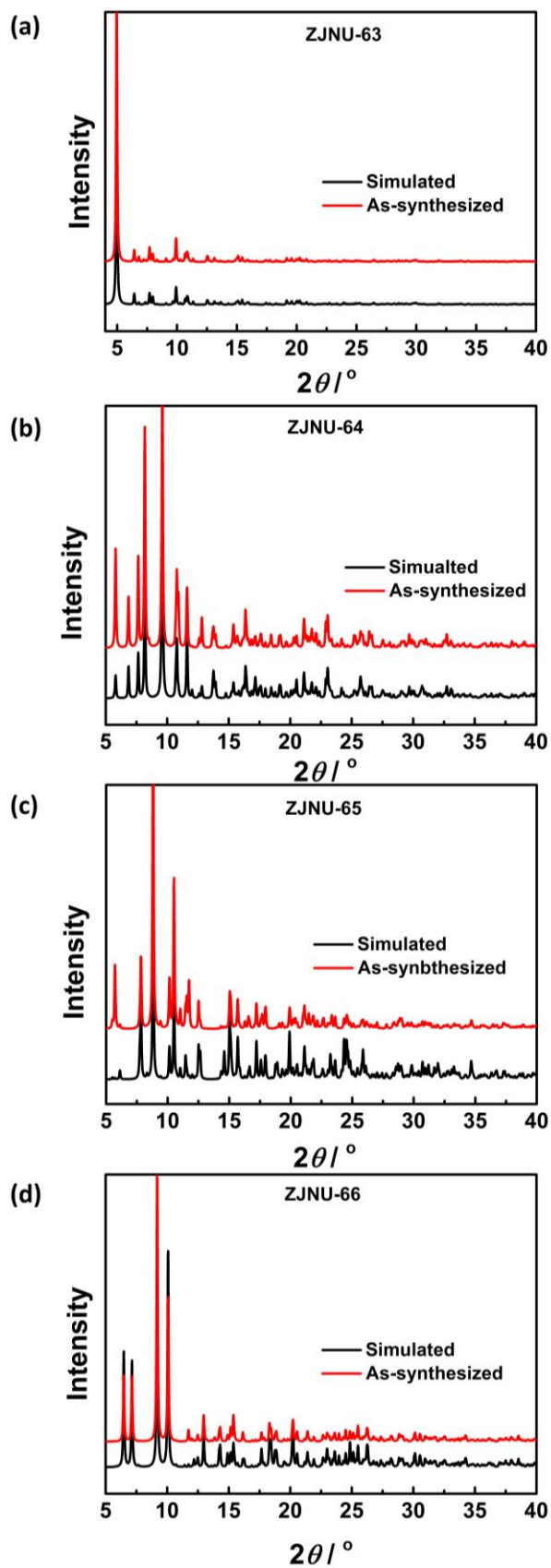


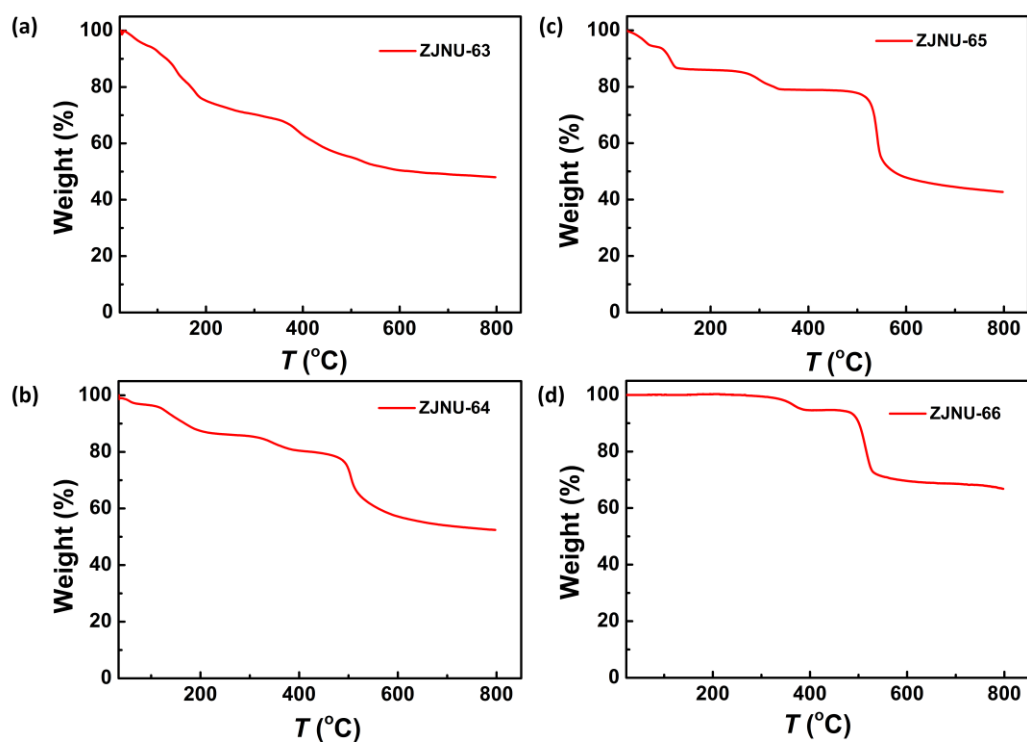
**Structural Diversities and Gas Adsorption Properties of a Family of  
Rod-Packing Lanthanide-Organic Frameworks Based on  
Cyclotriphosphazene-Functionalized Hexacarboxylate Derivatives**

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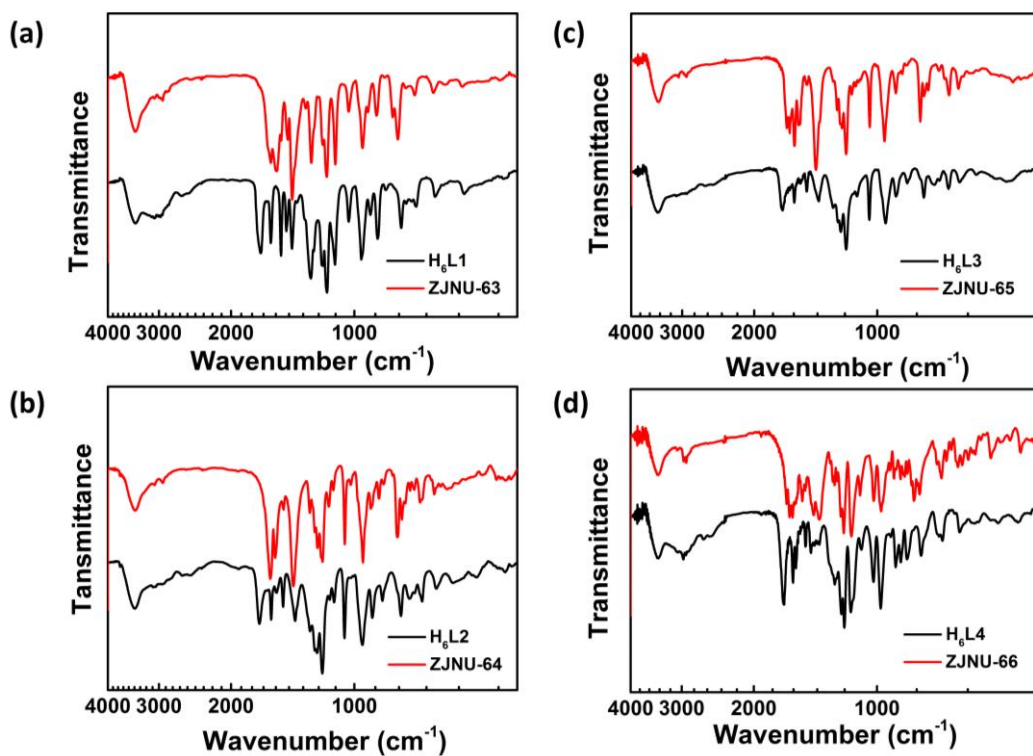
Key Laboratory of the Ministry of Education for Advanced Catalysis Materials,  
College of Chemistry and Life Sciences, Zhejiang Normal University, Jinhua 321004,  
China. E-mail: heyabing@zjnu.cn



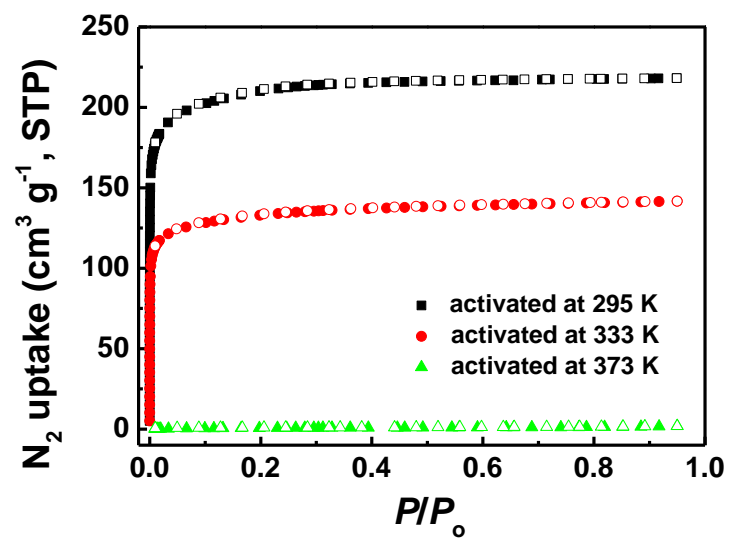
**Fig. S1** The experimental and simulated PXRD patterns for (a) **ZJNU-63**, (b) **ZJNU-64**, (c) **ZJNU-65**, and (d) **ZJNU-66**.



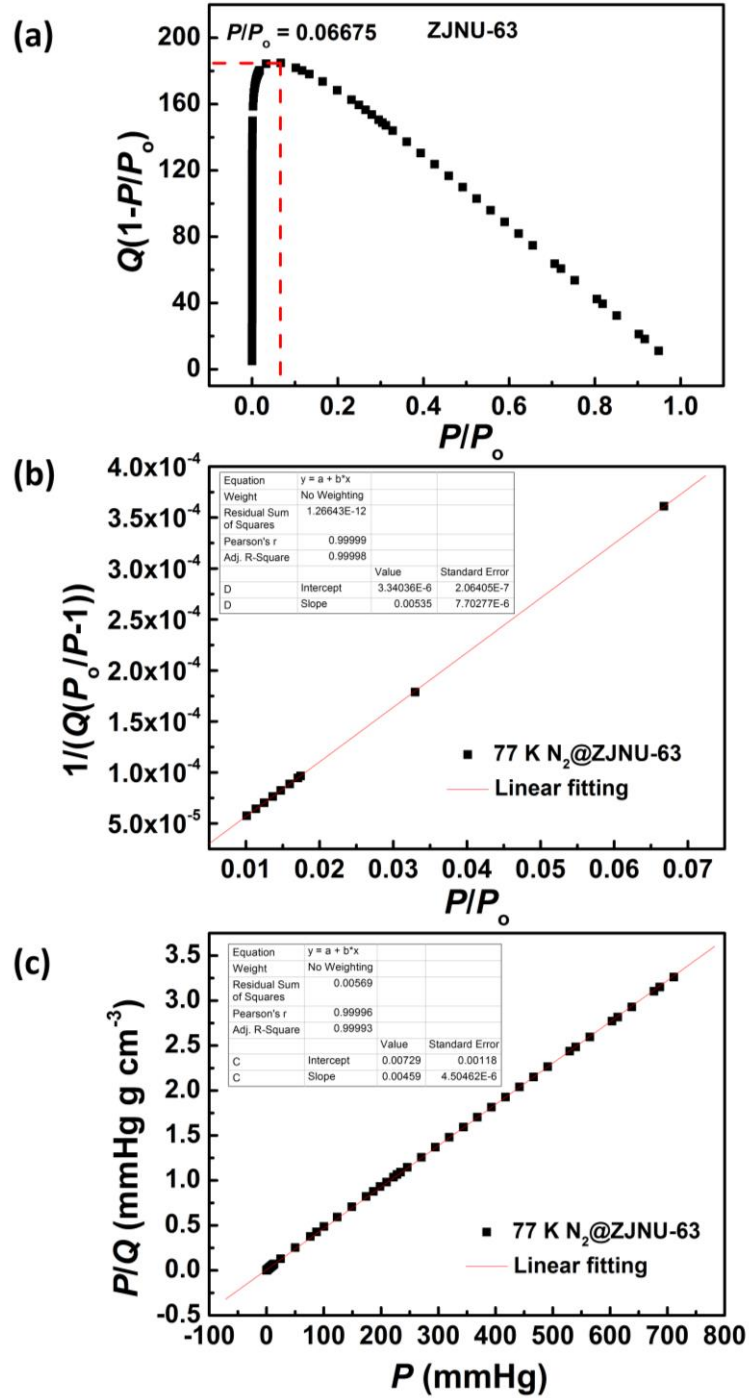
**Fig. S2** TGA curves of the as-synthesized **ZJNU-63**, **ZJNU-64**, **ZJNU-65** and **ZJNU-66** under nitrogen atmosphere.



**Fig. S3** Comparison of FTIR spectra of the organic ligands and their corresponding MOFs.



**Fig. S4** Comparison of  $N_2$  isotherms of ZJNU-63, which was activated at different temperatures. The solid and open symbols represent adsorption and desorption, respectively.



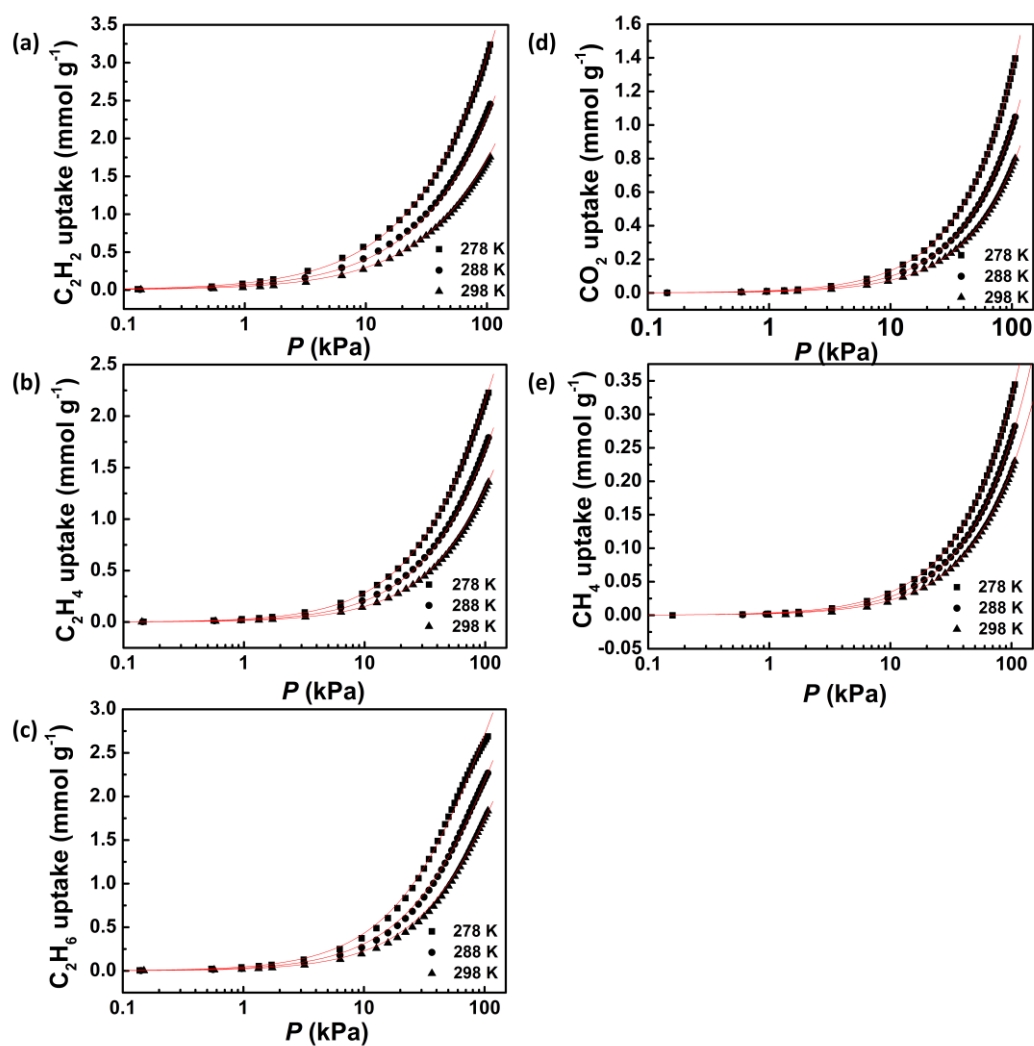
$$S_{\text{BET}} = 1/(3.34036 \times 10^{-6} + 0.00535)/22414 \times 6.023 \times 10^{23} \times 0.162 \times 10^{-18} = 813 \text{ m}^2 \text{ g}^{-1}$$

$$S_{\text{Langmuir}} = (1/0.00459)/22414 \times 6.023 \times 10^{23} \times 0.162 \times 10^{-18} = 948 \text{ m}^2 \text{ g}^{-1}$$

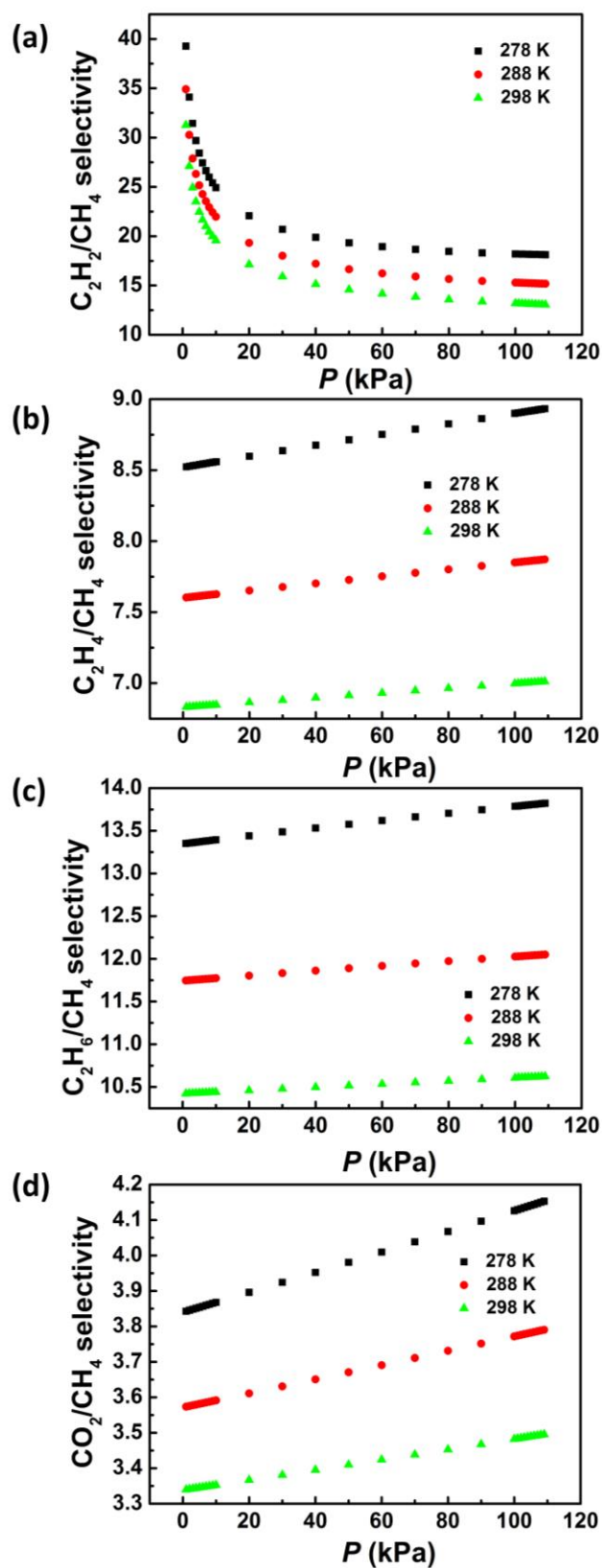
$$\text{BET constant } C = 1 + 0.00535/3.34036 \times 10^{-6} = 1603$$

$$(P/P_o)_{n_m} = \frac{1}{\sqrt{C} + 1} = 0.02437$$

**Fig. S5** The consistency plot (a), BET plot (b) and Langmuir plot (c) for **ZJNU-63**.

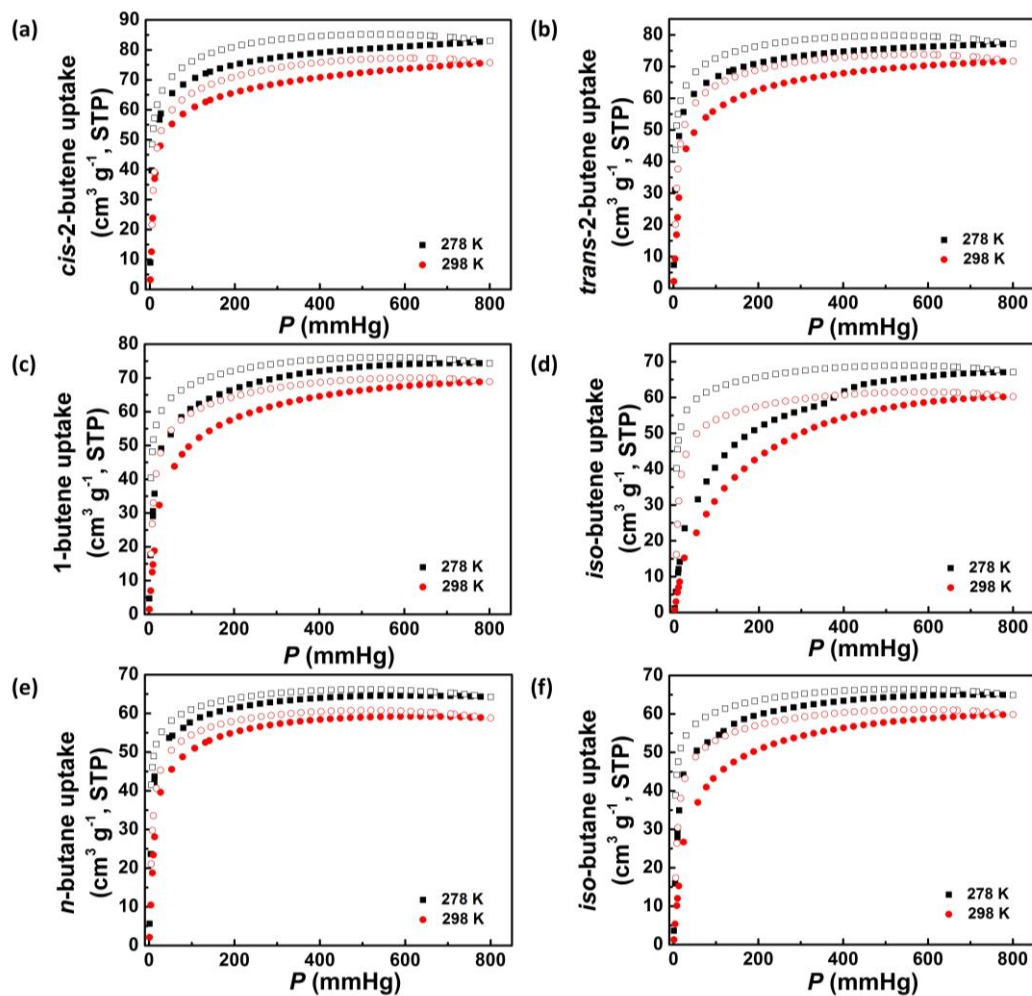


**Fig. S6** Comparison of the pure-component isotherm data for (a)  $C_2H_2$ , (b)  $C_2H_4$ , (c)  $C_2H_6$ , (d)  $CO_2$ , and (e)  $CH_4$  in **ZJNU-63** with the fitted isotherms at 278 K, 288 K and 298 K.

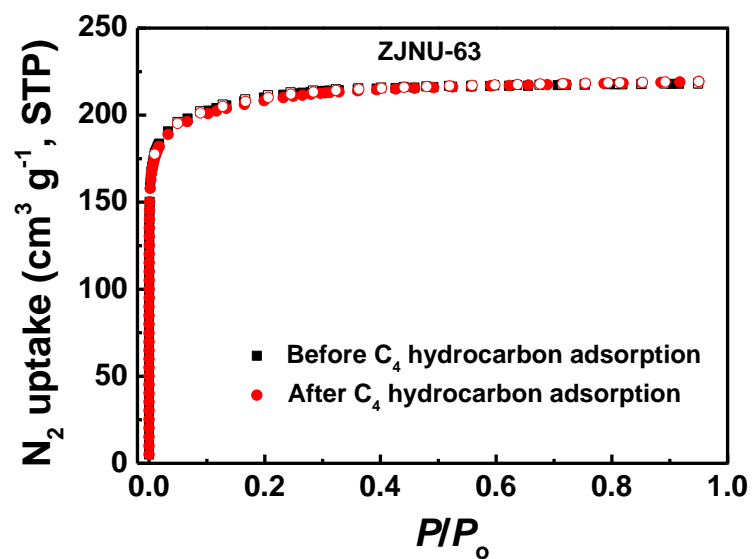


**Fig. S7** IAST selectivities for the equimolar (a)  $C_2H_2/CH_4$ , (b)  $C_2H_4/CH_4$ , (c)  $C_2H_6/CH_4$ , (d)  $CO_2/CH_4$  and (e)  $CO_2/CH_4$  binary gas mixtures in **ZJNU-63** at three different temperatures of 278 K, 288 K, and 298 K.

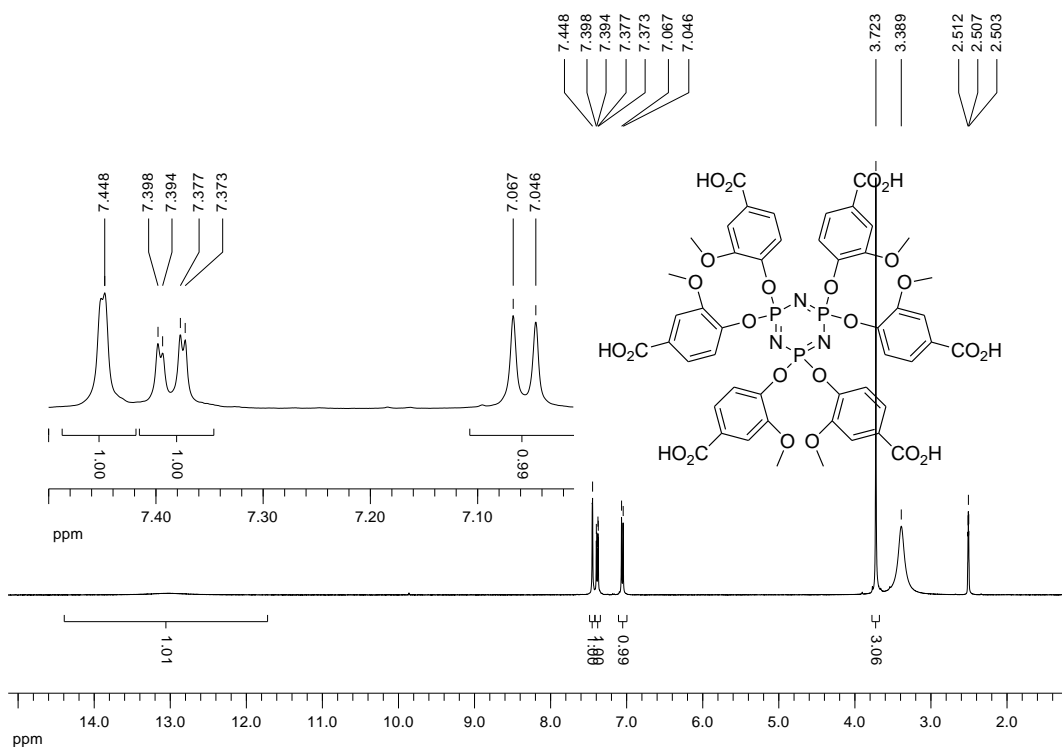
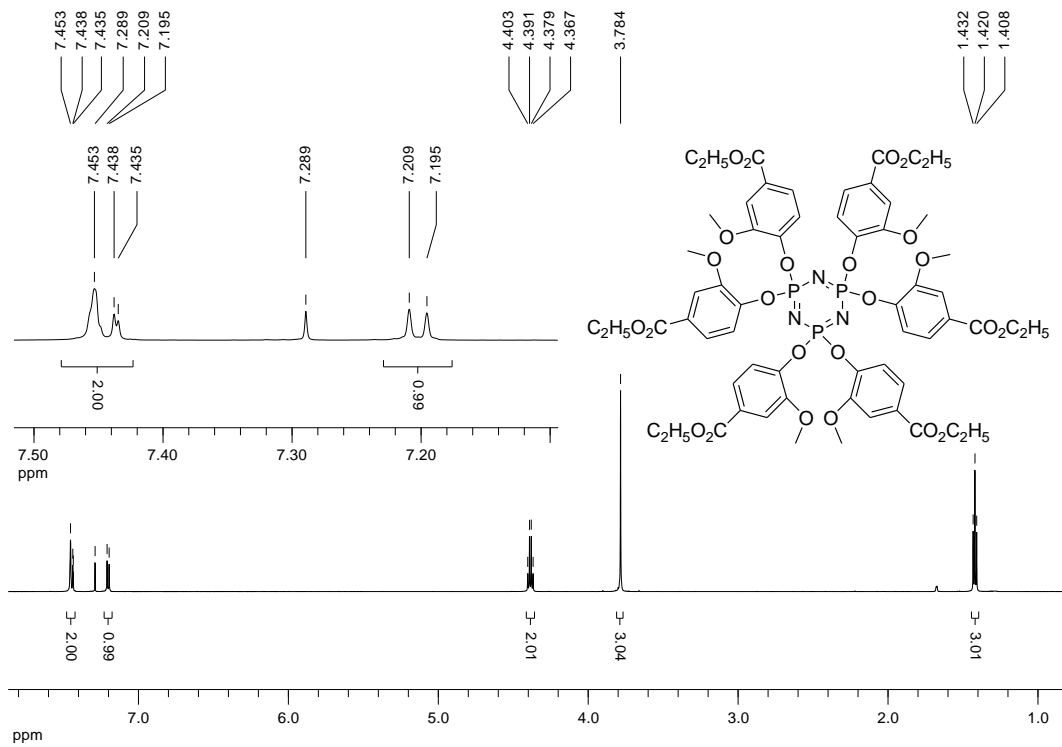


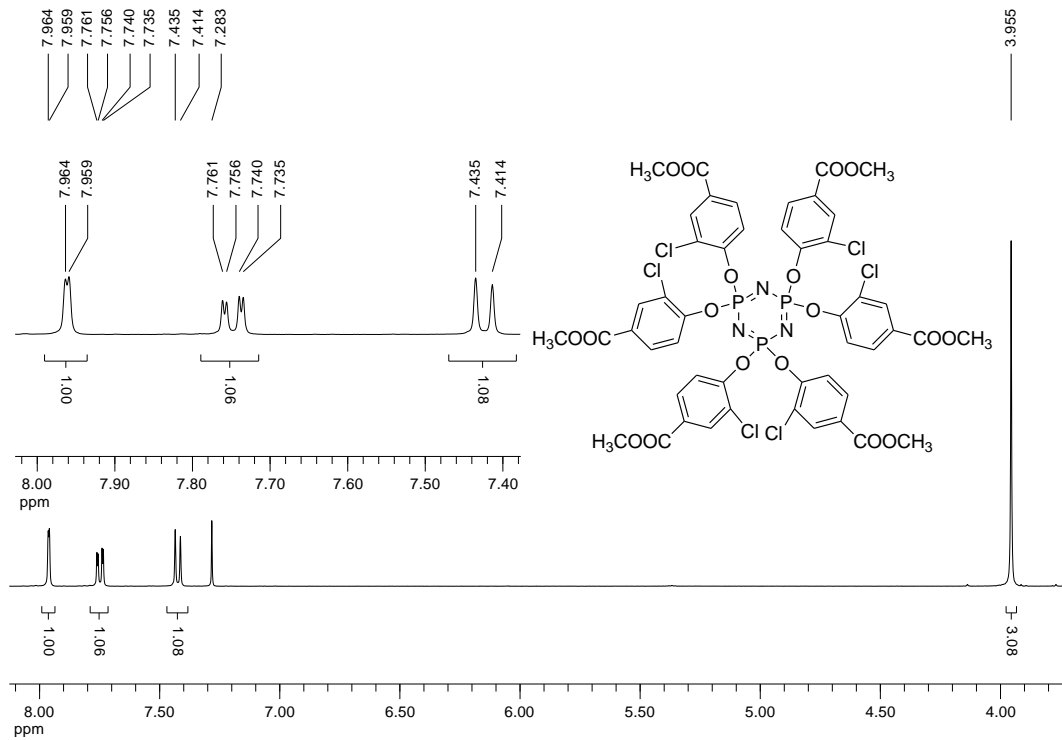
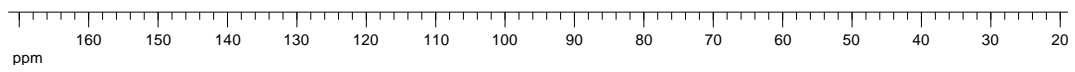
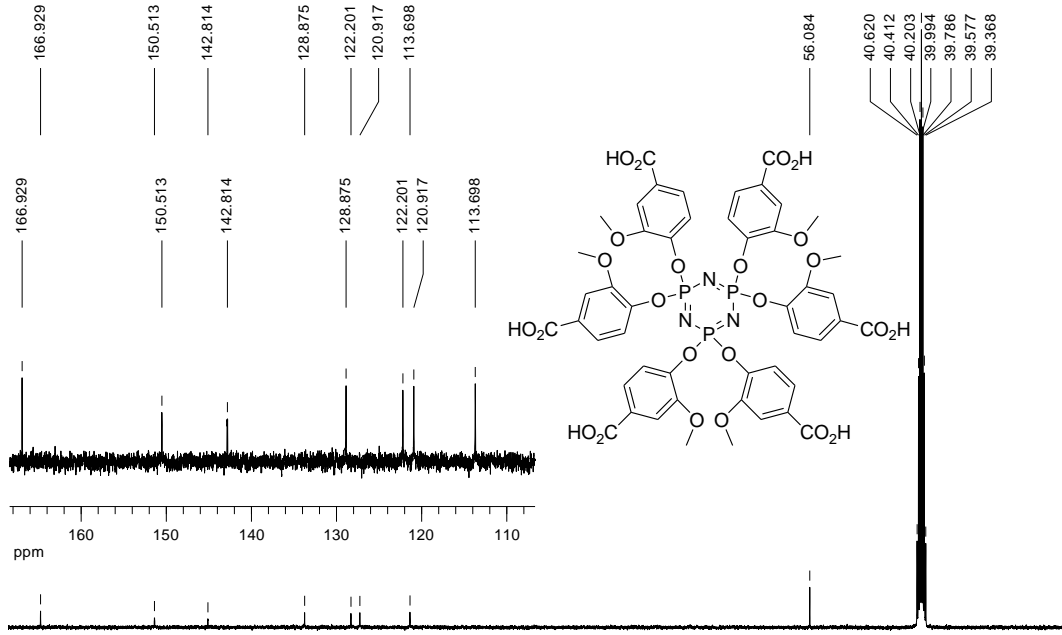


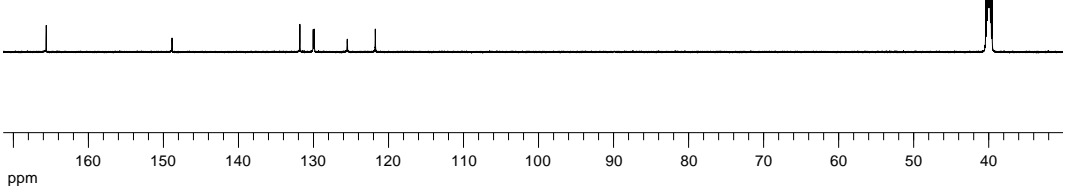
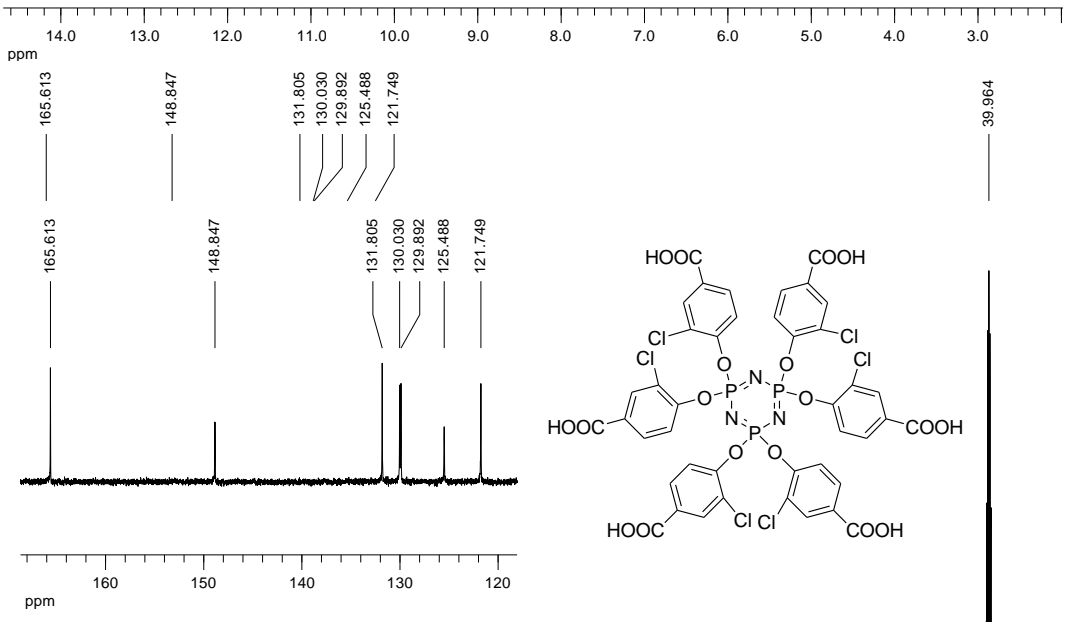
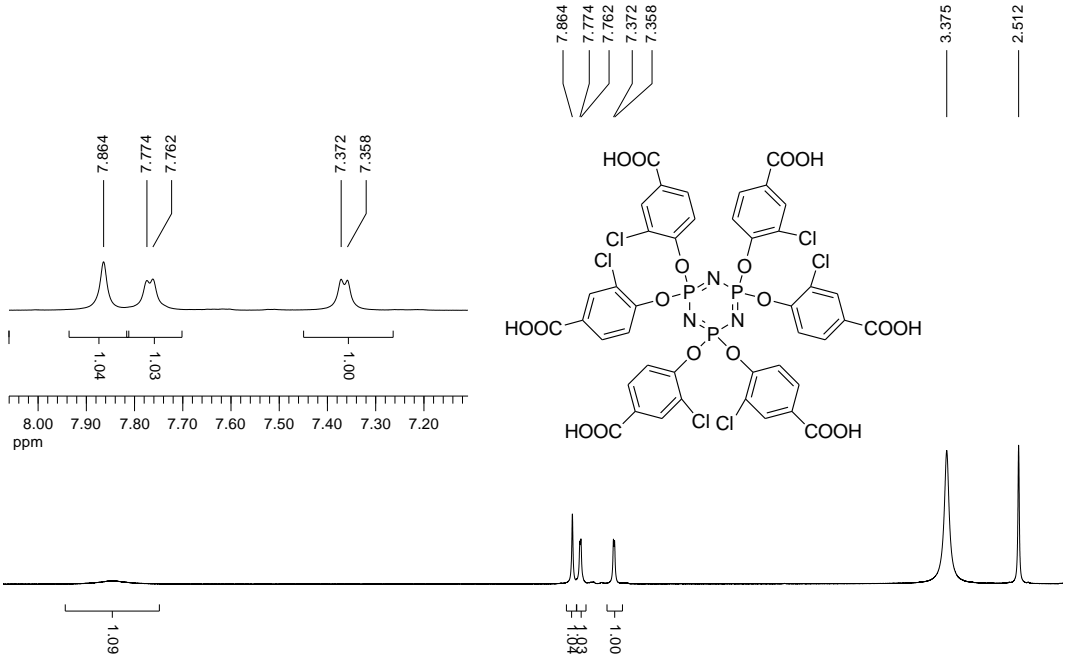
**Fig. S8** (a) *cis*-2-butene, (b) *trans*-2-butene, (c) 1-butene, (d) *iso*-butene, (e) *n*-butane and (f) *iso*-butane isotherms of **ZJNU-63** at two different temperatures of 278 K and 298 K. The solid and open symbols represent adsorption and desorption, respectively. STP = standard temperature and pressure.

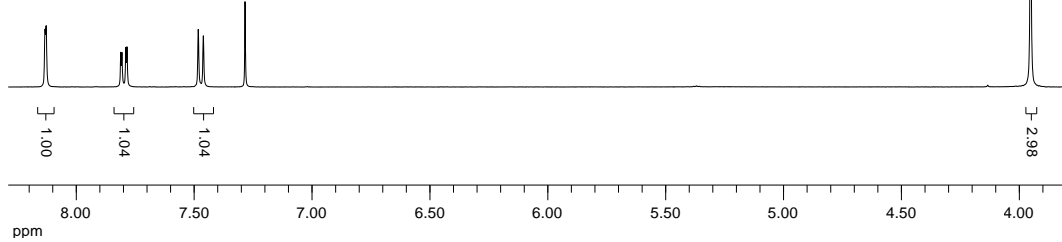
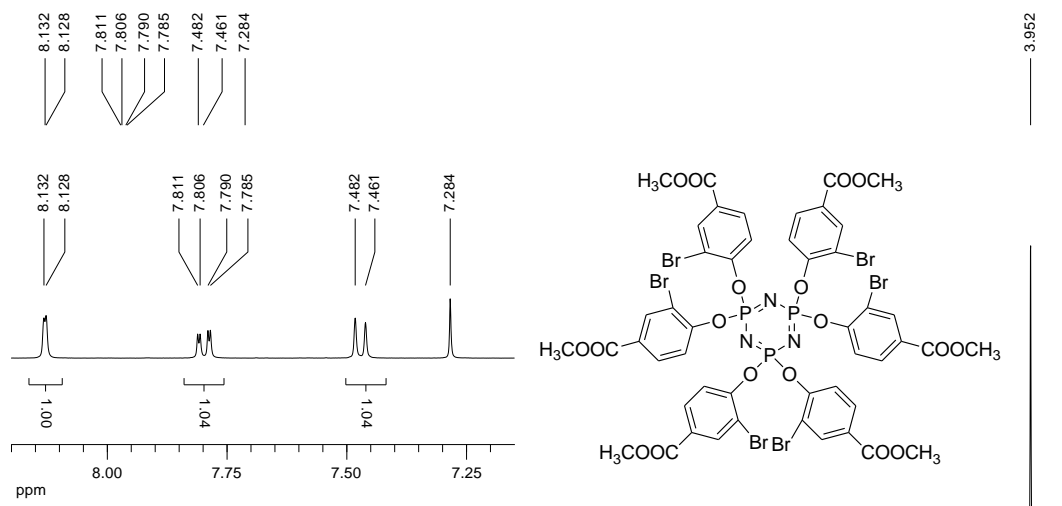
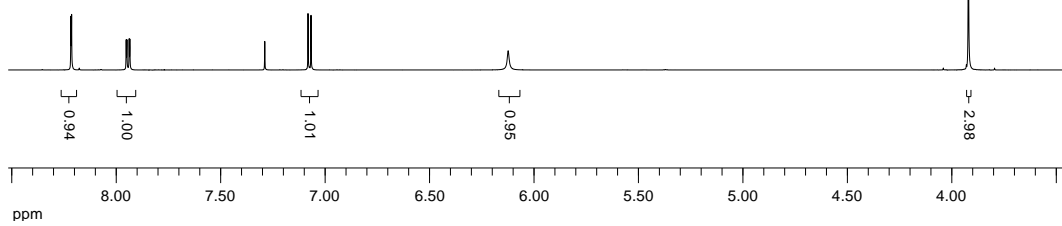
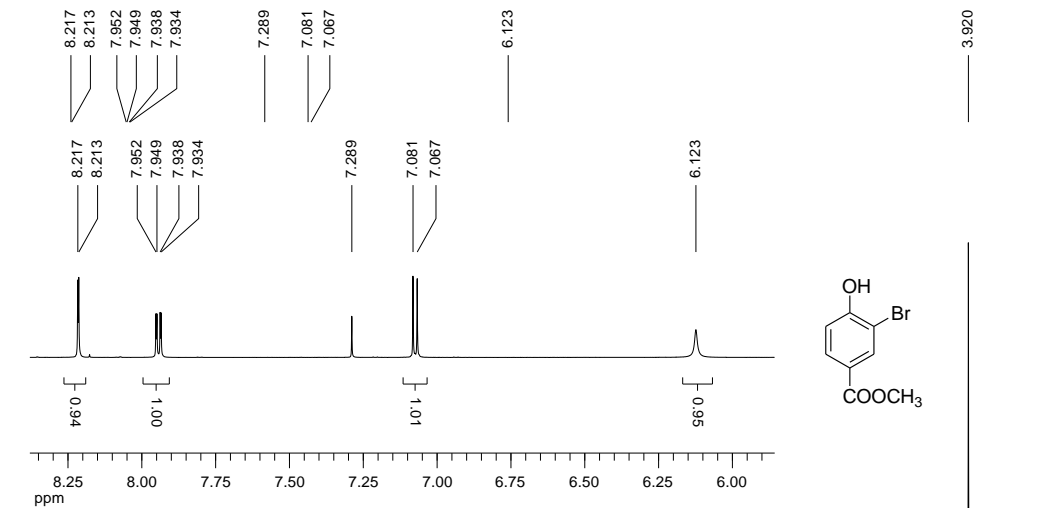


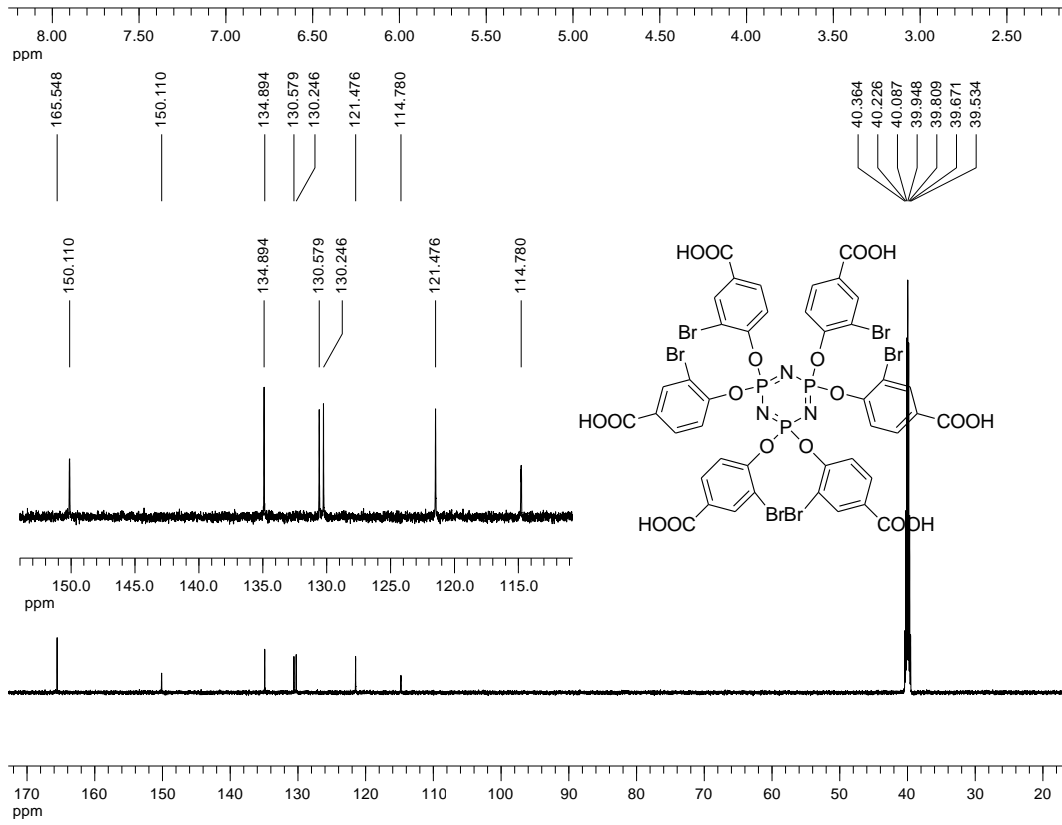
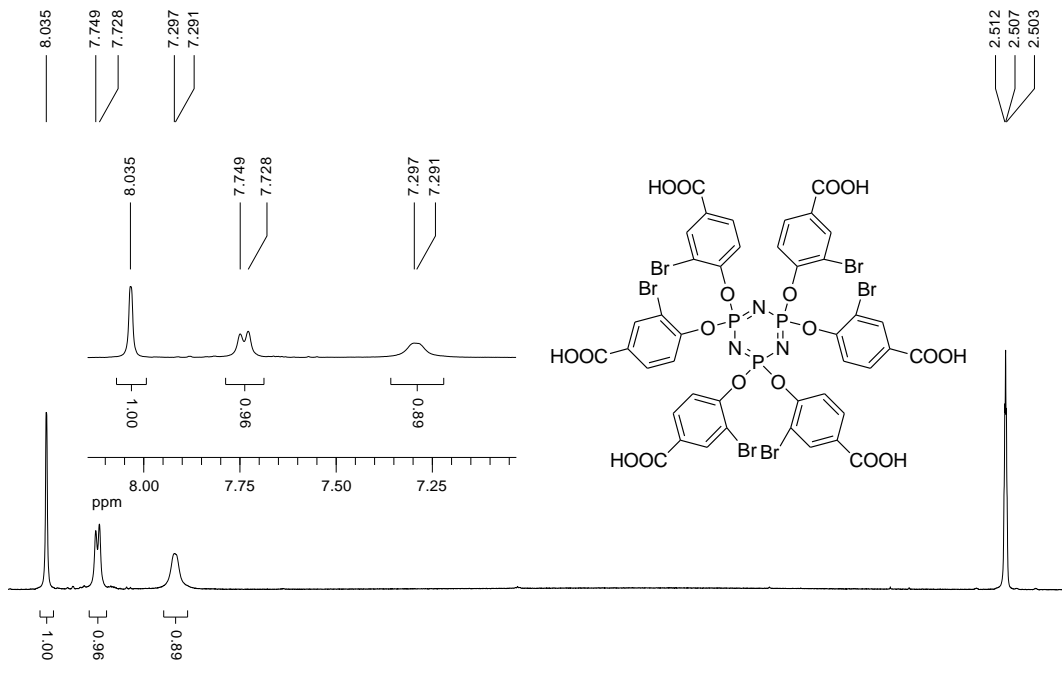
**Fig. S9** N<sub>2</sub> adsorption and desorption isotherms of **ZJNU-63** before and after measurements of all C<sub>4</sub> hydrocarbon isotherms. Solid and open symbols represent adsorption and desorption, respectively. STP = standard temperature and pressure.

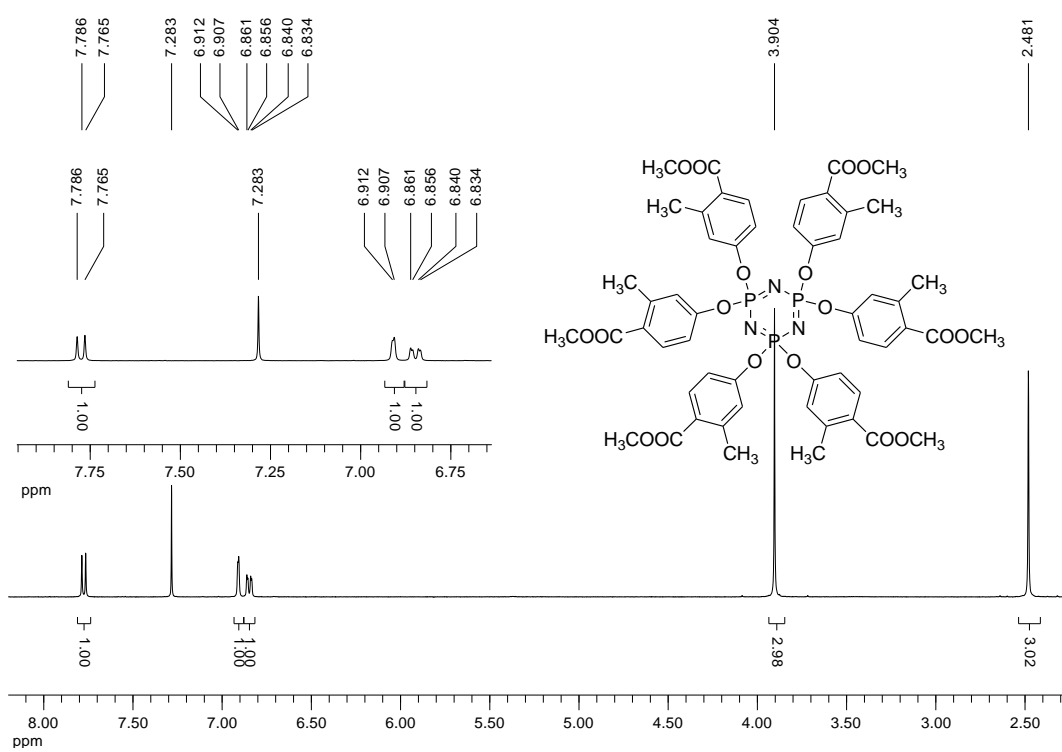
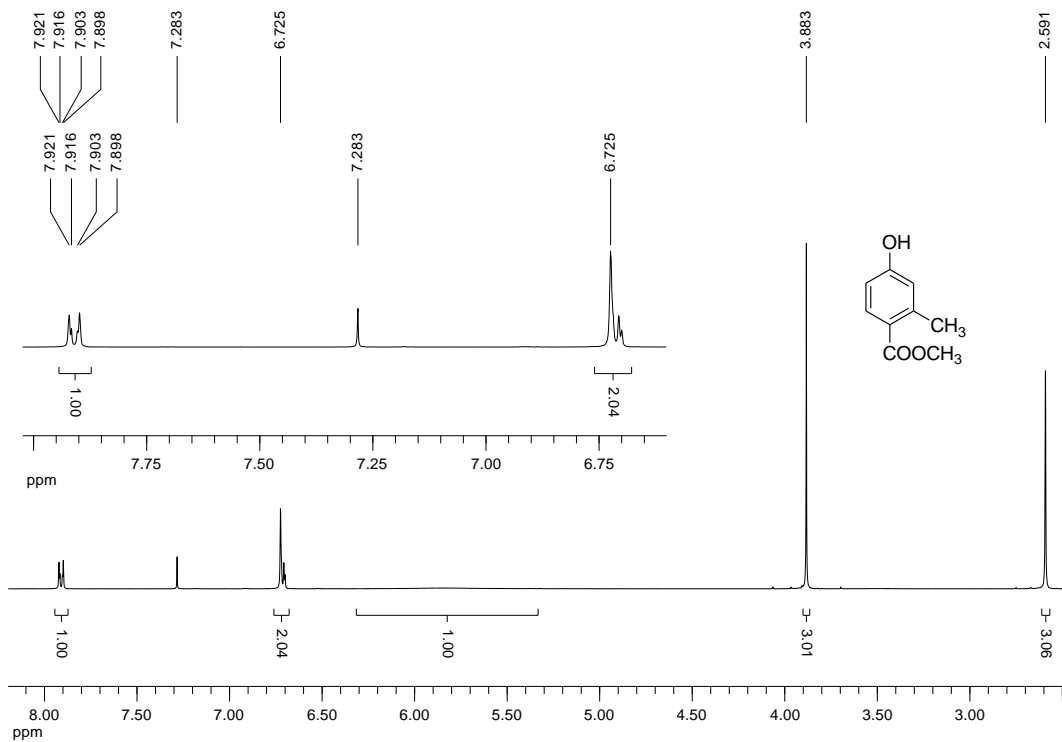














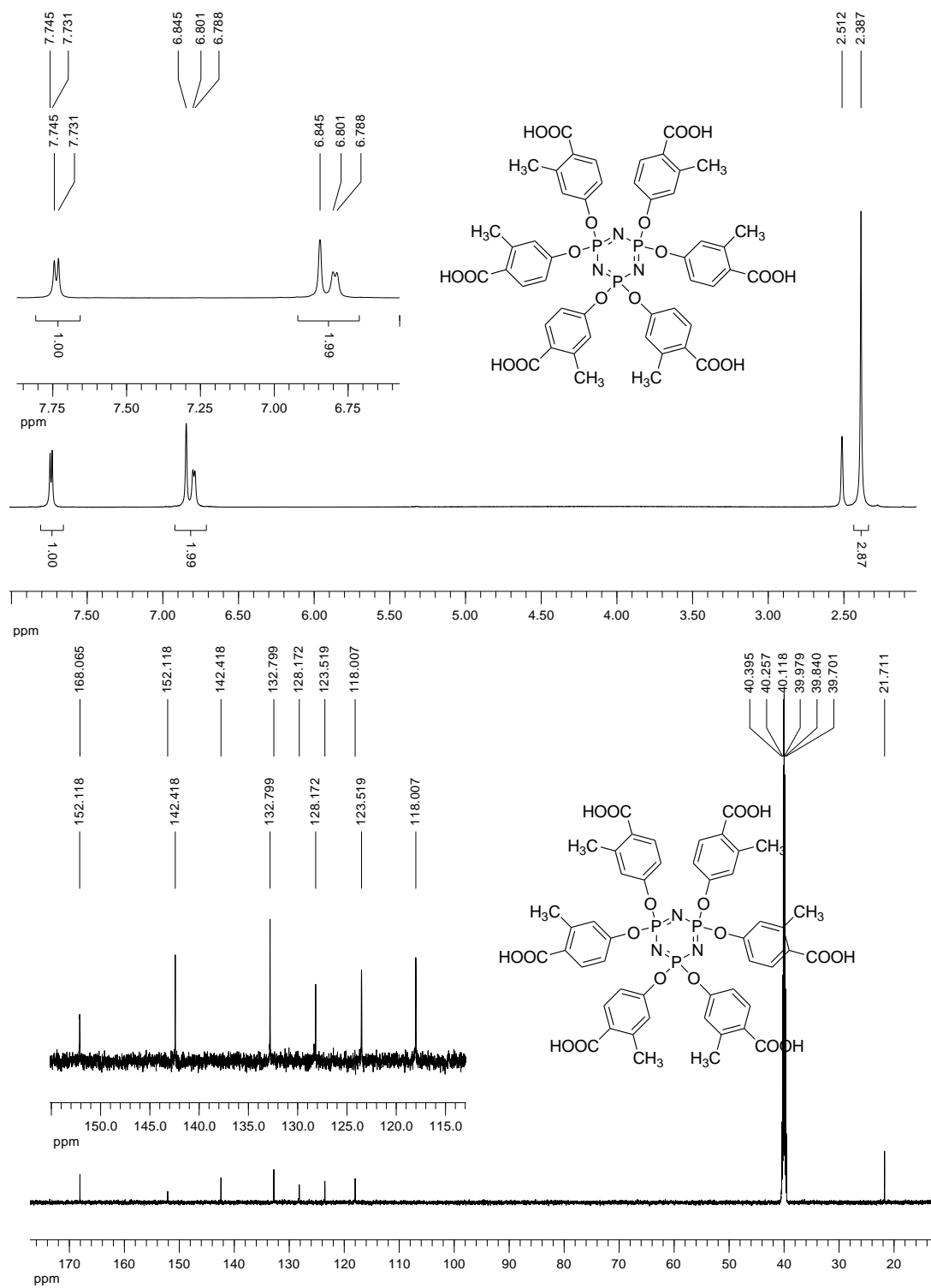


Fig. S8  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra.

**Table S1** Crystal data and structure refinement for **ZJNU-63**, **ZJNU-64**, **ZJNU-65** and **ZJNU-66**.

MOFs	ZJNU-63	ZJNU-64	ZJNU-65	ZJNU-66
Empirical formula	C <sub>49</sub> H <sub>39</sub> Ho <sub>2</sub> N <sub>3</sub> O <sub>27</sub> P <sub>3</sub>	C <sub>42</sub> H <sub>22</sub> Cl <sub>6</sub> Ho <sub>2</sub> N <sub>3</sub> O <sub>20</sub> P <sub>3</sub>	C <sub>48</sub> H <sub>36</sub> Br <sub>6</sub> Ho <sub>2</sub> N <sub>3</sub> O <sub>22</sub> P <sub>3</sub>	C <sub>51</sub> H <sub>43</sub> Ho <sub>2</sub> N <sub>4</sub> O <sub>19</sub> P <sub>3</sub>
Formula weight	1524.60	1524.09	1937.05	1438.66
$\lambda$ (Å)	0.71073	0.71073	0.71073	0.71073
Crystal system	Monoclinic	Orthorhombic	Triclinic	Monoclinic
Space group	<i>P2<sub>1</sub>/c</i>	<i>Pccn</i>	<i>P-1</i>	<i>C2/c</i>
Unit cell dimensions	$a = 14.0524(4)$ Å $b = 24.5135(7)$ Å $c = 26.5196(8)$ Å $\alpha = 90^\circ$ $\beta = 101.943(2)^\circ$ $\gamma = 90^\circ$	$a = 30.427(6)$ Å $b = 18.048(4)$ Å $c = 23.107(4)$ Å $\alpha = 90^\circ$ $\beta = 90^\circ$ $\gamma = 90^\circ$	$a = 12.3748(9)$ Å $b = 17.8829(11)$ Å $c = 18.3320(12)$ Å $\alpha = 105.840(3)^\circ$ $\beta = 106.771(3)^\circ$ $\gamma = 109.139(3)^\circ$	$a = 54.516(6)$ Å $b = 7.8411(8)$ Å $c = 27.411(3)$ Å $\alpha = 90^\circ$ $\beta = 114.783(5)^\circ$ $\gamma = 90^\circ$
$V$ (Å <sup>3</sup> )	8937.6(5)	12689(4)	3353.0(4)	10638(2)
$Z$	4	8	2	8
$D_c$ (g cm <sup>-3</sup> )	1.133	1.596	1.919	1.797
$\mu$ (mm <sup>-1</sup> )	1.869	2.869	6.061	3.124
$F(000)$	2996	5888	1848	5664
Crystal size (mm)	0.180×0.120×0.100	0.110×0.090×0.080	0.23×0.15×0.12	0.200×0.150×0.100
$\theta$ range for data collection (°)	1.143 to 27.617	3.162 to 27.525	1.270 to 27.845	1.492 to 27.614
Limiting indices	$-18 \leq h \leq 18$ $-31 \leq k \leq 31$ $-34 \leq l \leq 34$	$-39 \leq h \leq 38$ $-22 \leq k \leq 23$ $-30 \leq l \leq 30$	$-16 \leq h \leq 16$ $-23 \leq k \leq 23$ $-24 \leq l \leq 23$	$-70 \leq h \leq 70$ $-10 \leq k \leq 10$ $-35 \leq l \leq 35$
Reflections collected / unique	228777 / 20710	86826 / 14560	100728 / 15737	103761 / 12331
$R_{int}$	0.0523	0.0597	0.0995	0.0499
Max. and min. transmission	0.830 and 0.764	0.8030 and 0.7432	0.483 and 0.349	0.732 and 0.576
Refinement method	Full-matrix least-squares on $F^2$	Full-matrix least-squares on $F^2$	Full-matrix least-squares on $F^2$	Full-matrix least-squares on $F^2$
Data/restraints/parameters	20710 / 456 / 733	14560 / 36 / 687	15737 / 0 / 613	12331 / 318 / 725
Goodness-of-fit on $F^2$	1.075	1.150	1.059	1.010
Final $R$ indices [ $I > 2\sigma(I)$ ]	$R_1 = 0.0350$ $wR_2 = 0.1033$	$R_1 = 0.0775$ $wR_2 = 0.2602$	$R_1 = 0.0781$ $wR_2 = 0.2618$	$R_1 = 0.0251$ $wR_2 = 0.0842$
$R$ indices (all data)	$R_1 = 0.0407$ $wR_2 = 0.1058$	$R_1 = 0.1004$ $wR_2 = 0.2794$	$R_1 = 0.1002$ $wR_2 = 0.28384$	$R_1 = 0.0297$ $wR_2 = 0.0943$
Largest diff. peak and hole (e Å <sup>-3</sup> )	3.682 and -2.487	4.476 and -2.502	7.278 and -4.558	1.296 and -1.142
CCDC	1850271	1850269	1850272	1850270

**Table S2** Langmuir-Freundlich parameters for adsorption of C<sub>2</sub>H<sub>2</sub>, C<sub>2</sub>H<sub>4</sub>, C<sub>2</sub>H<sub>6</sub>, CO<sub>2</sub>, and CH<sub>4</sub> in **ZJNU-63**.

Guest	$q_{\text{sat}}$ (mmol g <sup>-1</sup> )	$b_0$ (kPa) <sup>-<math>\nu</math></sup>	$E$ (kJ mol <sup>-1</sup> )	$\nu$
C <sub>2</sub> H <sub>2</sub>	22.10211	2.59039×10 <sup>-7</sup>	22.412	0.79218
C <sub>2</sub> H <sub>4</sub>	7.95226	2.49709×10 <sup>-7</sup>	22.198	1
C <sub>2</sub> H <sub>6</sub>	6.53751	3.17898×10 <sup>-7</sup>	23.129	1
CO <sub>2</sub>	106.97254	2.75326×10 <sup>-8</sup>	19.444	1
CH <sub>4</sub>	5.48745	1.12506×10 <sup>-6</sup>	14.623	1