

Electronic Supplementary Information

A Simple and Versatile Route to Amphiphilic Polymethacrylates: Catalytic Chain Transfer Polymerisation (CCTP) Coupled with Post- Polymerisation Modifications

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Additional Figures and Schemes

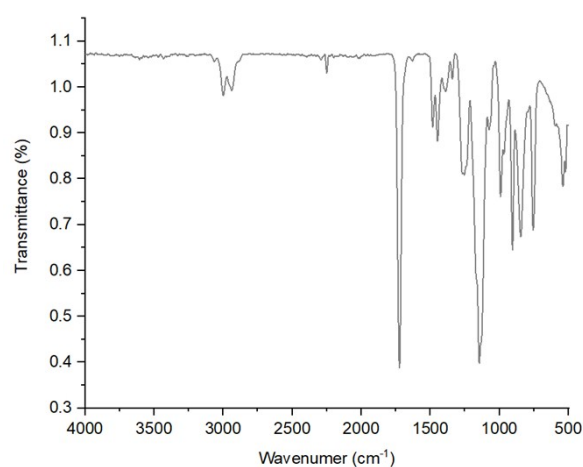


Figure S1. Typical FT-IR spectra obtained after CCTP of glycidyl methacrylate

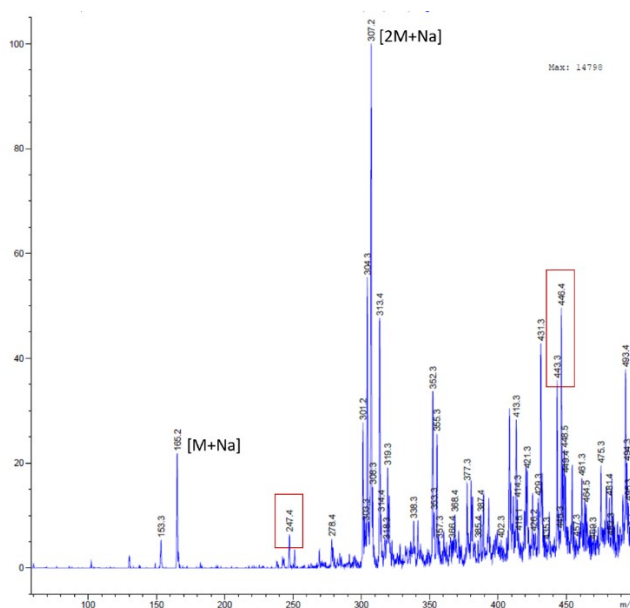


Figure S2. Electro-Spray Ionisation Mass Spectrometry (ESI-MS) spectrum of glycidyl methacrylate. In red are indicated peak not present in the blank, thought to be impurities visible in MALDI-ToF analysis.

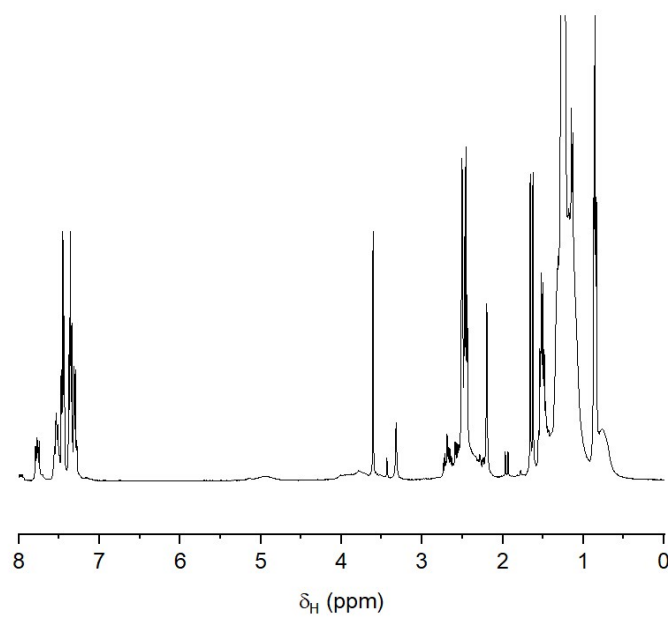


Figure S3. ^1H -NMR of unpurified p(GMA) reacted with dodecanethiol after 12 hours of storage, with disappearance of epoxide peaks due to side-reactions.

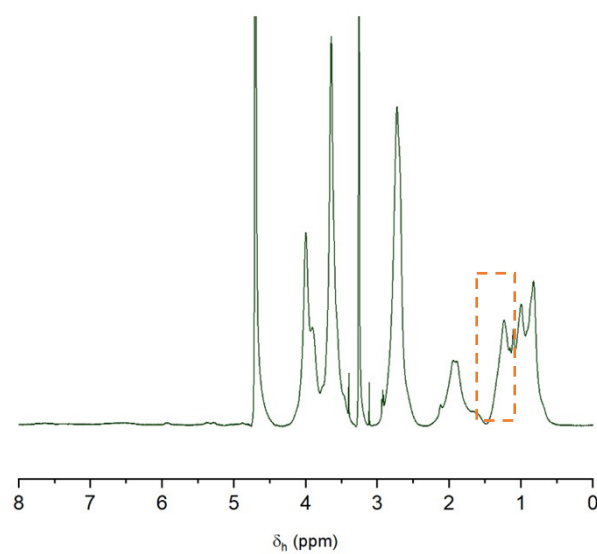


Figure S4. ^1H -NMR spectra of **P9** ran with D_2O as NMR solvent, with disappearance of dodecane peak between 1 and 1.5 ppm due to self-assembly in solution.

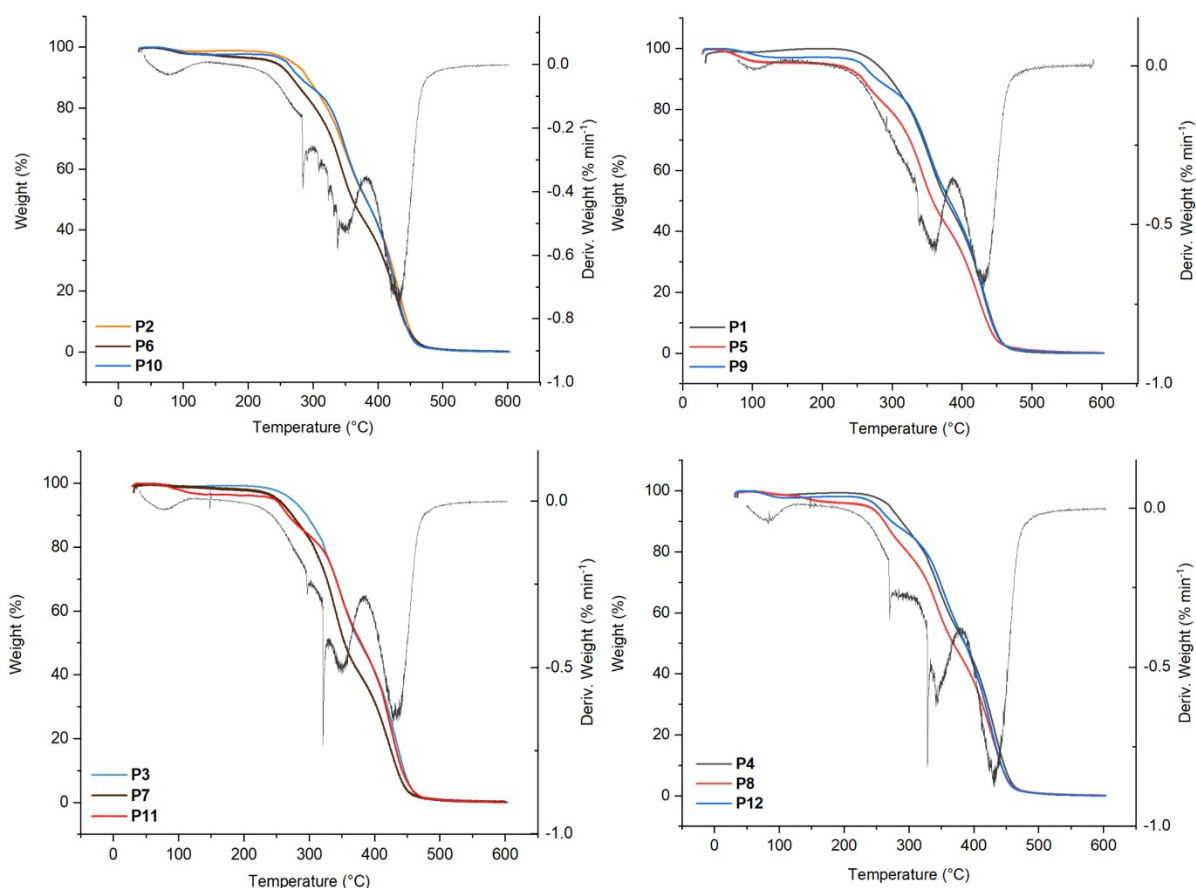


Figure S5. Thermogravimetric and differential thermogravimetric (light blue) data of p(GMA), and polymers **P1** to **P12**.

Polymer Characterisation

p(GMA):

$^1\text{H-NMR}$, (500 MHz at 25°C in CDCl_3): 0.90-1.20 and 1.10-1.20 (m, backbone CH_3), 1.60-1.70 (br, backbone CH_2), 2.60-2.70 (m, epoxide CH_2), 2.8-2.84 (m, epoxide CH_2), 3.20-3.30 (m, epoxide CH), 3.80-3.90 (m, ester- CH_2 -epoxide), 4.30-3.40 (m, ester- CH_2 -epoxide), 5.60-5.70 (br, terminal vinyl), 6.30-6.40 (br, terminal vinyl)

$^{13}\text{C-NMR}$ (/ppm, 400 MHz at 25°C in CDCl_3): 14.08-14.93 (backbone $-\text{CH}_3$), 43.34 (backbone $>\text{C}<$), 47.06 (epoxide $-\text{CH}_2-$), 52.31 (epoxide $-\text{CH}<$), 64.17 (ester- CH_2 -epoxide), 115.02 (terminal vinyl $-\text{CH}=\text{CH}_2$), 124.44 (terminal vinyl $-\text{CH}=\text{CH}_2$), 175.58-175.25-174.42 (carbonyl $>\text{C}=\text{O}$)

GPC (CHCl_3):
A: $M_n = 6670 \text{ g.mol}^{-1}$, $M_w = 14500 \text{ g.mol}^{-1}$, $\bar{D} = 2.17$
B: $M_n = 2600 \text{ g.mol}^{-1}$, $M_w = 4920 \text{ g.mol}^{-1}$, $\bar{D} = 1.87$
C: $M_n = 2170 \text{ g.mol}^{-1}$, $M_w = 3380 \text{ g.mol}^{-1}$, $\bar{D} = 1.55$

FT-IR (neat, $/\text{cm}^{-1}$): 2962 (C-H, s, medium), 1722 (C=O, s, strong), 1622 (C=C, s, medium), 1445 (CH_2 , medium), 1267 (s), 1141 (s, C-O), 905 (s, epoxide)

p(GMA)_n(DDT)(ETA): **P1/P5/P9**

$^1\text{H-NMR}$, (/ppm, 500 MHz at 25°C in DMSO-d_6): 0.78 (br, dodecyl terminal $-\text{CH}_3$), 0.86-0.96 (br, backbone + terminal $-\text{CH}_3$), 1.24 (br, dodecyl $-\text{CH}_2-$), 1.52 (br, $-\text{S-CH}_2\text{-CH}_2\text{-CH}_2-$), 1.65 (d, $-\text{S-CH}_2\text{-CH}_2\text{-CH}_2-$), 1.84 (br, backbone $-\text{CH}_2-$), 2.60 (br, $-\text{CH}_2\text{-S-}$ and $-\text{CH}_2\text{-NH-CH}_2-$), 3.42 (br, $-\text{CH}_2\text{-CH}_2\text{-OH}$), 3.80 (br, ester- $\text{CH}_2\text{-CH-}$), 4.41 (br, $-\text{CH}_2\text{-OH}$), 5.26 (br, $-\text{CH}_2\text{-CH(OH)-CH}_2-$), 5.92 (br, $\text{CH}_2\text{-NH-CH}_2$)

^{13}C -NMR, (/ppm, 500 MHz at 25°C in DMSO- d_6): 14.43 (dodecyl $-\text{CH}_3$), 22.59 (dodecyl $-\text{CH}_2-\text{CH}_3$), 28.72 (backbone $-\text{CH}_3$), 29.21-29.52 (dodecyl $-\text{CH}_2-$), 29.69 ($-\text{S}-\text{CH}_2-\text{CH}_2-$), 31.79 ($-\text{CH}_2-\text{S}-$), 32.46 ($-\text{S}-\text{CH}_2-$), 35.64 (backbone $>\text{C}<$), 52.18-52.71 ($-\text{CH}_2-\text{NH}-\text{CH}_2-$), 58.02-59.56 (backbone $-\text{CH}_2-$), 60.71 ($-\text{CH}_2-\text{CH}_2-\text{OH}$), 177.33 (carbonyl $>\text{C}=\text{O}$)

GPC (CHCl_3): **P1**: $M_n = 4652 \text{ g.mol}^{-1}$, $M_w = 13370 \text{ g.mol}^{-1}$, $\bar{D} = 2.87$
P5: $M_n = 5850 \text{ g.mol}^{-1}$, $M_w = 11750 \text{ g.mol}^{-1}$, $\bar{D} = 2.00$
P9: $M_n = 2420 \text{ g.mol}^{-1}$, $M_w = 4022 \text{ g.mol}^{-1}$, $\bar{D} = 1.66$

DLS (in water at room temperature, 5 runs average): **P1**: $d = 7.48 \text{ nm}$, $PDI = 0.067$
P5: $d = 8.60 \text{ nm}$, $PDI = 0.889$
P9: $d = 41.01 \text{ nm}$, $PDI = 0.0961$

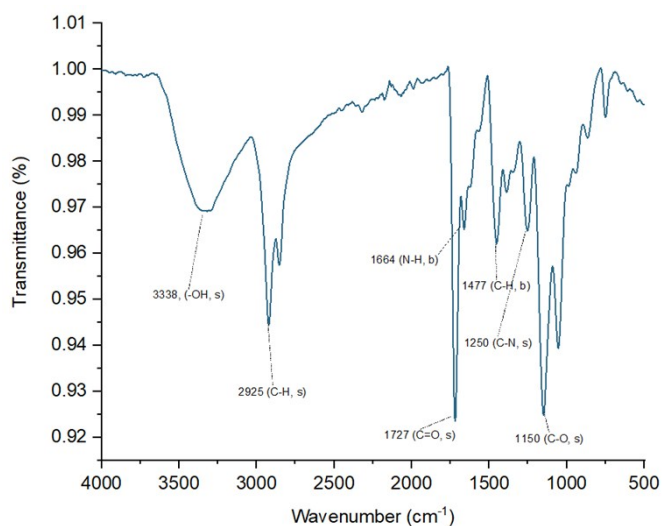


Figure S6. Typical FT-IR spectra of **P1/P5/P9**.

FT-IR (neat, $/\text{cm}^{-1}$): 3338 ($-\text{OH}$, s, broad/medium), 2925 (C-H , s, strong), 1727 ($\text{C}=\text{O}$, s, strong), 1664 (N-H , b, medium), 1477 (C-H , b, medium), 1250 (C-N , s, medium), 1150 (C-O , s, strong)

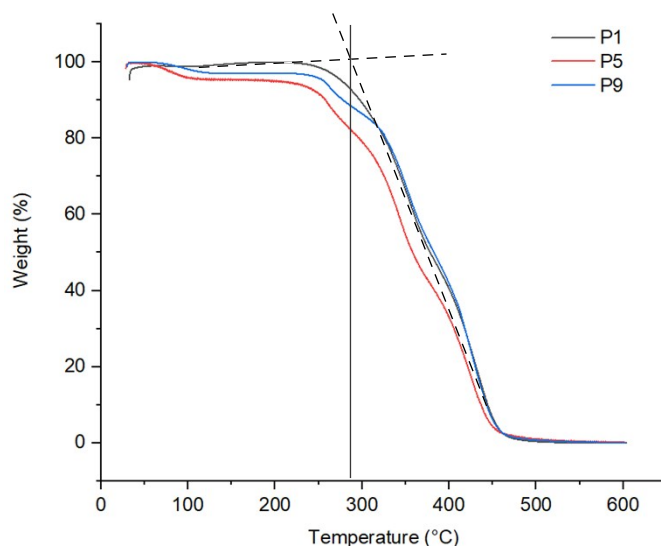


Figure S7. TGA graphs of **P1/P5/P9**.

TGA degradation onset (degradation steps temperatures): **P1**: $d_s = 287^\circ\text{C}$ (356°C , 435°C)
P5: $d_s = 270^\circ\text{C}$ (344°C , 424°C)
P9: $d_s = 290^\circ\text{C}$ (265°C , 353°C , 421°C)

p(GMA)_n(CHT)(ETA): P2/P6/P10

¹H-NMR, (/ppm, 500 MHz at 25°C in DMSO-d₆): 0.96-1.26 (br, backbone + terminal -CH₃), 1.57-1.70 (br, cyclohexane -CH₂-), 1.93 (br, backbone -CH₂-), 2.60 (br, -CH₂-S- and -CH₂-NH-CH₂-), 3.40 (br, -CH₂-CH₂-OH), 3.78 (br, ester-CH₂-CH-), 4.46 (br, -CH₂-OH), 5.27 (br, -CH₂-CH(OH)-CH₂-), 5.93 (br, CH₂-NH-CH₂)

¹³C-NMR, (/ppm, 500 MHz at 25°C in DMSO-d₆): 25.99 (cyclohexane -CH₂-), 33.76 (backbone >C< and -CH₃), 43.73 (cyclohexane, -S-CH<), 52.23 (-CH₂-NH-CH₂-), 52.81 (-CH₂-NH-CH₂-), 60.75 (backbone -CH₂-), 62.48 (-CH₂-CH₂-OH), 67.65 (ester-CH₂-CH(OH)-), 68.55 (ester-CH₂-CH(OH)-), 177.65 (carbonyl >C=O)

GPC (CHCl₃):
P2: $M_n = 4540 \text{ g.mol}^{-1}$, $M_w = 8300 \text{ g.mol}^{-1}$, $\bar{D} = 1.83$
P6: $M_n = 2800 \text{ g.mol}^{-1}$, $M_w = 5330 \text{ g.mol}^{-1}$, $\bar{D} = 1.91$
P10: $M_n = 2560 \text{ g.mol}^{-1}$, $M_w = 5030 \text{ g.mol}^{-1}$, $\bar{D} = 1.96$

DLS (in water at room temperature, 5 runs average):
P2: $d = 121.65 \text{ nm}$, $PDI = 0.294$
P6: $d = 118.15 \text{ nm}$, $PDI = 1.28$
P10: $d = 11.36 \text{ nm}$, $PDI = 0.0870$

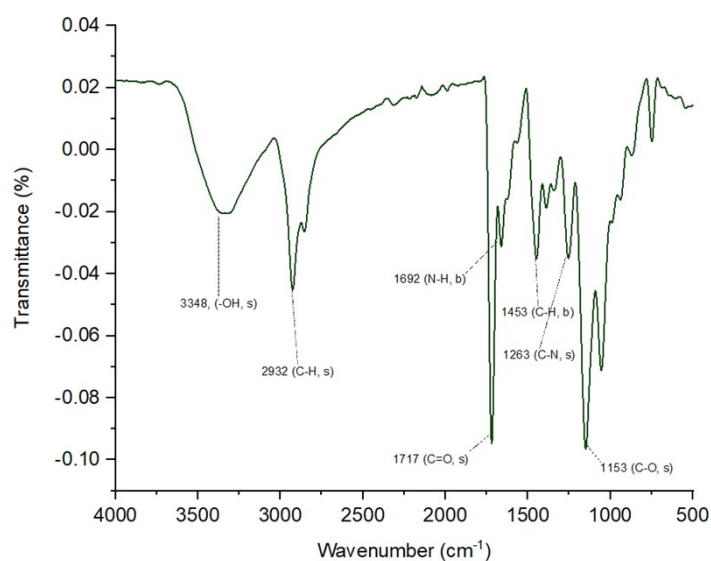


Figure S8. Typical FT-IT spectra of **P2/P6/P10**.

FT-IR (neat, /cm⁻¹): 3348 (-OH, s, broad/medium), 2932 (C-H, s, strong), 1717 (C=O, s, strong), 1692 (N-H, b, medium), 1453 (C-H, b, medium), 1263 (C-N, s, medium), 1153 (C-O, s, strong)

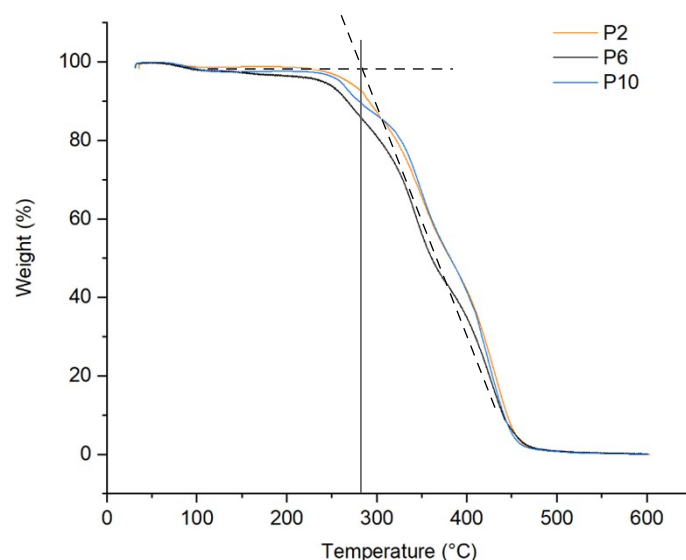


Figure S9. TGA graphs of **P2/P6/P10**.

TGA (degradation onset /°C, degradation steps temperatures):
P2: $d_s = 286^\circ\text{C}$ (351°C , 435°C)
P6: $d_s = 273^\circ\text{C}$ (343°C , 429°C)
P10: $d_s = 293^\circ\text{C}$ (265°C , 347°C , 421°C)

p(GMA)_n(PET)(ETA): P3/P7/P11

$^1\text{H-NMR}$, (/ppm, 500 MHz at 25°C in DMSO-d_6): 0.78-1.07 (br, backbone + terminal $-\text{CH}_3$), 1.85 (br, backbone $-\text{CH}_2-$), 2.60 (br, $-\text{CH}_2-\text{S}-\text{CH}_2-\text{CH}_2$ and $-\text{CH}_2-\text{NH}-\text{CH}_2-$), 2.81 (br, $-\text{CH}_2-\text{NH}-\text{CH}_2-$), 3.47 (br, $-\text{CH}_2-\text{CH}_2-\text{OH}$), 3.83 (br, ester- $\text{CH}_2-\text{CH}-$), 4.43 (br, $-\text{CH}_2-\text{OH}$), 5.24 (br, $-\text{CH}_2-\text{CH}(\text{OH})-\text{CH}_2-$), 5.48 (br, $\text{CH}_2-\text{NH}-\text{CH}_2$), 7.24 (br, aromatic $-\text{CH}=\text{}$)

$^{13}\text{C-NMR}$, (/ppm, 500 MHz at 25°C in DMSO-d_6): 33.97 (backbone $-\text{CH}_3$), 35.66-35.91 ($-\text{S}-\text{CH}_2-\text{CH}_2-$), 36.07 (backbone $>\text{C}<$), 42.86 ($>\text{CH}-\text{CH}_2-\text{S}-$), 52.19 ($-\text{CH}_2-\text{NH}-\text{CH}_2-$), 52.71 ($-\text{CH}_2-\text{NH}-\text{CH}_2-$), 58.02-58.99-59.52 (backbone $-\text{CH}_2-$), 60.73 ($-\text{CH}_2-\text{CH}_2-\text{OH}$), 67.76 (ester- $\text{CH}_2-\text{CH}(\text{OH})-$), 68.28 (ester- $\text{CH}_2-\text{CH}(\text{OH})-$), 126.58-128.58-128.95 (aromatic $=\text{CH}-$), 140.94 (aromatic $-\text{C}\leq$), 177.45 (carbonyl $>\text{C}=\text{O}$)

GPC (CHCl_3): **P3:** $M_n = 5270 \text{ g.mol}^{-1}$, $M_w = 9050 \text{ g.mol}^{-1}$, $\bar{D} = 1.72$

P7: $M_n = 3250 \text{ g.mol}^{-1}$, $M_w = 6274 \text{ g.mol}^{-1}$, $\bar{D} = 1.93$

P11: $M_n = 3530 \text{ g.mol}^{-1}$, $M_w = 6100 \text{ g.mol}^{-1}$, $\bar{D} = 1.73$

DLS (in water at room temperature, 5 runs average): **P3:** $d = 119.43 \text{ nm}$, $PDI = 0.306$

P7: $d = 15.27 \text{ nm}$, $PDI = 0.0851$

P11: $d = 6.16 \text{ nm}$, $PDI = 0.0663$

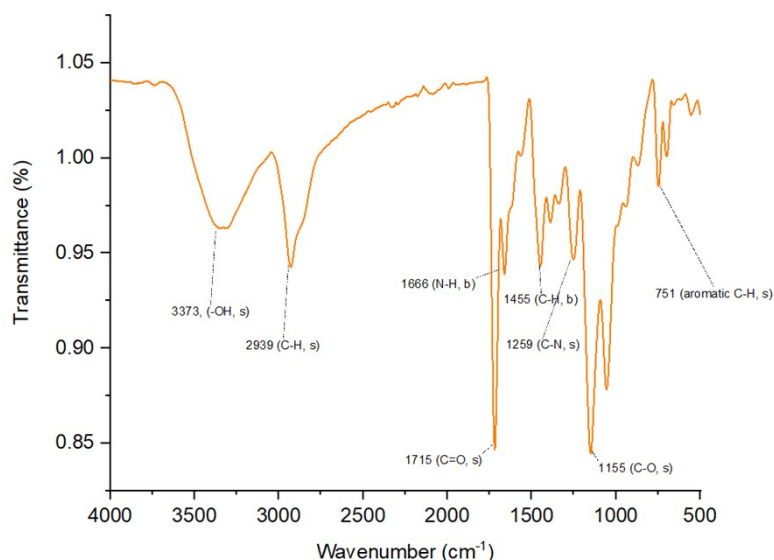


Figure S10. Typical FT-IR spectra of **P2/P6/P10**.

FT-IR (neat, /cm⁻¹): 3373 (-OH, s, broad/medium), 2939 (C-H, s, strong), 1715 (C=O, s, strong), 1666 (N-H, b, medium), 1455 (C-H, b, medium), 1259 (C-N, s, medium), 1155 (C-O, s, strong), 751 (aromatic C-H, s, medium)

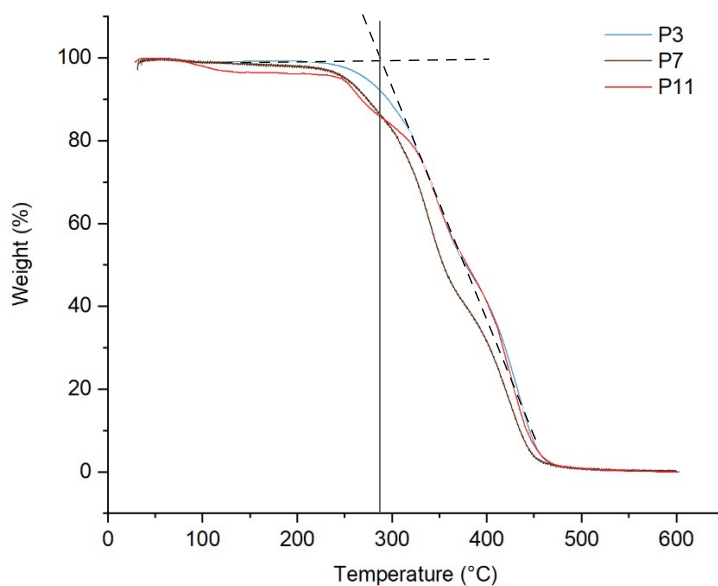


Figure S11. TGA graphs of **P3/P7/P11**.

TGA (degradation onset /°C, degradation steps temperatures):
P3: d_s = 286°C (351°C, 433°C)
P7: d_s = 270°C (342°C, 422°C)
P11: d_s = 296°C (260°C, 347°C, 423°C)

p(GMA)_n(i-PT)(ETA): P4/P8/P12

¹H-NMR, (/ppm, 500 MHz at 25°C in DMSO-d₆): 0.78-1.07 (br, backbone + terminal -CH₃), 1.21 (isopropyl -CH(CH₃)₂) 1.85 (br, backbone -CH₂-), 2.60 (br, -CH₂-S-CH₂-CH₂ and -CH₂-NH-CH₂-), 2.73 (isopropyl -CH<), 2.98 (br, -CH₂-NH-CH₂-), 3.47 (br, -CH₂-CH₂-OH), 3.79 (br, ester-CH₂-CH-), 4.45 (br, -CH₂-OH), 5.25 (br, -CH₂-CH(OH)-CH₂-), 5.92 (br, CH₂-NH-CH₂), 7.59 (aromatic >CH-, small, DMPP contamination **P4** and **P8**)

^{13}C -NMR, (/ppm, 500 MHz at 25°C in DMSO- d_6): 23.86 (isopropyl $-\text{CH}(\text{CH}_3)_2$), 34.19 (backbone $-\text{CH}_3$), 35.18 (isopropyl $-\text{CH}<$), 43.95 ($>\text{CH}-\text{CH}_2-\text{S}-$), 52.23 ($-\text{CH}_2-\text{NH}-\text{CH}_2-$), 52.78 ($-\text{CH}_2-\text{NH}-\text{CH}_2-$), 60.75 (backbone $-\text{CH}_2-$), 62.60 ($-\text{CH}_2-\text{CH}_2-\text{OH}$), 67.92 (ester- $\text{CH}_2-\text{CH}(\text{OH})-$), 68.38 (ester- $\text{CH}_2-\text{CH}(\text{OH})-$), 177.76 (carbonyl $>\text{C}=\text{O}$)

GPC (CHCl_3): **P4**: $M_n = 4510 \text{ g.mol}^{-1}$, $M_w = 8100 \text{ g.mol}^{-1}$, $\bar{D} = 1.79$
P8: $M_n = 3210 \text{ g.mol}^{-1}$, $M_w = 5920 \text{ g.mol}^{-1}$, $\bar{D} = 1.85$
P12: $M_n = 3000 \text{ g.mol}^{-1}$, $M_w = 4050 \text{ g.mol}^{-1}$, $\bar{D} = 1.35$

DLS (in water at room temperature, 5 runs average): **P4**: $d = 101.42 \text{ nm}$, $PDI = 0.467$
P8: $d = 13.53 \text{ nm}$, $PDI = 0.0896$
P12: $d = 9.66 \text{ nm}$, $PDI = 0.0769$

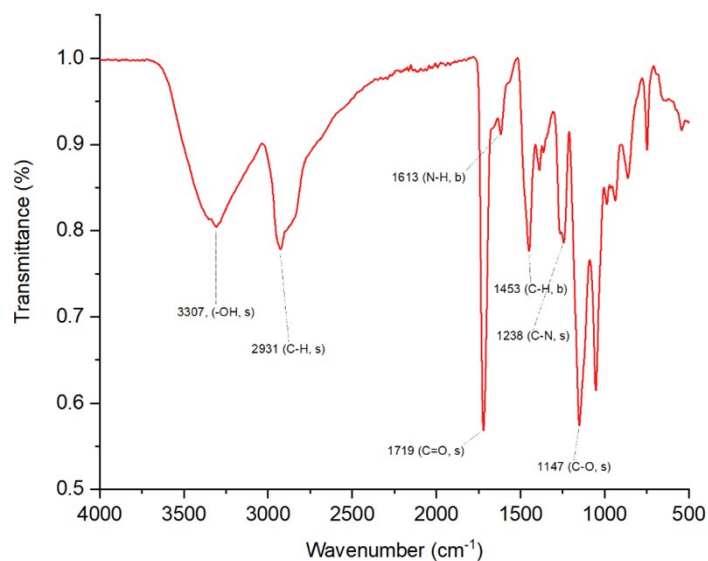


Figure S12. Typical FT-IR spectra of **P4/P8/P12**.

FT-IR (neat, $/\text{cm}^{-1}$): 3307 ($-\text{OH}$, s, broad/medium), 2931 (C-H , s, strong), 1719 ($\text{C}=\text{O}$, s, strong), 1613 (N-H , b, medium), 1453 (C-H , b, medium), 1238 (C-N , s, medium), 1147 (C-O , s, strong)

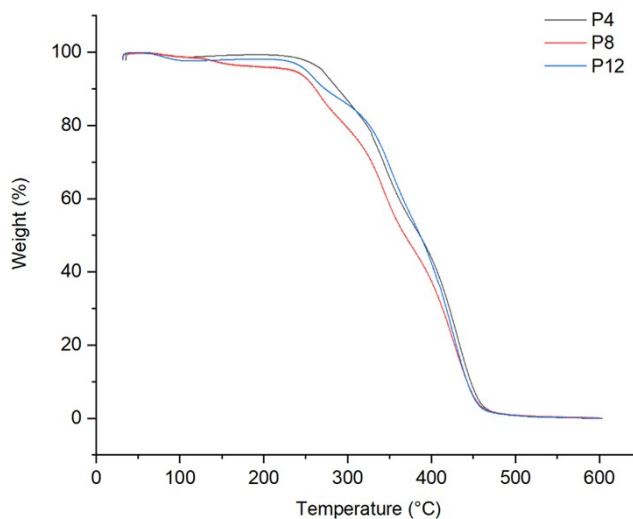


Figure S13. TGA graphs of **P4/P8/P12**.

TGA (degradation onset $/^\circ\text{C}$, degradation steps temperatures): **P4**: $d_s = 289^\circ\text{C}$ (344°C , 437°C)
P8: $d_s = 267^\circ\text{C}$ (264°C , 340°C , 431°C)
P12: $d_s = 292^\circ\text{C}$ (259°C , 349°C , 429°C)