

Supporting Information for

In Situ Synthesis and Photocatalytic Mechanism of Cyano bridged Cu(I) polymer

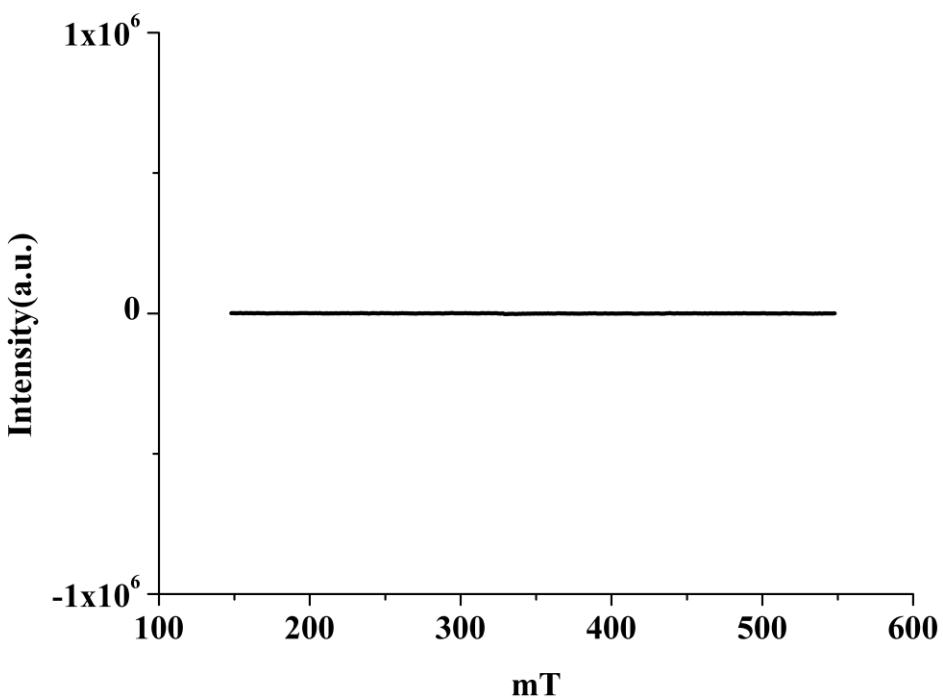
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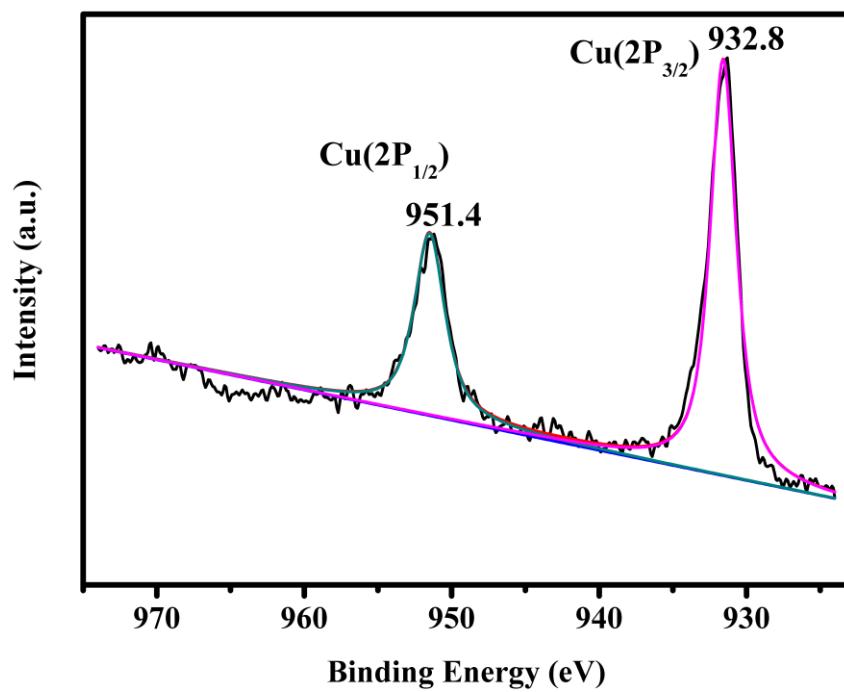
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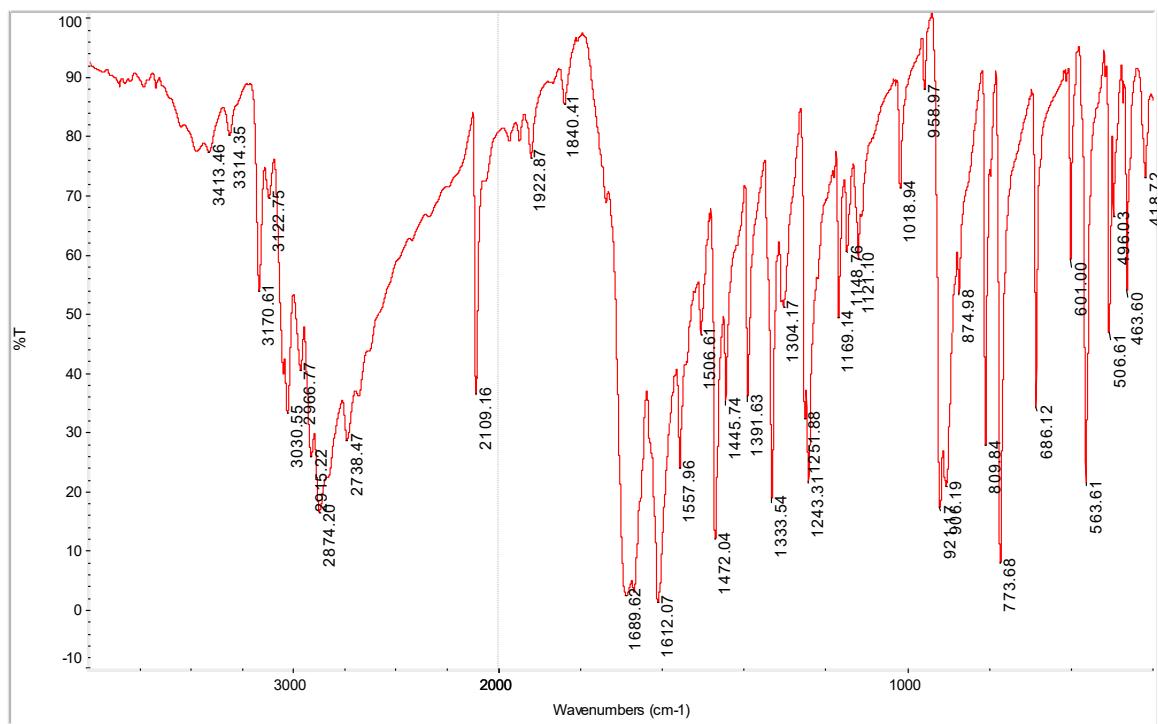
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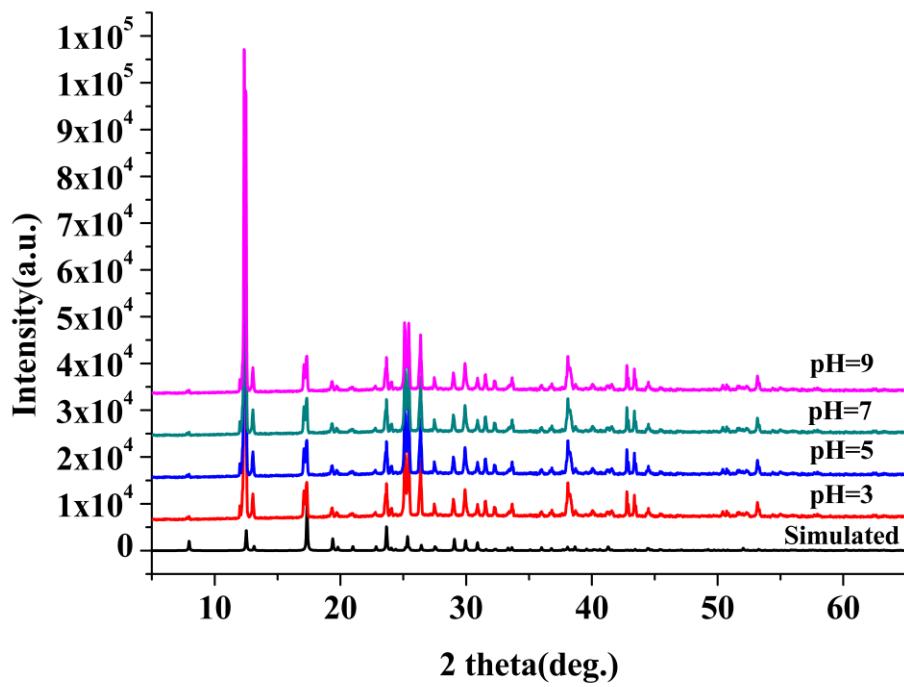
Supplementary Figure 1. EPR spectrum of powdered **1** at 173 K.



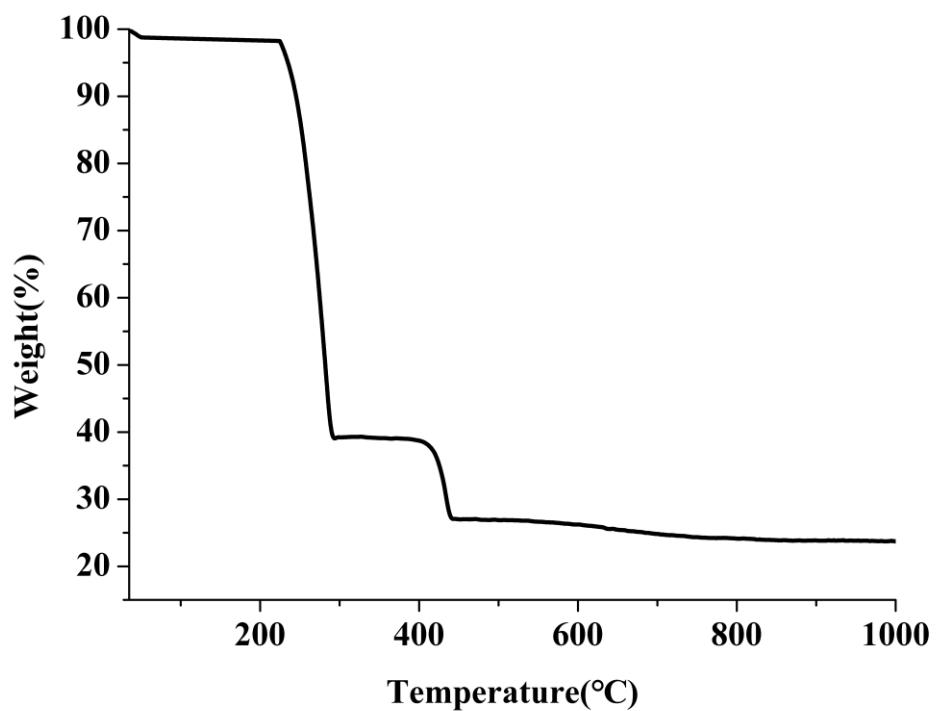
Supplementary Figure 2. XPS spectra of Cu2p region of the **1**.



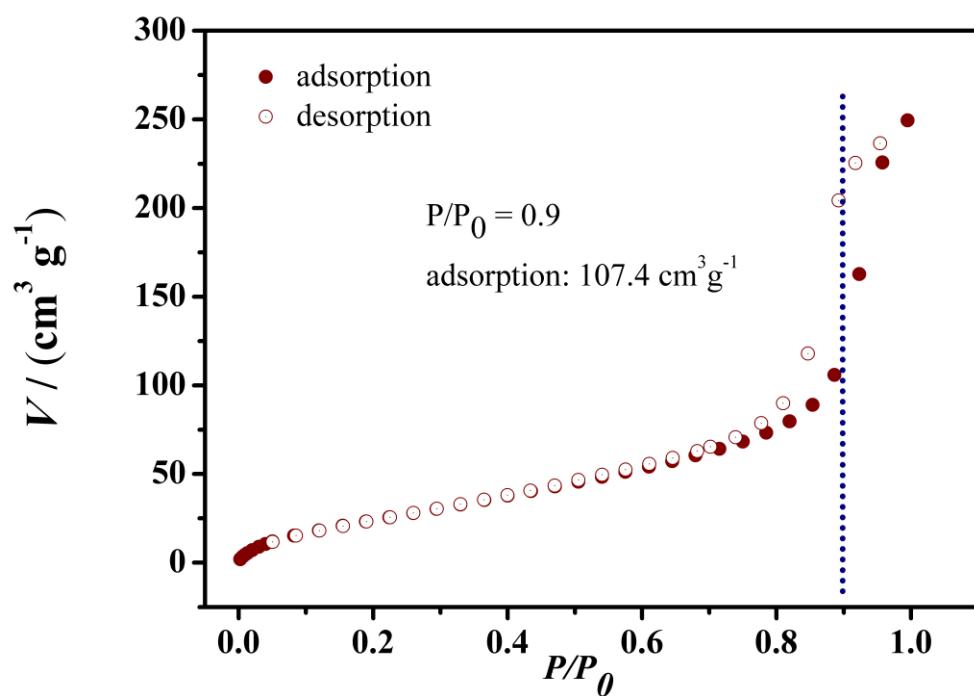
Supplementary Figure 3. IR spectrum of **1**.



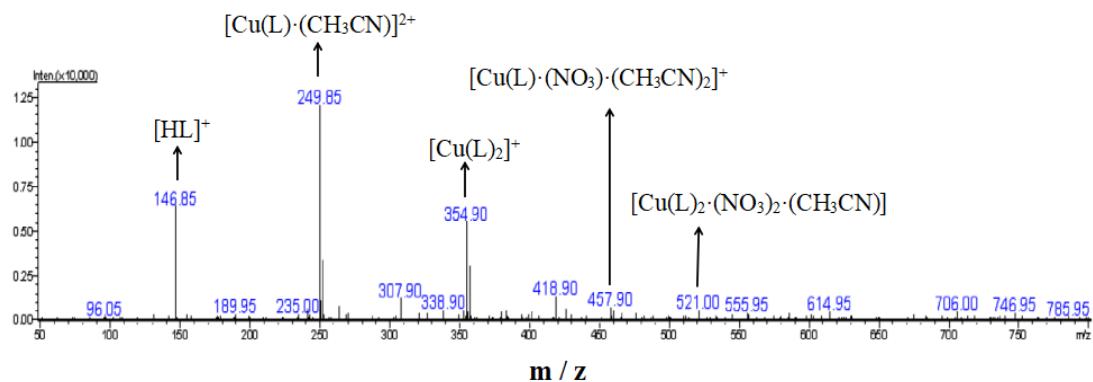
Supplementary Figure 4. Simulated and measured XRPD patterns of **1**.



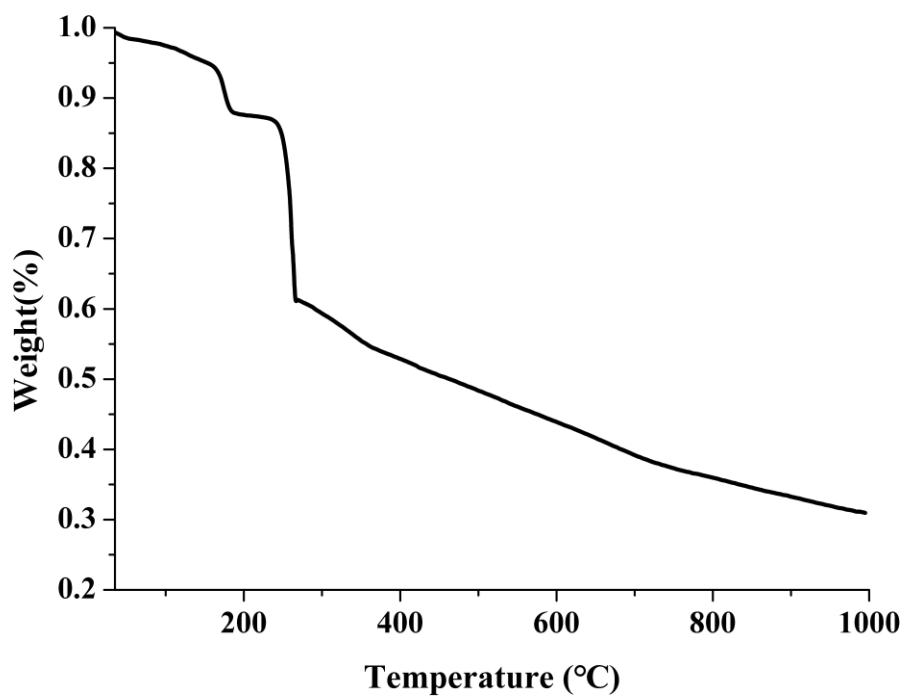
Supplementary Figure 5. TGA of 1.



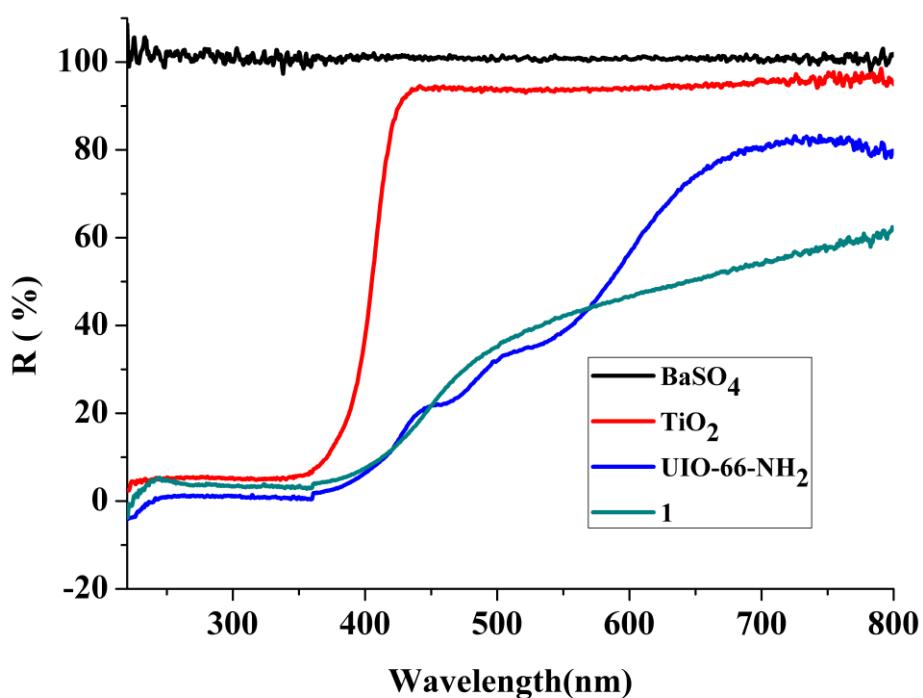
Supplementary Figure 6. The N_2 adsorption/desorption experiment of 1.



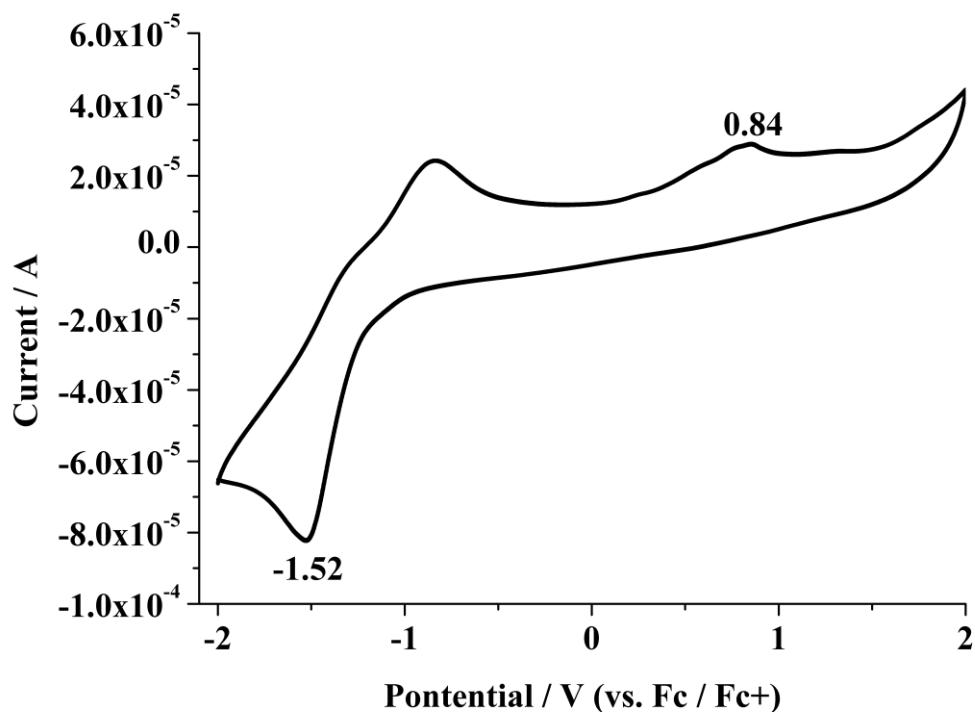
Supplementary Figure 7. ESI mass spectra of $[\text{Cu}(\text{L})_2 \cdot (\text{H}_2\text{O})_2 \cdot (\text{NO}_3)_2]$ in acetonitrile after heating at 130°C .



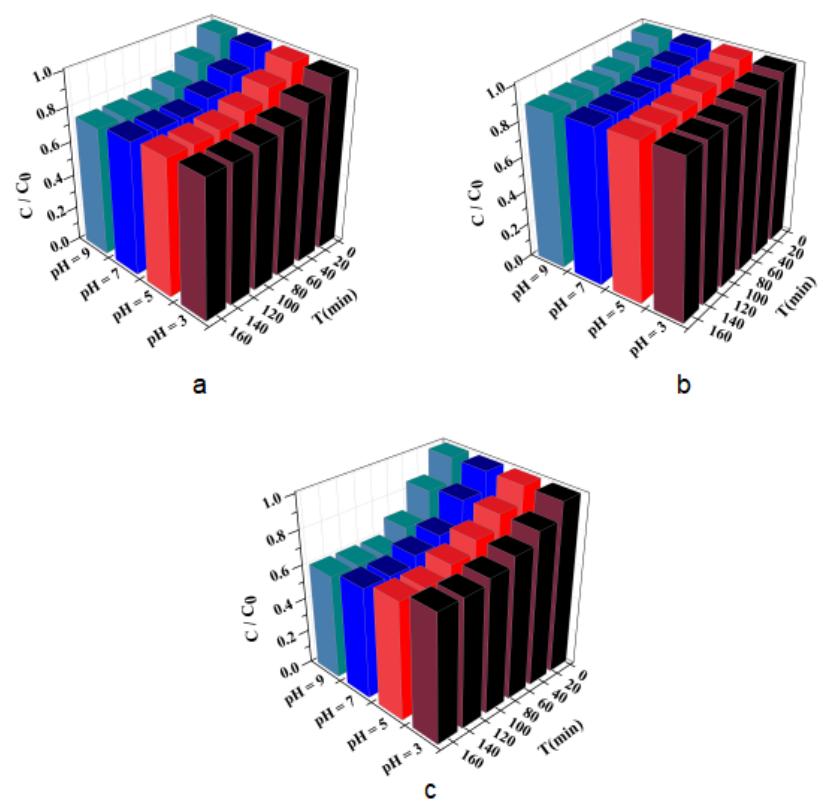
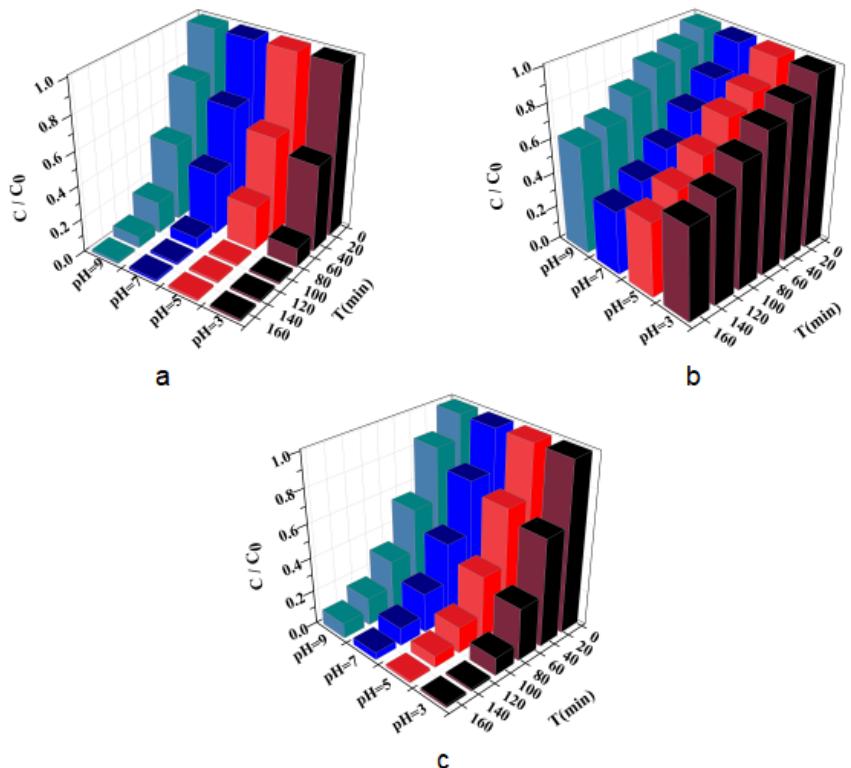
Supplementary Figure 8. TGA of $[\text{Cu}(\text{L})_2 \cdot (\text{H}_2\text{O})_2 \cdot (\text{NO}_3)_2]$.

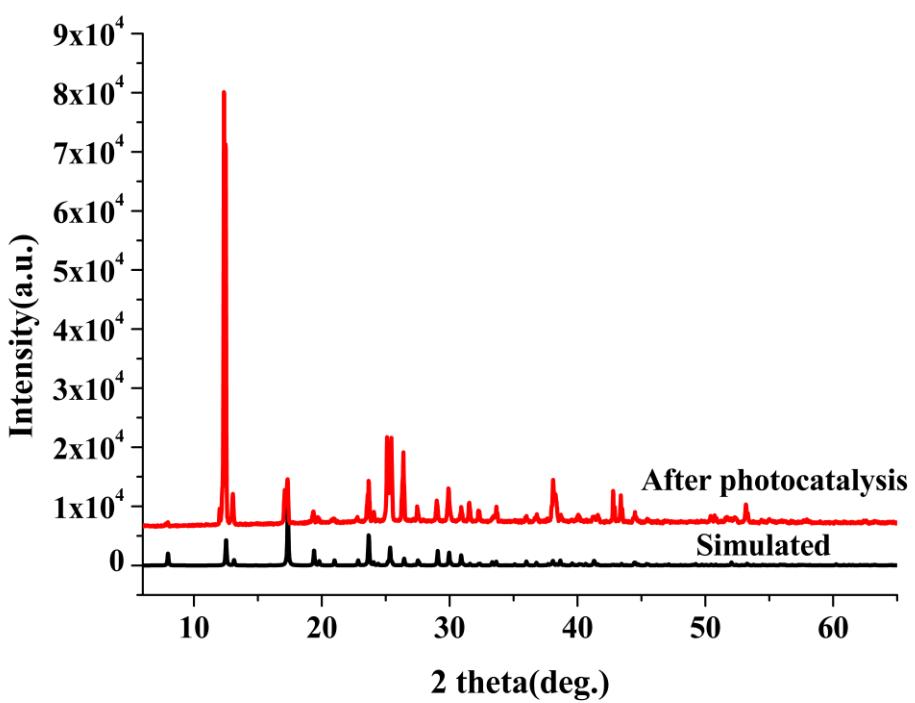


Supplementary Figure 9. UV-Vis DRS of **1**, UIO-66-NH₂ and TiO₂.

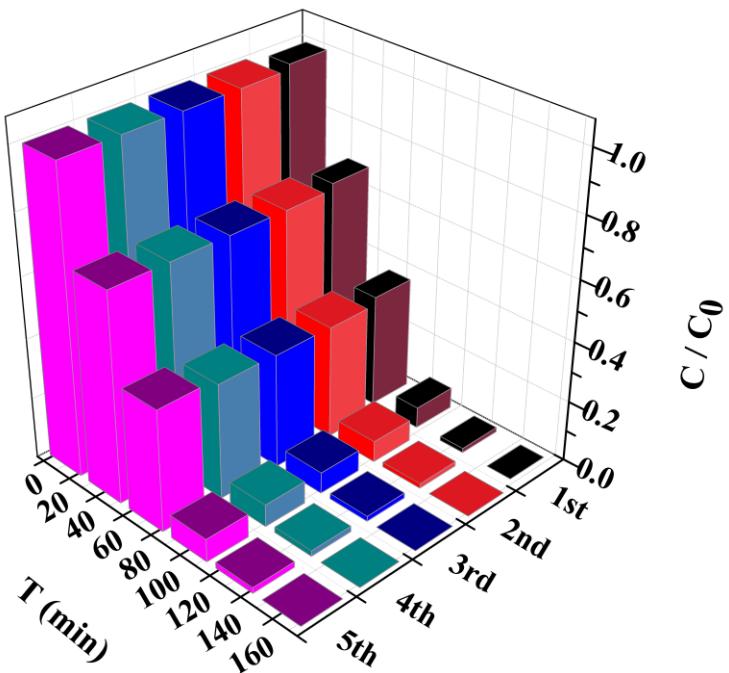


Supplementary Figure 10. Cyclic voltammograms of **1** on platinum plates in an acetonitrile solution of 0.1 mol/L Bu₄NPF₆ at a sweep rate of 0.1 V/s.

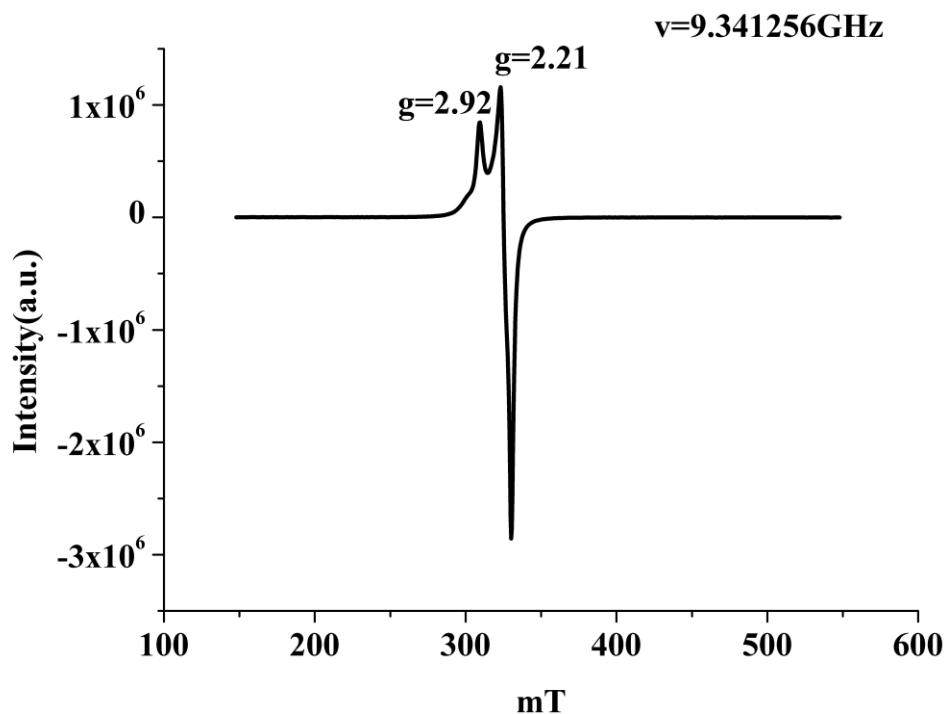




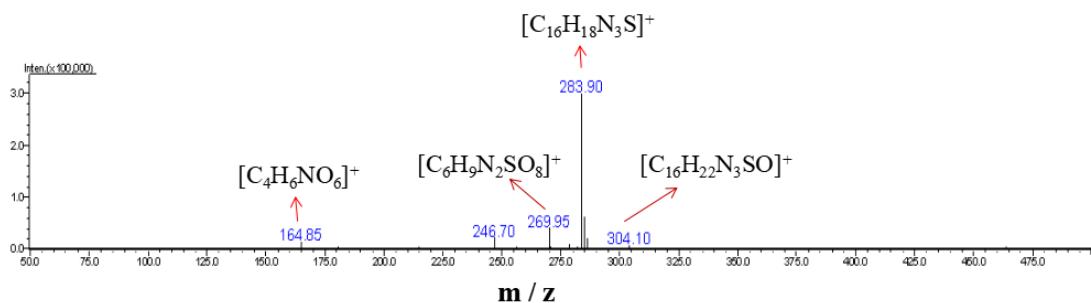
Supplementary Figure 13. Simulated and after photocatalysis XRPD patterns of **1**.



Supplementary Figure 14. Cycle runs of **1** for the degradation of MB at pH=7.



Supplementary Figure 15. In situ EPR spectrum of **1** sample under 175 W high-pressure mercury lamp radiation catalyzed the MB at 173 K.



Supplementary Figure 16. ESI mass spectra of **1** photocatalytic degradation of MB.

Supplementary Table 1. The zeta potential vs. pH curves of **1**, UIO-66-NH₂ and TiO₂.

pH	Zeta potential (mV)		
	1	UIO-66-NH ₂	TiO ₂
3	38.52	27.5	18.84
5	23.5	16.59	5.81
7	11.12	5.74	-9.84
9	-5.68	-9.56	-22.35

Supplementary Table 2. Frontier electron densities on atoms of MB calculated by DFT analysis.

Atom (number)	2FED ² _{HOMO}	FED ² _{HOMO} + FED ² _{LUMO}
C(1)	0.0025	0.0937
C(2)	0.0076	0.0780
C(3)	0.0844	0.0525
C(4)	0.0981	0.2257
C(5)	0.0260	0.0167
C(6)	0.0009	0.0799
C(7)	0.0981	0.2258
C(8)	0.0844	0.0525
C(9)	0.0076	0.0780
C(10)	0.0025	0.0934
C(11)	0.0009	0.0799
C(12)	0.0259	0.0167
S(13)	0.0019	0.0009
N(14)	1.4971	0.7485
C(15)	0.0000	0.0171
C(16)	0.0000	0.0171
C(17)	0.0000	0.0171
C(18)	0.0000	0.0171
N(19)	0.0002	0.0217
N(20)	0.0002	0.0216

The chemical structure of MB is shown with atom numbering. A reaction scheme indicates the conversion of MB to a substituted product where the S(13) atom is replaced by a red oxygen atom.

Supplementary Table 3. Crystallographic Data of **1**.

Polymer 1			
Empirical formula	C ₁₇ H ₁₂ CuN ₅ O ₂	Z	4
Formula weight	381.86	Dcalcd, (g/cm ³)	1.676
Temperature (K)	293(2)	S	1.05
Wavelength (Å)	0.71073	Limiting indices	-6 ≤ h ≤ 5; -17 ≤ k ≤ 18; -18 ≤ l ≤ 28
Crystal system	Orthorhombic	F(000)	776
Space group	Pbcn	θ(°)	3.0 to 28.3
a (Å)	4.8345(1)	Refs collected total / unique	4432 / 1746
b (Å)	14.1398(5)	(Δ/σ) _{max}	0.001
c (Å)	22.1826(8)	Data/restraints/paramets	1746 / 0 / 114
α (°)	90	μ(mm ⁻¹)	1.46
β (°)	90	Final R indices; [I>2σ(I)]	R ₁ = 0.042; wR ₂ = 0.106
γ (°)	90	Δρ _{max} /Δρ _{min}	0.57 / -0.86

Volume (Å ³)	1516.38(8)	R _(int)	0.022
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Supplementary Table 4. Selected Bond Lengths (nm) and Bond Angles (°) for **1**.

Bond	Dist. (Å)	Bond	Dist. (Å)	Bond	Dist. (Å)
N(1)-Cu(1)	2.343(2)	C(9)-Cu(1)	1.918(2)	N(3)-Cu(1)	1.918(2)
Angle	(°)	Angle	(°)	Angle	(°)
N(1) ^A -Cu(1)-N(1)	113.89(11)	N(3)-Cu(1)-N(1)	90.68(9)	N(3) ^A -Cu(1)-N(1)	106.63(9)
N(3) ^A -Cu(1)-N(3)	148.29 (14)	C(9)-Cu(1)-N(1)	90.68(9)	O(1)-C(2)-N(2)	120.8 (2)

Symmetry codes: (A) -x+1, y, -z+3/2.

Supplementary Table 5. Hydrogen bonds of **1**.

D-H···A	D-H/ Å	H···A/ Å	D···A/ Å	D-H···A
N(2)-H(2)···O(1) ^B	0.86	1.94(1)	2.802(3)	177(1)

Symmetry codes: (B) -x+1, -y+2, -z+1.