

**Table S1** Physicochemical properties of Sn-Beta samples

Sample	Pore volume ( $\text{cm}^3 \text{ g}^{-1}$ )			SSA <sup>c</sup> ( $\text{m}^2 \text{ g}^{-1}$ )	
	$V_{\text{total}}$	$V_{\text{micro}}^{\text{a}}$	$V_{\text{meso}}^{\text{b}}$	$S_{\text{BET}}$	$S_{\text{ext}}^{\text{a}}$
Sn-Beta-200-AIE-cal	0.40	0.13	0.27	578	255
Sn-Beta-200-DC	0.23	0.11	0.12	388	117
Sn-Beta-150-AIE-cal	0.33	0.14	0.19	577	252
Sn-Beta-150-DC	0.27	0.12	0.15	416	124

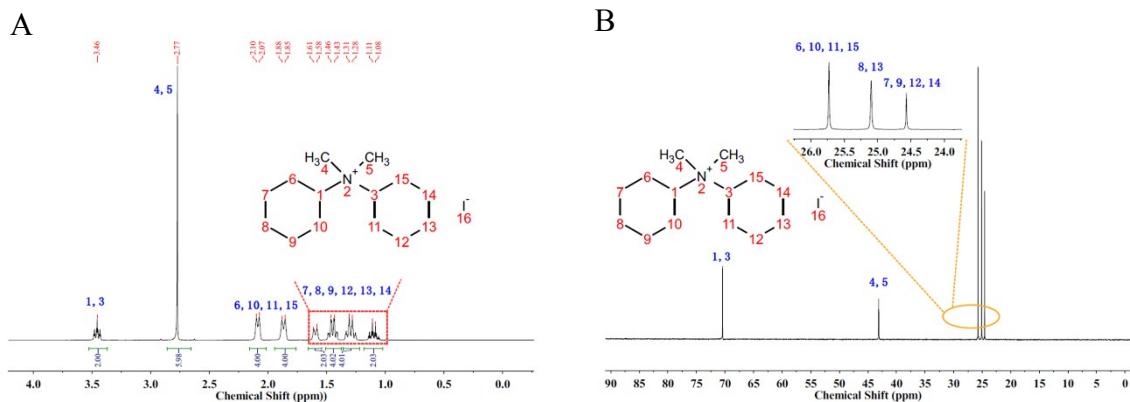
<sup>a</sup> Calculated by *t*-plot method

<sup>b</sup>  $V_{\text{meso}} = V_{\text{total}} - V_{\text{micro}}$

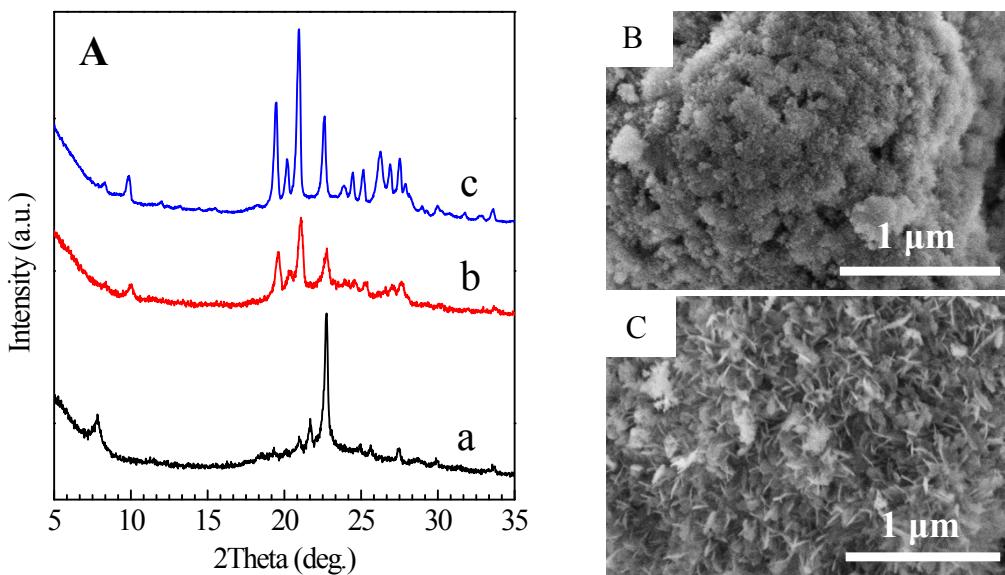
<sup>c</sup> Specific surface area (SSA) was determined by  $\text{N}_2$  adsorption at 77 K.

**Table S2**  $^{29}\text{Si}$  MAS NMR spectra deconvolution for different Sn-Beta zeolites

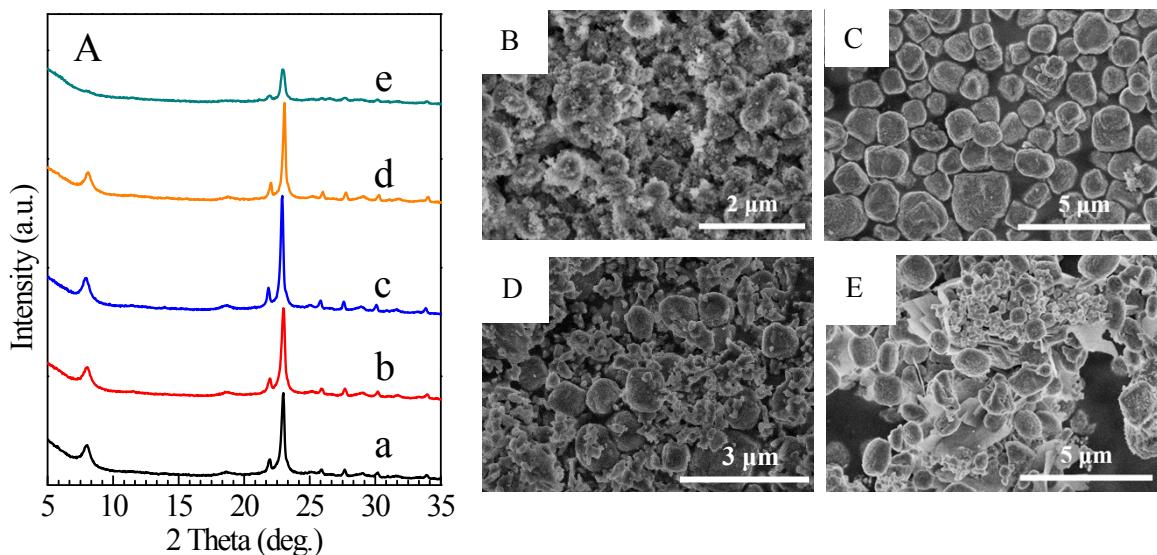
Sn-Beta-150-F			Sn-Beta-200		
$\delta$ (ppm)	attribution	area (%)	$\delta$ (ppm)	attribution	area (%)
-	-	-	-94.9	$\text{Q}^2$	5.58
-101.4	$\text{Q}^3$	4.79	-103.0	$\text{Q}^3$	24.37
-104.6		7.22			
-108.0	$\text{Q}^4$	13.77	-111.6	$\text{Q}^4$	70.04
-111.7		48.29			
-115.5		25.33			



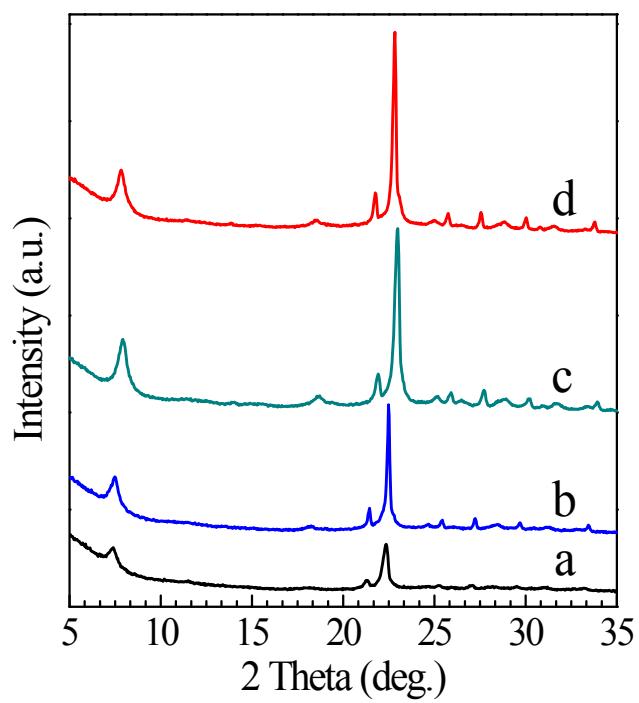
**Figure S1**  $^1\text{H}$  (A) and  $^{13}\text{C}$  (B) NMR spectra of *N*-cyclohexyl-*N,N*-dimethylcyclohexanaminium iodide



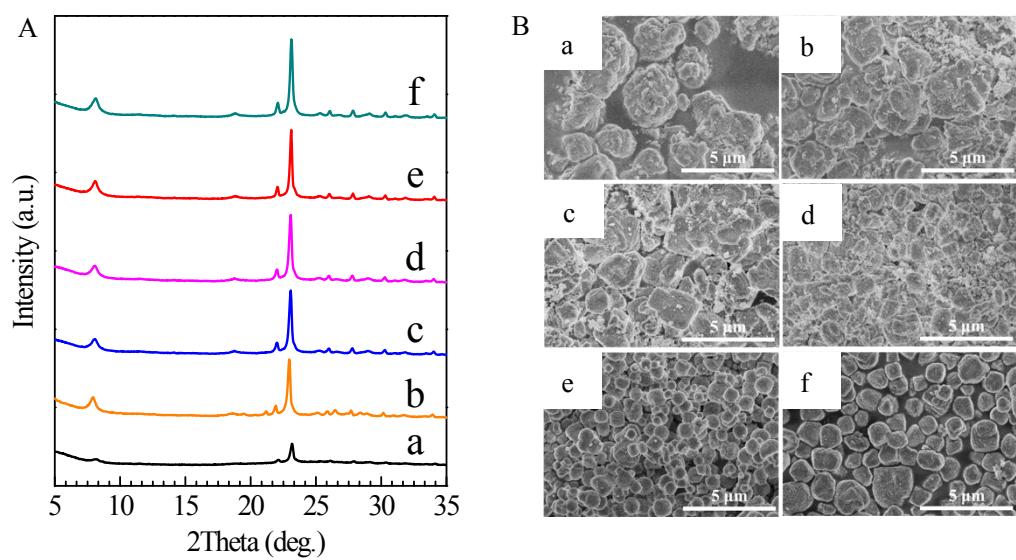
**Figure S2** XRD patterns (A) of the samples crystallized at 413 K (a), 423 K (b) and 433 K (c). SEM images of the samples crystallized at 423 K (B) and 433 K (C). Other crystallization conditions: 1 SiO<sub>2</sub> : 0.5 M<sub>2</sub>Cy<sub>2</sub>OH : 0.01 SnCl<sub>4</sub> : 0.05 NaOH : 22 H<sub>2</sub>O; 5 wt% seed; time, 14 d.



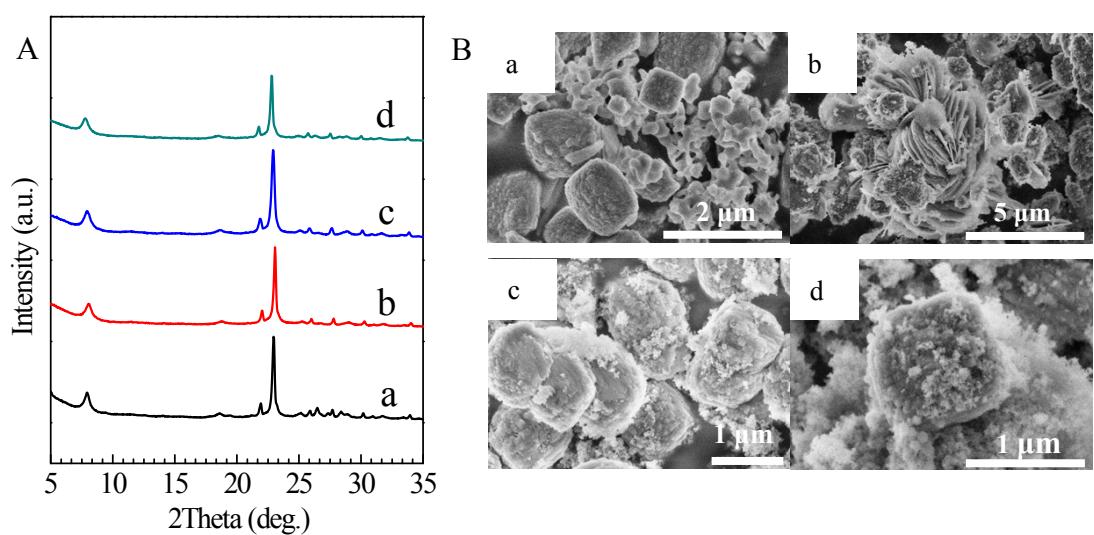
**Figure S3** XRD patterns (A) of the samples crystallized with a Na/Si ratio of 0.04 (a), 0.05 (b), 0.08 (c), 0.10 (d) and 0.15 (e), respectively. SEM images of the samples crystallized with a Na/Si ratio of 0.05 (B), 0.08 (C), 0.10 (D) and 0.15 (E), respectively. Other crystallization conditions: 1 SiO<sub>2</sub> : 0.5 M<sub>2</sub>Cy<sub>2</sub>OH : 0.01 SnCl<sub>4</sub> : (0.04 - 0.15) NaOH : 22 H<sub>2</sub>O; 5 wt% seed; temp., 413 K; time, 14 d.



**Figure S4** XRD patterns of the samples crystallized with a H<sub>2</sub>O/Si ratio of 19 (a), 15 (b), 10 (c), 7 (d), respectively. Other crystallization conditions: 1 SiO<sub>2</sub> : 0.5 M<sub>2</sub>Cy<sub>2</sub>OH : 0.01 SnCl<sub>4</sub> : 0.08 NaOH : (7 - 19) H<sub>2</sub>O; 5 wt% seed; temp., 413 K; time, 7 d.



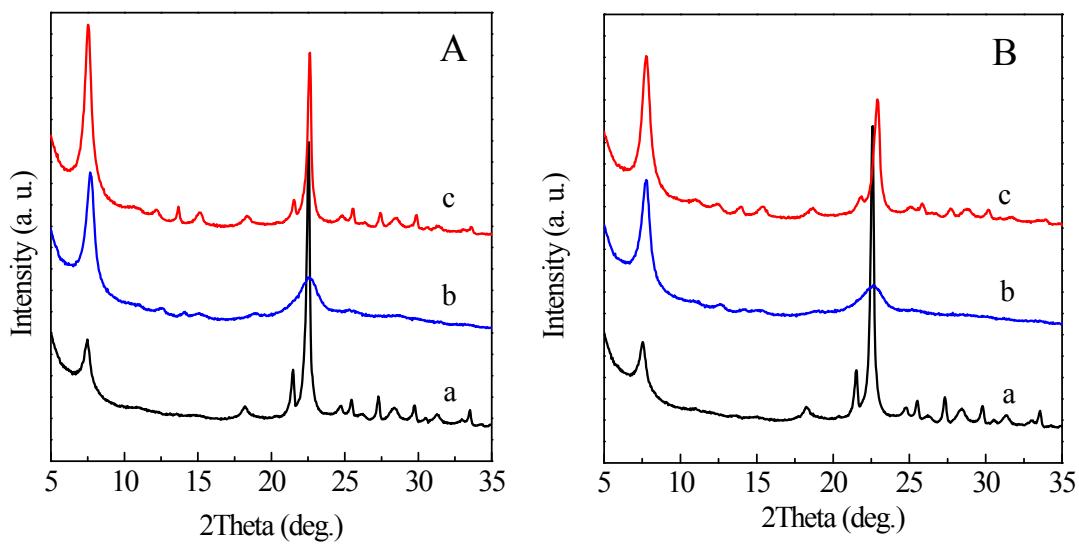
**Figure S5** XRD patterns of the samples crystallized for 2 d (a), 3 d (b), 5 d (c), 6 d (d), 7 d (e) and 10 d (f), respectively. Other crystallization conditions: 1 SiO<sub>2</sub> : 0.5 M<sub>2</sub>Cy<sub>2</sub>OH : 0.01 SnCl<sub>4</sub> : 0.08 NaOH : 7 H<sub>2</sub>O; 5 wt% seed; temp., 413 K.



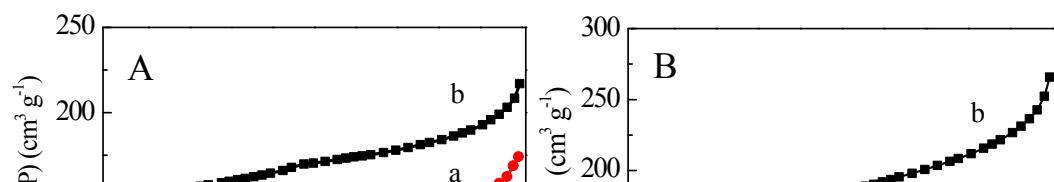
**Figure S6** XRD patterns (A) and SEM images (B) of the samples obtained with a Si/Sn ratio of 200 (a), 150 (b), 80 (c), 60 (d), respectively. Other crystallization conditions: 1 SiO<sub>2</sub> : 0.5 M<sub>2</sub>Cy<sub>2</sub>OH : (0.005 - 0.017) SnCl<sub>4</sub> : 0.08 NaOH : 7 H<sub>2</sub>O; 5 wt% seed; temp., 413 K; time, 7 d.



**Figure S7** XRD patterns of as-synthesized Sn-Beta-150-F.



**Figure S8** XRD patterns of of Sn-Beta-150 (A) and Sn-Beta-200 (B) in as-made form  
 (a), directly calcined form (b), and ion-exchanged and subsequently calcined form (c),  
 respectively.



**Figure S9** N<sub>2</sub> adsorption isotherms of Sn-Beta-150 (A) and Sn-Beta-200 (B) in directly calcined form (a), and ion-exchanged and subsequently calcined form (b).