

Supplementary Information

An efficient yellow thermally activated delayed fluorescence emitter with universal applications in both doped and non-doped organic light-emitting diodes

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Table S1. Summary of device performance of yellow TADF-based OLEDs reported in the literature.

Yellow TADF emitter	λ_{EL}^a (nm)	L ^b (cd m ⁻²)	CE ^b (cd A ⁻¹)	PE ^b (lm W ⁻¹)	EQE ^b (%)	EQE ^c (%)	RO ^d (%)	Ref.
OPDPO	560	32590	73.1	38.2	26.7	21.8	18	This work
<i>p</i> -Px2BBP	548	57120	20.1	/	6.9	/	/	1
3,6-2TPA-TXO	551	20000	82.7	78.9	24.3	3.7	75	2
4CzPNPh	560	~30000	63.3	53.7	22.5	20.9	8	3
TXO-TPA	552	/	43.3	47.4	18.5	~6.0	68	4
DBT-BZ-PTZ	539	~10000	46.0	43.3	15.1	9.7	36	5
DBT-BZ-PTZ	563	~24000	26.5	29.1	9.7	8.5	13	5
CP-BP-PTZ	542	32250	62.4	57.3	19.6	13.6	31	6
CP-BP-PTZ	554	46820	46.1	55.7	15.3	12.7	17	6
Py56	562	~10000	96.3	105.5	29.2	20.6	30	7
CzSOXO	550	8466	43.1	/	13.6	8.8	35	8
PXZDSO2	560	17000	49.3	38.5	16.7	13.7	18	9
b3	574	~30000	/	/	9.0	~6.2	31	10
TPXZ- <i>as</i> -TAZ	560	11270	34.8	31.2	13.0	11.5	12	11
tri-PXZ-TRZ	550	/	/	/	13.3	/	/	12
TA-3Cz	546	23145	39.0	40.8	11.8	~6.0	50	13

Abbreviations: L, Luminance; CE, current efficiency; PE, power efficiency; EQE, external quantum efficiency; RO, efficiency roll-off; Ref, reference.

^a EL peak wavelength. ^b Measured at maximum value. ^c Measured at a luminance of 1000 cd m⁻². ^d RO = EQE roll-off from maximum value to that at 1000 cd m⁻².

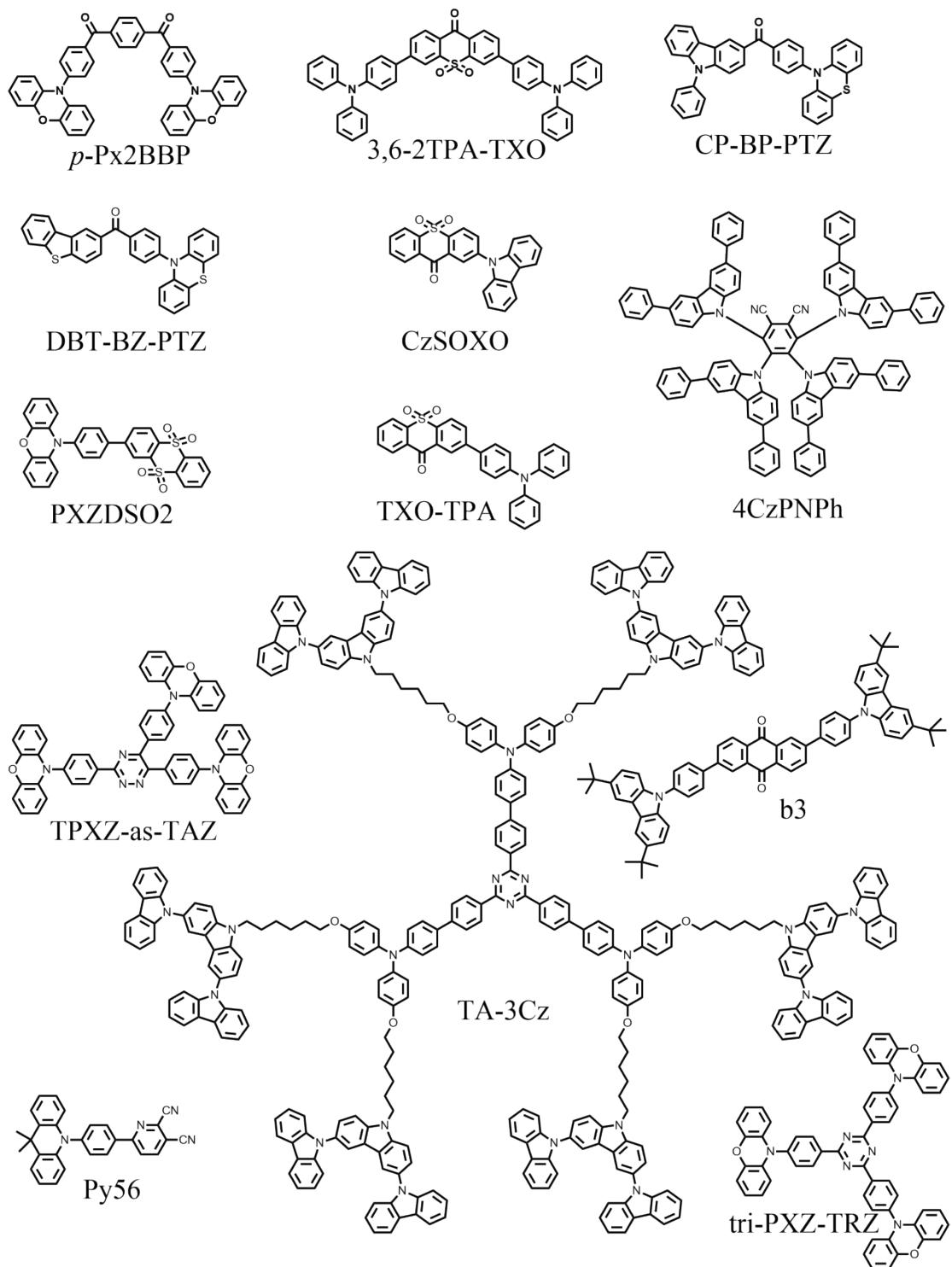


Figure S1. Chemical structures of previous yellow TADF emitter materials listed in Table S1.

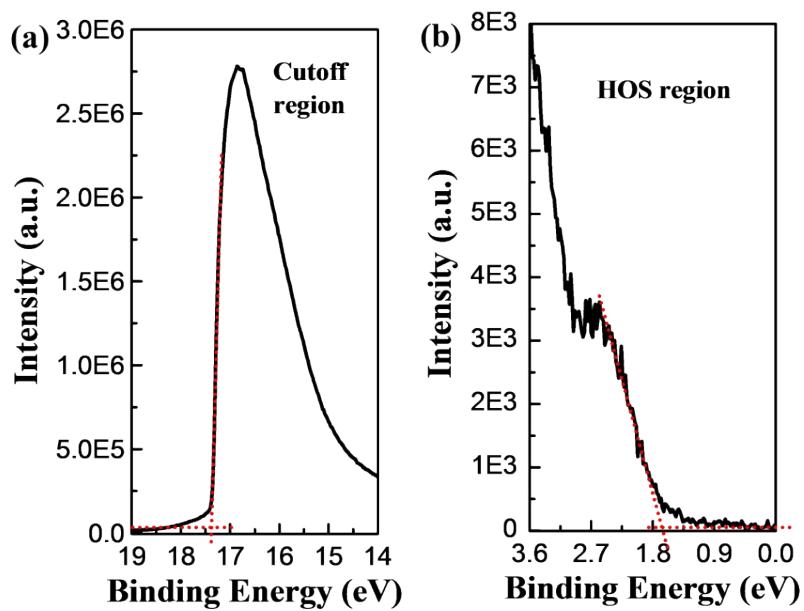


Figure S2. Ultraviolet photoelectron spectroscopy spectra of OPDPO in (a) the cutoff region and (b) the highest-occupied states (HOS) region. The dotted red lines in (a) and (b) determine two energy points of 17.37 eV and 1.65 eV for OPDPO, and thus the HOMO of OPDPO can be determined by subtracting the energy separation by 21.22 eV and calculated to be $21.22 - (17.37 - 1.65) = 5.5$ (eV).

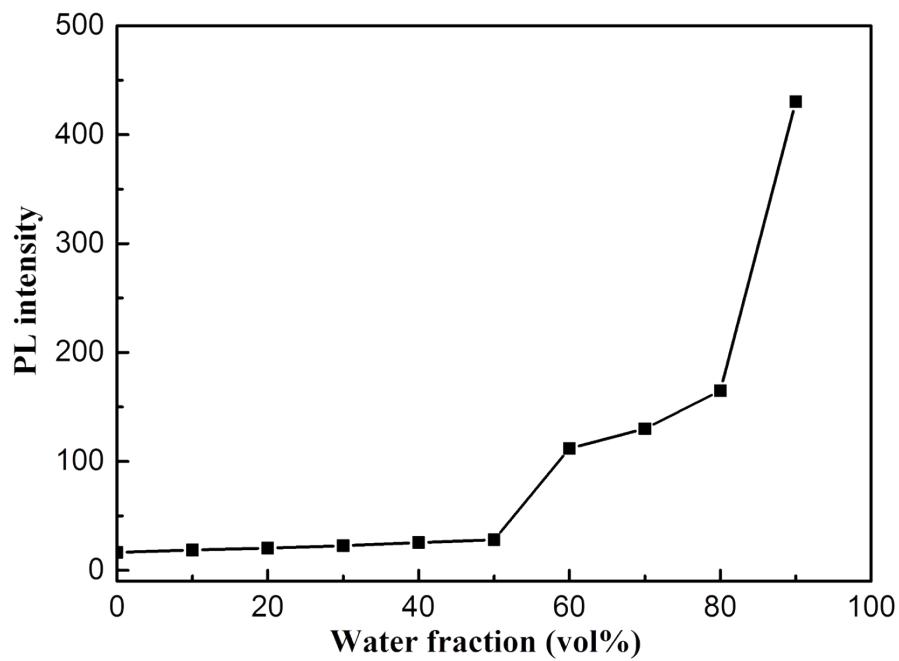


Figure S3. PL intensity of 2CzTPEPCz versus the water fraction in THF/water mixture.

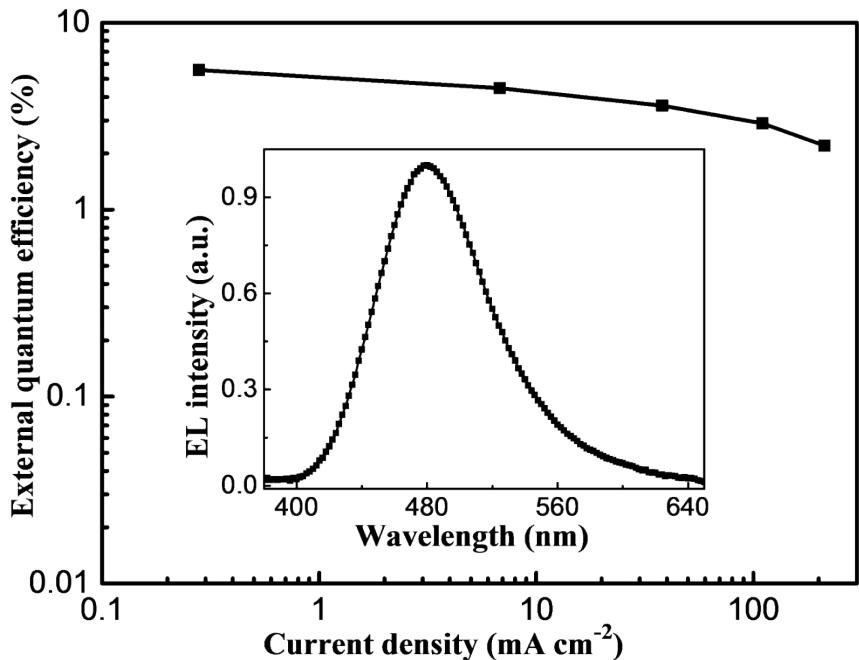
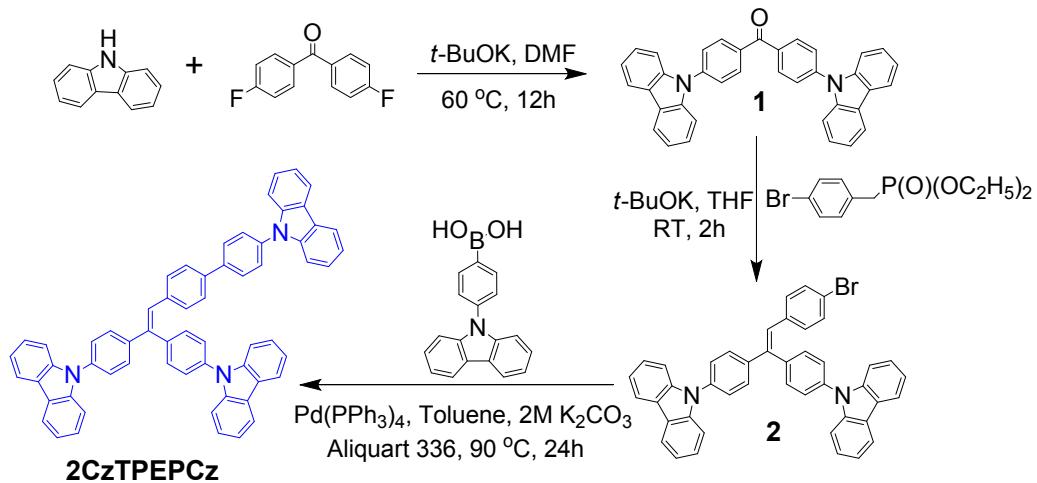


Figure S4. EQE–luminance curves of device B1 with a device configuration as: ITO/PEDOT:PSS(40 nm)/CBP(20 nm)/2CzTPEPCz (15 nm)/TPBI(40 nm)/Mg:Ag. Inset: the normalized EL intensity of device B1.

Synthesis of 2CzTPEPCz

Scheme S1 shows the synthetic route for **2CzTPEPCz**, wherein compound **1** and compound **2** were synthesized according to our previous procedures.^[3]



Scheme S1. Synthetic route for **2CzTPEPCz**.

A 2.0 M aqueous K₂CO₃ solution (3.0 mL) was added to a toluene solution (20 mL) containing compound **2** (1.33 g, 2.0 mmol) and [4-(9H-carbazol-9-yl)phenyl]boronic acid (0.58 g, 2.0 mmol) and stirred for 30 mins under an argon atmosphere. Then,

Pd(PPh₃)₄ catalyst (0.05 g) was added to the reaction mixture and stirred at 90 °C for 24 h. After cooling down to room temperature, the product was extracted with dichloromethane/water. The organic layers were collected, dried over Na₂SO₄, concentrated using a rotary evaporator and purified by silica gel column chromatography (dichloromethane/n-hexane, gradient elution from 1:5 to 1:1 v/v) to yield 1.39 g white powder (84% yield). ¹H NMR (500 MHz, DMSO-*d*₆) δ = 8.26 (dd, *J* = 12.4, 5.1, 6H), 7.97 (d, *J* = 8.3, 2H), 7.78 (t, *J* = 7.3, 4H), 7.71 (dd, *J* = 16.5, 8.1, 6H), 7.62 (d, *J* = 8.1, 2H), 7.56 – 7.40 (m, 13H), 7.37 – 7.26 (m, 8H). ¹³C NMR (125 MHz, deuterated chloroform (CDCl₃)) δ = 141.87, 141.38, 140.86, 140.79, 140.76, 139.51, 139.21, 138.89, 137.37, 137.29, 137.05, 136.56, 132.09, 130.36, 129.04, 128.97, 128.29, 127.45, 127.40, 126.89, 126.83, 126.06, 123.60, 123.56, 123.50, 120.50, 120.43, 120.40, 120.19, 120.15, 120.07, 110.04, 109.91, 109.86, 109.83. HRMS, *m/z*: [M]⁺ 827.3291; calcd. for C₆₂H₄₁N₃, 827.3295. Anal. Calcd. for C₆₂H₄₁N₃: C 89.93%, H 4.99%, N 5.07%; found: C 89.90%, H 5.01%, N 5.03%.

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