

A facile dip-coating method for the preparation of separable MoS₂ sponges and their high-efficient adsorption behaviors of Rhodamine B

Supporting Information

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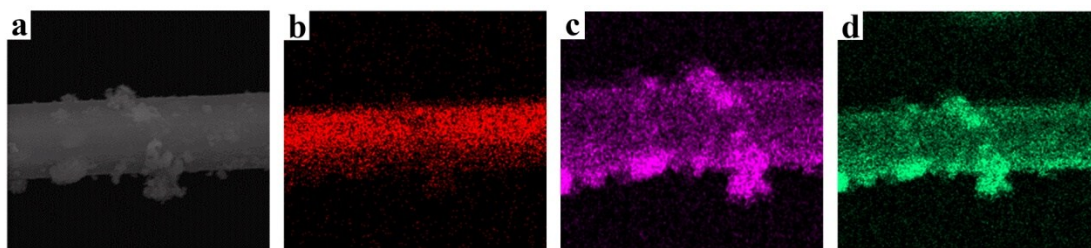


Figure S1. EDS mapping of the distribution of the (b) N element, (c) Mo element and (d) S element on the surface of MoS₂ sponge (a).

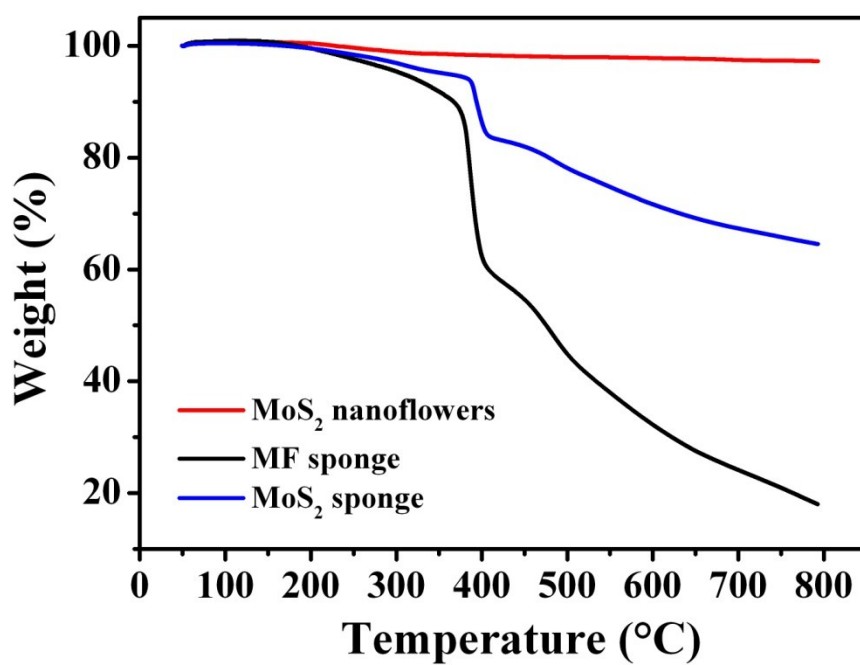


Figure S2. TG curves of MoS₂ nanoflowers, MF sponge and MoS₂ sponge.

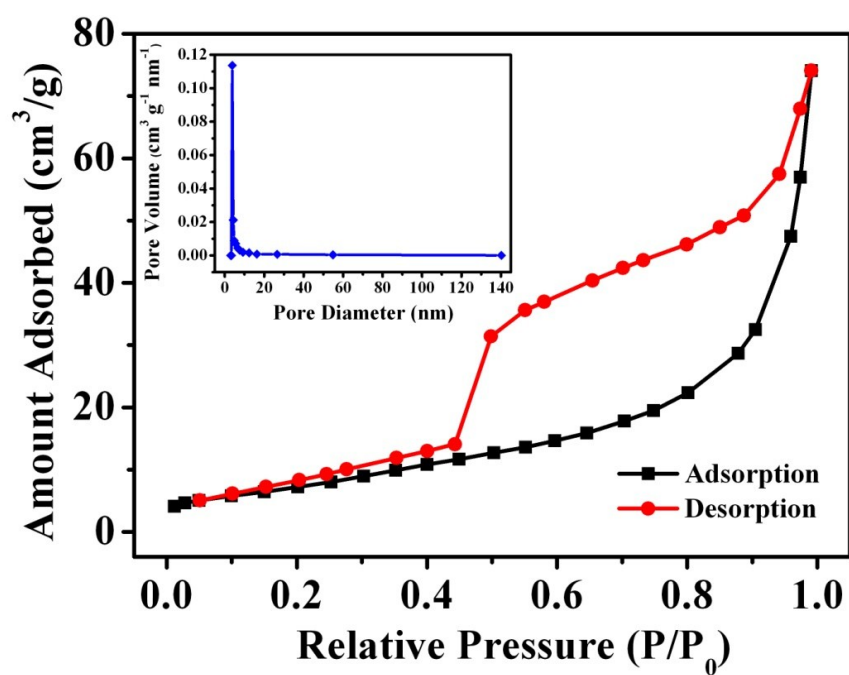


Figure S3. Nitrogen adsorption-desorption isotherms and pore size distribution (inset) of MoS₂ nanoflowers.

Table S1. The experimental zeta potential data of MoS₂ nanoflowers.

Experimental zeta potential/mV	Average zeta potential/mV
-20.51	-21.0
-21.62	
-21.21	
-20.10	
-21.56	

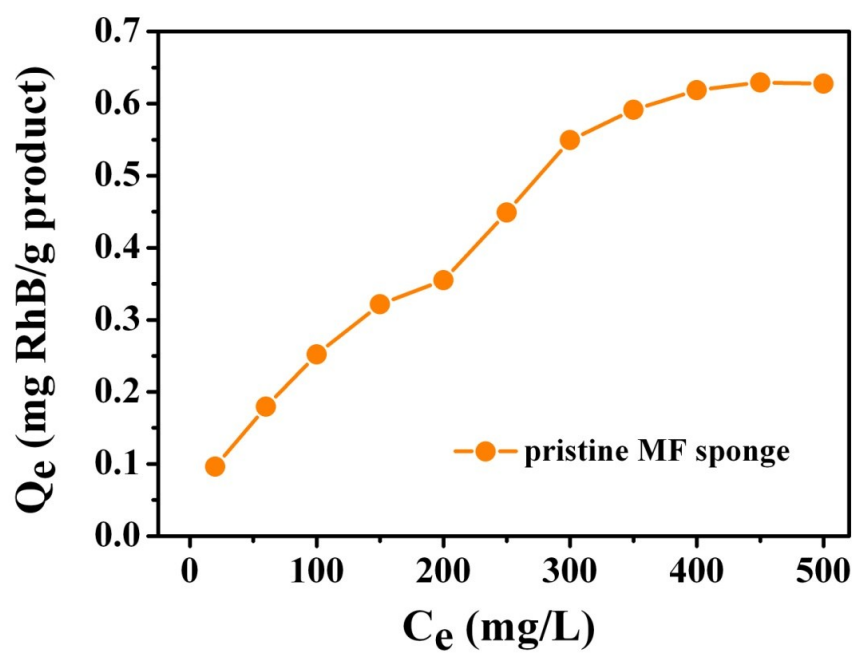


Figure S4. Adsorption isotherms of RhB onto the pristine MF sponges.