

*Supporting Information for*

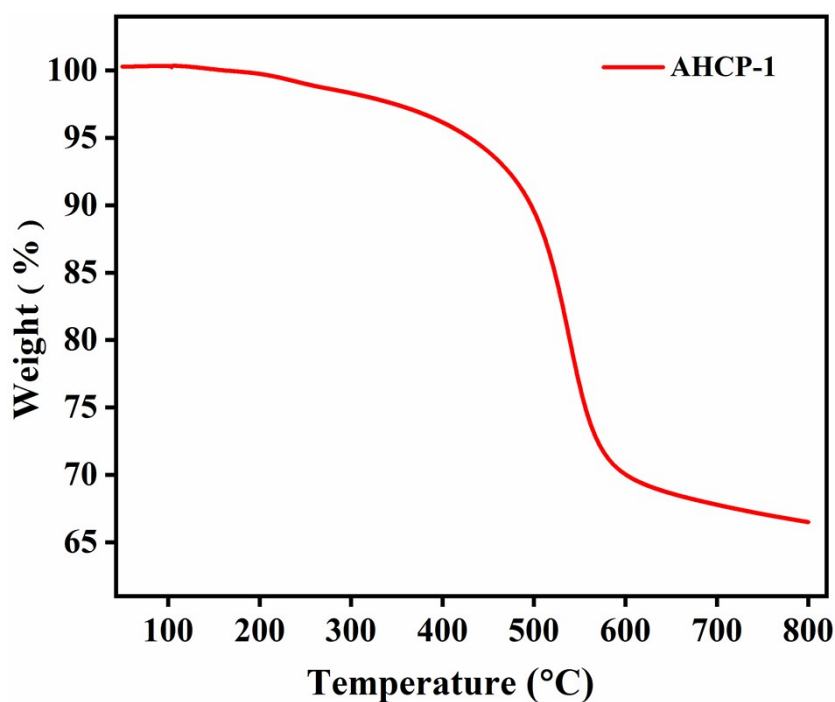
**One-pot synthesis of highly porous anionic hypercrosslinked polymer for ultrafast adsorption of organic pollutants**

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**Fig. S1** TG curve of AHCP-1 under N<sub>2</sub> atmosphere.

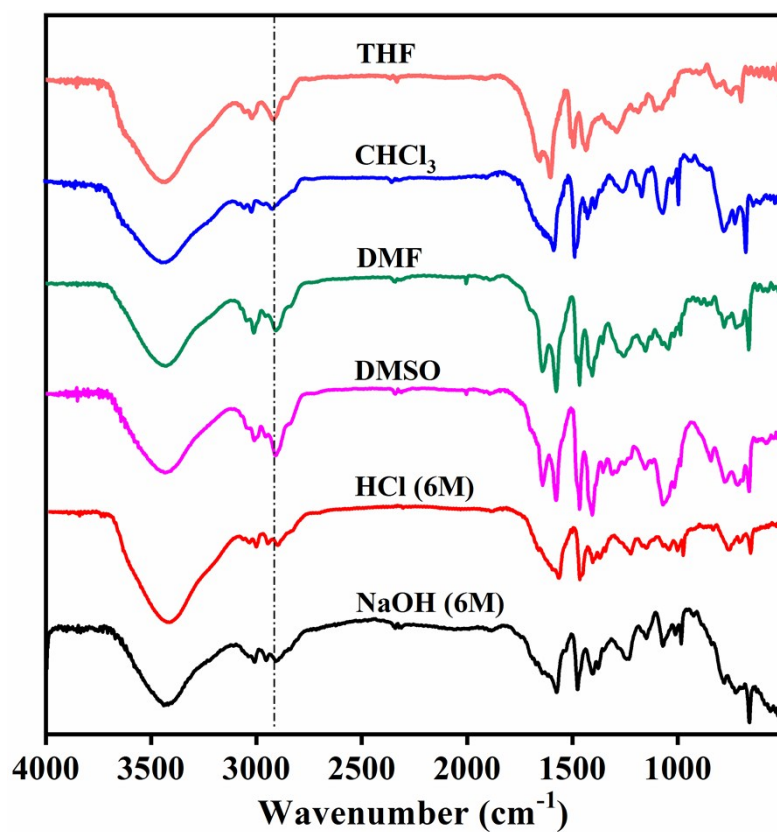


Fig. S2 FT-IR of AHCP-1 after treatment for one week in different solvents.

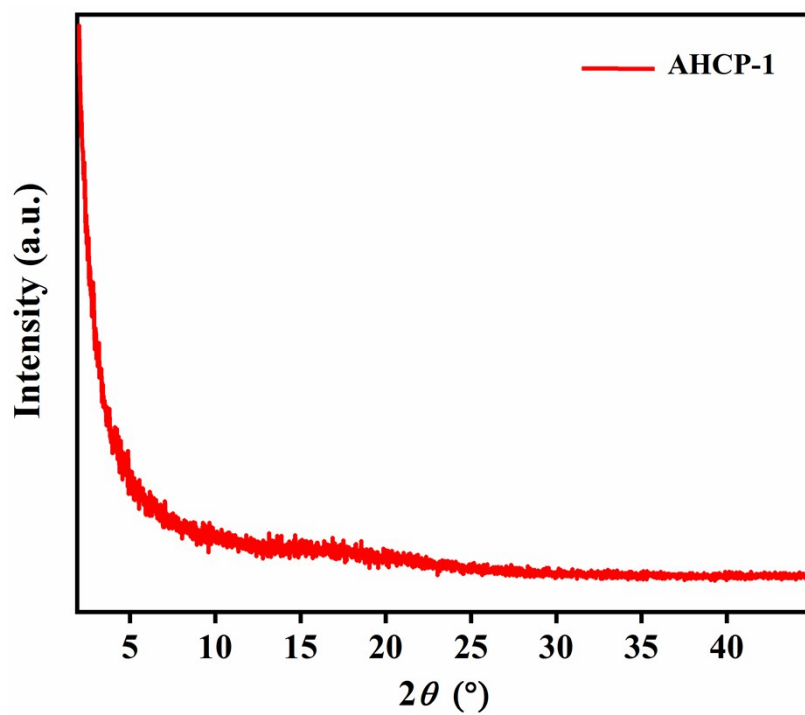


Fig. S3 The PXRD pattern of AHCP-1 at wide angle showing broad diffraction for thick amorphous pore wall.

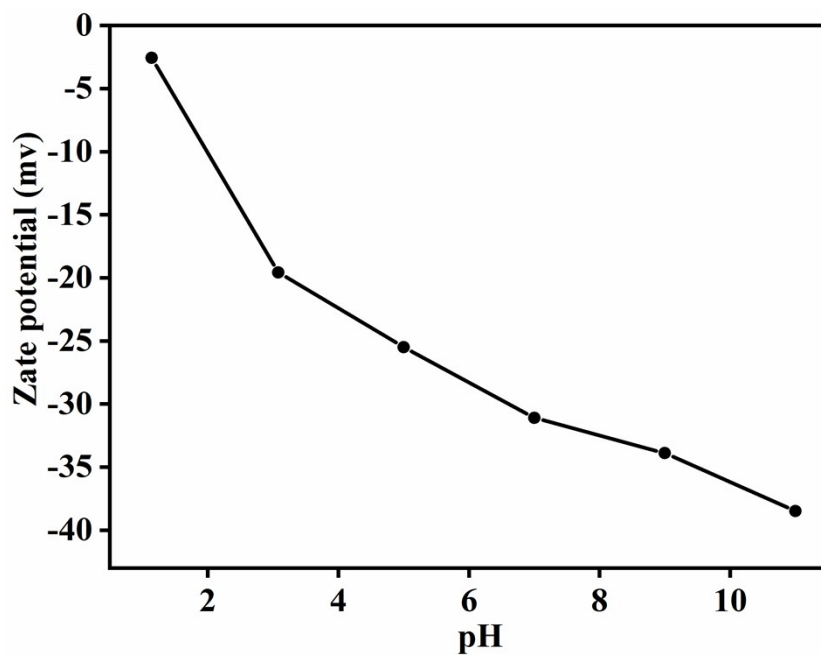


Fig. S4 Zeta potential vs. pH value of AHCP-1.

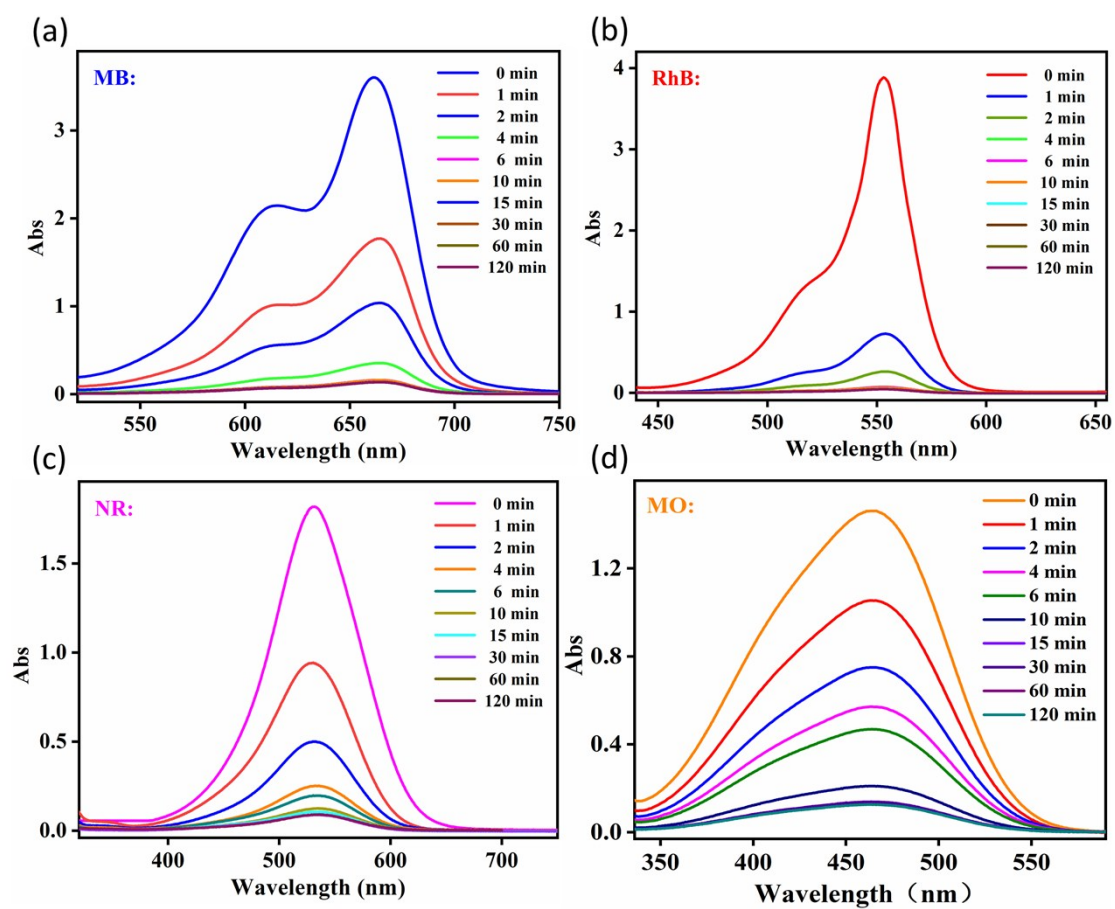
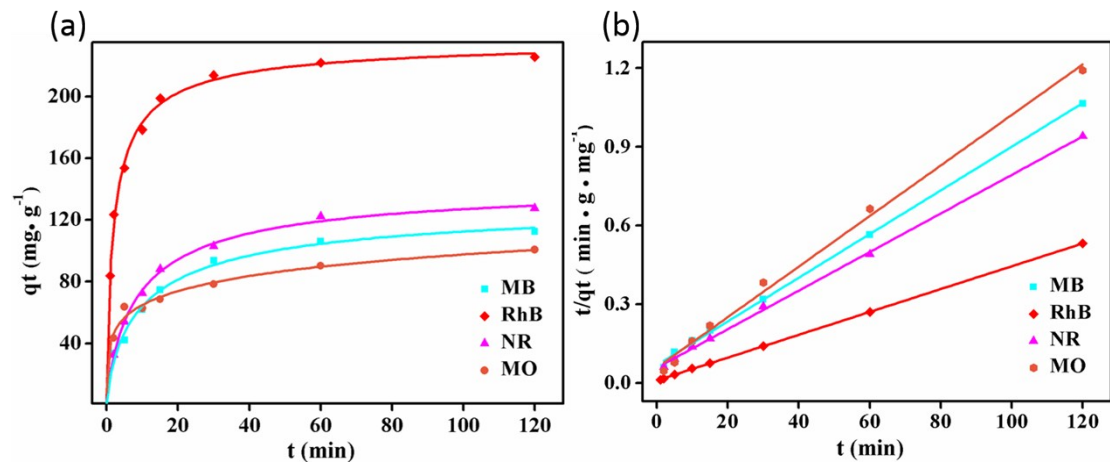
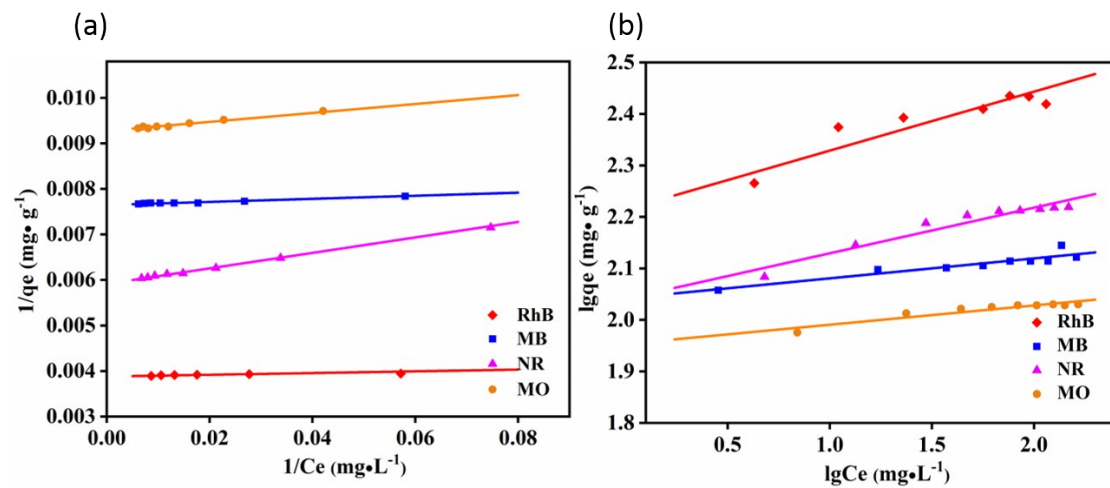


Fig. S5 Time-dependant electronic absorption spectral change (from 0 to 120 minutes) of the dyes at initial concentration of  $20 \text{ mg} \cdot \text{L}^{-1}$ . (a) MB, (b) RhB, (c) NR, (d) MO.



**Fig. S6** (a) Effect of contact time on the dyes adsorption at initial concentration of  $200 \text{ mg L}^{-1}$ . (b) Pseudo-second-order kinetics of dyes adsorption.



**Fig. S7** Linear regression using Langmuir isotherm model (a) and Freundlich isotherm model (b) for RhB, MB, NR, and MO on AHCP-1.

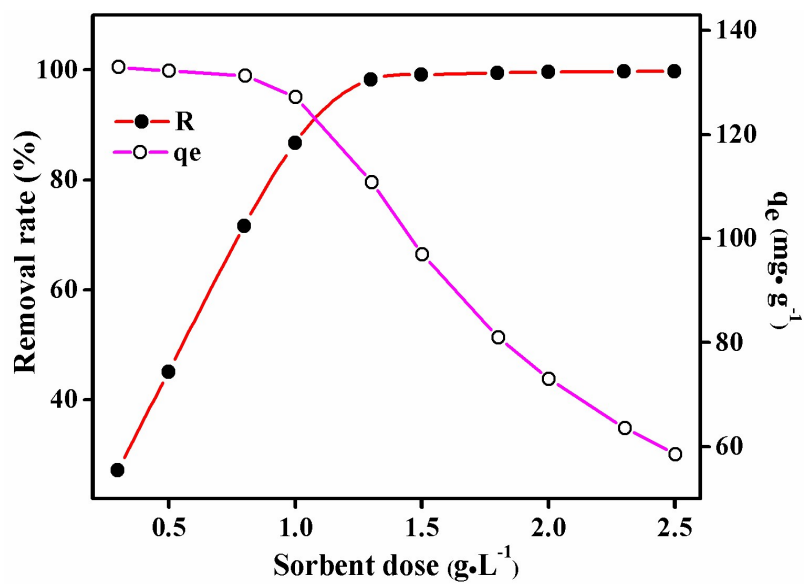


Fig. S8 The effect of AHCP-1 dose on the adsorption of MB.

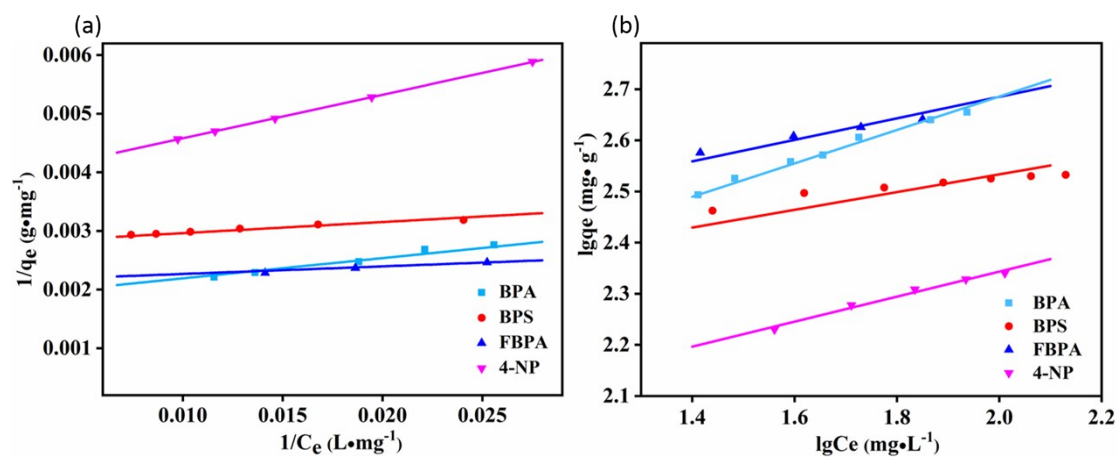
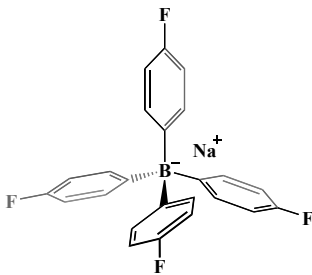
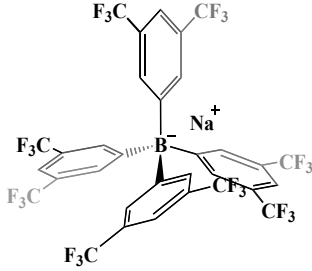
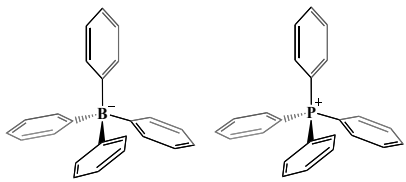
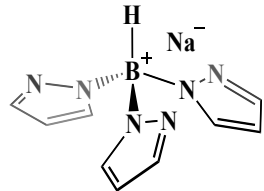


Fig. S9 Linear regression using Langmuir isotherm model (a) and Freundlich isotherm model (b) for BPA, BPS, FBPA, and 4-NP on AHCP-1.

**Table S1.** Additional six building blocks and yields of hypercrosslinking reactions.

No.	Chemical names	Building units	Yield
1	Sodium tetrakis(4-fluorophenyl)borate		33.5%
2	Sodium tetrakis[3,5-bis(trifluoromethyl)phenyl] borate		0.4%
3	Tetraphenylphosphonium tetraphenylborate		74%
4	Sodium tris(1-pyrazolyl)borohydride		0%

**Table S2.** Average weight change before and after soaking in solvent for one week.

Solvents	Original weight	Soaked weight	Weight retention rate
	(mg)	(mg)	
THF	20.3	19.6	96.6%
CHCl <sub>3</sub>	20.1	20.0	98.0%
DMF	20.7	20.0	96.6%
DMSO	19.8	18.9	95.4%
aqueous HCl (6 M)	19.5	19.1	97.9%
aqueous NaOH (6 M)	20.2	19.8	98.0%

**Table S3.** Isotherm Parameters for the Adsorption of dyes onto **AHCP-1**

	Langmuir isotherm			Freundlich isotherm		
	$b$ (L·mg <sup>-1</sup> )	$q_{\max}$ (mg·g <sup>-1</sup> )	$R^2$	$K_f$ (L·mg <sup>-1</sup> )	$n$	$R^2$
MB	2.263	130.7	0.996	112.5	30.99	0.838
RhB	1.879	258.4	0.997	158.4	9.081	0.916
NR	0.348	169.2	0.975	96.36	54.17	0.918
MO	0.944	107.9	0.981	88.81	25.08	0.865

**Table S4.** Summary of the maximum adsorption capacity ( $q_{\max}$ ) of Rhodamine B (RhB) on various adsorbents.

Adsorbent	$q_{\max}/\text{mg g}^{-1}$	References
C_carnauba_CaCl <sub>2</sub>	39.218	S1
Char AC	189.8	S2
MoS <sub>2</sub>	49.2	S3
MWCNT-COOH	42.68	S4
Carboxy-GO/zeolite	34.13	S5
FGNC	33.2	S6
Cd <sub>6</sub> (L) <sub>2</sub> (bib) <sub>2</sub> (DMA) <sub>4</sub>	67	S7
<b>AHCP-1</b>	<b>255</b>	<b>This work</b>

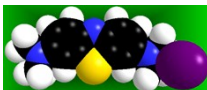
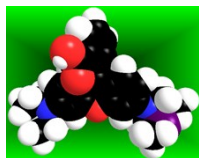
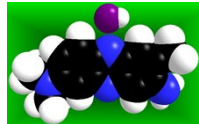
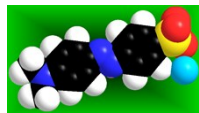

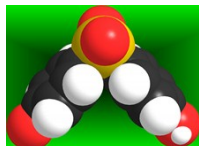
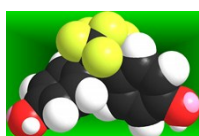
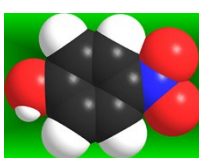
**Table S5.** Isotherm Parameters for the Adsorption of Phenolic derivatives micropollutants onto **AHCP-1**.

	Langmuir isotherm			Freundlich isotherm		
	$b$ (L · mg <sup>-1</sup> )	$q_{\max}$ (mg · g <sup>-1</sup> )	$R^2$	$K_f$ (L · mg <sup>-1</sup> )	$n$	$R^2$
BPA	0.054	540.5	0.992	0.310	3.059	0.986
BPS	0.146	361.1	0.994	0.339	5.767	0.877
FBPA	0.165	467.2	0.993	0.354	4.751	0.987
4-NP	0.052	260.4	0.998	0.268	4.088	0.980

**Table S6.** Summary of the maximum adsorption capacity ( $q_{\max}$ ) of Bisphenol A (BPA) on various adsorbents.

Adsorbent	$q_{\max}/\text{mg g}^{-1}$	References
P-CDP	88	S8
magnetic rGOs	48.74	S9
GQ-02	313.5	S10
W20N	432.34	S11
graphene	181.82	S12
EPI-CDP	84	S13
PP-g-GMA-OA	45.59	S14
<b>AHCP-1</b>	<b>540.5</b>	<b>This work</b>

**Table S7.** Structural parameters for the eight organic pollutants.

Organic Pollutants	Space-filling model	Chemical class	Molar mass (g·mol <sup>-1</sup> )	Molecular 3-D size (Å)
Methylene blue (MB)		Cationic dye	319.8	17.043×8.243×6.815
Rhodamine B (RhB)		Cationic dye	479.01	18.089×14.296×8.265
Neutral red (NR)		Neutral dye	288.77	14.628×7.966×4.640
Methyl orange (MO)		Anionic dye	327.33	19.803×7.084×6.346
Bisphenol A (BPA)		Bisphenol A analogues	228.29	12.215×7.701×6.697
Bisphenol S (BPS)		Bisphenol A analogues	250.27	10.813×7.907×6.501
Hexafluorobisphenol A (FBPA)		Bisphenol A analogues	336.23	12.393×8.494×7.445
4-Nitrophenol (4-NP)		Phenolics	139.11	9.026×6.537×3.335



## References

- S1 V. da Silva Lacerda, J. B. López-Sotelo, A. Correa-Guimarães, S. Hernández-Navarro, M. Sánchez-Báscones, L. M. NavasGracia, P. Martín-Ramos and J. Martín-Gil, *J. Environ. Manage.*, 2015, **155**, 67–76.
- S2 T. Maneerung, J. Liew, Y. Dai, S. Kawi, C. Chong and C. H. Wang, *Bioresour. Technol.*, 2016, **200**, 350–359.
- S3 X. Wang, J. Ding, S. Yao, X. Wu, Q. Feng, Z. Wang and B. Geng, *J. Mater. Chem. A.*, 2014, **2**, 15958.
- S4 O. A. Oyetade, V. O. Nyamori, B. S. Martincigh and S. B. Jonnalagadda, *RSC Adv.*, 2015, **5**, 22724–22739.
- S5 Y. Yu, B. N. Murthy, J. G. Shapter, K. T. Constantopoulos, N. H. Voelcker and A. V. Ellis, *J. Hazard. Mater.*, 2013, **260**, 330–338.
- S6 J. Ding, B. Li, Y. Liu, X. Yan, S. Zeng, X. Zhang and J. Zhang, *J. Mater. Chem. A.*, 2015, **3**, 832–839.
- S7 F. Y. Yi, J. P. Li, D. Wu and Z. M. Sun, *Chem.–Eur. J.*, 2015, **21**, 11475–11482.
- S8 A. Alsbaiee, B. J. Smith, L. Xiao, Y. Ling, D. E. Helbling and W. R. Dichtel, *Nature.*, 2015, **529**, 190.
- S9 Z. X. Jin, X. X. Wang, Y. B. Sun, Y. J. Ai and X. K. Wang, *Environ. Sci. Technol.*, 2015, **49**, 9168–9175.
- S10 G. Q. Xiao, L. C. Fu and A. M. Li, *Chem. Eng. J.*, 2012, **191**, 171–176.
- S11 G. Liu, J. Ma, X. Li and Q. Qin, *J. Hazard. Mater.*, 2009, **164**, 1275–1280.
- S12 J. Xu, L. Wang and Y. F. Zhu, *Langmuir*, 2012, **28**, 8418–8425.
- S13 N. Morin-Crini and G. Crini, *Prog. Polym. Sci.*, 2013, **38**, 344–368.
- S14 X. Y. Zhou, J. F. Wei, K. Liu, N. N. Liu and B. Zhou, *Langmuir.*, 2014, **30**, 13861–13868.