SUPPORTING INFORMATION

Facile Preparation of Stereochemistry-controllable Biobased Poly(butylene maleate-co-butylene fumarate) Unsaturated Copolyesters: A Chemoselective Polymer Platform for Versatile Functionalization via Aza-Michael Addition

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Calculation of average sequence length of butylene maleate (L_{ncis}) and butylene fumarate (L_{ntrans}) segments, and the degree of polyester randomness (R). The resonance absorptions of the α -carbon of butylene segment were resolved with a curve-fitting program based on the least squares parameter adjustment criterion using the Gauss-Newton iteration procedure. The fitting process adjusts the peak position, the curve shape (a Gaussian fraction; whereby zero represents a pure Lorentzian and unity represents a pure Gaussian), and the peak width and the height in such a way that a best fit is obtained. The excellent agreement between the experimental and fitted spectra were founded with a regression coefficient R²>0.995, Figure S1 displayed the resolution result of Entry 12, as an instance. The four triads are assigned to fumarate butylene maleate (*trans/cis*) at 64.75 ppm, fumarate butylene fumarate (*trans/trans*) at 64.70 ppm, maleate butylene maleate (*cis/cis*) at 64.62 ppm, and maleate butylene fumarate (*cis/trans*) at 64.56 ppm, respectively. From the four peak areas ($A_{trans/cis}$, $A_{tis/cis}$, and $A_{cis/trans}$), molar ratios of butylene fumarate (P_{trans}) and butylene maleate (P_{cis}) groups were firstly calculated by the following equations:

$$P_{trans} = [(A_{trans/cis} + A_{cis/trans})/2 + (A_{trans/trans})]$$
$$P_{cis} = [(A_{trans/cis} + A_{cis/trans})/2 + (A_{cis/cis})]$$

Next, the following equations were used to calculate number average sequence length of butylene maleate (L_{ncis}) and butylene fumarate (L_{ntrans}) segments:

 $L_{ntrans} = 2 P_{trans} / (A_{trans/cis} + A_{cis/trans})$ $L_{ncis} = 2 P_{cis} / (A_{trans/cis} + A_{cis/trans})$

Then, the following equation was used to calculate the degree of randomness (R):

 $R= 1/L_{ntrans} + 1/L_{ncis}$

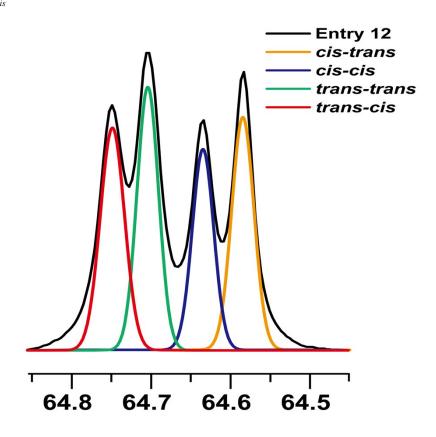


Figure S1. Peak fitting of the methylene carbon resonance in quantitative ¹³C NMR spectrum of Entry12 as an

example.

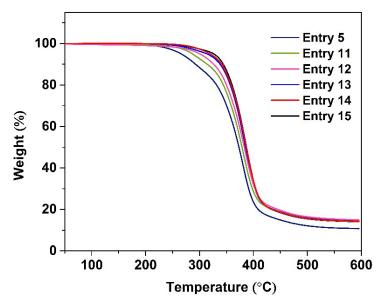


Figure S2. Thermoanalytical curves of poly(BM-co-BF) copolyesters.

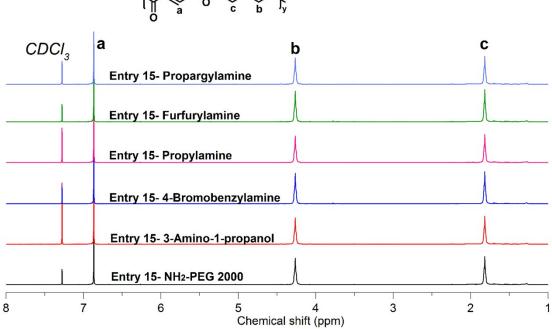


Figure S3. ¹H NMR spectra of model reaction. Entry 15 reacted with primary amine compounds for 10h at room temperature, and no aza- Michael addition occurred.

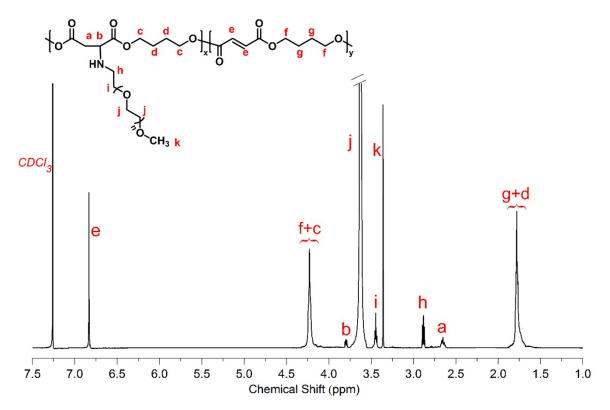


Figure S4. ¹H NMR spectrum of Entry 16 prepared by Michael addition of Entry5 with NH₂-PEG 2000.

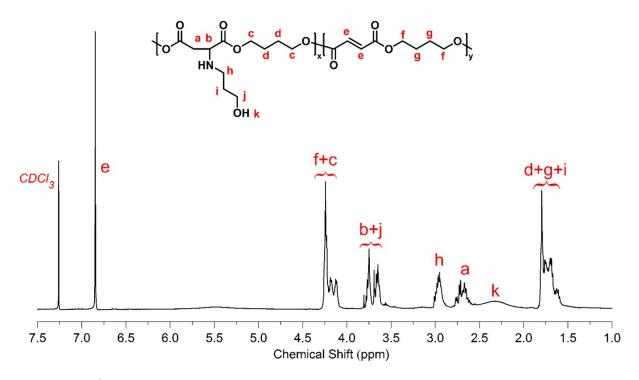


Figure S5. ¹H NMR spectrum of Entry 17 prepared by Michael addition of Entry5 with 3-Amino-1-propanol.

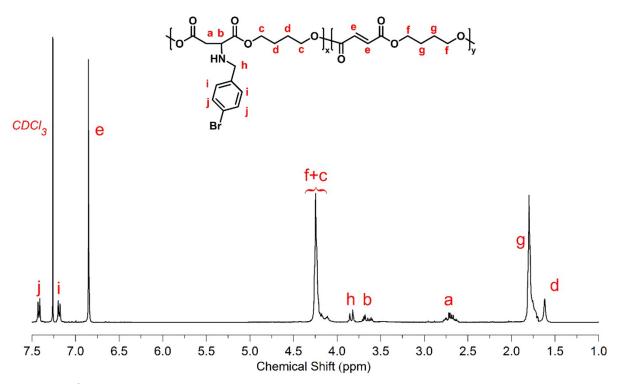


Figure S6. ¹H NMR spectrum of Entry 18 prepared by Michael addition of Entry5 with 4-Bromobenzylamine.

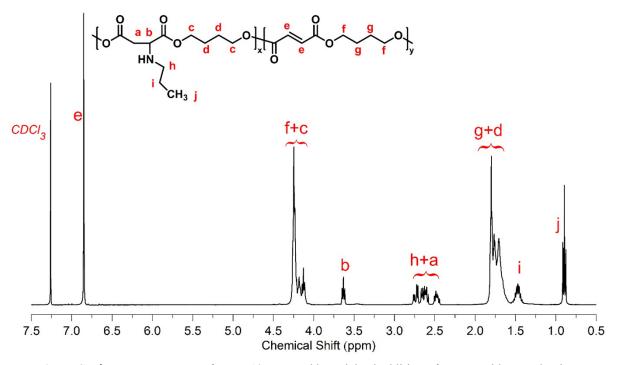


Figure S7. ¹H NMR spectrum of Entry 19 prepared by Michael addition of Entry5 with propylamine.

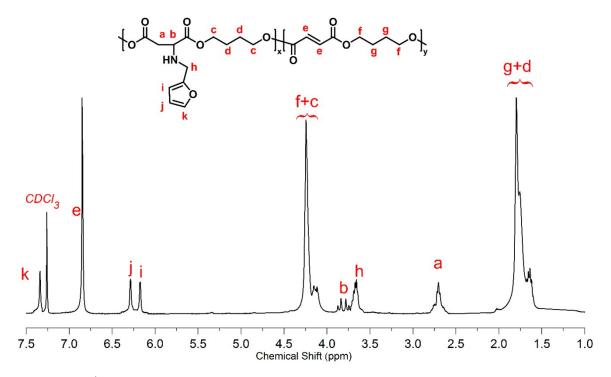


Figure S8. ¹H NMR spectrum of Entry 20 prepared by Michael addition of Entry5 with furfurylamine.

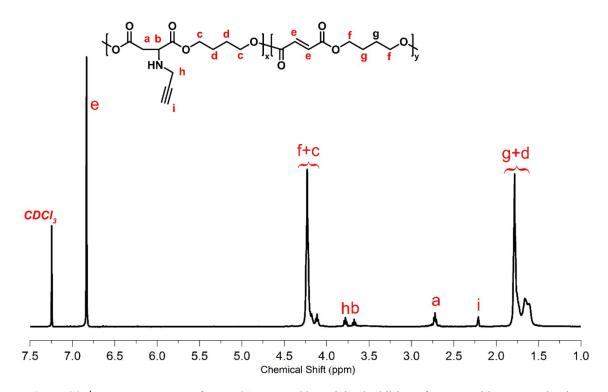


Figure S9. ¹H NMR spectrum of Entry 21 prepared by Michael addition of Entry5 with propargylamine.

