## SUPPORTING INFORMATION

Correlation between polymer architecture and Polyion Complex Micelle Stability with Proteins in Spheroid Cancer Models as seen by Light Sheet Microscopy

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Materials. Chemicals were used as received unless otherwise specified. Poly (ethylene glycol) methyl ether acrylate (PEGMEA, $M_{\mathrm{n}}=480 \mathrm{~g} \mathrm{~mol}^{-1}$ ), 2-carboxyethyl acrylate (CEA, $M_{\mathrm{n}}=144.13 \mathrm{~g} \mathrm{~mol}^{-1}$ ), 1,4-dioxane ( $>99 \%$ ), fluorescein isothiocyanate (FITC, $\geqslant 90 \%$ ), lysozyme from chicken egg white (HEWL, >90\%), $N$-hydroxysuccinimide (NHS, 98\%), N -(3-(dimethylamino)propyl)- $N$ '-ethylcarbodiimide hydrochloride (EDC, BioXtra) were purchased from Sigma-Aldrich. 2,2'-Azobis(isobutyronitrile) (AIBN, Fluka, 98\%). Diethyl ether ( $99 \%$ ), toluene ( $>99.5 \%$ ), dimethyl sulfoxide (DMSO, 98\%) were purchased from Ajax Finechem. 3-benzylsulfanylthiocarbonylsufanylpropionic acid (BSPA) was synthesized according to a previous report. ${ }^{1}$ Cyanine5 maleimide (Cy5) was purchased from Lumiprobe. Dialysis tubings (molecular weight cut-off 3.5 kDa ) were purchased from Thermo Scientific. The cell line MCF-7 was purchased from CellBank Australia.

## Synthesis of polymers

## Synthesis of PPEGMEA

PEGMEA ( $5 \mathrm{~g}, 1.04 \times 10^{-2} \mathrm{~mol}$ ), BSPA $\left(56.75 \mathrm{mg}, 2.08 \times 10^{-4} \mathrm{~mol}\right)$ and AIBN $(3.42 \mathrm{mg}$, $\left.2.08 \times 10^{-5} \mathrm{~mol}\right)$ were dissolved in toluene ( 5.83 mL ) in a clean 50 mL round-bottom flask. The mixed solution was purged with nitrogen for 40 min , followed by the polymerisation at $60{ }^{\circ} \mathrm{C}$ for 2 h . The polymerisation was terminated by placing the vial into ice bath and exposure to air. The conversion of the monomers was determined by ${ }^{1} \mathrm{H}$ NMR in $\mathrm{CDCl}_{3}$ on a Bruker Avance III HD ( 400 MHz ). The crude solution was purified by precipitating in a mixture of diethyl ether and hexane ( $1: 1, \mathrm{v} / \mathrm{v}$ ) once and another two times in diethyl ether alone. The molecular weight was determined by size exclusion chromatography (SEC) to be $11,000 \mathrm{~g} \mathrm{~mol}^{-1}$ with a $Đ$ of 1.13 using $N, N$-dimethylacetamide (DMAc) as the mobile phase and poly(methyl methacrylate) (PMMA) standards.

## Synthesis of PPEGMEA-b-PCEA

The PPEGMEA- $b$-PCEA polymers with different lengths of the second charged blocks were prepared by chain extension of PPEGMEA ${ }_{19}$ as a macro-RAFT agent. PPEGMEA ( 391 mg ,
$4.16 \times 10^{-5} \mathrm{~mol}$ ) was mixed with CEA ( $600 \mathrm{mg}, 4.16 \times 10^{-3} \mathrm{~mol}$ ) and AIBN ( $1.37 \mathrm{mg}, 8.33 \times$ $10^{-6} \mathrm{~mol}$ ) in 1,4 -dioxane ( 3.69 mL ) in a feed ratio of $1: 100: 0.2$. The solution was equally split into three 5 mL round-bottom flask and purged with nitrogen for 40 min . Polymerizations were carried out at $60^{\circ} \mathrm{C}$ in an oil bath for $1 \mathrm{~h}, 1.5 \mathrm{~h}$ and 2 h respectively before terminated by immersing the flasks into ice bath and exposure to air. The crude was extensively dialysed against Milli-Q water for two days followed by lyophilisation. The number of repeating units of the CEA block was determined by ${ }^{1} \mathrm{H}$ NMR in DMSO-d6. The dispersities $(\boxplus)$ of the polymers were obtained from SEC using DMAc as the mobile phase and PMMA standards.

## Synthesis of PPEGMEA-b-PAA

The same procedures were used as PPEGMEA-b-PCEA, where CEA was replaced by AA ( $300 \mathrm{mg}, 4.16 \times 10^{-3} \mathrm{~mol}$ ).

## Synthesis of PPEGMEA-b-(PCEA-co-PBA)

PPEGMEA ( $261 \mathrm{mg}, 2.78 \times 10^{-5} \mathrm{~mol}$ ) was mixed with CEA ( $200 \mathrm{mg}, 1.39 \times 10^{-3} \mathrm{~mol}$ ), BA $\left(71 \mathrm{mg}, 5.54 \times 10^{-4} \mathrm{~mol}\right)$ and $\operatorname{AIBN}\left(0.91 \mathrm{mg}, 5.54 \times 10^{-6} \mathrm{~mol}\right)$ in 1,4 -dioxane $(2 \mathrm{~mL})$ in a 5 mL round-bottom flask. The mixture was degassed by purging nitrogen for 40 min before the polymerised at $60^{\circ} \mathrm{C}$ overnight. The reaction was terminated by immersing the flask into ice bath and exposure to air. The repeating units of CEA and BA were determined by ${ }^{1} \mathrm{H}$ NMR in DMSO-d6. The crude was transferred to 3.5 kDa MWCO dialysis tubings and purified by dialysing against methanol for one day and Milli-Q for another day followed by lyophilisation. $M_{\mathrm{n}, \mathrm{SEC}}=34,000 \mathrm{~g} \mathrm{~mol}^{-1}, ~ Đ=1.14$.

## Synthesis of PPEGMEA-b-PBA

PPEGMEA ( $293 \mathrm{mg}, 3.12 \times 10^{-5} \mathrm{~mol}$ ) was mixed with BA ( $200 \mathrm{mg}, 1.56 \times 10^{-3} \mathrm{~mol}$ ) and AIBN ( $1.02 \mathrm{mg}, 6.24 \times 10^{-6} \mathrm{~mol}$ ) in toluene $(2 \mathrm{~mL})$ in a 5 mL round-bottom flask. The mixture was degassed by purging nitrogen for 40 min before the polymerised at $60^{\circ} \mathrm{C}$ for 1 h 45 min . The reaction was terminated by immersing the flask into ice bath and exposure to air. The repeating units of BA were determined by ${ }^{1} \mathrm{H} \mathrm{NMR}$ in $\mathrm{CDCl}_{3}$. The crude was transferred to 3.5 kDa MWCO dialysis tubings and purified by dialysing against methanol for one day and Milli-Q for another day followed by lyophilisation. $M_{\mathrm{n}, \mathrm{SEC}}=13,000 \mathrm{~g} \mathrm{~mol}^{-1}, ~ Ð=1.12$.

## Synthesis of PPEGMEA-b-PBA-b-PCEA

PPEGMEA-b-PBA ( $204 \mathrm{mg}, 1.66 \times 10^{-5} \mathrm{~mol}$ ) was mixed with CEA ( $239.5 \mathrm{mg}, 1.66 \times 10^{-3}$ $\mathrm{mol})$ and $\operatorname{AIBN}\left(0.55 \mathrm{mg}, 3.32 \times 10^{-6} \mathrm{~mol}\right)$ in 1,4-dioxane $(1.5 \mathrm{~mL})$ in a 5 mL round-bottom flask. The mixture was degassed by purging nitrogen for 40 min before the polymerised at 60 ${ }^{\circ} \mathrm{C}$ for 2 h 10 min . The reaction was terminated by immersing the flask into ice bath and exposure to air. The repeating units of BA were determined by ${ }^{1} \mathrm{H}$ NMR in $\mathrm{CDCl}_{3}$. The crude
was transferred to 3.5 kDa MWCO dialysis tubings and purified by dialysing against Milli-Q for 2 days followed by lyophilisation. $M_{\mathrm{n}, \mathrm{SEC}}=26,000 \mathrm{~g} \mathrm{~mol}^{-1}, ~ D=1.20$.

## Synthesis of PPEGMEA-b-PCEA-b-PBA

PPEGMEA- $b$-PCEA ( $200 \mathrm{mg}, 1.25 \times 10^{-5} \mathrm{~mol}$ ) was mixed with BA ( $80.1 \mathrm{mg}, 6.25 \times 10^{-4}$ $\mathrm{mol})$ and $\operatorname{AIBN}\left(0.41 \mathrm{mg}, 2.50 \times 10^{-6} \mathrm{~mol}\right)$ in 1,4 -dioxane $(1.7 \mathrm{~mL})$ in a 5 mL round-bottom flask. The mixture was degassed by purging nitrogen for 40 min before the polymerised at 60 ${ }^{\circ} \mathrm{C}$ for 4 h 40 min . The reaction was terminated by immersing the flask into ice bath and exposure to air. The repeating units of BA were determined by ${ }^{1} \mathrm{H} N M R$ in $\mathrm{CDCl}_{3}$. The crude was transferred to 3.5 kDa MWCO dialysis tubings and purified by dialysing against methanol for one day and Milli-Q for another day followed by lyophilisation. $M_{\mathrm{n}, \mathrm{SEC}}=31,000$ $\mathrm{g} \mathrm{mol}^{-1}, ~ Đ=1.12$.

Micellization. The polymers were dissolved in Milli-Q water (500 $\mu \mathrm{L}$ ) at various concentrations. HEWL ( 0.7 mg ) was also dissolved in Milli-Q water ( $500 \mu \mathrm{~L}$ ). PIC micelles were formed by dropwise adding HEWL solution into polymer solution under stirring.

Size Exclusion Chromatography (SEC). SEC was conducted using a Shimadzu modular system containing a DGU-12A degasser, a LC-10AT pump, a SIL-10AD automatic injector, a CTO-10A column oven, and a RID-10A refractive index detector. A $50 \times 7.8 \mathrm{~mm}$ guard column and four $300 \times 7.8 \mathrm{~mm}$ linear columns $\left(10^{5}, 10^{4}, 10^{3}\right.$, and $10^{2} \AA$ pore size, $5 \mu \mathrm{~m}$ particle size) were used for the analyses. $N, N$-Dimethylacetamide (DMAc, HPLC grade, $0.05 \% \mathrm{w} / \mathrm{v}$ of 2,6- dibutyl-4-methylphenol (BHT), $0.03 \% \mathrm{w} / \mathrm{v}$ of LiBr ) with a flow rate of 1 $\mathrm{mL} \mathrm{min}^{-1}$ was used for analyses. The calibration of the instrument was conducted by commercial polymethyl methacrylate standards ( $0.5-1000 \mathrm{kDa}$, Polymer Laboratories). Chromatograms were processed using Cirrus 2.0 software (Polymer Laboratories).

## Dynamic light scattering (DLS)

The measurements were performed at $25^{\circ} \mathrm{C}$ using Malvern Nano-ZS equipped with a $\mathrm{He}-\mathrm{Ne}$ laser ( $4 \mathrm{~mW}, \lambda=632 \mathrm{~nm}$ ) and scattering light was detected at an angle of $173^{\circ}$. All measurements were taken in triplicate.

## Isothermal titration calorimetry

Stock solutions of PPEGMEA-b-PCEA ( 0.02 mM ), PPEGMEA-b-PAA $(0.02 \mathrm{mM})$ and Lys $(0.5 \mathrm{mM})$ were separately prepared in Milli-Q water. For each measurement, polymer solution $(200 \mu \mathrm{~L})$ was added in the sample cell of a GE ITC200 isothermal calorimeter and equal volume of Milli-Q water was place in the reference cell. Lys solution ( $40 \mu \mathrm{~L}$ ) was loaded into the syringe. Measurement was carried out at $25^{\circ} \mathrm{C}$ and titrant $(1 \mu \mathrm{~L})$ was injected into the sample cell 20 times in a row with 180 s spacing between 2 injections. The mixture was constantly stirred at 750 rpm throughout the experiment. Control experiments of Lys into Milli-Q water, Milli-Q water into polymer and Milli-Q to Milli-Q were also conducted using the same procedure. The raw data was analysed using a MicroCal LLC ITC program in Origin ${ }^{\circledR}$ and the data from the control experiments were subtracted from sample data before they were fitted.

## Disassembly at high ionic strength

Stock solutions of the polymers $(1 \mathrm{mg} / \mathrm{ml})$ and lysozyme $(1.4 \mathrm{mg} / \mathrm{ml})$ were prepared in Milli-Q water. For the PIC micelle preparation, $500 \mu 1$ of lysozyme solution was added dropwise to varying concentrations of polymer solution such that ratio of the negative to positive charges was 1:1. After stirring for 3 hours, the PIC micelles were left to equilibrate for 2 hours before addition of NaCl solution. $20 \mu \mathrm{l}$ of 1 M NaCl was added, each time, followed by stirring for 60 secs and equilibration for 120 secs at $25^{\circ} \mathrm{C}$ before measuring the derived count rate using Malvern Nano-ZS equipped with a He-Ne laser ( $4 \mathrm{~mW}, \lambda=632 \mathrm{~nm}$ ) with scattering light detection at an angle of $173^{\circ}$.

## Cell viability assay

The nanoparticles were prepared in Milli-Q water at a concentration of $1 \mathrm{mg} / \mathrm{mL}$. The biocompatibility and cytotoxicity were evaluated by sulforhodamine B (SRB) assay using MCF-7 human breast cancer cell line. The cells were seeded into a 96 -well plate ( 4000 cells/well) in $100 \mu \mathrm{~L}$ Dulbecco's modified eagle medium (DMEM) containing $10 \%$ fetal bovine serum (FBS), $0.002 \%$ plasmocin at $37{ }^{\circ} \mathrm{C}$ in a $5 \% \mathrm{CO}_{2}$ and $100 \%$ humidity environment for 24 h . The medium was then replaced by $2 \times$ concentrated fresh DMEM media. The micelles were sterilized by UV irradiation for 15 min and then serially diluted ( $2 \times$ dilution) by sterile Milli-Q water. $100 \mu \mathrm{~L}$ of each solution were separately added into 4 wells while 4 control samples were fed with $100 \mu \mathrm{~L}$ Milli-Q water. The final concentration of micelles ranged from $2 \mu \mathrm{~g} / \mathrm{mL}$ to $500 \mu \mathrm{~g} / \mathrm{mL}$. After 72 h of incubation, the culture medium
was discarded and $100 \mu \mathrm{~L}$ of cold trichloroacetic acid (TCA) was added into each well, followed by an additional incubation at $4{ }^{\circ} \mathrm{C}$ for 30 min . After that, the cells were washed 5 times with Milli-Q water and dyed with $100 \mu \mathrm{~L}$ of SRB solution $(0.4 \% \mathrm{w} / \mathrm{v}$ in $1 \%$ acetic acid). After incubated at room temperature for 20 min , unbound dye was washed out by $1 \%$ acetic acid $(100 \mu \mathrm{~L})$ for 5 times. The plate was then allowed to air dry in the dark. Tris buffer $(200 \mu \mathrm{~L}, 10 \mathrm{mM})$ was added into each well to solubilize bound dyes. The absorbance of the solution was measured on a Bio-Rad BenchMark microplate reader at 490 nm . The relative viability of cell (\%) was determined as a percentage of the treated cells to the untreated control cells.

## Polymer labelling with Cy5

For each polymer, it was firstly dissolved in DMSO ( 1.5 mL ) at a concentration of $1.53 \mu \mathrm{M}$ 50 eq , containing EDC ( $11.7 \mu \mathrm{~g}, 61.2 \mathrm{nM}, 2 \mathrm{eq}$ ) and NHS ( $3.1 \mu \mathrm{~g}, 30.6 \mathrm{nM}, 1 \mathrm{eq}$ ) and stirred for 20 min at room temperature. Cy5 stock solution ( $20 \mu \mathrm{~g}$ in 2 mL DMSO, 1 eq ) was then added into the polymer solution and continuously stirred overnight in the dark. Unreacted cy 5 was removed by dialysing against methanol for one day and Milli-Q water for another day, followed by lyophilization. Lys was labelled with FITC as follows: Lys ( 14 mg ) was dissolved in 7 mL 0.05 M carbonate buffer ( $7 \mathrm{~mL}, \mathrm{pH} 9.5$ ) while FITC ( 2.34 mg ) was dissolved in DMSO ( 1 mL ). The reaction was carried out by mixing both the solution at $4{ }^{\circ} \mathrm{C}$ overnight in the dark. The crude product was purified by dialyzing against Milli-Q water for 2 days at $4^{\circ} \mathrm{C}$, followed by lyophilization.

## Flow cytometry

MCF-7 breast cancer cells were seeded into 6 -well plate at a concentration of $5 \times 10^{5}$ cells/well and cultured for 2 days. The cells were then incubated with 3 mL fresh DMEM culture media containing FITC labelled micelles (Lys $15 \mu \mathrm{M}$, polymer $30 \mu \mathrm{M}$ ) for (Lys 7.1 $\mu \mathrm{M}$, polymer $14.2 \mu \mathrm{M}) 2 \mathrm{~h}$ before the cells were washed with cold PBS for 4 times and detached from the plate well using trypsin/EDTA solution. Cell suspensions were then centrifuged and resuspended in cold Hank's buffer. The cellular uptake was determined by measuring fluorescence intensity from 20000 cell events per well ( $\lambda_{\mathrm{ex}}=488 \mathrm{~nm}, \lambda_{\mathrm{ex}}=530 / 30$ nm ) using flow cytometry (BD FACSCanto ${ }^{\mathrm{TM}}$ II Analyser). Median fluorescence intensity (MFI) was averaged from 3 individual wells for one sample using a FlowJo® software.

## Protein labelling with FITC

FITC was used to fluorescently label Lys using a previously reported method. Briefly, Lys was dissolved in 0.05 M carbonate buffer ( $7 \mathrm{~mL}, \mathrm{pH} 9$ ) at a concentration of $2 \mathrm{mg} / \mathrm{mL}$ while FITC stock solution was prepared in DMSO $(1 \mathrm{~mL})$ at a concentration of $2.34 \mathrm{mg} / \mathrm{mL}$. The reaction was carried out by dropwise adding the FITC solution into Lys solution. The mixture was then stirred in the dark at room temperature for 1 h . Unreacted FITC was removed by extensive dialysis against Milli-Q water for 2 days at $4^{\circ} \mathrm{C}$, followed by lyophilisation.

PIC micelles were then prepared using FITC labelled Lys. All the samples were prepared at a polymer/Lys molar ratio of 2. The fluorescently labelled PIC micelles were then incubated with MCF-7 cancer cells for 2 h . The fluorophore was excited by a 488 nm laser and fluorescence emission was collected using a $530 / 30 \mathrm{~nm}$ band-pass filter.


Figure S1. Synthesis scheme of RAFT polymerised PPEGMEA- $b$-PCEA and PPEGMEA- $b$ PAA with various lengths of the charged blocks.

Table S1. Characterisation of macroRAFT agent and block copolymers.

| Polymer | Polymerisation conditions |  | $\begin{gathered} M_{\mathrm{n}} \\ \left(\mathrm{~g} \mathrm{~mol}^{-}\right. \\ \left.{ }^{1}\right)^{\mathrm{b}} \end{gathered}$ | $\begin{gathered} M_{\mathrm{n}} \\ \left(\mathrm{~g} \mathrm{~mol}^{-}\right. \\ \left.{ }^{1}\right)^{\mathrm{c}} \end{gathered}$ | $\boldsymbol{D}^{\text {c }}$ | Conversion ${ }^{\text {b }}$ (\%) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| PPEGMEA $_{19}$ | 50:1:0.1 | 2 | 9400 | 11000 | 1.13 | 38 |
| $\begin{gathered} \text { PPEGMEA }_{19}-b- \\ \text { PCEA }_{30} \end{gathered}$ | 100:1:0.2 | 1 | 14000 | 15000 | 1.17 | 30 |
| PPEGMEA $_{19}-b-$ PCEA $_{52}$ | 100:1:0.2 | 1.3 | 17000 | 31000 | 1.19 | 52 |
| PPEGMEA $_{19}-b-$ PCEA $_{72}$ | 100:1:0.2 | 2 | 20000 | 34000 | 1.23 | 72 |
| $\begin{gathered} \text { PPEGMEA }_{19}-b- \\ \text { PAA }_{30} \end{gathered}$ | 100:1:0.2 | 1 | 12000 | 25000 | 1.17 | 30 |
| $\begin{gathered} \text { PPEGMEA }_{19}-b- \\ \text { PAA }_{50} \end{gathered}$ | 100:1:0.2 | 1.3 | 13000 | 28000 | 1.21 | 50 |
| PPEGMEA $_{19}-b-$ $\mathrm{PAA}_{72}$ | 100:1:0.2 | 2 | 22000 | 31000 | 1.26 | 72 |
| ${ }^{\mathrm{a}}[\mathrm{M}]:[\mathrm{C}]:[\mathrm{I}]=[\mathrm{mo}$ using DMAc as elue was polymerised at | mer]:[CTA]:[A <br> and PMMA <br> ${ }^{\circ} \mathrm{C}$ in 1,4-diox |  | ults from GMEA- $b$ | H NMR. PCEA an | $\begin{aligned} & { }^{\mathrm{c} \mathrm{Res}} \\ & \mathrm{qd} \mathrm{PPE} \end{aligned}$ | ults from SEC GMEA-b-PAA |

$a, b$






Figure S2. ${ }^{1} \mathrm{H}$ NMR spectra for determining the conversion of $n$-butyl acrylate.


Figure S3. ${ }^{1} \mathrm{H}$ NMR spectra for determining the conversion of 2-carboxyethyl acrylate.
The repeating units (RU) of $n$-butyl acrylate (BA) and 2-carboxyethyl acrylate (CEA) were calculated as following:

Conversion $_{B A}=\frac{4.85-2.87}{4.85}=40 \%$
$R U_{B A}=50 \times 40 \%=20$
Conversion $_{\text {CEA }}=\frac{2.10-1.22}{2.10}=42 \%$
$R U_{C E A}=100 \times 42 \%=42$


Figu
re S4. DMAc SEC traces of (A) PPEGMEA ${ }_{19}-b$-PCEA series and (B) PPEGMEA $_{19}-b-$-PAA series.

(B)





PPEGMEA $_{19}-b-$ PCEA $_{46}-b-$ PBA $_{20}$




PPEGMEA $_{19}-b-$ PBA $_{20}-b-$ PCEA $_{42}$

Figure S5. Synthetic routes of (A) PPEGMEA $19-b-\left(\mathrm{PCEA}_{50}-c o-\mathrm{PBA}_{20}\right)$, (B) PPEGMEA ${ }_{19}-b-$ PCEA $_{46}-b-$ PBA $_{20}$ and (C) PPEGMEA $19-b-$ PBA $_{20}-b-$ PCEA $_{42}$.

Table S2. Conditions used to prepare polymers via RAFT polymerisation to study the effect of hydrophobic units (butyl acrylate) on micelles' stability.

| Polymer | Polymerisation conditions |  |  | $\begin{gathered} M_{\mathrm{n}, \mathrm{NMR}} \\ (\mathrm{~g} \mathrm{~mol} \\ \left.\mathrm{mol}^{-1}\right)^{\mathrm{c}} \end{gathered}$ | $\begin{gathered} M_{\mathrm{n}, \mathrm{SEC}} \\ \left(\mathrm{~g} \mathrm{~mol}^{-1}\right)^{\mathrm{d}} \end{gathered}$ | D |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | $[\mathrm{M}]:[\mathrm{C}]:[\mathrm{I}]^{\text {a }}$ | Time <br> (h) | Temperature $\left({ }^{\circ} \mathrm{C}\right)$ |  |  |  |
| PPEGMEA $_{19}$ | 50:1:0.1 | 2 | 60 | 9392 | 11000 | 1.13 |
| P (C-co-B) | 50: $20: 1: 0.2^{\text {b }}$ | 16 | 60 | 19152 | 34000 | 1.14 |
| PC | 100:1:0.2 | 1.5 | 60 | 16016 | 25000 | 1.13 |
| PCB | 50:1:0.2 | 4.67 | 60 | 18576 | 31000 | 1.12 |
| PB | 50:1:0.2 | 1.75 | 60 | 11952 | 13000 | 1.12 |
| PBC | 100:1:0.2 | 2.17 | 60 | 18000 | 26000 | 1.20 |

${ }^{\mathrm{a}}[\mathrm{M}]:[\mathrm{C}]:[\mathrm{I}]=$ [monomer]:[CTA]:[AIBN]. ${ }^{\mathrm{b}}[\mathrm{CEA}]:[\mathrm{BA}]:[\mathrm{CTA}]:[\mathrm{I}]=$ 50:20:1:0.2.
${ }^{\mathrm{c}}$ Molecular weights calculated from ${ }^{1} \mathrm{H}$ NMR. ${ }^{\mathrm{d}}$ Molecular weights obtained from DMAc SEC using PMMA standards. PPEGMEA ${ }_{19}$, PCB and PB were polymerised in toluene. $\mathrm{P}(\mathrm{C}-c o-\mathrm{B}), \mathrm{PC}, \mathrm{PBC}$ were polymerised in 1,4-dioxane. $\mathrm{P}(\mathrm{C}-c o-\mathrm{B})=$ PPEGMEA $_{19}-b-$ $\left(\right.$ PCEA $_{50}-c o-$ PBA $\left._{20}\right) ; \mathrm{PC}=$ PPEGMEA $_{19}-b-$ PCEA $_{46} ; \mathrm{PCB}=\mathrm{PPEGMEA}_{19}-b-\mathrm{PCEA}_{46}-b-$ $\mathrm{PBA}_{20} ; \mathrm{PB}=$ PPEGMEA $_{19}-b-\mathrm{PBA}_{20} ; \mathrm{PBC}=\mathrm{PPEGMEA}_{19}-b-\mathrm{PBA}_{20}-b-\mathrm{PCEA}_{42}$. The monomer concentration for all polymerisations was 1 M .

A


B


C


Figure S6. $N, N$-dimethylacetamide size exclusion chromatography (DMAc SEC) traces of polymers used for synthesis of (A) PPEGMEA ${ }_{19}-b-\left(\mathrm{PCEA}_{50}-c o-\mathrm{PBA}_{20}\right)$, (B) PPEGMEA ${ }_{19}-b-$ PCEA $_{46}-b-\mathrm{PBA}_{20}$ and (C) PPEGMEA $_{19}-b-\mathrm{PBA}_{20}-b-$ PCEA $_{42}$.


Figure S7. Cryo-TEM image of PPEGMEA ${ }_{19}-b-$ PCEA $_{52} /$ lysozyme micelles at a molar ratio $^{2}$ of 2 . The image was taken at a slight defocus to improve image contrast.


Figure S8. SAXS analysis of (A) PPEGMEA $_{19}-b-$ PCEA $_{52} /$ lysozyme micelles and (B) $^{\text {(B) }}$ PPEGMEA $_{19}-b-$ PAA $_{50}$ lysozyme micelles at a molar ratio of 2 in Milli-Q water.

Table S3. Fitting parameters of SAXS data.

| Parameters | CEA ${ }_{52} /$ LYS $^{*}$ | $\mathrm{AA}_{50} / \mathbf{L Y S}{ }^{*}$ |
| :---: | :---: | :---: |
| core radius (nm) | 6.29 | 7.92 |
| Core Polydispersity(0,1) | 0.47 | 0.43 |
| Core SLD ( ${ }^{-2}$ ) | $1.09 * 10^{-5}$ | $1.09 * 10^{-5}$ |
| Shell 1 thickness (nm) | 4.56 | 4.56 |
| Shell 1 SLD ( ${ }^{-2}$ ) | $1.08 * 10^{-5}$ | $1.06 * 10^{-5}$ |
| Shell 2 thickness (nm) | 3.56 | 3.57 |
| Shell 2 SLD ( ${ }^{-2}$ ) | $1.08 * 10^{-5}$ | $9.34 * 10^{-6}$ |
| $\begin{aligned} & { }^{*} \mathrm{CEA}_{52} / \mathrm{LYS}=\text { PPEGMEA }_{19}-b-\text { PCEA }_{52} / \text { lysozyme micelles, } \mathrm{AA}_{50} / \mathrm{LYS} \\ & \text { PPEGMEA } \\ & 19-b-\mathrm{PAA}_{50} / \text { lysozyme, molar ratio of polymer to protein was } 2 . \end{aligned}$ |  |  |



Figure S9. $z$-Averaged hydrodynamic diameters of polyion complex (PIC) micelles formed by (A) PPEGMEA- $b$-PCEA (CEA) and (B) PPEGMEA- $b$-PAA (AA) with Lysozyme (LYS, 0.05 mM ) respectively at various molar ratios of negative charges to positive charges Zeta potential of (C) CEA/LYS PIC micelles and (D) AA/LYS PIC micelles. Data was averaged from three measurements.


Figure S10. PDI of the DLS distribution of polyion complex (PIC) micelles formed by (A) PPEGMEA- $b$-PCEA (CEA) and (B) PPEGMEA- $b$-PAA (AA) with Lysozyme (LYS, 0.05 $\mathrm{mM})$ respectively at various molar ratios of negative charges to positive charges


Figure S11. Isothermal titration calorimetry (ITC) profiles of titrating lysozyme (Lys) into (A) PPEGMEA ${ }_{19}-b-$ PCEA $_{52}$ and (B) PPEGMEA $19-b-$ PAA $_{50}$ at $25^{\circ} \mathrm{C}$.

Table S4. Thermodynamic parameters obtained from isothermal titration calorimetry (ITC) measurements.

| Sample | $\boldsymbol{N}$ | $\boldsymbol{K}_{\mathrm{a}}$ | $\Delta \boldsymbol{H}$ | $\Delta \boldsymbol{S}$ | $\Delta \boldsymbol{G}^{\mathrm{c}}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | $\left(\times 10^{5} \mathrm{M}^{-1}\right)$ | $\left(\mathrm{kcal} \mathrm{mol}^{-1}\right)$ | $\left(\mathrm{cal} \mathrm{mol}^{-1} \mathrm{~K}^{-1}\right)$ | $\left(\mathrm{kcal} \mathrm{mol}{ }^{-1}\right)$ |
| $\mathrm{CEA}_{52}{ }^{\mathrm{a}}$ | $1.73 \pm 0.01$ | $4.04 \pm 0.26$ | $-22.22 \pm 0.27$ | -48.8 | -7.67 |
| $\mathrm{AA}_{50}{ }^{\mathrm{b}}$ | $1.16 \pm 0.03$ | $1.13 \pm 0.10$ | $-19.14 \pm 0.74$ | -41.1 | -12.25 |

*Titrations of lysozyme (LYS) into ${ }^{\text {a }}{ }^{\text {PPEGMEA }}{ }_{19}-b-$ PCEA $_{52}$ and ${ }^{\text {aPPEGMEA }} 19-b-$ $\mathrm{PAA}_{50} . N$, binding stoichiometry; $K_{\mathrm{a}}$, binding affinity; $\Delta H, \Delta S$ and $\Delta G$, changes in enthalpy, entropy and Gibbs free energy respectively. ${ }^{\text {a }}$ The $\Delta G$ value was calculated using Gibbs free energy equation: $\Delta G=\Delta H-T \Delta S$.


Figure S12. The size changes of PPEGMEA $_{19}-b-$ PCEA $_{52} /{ }^{\prime}$ lysozyme micelles in buffer (HBSS) over a period of 7 days as determined by DLS. The concentrations of polymer and lysozyme are 7.14 and $3.57 \mu \mathrm{M}$, respectively.


Figure S13. Evaluation of cytotoxicity of PPEGMEA-b-PCEA and PPEGMEA-b-PAA at concentrations up to $500 \mu \mathrm{~g} / \mathrm{mL}$ against MCF-7 human breast cancer cells after incubation of 72 h .


Figure S14. Fluorescence spectra of cy5 labelled PPEGMEA ${ }_{19}-b-$ PCEA $_{\mathrm{n}}(\mathrm{n}=30,52,72)$ and PPEGMEA $_{19}-b-$ PAA $_{\mathrm{m}}(\mathrm{m}=30,50,72)$ in aqueous solutions $(0.1 \mathrm{mg} / \mathrm{mL})$. Excitation wavelength $=600 \mathrm{~nm}$. The dotted line shows wavelength of the filter ( 660 nm longpass) used on LSFM.

Table S5. Fluorescence intensity of polymers used for light-sheet fluorescence microscopy (LSFM).

## Polymer

PPEGMEA $_{19}-b-$ PCEA $_{30}$ Fluorescence intensity at $\mathbf{6 6 0} \mathbf{~ n m ~ ( a . ~ u . ) ~}$

PPEGMEA $_{19}-b-$ PCEA $_{52}$ 496

PPEGMEA $_{19}-b-$ PCEA $_{72}$ .

PPEGMEA $_{19}-b-$ PAA $_{30}$ 285

PPEGMEA $_{19}-b-$ PAA $_{50} \quad 190$
PPEGMEA $_{19}-b-$ PAA $_{72}$ 114

Polymer concentration was $0.1 \mathrm{mg} / \mathrm{mL}$ for all measurements. Excitation wavelength was 600 nm .


Figure S15. Normalised UV-Vis absorption (abs) spectra and fluorescence emission (emi) spectra of Texas Red labelled lysozyme and cy 5 labelled CEA polymer in Milli-Q water. The block dotted line shows the wavelength of the laser subsequently used in confocal laser scanning microscopy and light sheet microscopy.

Table S6. Contents of PIC micelles for FRET studying.

|  | Concentration $(\mu \mathrm{M})$ |  |  |
| :---: | :---: | :---: | :---: |
| Sample | Lysozyme <br> (Texas Red) | CEA (no label) ${ }^{\mathbf{a}}$ | CEA (cy 5) ${ }^{\mathbf{a}}$ |
| $8: 2$ | 2.5 | 4 | 1 |
| $6: 4$ | 2.5 | 3 | 2 |
| $4: 6$ | 2.5 | 2 | 3 |
| $2: 8$ | 2.5 | 1 | 4 |
| $2: 8$ (control) | 0 | 1 | 4 |

${ }^{\mathrm{a}}$ CEA $=$ PPEGMEA $_{39}-b-$ PCEA $_{35}$. All samples were prepared in 2 mL Milli-Q water.


Figure S16. Fluorescence spectra of the mixture of CEA, CEA (cy5) and lysozyme (Texas Red) with increasing amount of CEA (cy5) polymer (from 1 to $4 \mu \mathrm{M}$ ) in 2 mL Milli-Q water. The concentration of Lys (Texas Red) was kept constant ( $2.5 \mu \mathrm{M}$ ) in all samples. A wavelength of 530 nm laser was used to excite fluorophores for all measurements. The control is CEA (cy5) only without Lys. Ratios are shown in moles. CEA $=$ PPEGMEA $_{39}-b-$ PCEA $_{35}$.


Figure S17. Images of MCF-7 breast cancer cell spheroids obtained using light-sheet fluorescence microscopy (LSFM) showing the fluorescence intensity of (negative control) PIC micelles of unlabelled CEA polymer/LYS (Texas Red), (positive control) PIC micelles of unlabelled CEA polymer/LYS (labelled with both Texas Red and cy5) after an incubation period of 2 h . The donor fluorophore (Texas Red) was excited using a 561 nm laser and the acceptor fluorophore (cy5) was exited at 638 nm . Fluorescence emission was collected with a $595 / 20 \mathrm{~nm}$ bandpass filter for the donor and a 660 nm longpass filter for the acceptor.

