Supplementary material

A self-calibrating bimetallic lanthanide metal-organic luminescent sensor integrated with logic gate operation for detecting N-methylformamide

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Figure S1. The FT-IR images of $Eu_xTb_{2-x}(FDA)_3$. The peaks around 3000 cm⁻¹ were attributed to the O-H vibration. The peaks around 1680 cm⁻¹ were C=O stretch, which were extremely weak owing to the coordinating with lanthanide ions.



Figure S2. SEM images of Eu_{0.1}Tb_{1.9} (FDA)₃.



Figure S3. TEM images of Eu_{0.1}Tb_{1.9}(FDA)₃.



Figure S4. Thermal gravimetric analysis curves for Eu-MOF, Tb-MOF, and $Eu_{0.1}Tb_{1.9}$ -MOF





Figure S5. The spectra of (a) $Tb_2(FDA)_3$, (b) $Eu_2(FDA)_3$, (c) $Eu_{0.02}Tb_{1.98}(FDA)_3$, and (d) $Eu_{0.2}Tb_{1.8}(FDA)_3$ in solid-state, inset shows the powder under UV-light irradiation.





Figure S6. The suspension-state luminescent spectra of (a) $Eu_2(FDA)_3$, (b) $Tb_2(FDA)_3$, (c) $Eu_{0.02}Tb_{1.98}(FDA)_3$, and (d) $Eu_{0.2}Tb_{1.8}(FDA)_3$ within different volume content of NMF in DMF from 0 to 1, excited at 300 nm at RT.



Figure S7. The CIE chromaticity diagram shows the change at different volume ratios of NMF.



Figure S8. The (a) suspension-state luminescent spectra and (b) the corresponding curve ($I_{544nm}/I_{614nm}=3.836-53.63X+175.25X^2$, R²=0.9845) of the probe in different volume content of NMF in DMA from 0 to 1.



Figure S9. The (a) suspension-state luminescent spectra and (b) the linear model $(Log(I_{544nm}/I_{614nm})=-0.993+3.26X, R^2=0.969)$ of the sensor in different volume content of NMF in ethyl acetate from 0 to 1, excited at 300 nm at RT.



Figure S10. The (a) suspension-state luminescent spectra and (b) the normalization curve $(\text{Log}(I_{544\text{nm}}/I_{614\text{nm}})=-1.77+1.63X+0.83X^2, \text{ R}^2=0.952)$ of the sensor in N-Acetyl-N-Methylamine (NMA) different volume content of NMF in from 0 to 1, excited at 300 nm at RT.



Figure S11. The integrated intensity ratio of the $Eu_{0.1}Tb_{1.9}(FDA)_3$ changed with the contact time of NMF.



Figure S12. Integrated intensity ratio I_{544nm}/I_{612nm} of $Eu_{0.1}Tb_{1.9}(FDA)_3$ in the solvents for five runs.



Figure S13. The XRD patterns of $Eu_{0.1}Tb_{1.9}(FDA)_3$ after five runs.



Figure S14. The IR patterns of the sensing probe after cycling.



Figure S15. UV-vis spectra of (a) DMF and NMF; (b) ligand, and MOF.

Table S1.	The ICP	analysis	result of	of mixed	Eu _x Tb ₁₋	x-MOF
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LMOF	Eu(mg/L)	Tb(mg/L)	Ratio	Ratio in theory
[Eu _{0.02} Tb _{1.98} (FDA) ₃ (DMF) ₂]·2DMF	0.736	49.65	0.0148	0.0101
[Eu _{0.1} Tb _{1.9} (FDA) ₃ (DMF) ₂]·2DMF	3.815	91.10	0.0418	0.0526
[Eu _{0.2} Tb _{1.8} (FDA) ₃ (DMF) ₂]·2DMF	5.394	56.84	0.0949	0.111

Sensor system	Linear range	Advantage	Disadvantage	Ref
Liquid chromatography-mass spectrometry (LC- MS)	0.004-8 μg/mL	Accurate, and reliable.	Complicated process.	1
Gas chromatography- mass spectrometry (GC-MS)	0.3-30 μg/mL	Non-invasive, and reliable.	Requiring highly skilled personnel to operate.	2
Eu@MOF-1	0-100 μΜ	Fast response, and high selectivity.	Monometallic MOF, non- self reference.	3
[Eu _{0.1} Tb _{1.9} (FDA) ₃]	0-100%	Self-calibrating, visibility and good reusability.	Preliminary test.	This work

Table S2. Performances of different sensor systems for detection of NMF.

Table S3. Luminescence lifetimes of sensor $Eu_{0.1}Tb_{1.9}$ (FDA)₃ with excitation at 300 nm in different volume ratio of NMF.

The ratio of NMF	⁵ D ₄ of Tb ³⁺ /μs	⁵ D ₀ of Eu ³⁺ /µs	η
0	69	1528	0.96
0.1	100	1512	0.94
0.2	218	1511	0.87
0.3	331	1457	0.80
0.4	549	1487	0.66
0.5	1375	1706	0.15
0.6	1377	1476	0.15
0.7	1446	1449	0.11
0.8	1436	-	0.10
0.9	2436	-	-
1	1425	-	0.1

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