

Electronic Supplementary Information (ESI) for

**A size-matched POM@MOF composite catalyst for highly efficient
and recyclable ultra-deep oxidative fuel desulfurization**

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Characterization methods

All reagents and solvents employed were commercially available and used as received without further purification. Fourier transform Infrared spectra (FT-IR) were obtained in KBr disks on a Nicolet Avatar 360 FT-IR spectrometer in the range of 4000-400 cm^{-1} . Elemental analyses (C, H, N) was performed using a Vario EL III CHNS elemental analyzer. Thermogravimetric analysis (TGA) was performed on TA Instruments Q50. Thermogravimetric Analyzer was performed under nitrogen flow of (40 mL min^{-1}) at a typical heating rate of 10 $^{\circ}\text{C min}^{-1}$. Powder X-ray diffraction (PXRD) experiments were performed on a D8 Advance X-ray diffractometer with Cu $K\alpha$ radiation ($\lambda = 1.5418 \text{ \AA}$). High Performance Liquid Chromatography (HPLC) experiments were performed on Agilent Technologies 1100 with a Hypersil ODS column. Atomic absorption spectroscopy (AAS) was performed on Solaar M6 MK2 equipment. Scanning electron microscopy (SEM) and energy dispersive spectroscopy (EDS) analysis were carried out on the S-4800 Field Emission Scanning Electron Microscope (Hitachi, Japan).

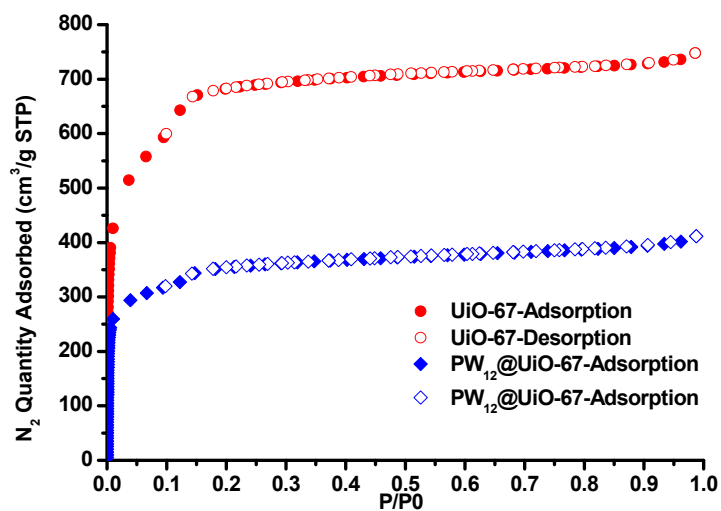


Fig. S1 N₂ adsorption/desorption isotherms of UiO-67 and PW₁₂@UiO-67.

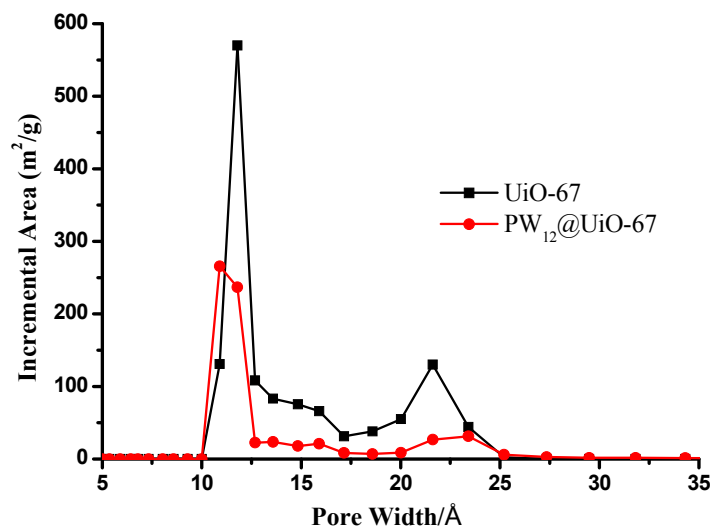


Fig. S2 Density functional theory pore size distribution (PSD) profiles of UiO-67 and PW₁₂@UiO-67.

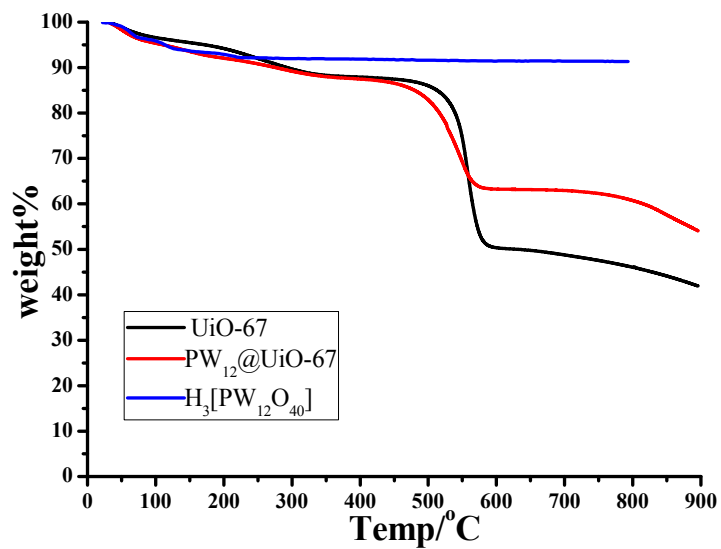


Fig. S3 TGA plots of H₃PW₁₂O₄₀, UiO-67 and PW₁₂@UiO-67.

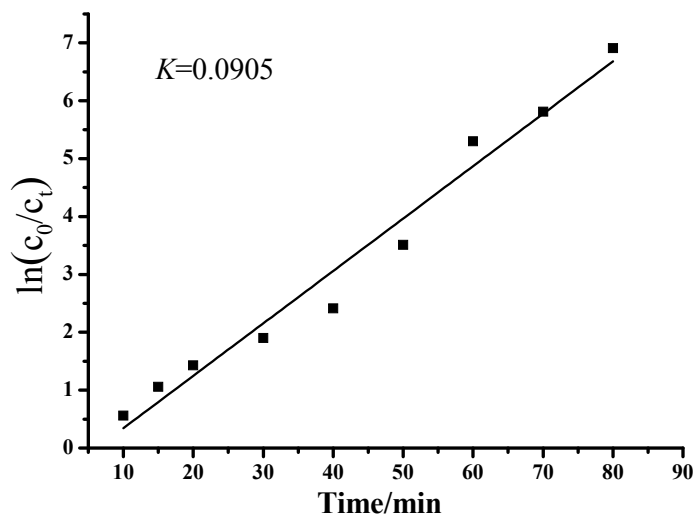


Fig. S4 Pseudo-first-order kinetic plot for the oxidation of DBT catalyzed by PW₁₂@UiO-67 (reaction condition: 70 °C; O/S = 13).

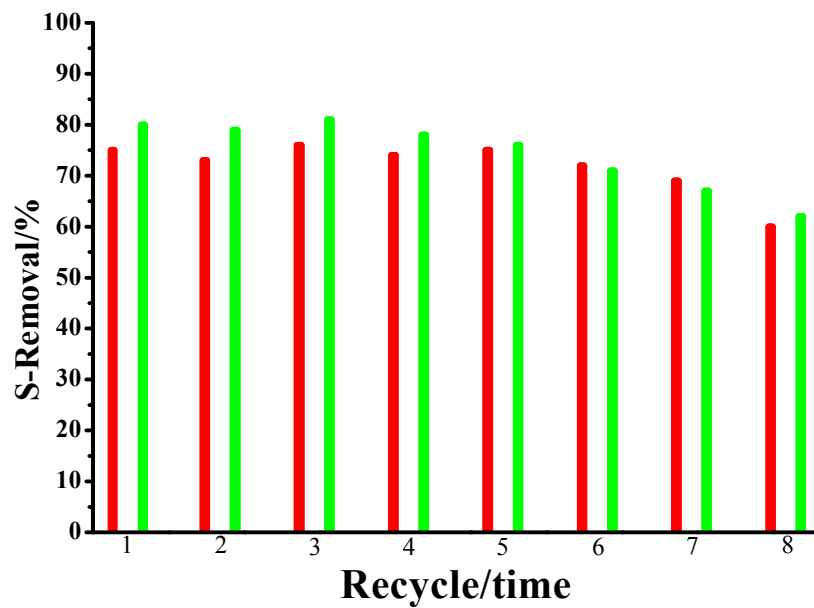


Fig. S5 Removal rate of DBT (red) and 4, 6-DMDBT (green) among 8 consecutive catalytic reactions.

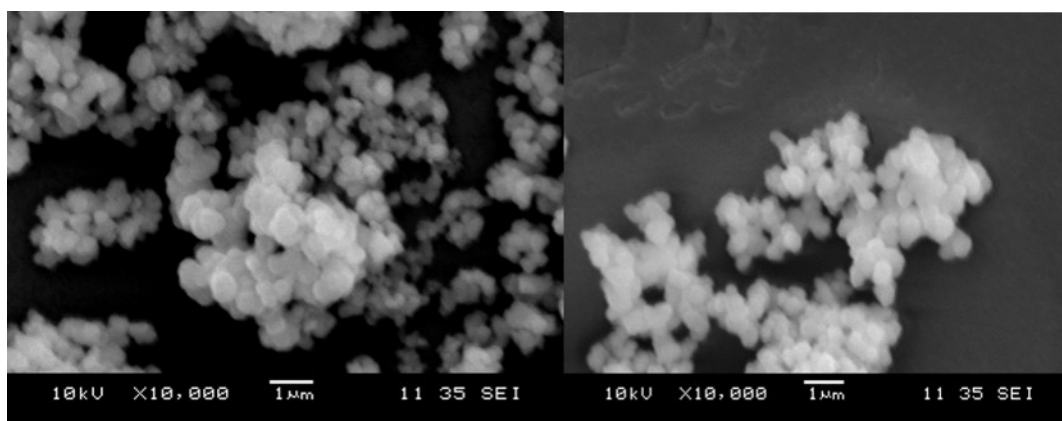


Fig. S6 SEM images of UiO-67 (left) and PW₁₂@UiO-67 (right).