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Supporting Information

A facile metal-phenolic-amine strategy for dualfunctionalization of blood-contacting devices with antibacterial and anticoagulant properties†

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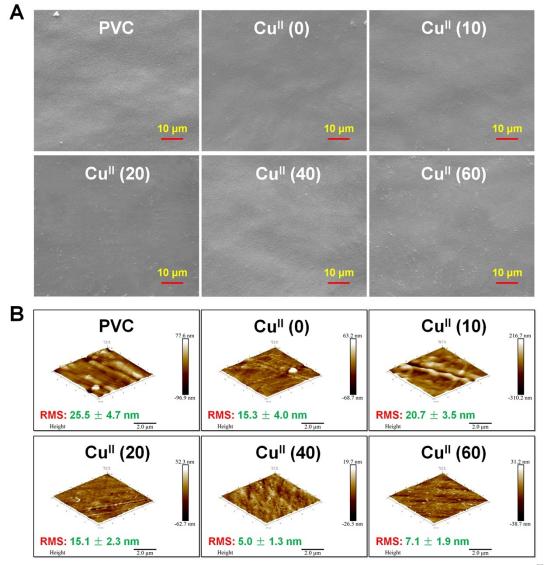


Fig. S1. SEM (A) and AFM (B) of the PVC before and after modification via Cu^{II} -GA coatings.

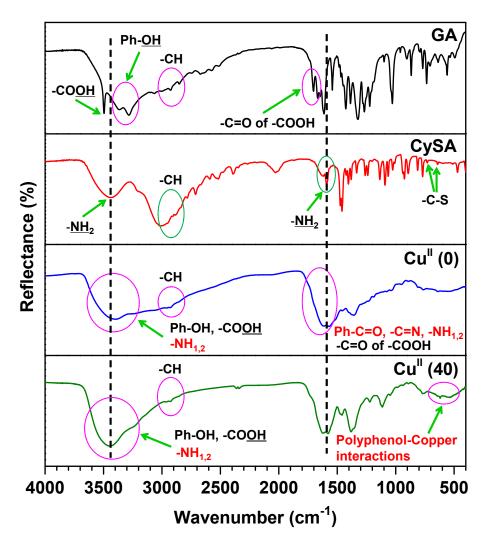


Fig. S2. IR spectra of GA, CySA, Cu^{II}(0)-GA/CySA and Cu^{II}(40)-GA polymers.

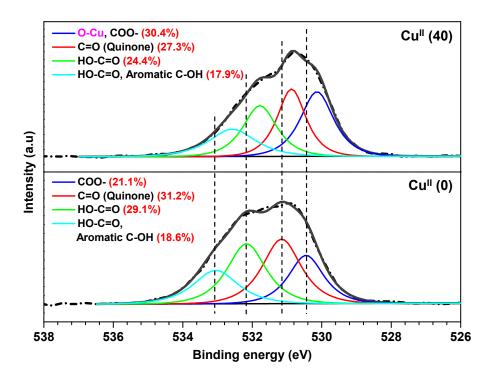


Fig. S3. High-resolution O1s spectra of $Cu^{II}(0)$ and $Cu^{II}(40)$ coatings.

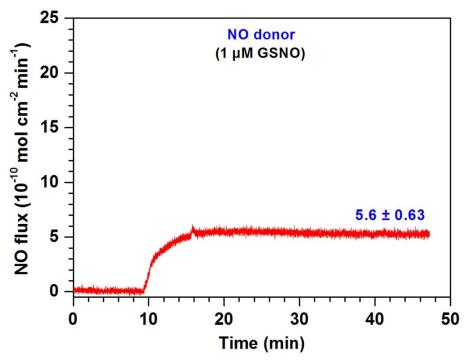


Fig. S4. Catalytic NO generation patterns induced by $Cu^{II}(40)$ coating prepared using 40 µg/mL feed concentrations of $CuCl_2 \cdot 2H_2O$ in PBS (pH 7.4) containing NO donor consisting of 1 µM GSNO and 1 µM GSH.

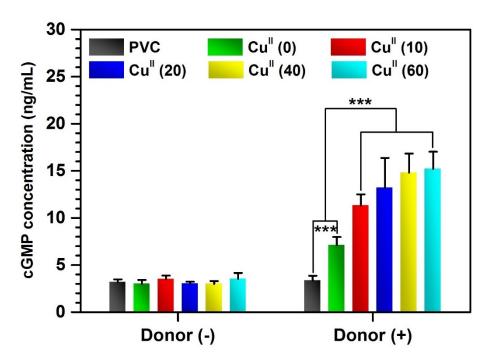


Fig. S5. The concentration of cGMP synthesized by platelets incubated with uncoated and Cu^{II} -GA/CySA coated PVC for 2 hours. Data presented as mean \pm SD (n=4) and analyzed using a one–way ANOVA, ***p < 0.001.

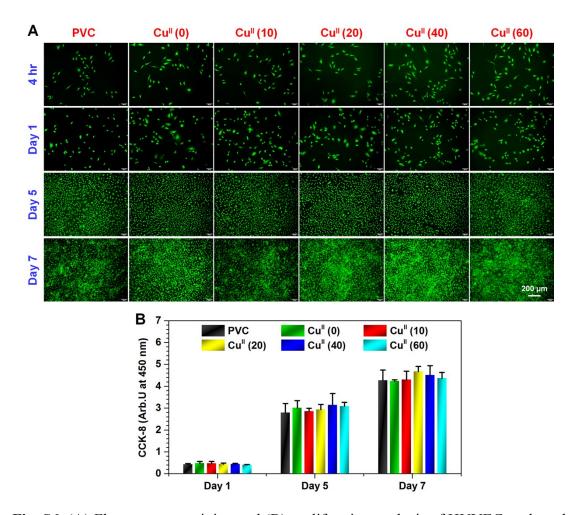


Fig. S6. (A) Fluorescence staining and (B) proliferation analysis of HUVECs cultured on the surfaces of PVC before and after modification via Cu^{II} -GA/CySA coatings after 1, 5 and 7 days of culture. Data presented as mean \pm SD (n = 4).

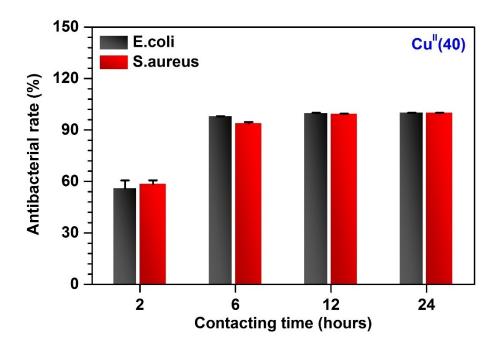


Fig. S7. Antibacterial rate of Cu^{II}(40) at different contacting durations (detailed values of antibacterial rates are available in **Table S2**).

Table S1. Hemolysis ratio of the PVC before and after modification by different Cu^{II} -GA/CySA coatings.

Sample	Hemolysis ratio (%)		
PVC	0.32 ± 0.08		
$Cu^{II}(0)$	0.34 ± 0.22		
$Cu^{II}(10)$	0.27 ± 0.09		
Cu ^{II} (20)	0.54 ± 0.17		
Cu ^{II} (40)	0.72 ± 0.41		
Cu ^{II} (60)	0.36 ± 0.13		

Table S2. Antibacterial rate of $Cu^{II}(40)$ at different contacting durations (%).

	2 h	6 h	12 h	24 h
E.coli	55.87	97.83	99.75	>99.99
S.aureus	58.43	93.82	99.29	99.88