

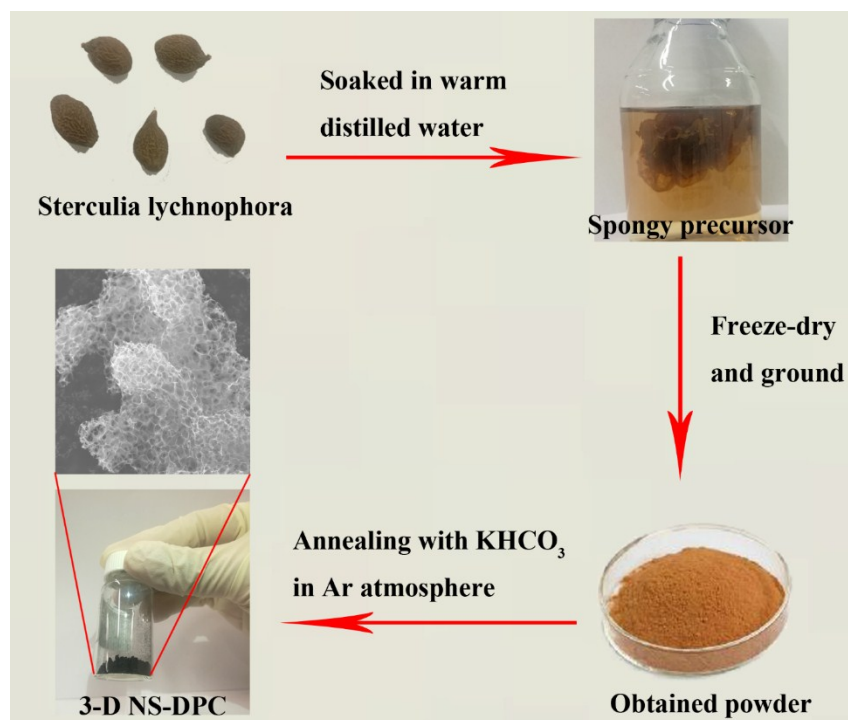
A Novel Lithium-Ion Hybrid Capacitor based on the Aerogel-like MXene Wrapped Fe_2O_3 Nanosphere Anode and the 3D Nitrogen Sulphur Dual-Doped Porous Carbon Cathode

Xiao Tang ^a, Hao Liu ^a, Xin Guo ^a, Shijian Wang ^b, Wenjian Wu ^b, Anjon Kumar Mondal ^a, Chengyin Wang ^c, Guoxiu Wang ^{a,}*

^a Centre for Clean Energy Technology, Faculty of Science, University of Technology Sydney, NSW 2007, Sydney, Australia.

^b Department of Materials Science and Engineering, Dongguan University of Technology, Dongguan, China.

^c College of Chemistry and Chemical Engineering, Yangzhou University, Yangzhou, China.



Scheme S1. Schematic illustration of the synthesis procedure for the three-dimensional nitrogen sulphur dual-doped porous carbon.

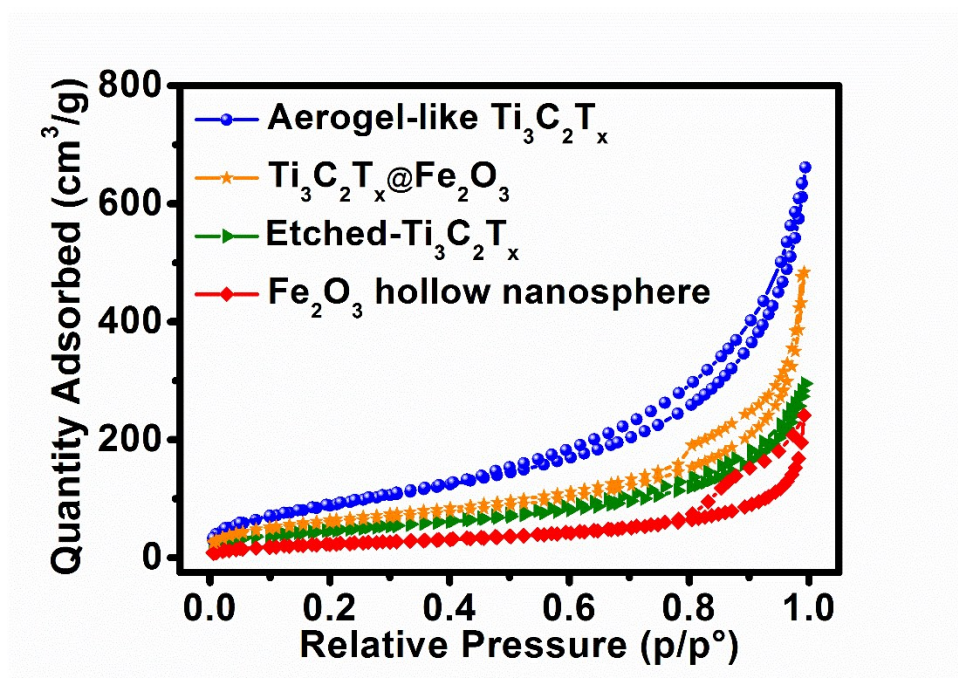


Figure S1. The BET surface area of Aerogel-like $\text{Ti}_3\text{C}_2\text{T}_x$, $\text{Ti}_3\text{C}_2\text{T}_x@\text{Fe}_2\text{O}_3$, Etched- $\text{Ti}_3\text{C}_2\text{T}_x$ and Fe_2O_3 hollow nanospheres.

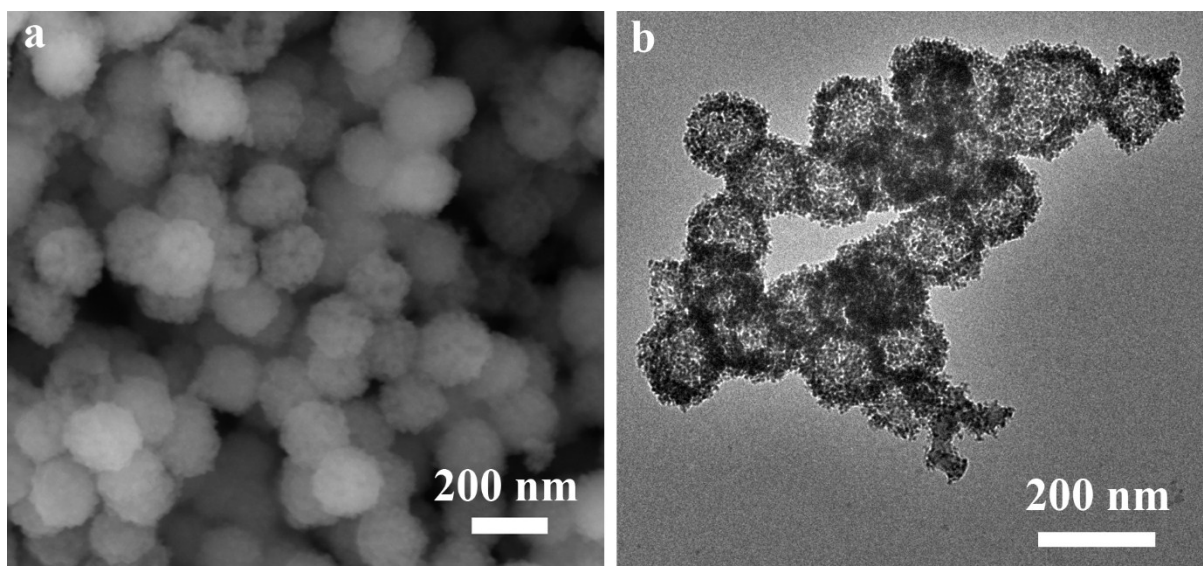


Figure S2. a) SEM image and b) TEM image of Fe_2O_3 hollow nanospheres.

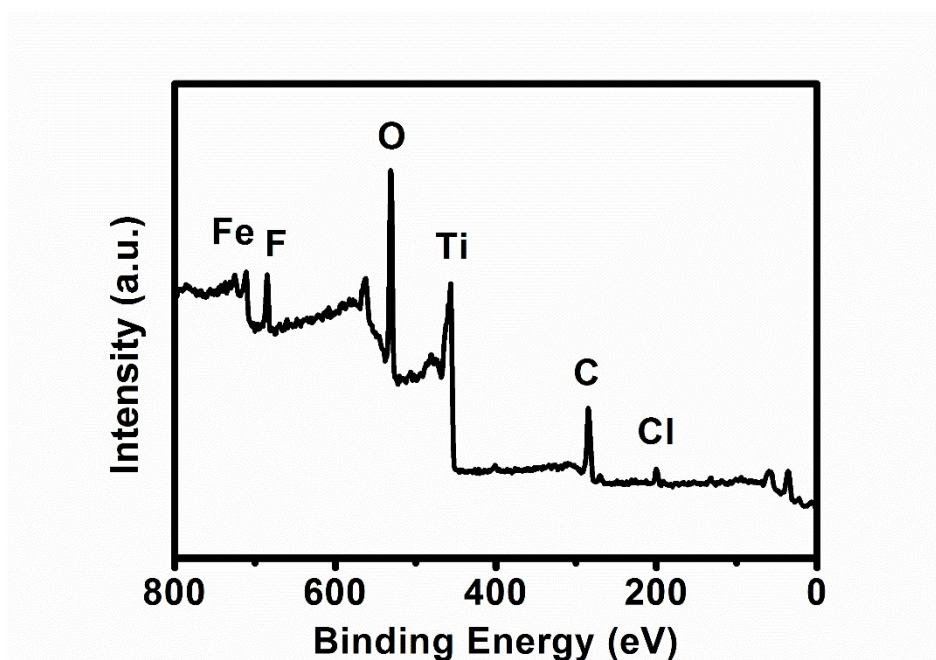


Figure S3. Survey-scan XPS spectrum of $\text{Ti}_3\text{C}_2\text{T}_x@\text{Fe}_2\text{O}_3$ nanocomposite.

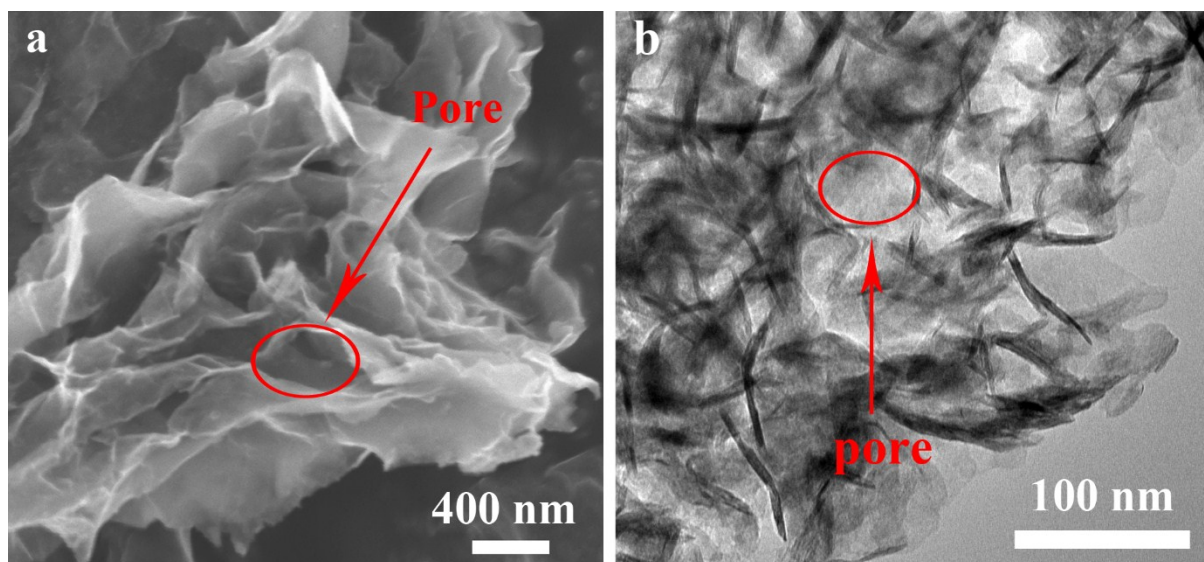


Figure S4. a) SEM image and b) TEM image of the aerogel-like Ti₃C₂T_x MXene.

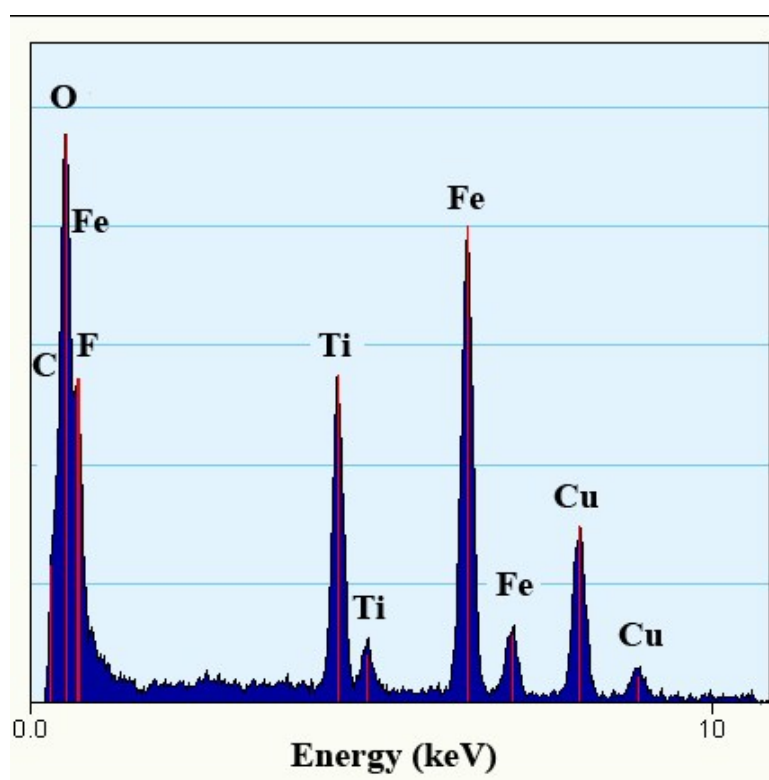


Figure S5. EDS spectrum of the $\text{Ti}_3\text{C}_2\text{T}_x@\text{Fe}_2\text{O}_3$ composite.

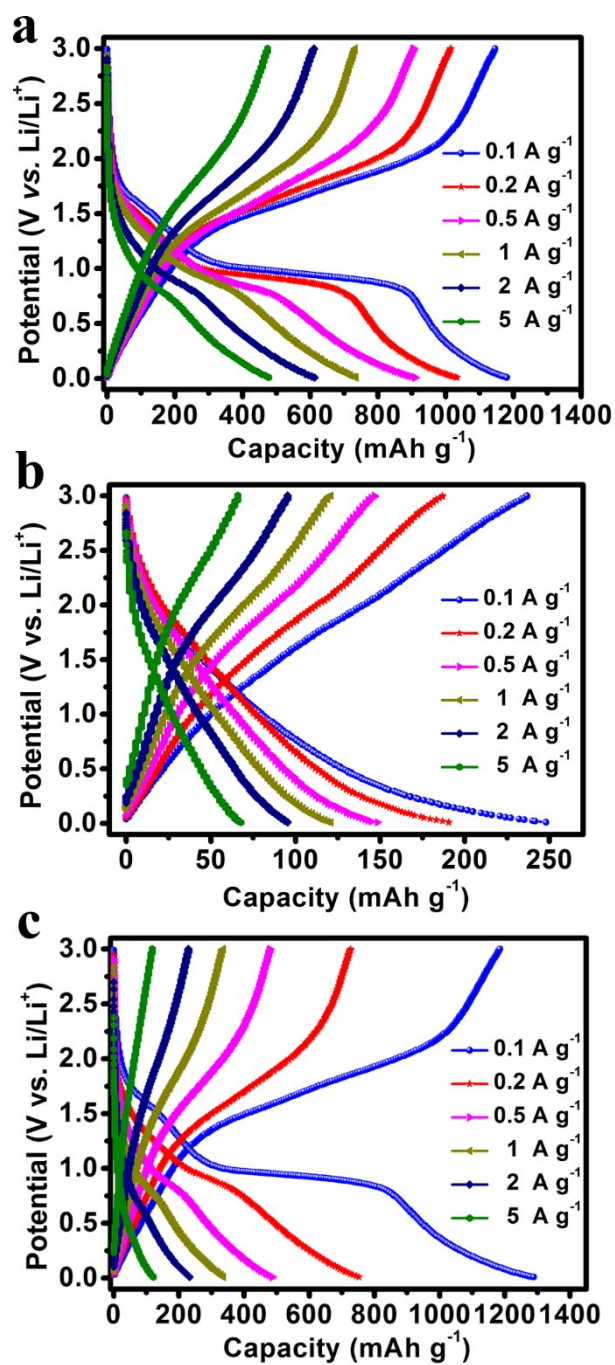


Figure S6. The charge-discharge curves of a) the $\text{Ti}_3\text{C}_2\text{T}_x@\text{Fe}_2\text{O}_3$ anode, b) the aerogel-like MXene anode and c) the pristine Fe_2O_3 anode at different current density.

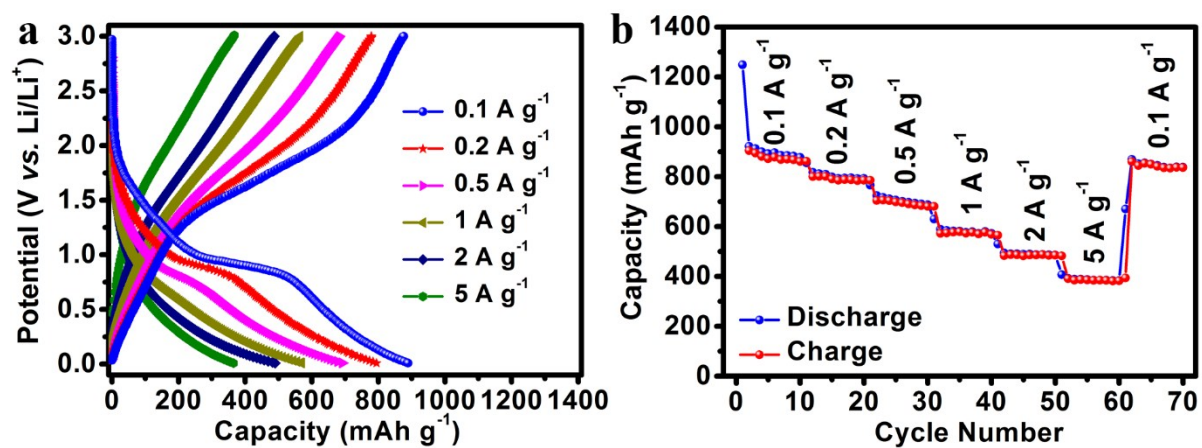


Figure S7. (a) Charge/discharge curves of a Ti₃C₂T_x@Fe₂O₃ (1:1) anode at different current densities. (b) Rate performance of the Ti₃C₂T_x@Fe₂O₃ (1:1) anode.

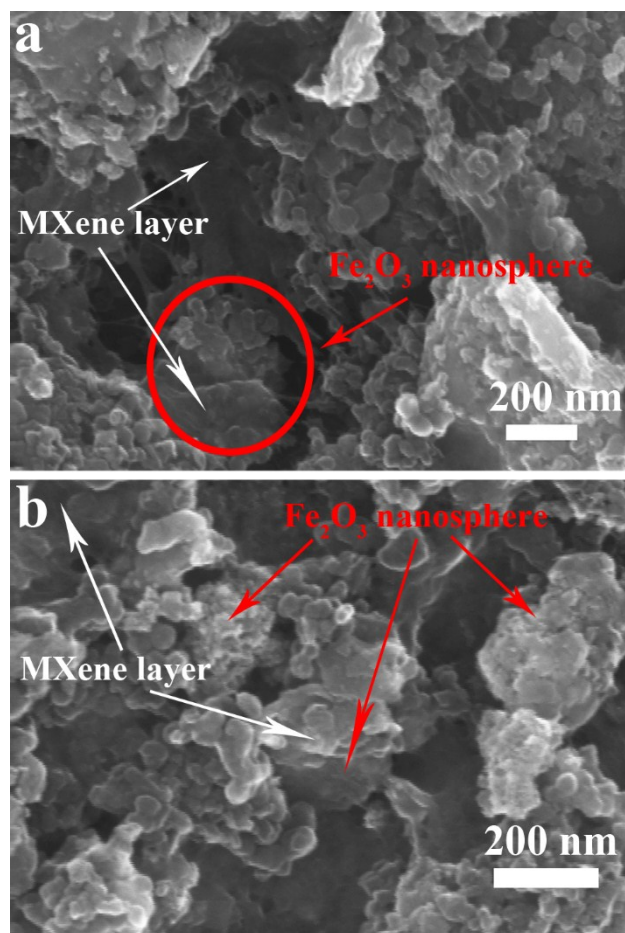


Figure S8. SEM images of $\text{Ti}_3\text{C}_2\text{T}_x@ \text{Fe}_2\text{O}_3$ electrode after 500 cycles.

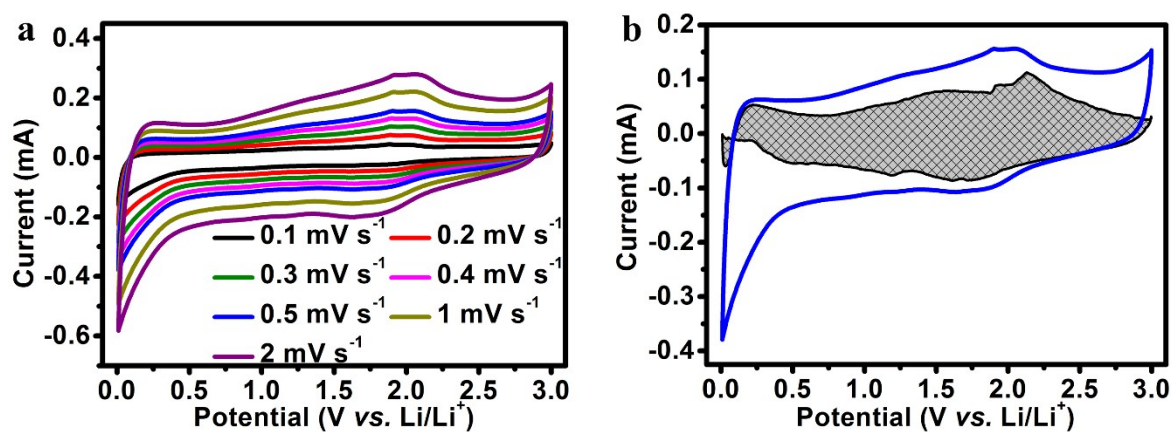


Figure S9. a) The CV curves of the aerogel-like $\text{Ti}_3\text{C}_2\text{T}_x$ MXene anode at different scan rates; b) the capacity separation of the aerogel-like $\text{Ti}_3\text{C}_2\text{T}_x$ MXene anode at the scan rate of 0.5 mV s^{-1} .

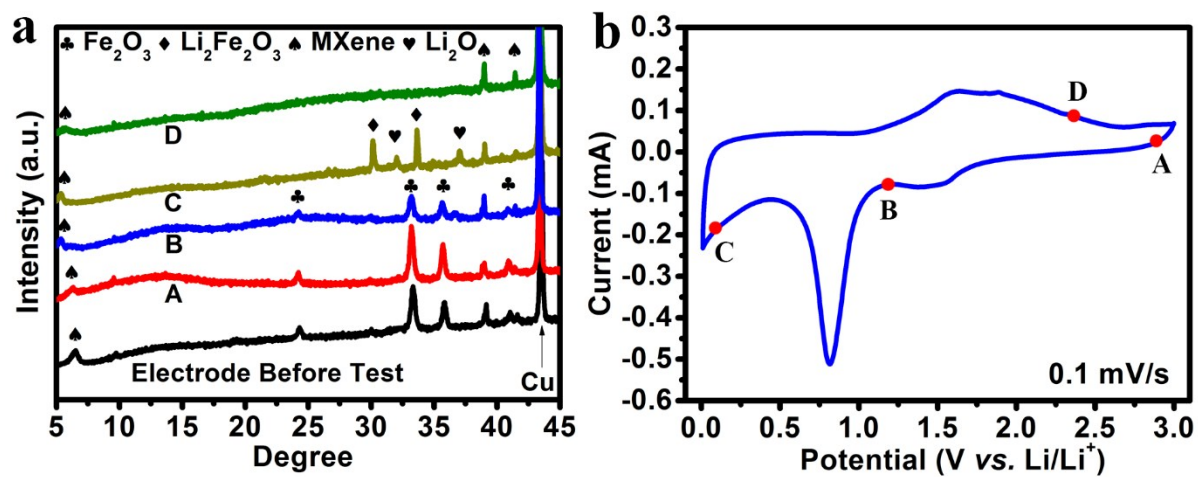


Figure S10. a) *Ex situ* XRD patterns and b) corresponding CV curve collected at a scan rate of 0.1 mV s^{-1} of the hybrid $\text{Ti}_3\text{C}_2\text{T}_x@\text{Fe}_2\text{O}_3$ electrode.

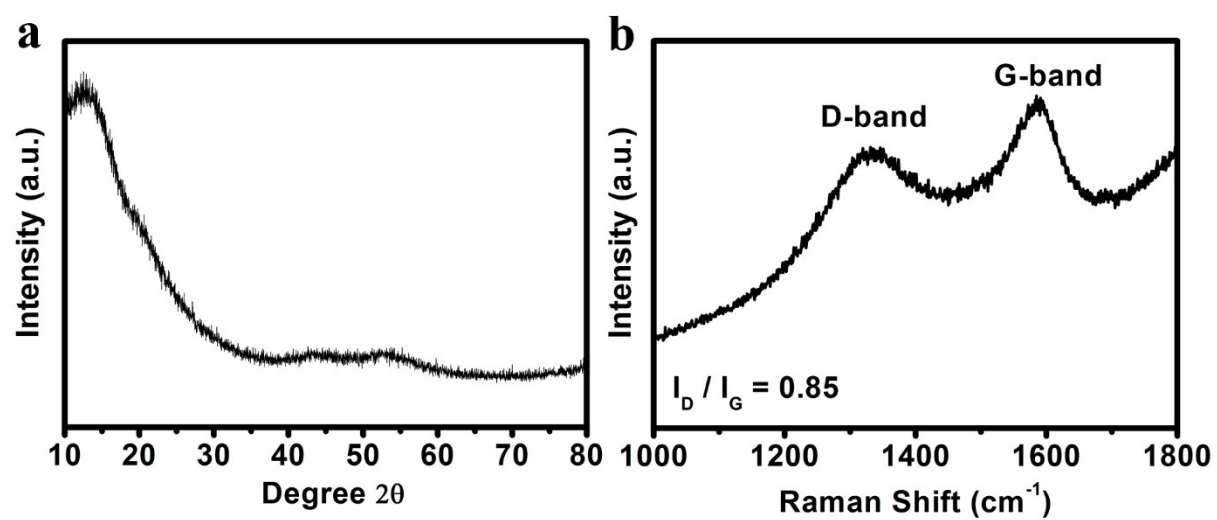


Figure S11. a) XRD pattern and b) Raman spectrum of as-prepared NS-DPC.

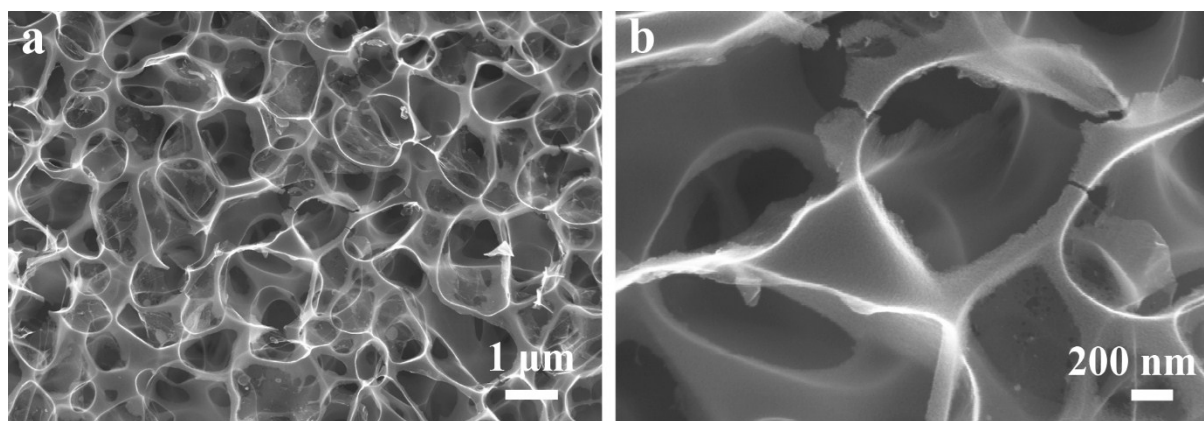


Figure S12. a) Low-resolution SEM and b) high-resolution SEM of nitrogen and sulphur dual-doped porous carbon.

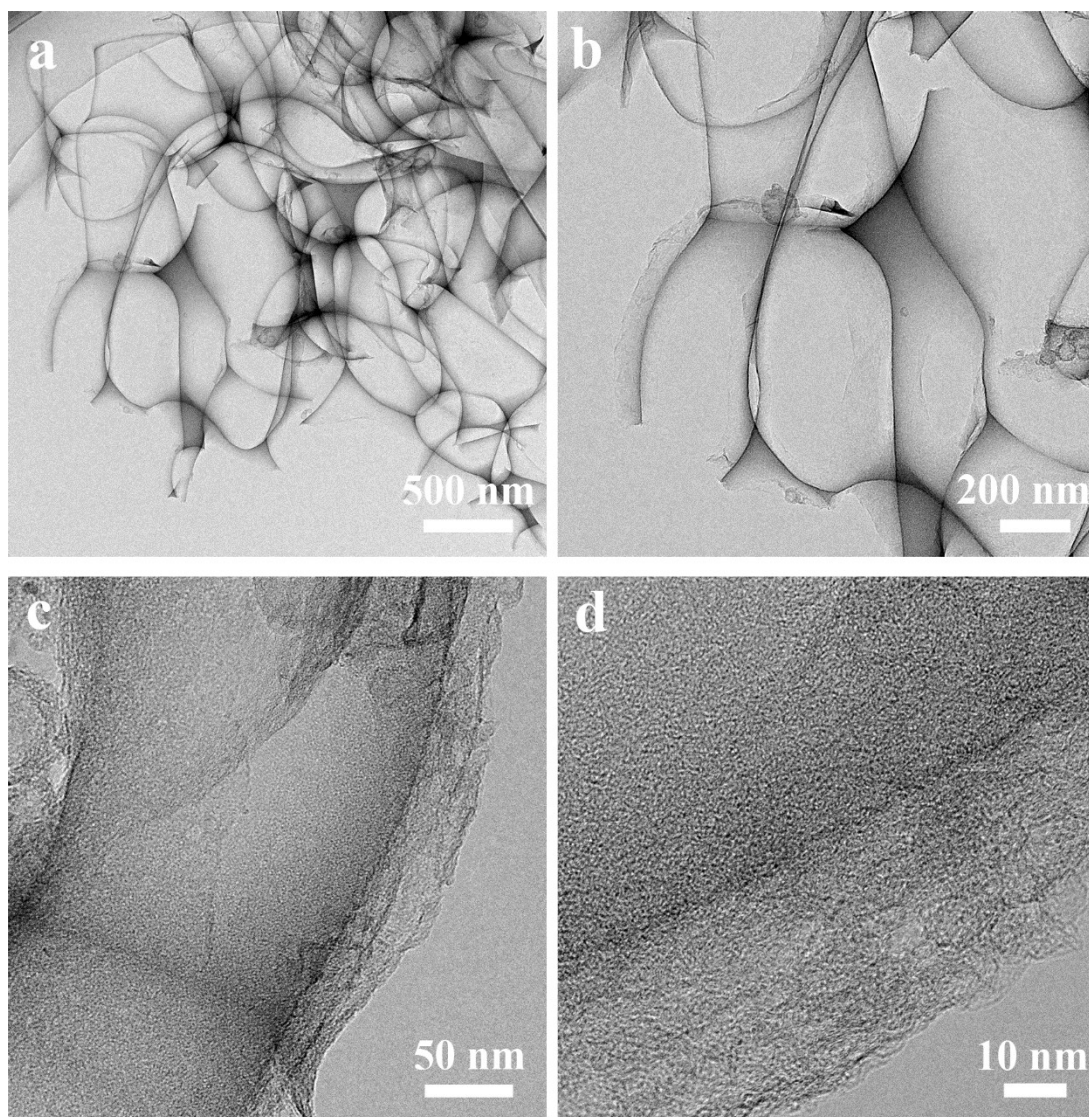


Figure S13. a, b) TEM images and c, d) HRTEM images of NS-DPC.

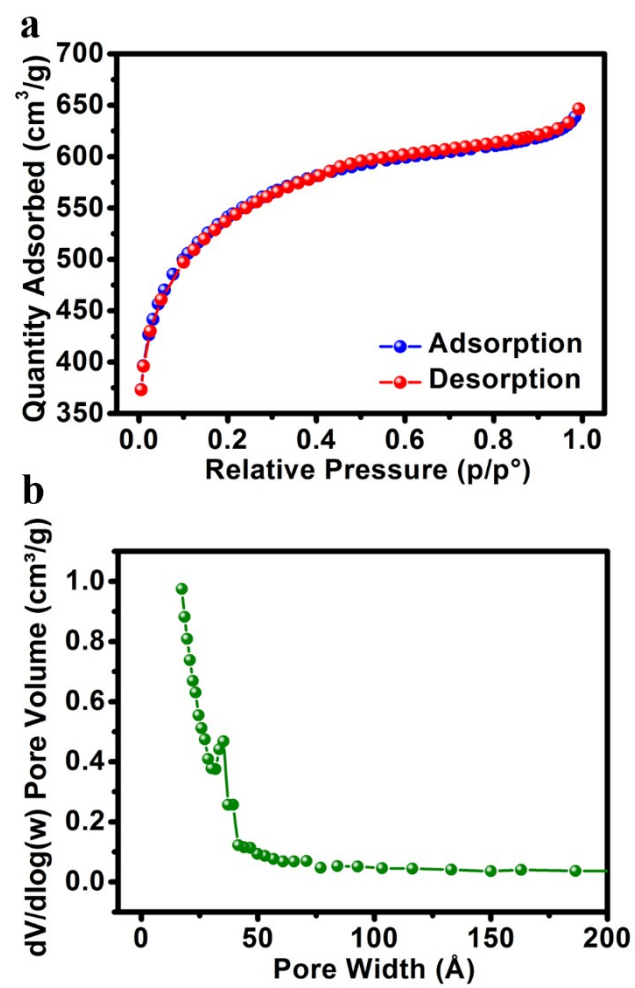


Figure S14. a) The BET surface area of NS-DPC sample. b) Pore size distribution of NS-DPC sample.

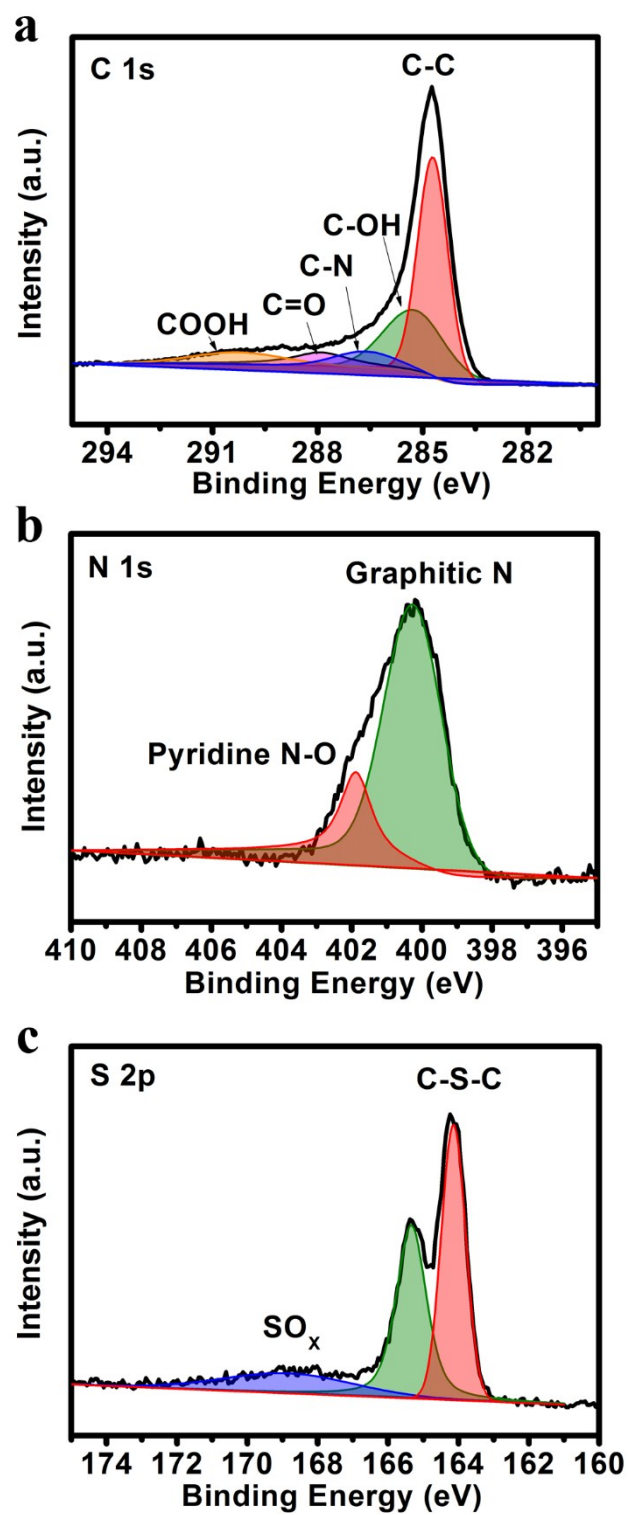


Figure S15. High-resolution a) C 1s, b) N 1s and c) S 2p XPS spectrum of NS-DPC.

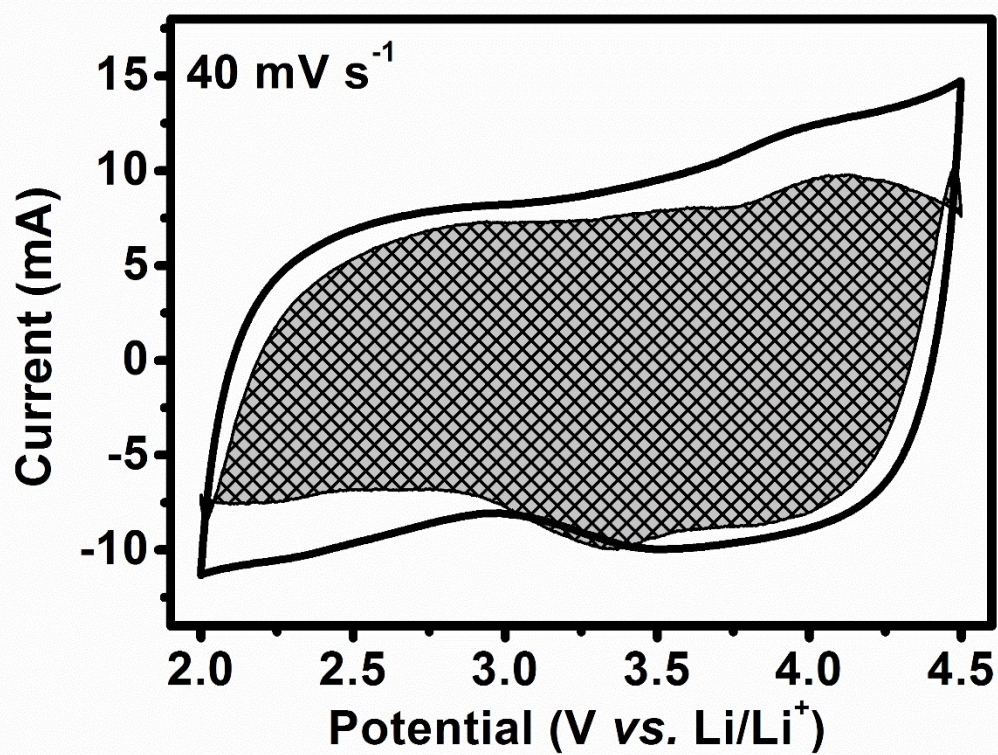


Figure S16. Capacity separation of nitrogen and sulphur dual-doped porous carbon cathode at the scan rate of 40 mV s⁻¹.

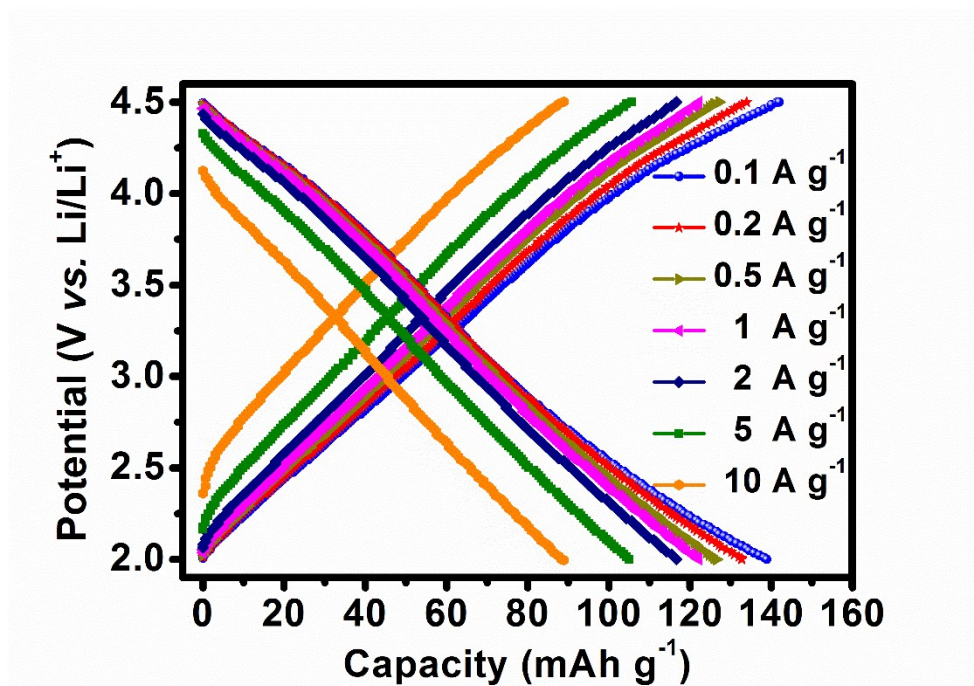


Figure S17. Charge-discharge curves of a NS-DPC cathode tested in LiPF_6 electrolyte at different current densities.

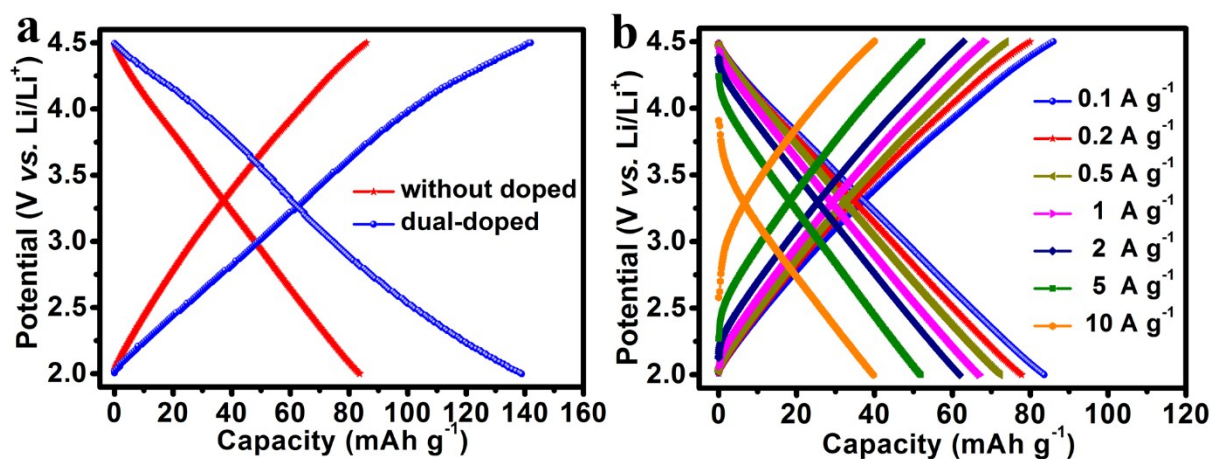


Figure S18. (a) The charge-discharge curves of dual-doped porous carbon cathode and porous carbon cathode without doping (average pore diameter $100 \text{ \AA} \pm 10 \text{ \AA}$ (typical), >99.95% trace metals basis (Aldrich)) at a current density of 0.1 A g^{-1} . (b) The charge-discharge curves of porous carbon without doping at different current densities.

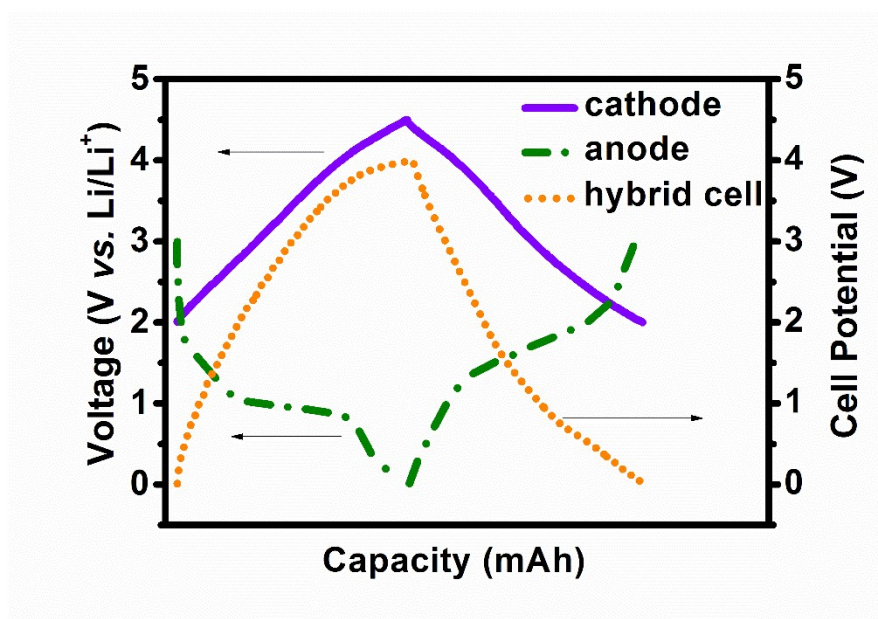


Figure S19. The mechanism of hybrid capacitor to achieve a 4 V potential window.

Table S1. Power and energy densities of $\text{Ti}_3\text{C}_2\text{T}_x@\text{Fe}_2\text{O}_3//\text{NS-DPC}$ LIC device obtained at different current density.

Current Density (A g^{-1})	0.2	0.5	1	2	5	10
Power Density (W kg^{-1})	400	1000	2000	4000	10000	20000
Energy Density (Wh Kg^{-1})	216.16	190.08	168.48	142.56	112.32	96.5

Table S2. Comparison of the electrochemical performances between the $\text{Ti}_3\text{C}_2\text{T}_x@\text{Fe}_2\text{O}_3/\text{NS}$ -DPC LIC device and previously reported LICs.

	Voltage Window	Energy Density	Power Density	Cycling Stability	Reference
graphite//AC	1.5-4.5	103.8 Wh kg ⁻¹	10 kW kg ⁻¹	85%, 10000 cycles	1
hard carbon//AC	2.0-4.2	85.7 Wh kg ⁻¹	7.6 kW kg ⁻¹	96%, 5000 cycles	2
$\text{Li}_4\text{Ti}_5\text{O}_{12}$ /graphene //AC	1-2.5	30 Wh kg ⁻¹ at 1000 W kg ⁻¹	15 kW kg ⁻¹ at 4 Wh kg ⁻¹	84%, 10000 cycles	3
TiO_2 // graphene	0.0-3.8	82 Wh kg ⁻¹ at 570 W kg ⁻¹	19 kW kg ⁻¹ at 21 Wh kg ⁻¹	73%, 600 cycles	4
$\text{Nb}_2\text{O}_5/\text{C}$ //AC	1.0-3.5	63 Wh kg ⁻¹ at 70 W kg ⁻¹	16.528 kW kg ⁻¹ at 5 Wh kg ⁻¹	nearly no fading, 1000 cycles	5
Fe_3O_4 /Graphene//Graphene	1.0-4.0	147 Wh kg ⁻¹ at 150 W kg ⁻¹	2.587 kW kg ⁻¹ at 86 Wh kg ⁻¹	70%, 1000 cycles	6
MnO/CNS//CNS	1.0-4.0	184 Wh kg ⁻¹ at 183 W kg ⁻¹	15 kW kg ⁻¹ at 90 Wh kg ⁻¹	76%, 5000 cycles	7
VN-RGO//APDC	0.0-4.0	162 Wh kg ⁻¹ at 200 W kg ⁻¹	10 kW kg ⁻¹ at 64 Wh kg ⁻¹	83%, 1000 cycles	8
TiC//PHPNC	0.0-4.5	112 Wh kg ⁻¹ at 450 W kg ⁻¹	67.5 kW kg ⁻¹ at 35.6 Wh kg ⁻¹	83%, 5000 cycles	9
NiCo_2O_4 //AC	0-4.5	39.4 Wh kg ⁻¹	554 W kg ⁻¹	no loss, 2000 cycles	10
$\text{Ti}_3\text{C}_2\text{T}_x@\text{Fe}_2\text{O}_3/\text{NS}$ -DPC	0.1-4.0	216 Wh kg ⁻¹ at 400 W kg ⁻¹	20 kW kg ⁻¹ at 96.5 Wh kg ⁻¹	87%, 5000 cycles	This work

References:

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