

Supporting Information

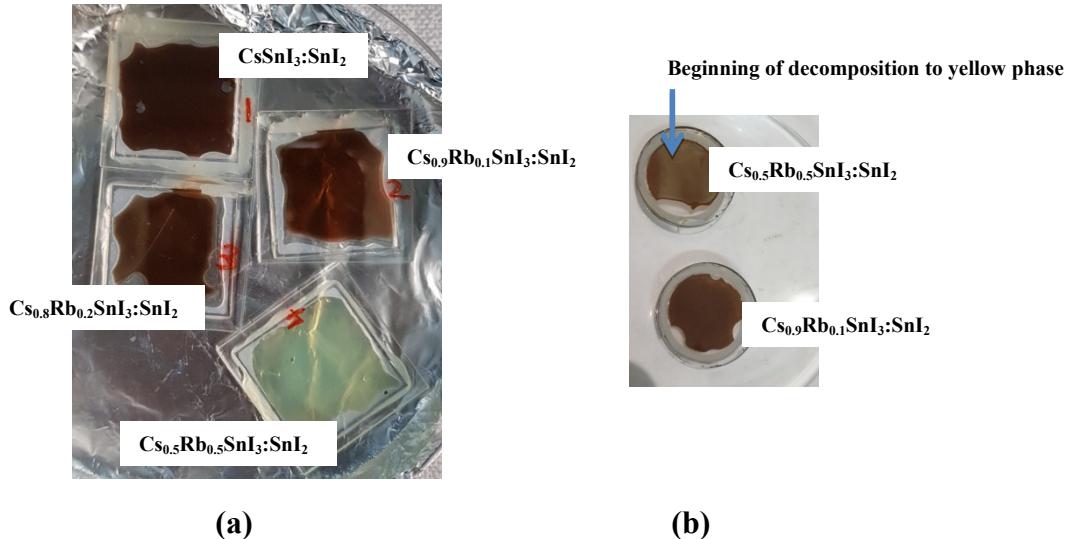


Figure S1: (a) Photograph of $\text{Cs}_{1-x}\text{Rb}_x\text{SnI}_3$ films ($x = 0, 0.1, 0.2, 0.5$) spin cast from 8 wt.% DMF solutions and encapsulated under nitrogen. Notably, even with encapsulation $\text{Cs}_{0.5}\text{Rb}_{0.5}\text{SnI}_3$ perovskite is unstable, decomposing to the yellow phase within hours/days.; (b) Photograph of $\text{Cs}_{0.9}\text{Rb}_{0.1}\text{SnI}_3$ and $\text{Cs}_{0.5}\text{Rb}_{0.5}\text{SnI}_3$ immediately after perovskite deposition and encapsulation under nitrogen. For the latter the start of degradation to the yellow phase is evident. All perovskite films were deposited with 10 mol% SnI_2 as a source of excess Sn.

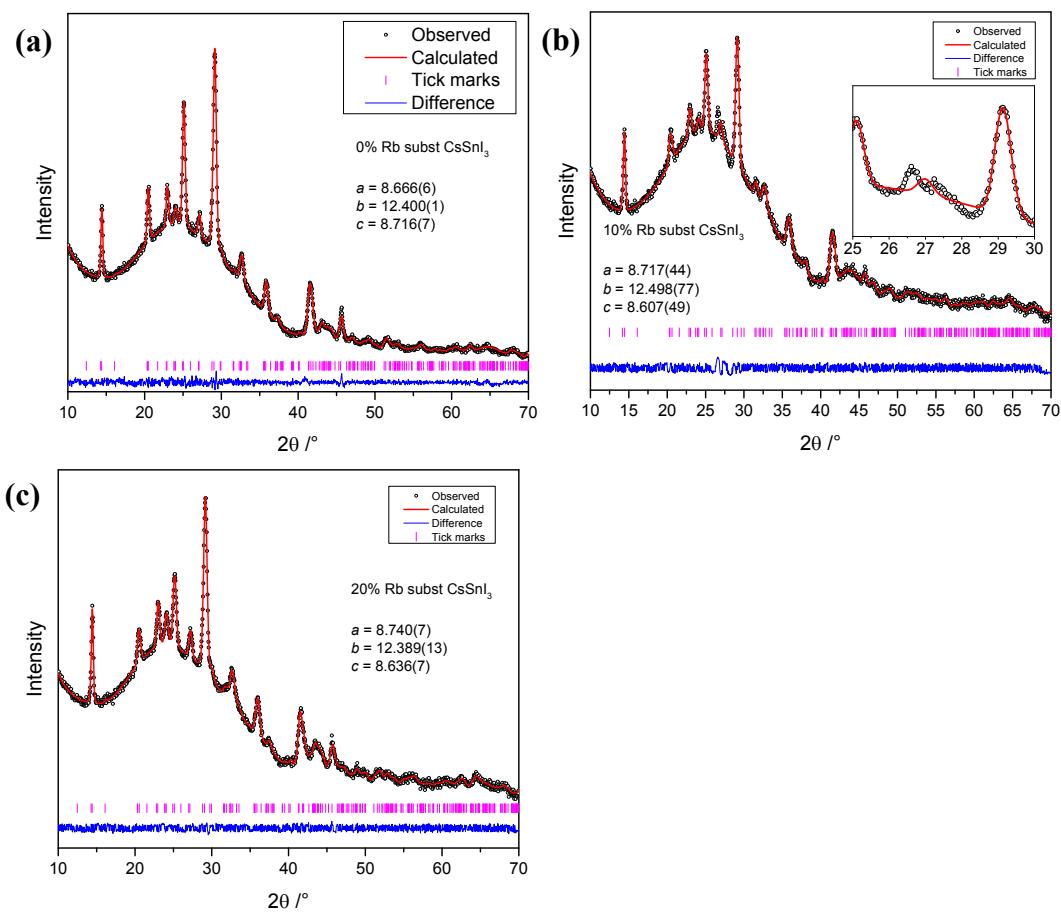


Figure S2: Lattice parameter fitting using space group *Pnma* of XRD patterns of $\text{Cs}_{1-x}\text{Rb}_x\text{SnI}_3$ where $x = 0$ **(a)**, 0.1 **(b)** (inset shows presence of possible impurity phase), or 0.2 **(c)**. The lattice parameters shown on the graphs have units of Ångströms.

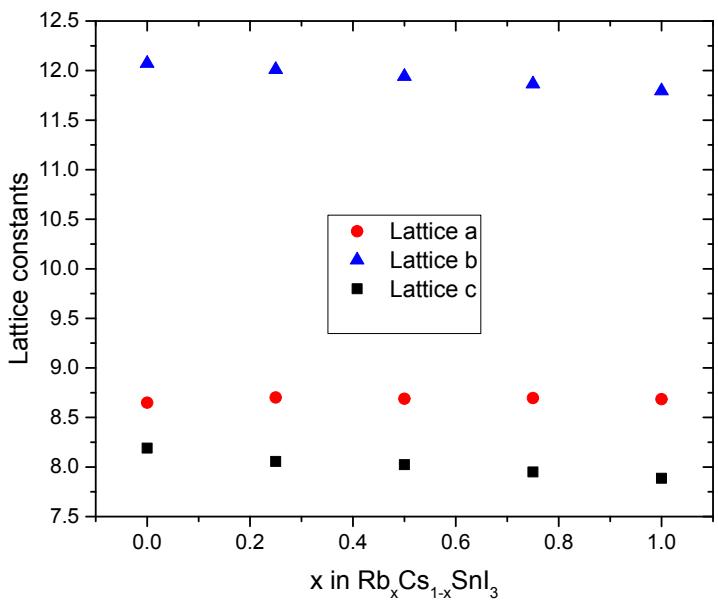


Figure S3: Calculated lattice constants evolution with increasing Rb in orthorhombic $\text{Cs}_{1-x}\text{Rb}_x\text{SnI}_3$ (in Å).

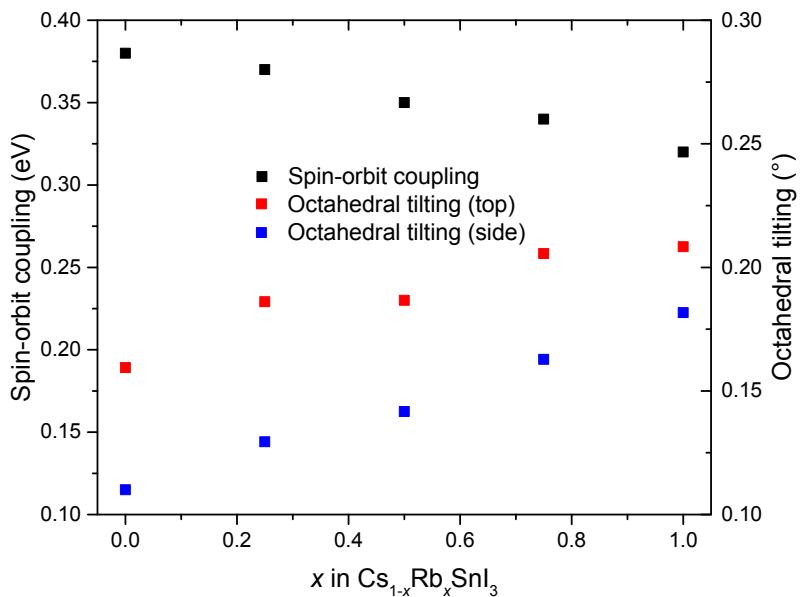


Figure S4: Calculated average octahedral tilting angles and the corresponding spin-orbit coupling effect derived from DFT-1/2 calculated band gaps with and without the spin-orbit coupling effect.

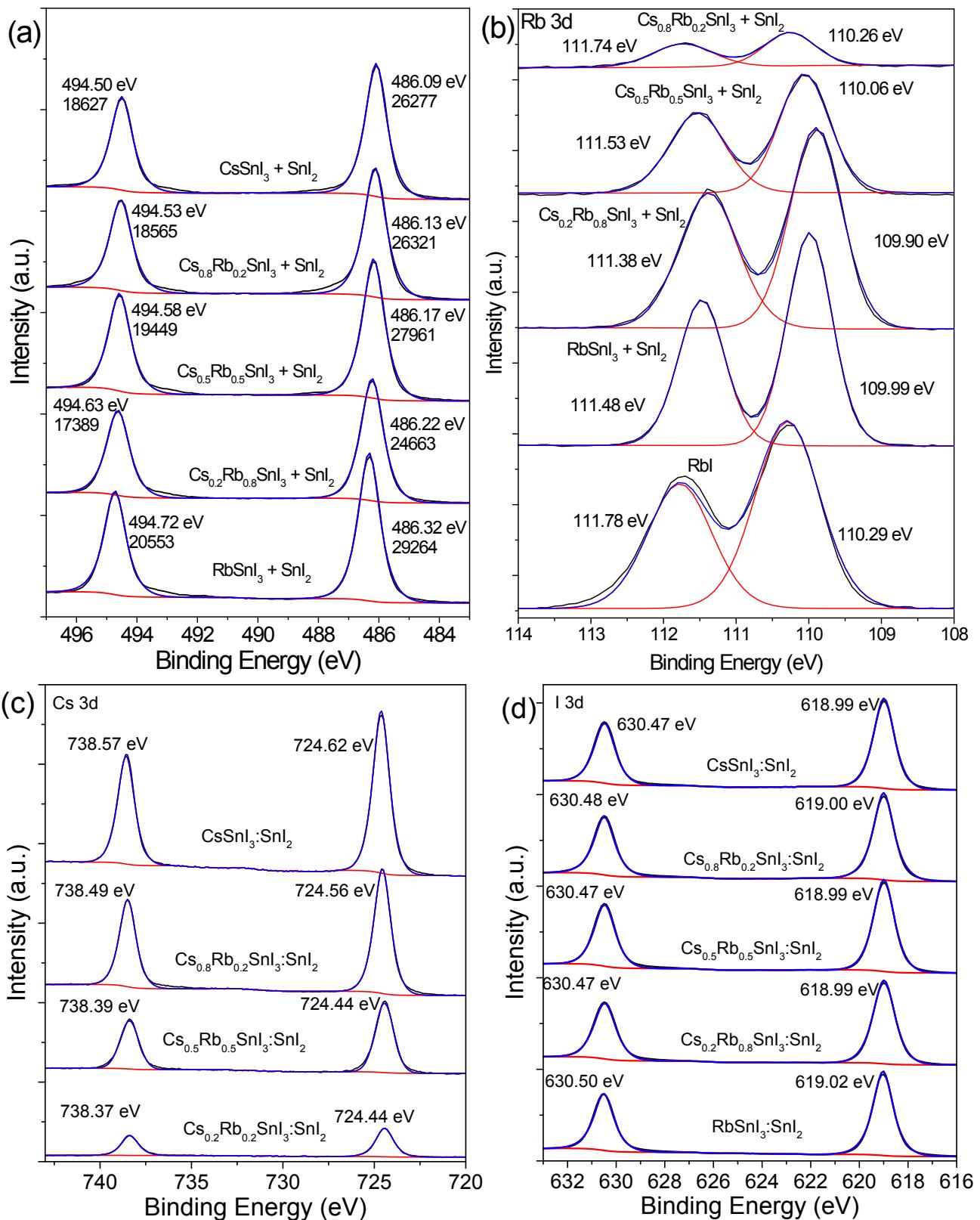


Figure S5: XPS spectra of $\text{Cs}_{1-x}\text{Rb}_x\text{SnI}_3 + 10$ mol% SnI_2 where $x = 0, 0.2, 0.5, 0.8$, and 1 , showing **(a)** Sn 3d; **(b)** Rb 3d; **(c)** Cs 3d; **(d)** I 3d. Also shown in (b) is the binding energy for Rb 3d in RbI.

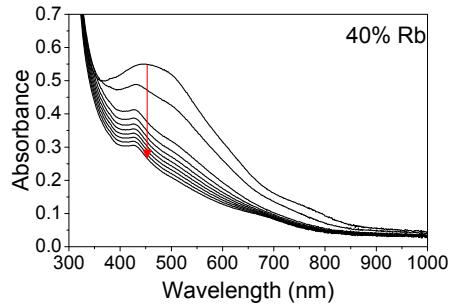


Figure S6: Electronic absorption spectra of $\text{Cs}_{0.6}\text{Rb}_{0.4}\text{SnI}_3:\text{SnI}_2$ taken over time exposed to ambient air. Measurements were taken every 5 minutes

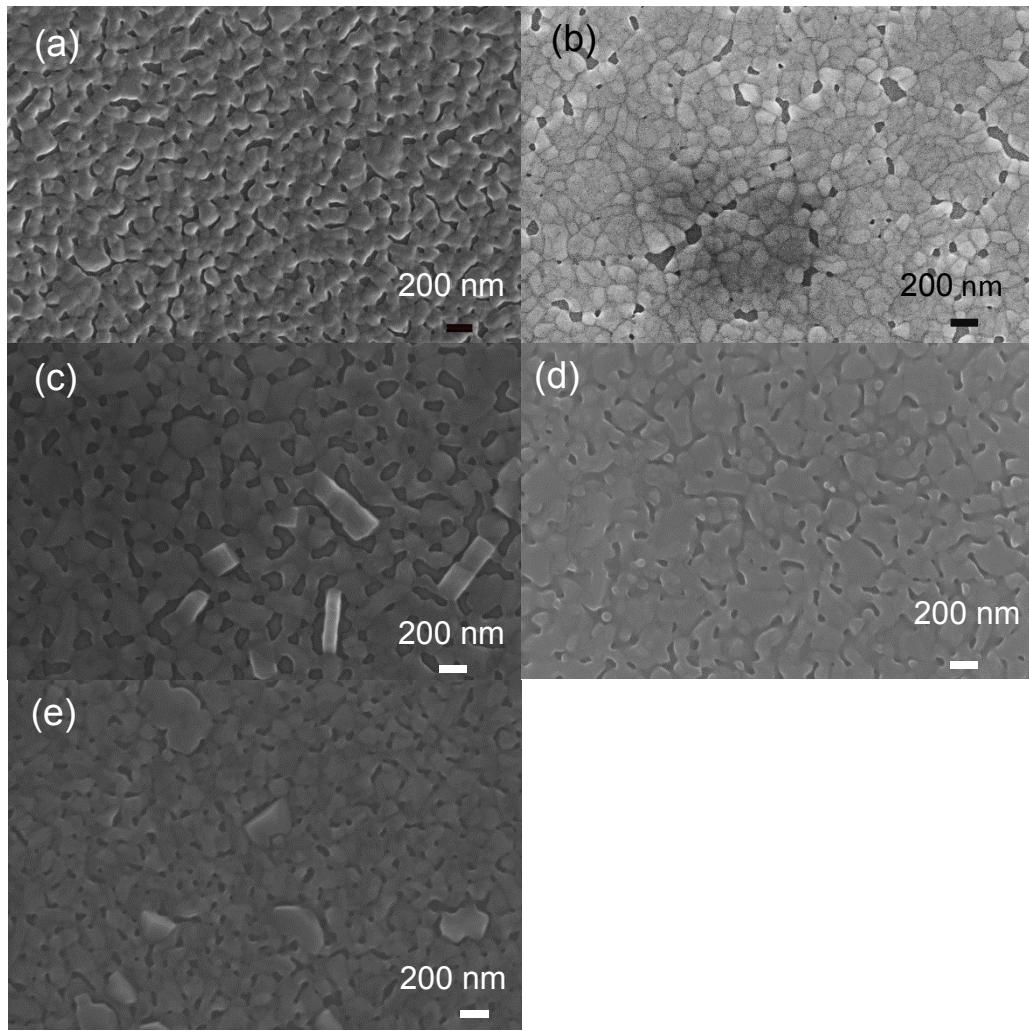


Figure S7: SEM image of a film of: (a) $\text{Cs}_{0.8}\text{Rb}_{0.2}\text{SnI}_3:\text{SnI}_2$ ($\sim 6\%$) (b) $\text{CsSnI}_3:\text{SnI}_2$ ($\sim 1.6\%$) (c) $\text{Cs}_{0.5}\text{Rb}_{0.5}\text{SnI}_3:\text{SnI}_2$ (15.9%) (d) $\text{Cs}_{0.8}\text{Rb}_{0.2}\text{SnI}_3:\text{SnCl}_2$, (4.3%) and (e) $\text{Cs}_{0.5}\text{Rb}_{0.5}\text{SnI}_3:\text{SnCl}_2$ (6.8%) on an ITO glass substrates. For each image the values in brackets correspond to the area of the substrate not covered by the perovskite film.

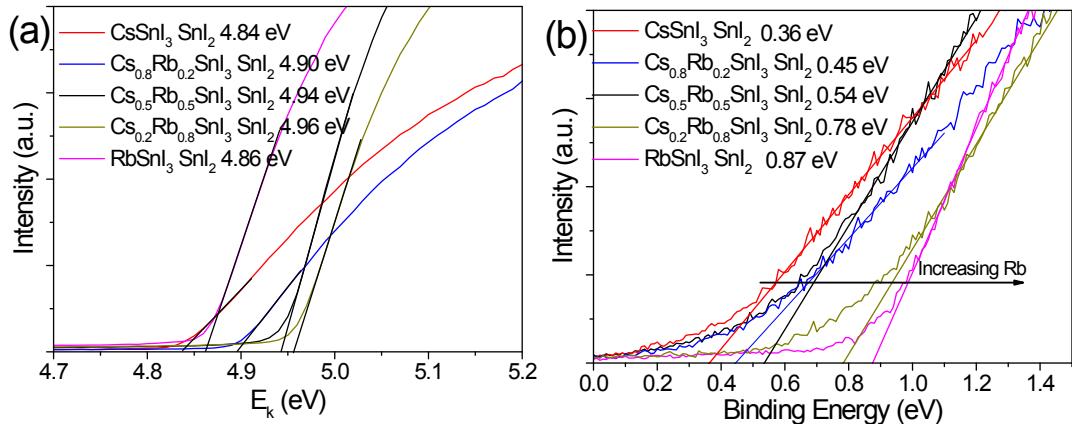


Figure S8: UPS of $\text{Cs}_{1-x}\text{Rb}_x\text{SnI}_3$, where $x = 0, 0.2, 0.5, 0.8$ and 1 , showing **(a)** low kinetic energy cut-off and **(b)** high kinetic energy cut-off. The ionisation potential is equal to the sum of the work function (deduced from the secondary electron cut-off in (a)) and the difference in energy between the Fermi level (binding energy = 0) and valance band edge deduced from (b).

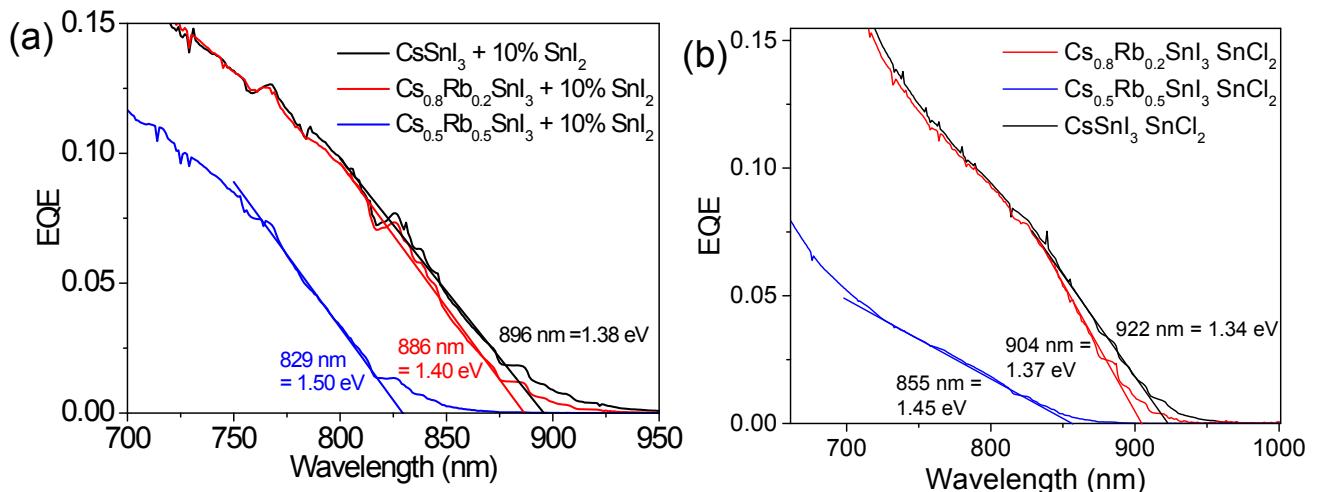


Figure S9: Low energy edge of EQE spectra shown in Figure 7(b) and 8(b) of main manuscript: **(a)** ITO| $\text{Cs}_{1-x}\text{Rb}_x\text{SnI}_3:\text{SnI}_2$ | C₆₀| BCP| Al, and **(b)** ITO| $\text{Cs}_{1-x}\text{Rb}_x\text{SnI}_3:\text{SnCl}_2$ | PC₆₁BM| BCP| Al.

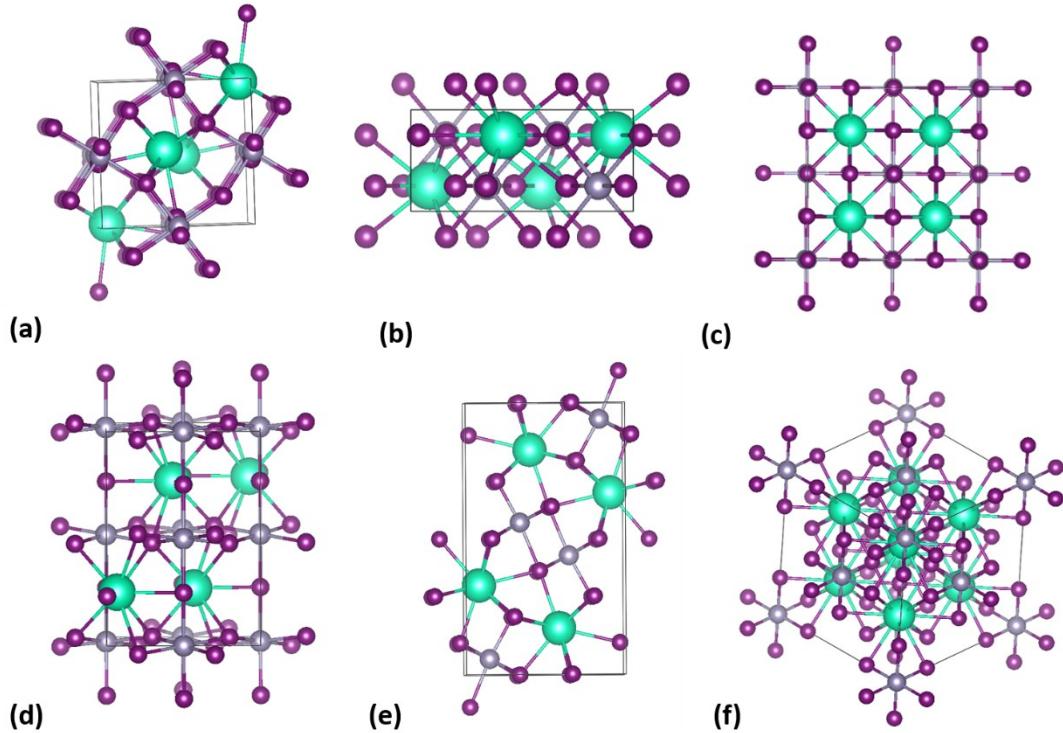


Figure S10: Top (a, b, and c) and side (e, f, and g (45 degree tilting)) views of orthorhombic, yellow phase CsSnI_3 and FCC Cs_2SnI_6 . Figures are made using VESTA^a. Purple, green, and grey spheres denote I, Cs and Sn, respectively.

Table S1: Relative proportions of Cs and Rb estimated from the XPS $3\text{d}_{5/2} + 3\text{d}_{3/2}$ peak intensities using the relative sensitivity factors for Cs and Rb 3d peaks, 40.22 and 4.21 respectively.

Sample	Cs / %	Rb / %
$\text{Cs}_{0.8}\text{Rb}_{0.2}\text{SnI}_3:\text{SnI}_2$	80.1	19.9
$\text{Cs}_{0.5}\text{Rb}_{0.5}\text{SnI}_3:\text{SnI}_2$	43.5	56.5
$\text{Cs}_{0.2}\text{Rb}_{0.8}\text{SnI}_3:\text{SnI}_2$	15.0	85.0

Table S2: Summary of JV characteristics of PV devices made using $\text{CsSnI}_3 + 10 \text{ mol\% SnI}_2$ with 0, 20% or 50% Cs substituted for Rb, and a C_{60} ETL.

Sample	n	J_{sc} (mA/cm ²)	V_{oc} (V)	FF	η (%)	Champion η (%)
$\text{CsSnI}_3:\text{SnI}_2$	18	9.84 ± 0.67	0.31 ± 0.01	0.46 ± 0.02	1.37 ± 0.14	1.55
$\text{Cs}_{0.8}\text{Rb}_{0.2}\text{SnI}_3:\text{SnI}_2$	17	8.79 ± 0.25	0.43 ± 0.02	0.45 ± 0.01	1.70 ± 0.09	1.84
$\text{Cs}_{0.5}\text{Rb}_{0.5}\text{SnI}_3:\text{SnI}_2$	18	8.11 ± 0.53	0.48 ± 0.04	0.46 ± 0.05	1.81 ± 0.30	2.25

Table S3: JV characteristics of PPV devices in the structure ITO| Cs_{1-x}Rb_xSnI₃ + 10 mol% SnCl₂| PC₆₁BM| BCP| Al, immediately after fabricating (black), after 5 days (red) and 12 days (blue) storage in an N₂ filled glovebox.

Sample	n	J_{sc} (mA/cm ²)	V_{oc} (V)	FF	η (%)	Champion η (%)
CsSnI ₃ :SnCl ₂	16	5.32 ± 0.78	0.37 ± 0.01	0.57 ± 0.04	1.14 ± 0.24	1.50
	15	5.30 ± 0.79	0.41 ± 0.02	0.61 ± 0.02	1.33 ± 0.26	1.67
	14	5.33 ± 0.97	0.44 ± 0.01	0.62 ± 0.03	1.46 ± 0.34	1.98
Cs _{0.8} Rb _{0.2} SnI ₃ : SnCl ₂	17	6.71 ± 0.71	0.50 ± 0.01	0.61 ± 0.03	2.05 ± 0.26	2.58
	16	6.42 ± 0.63	0.48 ± 0.09	0.60 ± 0.14	1.93 ± 0.64	2.64
	16	6.75 ± 0.68	0.48 ± 0.02	0.67 ± 0.02	2.19 ± 0.20	2.45
Cs _{0.5} Rb _{0.5} SnI ₃ : SnCl ₂	18	2.18 ± 0.17	0.53 ± 0.02	0.44 ± 0.03	0.50 ± 0.07	0.63
	16	2.22 ± 0.17	0.54 ± 0.01	0.58 ± 0.01	0.70 ± 0.06	0.79
	17	2.31 ± 0.17	0.53 ± 0.01	0.60 ± 0.02	0.74 ± 0.07	0.90