## **Electronic Supplementary Information for**

## Deflagration synthesis of nitrogen/fluorine co-doped hollow carbon

## nanoparticles with excellent oxygen reduction performance

Yousong Liu,<sup>a</sup> Bing Huang,<sup>a</sup> Shengjie Peng,<sup>bcd\*</sup> Tao Wang,<sup>b</sup> Guangbin Ji,<sup>b</sup> Guangcheng Yang<sup>a\*</sup> and Seeram Ramakrishna<sup>c</sup>

- a. Institute of Chemical Materials, China Academy of Engineering Physics, Mianyang, Sichuan, 621900, P.R. China.
- b. College of Material Science and Technology, Nanjing University of Aeronautics and Astronautics, Nanjing 210016, P. R. China.
- c. Department of Mechanical Engineering, National University of Singapore, 117574, Singapore.
- d. Key Laboratory of Advanced Energy Materials Chemistry (Ministry of Education), Nankai University, Tianjin 300071, China

\*Authors to whom correspondence should be addressed.

Electric mail: ygcheng@caep.cn; pengshengjie@nuaa.edu.cn.



Figure S1. (a) Mass spectra and (b) emission spectra of  $NaN_3/TiO_2$  deflagration generated by a 532-nm laser at 90 Jcm<sup>-2</sup>.<sup>1</sup>

Na nanoclusters and N radicals generated from NaN<sub>3</sub> deflagration have been detected in our previous work (Nano Research, just accepted, http://www.thenanoresearch.com/work\_just.asp) as shown in Fig. S1.



Figure S2. Emission spectra of Si/PTFE deflagration.<sup>2</sup>

The deflagration reaction of Na nanoclusters capturing partial F from  $C_5F_5N$  is difficult to be investigated because of its occurring in an air-tight reactor. So, we cannot obtain some reliable experimental proofs of N/F contained C radicals. However, from the emission spectra of Si/PTFE(polytetrafluoroethylene) deflagration as shown in Fig. S2, some F contained C radicals (CF<sub>2</sub>•) have been detected in the Si capturing F from PTFE deflagration process. Thus, it can be inferred that N/F contained C radicals will be generated in the Na capturing partial F from C<sub>5</sub>F<sub>5</sub>N deflagration process.



**Figure S3.** TEM (a-b) and STEM (c) images of the N/F-HC-2 sample without being washed with deionized water.



Figure S4. (a) TEM image and (b-f) C, N, F and Na mapping of unwashed N/F-HC-2 sample.



Figure S5. (a) TEM image and (b-f) C, N, F and Na mapping of N/F-HC-2 sample after washed.



**Figure S6.** XRD pattern (a), TEM images (b-c) and Carbon, Nitrogen, Fluorine, Sodium elemental mapping (d) of raw N/F-HC-0.5 sample without being washed with deionized water.



Figure S7. XPS spectra of the N/F-HC-X samples



**Figure S8.** XPS spectra of the obtained carbon product from NaN<sub>3</sub> deflagration with polytetrafluoroethylene. Inset is the C, N, F and O elemental percentage in the sample.



Figure S9. N<sub>2</sub> adsorption/desorption isotherms of N/F-HC-X samples.



Figure S10. CV curves of N/F-HC-X samples in  $O_2$  saturated 0.1 M KOH aqueous solution with a scan rate of 50 mV s<sup>-1</sup>.



**Figure S11.** LSV curves of Pt/C before and after cycling for 5000 cycles with a rotation rate of 1600 rpm.



**Figure S12.** (a) Raman spectra and (b) LSV of N/F-HC-0, N/F-HC-0.5 and N/F-HC-1 samples after calcinated at 900 °C for 2 h



Figure S13. (a) Raman spectra and (b) LSV of N-GR and N-RGO samples

Sample	C (at%)	N (at%)	O (at%)
N-RGO-1	76.476	1.808	21.725
N-RGO-2	84.292	2.181	13.526
N-RGO-3	85.163	2.994	11.843
N-GR	97.754	0.248	1.998

Table S1. The element compositions and contents in N-RGO and N-GR samples

Considering that the pressure will be further increased to cause some danger if we fabricated N/F-HC-X (X>2), a contrast experiment has been conducted to obtain N doped carbon materials with similar graphitization degree and incremental N doping lever to clarify the reason why we aimed at high doping level. We fabricated a series N doped reduced graphene oxides (labeled as N-RGO-X, where X represents the amount of  $NaN_3$  addition (g) when 0.3 g GO was added) via NaN<sub>3</sub>/GO deflagration. And N-GR referred to the graphene treated by NaN<sub>3</sub> deflagration. It can be observed from the almost same  $I_D/I_G$  intensity ratios in Fig. S13a that N-RGO samples showed similar graphitization degrees, which of course are worse than that of N-GR. As shown in Table S1, N-GR contained only 0.248 at% N dopant, which is far more less than those of N-RGO series samples (1.808 at%, 2.181 at% and 2.994 at% for N-RGO-1, N-RGO-2 and N-RGO-3, respectively). From Fig. S13b, it can be observed that N-RGO series samples show better ORR activity than N-GR due to their higher N doping levels although they exhibit lower graphitization degrees. Additionally, it also can be observed that the ORR activities of N-RGO increase with their N doping levels. All the results above indicate that the ORR activities of N doped carbon materials increase with their N doping levels when they possess similar graphitization degrees. Therefore, we designed the NaN<sub>3</sub>/ $C_5F_5N$  deflagration method to prepare N/F co-doped carbon hollow nanospheres with both high doping level and graphitization degree to exhibit excellent ORR activities.

## **References:**

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- 2 Y. S. Liu, B. Gao, Z. Q. Qiao, Y. J. Hu, W. F. Zheng, L. Zhang, Y. Zhou, G. B. Ji, G. C. Yang, Chem. Mater., 2015, 27, 4319-4327.