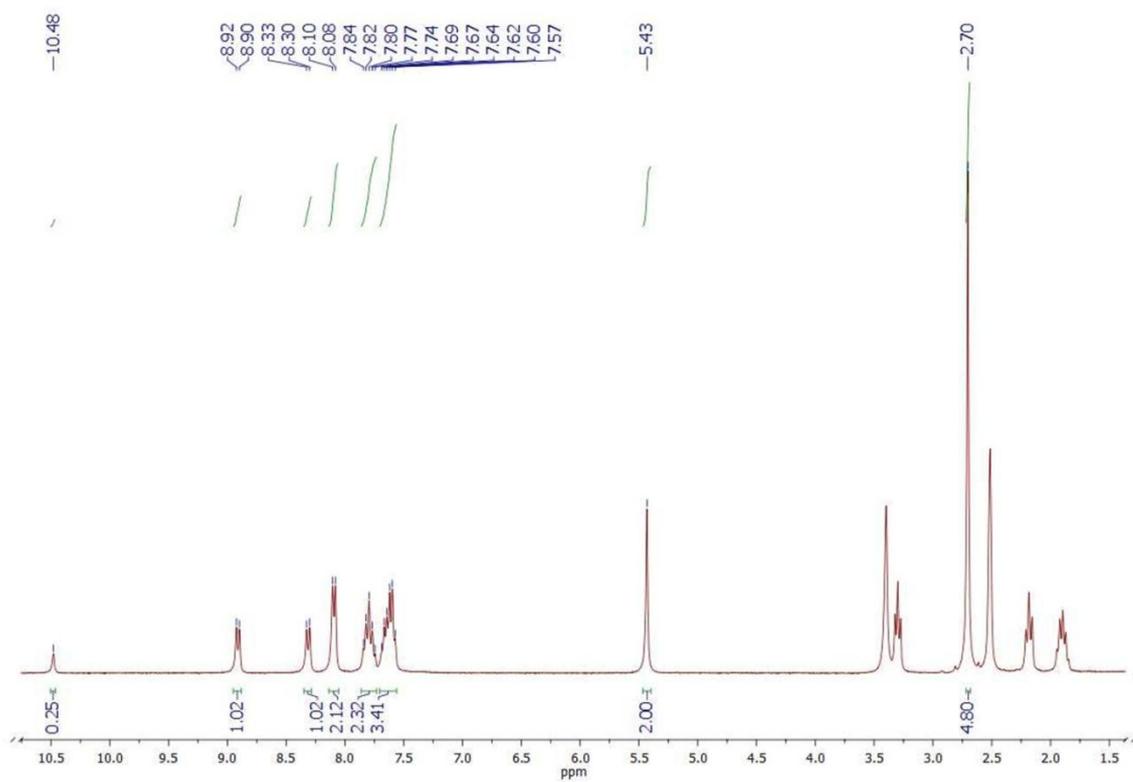
Solvent: CHCl₃



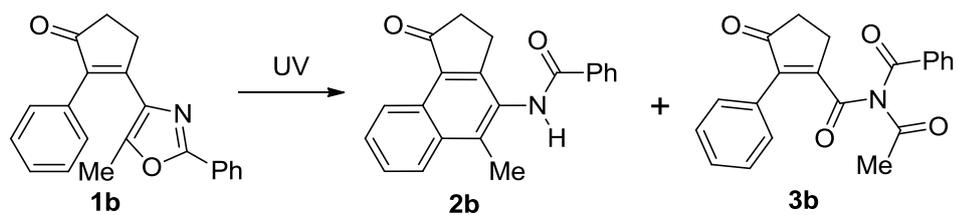
Solvent: acetone



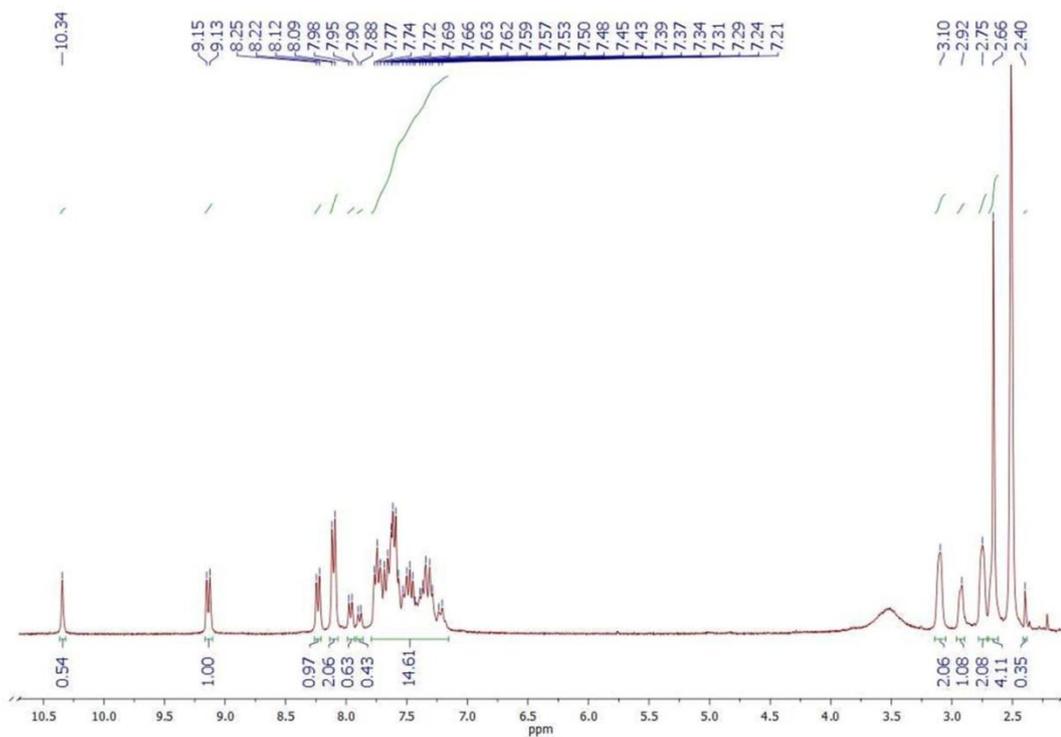
Solvent: *N*-methyl-2-pyrrolidone



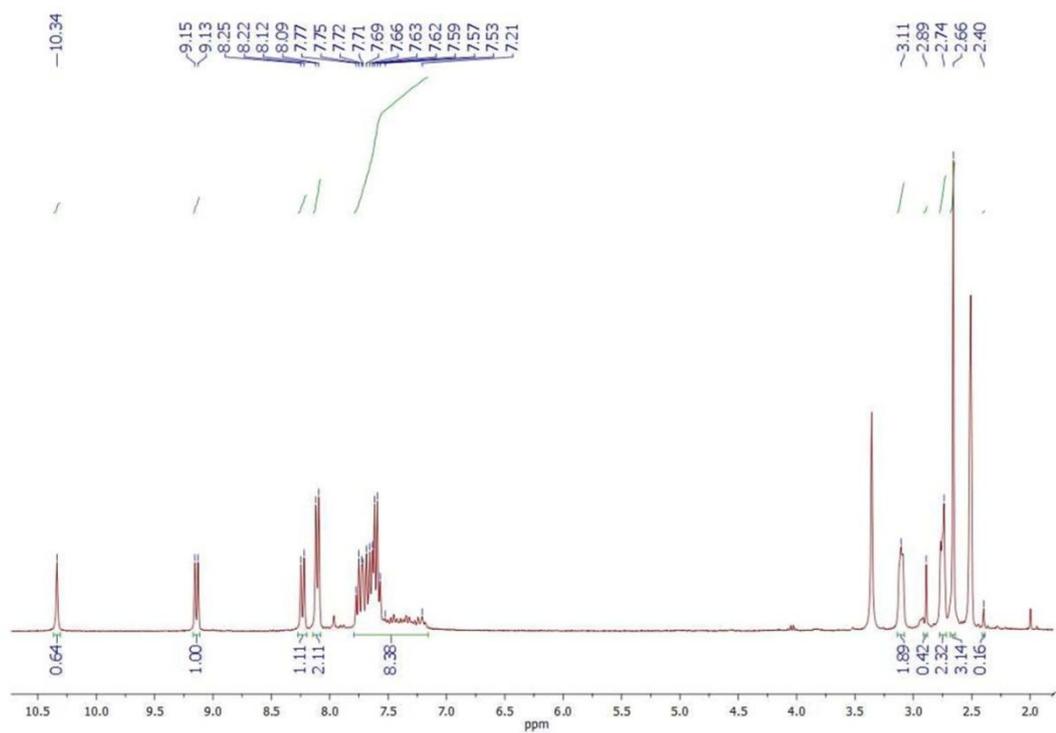
I.2. Diarylethene 1b



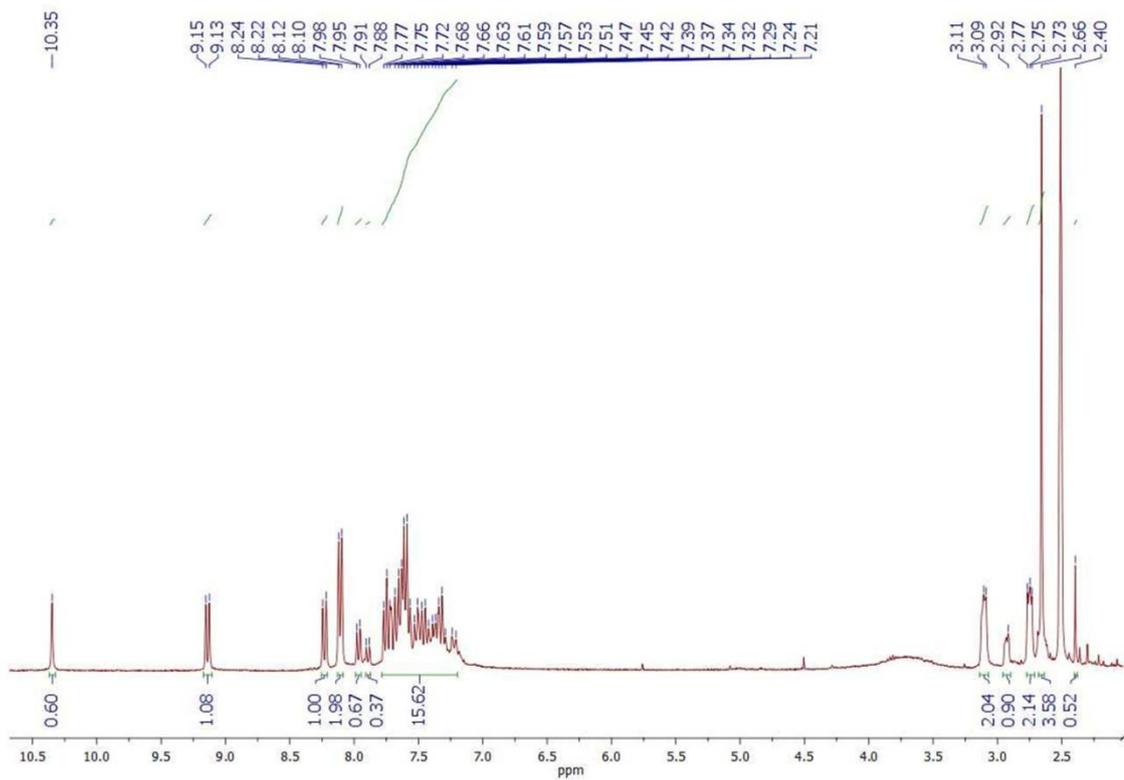
Solvent: MeCN



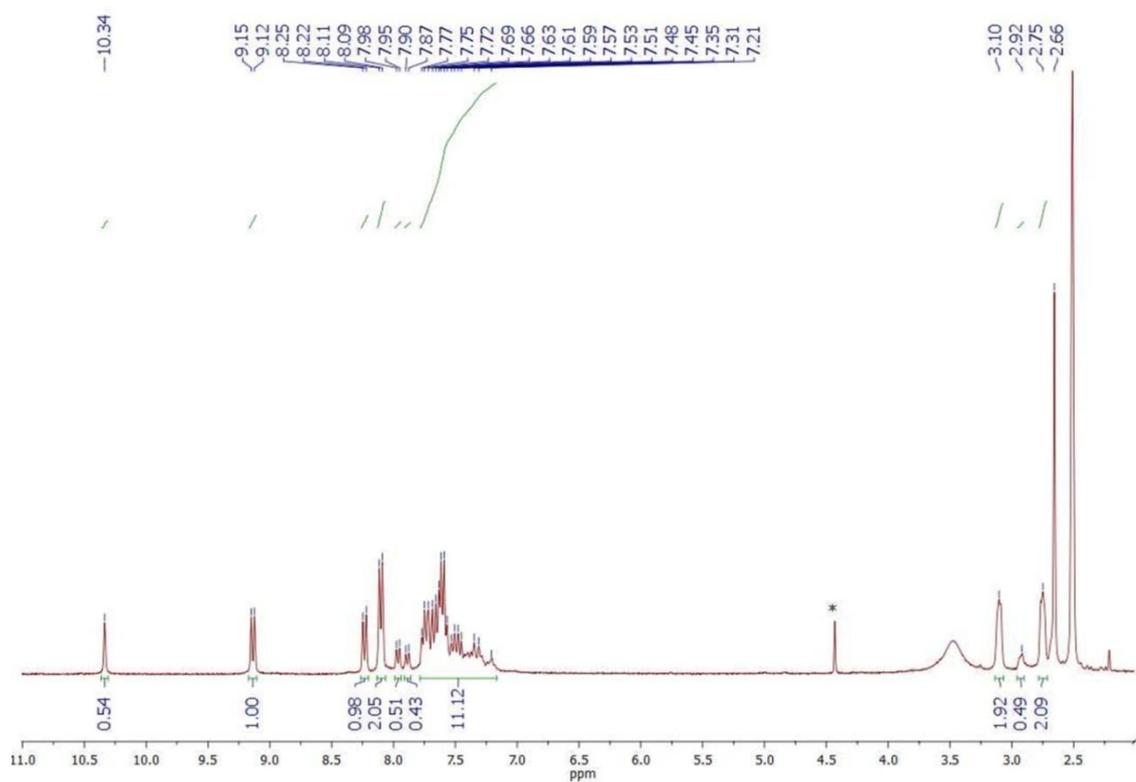
Solvent: DMF



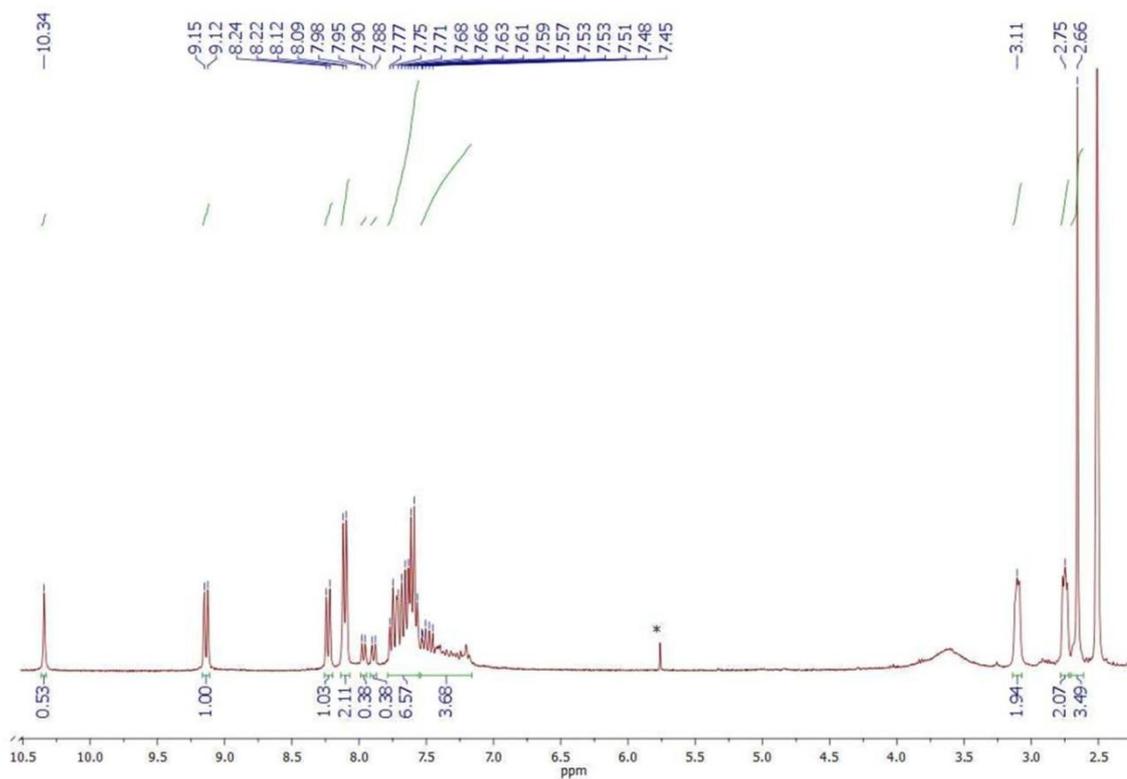
Solvent: toluene



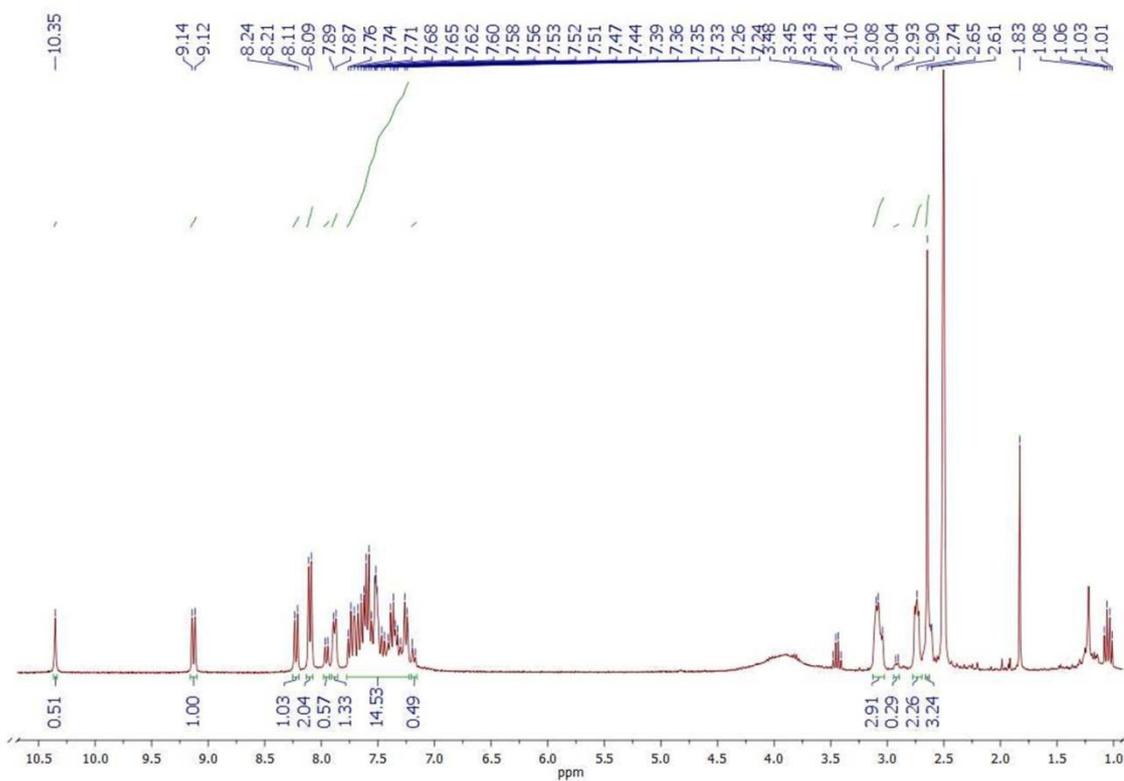
Solvent: MeNO₂



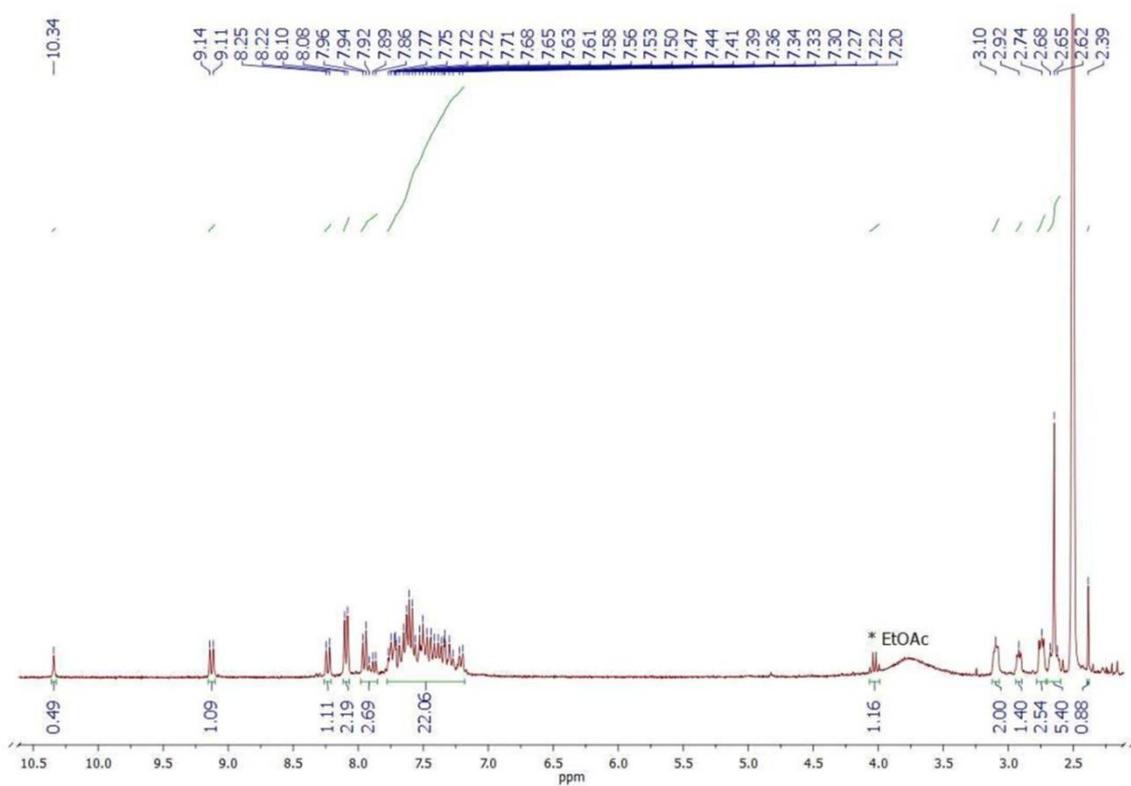
Solvent: CH₂Cl₂



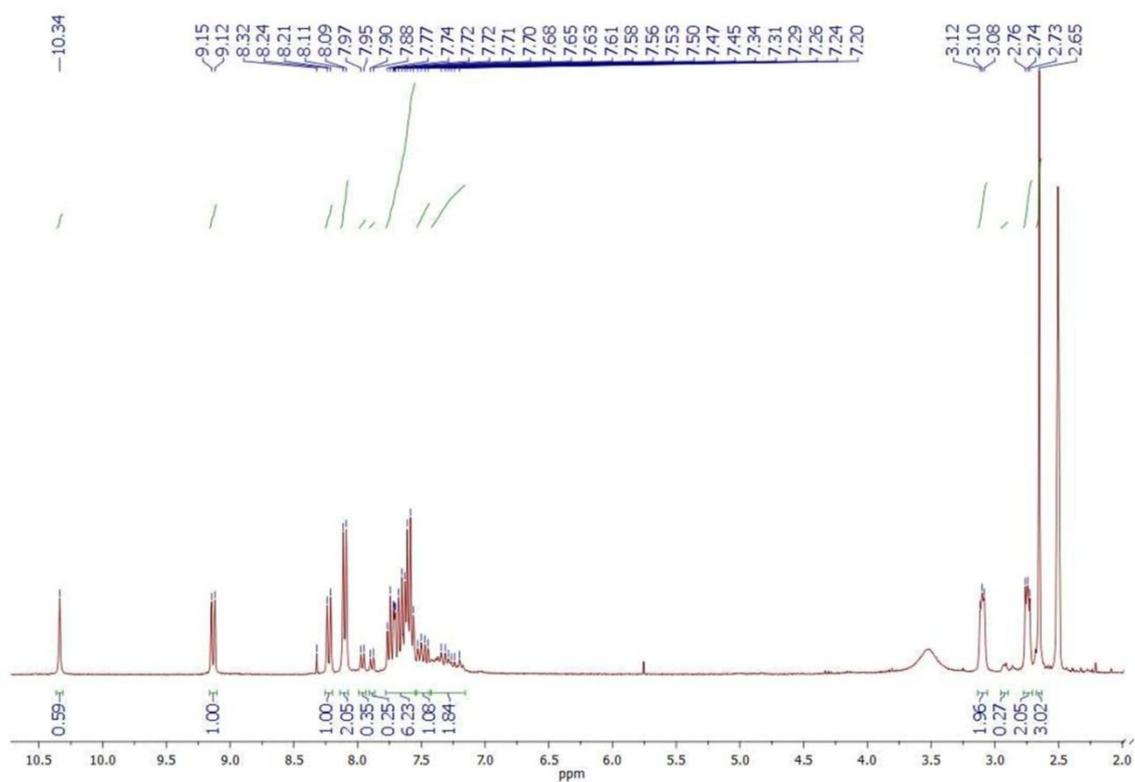
Solvent: EtOH



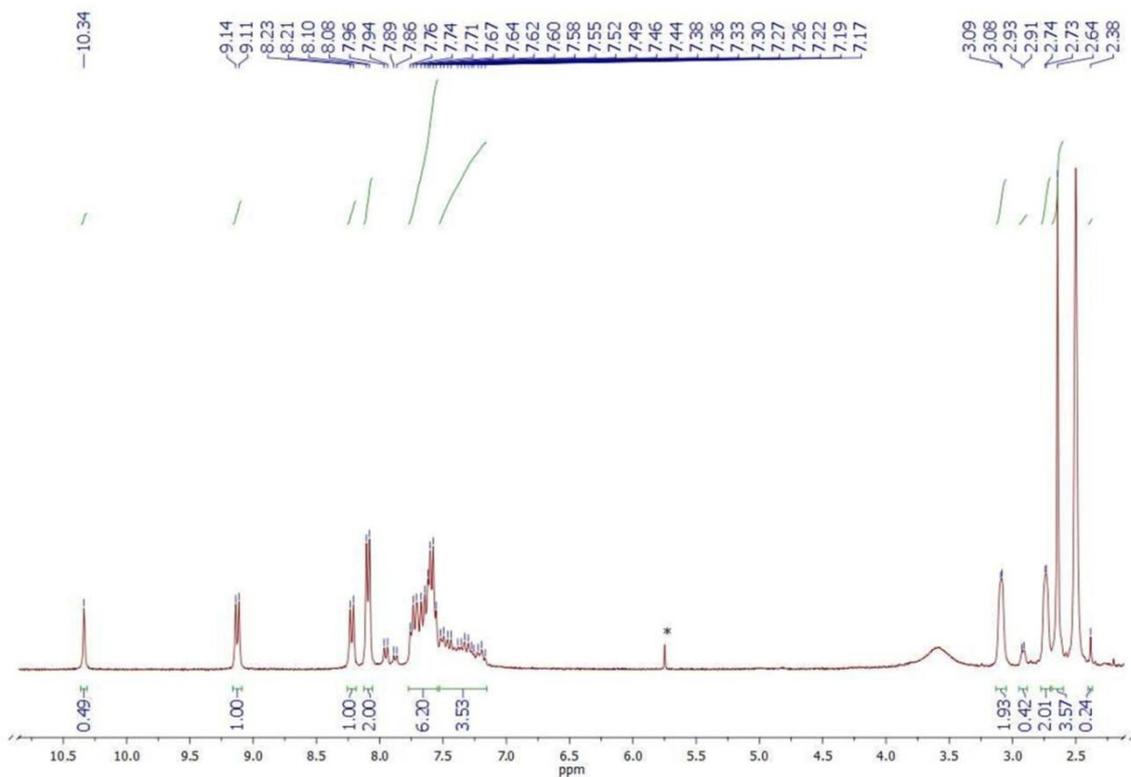
Solvent: EtOAc



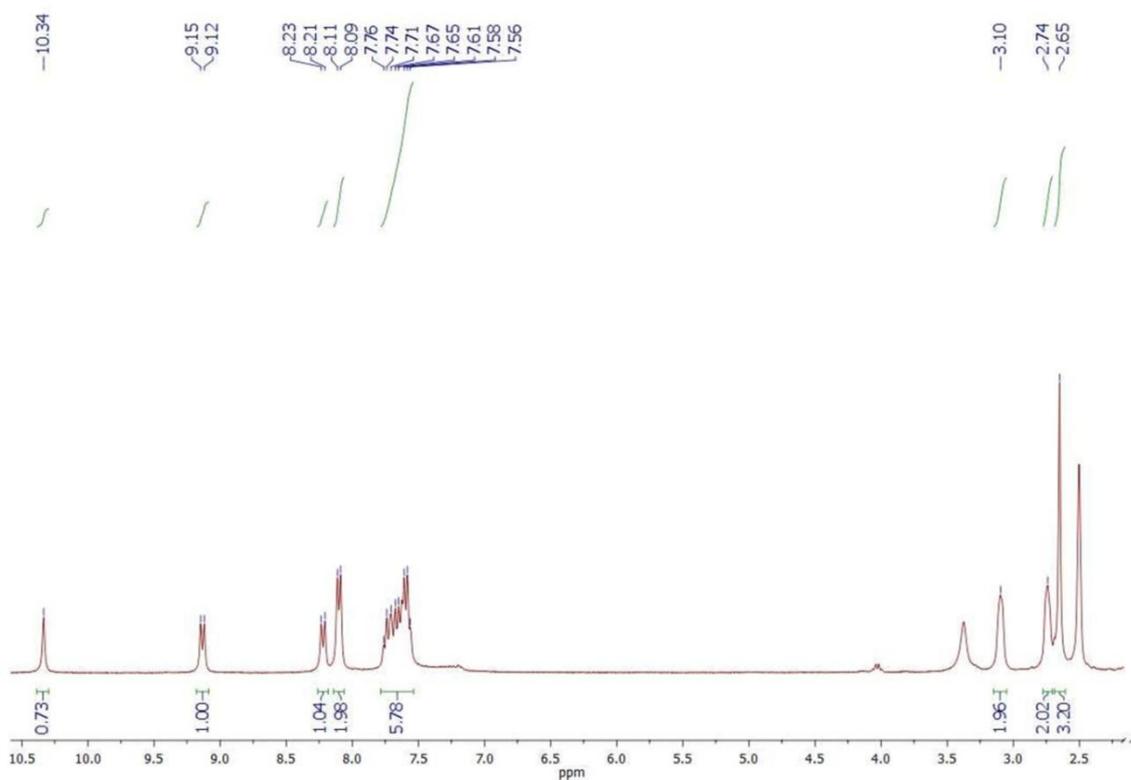
Solvent: CHCl₃



Solvent: acetone

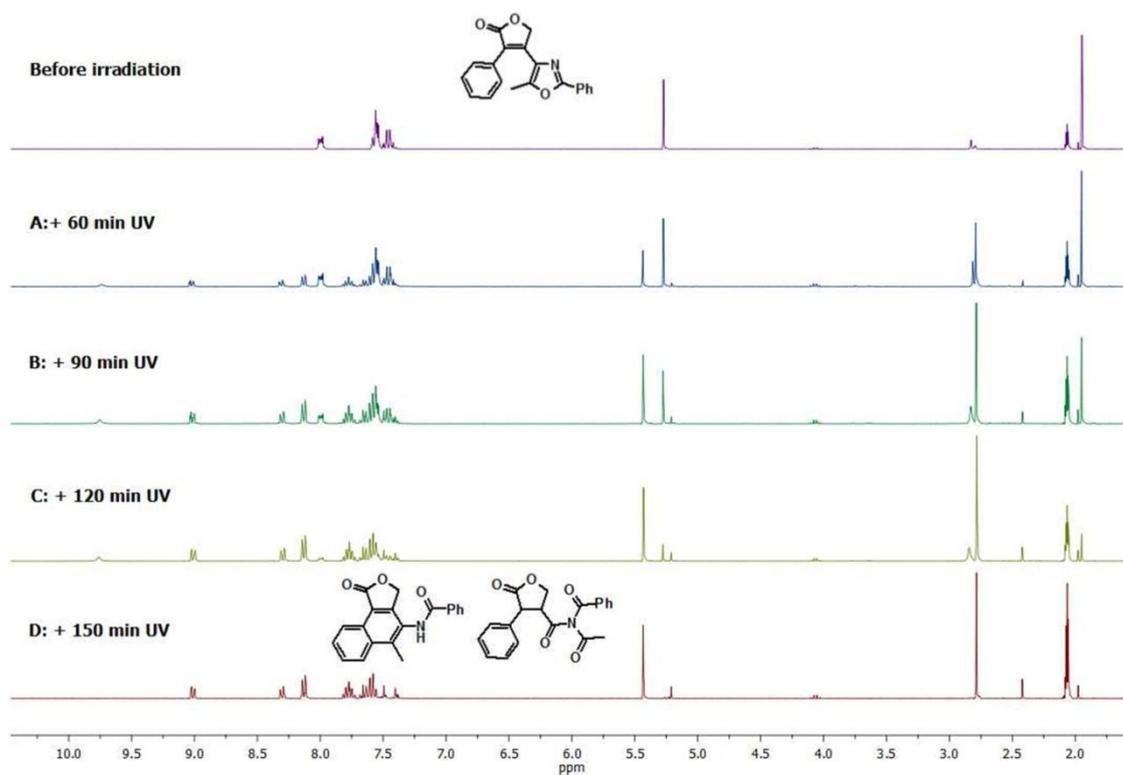


Solvent: *N*-methyl-2-pyrrolidone

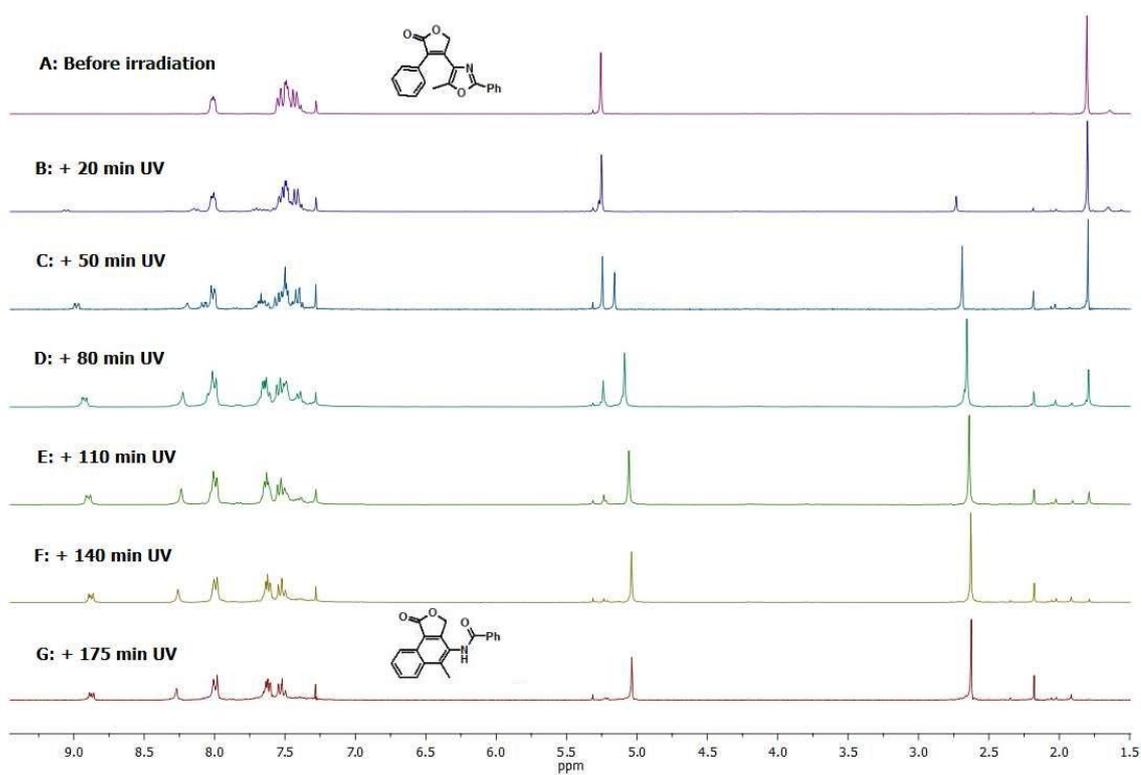


II. NMR monitoring of photoreactions of diarylethene 1a

Acetone- d_6 , C = 0.063 M.



Solvent: $CDCl_3$, C = 0.063 M.



III. Photosensitizer effect (NMR spectra of reaction mixtures)

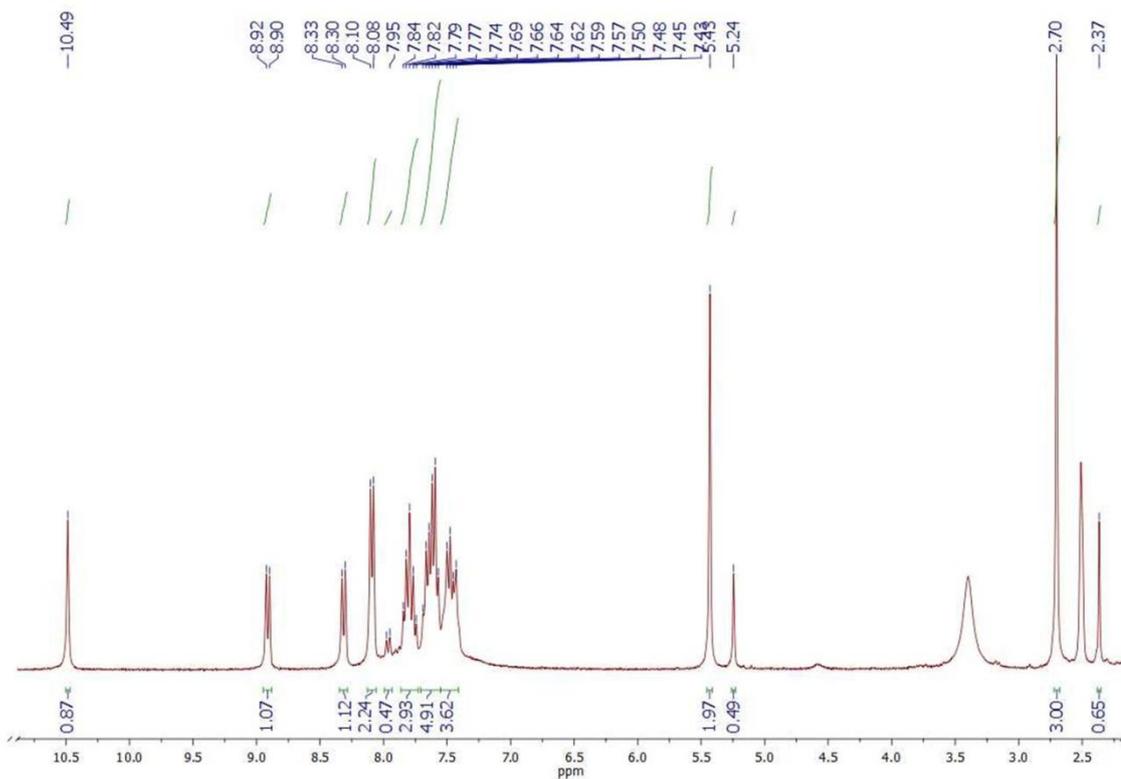
Table S1. The effect of UV sensitizers on the yields of photoproducts^a

No.	Sensitizers	Time (h)	Yields	
			2a	3a
1	Coumarin 30	17	80	20
2	Perylenetetracarboxylic dianhydride	17	83	17
3	Erythrosin B	17	82	18
4	Coumarin 1	17	81	19
5	6-Ethoxy-3-methyl-1 <i>H</i> -phenalen-1-one	17	80	20
6	Naphthalene ^b	17	69	30
7	Phenanthrene	15	78	22
8	Pyrene	26	81	19
9	4-Dimethylamino-4'-nitrostilbene	17	94	6
10	5,10,15,20-Tetrakis(4-bromophenyl)porphyrin	11	84	16
11	Benzophenone	36	85	15

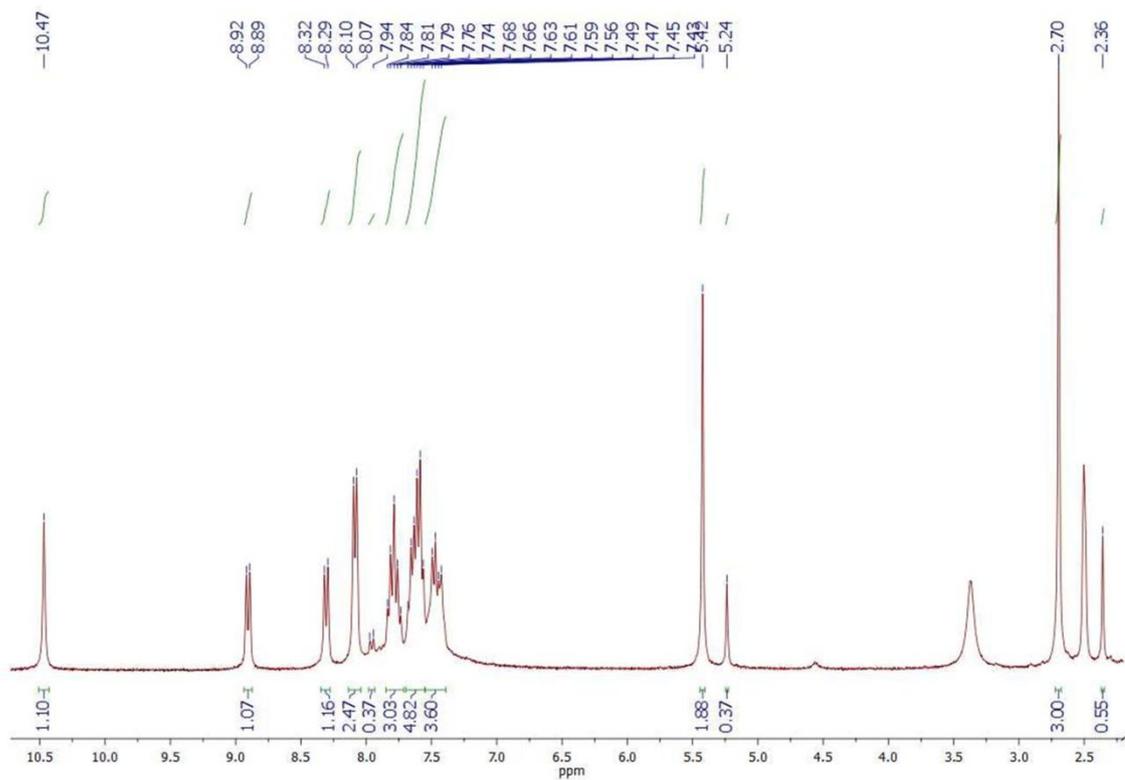
^a 0.1 eq. of sensitizer with 1 eq. of diarylethene in 2 ml acetone;

^b Scaling down does not affect the yield of the by-product.

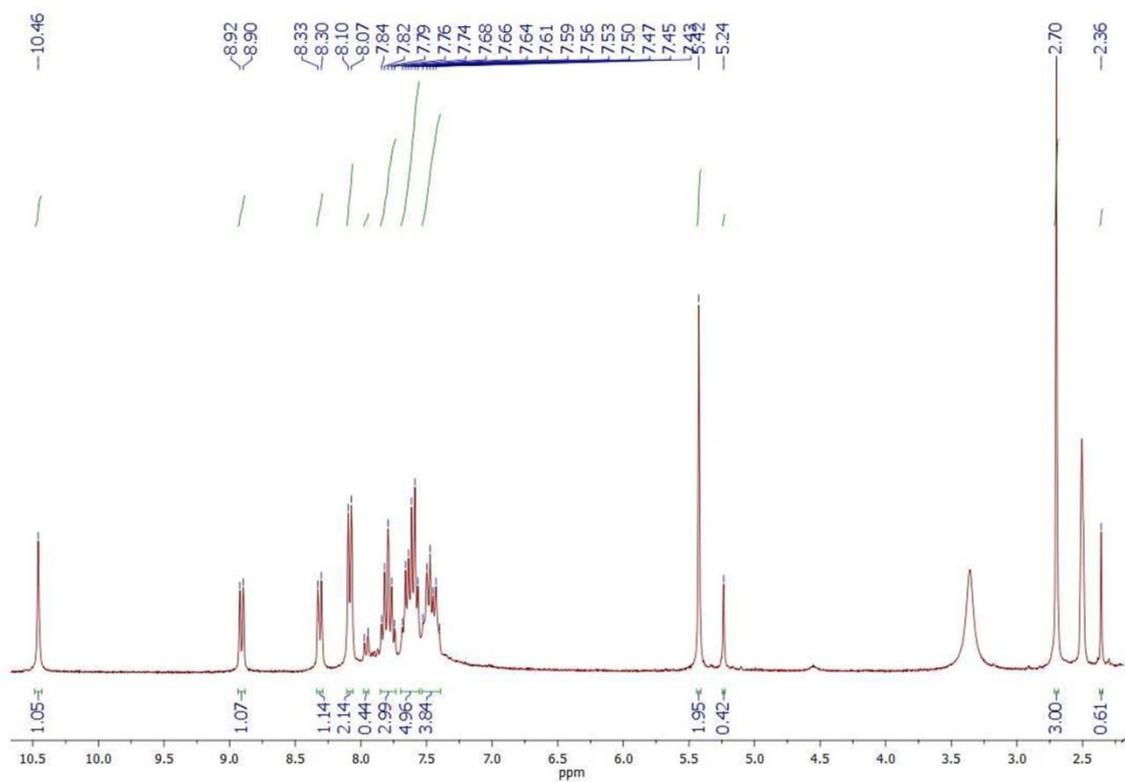
Sensitizer: coumarin 30



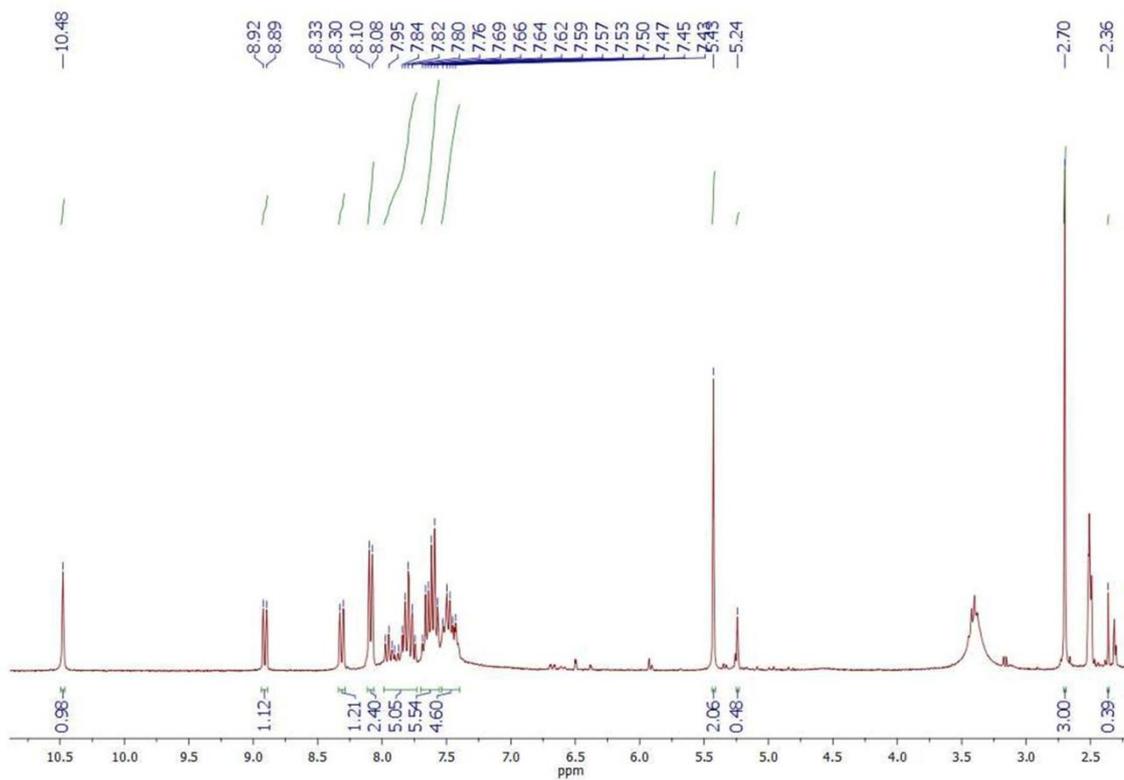
Sensitizer: perylenetetracarboxylic dianhydride



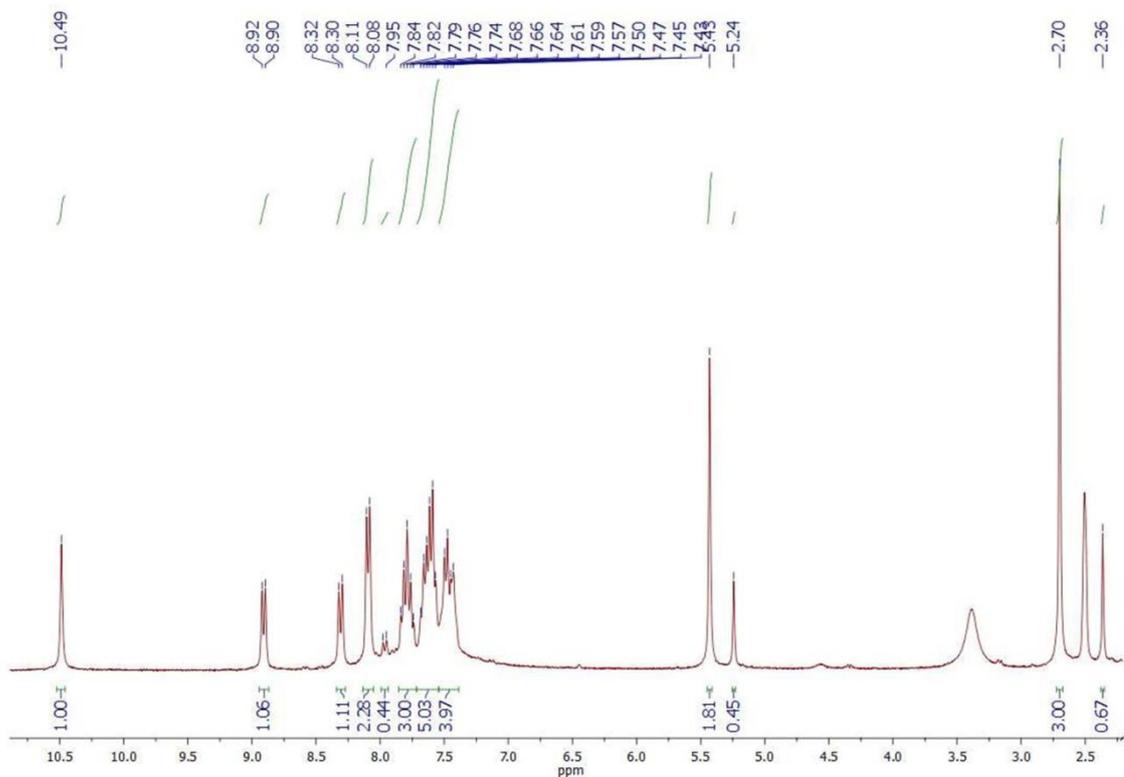
Sensitizer: erythrosine B



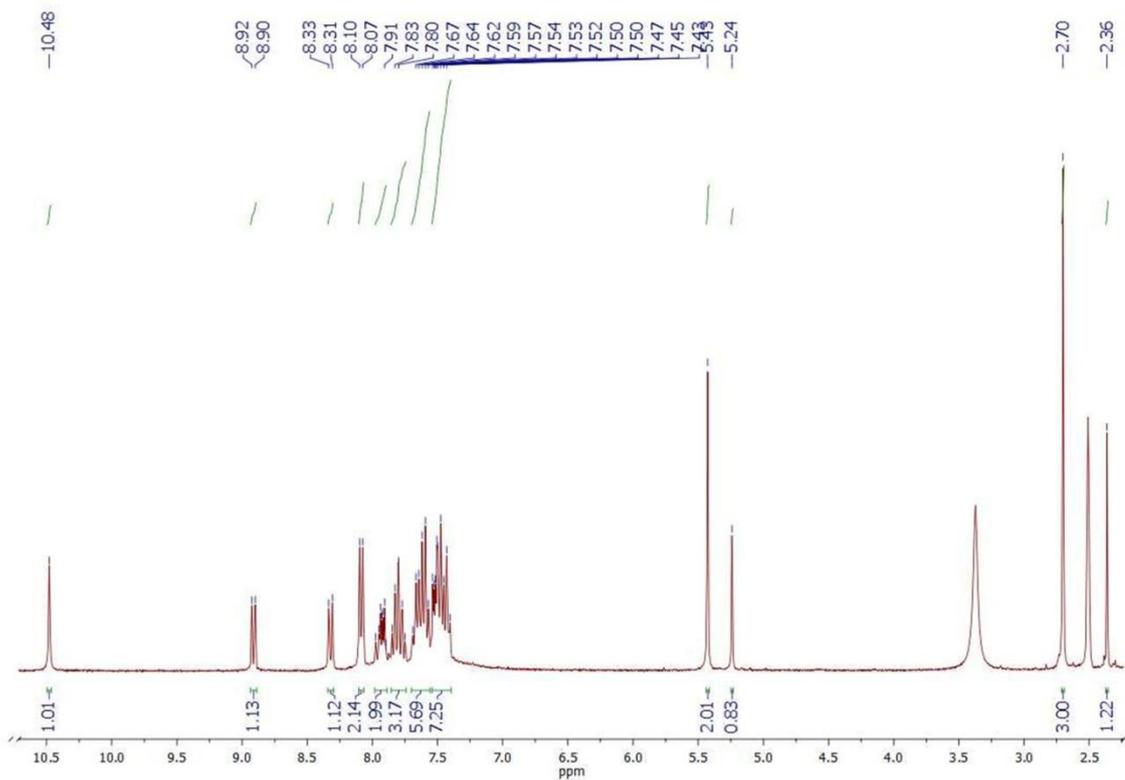
Sensitizer: coumarin 1



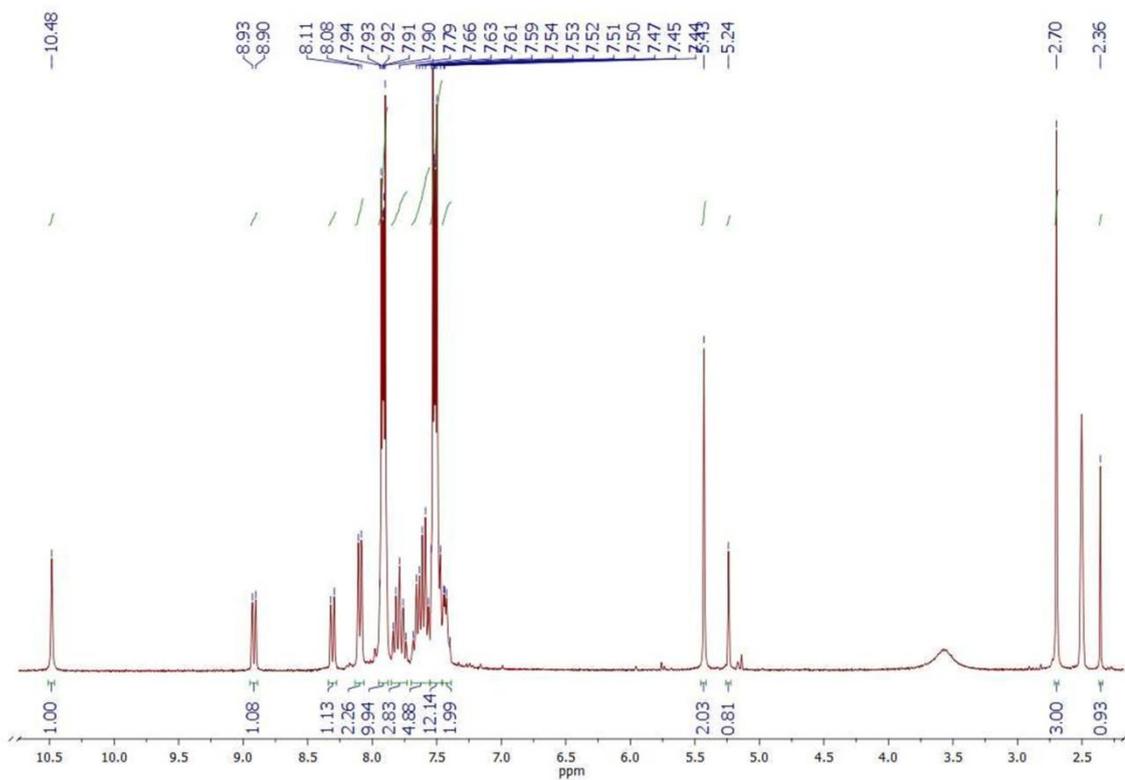
Sensitizer: 6-ethoxy-3-methyl-1H-phenalen-1-one



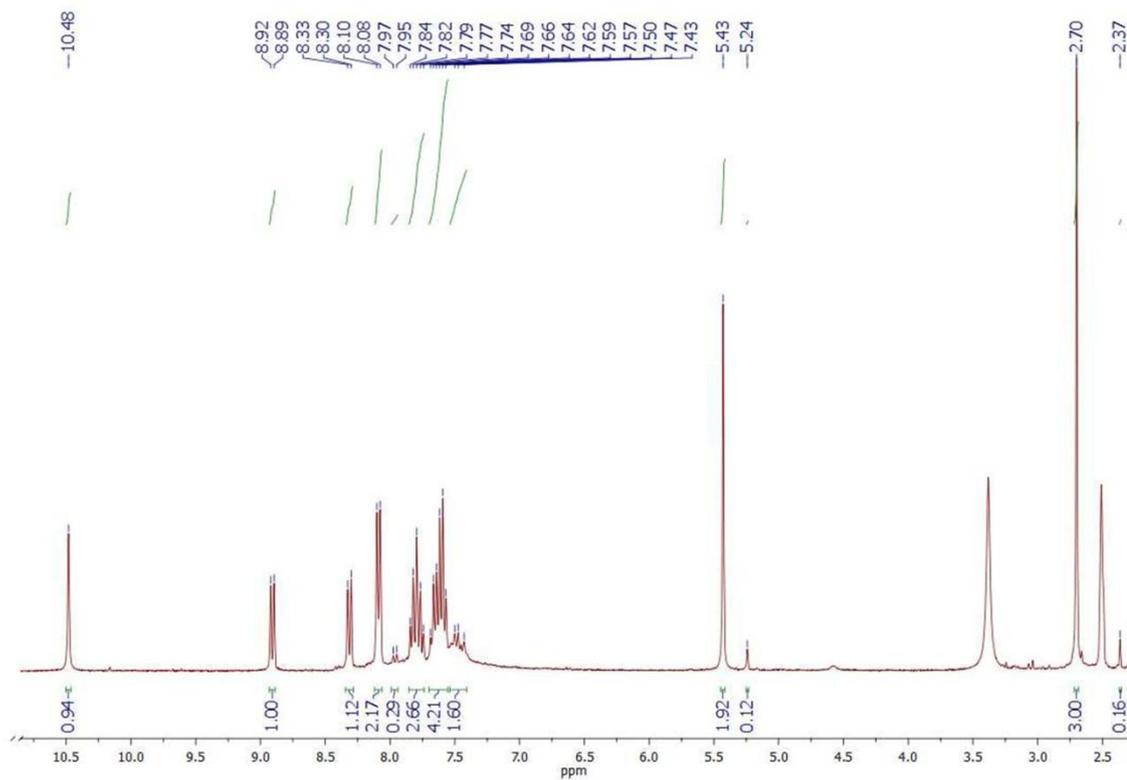
Sensitizer: naphthalene (0.1 eq.)



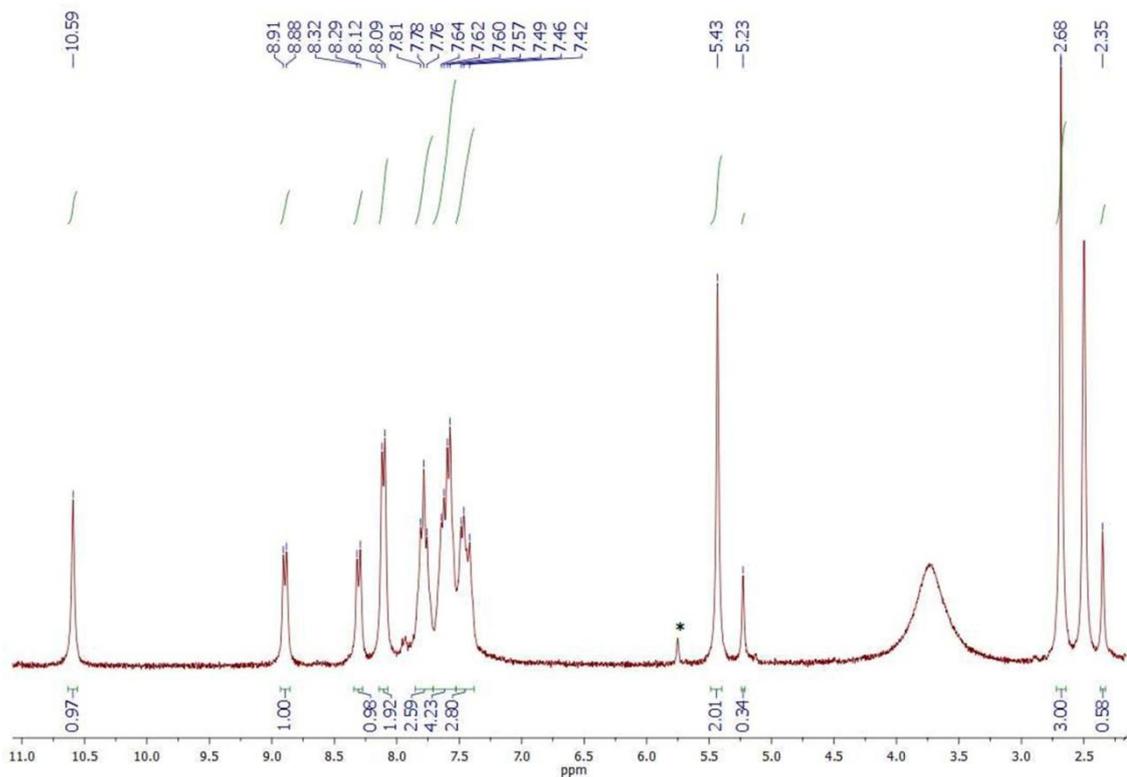
Sensitizer: naphthalene (1 eq.)



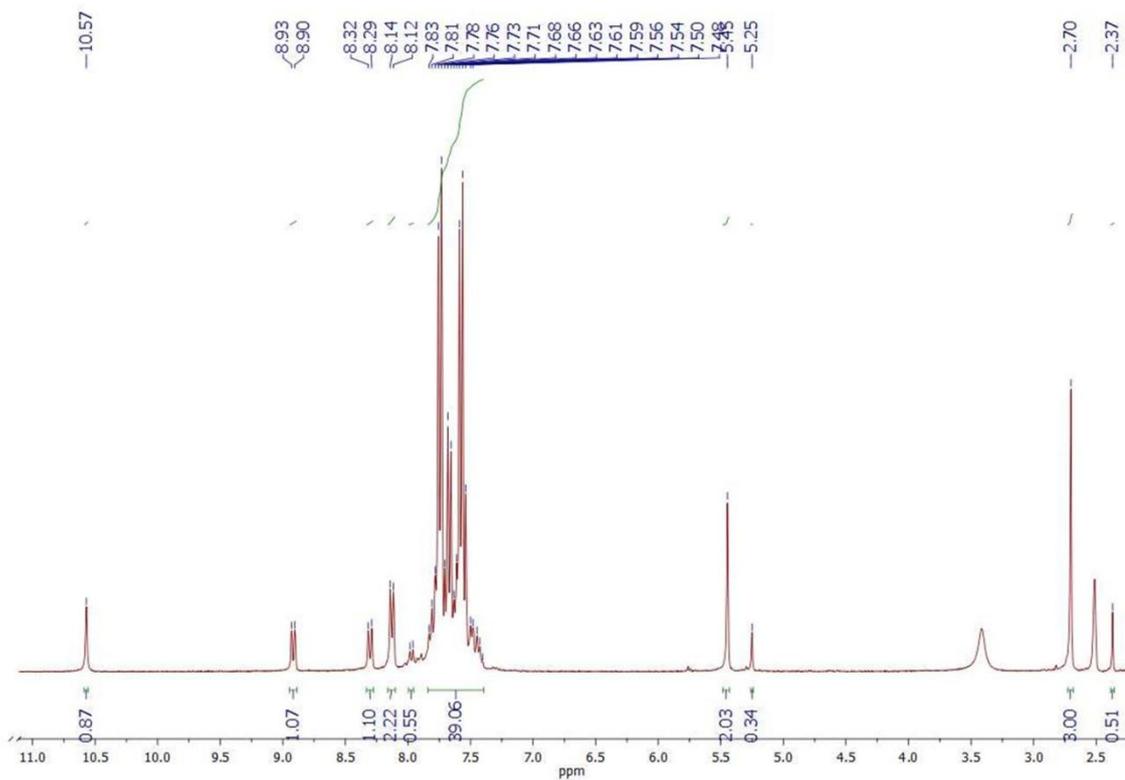
Sensitizer: 4-dimethylamino-4'-nitrostilbene



Sensitizer: 5,10,15,20-tetrakis(4-bromophenyl)porphyrin



Sensitizer: benzophenone

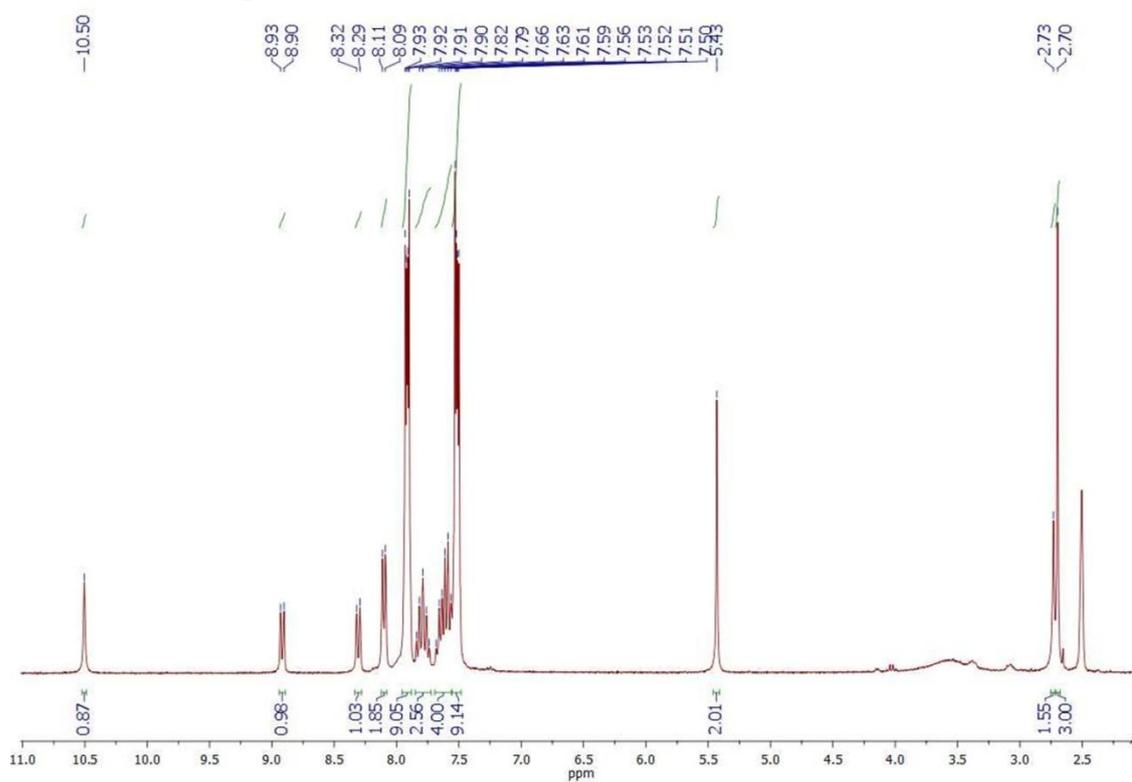


IV. Amine effect (NMR spectra of reaction mixtures)

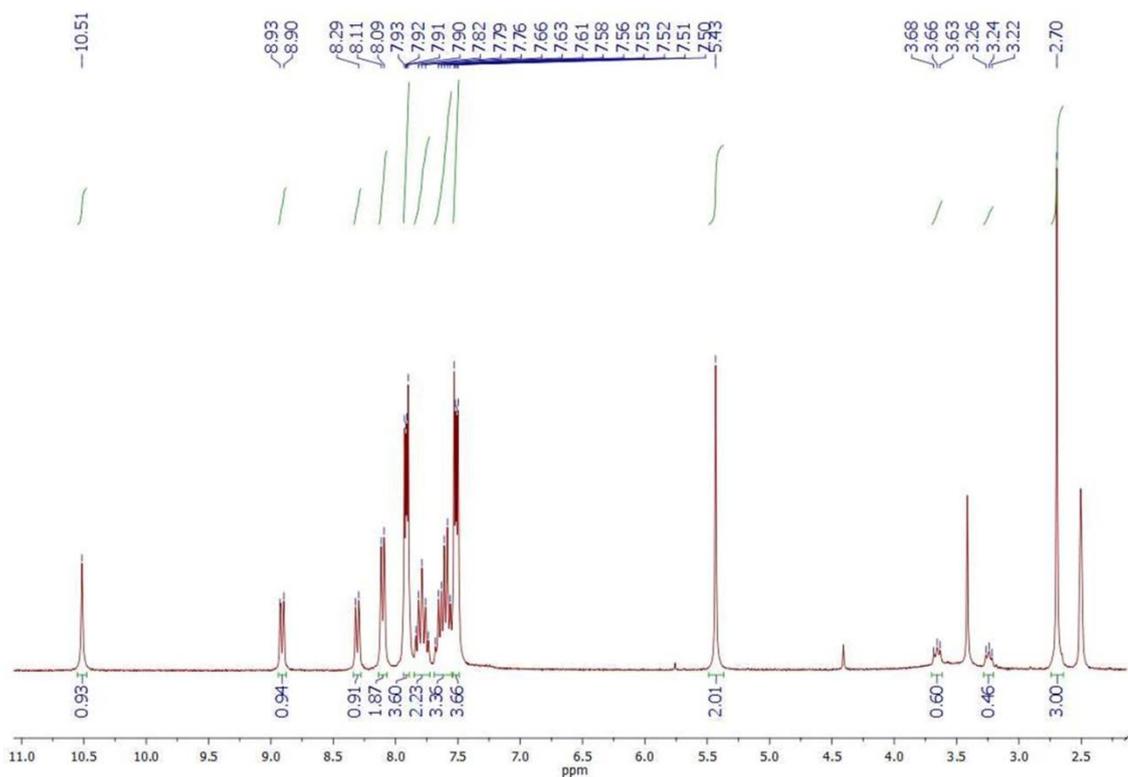
Table S2. ^1H NMR chemical shift signals (ppm) of tertiary amines after reaction completion.

Amines	Starting values	Values after photoreaction
Et_3N	0.93 (t, $J = 7.2$ Hz, 3H, CH_3)	1.20 (t, $J = 7.2$ Hz, 3H, CH_3)
	2.43 (q, $J = 7.2$ Hz, 2H, CH_2)	3.04 (q, $J = 7.2$ Hz, 2H, CH_2)
DABCO	2.73 (s, 12H, CH_2)	3.24 (m, 6H, CH_2)
		3.66 (m, 6H, CH_2)
1-Methylimidazole	3.66 (s, 3H, CH_3)	3.76 (s, 3H, CH_3)
	6.93 (s, 1H, H^{arom})	7.30 (s, 1H, H^{arom})
	7.14 (s, 1H, H^{arom})	7.43 (s, 1H, H^{arom})
	7.66 (s, 1H, H^{arom})	8.39 (s, 1H, H^{arom})

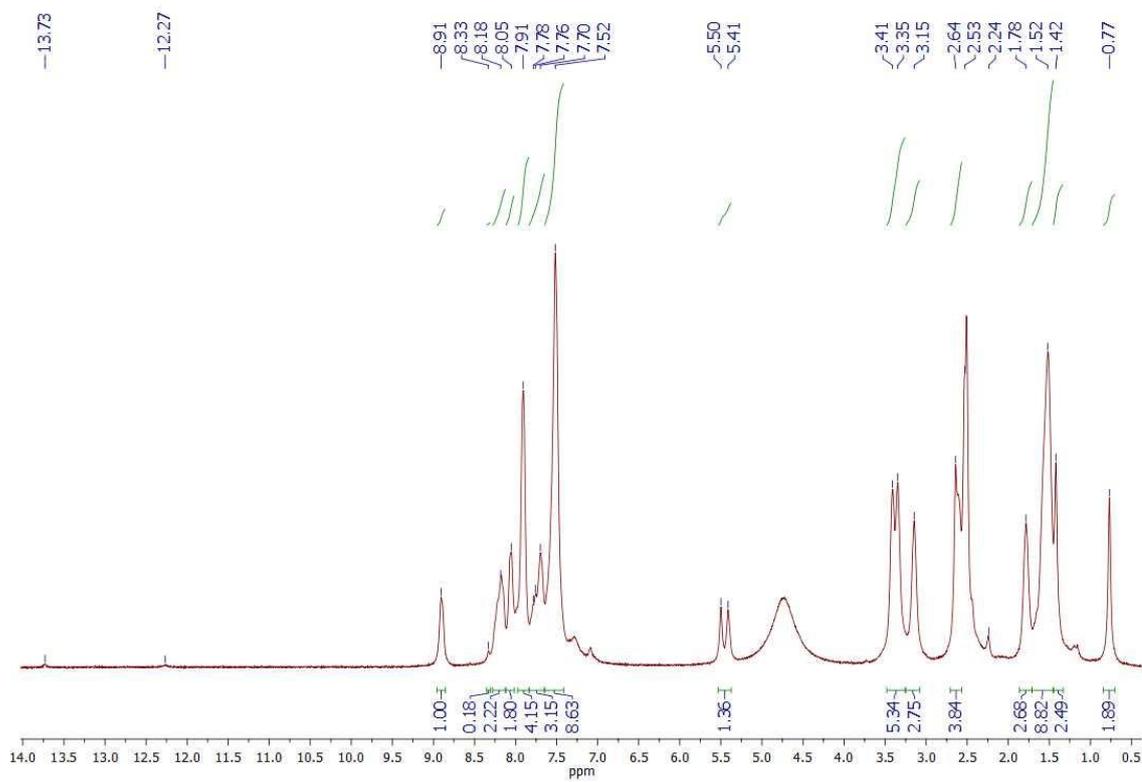
Amine: DABCO (1 eq.)



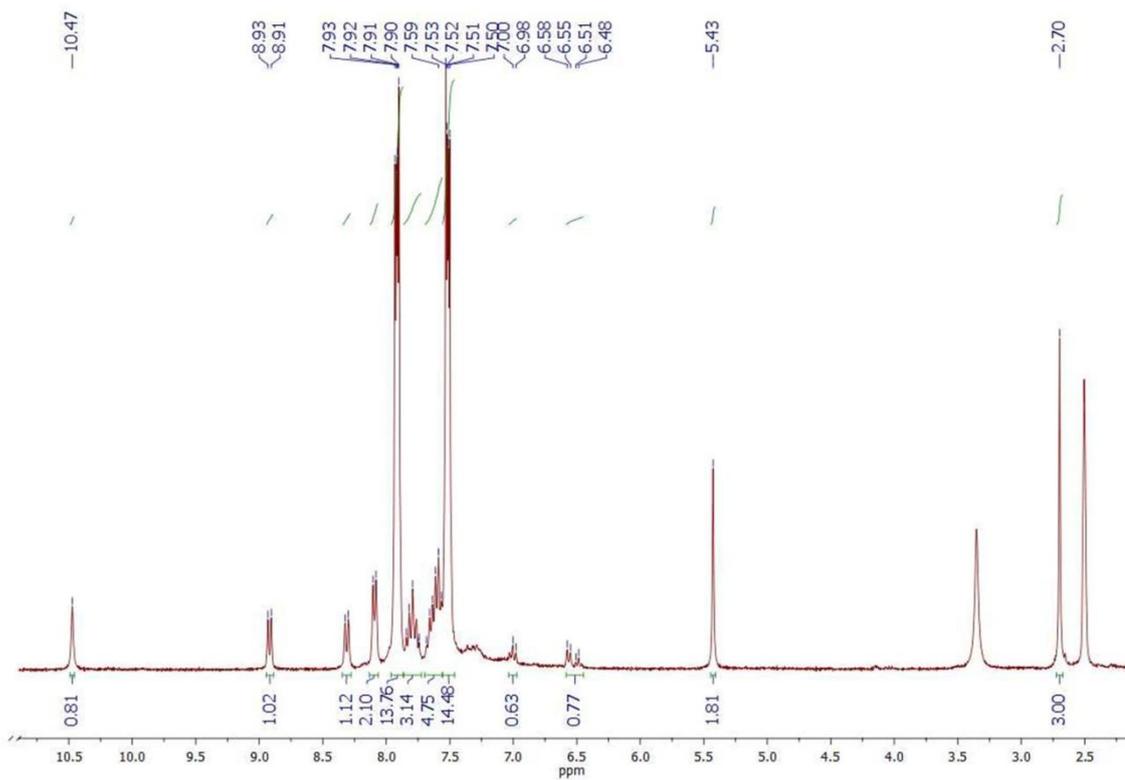
Amine: DABCO (0.1 eq.)



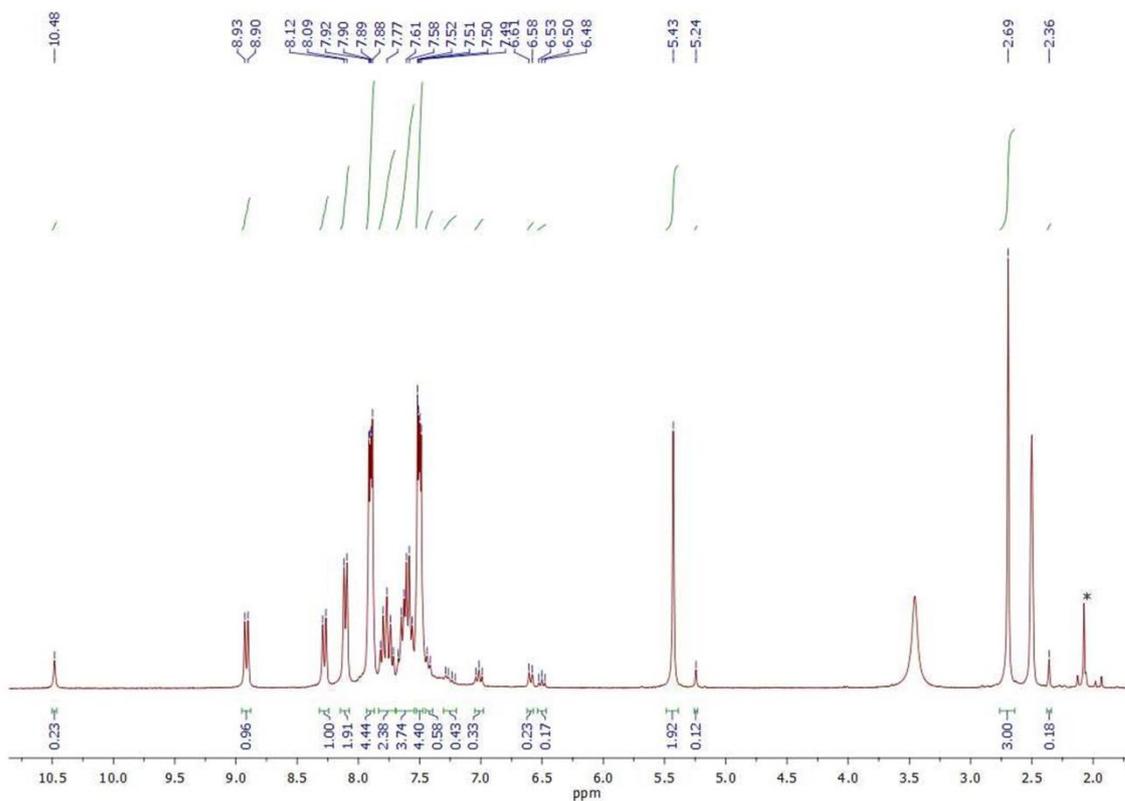
Amine: DBU



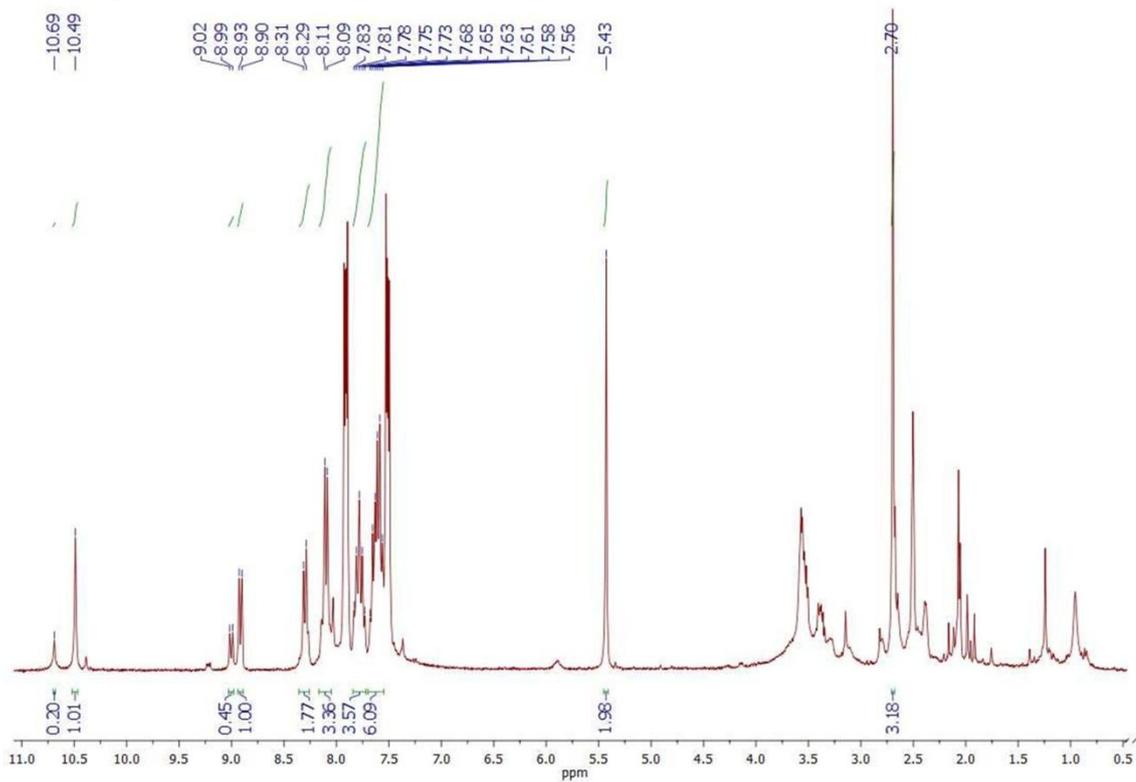
Amine: aniline (1 eq.)



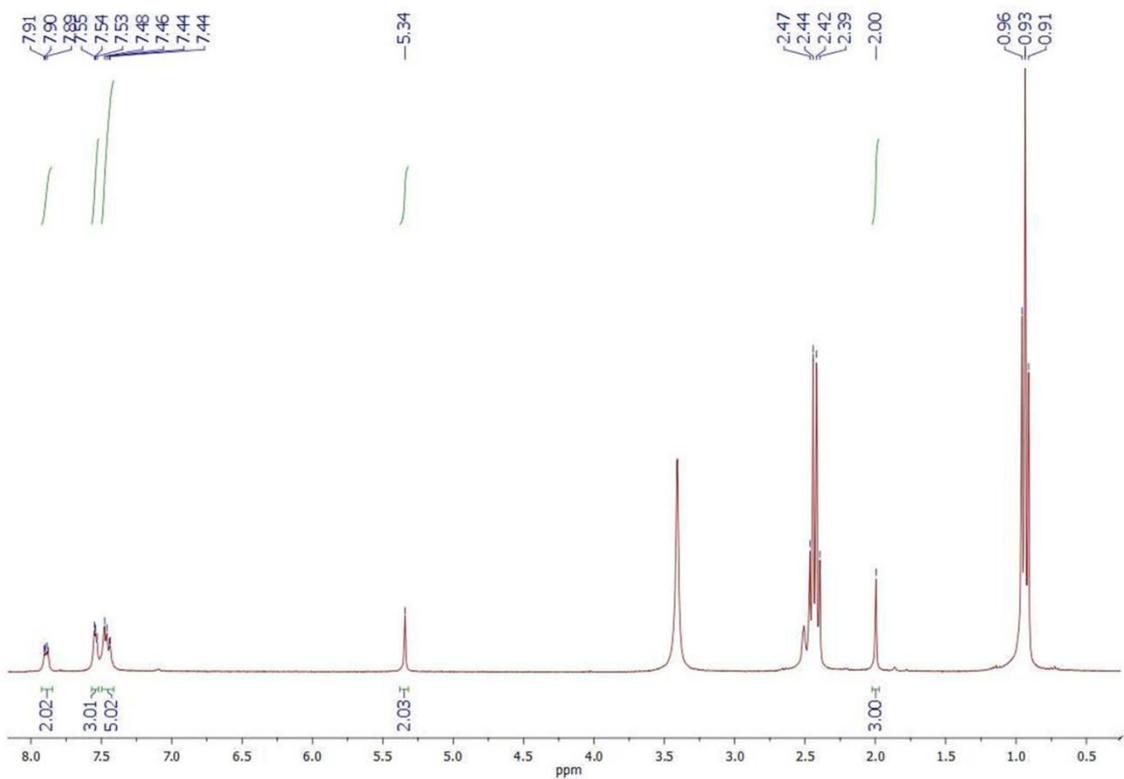
Amine: aniline (0.1 eq.)



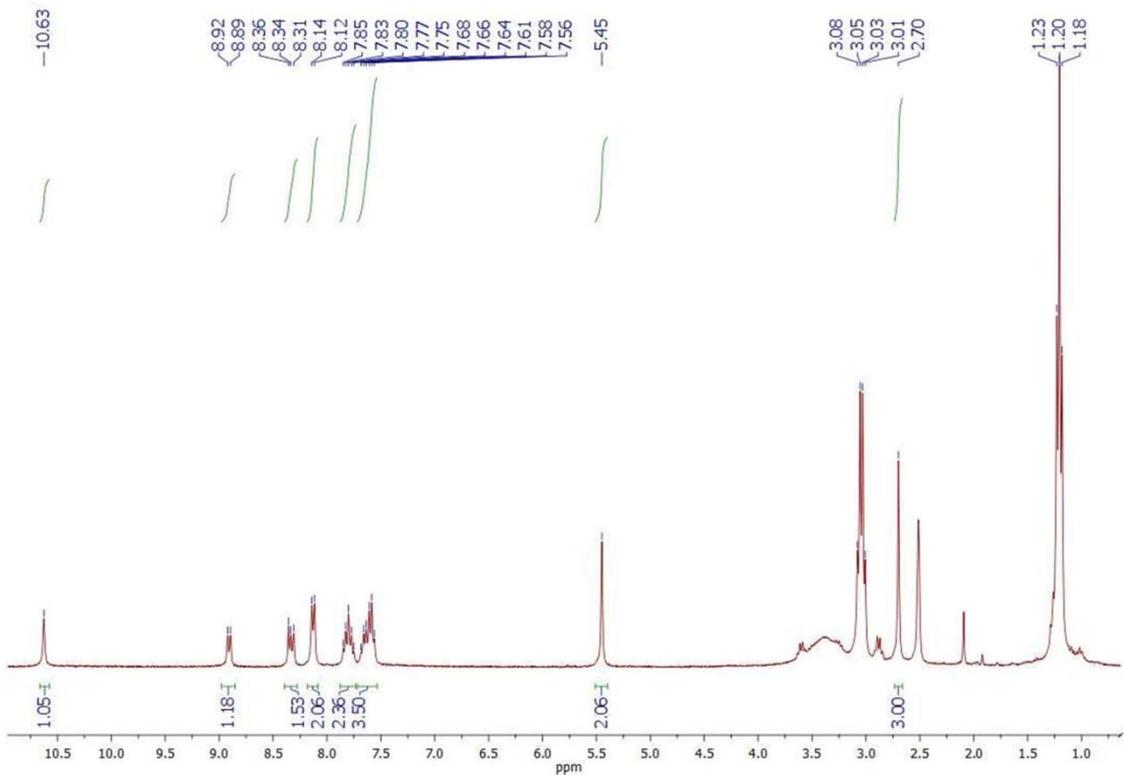
Amine: morpholine



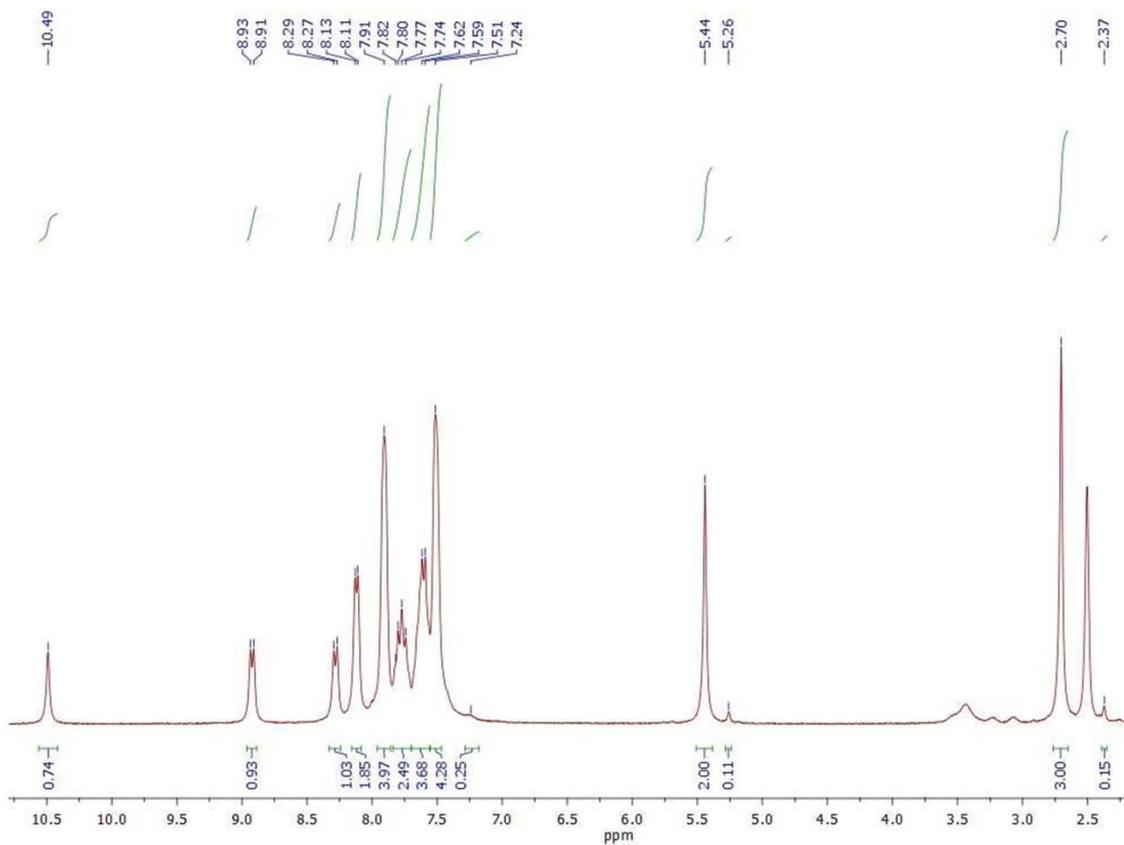
Amine: Et₃N (before irradiation)



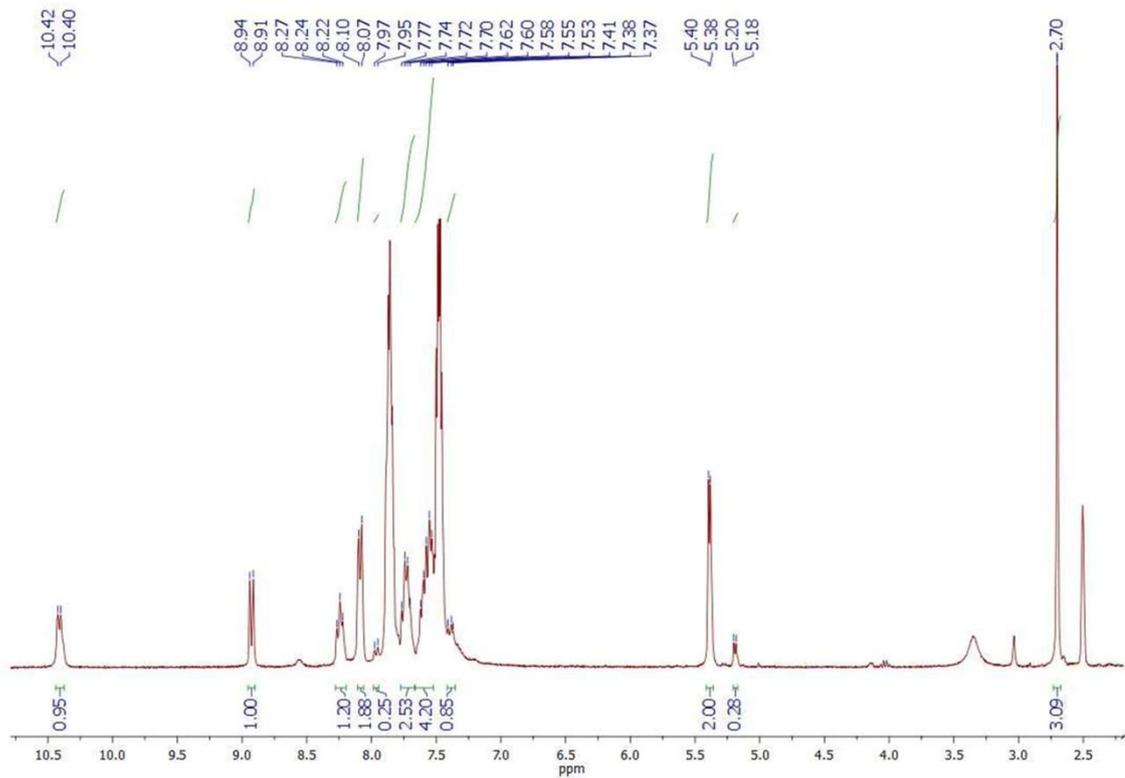
Amine: Et₃N (after irradiation)



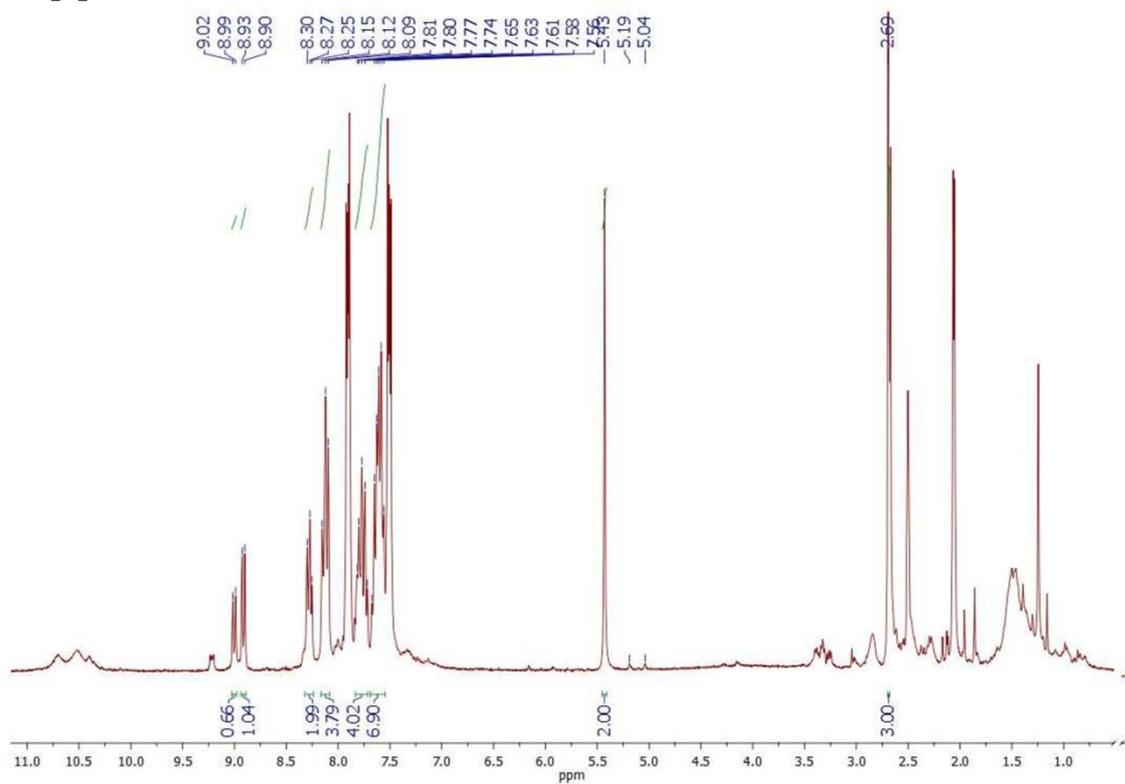
Amine: Et₃N (0.1 eq.)



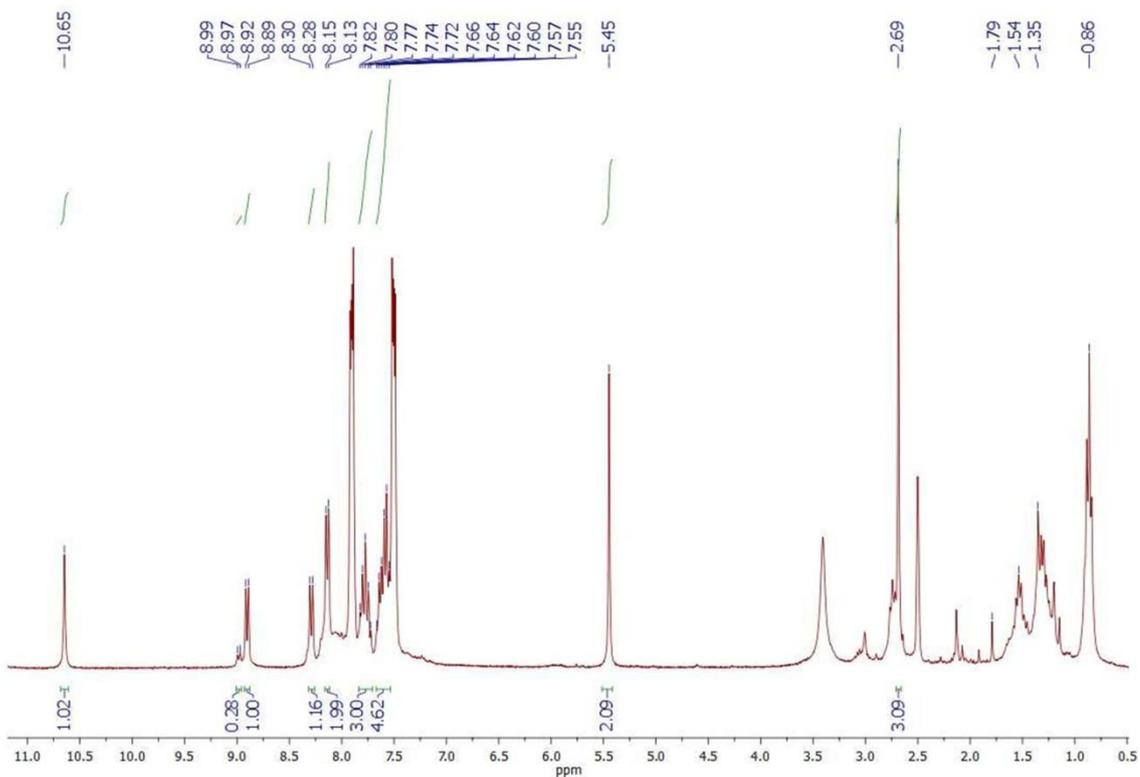
Amine: pyridine



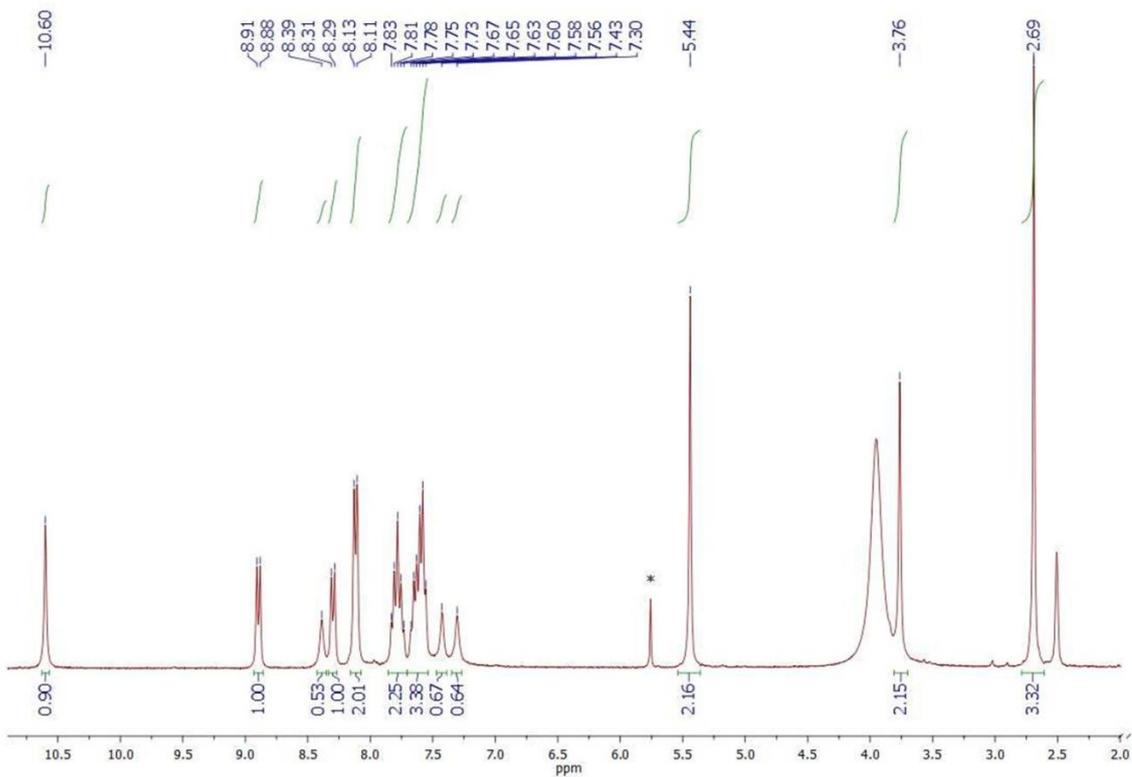
Amine: piperidine



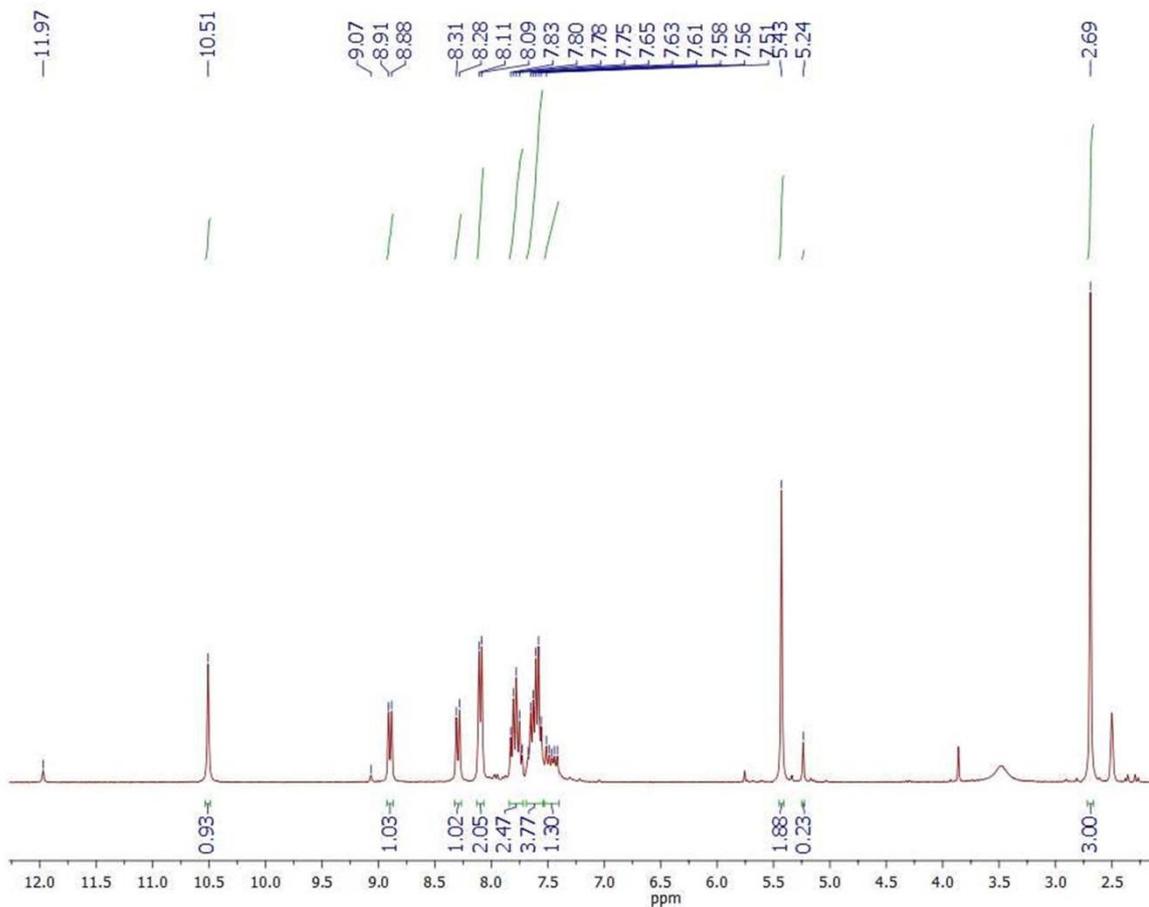
Amine: *n*-butylamine



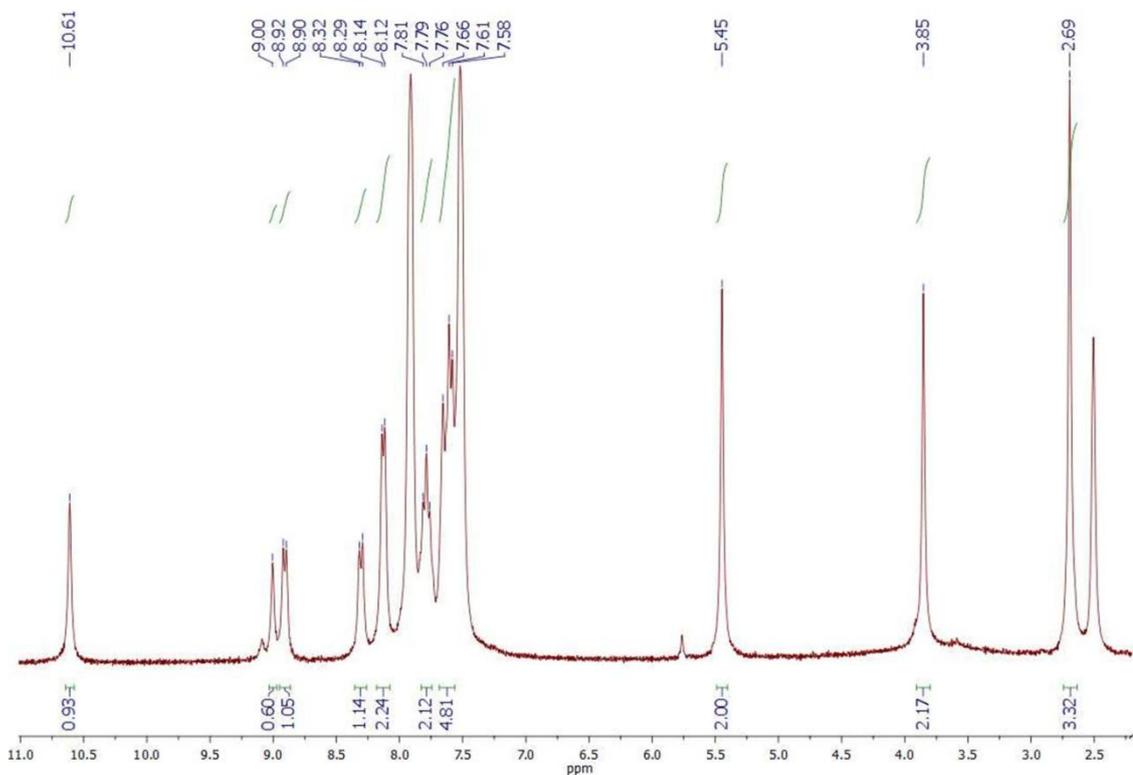
Amine: 1-methylimidazole (without naphthalene)



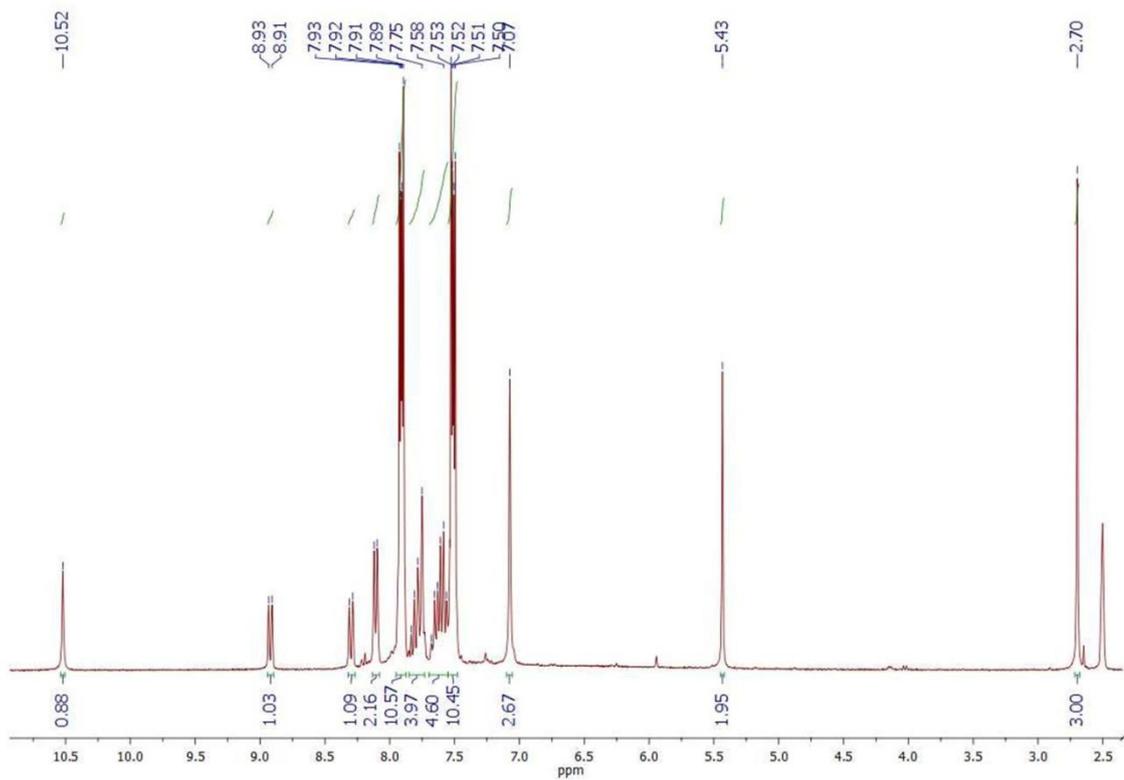
Amine: 1-methylimidazole (0.1 eq.)



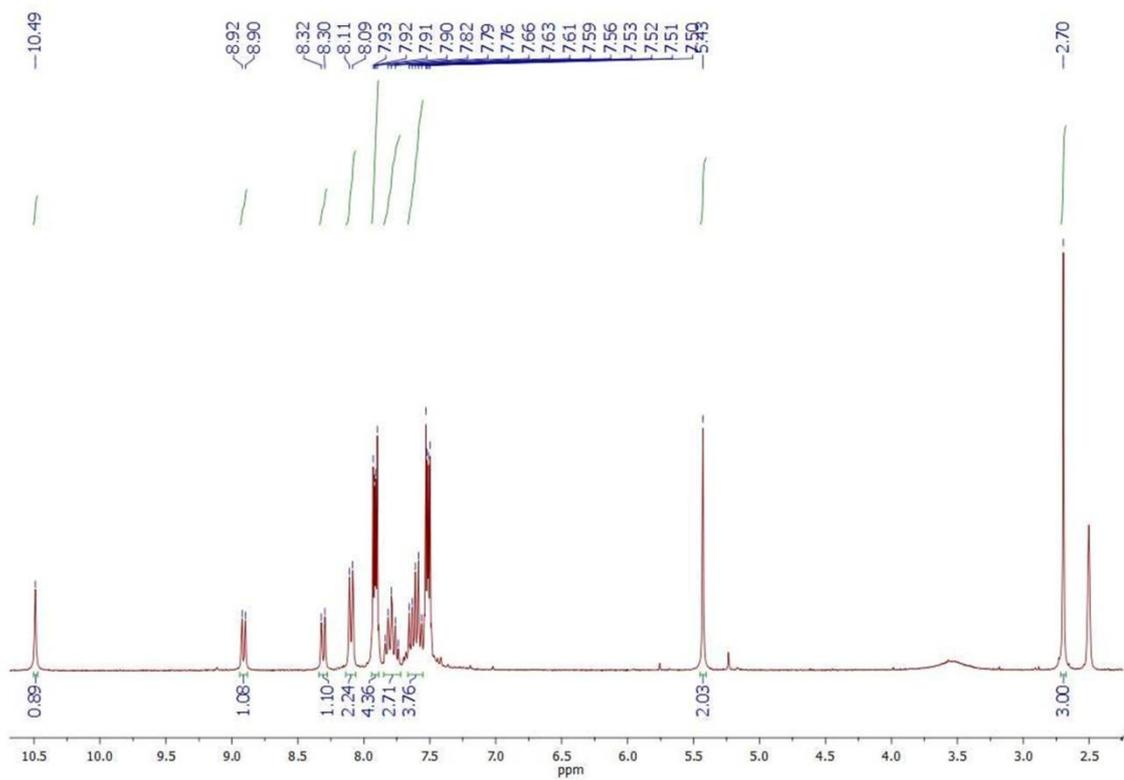
Amine: 1-methylimidazole (1 eq.)



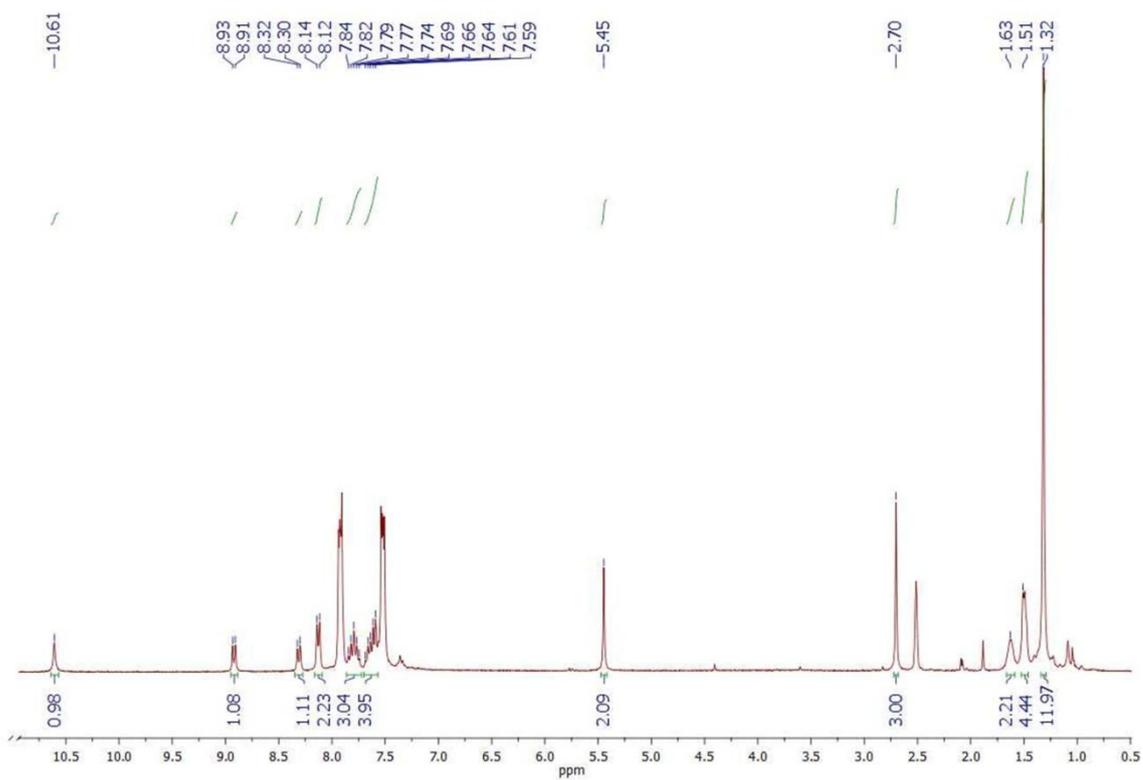
Amine: imidazole (1 eq.)



Amine: imidazole (0.1 eq.)

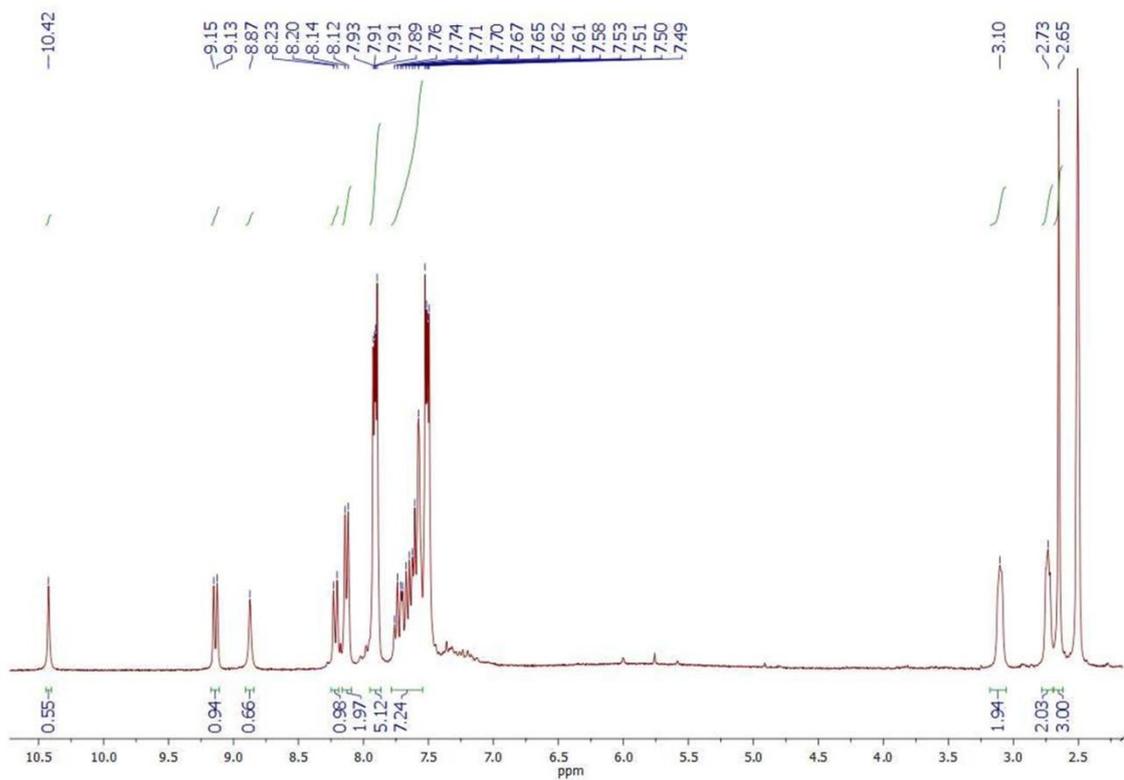


Amine: 2,2,6,6-tetramethylpiperidine

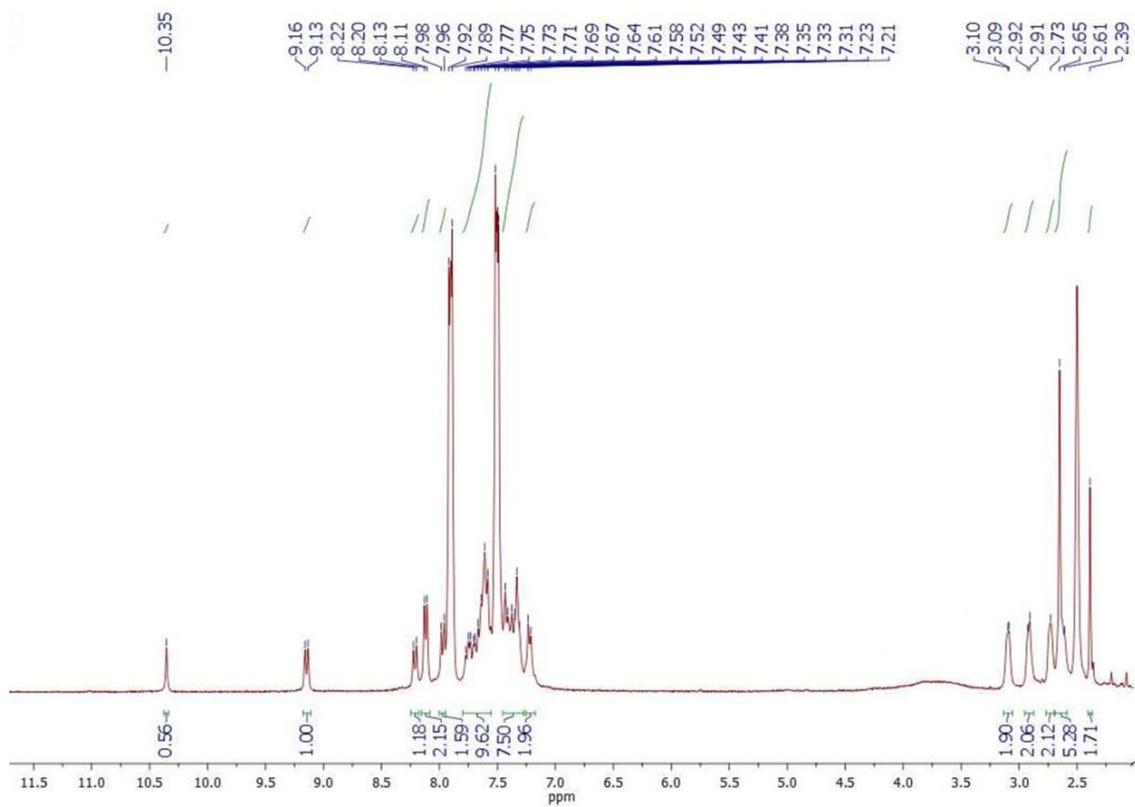


V. Effect of imidazole on the photoreaction of diarylethene **1b**

^1H NMR spectra of photoreaction of **1b** (40 mg in 2 ml of MeCN, 1 eq. of imidazole)

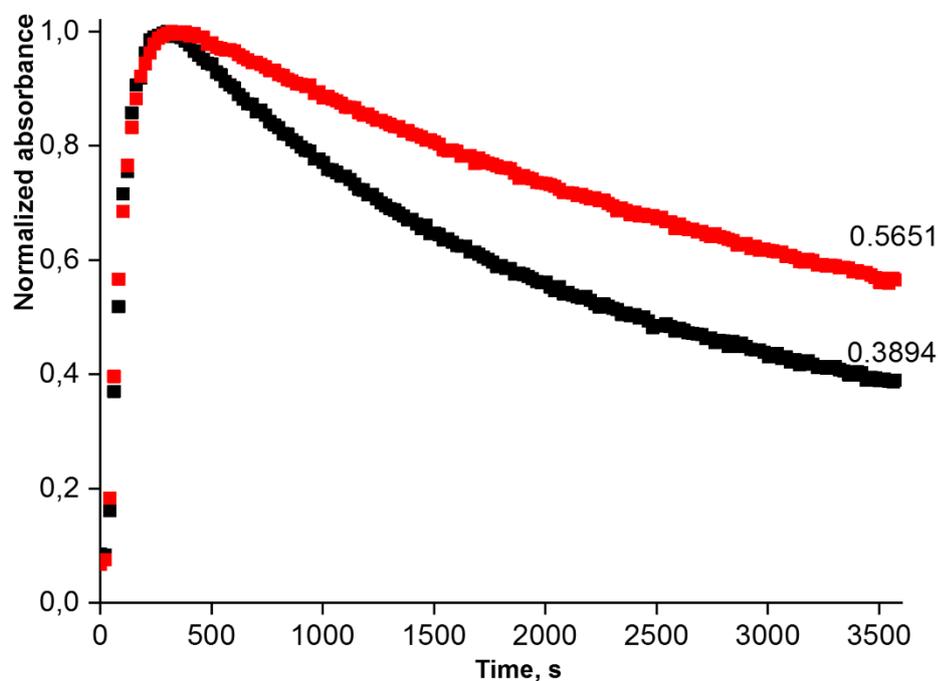


^1H NMR spectra of photoreaction of **1b** (40 mg in 2 ml of MeCN)

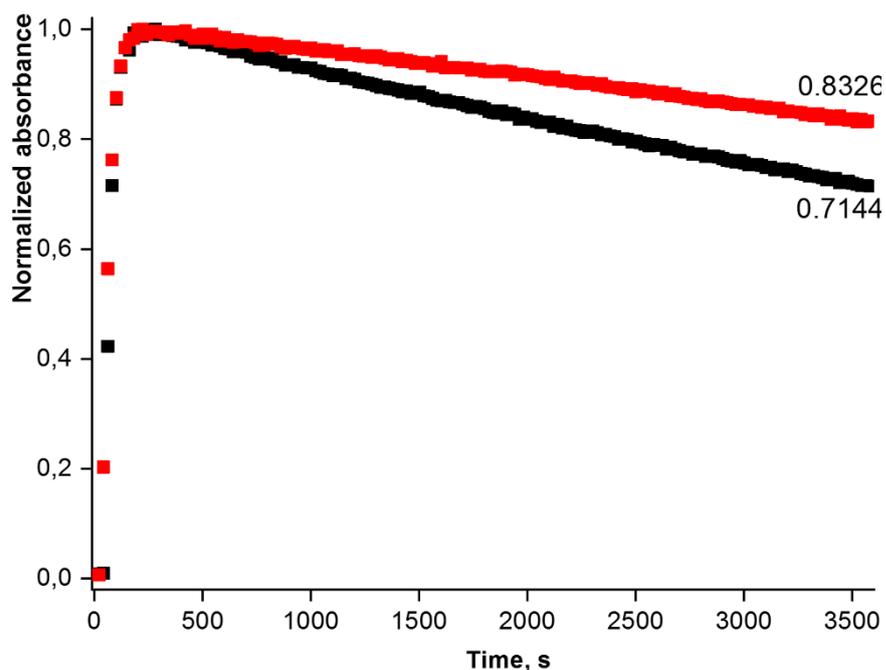


VI. Photostability of photochromic compounds

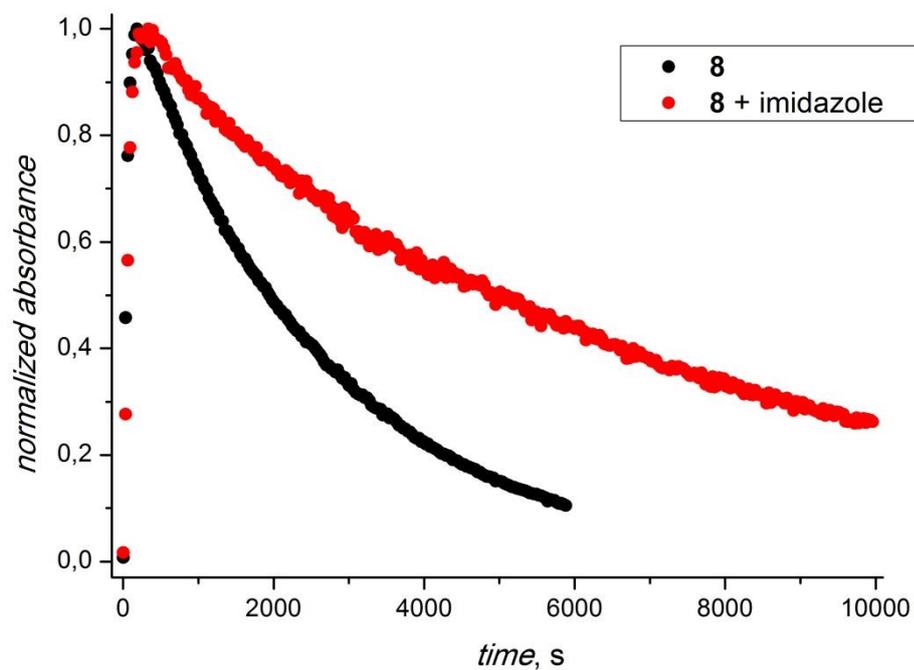
Dependence of the normalized optical density at the absorption maximum of photoinduced isomer **B** of diarylethene **5** on the irradiation time without (black points) and with imidazole (red points) (solvent – acetonitrile, $C = 1.4 \cdot 10^{-5}$ M, $C(\text{imidazole}) = 1.03 \cdot 10^{-3}$ M, $\lambda^{\text{irr}} = 365$ nm, $T = 293$ K).



Dependence of the normalized optical density at the absorption maximum of photoinduced isomer **B** of diarylethene **6** on the irradiation time without (black points) and with imidazole (red points) (solvent – acetonitrile, $C = 1.4 \cdot 10^{-5}$ M, $C(\text{imidazole}) = 1.03 \cdot 10^{-3}$ M, $\lambda^{\text{irr}} = 365$ nm, $T = 293$ K).

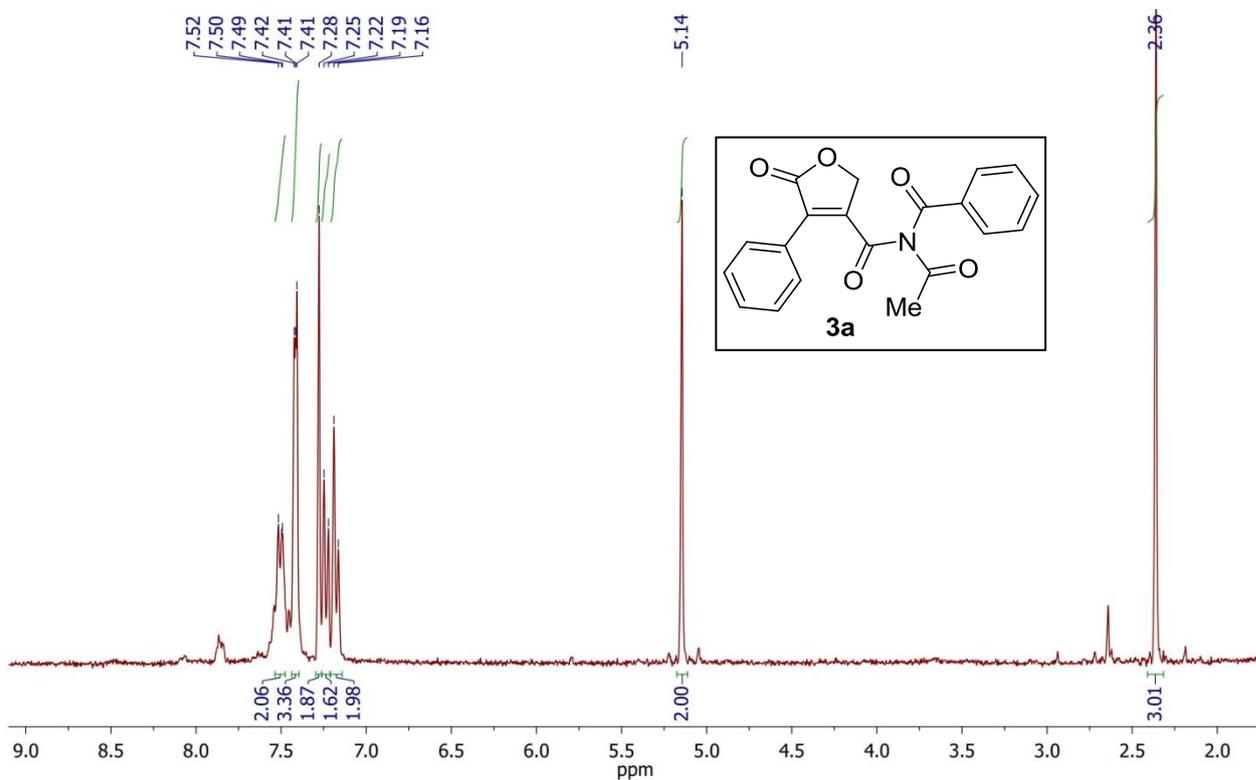


Dependence of the normalized optical density at the absorption maximum of photoinduced isomer **B** of spiropyran **8** on the irradiation time without (black points) and with imidazole (red points) (solvent – acetonitrile, $C = 1.4 \cdot 10^{-5}$ M, $C(\text{imidazole}) = 1.03 \cdot 10^{-3}$ M, $\lambda^{\text{irr}} = 365$ nm, $T = 293$ K).

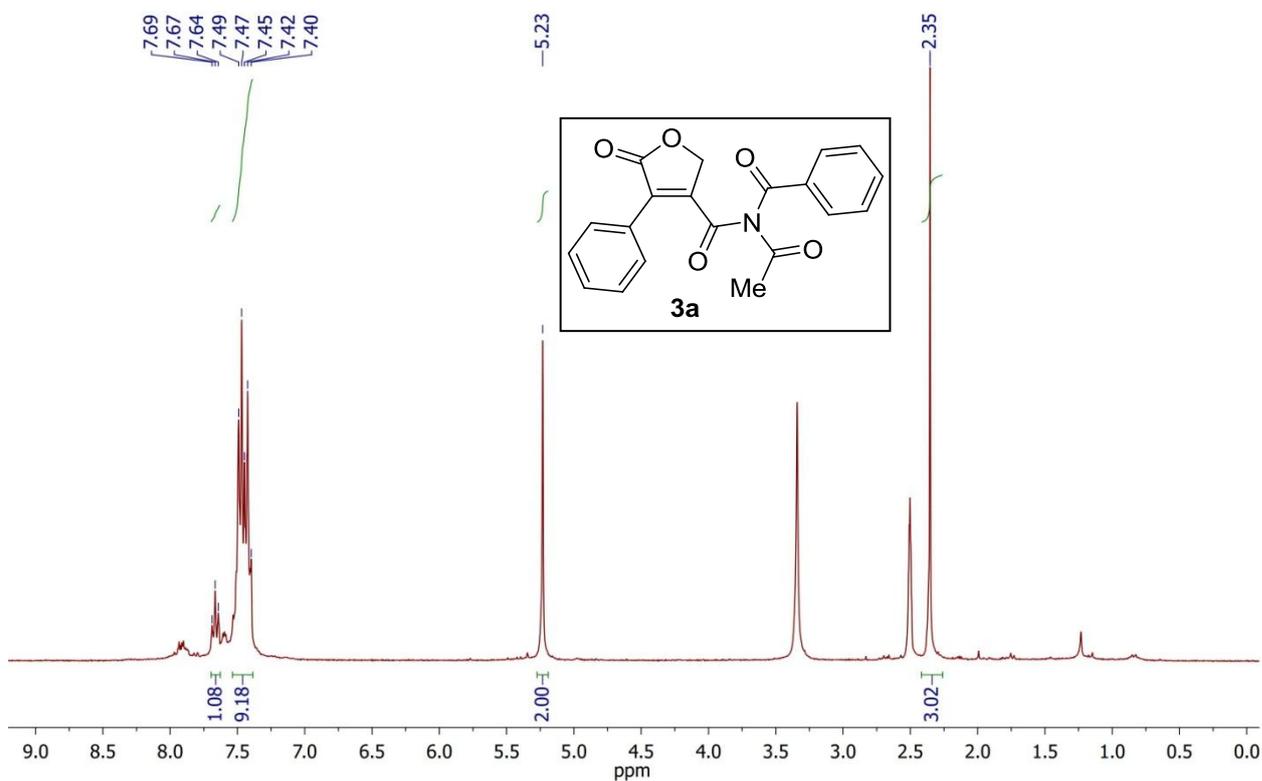


VII. Copies of NMR spectra

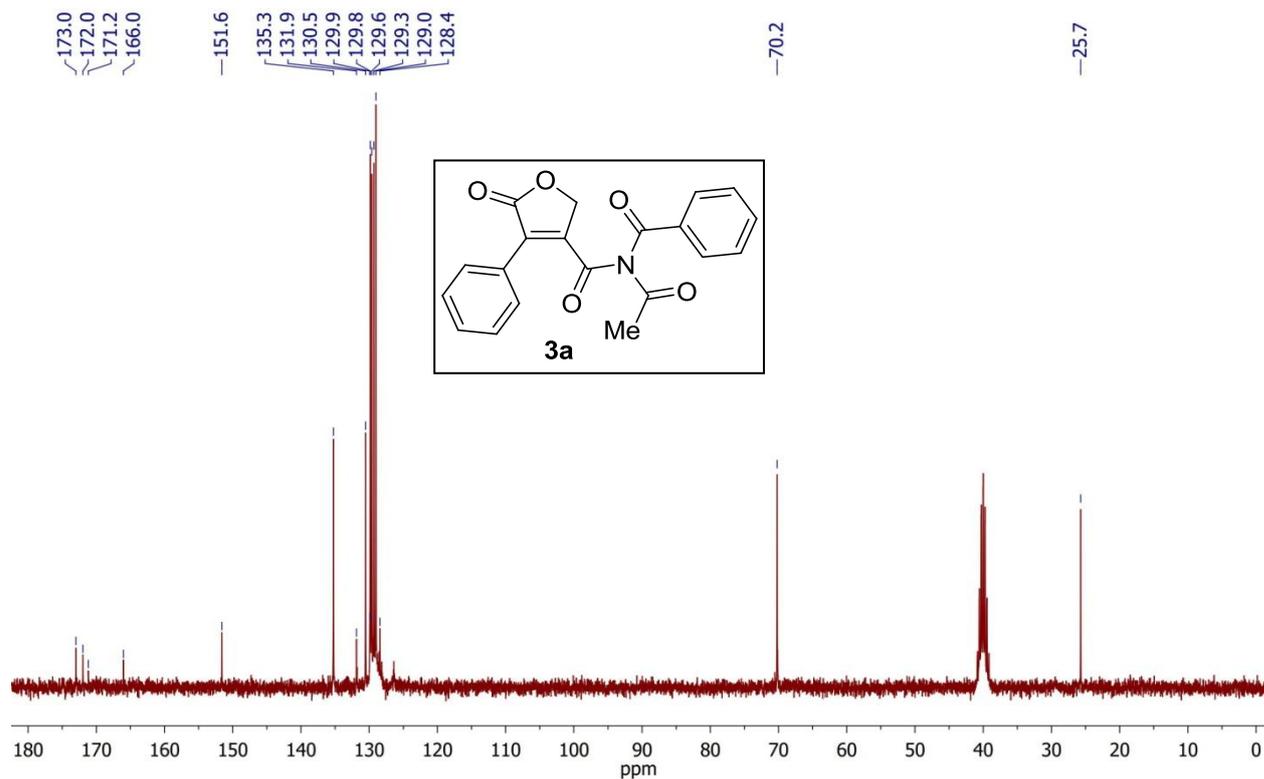
¹H NMR spectrum of compound 3a (CDCl₃)



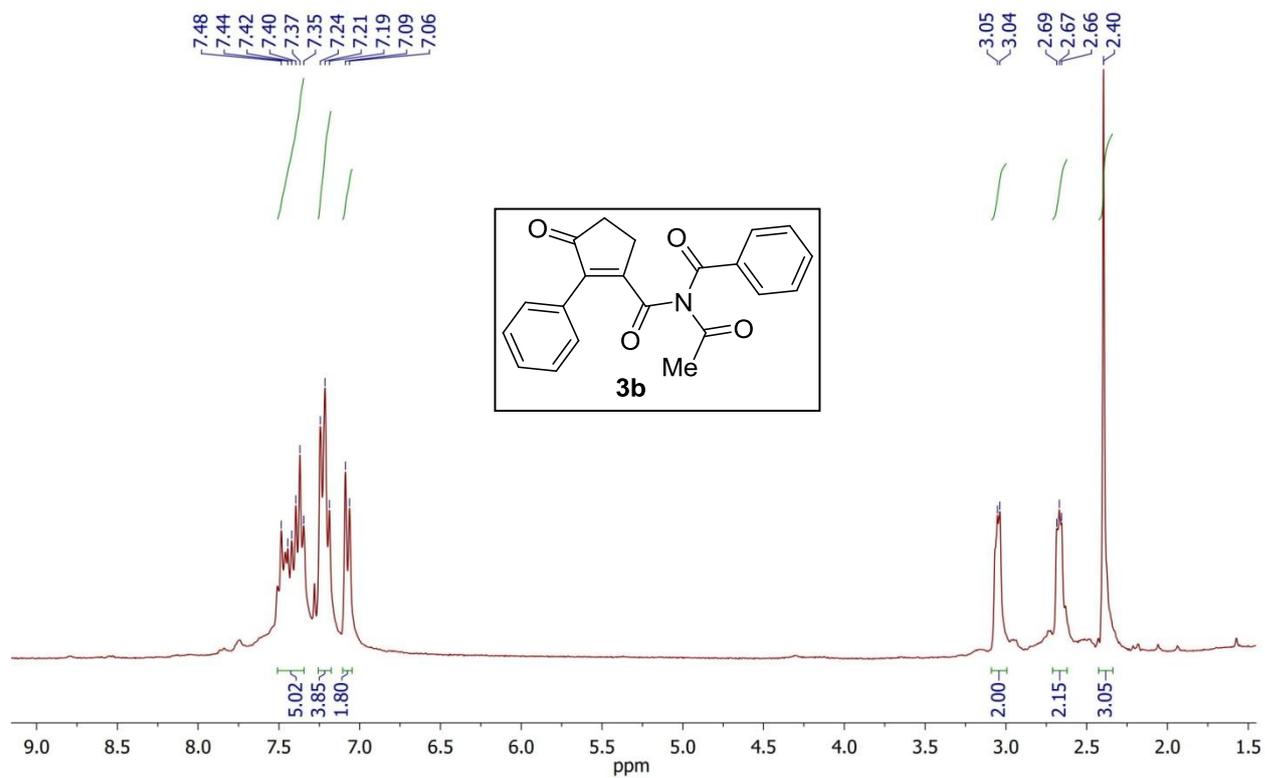
¹H NMR spectrum of compound 3a (DMSO-d₆)



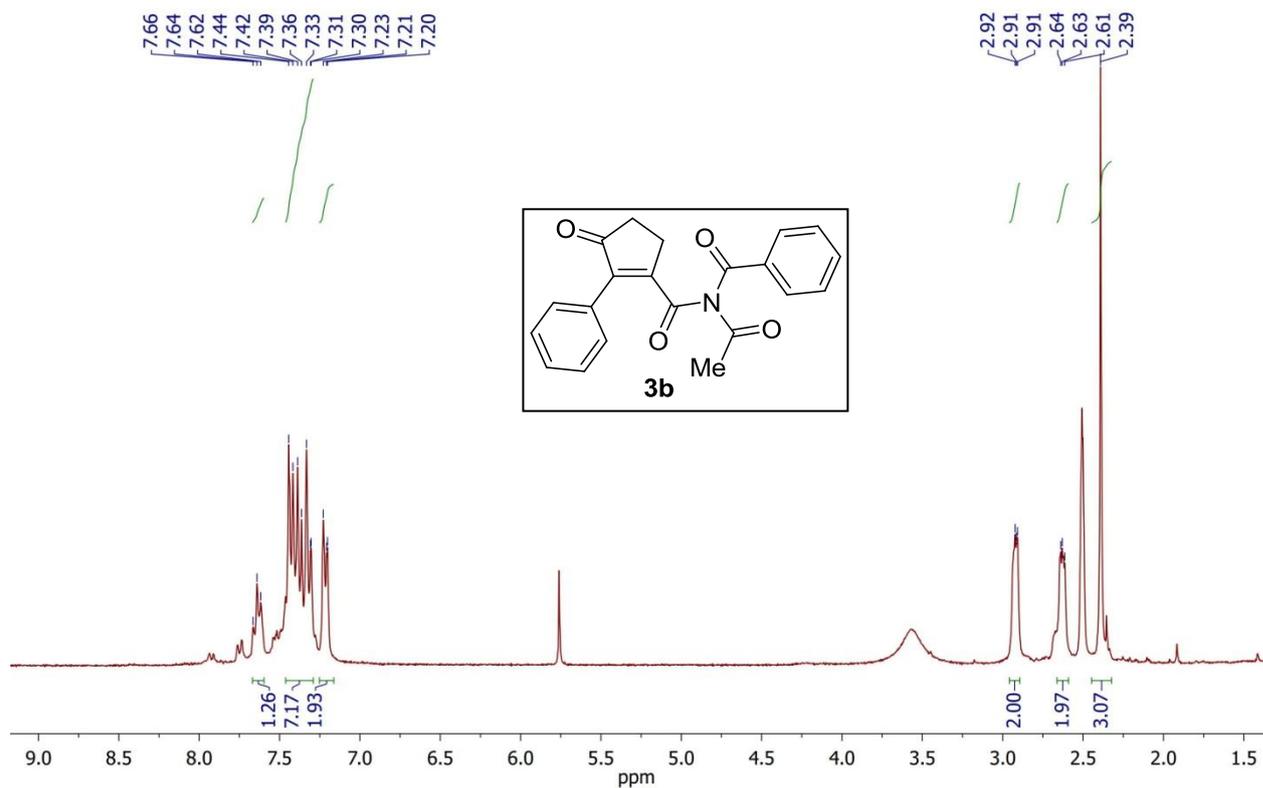
^{13}C NMR spectrum of compound 3a (DMSO- d_6)



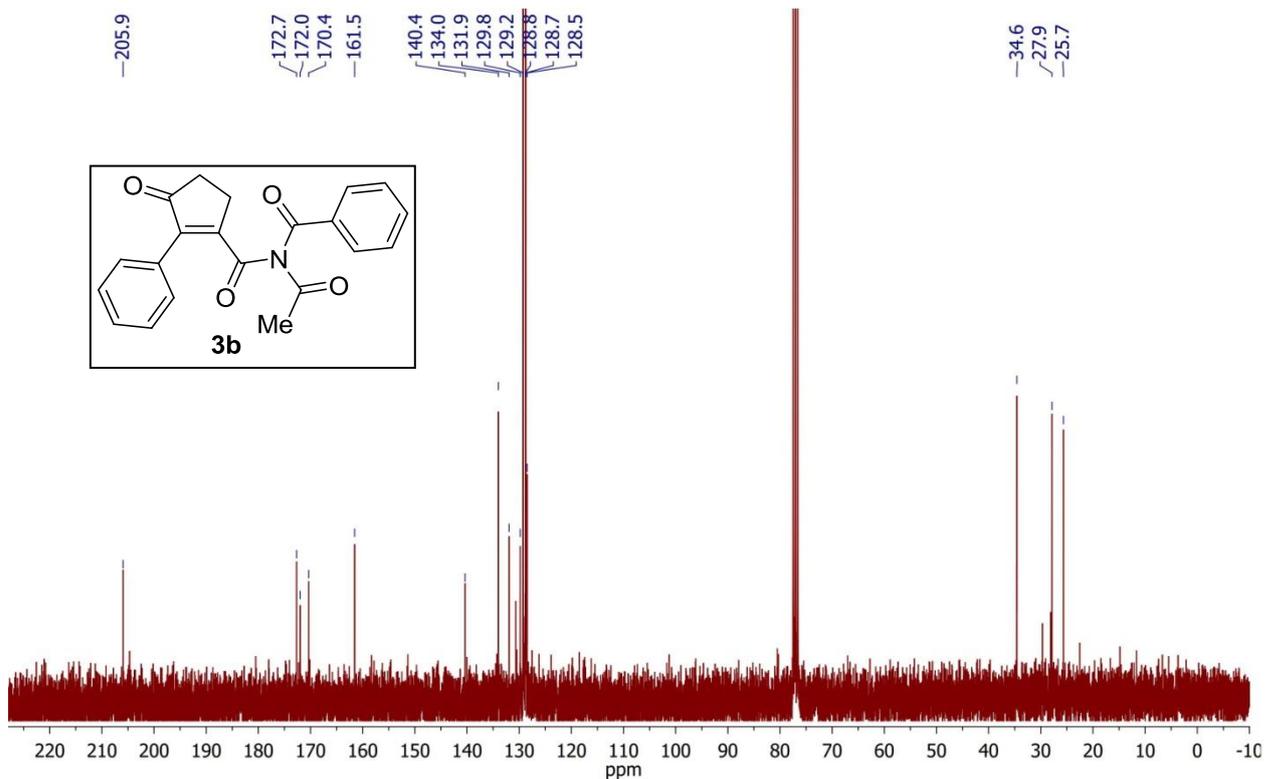
^1H NMR spectrum of compound 3b (CDCl_3)



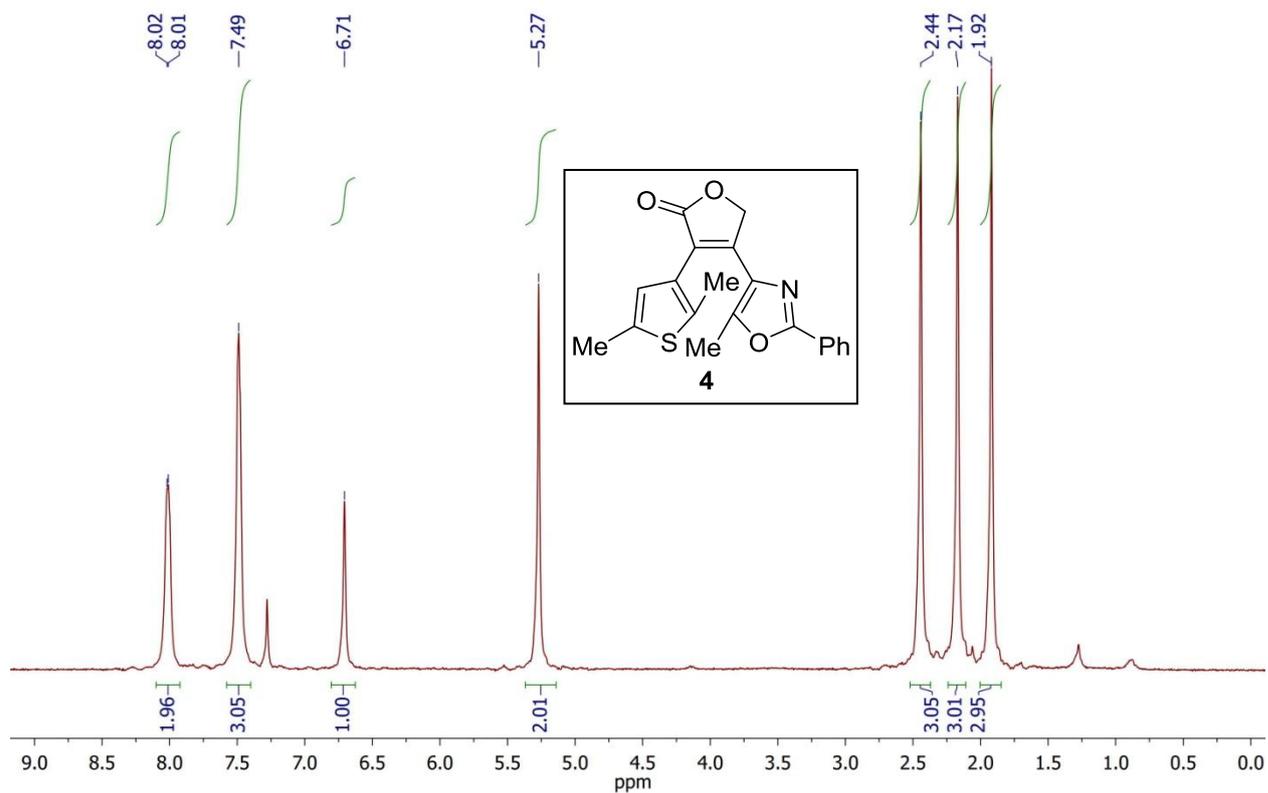
^1H NMR spectrum of compound 3b (DMSO- d_6)



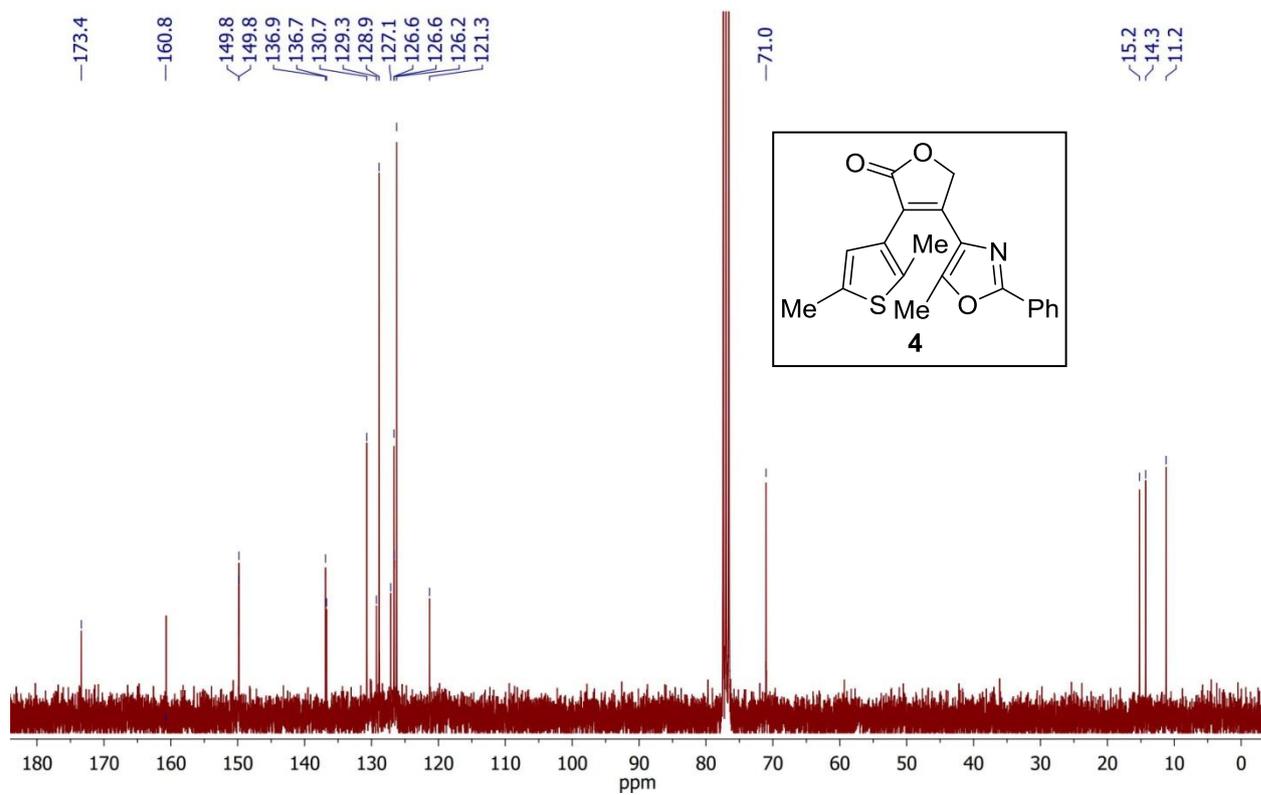
^{13}C NMR spectrum of compound 3b (CDCl_3)



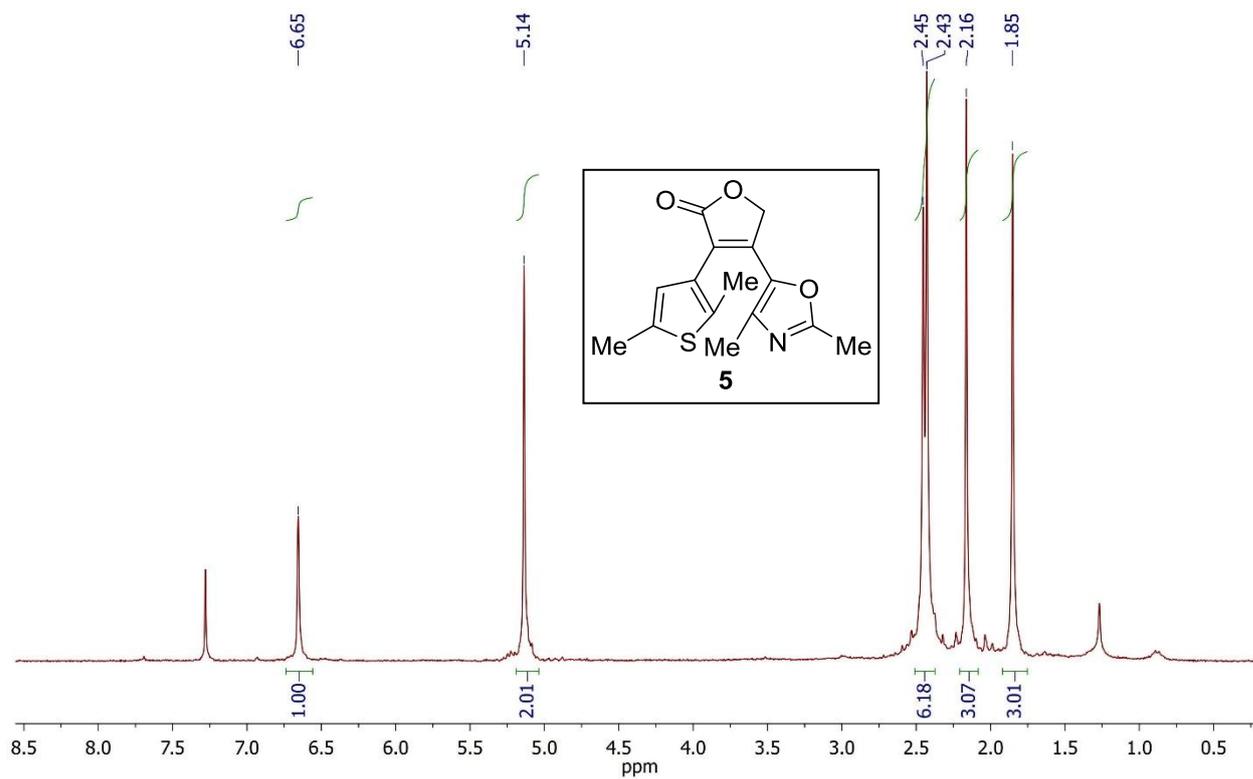
¹H NMR spectrum of compound 4



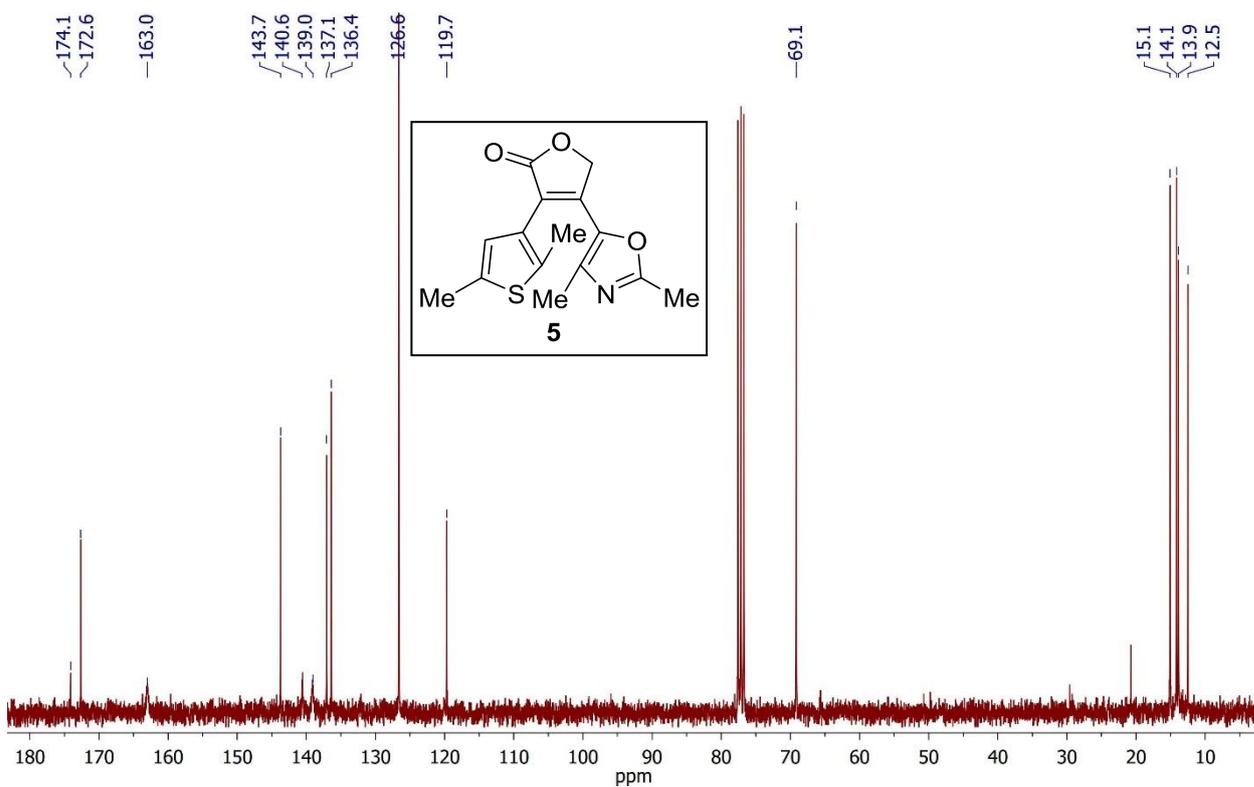
¹³C NMR spectrum of compound 4



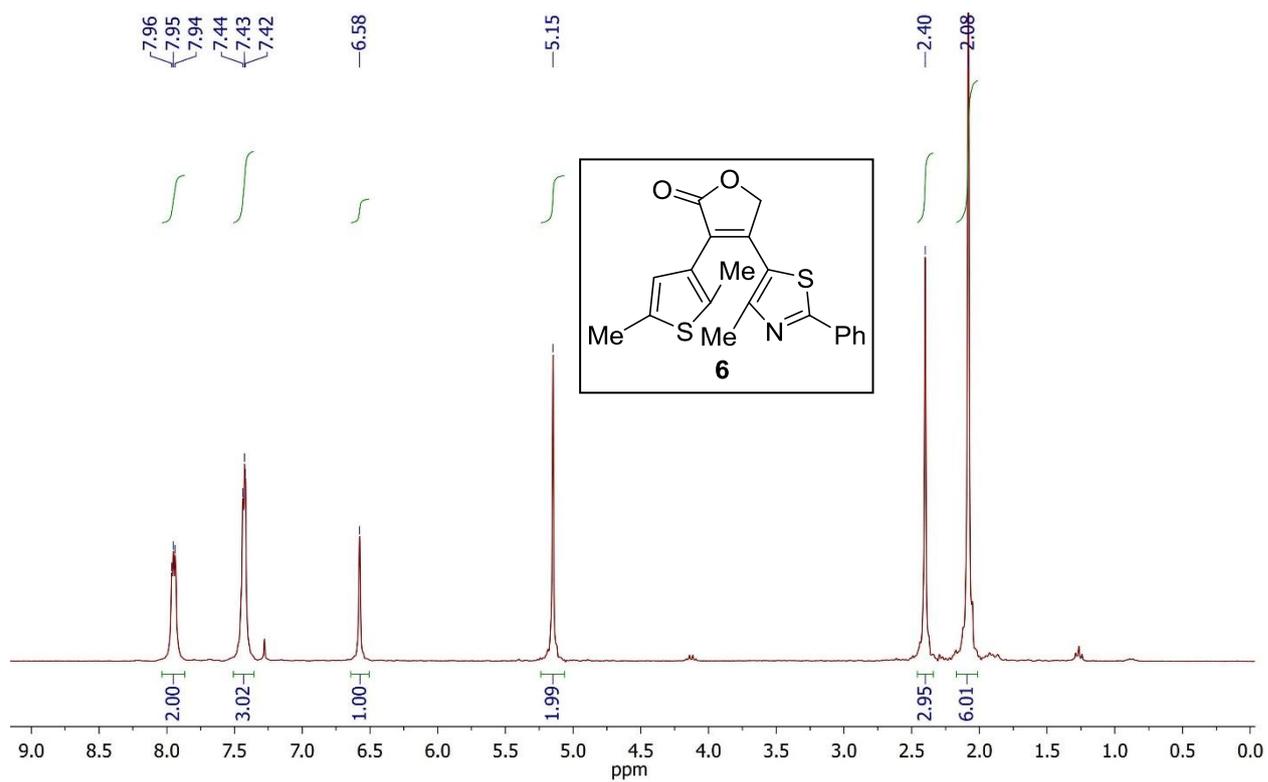
¹H NMR spectrum of compound 5



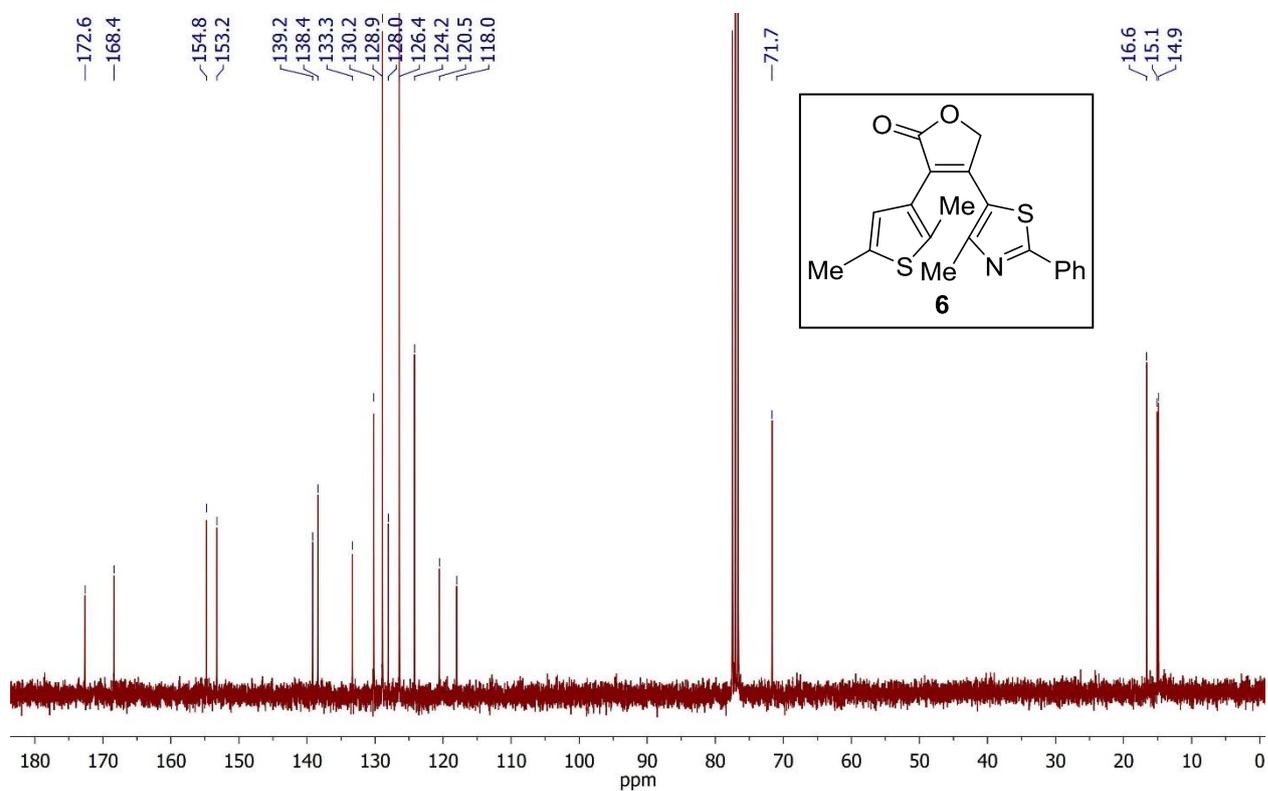
¹³C NMR spectrum of compound 5



¹H NMR spectrum of compound 6



¹³C NMR spectrum of compound 6

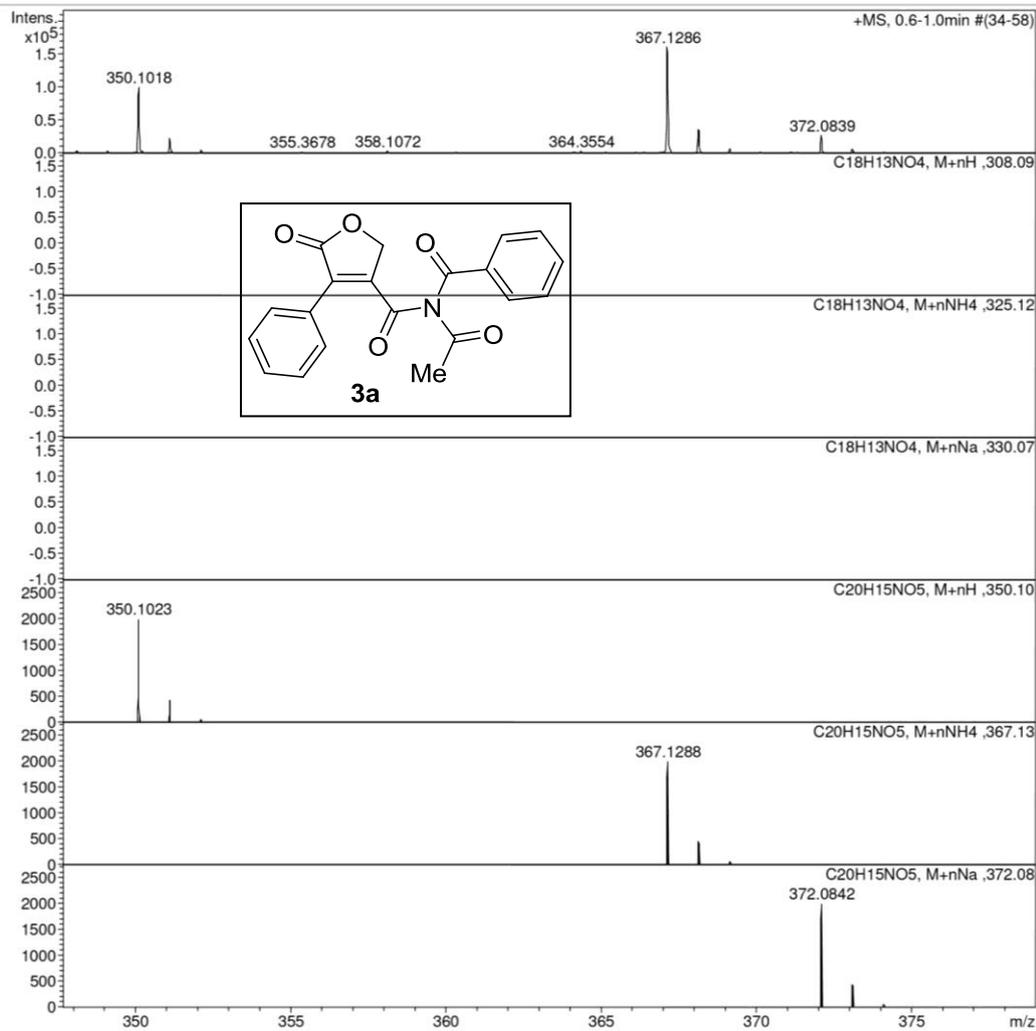


VIII. Copies of HRMS spectra

Compound 3a

Acquisition Parameter

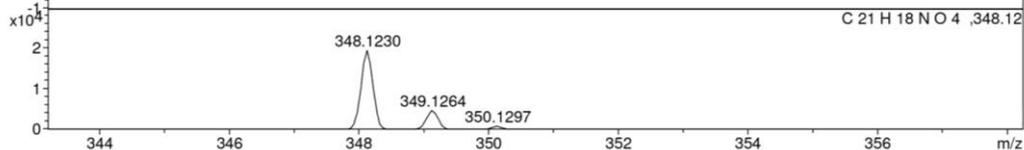
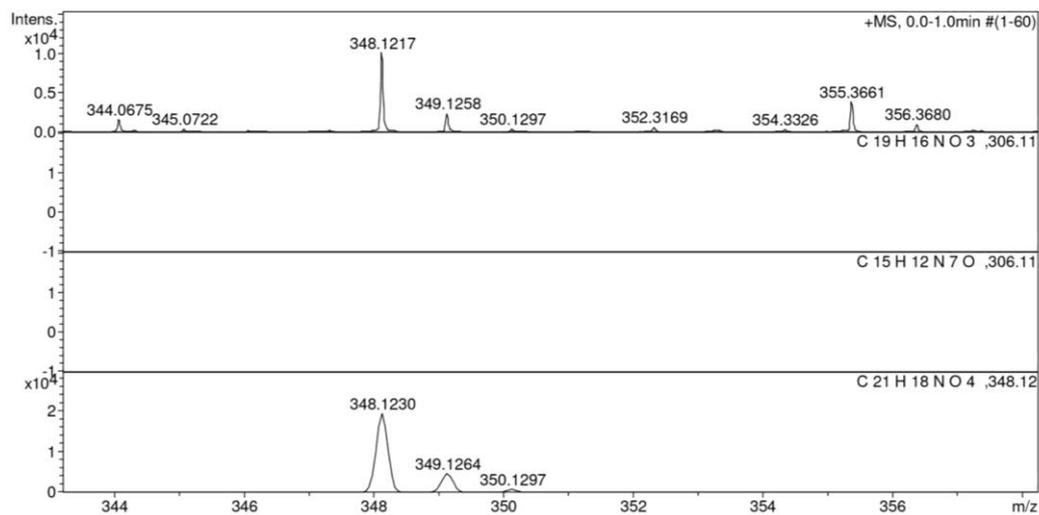
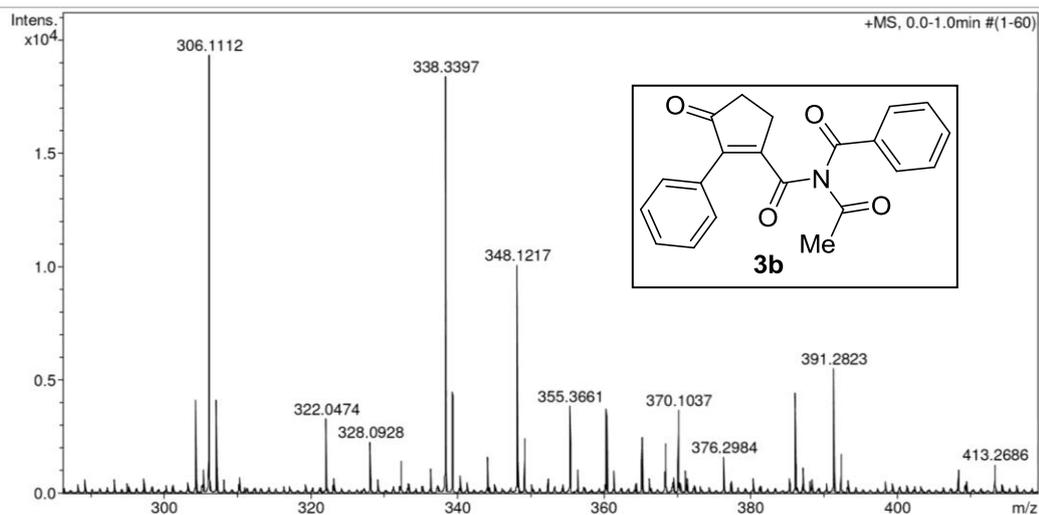
Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	4500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste



Compound 3b

Acquisition Parameter

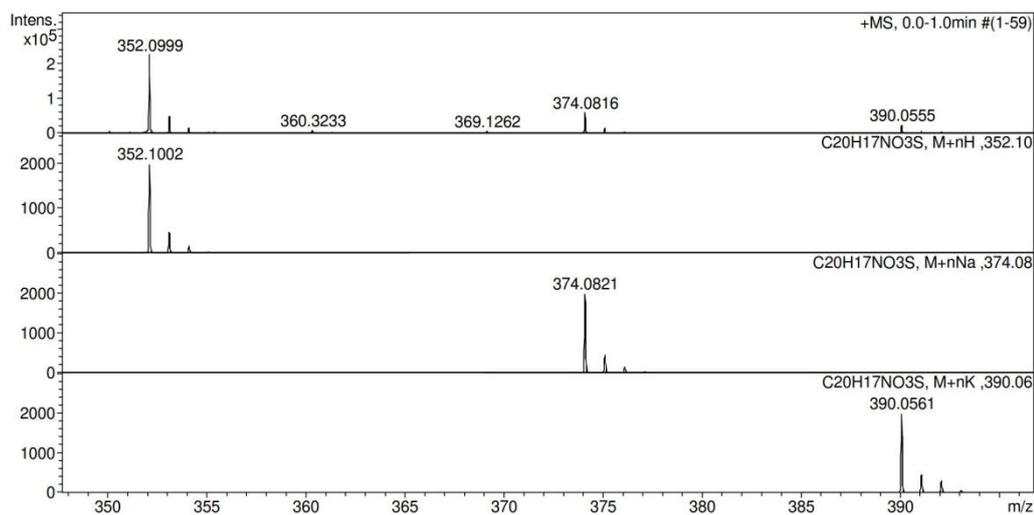
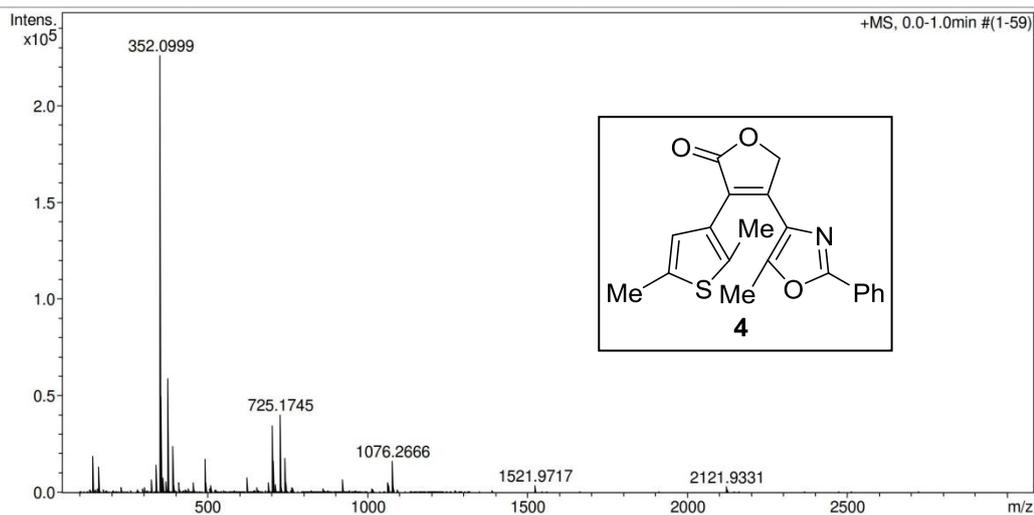
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Focus	Not active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	4500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste



Compound 4

Acquisition Parameter

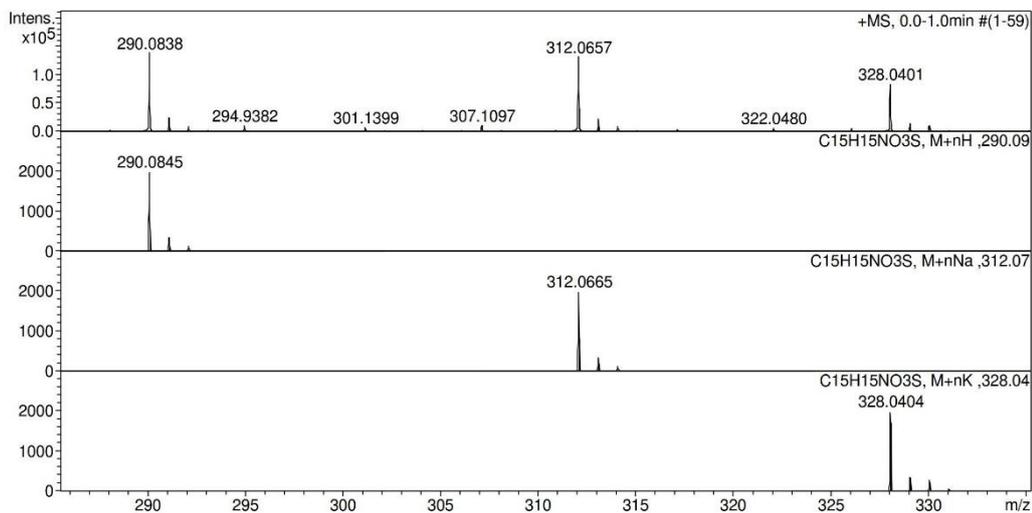
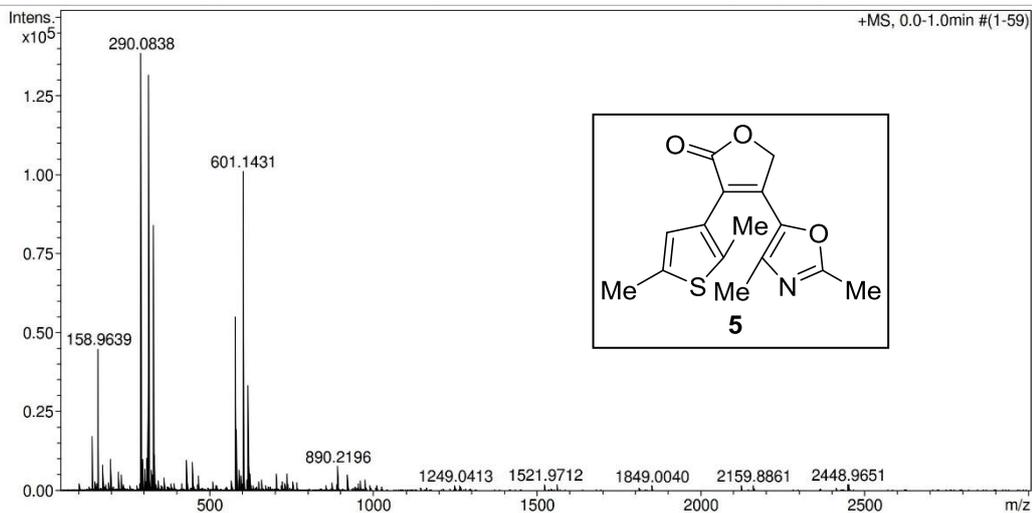
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Focus	Not active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	4500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste



Compound 5

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	4500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste



Compound 6

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	4500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste

