

## Supporting Information

### Synthesis and degradation of branched, photo-labile poly (acrylic acid) and polystyrene

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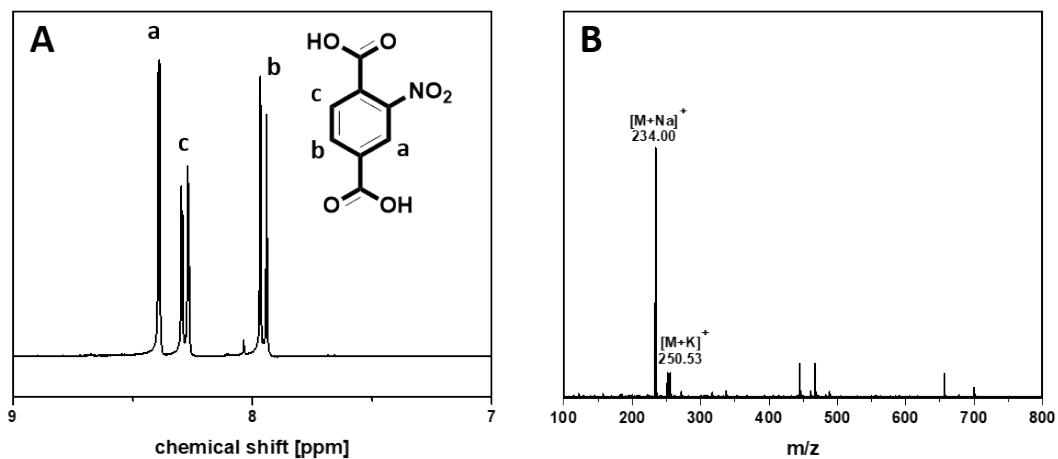


Figure S1:  $^1\text{H}$ -NMR spectrum in DMSO- $d_6$  (A) and ESI mass spectrum (B) of nitroterephthalic acid (NTA).

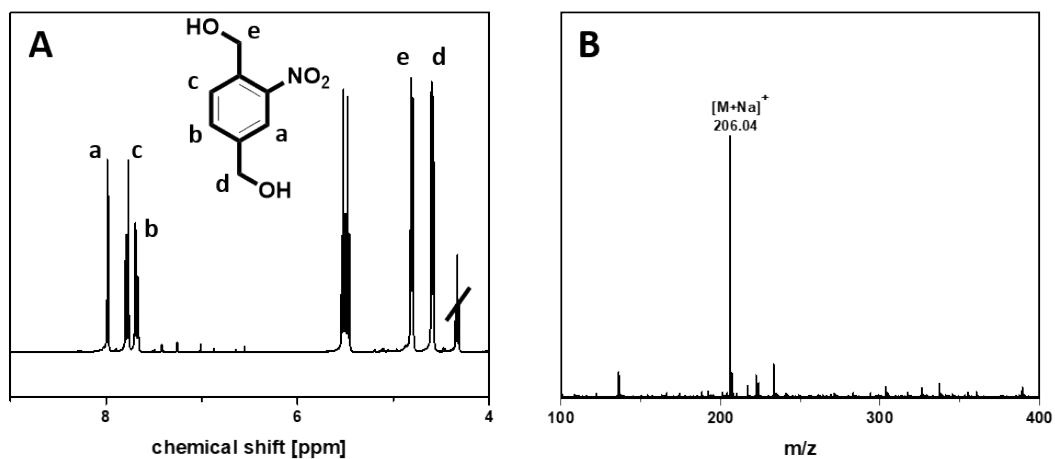


Figure S2:  $^1\text{H}$ -NMR spectrum in DMSO- $d_6$  (A) and ESI mass spectrum (B) of (2-nitro-1,4-phenylene)dimethanol (NPDM).

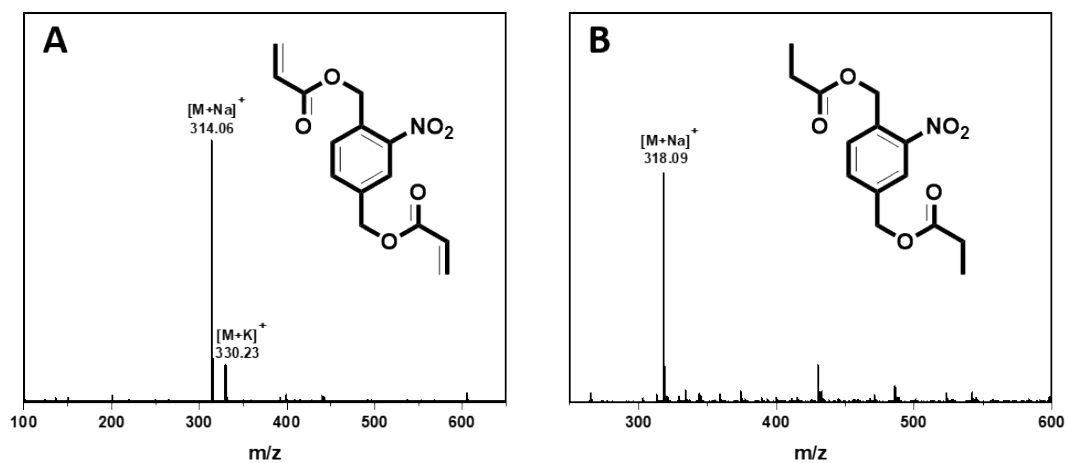
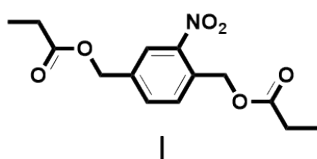
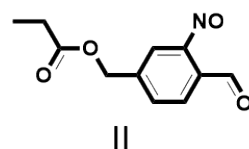


Figure S3: ESI mass spectra of (2-nitro-1,4-phenylene)bis(methylene) diacrylate (NPMDA) (A) and (2-nitro-1,4-phenylene)bis(methylene) dipropionate (NPMDP).

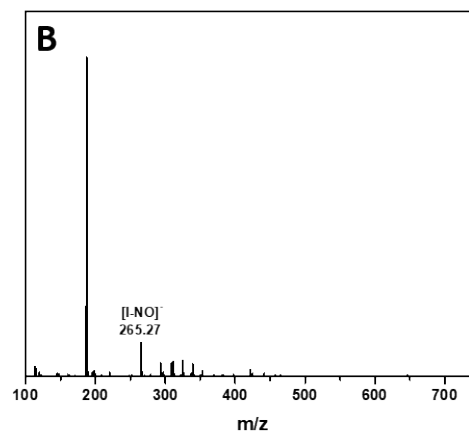
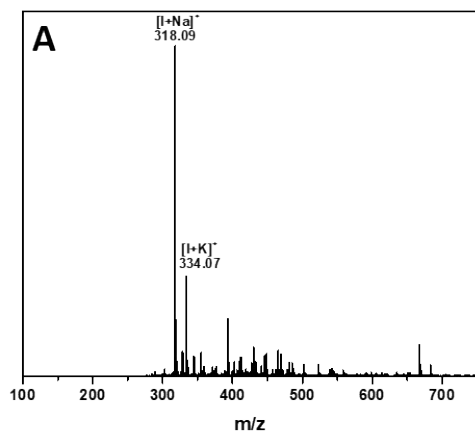


positive mode

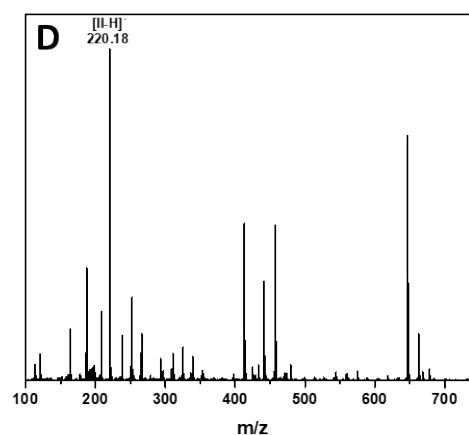
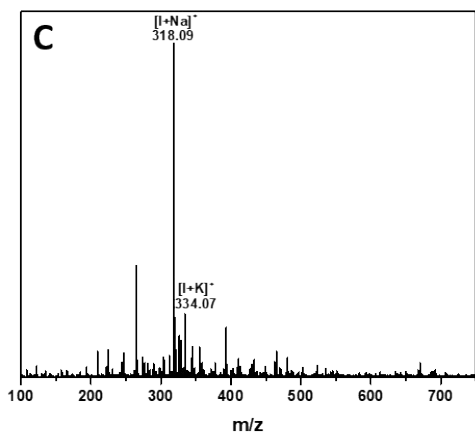


negative mode

30 min



300 min



600 min

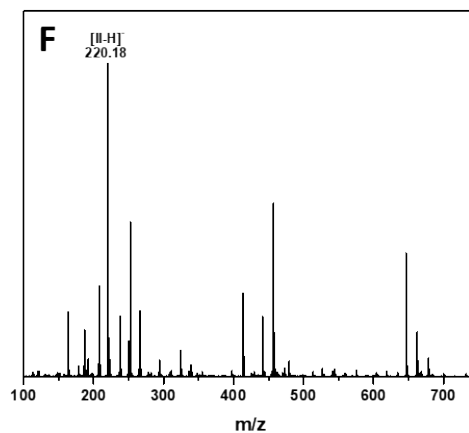
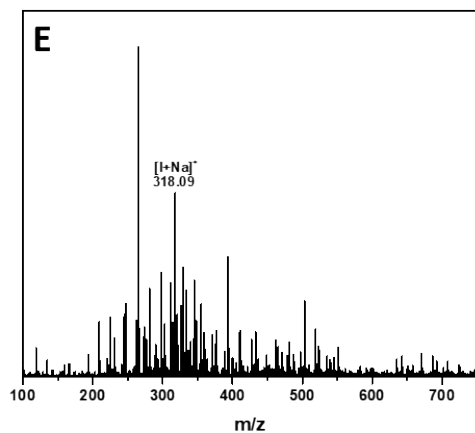


Figure S4: Positive mode ESI-MS data of NPMDP after 30 min (A), 300 min (C) and 600 min (E) of UV irradiation compared to negative mode ESI-MS spectra of NPMDP after 30 min (B), 300 min (D) and 600 min (F) of UV irradiation.

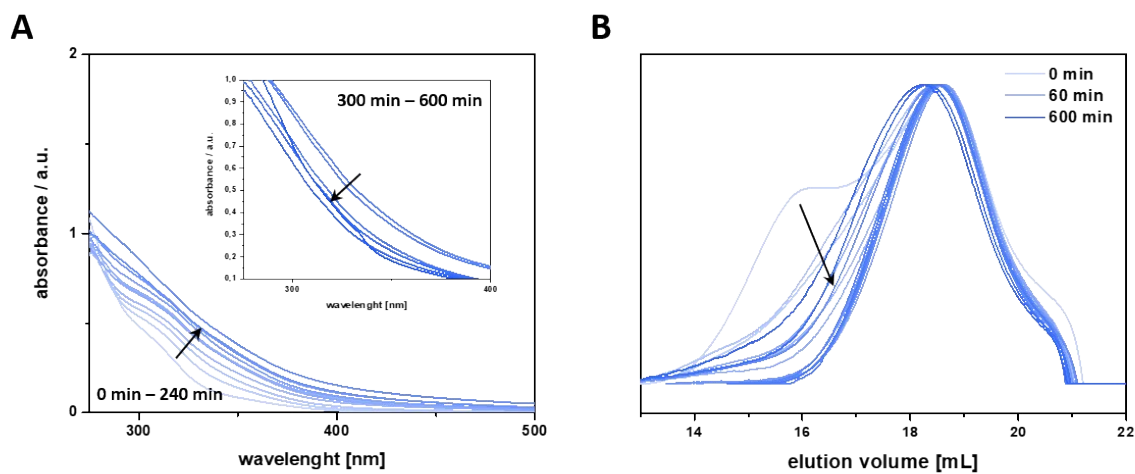


Figure S5: Change of UV-vis spectra (A) and SEC elution traces (B) of branched d-PAA1 during UV degradation in THF.

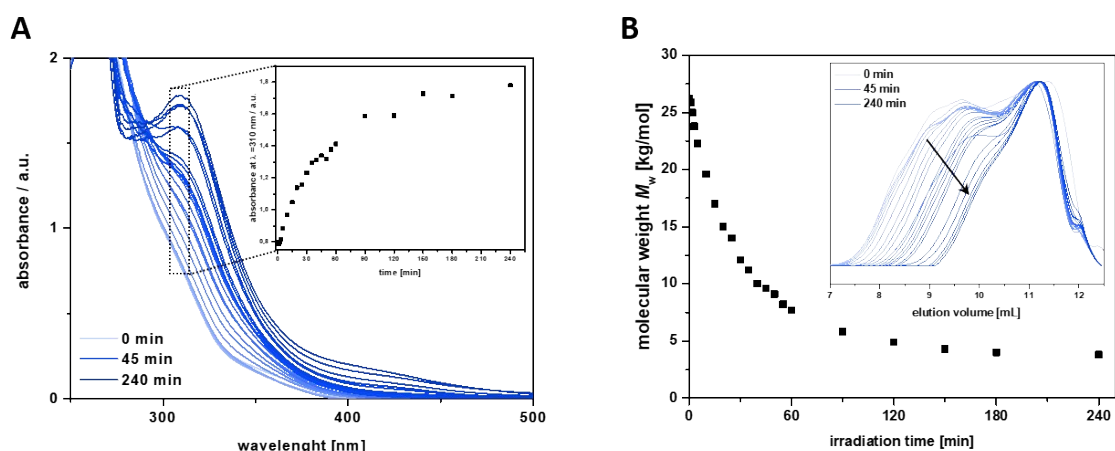


Figure S6: Change in the UV-vis spectra (A) and SEC elution traces (B) of branched d-PS2 with 4% photodegradable crosslinker during UV degradation.

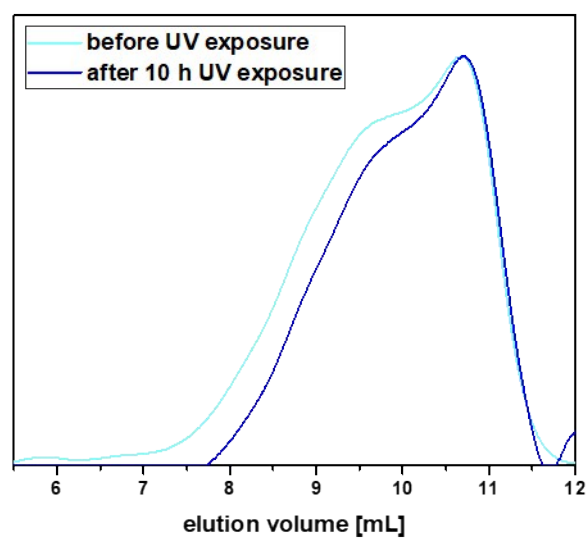


Figure S7: SEC traces of branched non-degradable PS3 before and after UV exposure.