

## Supplementary Information

# Hierarchical $\text{MnCo}_2\text{O}_4@\text{NiMoO}_4$ as Free-Standing Core-Shell Nanowire Arrays with Synergistic Effect for Enhanced Supercapacitor Performance

Jaafar Abdul-Aziz Mehrez<sup>a</sup>, Kwadwo Asare Owusu<sup>a</sup>, Qiang Chen<sup>a</sup>, Lun Li<sup>a</sup>, Khawla Hamwi<sup>c</sup>, Wen Luo<sup>\*,b</sup> & Liqiang Mai<sup>\*,a</sup>

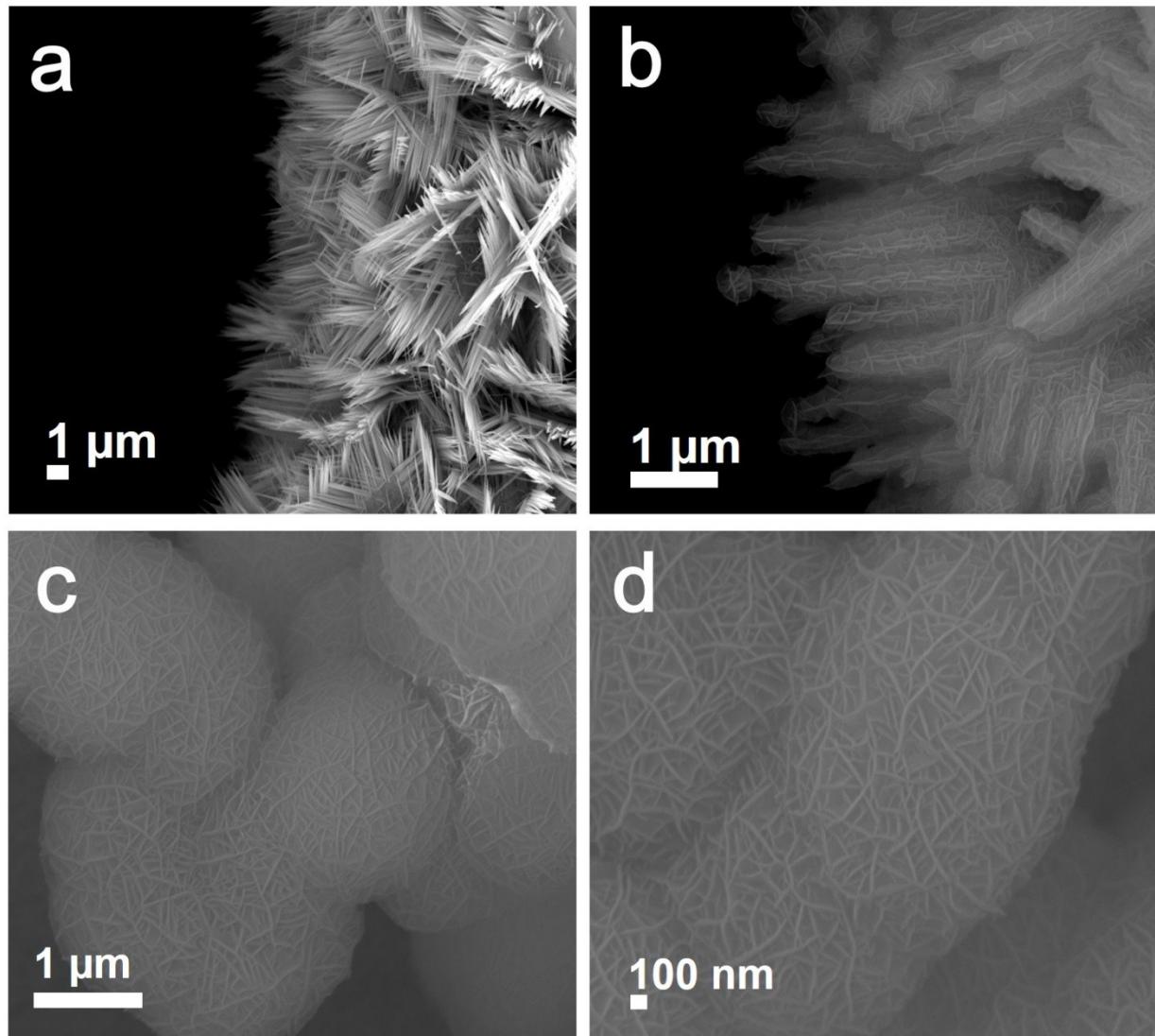
<sup>a</sup> State Key Laboratory of Advanced Technology for Materials Synthesis and Processing, International School of Materials Science and Engineering, Wuhan University of Technology, Wuhan 430070, China. E-mail: [mlq518@whut.edu.cn](mailto:mlq518@whut.edu.cn)

<sup>b</sup> Department of Physics, School of Science, Wuhan University of Technology, Wuhan 430070, China.

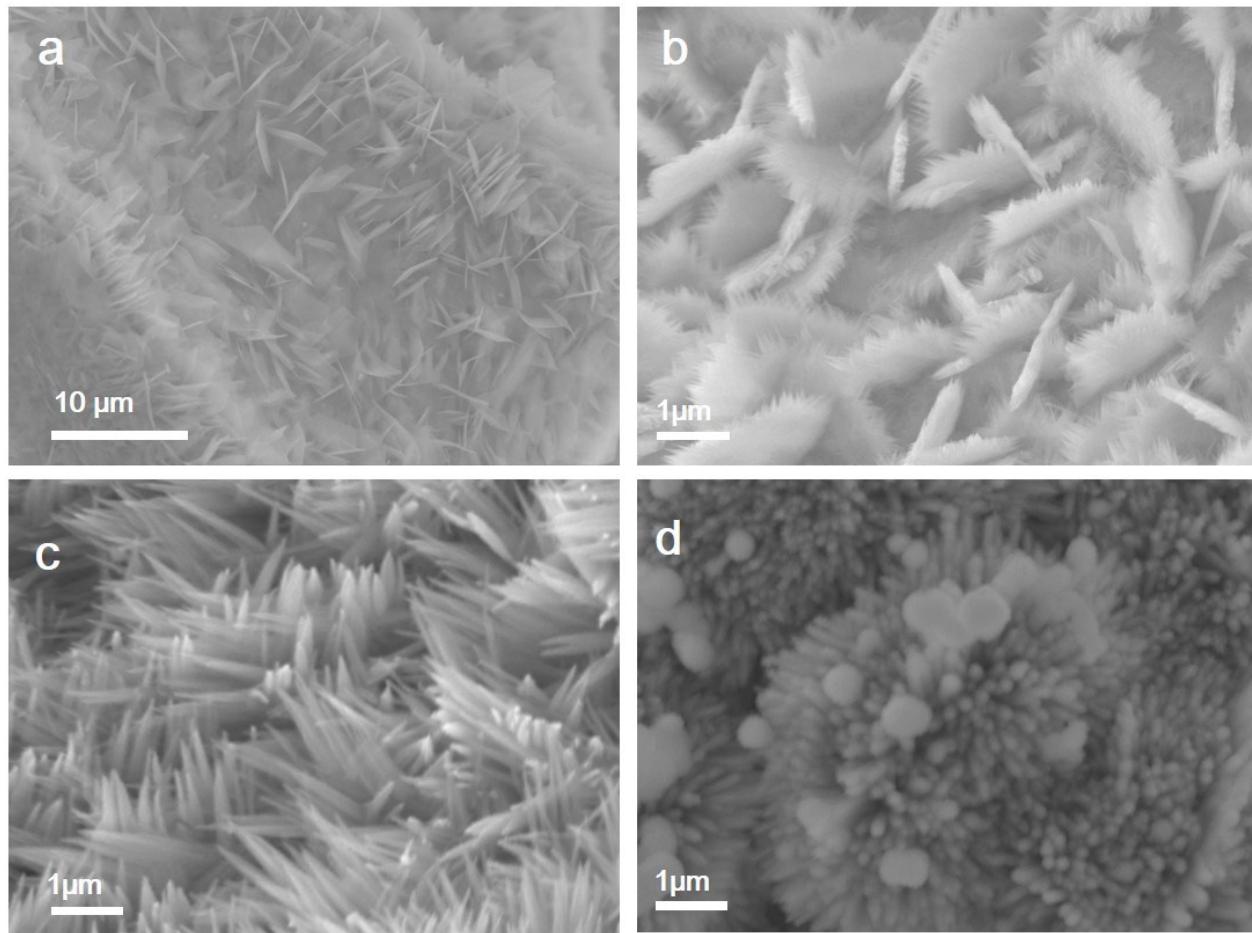
<sup>c</sup> Department of electronics engineering, Tishreen University, Lattakia, Syria.

\*Corresponding authors: Wen Luo (E-mail address: [luowen\\_1991@whut.edu.cn](mailto:luowen_1991@whut.edu.cn).)

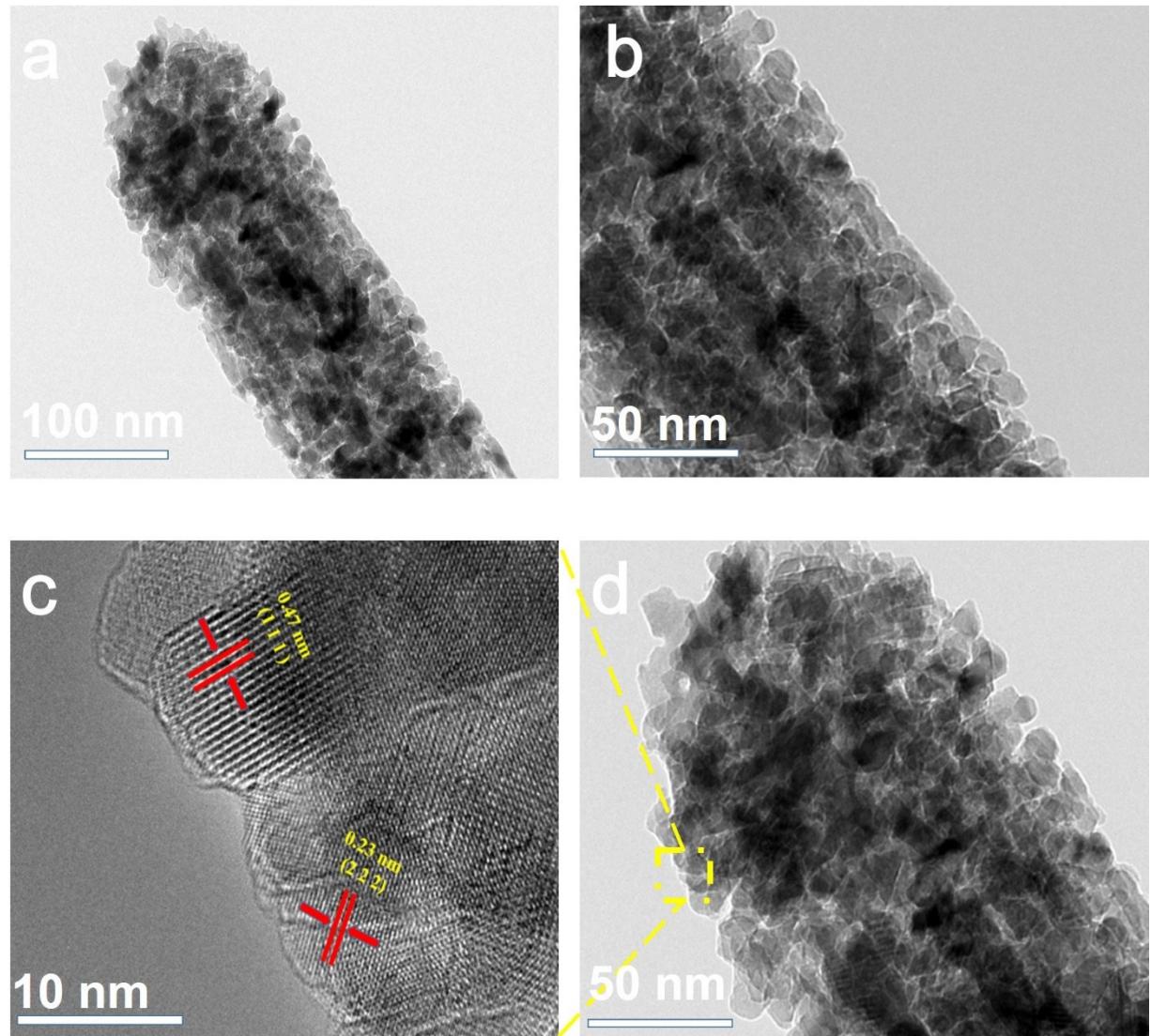
Liqiang Mai (Email address: [mlq518@whut.edu.cn](mailto:mlq518@whut.edu.cn) )



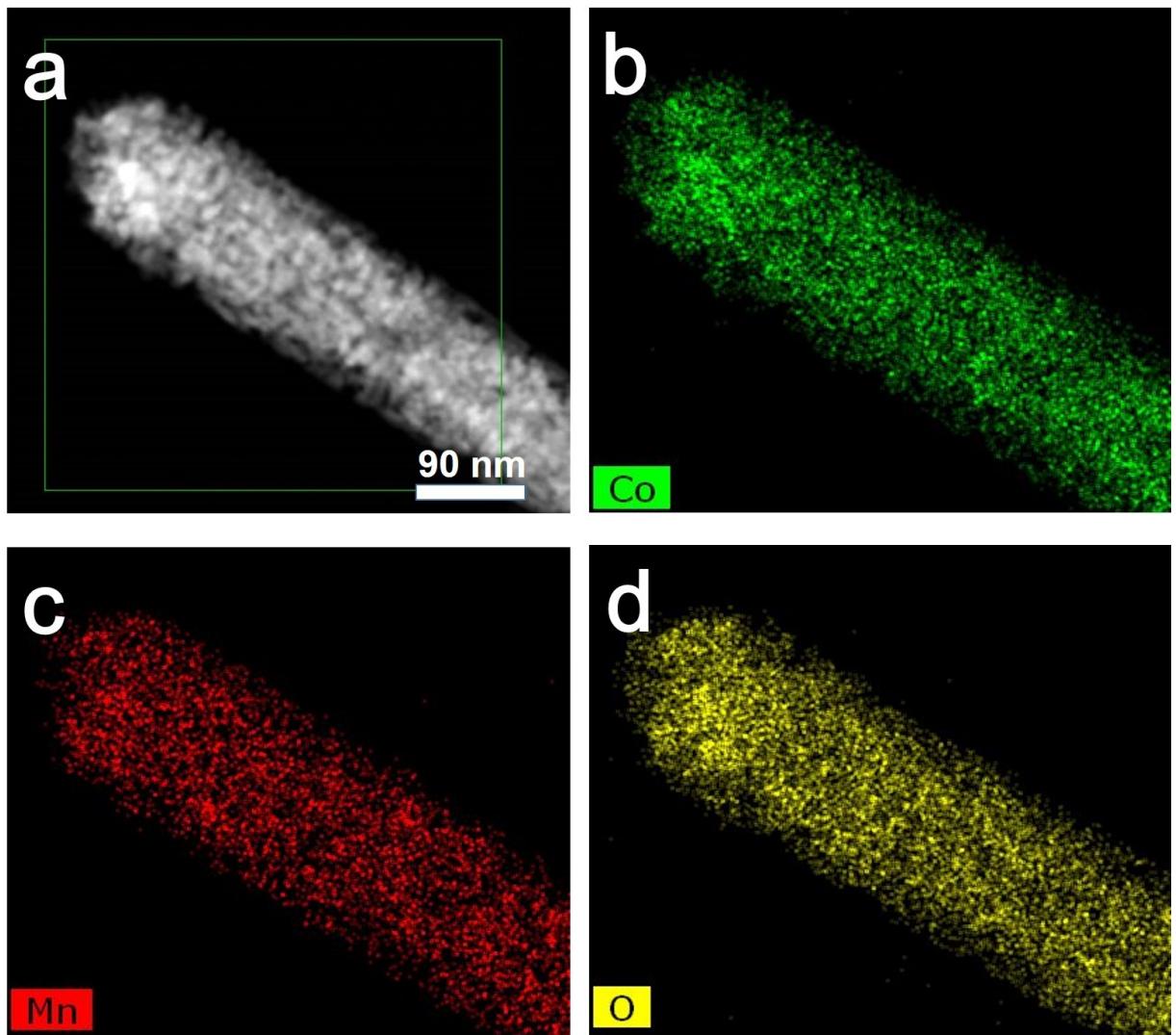
**Figure S1.** (a) typical SEM image of  $\text{MnCo}_2\text{O}_4$  NWAs in a low magnification, (b) SEM image of  $\text{MnCo}_2\text{O}_4@\text{NiMoO}_4$  core-shell NWAs in low magnification, (c) and (d) typical SEM images of pure  $\text{NiMoO}_4$  grown directly over the NF with high and low magnification.



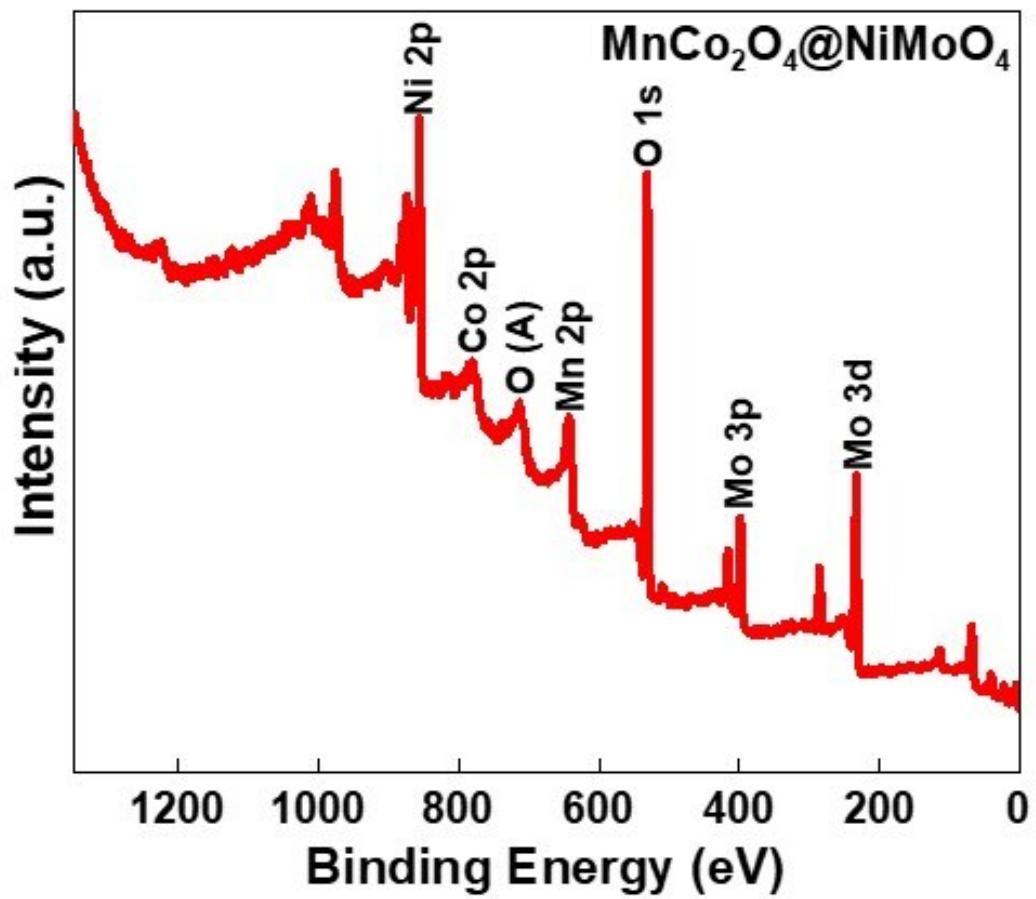
**Figure S2.** Typical (a) and (b) SEM images of  $\text{MnCo}_2\text{O}_4$  nanowires growth procedure with different reaction time duration (1 hour and 2 hours), respectively. (c) and (d) typical SEM images of  $\text{MnCo}_2\text{O}_4@\text{NiMoO}_4$  core-shell structure with different hydrothermal reaction duration (3 hours and 12 hours), respectively.



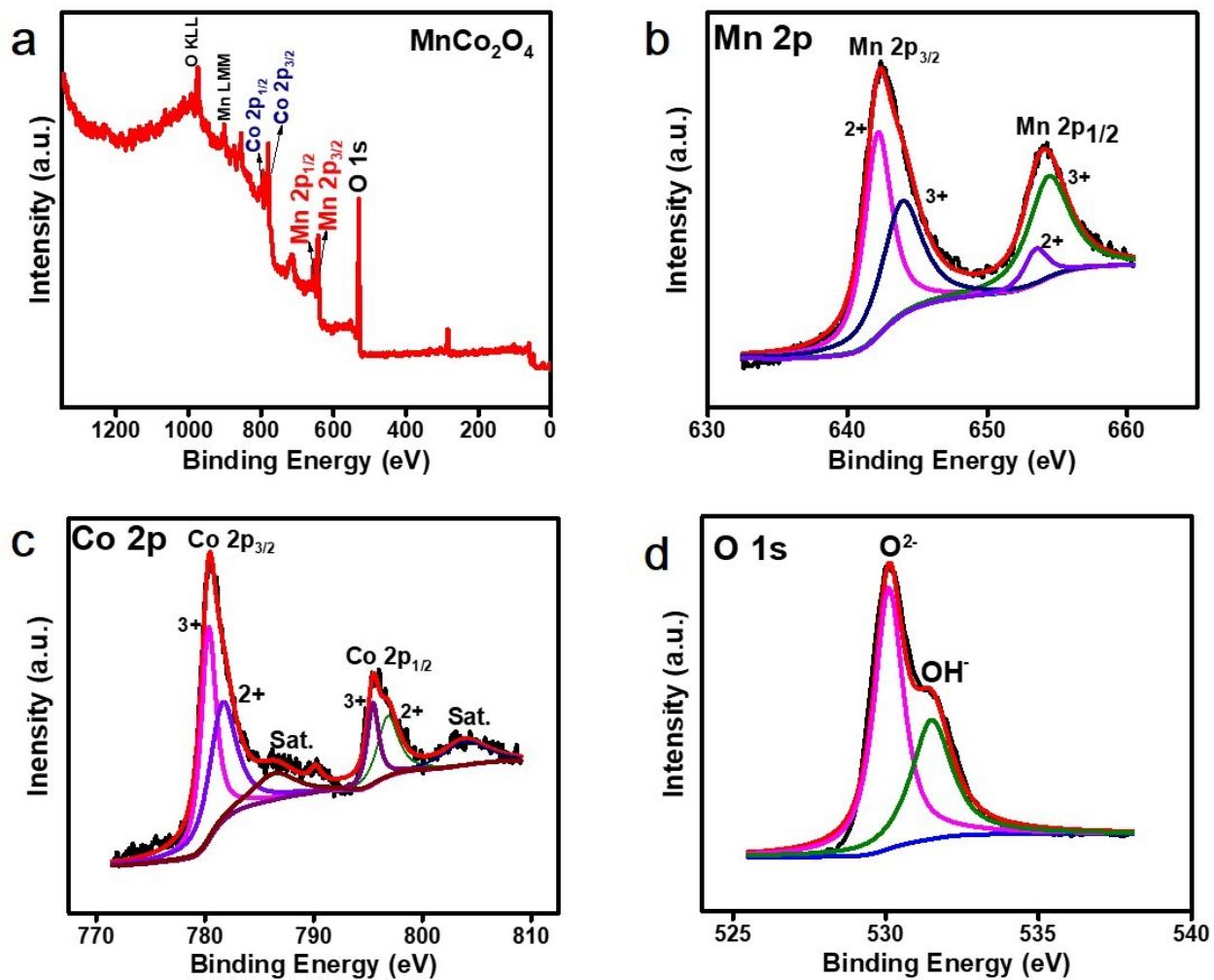
**Figure S3.** Typical (a),(b) and (d) TEM images of a single MnCo<sub>2</sub>O<sub>4</sub> nanowire, (c) HRTEM image of single MnCo<sub>2</sub>O<sub>4</sub> nanowire.



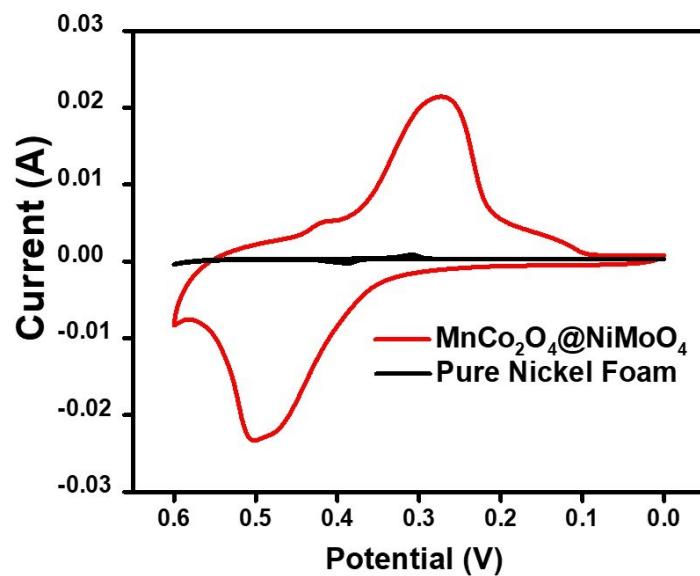
**Figure S4.** (a) Demonstrating  $\text{MnCo}_2\text{O}_4$  hierarchical structure of a single nanowire, (b-d) corresponding EDS mapping.



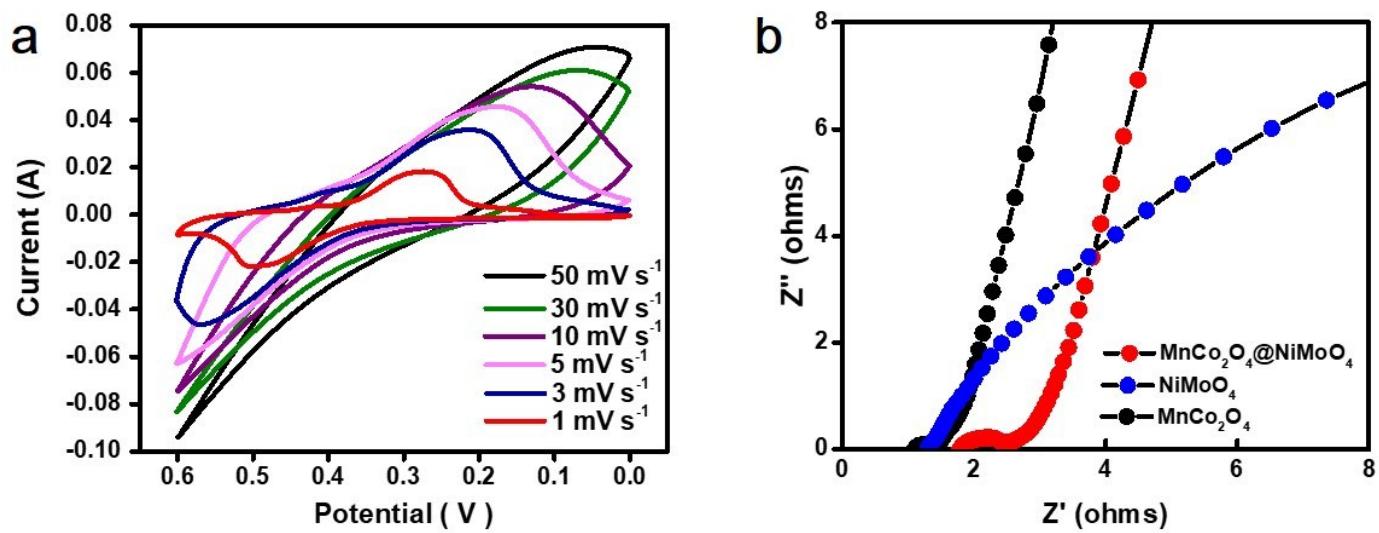
**Figure S5.** XPS spectra of as-synthesized  $\text{MnCo}_2\text{O}_4@\text{NiMoO}_4$  NWAs.



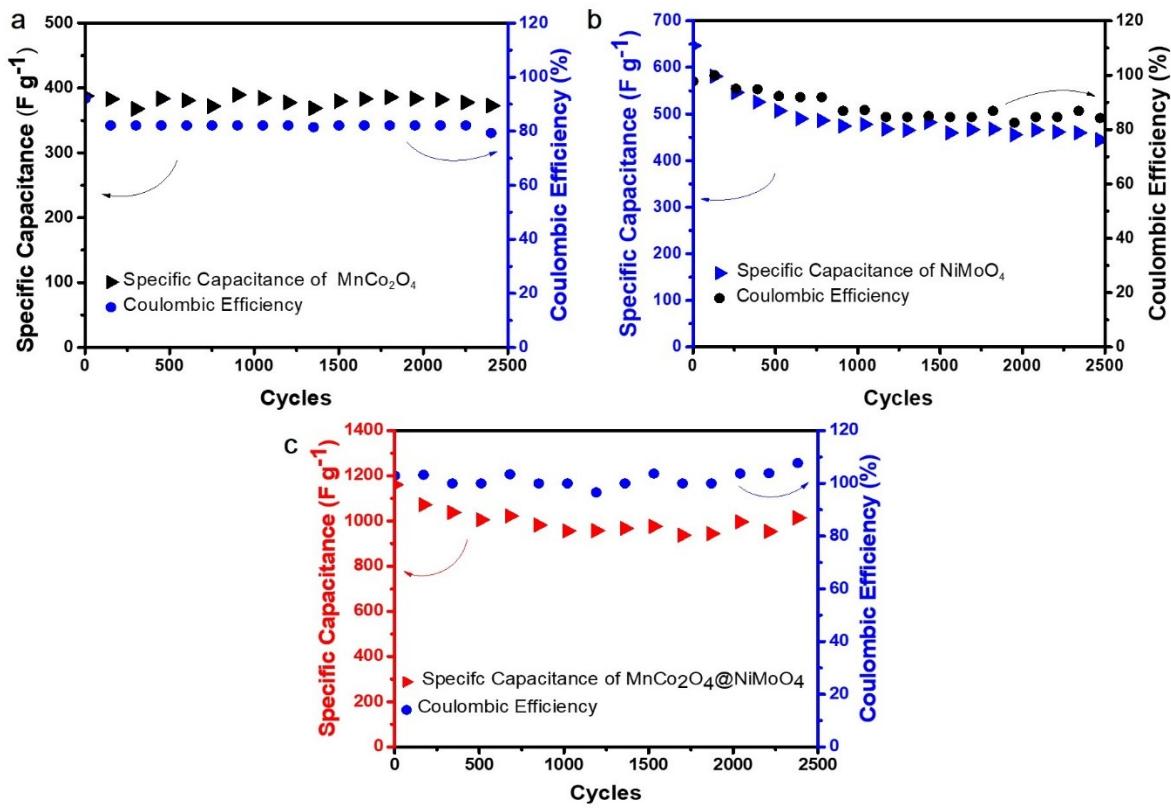
**Figure S6.** (a) XPS spectra of as-synthesized  $\text{MnCo}_2\text{O}_4$  NWAs, (b-d) XPS survey scan of Mn 2p, Co 2p and O 1s regions, respectively.



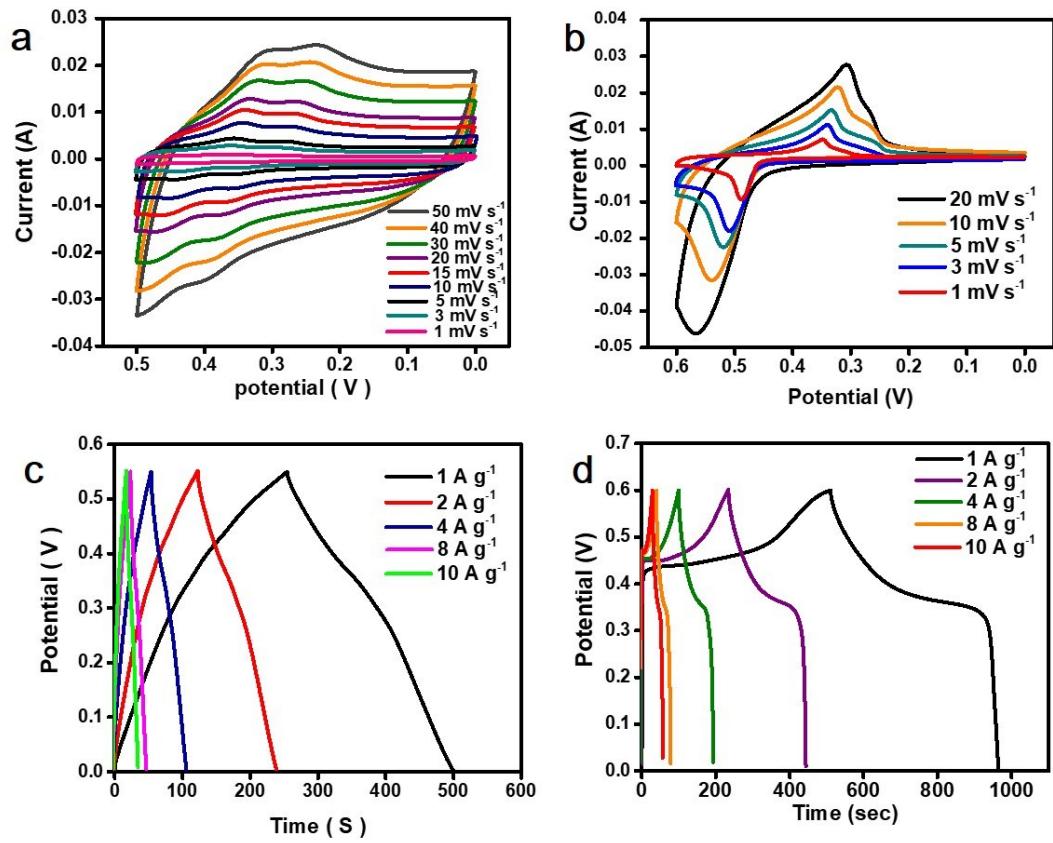
**Figure S7.** CV curves comparison of  $\text{MnCo}_2\text{O}_4@\text{NiMoO}_4$  and nickel foam current collector at scan rate of  $1 \text{ mV s}^{-1}$ .



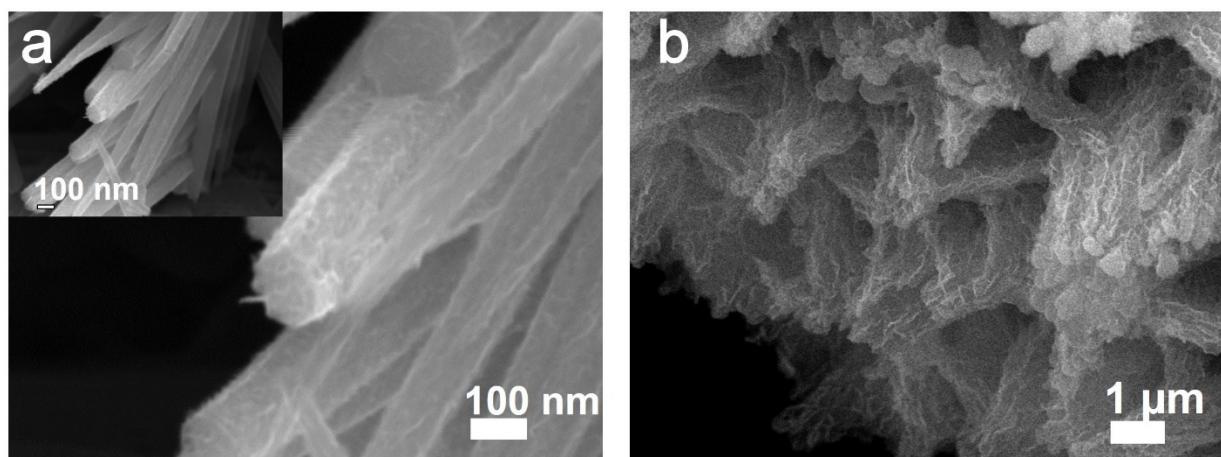
**Figure S8.** (a) CV curves of  $\text{MnCo}_2\text{O}_4@\text{NiMoO}_4$  obtained at different scanning rates, (b) EIS measurements of  $\text{MnCo}_2\text{O}_4$ ,  $\text{NiMoO}_4$  and  $\text{MnCo}_2\text{O}_4@\text{NiMoO}_4$  electrodes.



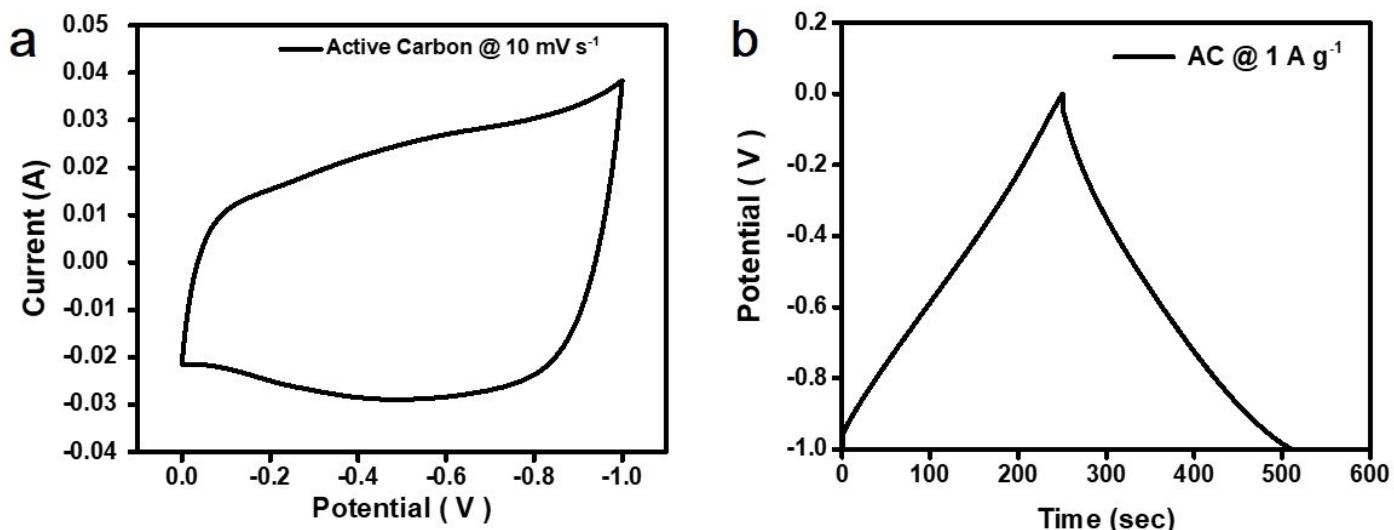
**Figure S9.** Coulombic efficiency profiles of (a) MnCo<sub>2</sub>O<sub>4</sub>, (b) NiMoO<sub>4</sub> and (c) MnCo<sub>2</sub>O<sub>4</sub>@NiMoO<sub>4</sub> during 2500 successive cycles at 5 A g<sup>-1</sup>.



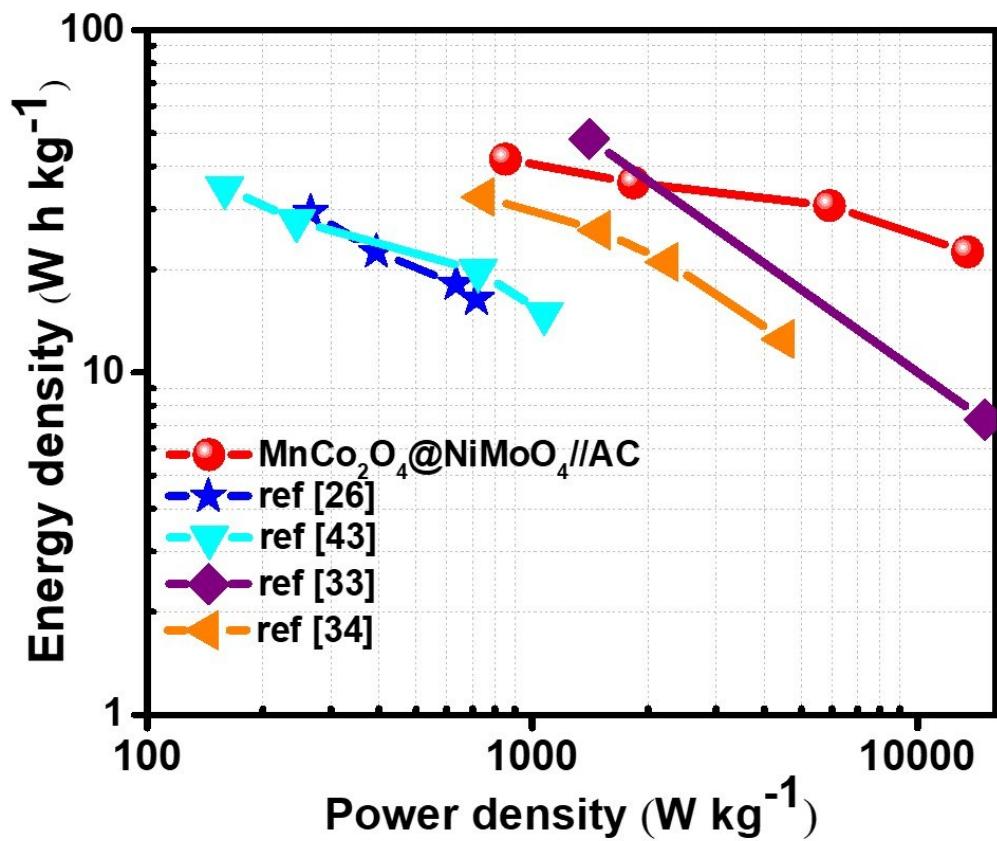
**Figure S10.** (a) CV curves of  $\text{MnCo}_2\text{O}_4$  at different scanning rates, (b) CV curves of  $\text{NiMoO}_4$  at different scanning rates, (c) and (d) GCD of  $\text{MnCo}_2\text{O}_4$  and  $\text{NiMoO}_4$ , respectively at different current densities.



**Figure S11.** (a) Typical SEM images of MnCo<sub>2</sub>O<sub>4</sub> after cycling (b) MnCo<sub>2</sub>O<sub>4</sub>@NiMoO<sub>4</sub> SEM images after cycling.



**Figure S12.** (a) CV curve of AC at a scanning rate of 10 mV s<sup>-1</sup>. (b) Galvanostatic charge-discharge of AC at a current density of 1 A g<sup>-1</sup>.



**Figure S13.** Ragone plot of  $\text{MnCo}_2\text{O}_4@\text{NiMoO}_4/\text{AC}$  device.

**Table S1.** Comparison of the maximum  $C_s$  of some reported manganese or cobalt oxide/hydroxide based composites as pseudocapacitor electrode material and the present work.

<i>Electrodes based on material</i>	<i>Electrolyte</i>	<i>Specific capacitance</i>	<i>Reference</i>
MnCo <sub>2</sub> O <sub>4</sub> Nanowire arrays	<b>1 M KOH</b>	<b>349.8 F g<sup>-1</sup> @ 1 Ag<sup>-1</sup></b>	<sup>1</sup>
MnCo <sub>2</sub> O <sub>4</sub> Nanosheet films	-	<b>400 F g<sup>-1</sup> @ 1 Ag<sup>-1</sup></b>	<sup>2</sup>
Urchin like MnCo <sub>2</sub> O <sub>4.5</sub>	-	<b>129.2 F g<sup>-1</sup> @ 0.1 Ag<sup>-1</sup></b>	<sup>3</sup>
MnCo <sub>2</sub> O <sub>4</sub> nanosheets	<b>6 M KOH</b>	<b>420 F g<sup>-1</sup> @ 1 Ag<sup>-1</sup></b>	<sup>4</sup>
MnCo <sub>2</sub> O <sub>4</sub> @Ni(OH) <sub>2</sub> hierarchical structure	<b>2 M KOH</b>	<b>2154 F g<sup>-1</sup> @ 5 Ag<sup>-1</sup></b>	<sup>5</sup>
MnCo <sub>2</sub> O <sub>4</sub> @MnO <sub>2</sub> hierarchical structure	<b>3 M KOH</b>	<b>858 F g<sup>-1</sup> @ 1 Ag<sup>-1</sup></b>	<sup>6</sup>
MnO <sub>2</sub> @NiMoO <sub>4</sub> core-shell hierarchical structure	<b>2 M KOH</b>	<b>582.2 F g<sup>-1</sup> @ 1 Ag<sup>-1</sup></b>	<sup>7</sup>
Nanosheet-based CoMoO <sub>4</sub> -NiMoO <sub>4</sub> nanotubes	<b>3 M KOH</b>	<b>751 F g<sup>-1</sup> @ 1 Ag<sup>-1</sup></b>	<sup>8</sup>
NiCo <sub>2</sub> O <sub>4</sub> -rGO composite	<b>2 M KOH</b>	<b>1222 F g<sup>-1</sup> @ 0.5 Ag<sup>-1</sup></b>	<sup>9</sup>
MnCo <sub>2</sub> O <sub>4</sub> @NiMoO <sub>4</sub> core-shell heterostructure	<b>1 M KOH</b>	<b>1244 F g<sup>-1</sup> @ 1 Ag<sup>-1</sup></b>	This work

## References

- 1 L. Li, Y. Q. Zhang, X. Y. Liu, S. J. Shi, X. Y. Zhao, H. Zhang, X. Ge, G. F. Cai, C. D. Gu, X. L. Wang and J. P. Tu, *Electrochimica Acta*, 2014, **116**, 467–474.
- 2 T. Nguyen, M. Boudard, L. Rapenne, O. Chaix-Pluchery, M. J. Carmezim and M. F. Montemor, *RSC Adv.*, 2015, **5**, 27844–27852.
- 3 W. Li, K. Xu, G. Song, X. Zhou, R. Zou, J. Yang, Z. Chen and J. Hu, *CrystEngComm*, 2014, **16**, 2335–2339.
- 4 L. Li, F. He, S. Gai, S. Zhang, P. Gao, M. Zhang, Y. Chen and P. Yang, *CrystEngComm*, 2014, **16**, 9873–9881.
- 5 Y. Zhao, L. Hu, S. Zhao and L. Wu, *Adv. Funct. Mater.*, 2016, **26**, 4085–4093.
- 6 X. Zheng, Y. Ye, Q. Yang, B. Geng and X. Zhang, *Dalton transactions (Cambridge, England : 2003)*, 2016, **45**, 572–578.
- 7 X. Wang, H. Xia, J. Gao, B. Shi, Y. Fang and M. Shao, *J. Mater. Chem. A*, 2016, **4**, 18181–18187.
- 8 Q. Yang and S.-Y. Lin, *RSC Adv.*, 2016, **6**, 10520–10526.
- 9 X. Wang, W. S. Liu, X. Lu and P. S. Lee, *J. Mater. Chem.*, 2012, **22**, 23114.