

## Supporting Information

### Efficient Catalyst of Co@CoP<sub>x</sub> Core-shell Nanochains for Oxygen Evolution Reaction

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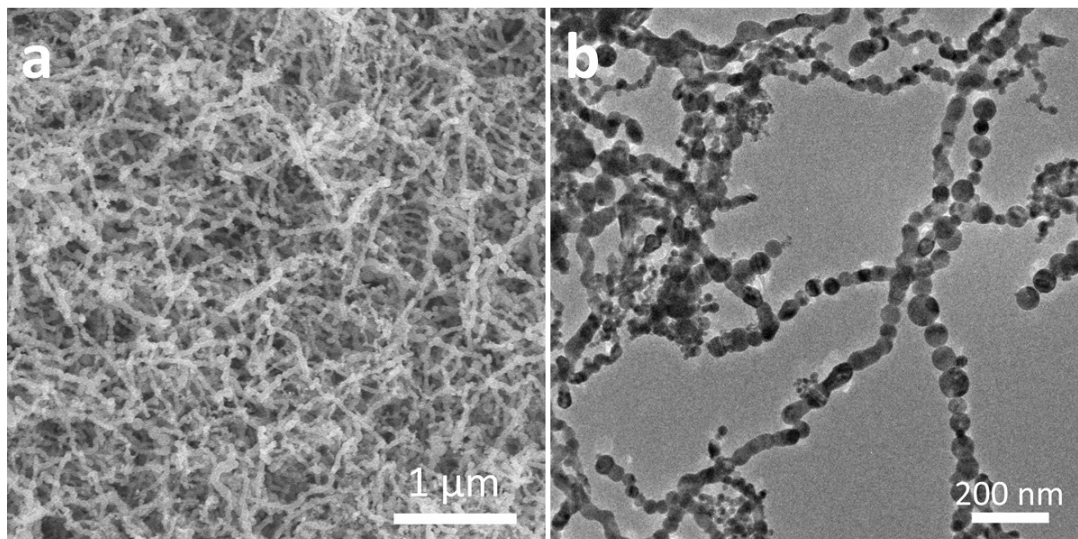
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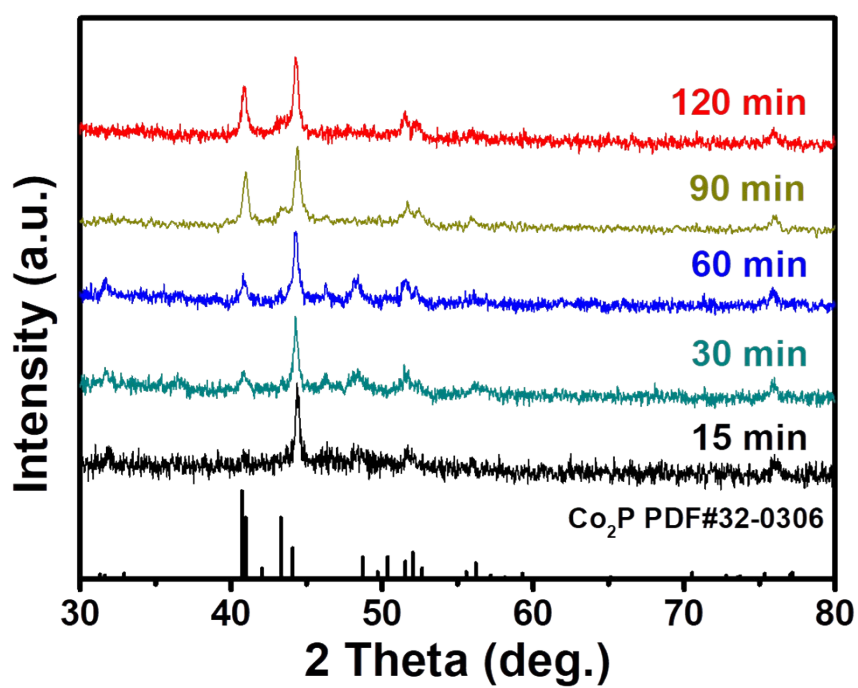
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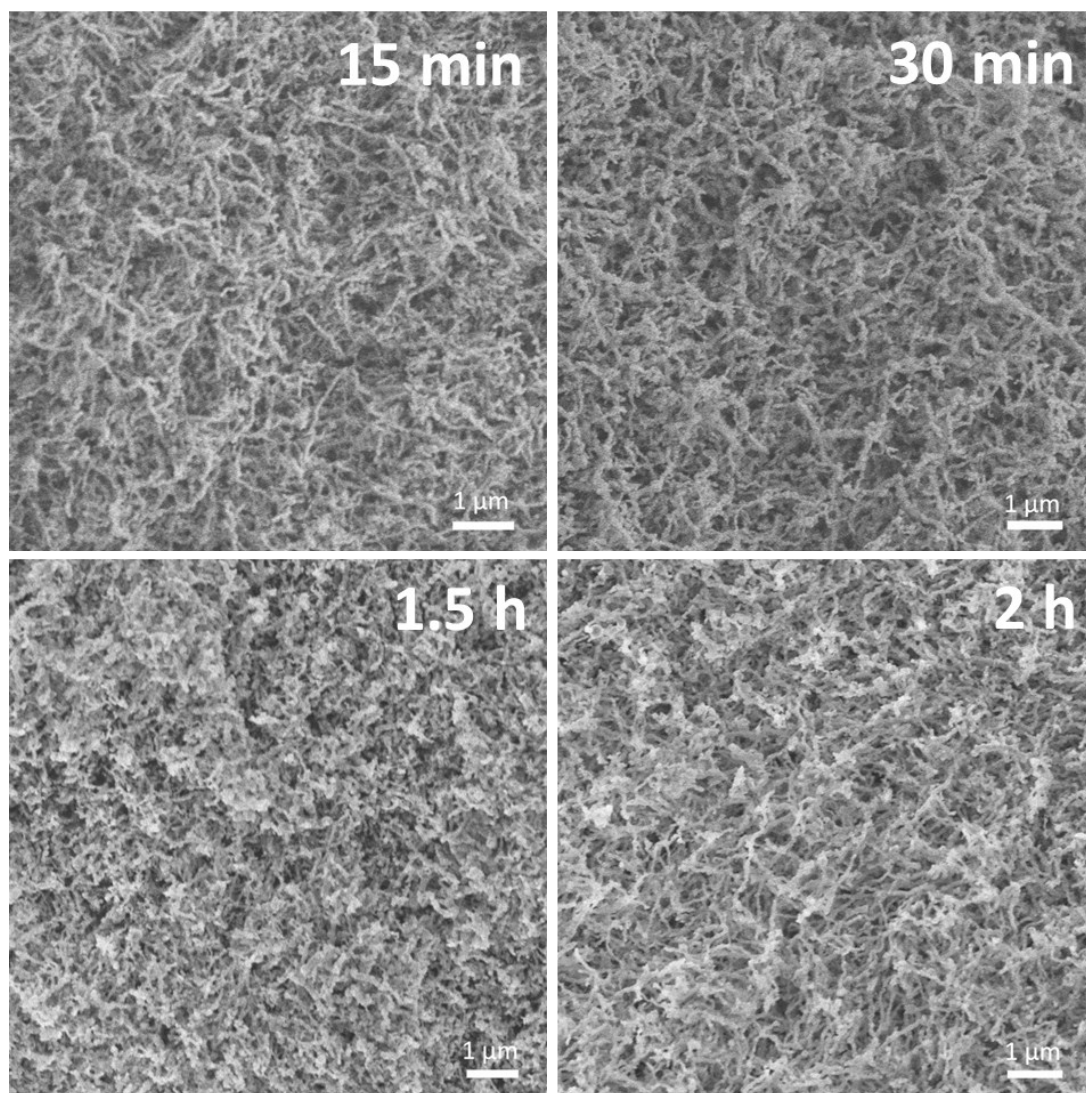
‡ Xiaotao Yuan and Zhe Zhang contribute equally to this work.



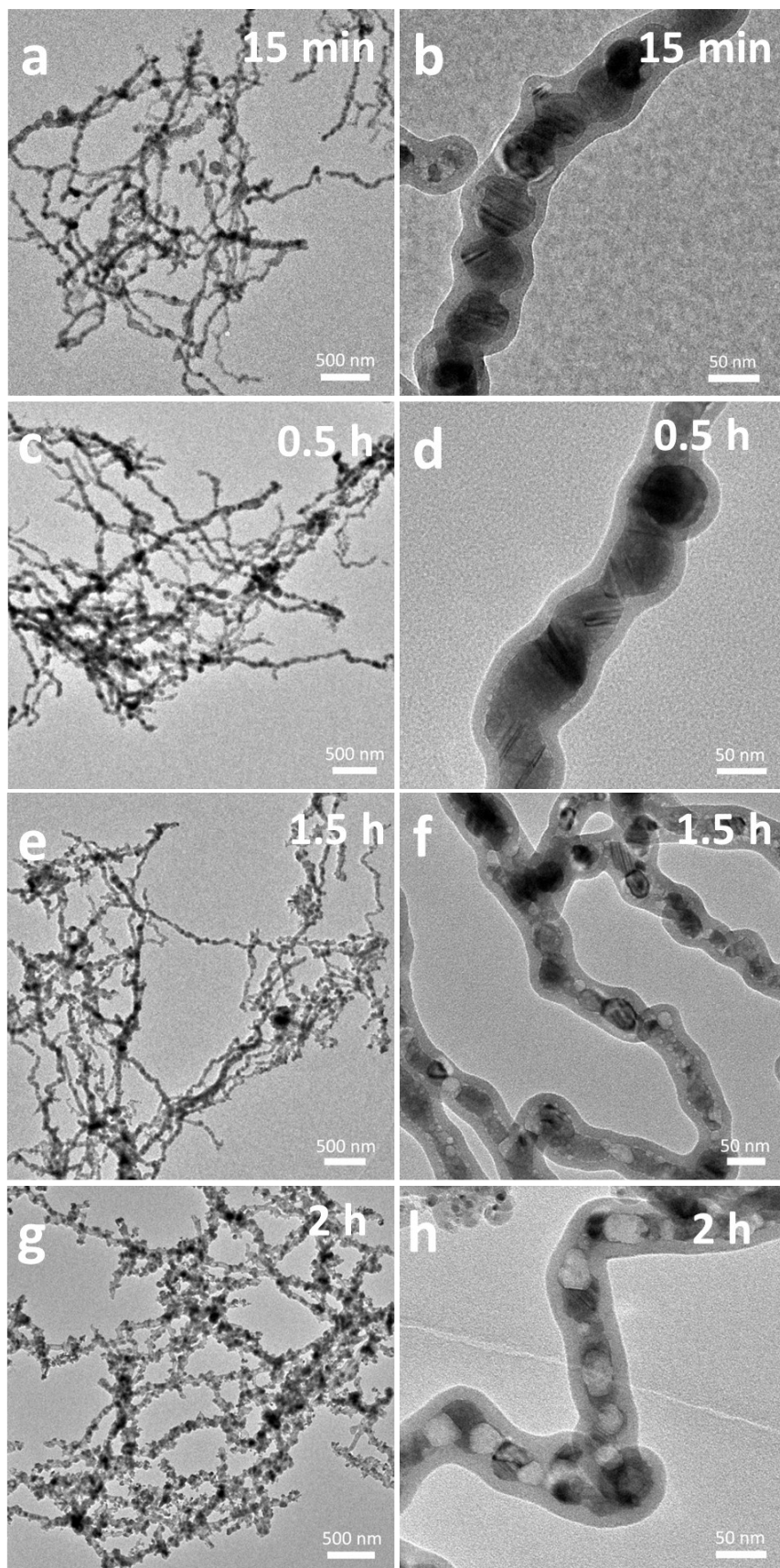
**Figure S1.** (a) SEM image and (b) TEM image of Co nanochains.



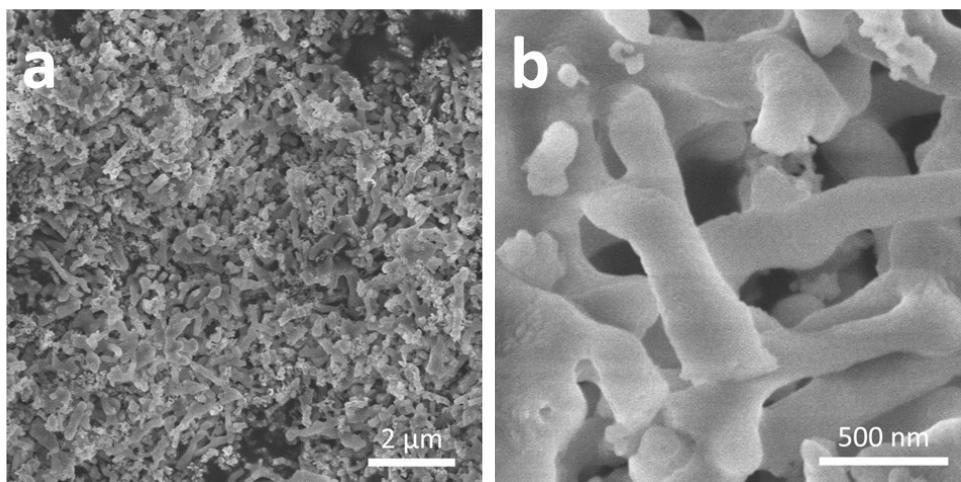
**Figure S2.** XRD patterns of Co@CoP<sub>x</sub> nanochains prepared under different phosphorization time.



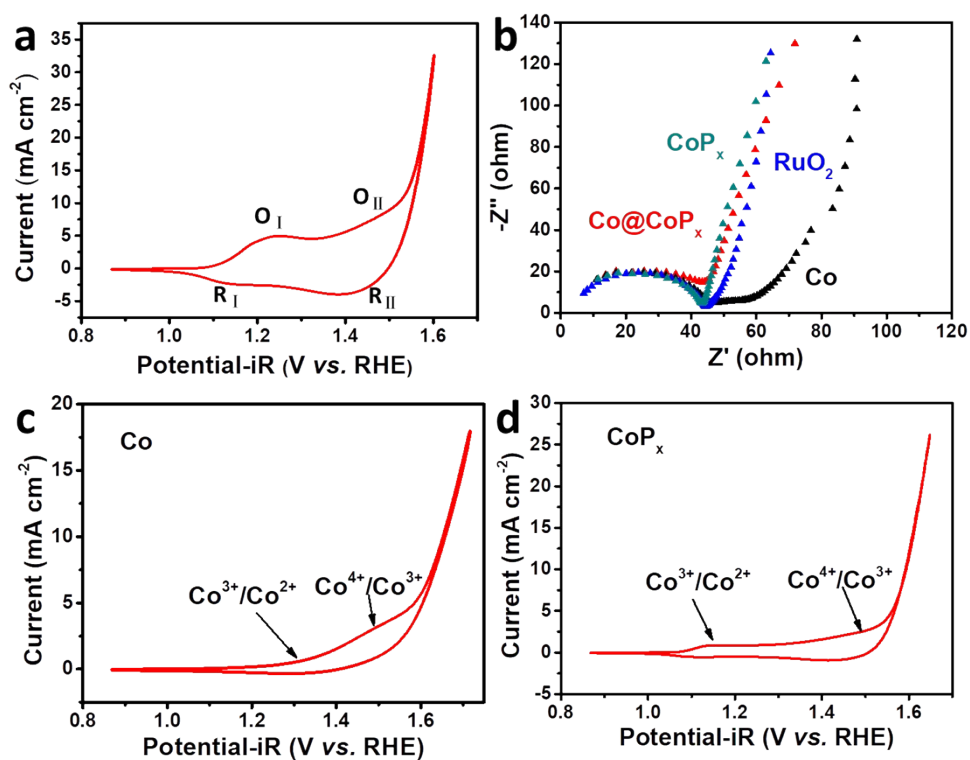
**Figure S3.** SEM images of Co@CoP<sub>x</sub> nanochains prepared under different phosphorization time (a, b) 15 min, (c, d) 0.5 h, (e, f) 1.5 h, (g, h) 2 h.



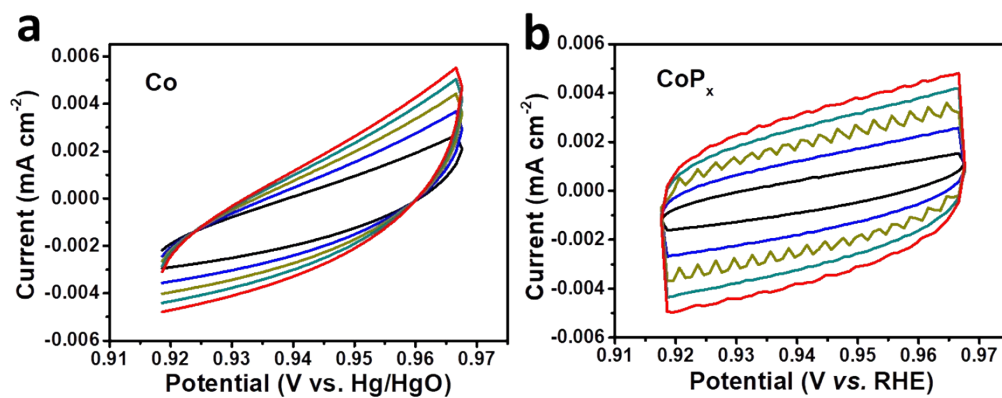
**Figure S4.** TEM images of Co@CoP<sub>x</sub> nanochains prepared under different phosphorization time (a, b) 15 min, (c, d) 0.5 h, (e, f) 1.5 h, (g, h) 2 h.



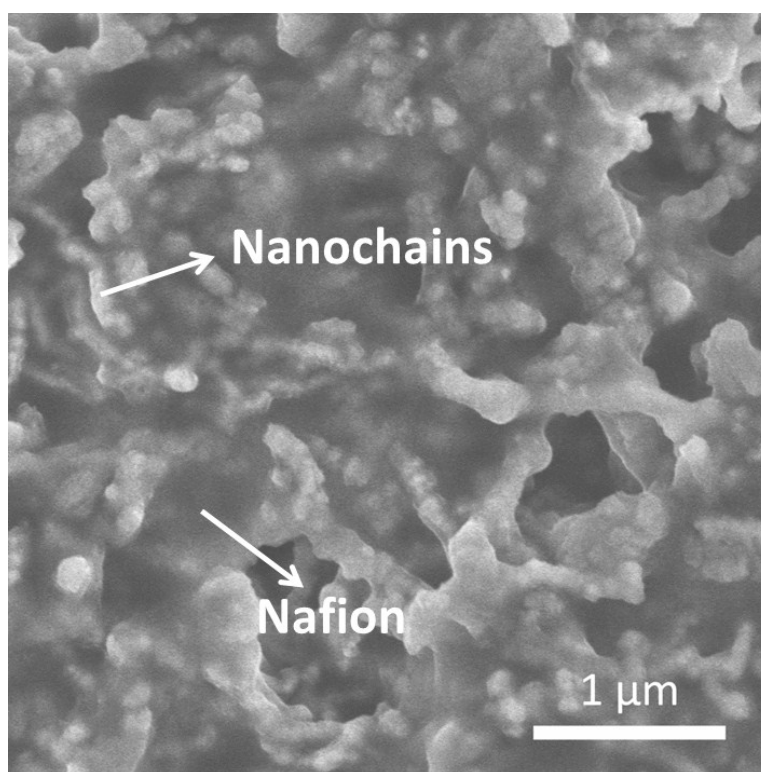
**Figure S5.** SEM images of  $\text{CoP}_x$ .



**Figure S6.** (a) Cyclic voltammety curves of  $\text{Co@CoP}_x$  nanochains; (b) Electrochemical impedance spectroscopy (EIS) of  $\text{Co}$ ,  $\text{CoP}_x$ ,  $\text{Co@CoP}_x$  and  $\text{RuO}_2$ . (c) Cyclic voltammety curves of  $\text{Co}$ ; (d) Cyclic voltammety curves of  $\text{CoP}_x$ .



**Figure S7.** Cyclic voltammetry curves of (a) Co nanochains and (b)  $\text{CoP}_x$  in the region where no redox reaction occurs, the scan rate are 10, 8, 6, 4, 2  $\text{mV s}^{-1}$ .



**Figure S8.** SEM image of  $\text{Co@CoP}_x$  after stability test.

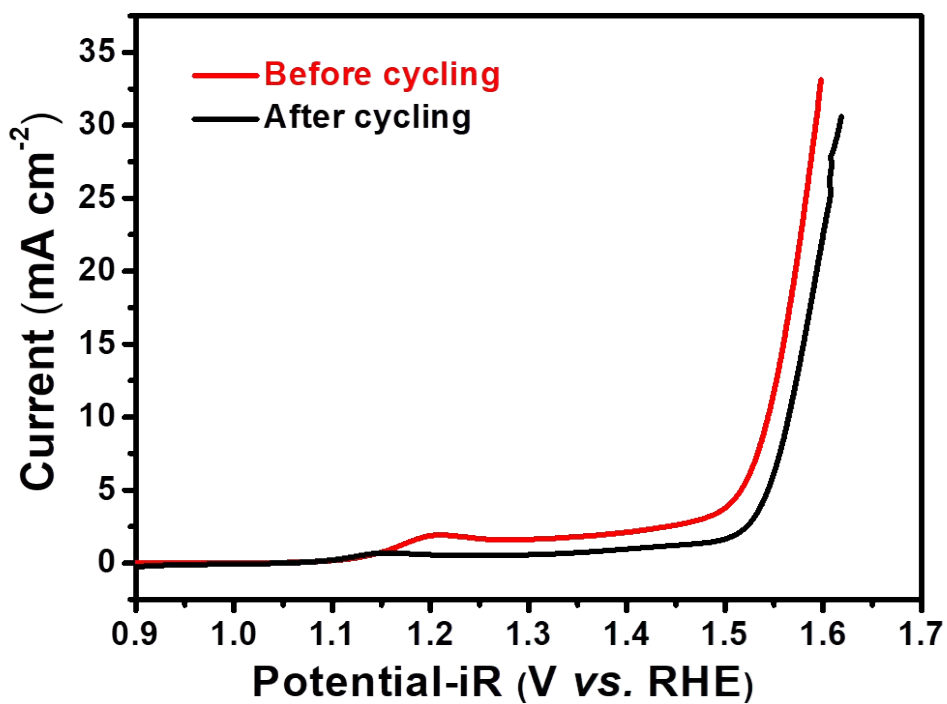


Figure S9. Polarization curves of Co@CoP<sub>x</sub> before and after stability test.

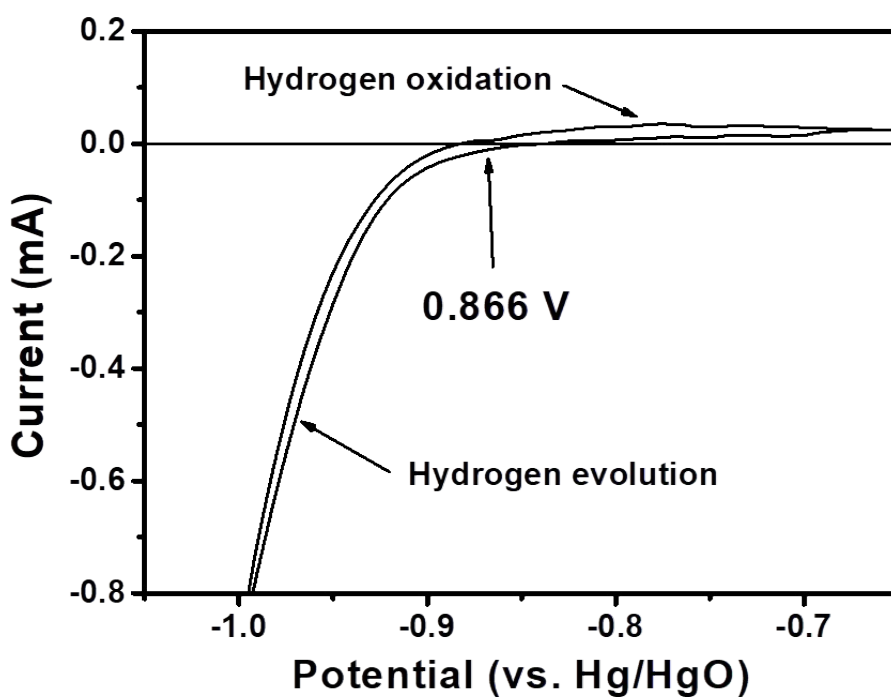


Figure S10. CV curve measured in H<sub>2</sub> saturated 0.1 M KOH for RHE calibration.