

† Electronic Supplementary Information

Copper-nickel embedded into nitrogen-doped carbon octahedron as effective bifunctional electrocatalyst

*Lulu Qiao, Anquan Zhu, Huan Yang, Weixuan Zeng, Rui Dong, Pengfei Tan, Donglin Zhong, Quanyin Ma and Jun Pan**

State Key Laboratory for Powder Metallurgy, Central South University, Changsha 410083, P. R. China

**Corresponding Author. Email: jun.pan@csu.edu.cn*

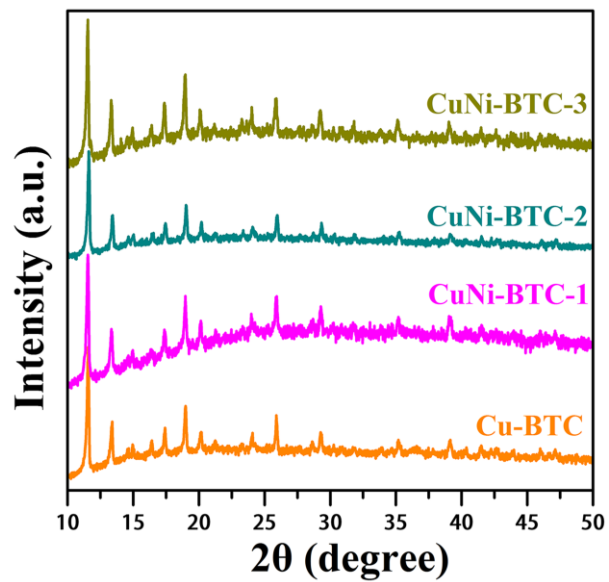


Fig. S1 XRD patterns of as-prepared CuNi-BTC and Cu-BTC precursors.

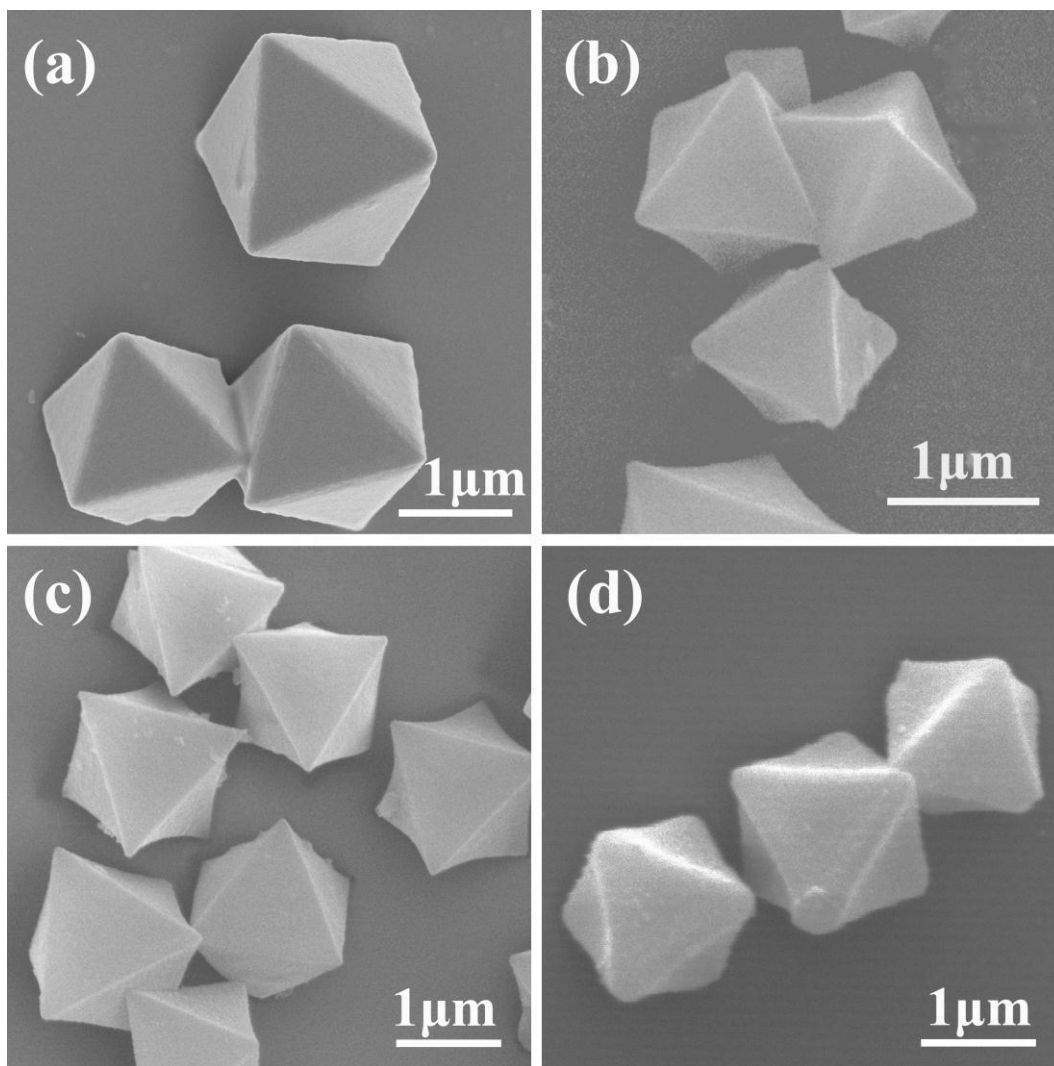


Fig. S2 SEM images of as-prepared precursors. (a) Cu-BTC, (b) CuNi-BTC-1, (c) CuNi-BTC-2, (d) CuNi-BTC-3.

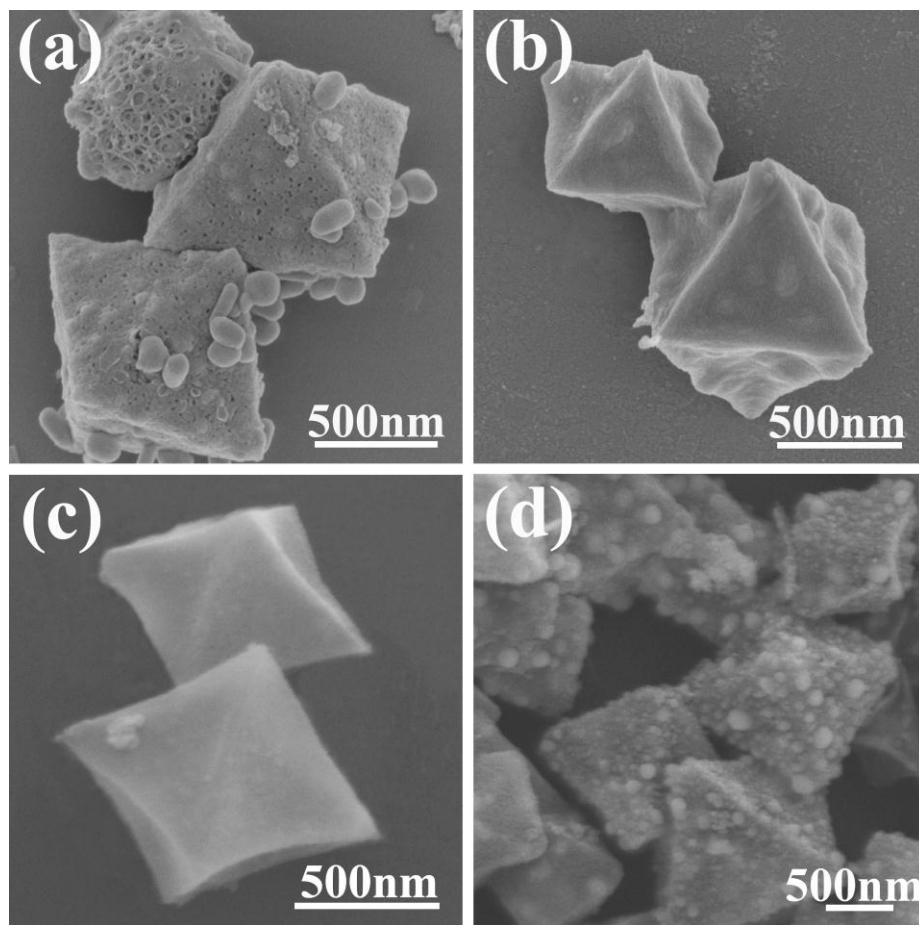


Fig. S3 SEM images of as-synthesized samples. (a) Cu-NC, (b) CuNi-NC-1, (c) CuNi-NC-2, (d) CuNi-NC-3.

Table S1 Summary of element contents of CuNi-NC-2 and Cu-NC.

Samples	Relative content (At. %)			
	C	N	Cu	Ni
Cu-NC	90.97	4.48	4.55	--
CuNi-NC-2	96.97	2.3	0.42	0.32

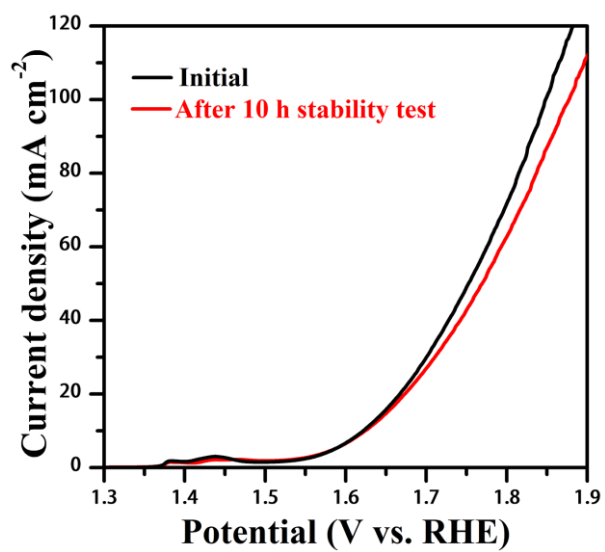


Fig. S4 LSV curves of CuNi-NC-2 sample before and after OER stability test for 10 h.

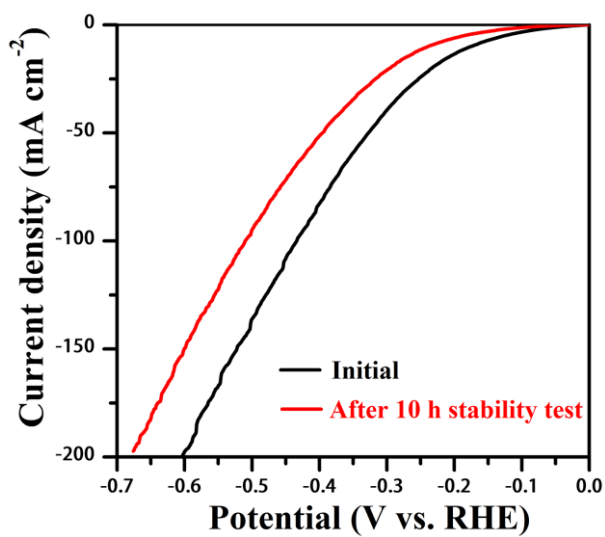


Fig. S5 LSV curves of CuNi-NC-1 sample before and after HER stability test for 10 h.

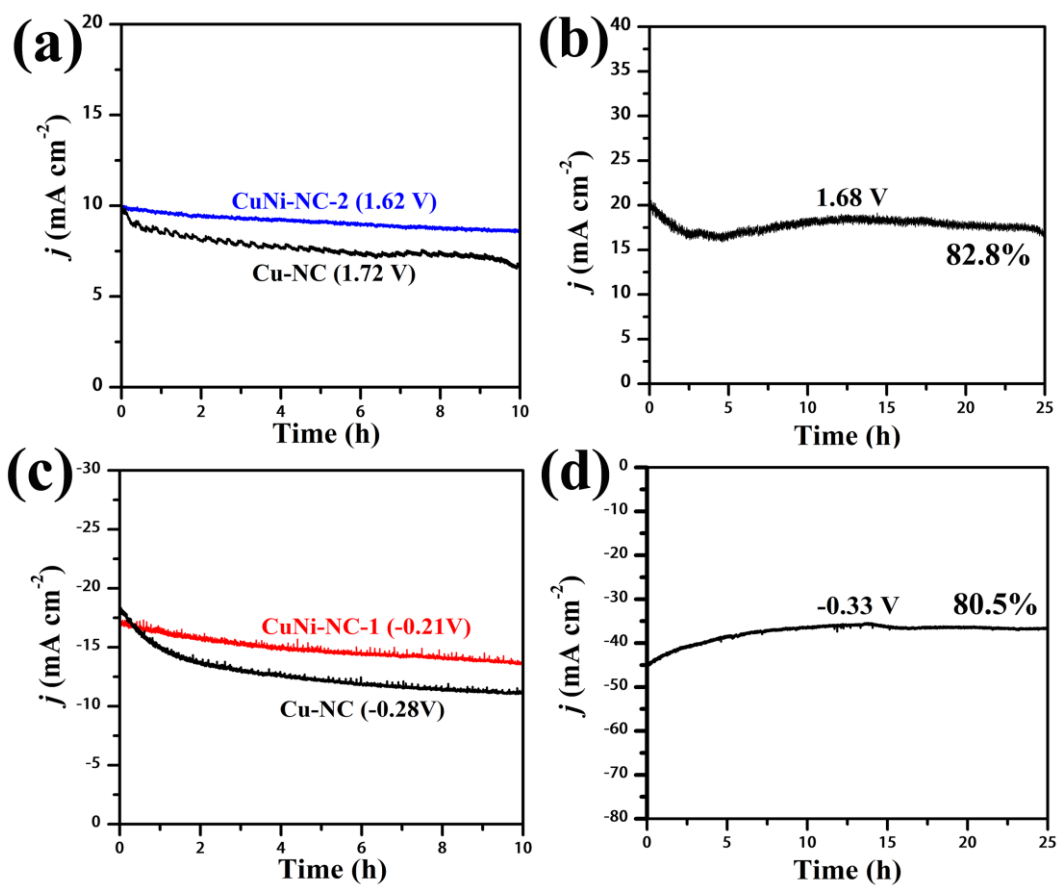


Fig. S6 Chronoamperometry curves of (a) Cu-NC and CuNi-NC-2 in OER process, (b) CuNi-NC-2 sample at a higher potential (1.68 V) for 25 h, (c) Cu-NC and CuNi-NC-1 in HER process, (d) CuNi-NC-1 sample at a higher potential (-0.33 V) for 25 h.

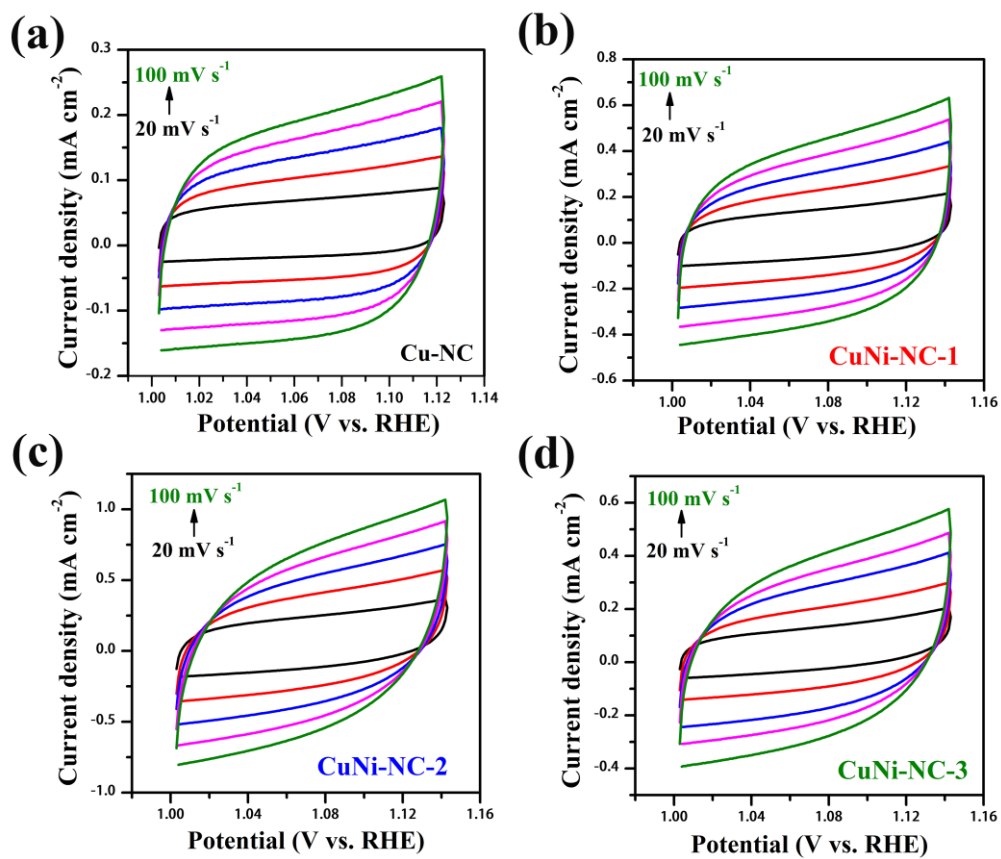


Fig. S7 CV curves of all samples. (a) Cu-NC, (b) CuNi-NC-1, (c) CuNi-NC-2, (d) CuNi-NC-3.

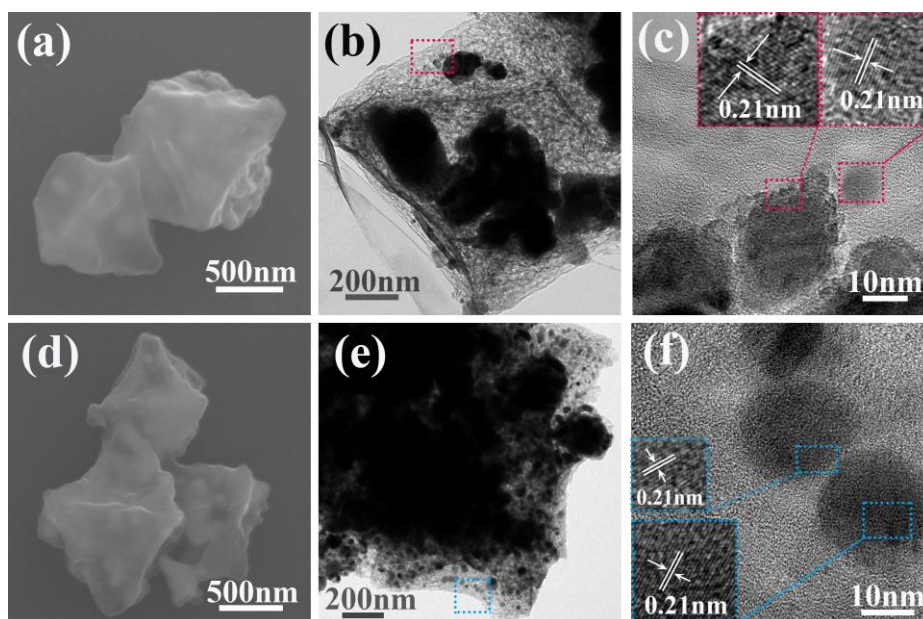


Fig. S8 (a) SEM image, (b) TEM and (c) HRTEM images (inset) of CuNi-NC-2 after 25 h stability test for OER, (d) SEM image, (e) TEM and (f) HRTEM images (inset) of CuNi-NC-1 after 25 h stability test for HER process.

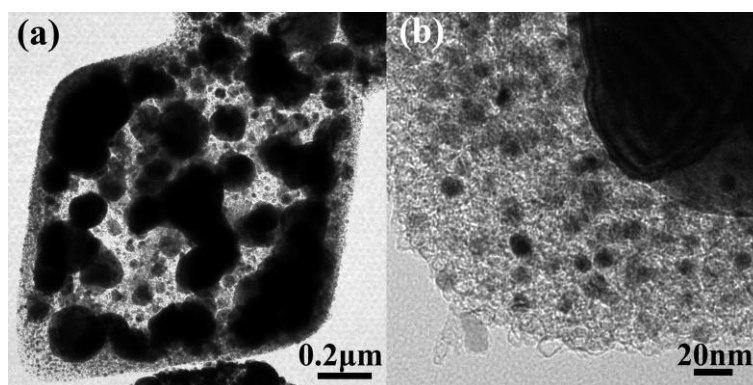


Fig. S9 (a) and (b) TEM images of as-obtained CuNi-NC-1 sample.