## Ionic liquid-supported 3DOM Silica for efficient heterogeneous oxidative desulfurization

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Fig. S1 EDS elemental mapping images of the IL-3DOM  $SiO_2$  catalyst.

| T 11 C1   | $\alpha$ $(1)$ | • 1 .•    | 1 10    | • ,•       | C           | C (1     |           |
|-----------|----------------|-----------|---------|------------|-------------|----------|-----------|
| Table NT  | ( atalytic     | oxidative | desulti | irization  | nertormance | of other | fungsten- |
| Tuble D1. | Cuturytic      | OMuulive  | uosuin  | ai iZution | periormanee | or other | tungsten  |

| Entry | Catalyst   | O/S molar | t/min | Sulfur   | Recycle | Ref.      |
|-------|--|-----------|-------|----------|---------|-----------|
|       |  | ratio     |       | removal/ | times   |           |
|       |  |           |       | %        |         |           |
| 1     | C <sub>4</sub> -IL@OMS   | 3         | 60    | 99.5     | 7       | 1         |
| 2     | Sw20-3.0   | 4         | 15    | 97.1     | 5       | 2         |
| 3     | 25wt% WO <sub>3</sub> /SBA-15  | 10        | 20    | 99       | 5       | 3         |
| 4     | 15% HPW-SPC  | 3         | 120   | 96.3     | 4       | 4         |
| 5     | PW <sup>-</sup> -H <sub>3</sub> N <sup>+</sup> -SBA-15                       | 62        | 120   | 100      | /       | 5         |
| 6     | 0.2W-SiO <sub>2</sub>  | 3         | 30    | 100      | 9       | 6         |
| 7     | [Eu(PW <sub>11</sub> O <sub>39</sub> ) <sub>2</sub> ] <sup>11-</sup> @SBA-15 | 12        | 120   | 92       | 10      | 7         |
| 8     | 0.25W-CeO <sub>2</sub> -400  | 5         | 50    | 99.2     | 3       | 8         |
| 9     | 550-WO <sub>3</sub> -SiO <sub>2</sub>  | 3         | 70    | 100      | 6       | 9         |
| 10    | HPW/SiO <sub>2</sub> -Al <sub>2</sub> O <sub>3</sub> (50)                    | 2         | 120   | 97       | 3       | 10        |
| 11    | LaW10/IL-SiO2  | 5         | 25    | >99      | 10      | 11        |
| 12    | HPW-TiO <sub>2</sub> -SiO <sub>2</sub> $(1:3)$                               | 12        | 120   | 96       | 3       | 12        |
| 13    | IL-3DOM SiO <sub>2</sub>   | 3         | 40    | 100      | 17      | This work |

containing mesoporous silica systems reported.

- 1. M. Zhang, M. Li, Q. Chen, W. Zhu, H. Li, S. Yin, Y. Li and H. Li, *RSC Adv.*, 2015, 5, 76048-76056.
- 2. D. Shen, Y. Dai, J. Han, L. Gan, J. Liu and M. Long, *Chem. Eng. J.*, 2018, **332**, 563-571.
- 3. J. González, J. A. Wang, L. F. Chen, M. E. Manríquez and J. M. Dominguez, *J. Phys. Chem. C*, 2017, **121**, 23988-23999.
- 4. B. Li, Z. Liu, C. Han, W. Ma and S. Zhao, J. Colloid Interface Sci., 2012, 377, 334-341.
- 5. X. N. Pham, D. L. Tran, T. D. Pham, Q. M. Nguyen, V. T. T. Thi and H. D. Van, *Adv. Powder Technol.*, 2018, **29**, 58-65.
- W. Zhu, Q. Gu, J. Hu, P. Wu, S. Yin, F. Zhu, M. Zhang, J. Xiong and H. Li, *J. Porous Mater.*, 2015, 22, 1227-1233.
- S. O. Ribeiro, L. S. Nogueira, S. Gago, P. L. Almeida, M. C. Corvo, B. d. Castro, C. M. Granadeiro and S. S. Balula, *Appl. Catal.*, *A*, 2017, 542, 359-367.
- Y. Li, M. Zhang, W. Zhu, M. Li, J. Xiong, Q. Zhang, Y. Wei and H. Li, *RSC Adv.*, 2016, 6, 68922-68928.
- S. Xun, W. Zhu, F. Zhu, Y. Chang, D. Zheng, Y. Qin, M. Zhang, W. Jiang and H. Li, *Chem. Eng. J.*, 2015, 280, 256-264.
- 10. X.M. Yan, Z. Mei, P. Mei and Q. Yang, J. Porous Mater., 2014, 21, 729-737.
- 11. Y. Chen and Y.F. Song, *ChemPlusChem*, 2014, **79**, 304-309.
- 12. X.M. Yan, P. Mei, L. Xiong, L. Gao, Q. Yang and L. Gong, *Catal. Sci. Technol.*, 2013, **3**, 1985.

The ionic liquid that prepared via ion exchange method. The content of C, H, N

element in the precursor is determined through elemental analyzer. The results and the

theoretical value are shown as follows (Table S2).

Table S2. Elemental composition of the  $[C_{16}mim]_6H_2W_{12}O_{40}$  in wt%.

| Element             | С     | Н    | N    |
|---------------------|-------|------|------|
| Actual value/%      | 30.15 | 5.08 | 3.33 |
| Theoretical value/% | 30.71 | 5.07 | 3.55 |

Table S3. The theoretical and actual content of the IL in the hybrid materials.

| Entry | Theoretical value (wt%) | Actual value <sup>a</sup> (wt%) |
|-------|-------------------------|---------------------------------|
| 1     | 24.3                    | 17.1                            |

<sup>a</sup> IL content based on the tungsten content measured by ICP-OES.