

Anchoring Nickel Nanoparticles on Three-Dimensionally Macro-/mesoporous Titanium Dioxide with Carbon Layer from Polydopamine Using Polymethylmethacrylate Microspheres as Sacrificial Templates

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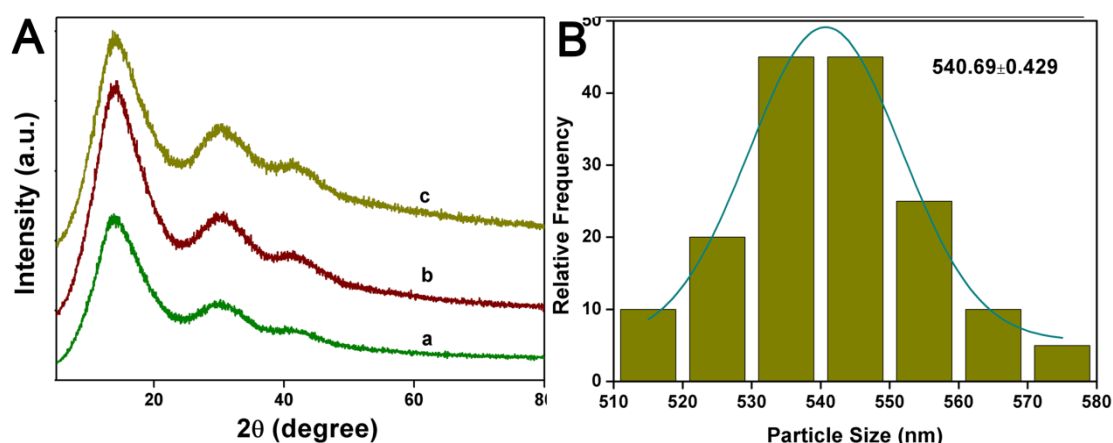


Figure S1. A: X-Ray diffraction patterns of PMMA (a), PMMA@TiO₂ (b), PMMA@TiO₂@PDA-Ni²⁺ (c); B: The particle size distribution of PMMA microspheres

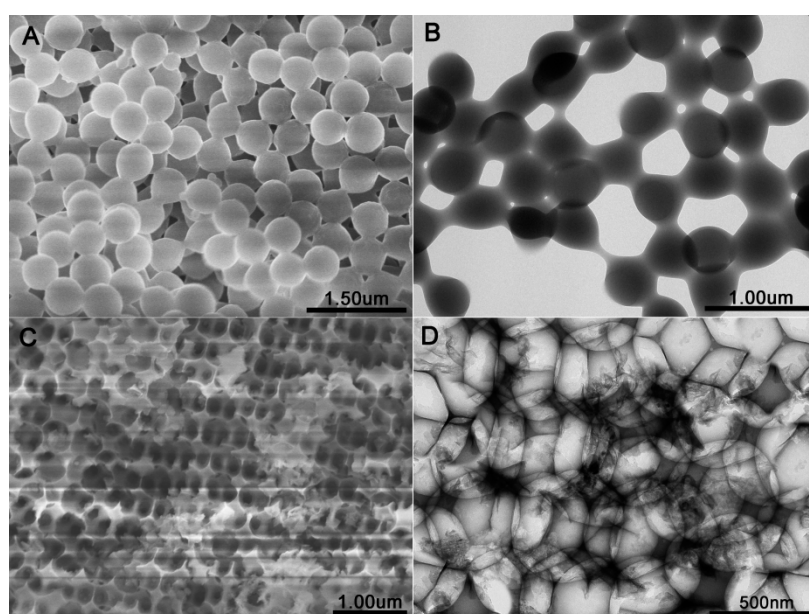


Figure S2. The SEM and TEM images of PMMA@TiO₂ (A, B) and 3D inverse-opal TiO₂ carbonized at 500 °C for 3 h (C,D)

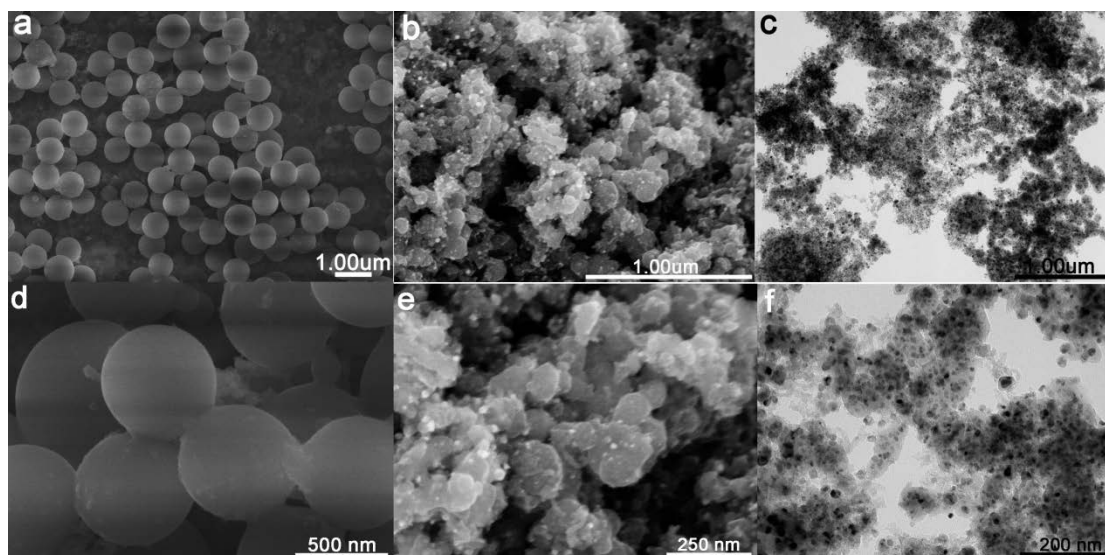


Figure S3. (a,d) SEM image of PMMA@PDA-Ni²⁺, and (b,e) SEM and (c,f) TEM images of 3D C-Ni annealed at 500 °C in N₂ atmosphere.

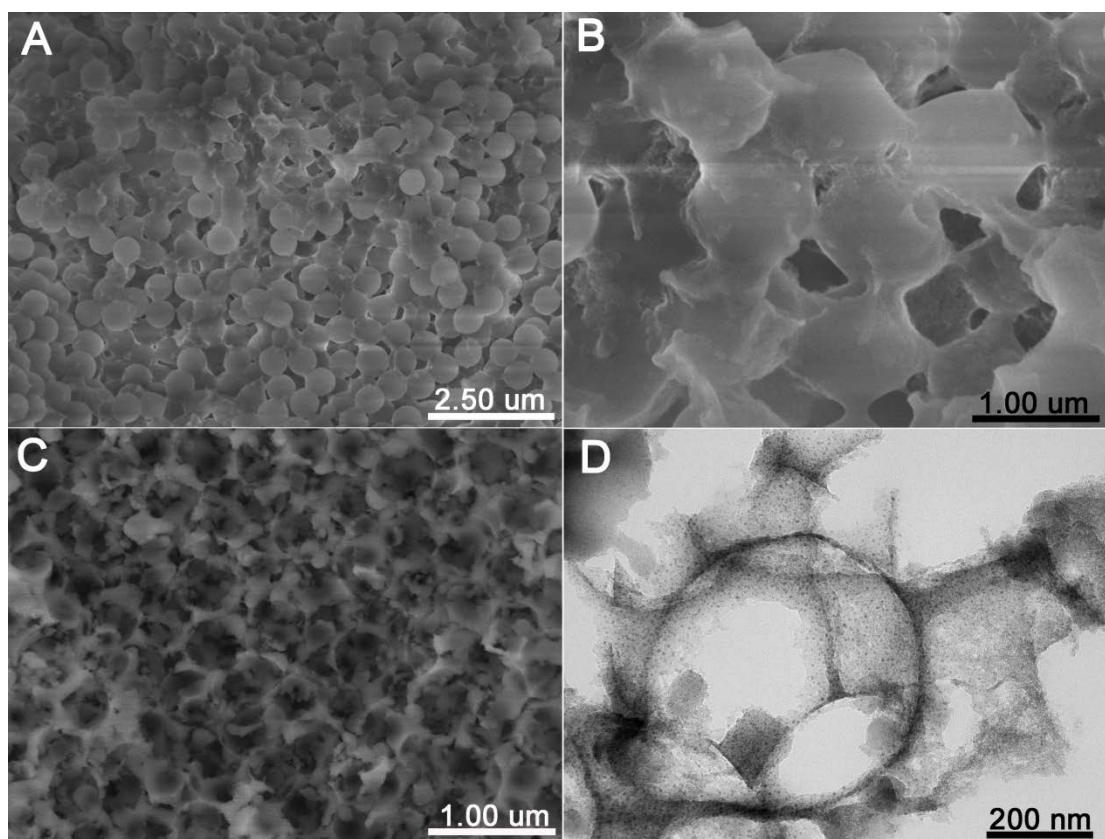


Figure S4. (a, b) SEM image of 3D PMMA@SiO₂@PDA-Ni²⁺ composites; (c) SEM image

and (d) TEM image of 3D SiO₂@C-Ni annealed at 500 °C in N₂ atmosphere.

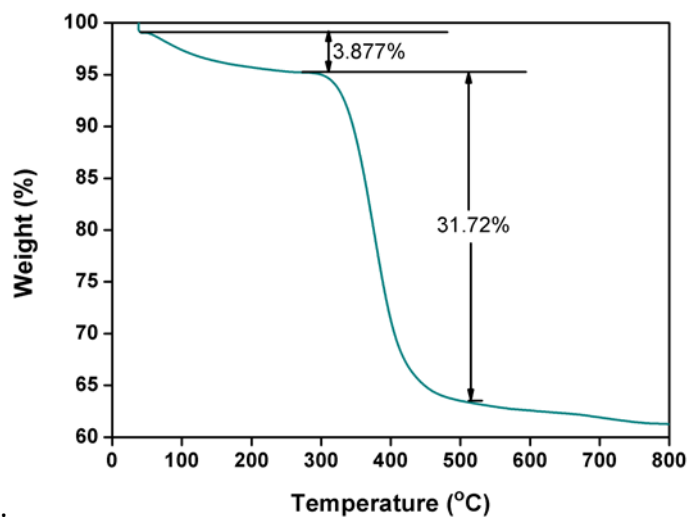


Figure S5. TGA curves of 3D TiO₂@C-Ni-500.

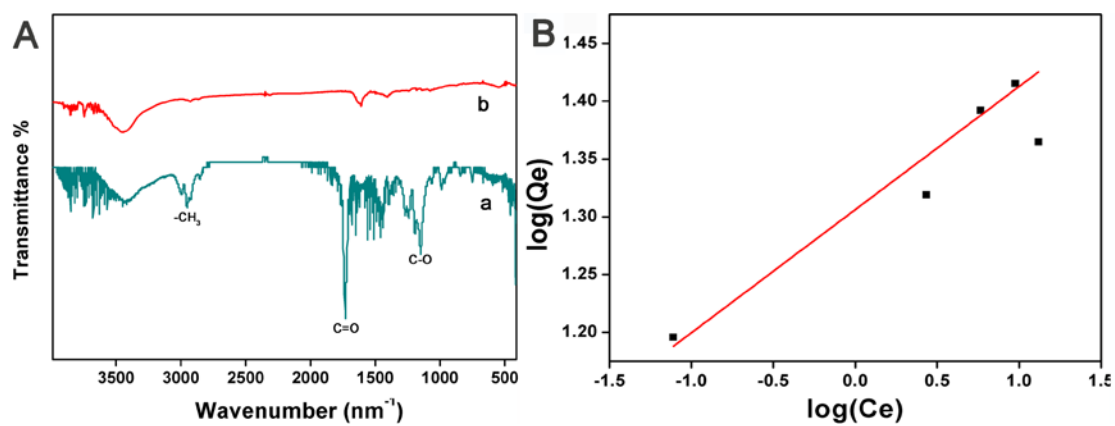


Figure S6. (A) The FT-IR spectra of PMMA@TiO₂@PDA-Ni²⁺(a) and 3D TiO₂@C-Ni-500(b) composites. (B) The fitting curve of Freundlich model.

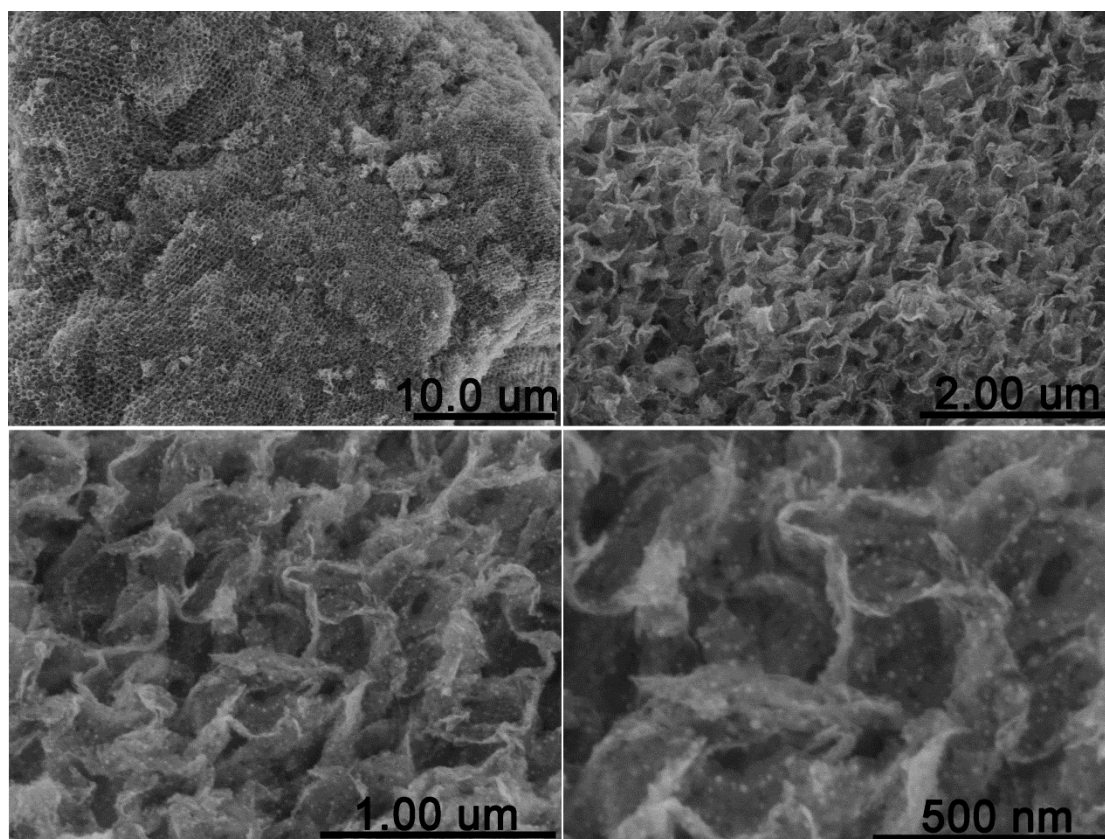


Figure S7. SEM images of 3D TiO₂@C-Ni-500 after seven catalytic reaction.

Table S1. Comparison of the activity parameter κ of metal catalysts for the reduction of 4-NP

Samples	$k(\times 10^{-3} \text{ s}^{-1})$	$\kappa(\times 10^{-3} \text{ mg}^{-1} \text{ s}^{-1})$	References
TiO ₂ @C-Ni-500	17.6	152.0	This work
Fe ₃ O ₄ @SiO ₂ -Au@mSiO ₂	7	105.0	1
P _{PAA}	15.46	38.43	2
C@CoNi/500	15.8	100.6	3
C-Ni/400	5.9	142.0	4
Ag/Cu	6.70	74	5
Ni/SNTs(23.0 wt%)	84	91	6
TiO ₂ @C-Ni/700	15.17	173.21	7

Table S2. Comparison of the adsorption capacity of different adsorbents for BHb

protein

Materials	Adsorption capacity (mg g ⁻¹)	References
TiO ₂ @C-Ni-500	1031.27 mg g ⁻¹	This work
Fe ₃ O ₄ @C/Ni-500	641.0 mg g ⁻¹	8
CuFe ₂ O ₄ MNCs	4475 mg g ⁻¹	9
Fe ₃ O ₄ @ZIF-8	>6000 mg g ⁻¹	10
Anionic PILs	983.4 mg g ⁻¹	11
Polydopamine-coated MIP silica NPs	321.7 mg g ⁻¹	12
C@CoNi	628.93 mg g ⁻¹	13
Fe ₃ O ₄ @BHB- MIPs	37.58 mg g ⁻¹	14

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