

Supporting Information

Fe_{1-x}S/Nitrogen and Sulfur Co-doped Carbon Composite Derived from Nanosized Metal-Organic Framework for High-Performance Lithium-Ion Batteries

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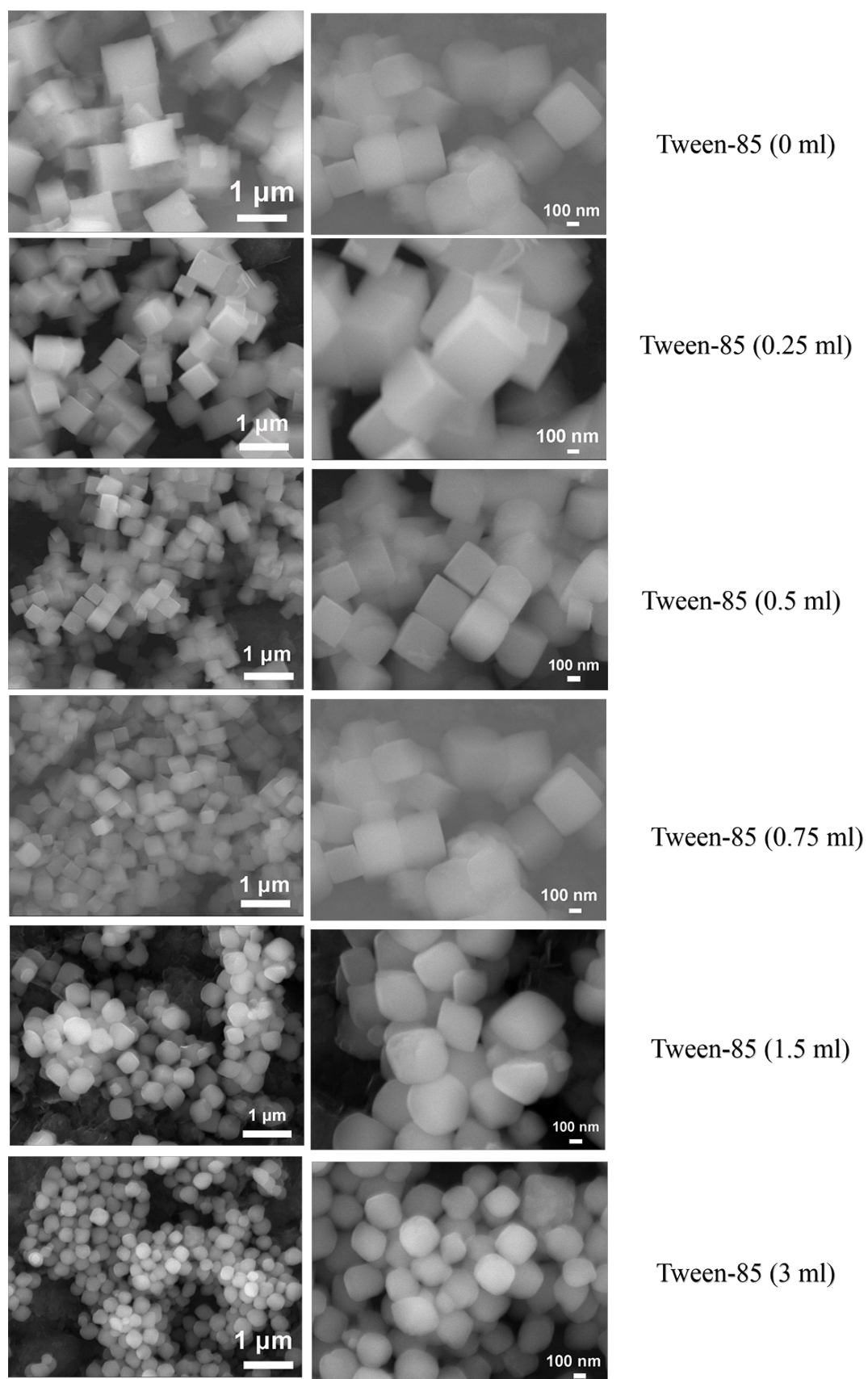


Fig. S1 The morphologies of Fe-MOF with adding different amounts of Tween-85.

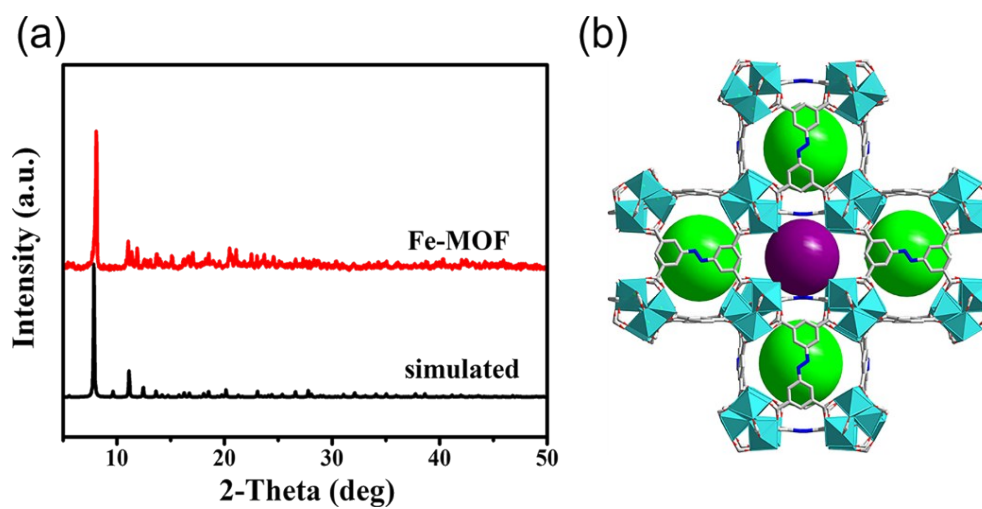


Fig. S2 (a)XRD pattern and (b) crystal structure of Fe-MOF.

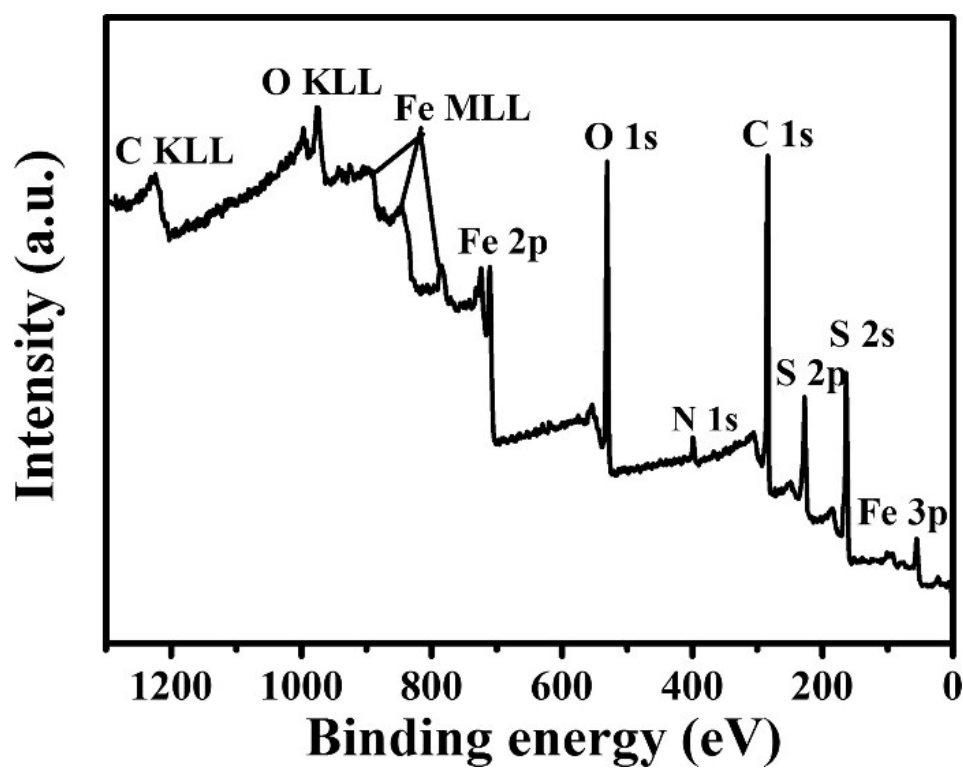


Fig. S3 The XPS survey spectrum of $\text{Fe}_{1-x}\text{S}/\text{NSC}$ composite.

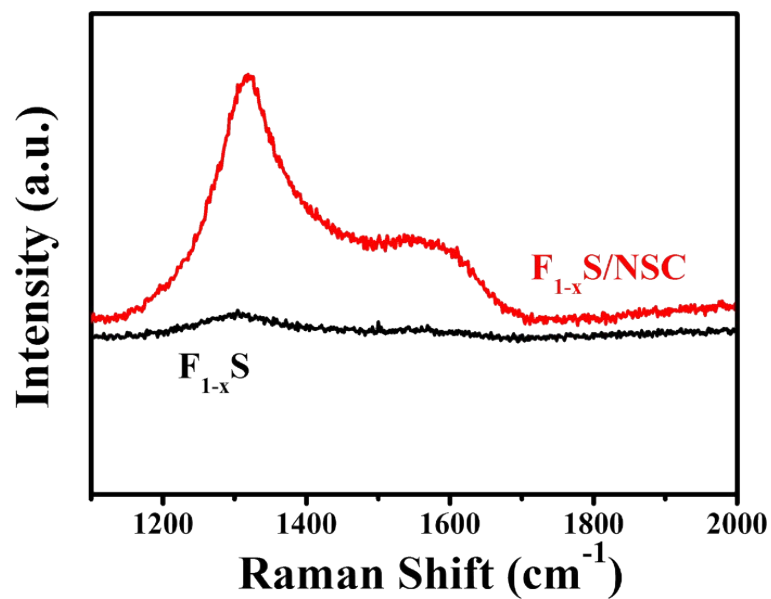


Fig. S4 Raman spectra of $\text{Fe}_{1-x}\text{S}/\text{NSC}$ and Fe_{1-x}S .

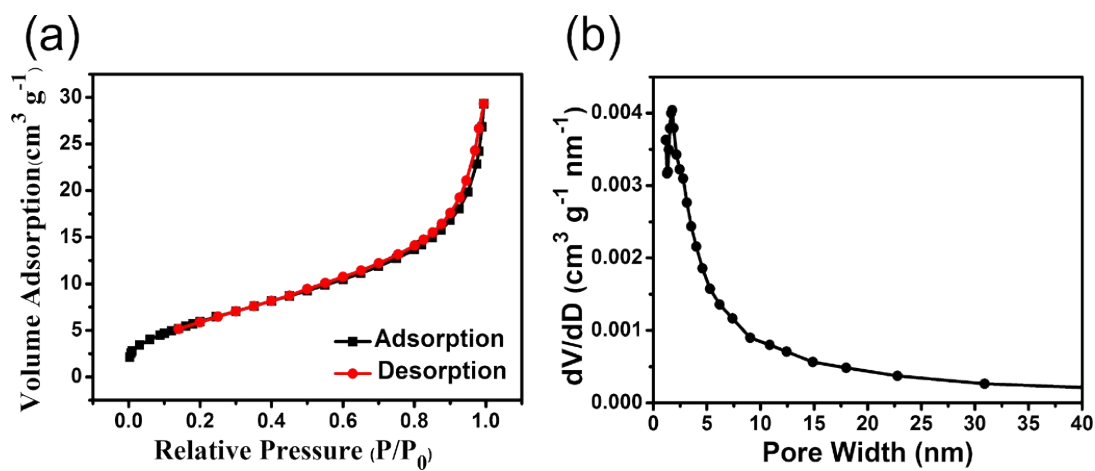


Fig. S5 N_2 sorption/desorption isotherms (a) and pore size distribution curve (b) of $\text{Fe}_{1-x}\text{S}/\text{NSC}$ nanocomposite.

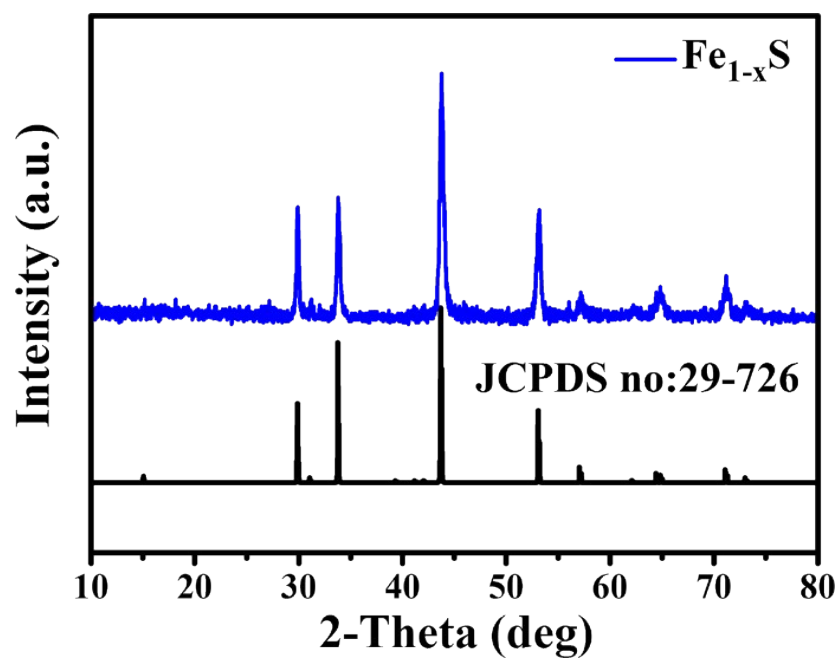


Fig. S6 The XRD pattern of Fe_{1-x}S .

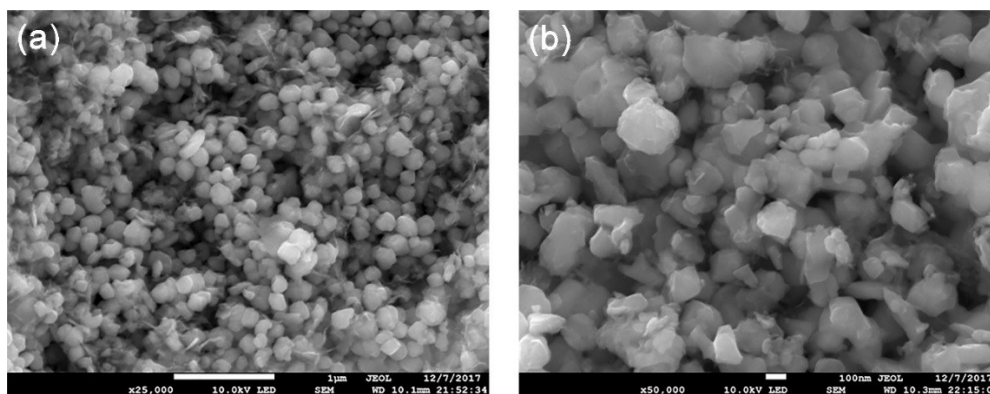


Fig. S7 The SEM images of Fe_{1-x}S .

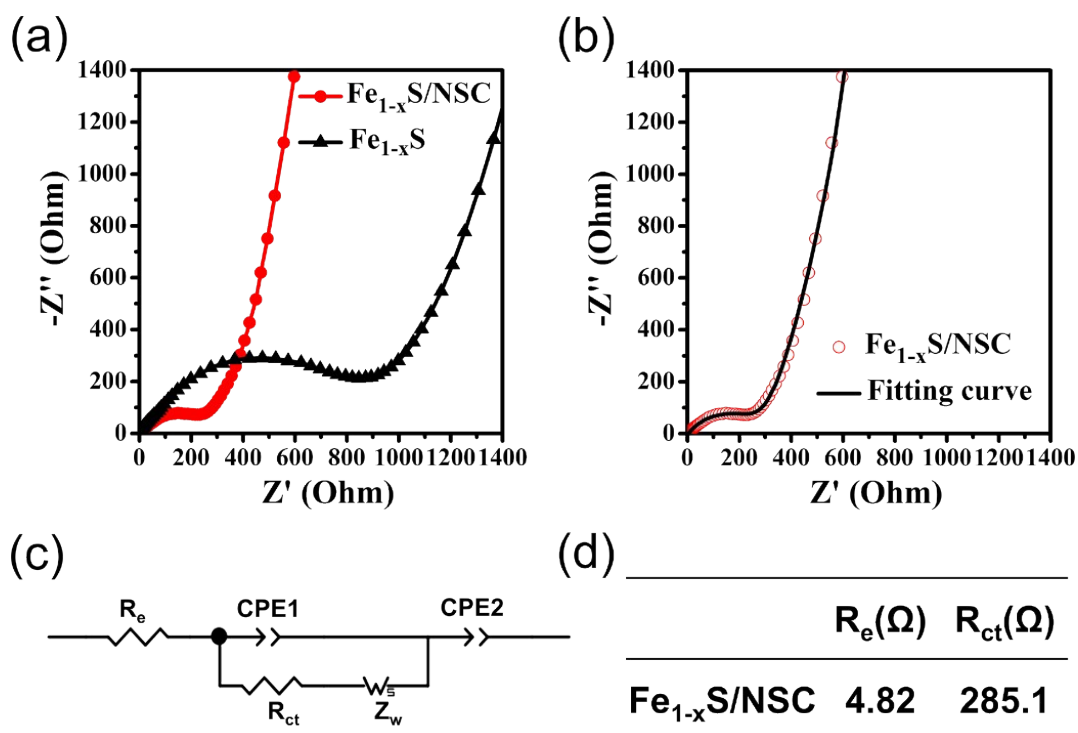


Fig. S8 (a) Nyquist plots of the $\text{Fe}_{1-x}\text{S/NSC}$ and Fe_{1-x}S electrodes; (b) Nyquist plots of the $\text{Fe}_{1-x}\text{S/NSC}$ and corresponding fitting curve; (c) Equivalent circuit model for interpreting Nyquist plots and (d) derived parameters of the $\text{Fe}_{1-x}\text{S/NSC}$. (R_e : electrolyte resistance; R_{ct} : charge transfer resistance; CPE: constant phase element; Z_w : Warburg impedance.)

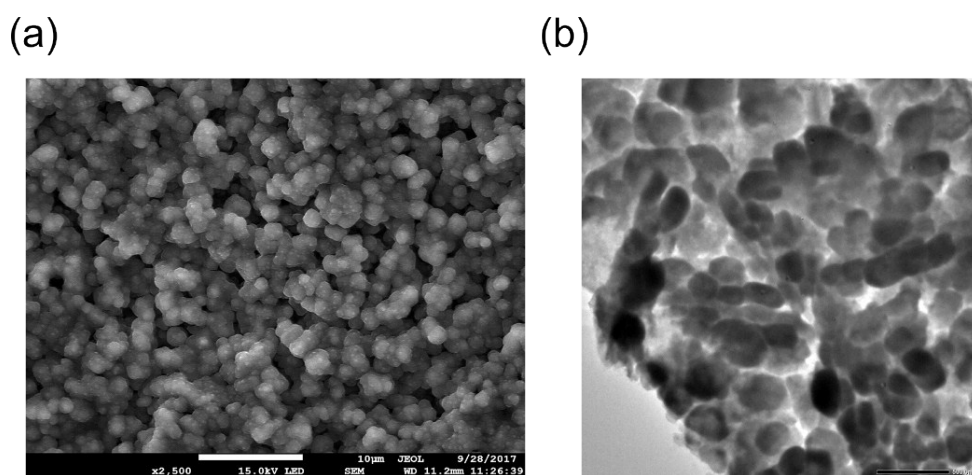


Fig. S9 (a) SEM images and (b) TEM images of the $\text{Fe}_{1-x}\text{S/NSC}$ electrode after 100 cycles.

Table S1. Comparison of the specific capacity and cycling performance for Fe_{1-x}S/NSC composite in this work with the previously reported iron sulfides-based electrodes.

Materials	Current density (A g⁻¹)	Specific Capacity (mAh g⁻¹)	Cycle number	Ref.
Fe_{1-x}S/NSC	0.1/1.0	1135/707	100/200	This work
Fe-S-CM	0.05	736	50	8
FeS@RGO	0.1	978	40	39
C@FeS	0.1	623	100	46
FeS@CNS	1.0	664	150	47
FeS@C	0.09	420	100	48
FeS₂@C	0.1	570	100	49
FeS@rGO	0.2/1.0	887/662	150/200	50