

## Supporting Information

### **Alkoxythiophene and alkylthiothiophene $\pi$ -bridges enhance the performance of A-D-A electron acceptors**

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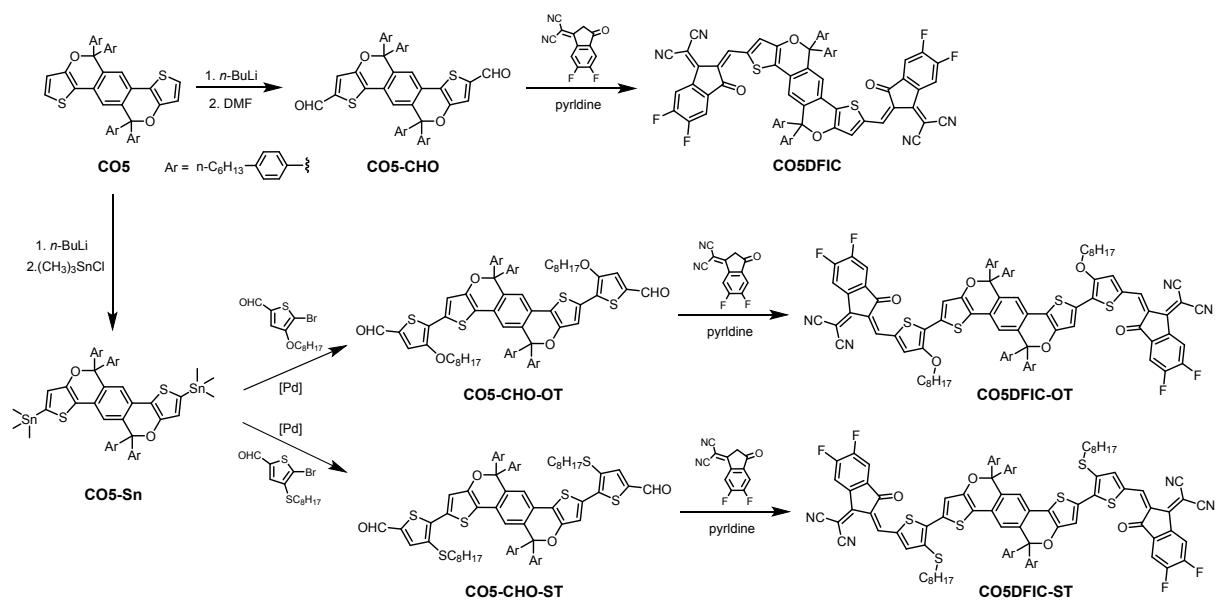
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## 1. General characterization

<sup>1</sup>H and <sup>13</sup>C NMR spectra were measured on a Bruker Avance-400 spectrometer. Absorption spectra were recorded on a Shimadzu UV-1800 spectrophotometer. Cyclic voltammetry was done by using a Shanghai Chenhua CHI620D voltammetric analyzer under argon in an anhydrous acetonitrile solution of tetra-n-butylammonium hexafluorophosphate (0.1 M). A glassy-carbon electrode was used as the working electrode, a platinum-wire was used as the counter electrode, and a Ag/Ag<sup>+</sup> electrode was used as the reference electrode. The donor or acceptor was coated onto glassy-carbon electrode and all potentials were corrected against Fc/Fc<sup>+</sup>. AFM was performed on a Multimode microscope (Veeco) by using tapping mode.

## 2. Synthesis

All reagents were purchased from J&K Co., Aladdin Co., Innochem Co., Derthon Co., Adamas-beta and other commercial suppliers. CO5<sup>[1]</sup>, CO5-CHO<sup>[1]</sup>, 2-bromo-3-(octylthio)thiophene<sup>[2]</sup> and 5-bromo-4-(octyloxy)thiophene-2-carbaldehyde<sup>[3]</sup> were prepared according to literature.



**Scheme S1** The synthetic routes for CO5DFIC, CO5DFIC-OT and CO5DFIC-ST.

**5-Bromo-4-(octylthio)thiophene-2-carbaldehyde.** To a solution of 2-bromo-3-(octylthio)thiophene (240 mg, 0.78 mmol) in dry THF (10 mL) was added LDA (0.47 mL, 2 M, 0.94 mmol) at -78 °C under argon. The resulting mixture was stirred for 2 h. To the

mixture was added N,N-dimethylformamide (75.4  $\mu$ L, 0.94 mmol). Then, the mixture was warmed to -40 °C and stirred for 1 h. It was then poured into water followed by extraction with  $\text{CH}_2\text{Cl}_2$  three times. The combined organic layer was dried over anhydrous  $\text{Na}_2\text{SO}_4$ . After removal of the solvent, the crude product was purified via column chromatography (silica gel) by using  $\text{CH}_2\text{Cl}_2$ :petroleum ether (1:2) as eluent to give **5-bromo-4-(octylthio)thiophene-2-carbaldehyde** as a colourless liquid (170 mg, 65%).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$   $\delta$ /ppm): 9.76 (s, 1H), 7.56 (s, 1H), 2.89 (t,  $J$  = 7.3 Hz, 2H), 1.49-1.74 (m, 2H), 1.13-1.47 (m, 10H), 0.87 (t,  $J$  = 4.4 Hz, 3H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ): 181.37, 143.08, 137.99, 135.87, 128.58, 35.13, 31.73, 29.41, 29.01, 29.04, 28.54, 22.60, 14.07.

**CO5DFIC.** To a solution of CO5-CHO (70 mg, 0.07 mmol) in  $\text{CHCl}_3$  (10 mL) was added DFIC (81 mg, 0.35 mmol) and pyridine (0.5 mL) at room temperature. The mixture was heated to reflux for 1 h. After removal of the solvent, the crude product was purified via column chromatography (silica gel) by using  $\text{CHCl}_3$  as eluent to give **CO5DFIC** as a black solid (100 mg, 95%).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$   $\delta$ /ppm): 8.62 (s, 2H), 8.48-8.55 (m, 2H), 7.58-7.67 (m, 4H), 7.17 (m, 16H), 6.92 (s, 2H), 2.63 (t,  $J$  = 7.7 Hz, 8H), 1.59-1.65 (m, 8H), 1.24-1.39 (m, 24H), 0.85 (t,  $J$  = 6.7 Hz, 12H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$ /ppm): 185.47, 157.94, 155.90, 155.77, 154.08, 153.29, 153.16, 143.56, 139.03, 137.30, 134.99, 134.89, 134.44, 134.31, 133.10, 128.52, 128.26, 128.23, 123.70, 122.94, 115.10, 114.88, 113.96, 113.77, 112.75, 89.09, 70.66, 70.65, 35.60, 31.68, 31.14, 29.01, 22.56, 14.06. MALDI-TOF MS (m/z): 1420.2 ( $\text{M}^+$ ).

**CO5-Sn.** To a solution of CO5 (300 mg, 0.32 mmol) in dry THF (20 mL) was added *n*-BuLi (0.5 mL, 1.6 M, 0.8 mmol) at -78 °C under argon. The resulting mixture was stirred for 1 h. The mixture was warmed to 0 °C and stirred for 2 h. To the resulting mixture was added chlorotrimethylstannane (0.96 mL, 1.0 M, 0.96 mmol). The mixture was stirred overnight. It was then poured into water followed by extraction with  $\text{CH}_2\text{Cl}_2$  three times. The combined organic layer was dried over anhydrous  $\text{Na}_2\text{SO}_4$ . After removal of the solvent to give **CO5-Sn** as an light yellow solid (310 mg, 77%).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$   $\delta$ /ppm): 7.10-7.16 (m, 16H), 6.80 (s, 2H), 6.54 (s, 2H), 2.59 (t,  $J$  = 7.7 Hz, 8H), 1.56-1.62 (m, 8H), 1.30-1.34 (m, 24H), 0.87 (t,  $J$  = 6.6 Hz, 12H), 0.29 (s, 18H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ , 100 MHz,  $\delta$ /ppm): 153.30, 142.45, 140.82, 135.99, 132.85, 128.56, 127.76, 126.66, 126.62, 121.96, 121.70, 88.70, 35.58, 31.71, 31.17, 28.97, 22.60, 14.11, -8.29. MALDI-TOF MS (m/z): 1264.9 ( $\text{M}^+$ ).

**CO5-CHO-OT.** To a solution of CO5-Sn (120 mg, 0.09 mmol) and 5-bromo-4-(octyloxy)thiophene-2-carbaldehyde (75 mg, 0.23 mmol) in dry toluene (10 mL) and DMF (1 mL) was added Pd(PPh<sub>3</sub>)<sub>4</sub> (11 mg, 0.01 mmol) under argon. The mixture was stirred at 110 °C for 12 h. After removal of the solvent, the crude product was purified via column chromatography (silica gel) by using CH<sub>2</sub>Cl<sub>2</sub>:petroleum ether (2:1) as eluent to give **CO5-CHO-OT** as a red solid (87 mg, 65%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub> δ/ppm): 9.71 (s, 2H), 7.39 (br, 2H), 7.11-7.18 (m, 16H), 6.52 (br, 2H), 4.10 (br, 4H), 2.57-2.61 (m, 8H), 1.76-1.84 (m, 4H), 1.57-1.64 (m, 8H), 1.25-1.47 (m, 44H), 0.84-0.88 (m, 18H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz, δ/ppm): 181.59 (br), 160.56, 153.70 (br), 152.24 (br), 142.85, 140.11, 136.20 (br), 133.92, 133.91, 131.17 (br), 128.50, 127.85, 126.65 (br), 123.34 (br), 121.37, 116.99 (br), 88.75, 72.05, 35.63, 31.76, 31.70, 31.24, 29.21, 29.09, 25.77, 22.64, 22.59, 14.09, 14.07. MALDI-TOF MS (m/z): 1415.2 (M<sup>+</sup>).

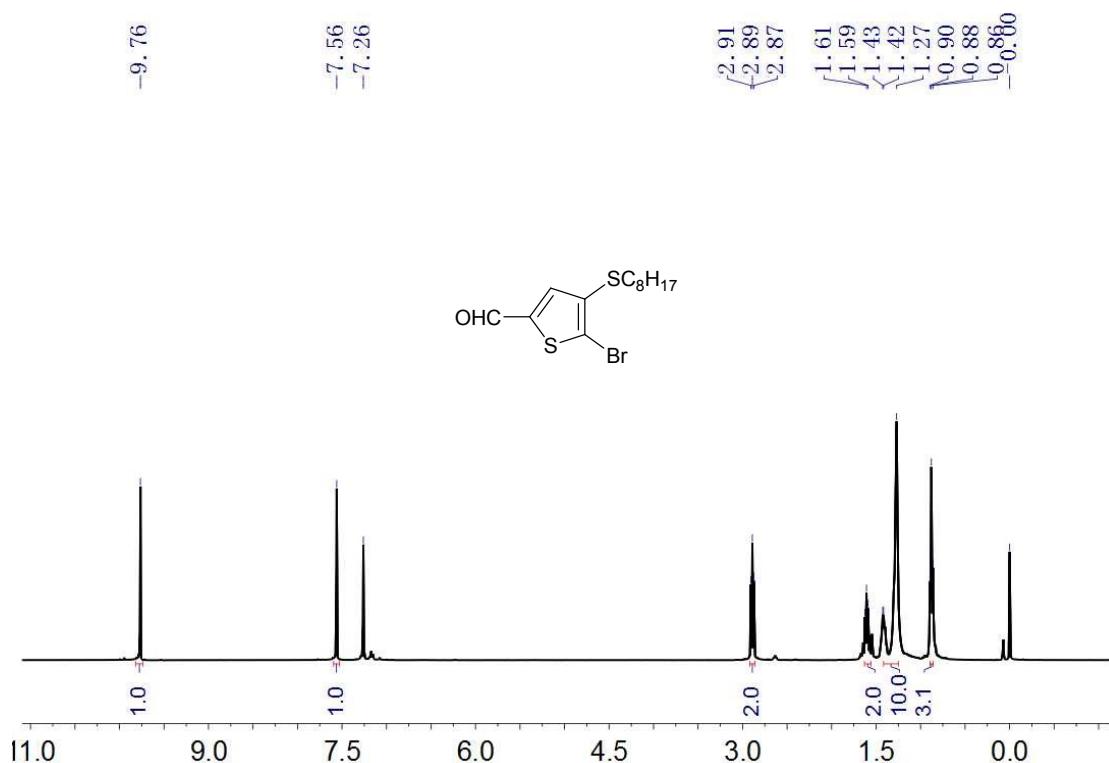
**CO5DFIC-OT.** To a solution of CO5OT-CHO (60 mg, 0.042 mmol) in CHCl<sub>3</sub> (10 mL) was added DFIC (49 mg, 0.21 mmol) and pyridine (0.5 mL) at room temperature. The mixture was heated to reflux for 1 h. After removal of the solvent, the crude product was purified via column chromatography (silica gel) by using CHCl<sub>3</sub> as eluent to give **CO5DFIC-OT** as a black solid (78 mg, 73%). <sup>1</sup>H NMR (400 MHz, δ/ppm): 8.64 (br, 2H), 8.48 (m, 2H), 7.62 (t, *J* = 7.5 Hz, 2H), 7.41 (br, 2H), 7.35 (br, 2H), 7.13-7.21 (m, 16H), 6.60 (br, 2H), 4.12 (br, 4H), 2.60-2.64 (m, 8H), 1.79-1.88 (m, 4H), 1.57-1.67 (m, 8H), 1.21-1.42 (m, 44H), 0.84-0.86 (m, 18H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz, δ/ppm): 185.75, 157.76 (br), 156.18, 155.60, 155.46, 153.00, 152.87, 142.99, 139.87, 136.52, 136.44, 135.90 (br), 135.66 (br), 134.39, 134.33, 134.30, 134.25, 132.00 (br), 131.41 (br), 129.88 (br), 129.37 (br), 128.40, 127.95, 127.15 (br), 122.57 (br), 121.59, 118.88 (br), 118.54 (br), 114.87, 114.65, 114.48, 112.44, 112.25, 88.77, 72.19, 68.83, 35.70, 31.78, 31.70, 31.25, 29.26, 29.16, 25.77, 22.66, 22.58, 14.10, 14.06. MALDI-TOF MS (m/z): 1840.6 (M<sup>+</sup>).

**CO5-CHO-ST.** To a solution of CO5-Sn (135 mg, 0.11 mmol) and 5-bromo-4-(octylthio)thiophene-2-carbaldehyde (89 mg, 0.27 mmol) in dry toluene (10 mL) and DMF (1 mL) was added Pd(PPh<sub>3</sub>)<sub>4</sub> (12 mg, 0.01 mmol) under argon. The mixture was stirred at 110 °C for 12 h. After removal of the solvent, the crude product was collected and purified via column chromatography (silica gel) by using CH<sub>2</sub>Cl<sub>2</sub>:petroleum ether (2:1) as eluent to give **CO5-CHO-ST** as a red solid (97 mg, 63%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub> δ/ppm): 9.78 (s, 2H), 7.60 (s, 2H), 7.32 (s, 2H), 7.13-7.18 (m, 16H), 6.59 (s, 2H), 2.82 (t, *J* = 7.3 Hz, 4H), 2.60 (t, *J*

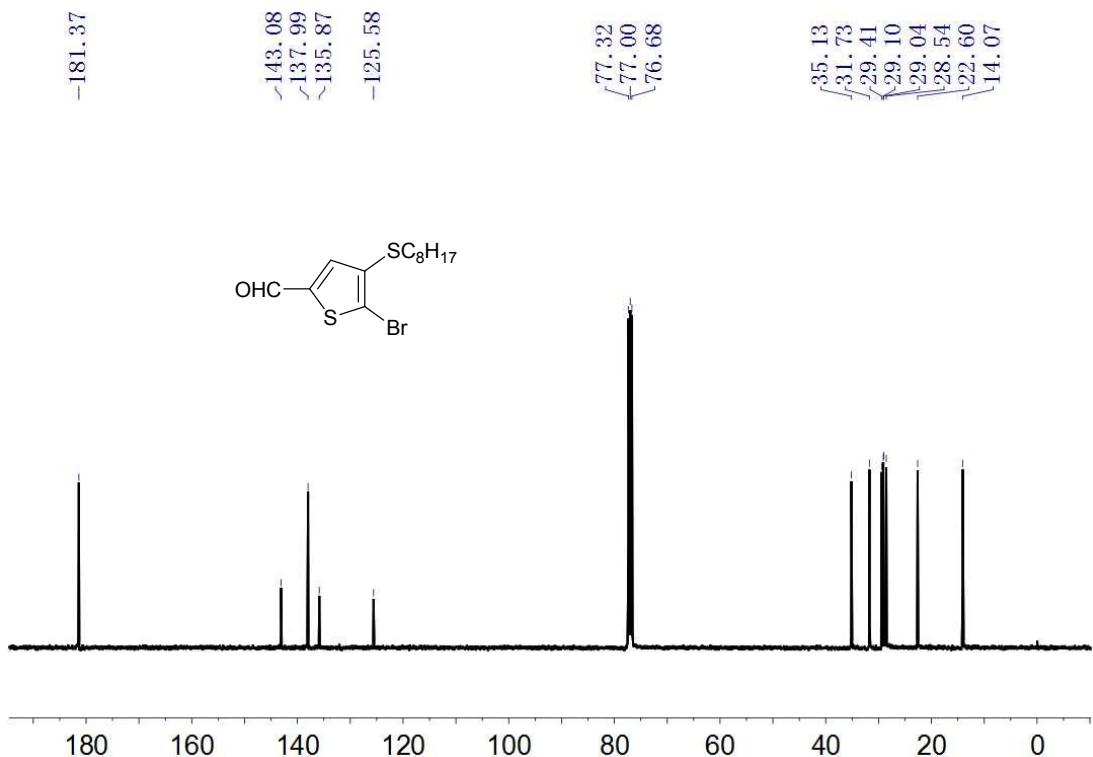
= 7.6 Hz, 8H), 1.53-1.65 (m, 12H), 1.22-1.38 (m, 44H), 0.85-0.88 (m, 18H).  $^{13}\text{C}$  NMR (CDCl<sub>3</sub>, 100 MHz,  $\delta$ /ppm): 181.85, 152.29, 144.44, 143.01, 140.68, 139.89, 139.11, 134.26, 131.65, 129.69, 128.52, 127.94, 126.72, 121.71, 120.24, 119.67, 88.90, 36.01, 35.61, 31.75, 31.69, 31.22, 29.38, 29.10, 29.04, 28.64, 22.60, 14.08. MALDI-TOF MS (m/z): 1448.5 (M<sup>+</sup>).

**CO5DFIC-ST.** To a solution of CO5ST-CHO (70 mg, 0.048 mmol) in CHCl<sub>3</sub> (10 mL) was added DFIC (56 mg, 0.24 mmol) and pyridine (0.5 mL) at room temperature. The mixture was heated to reflux for 1 h. After removal of the solvent, the crude product was purified via column chromatography (silica gel) by using CHCl<sub>3</sub> as eluent to give **CO5DFIC-ST** as a black solid (77 mg, 85%).  $^1\text{H}$  NMR (400 MHz, CDCl<sub>3</sub>  $\delta$ /ppm): 8.65 (s, 2H), 8.49-8.53 (m, 2H), 7.61-7.68 (m, 4H), 7.50 (s, 2H), 7.14-7.23 (m, 16H), 6.68 (s, 2H), 2.90 (t,  $J$  = 7.3 Hz, 4H), 2.61-2.65 (m, 8H), 1.58-1.67 (m, 12H), 1.26-1.42 (m, 44H), 0.85-0.88 (m, 18H).  $^{13}\text{C}$  NMR (CDCl<sub>3</sub>, 100 MHz,  $\delta$ /ppm): 185.89, 157.79, 155.87, 155.73, 153.20, 150.08, 148.51, 143.14, 139.74, 136.60, 136.33, 134.77, 134.43, 134.40, 134.36, 133.66, 131.98, 128.53, 128.03, 127.04, 122.75, 122.09, 121.98, 120.57, 155.06, 144.85, 144.05, 113.99, 112.77, 112.58, 89.01, 70.28, 70.27, 35.83, 35.64, 31.75, 31.68, 31.21, 29.16, 29.11, 29.08, 29.06, 28.69, 22.61, 22.58, 14.07. MALDI-TOF MS (m/z): 1872.2 (M<sup>+</sup>).

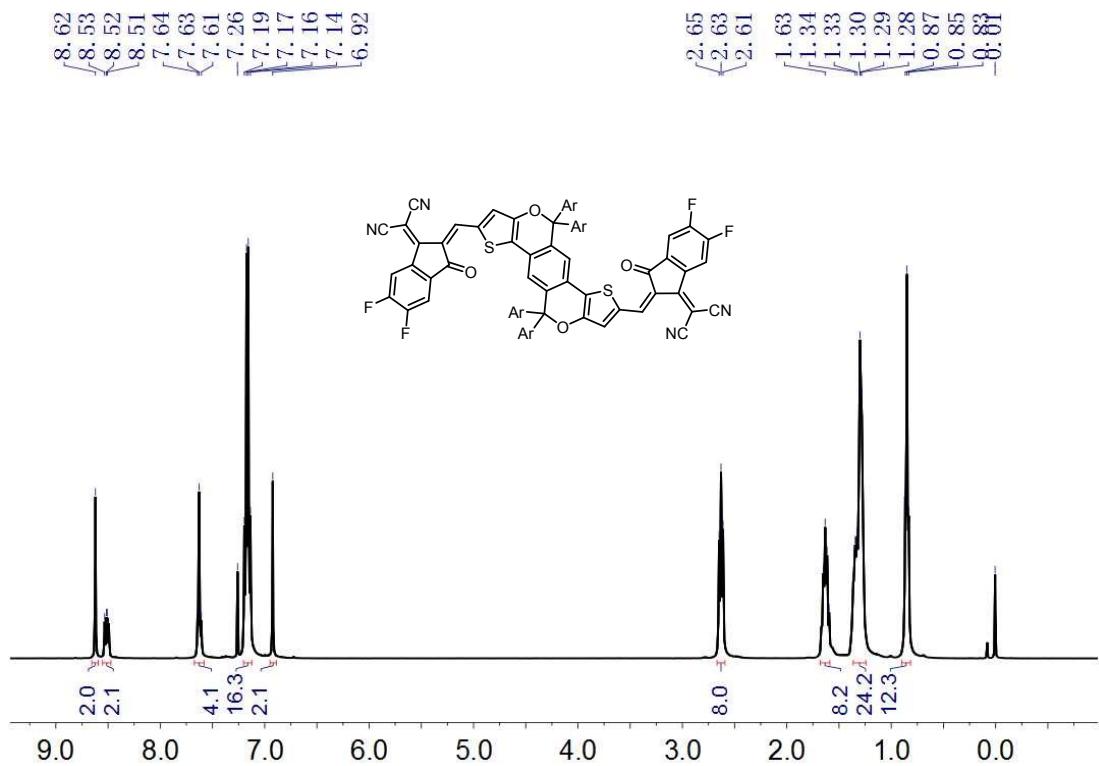
### 3. NMR



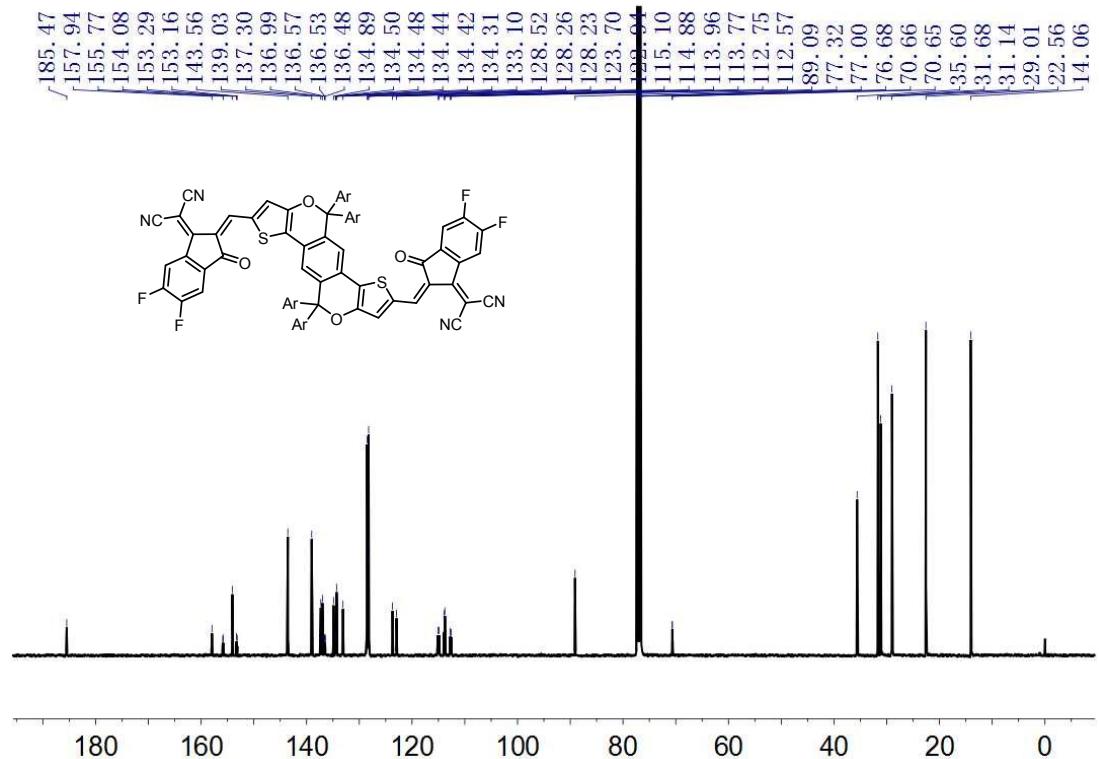
**Fig. S1**  $^1\text{H}$  NMR spectrum of **5-bromo-4-(octylthio)thiophene-2-carbaldehyde**.



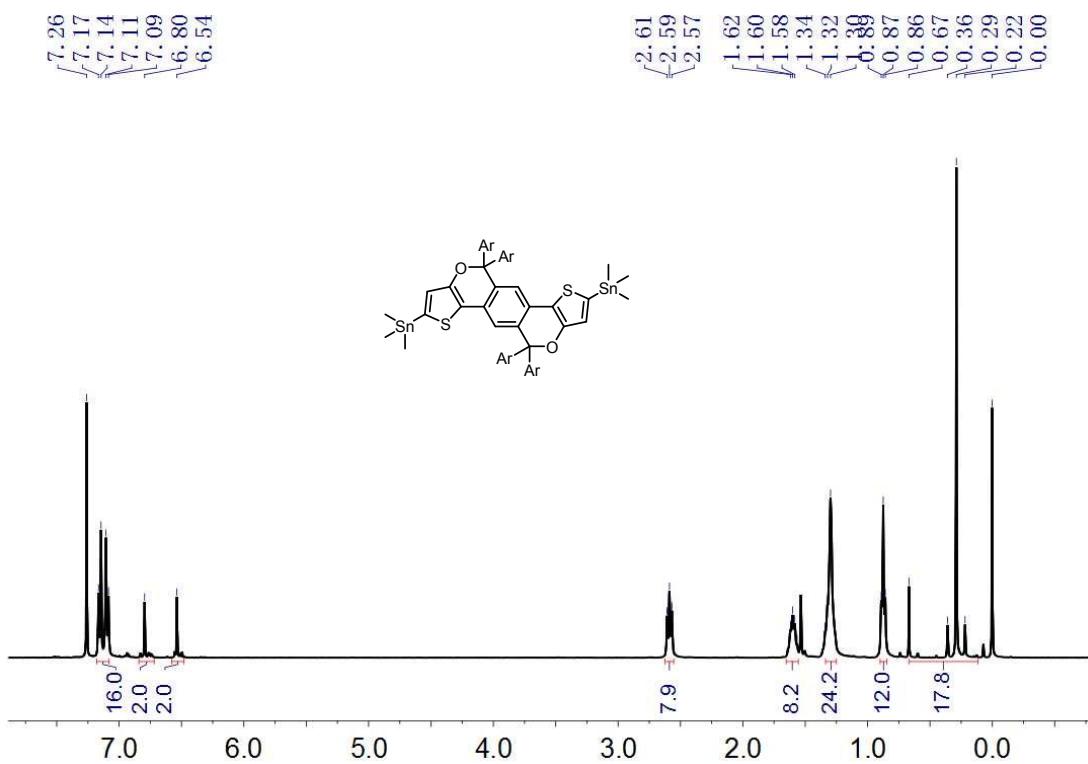
**Fig. S2**  $^{13}\text{C}$  NMR spectrum of **5-bromo-4-(octylthio)thiophene-2-carbaldehyde**.



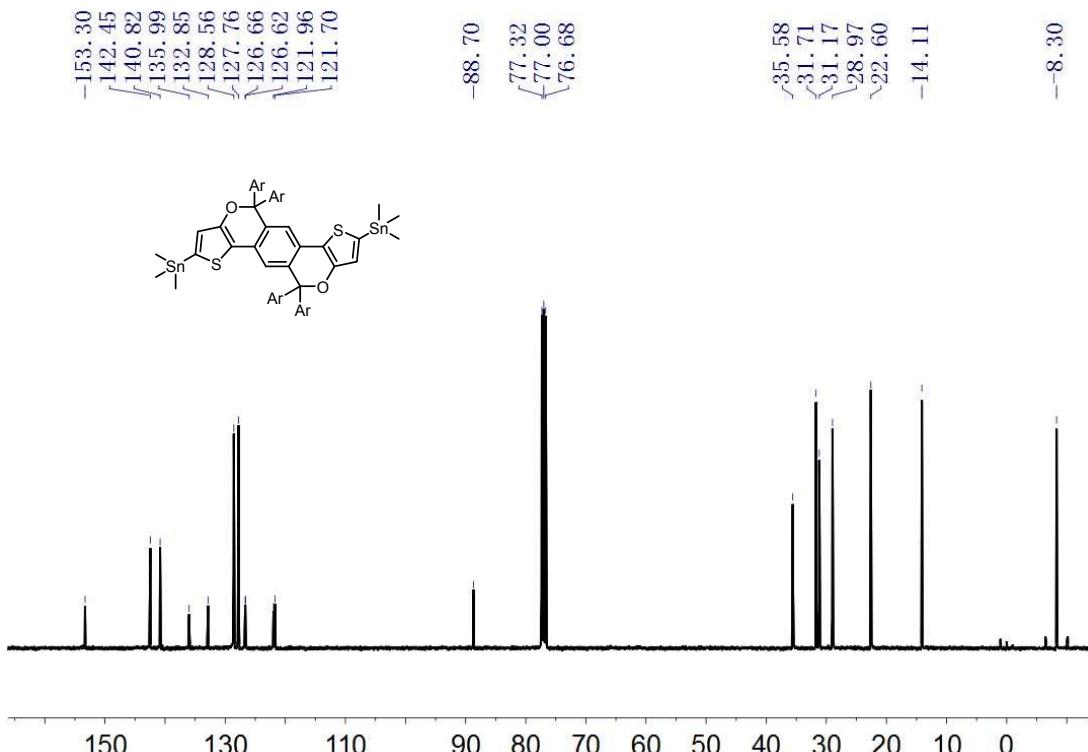
**Fig. S3**  $^1\text{H}$  NMR spectrum of CO5DFIC.



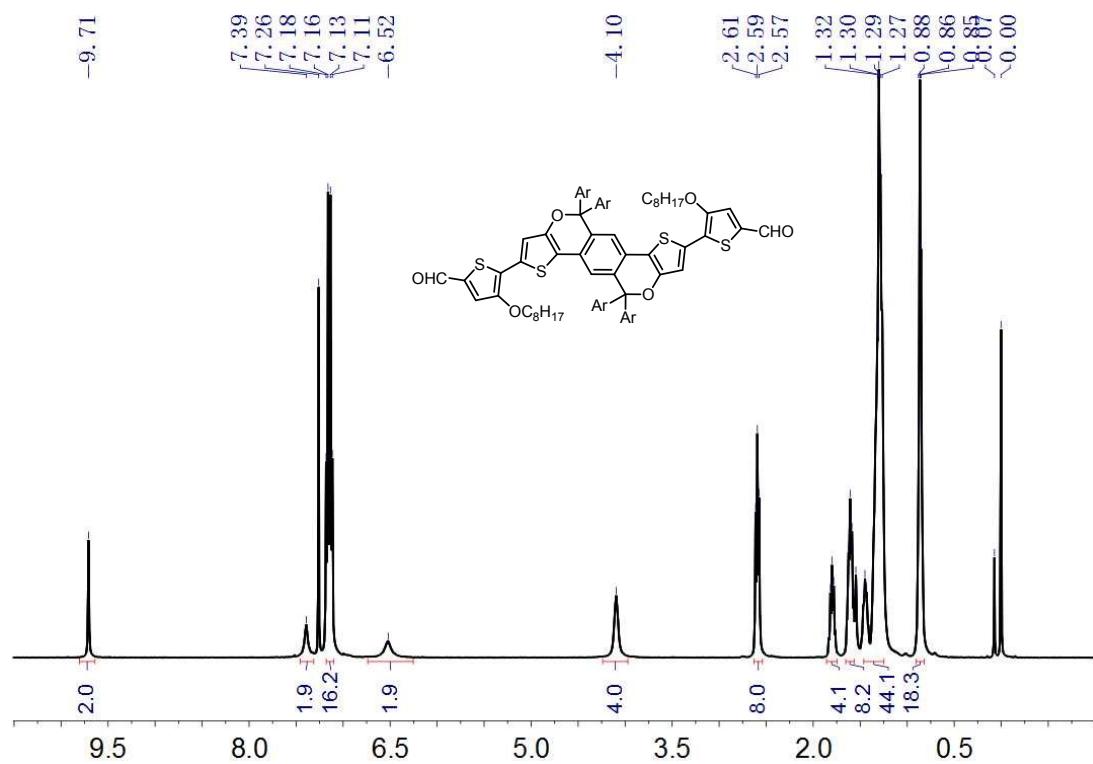
**Fig. S4**  $^{13}\text{C}$  NMR spectrum of CO5DFIC.



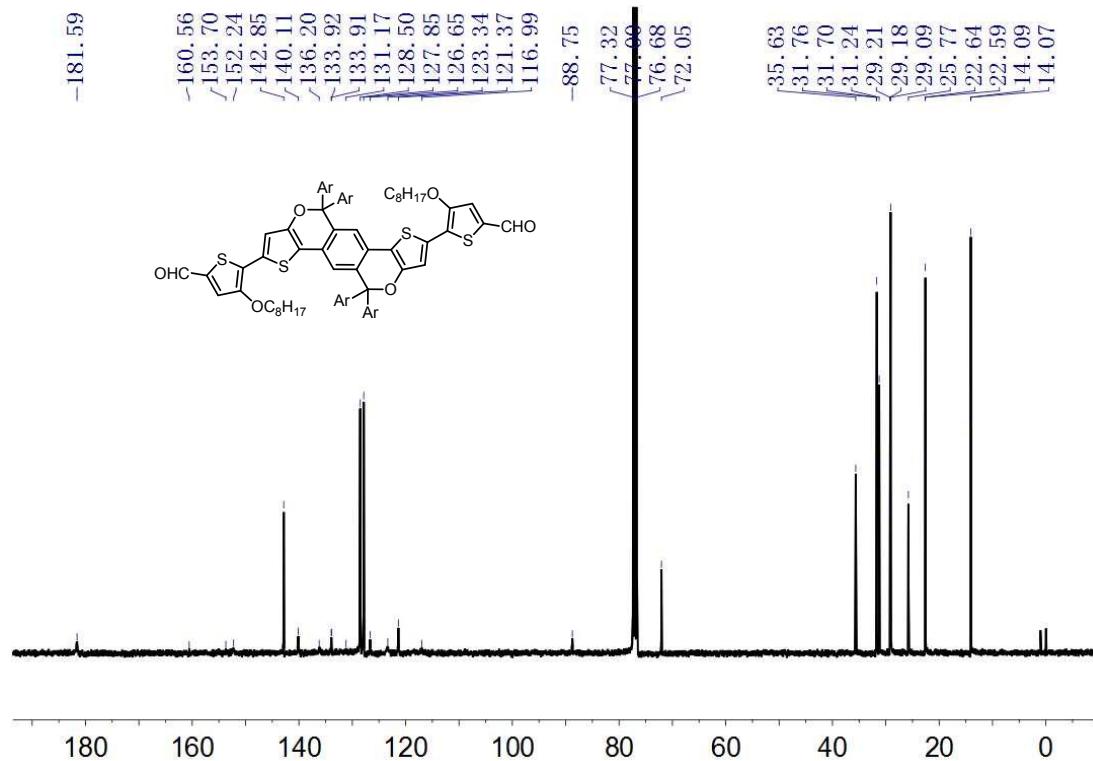
**Fig. S5**  $^1\text{H}$  NMR spectrum of **CO5-Sn**.



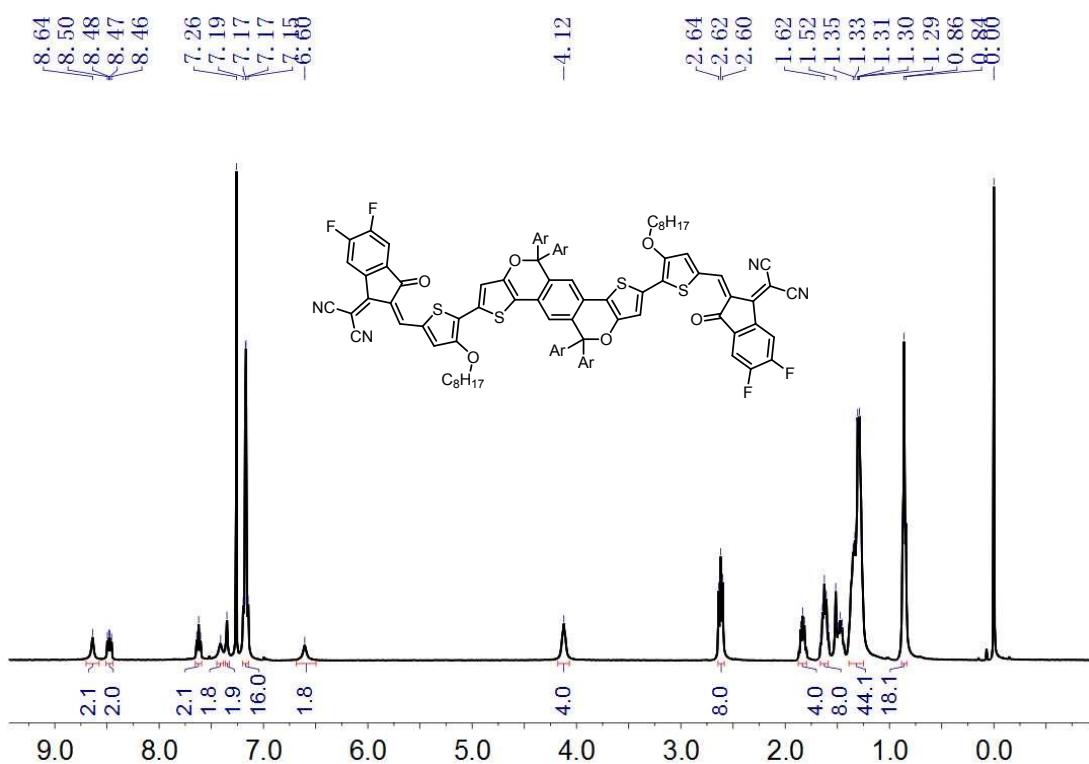
**Fig. S6**  $^{13}\text{C}$  NMR spectrum of **CO5-Sn**.



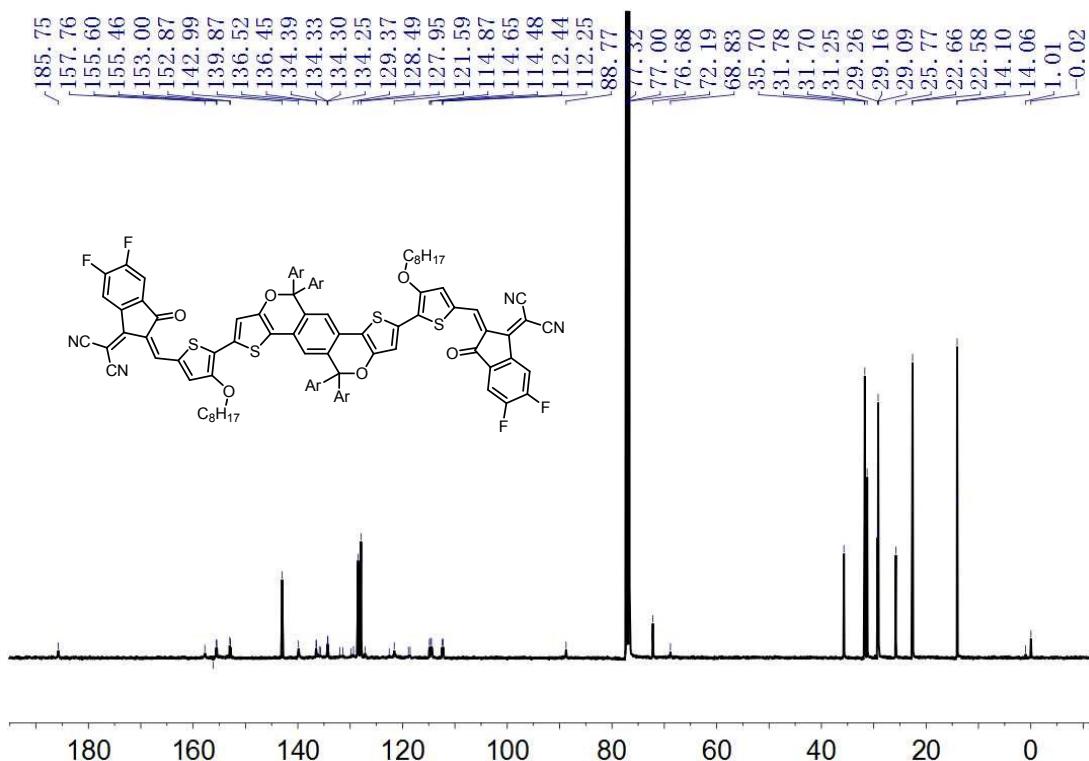
**Fig. S7**  $^1\text{H}$  NMR spectrum of **CO5-CHO-OT**.



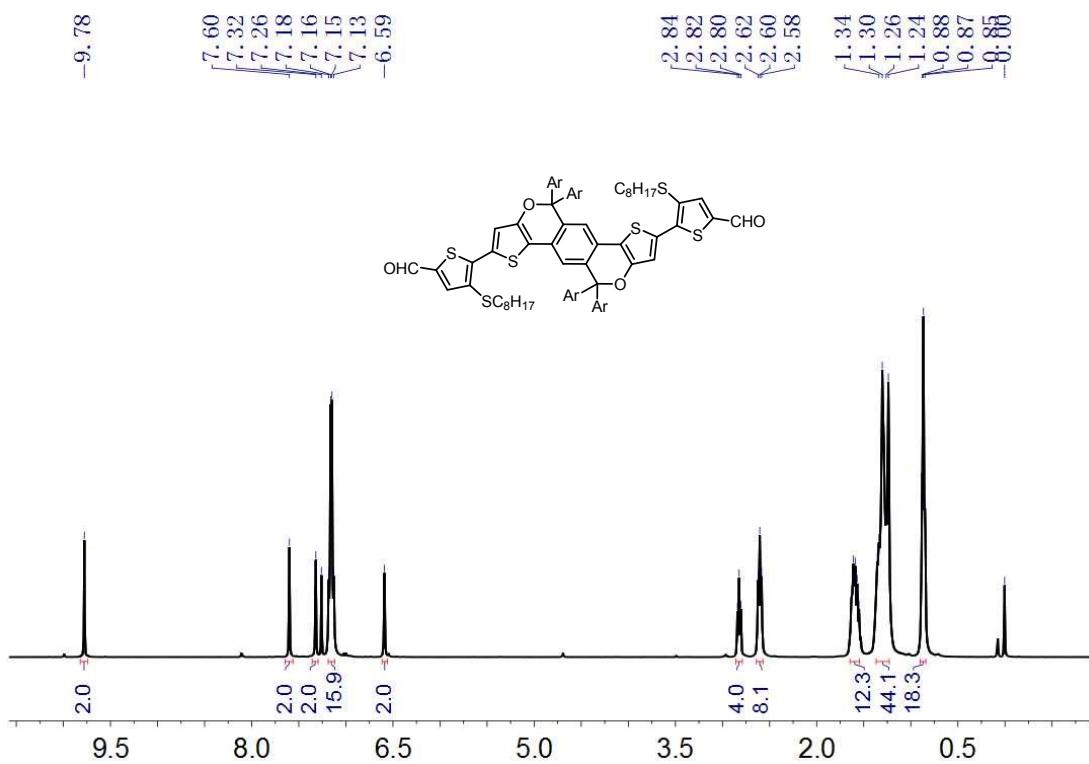
**Fig. S8**  $^{13}\text{C}$  NMR spectrum of **CO5-CHO-OT**.



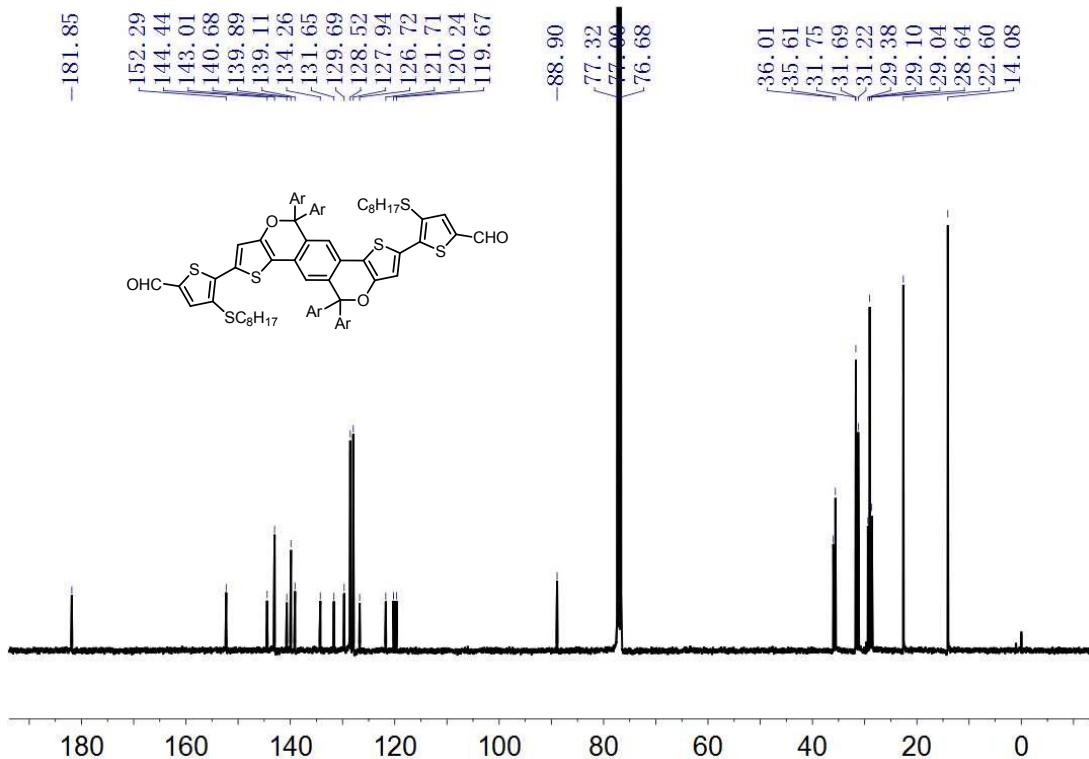
**Fig. S9**  $^1\text{H}$  NMR spectrum of **CO5DFIC-OT**.



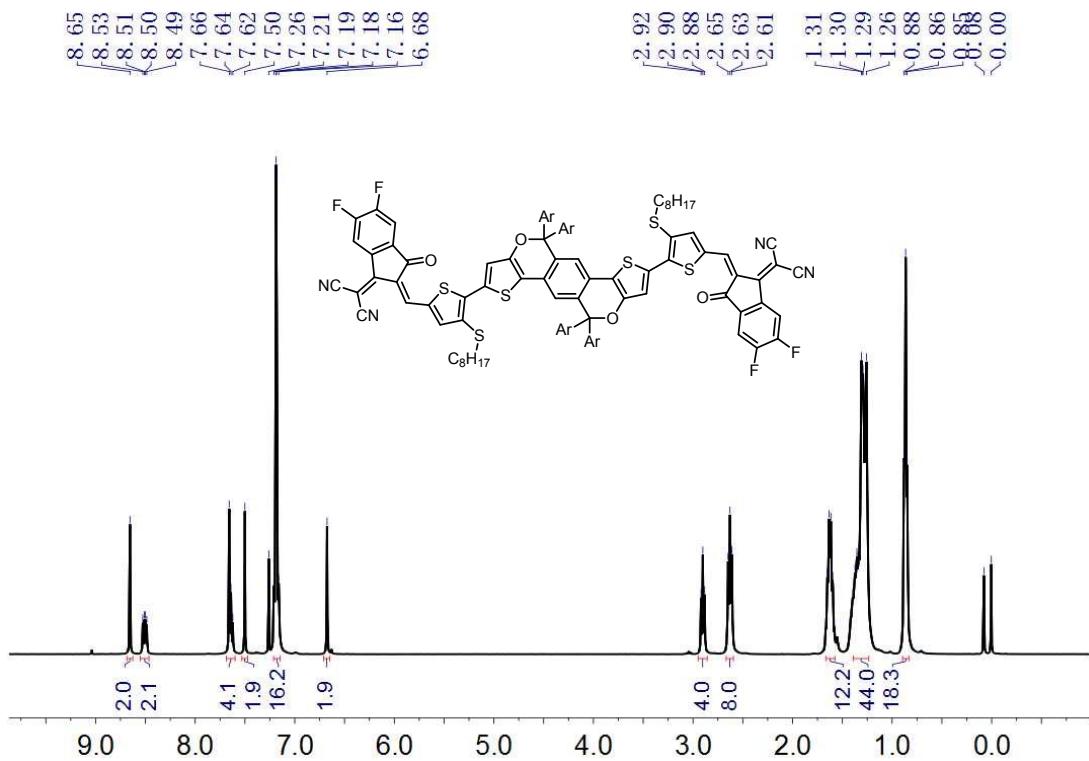
**Fig. S10**  $^{13}\text{C}$  NMR spectrum of **CO5DFIC-OT**.



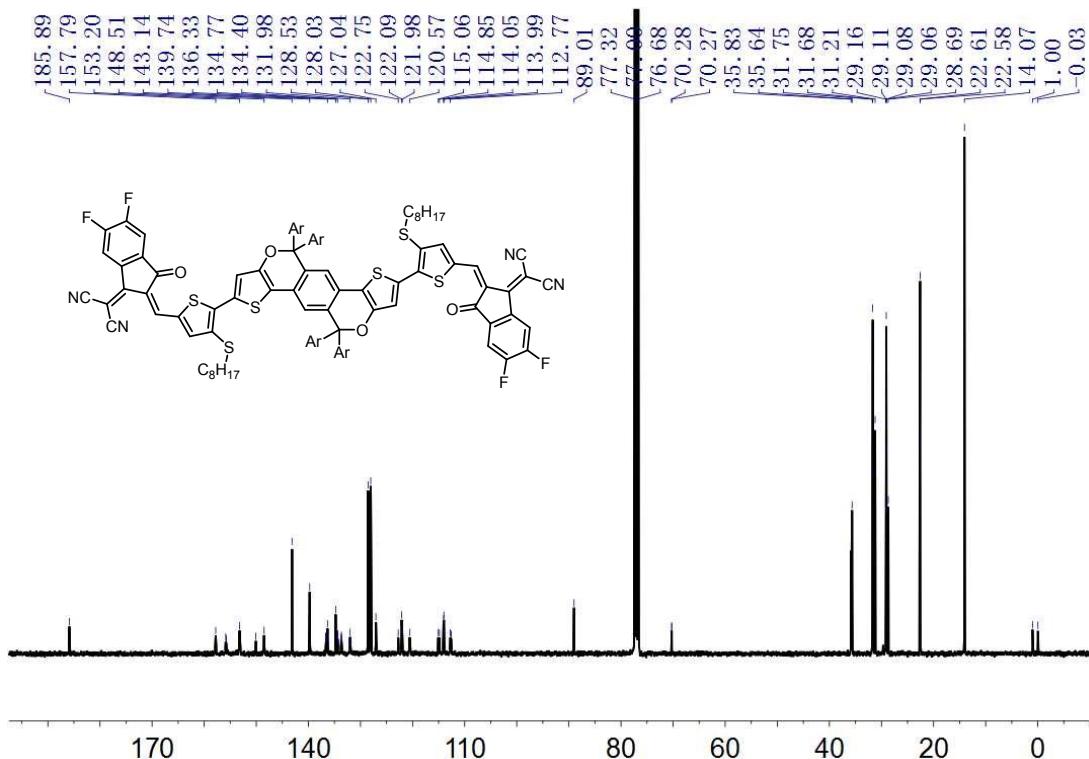
**Fig. S11**  $^1\text{H}$  NMR spectrum of **CO5-CHO-ST**.



**Fig. S12**  $^{13}\text{C}$  NMR spectrum of **CO5-CHO-ST**.

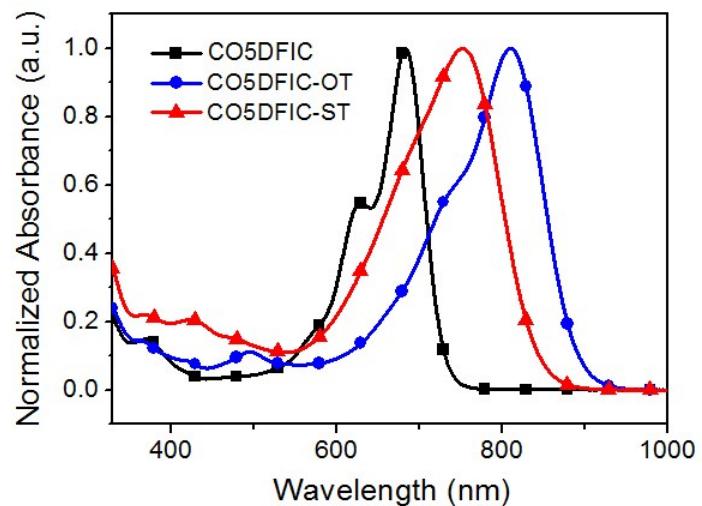


**Fig. S13**  $^1\text{H}$  NMR spectrum of **CO5DFIC-ST**.



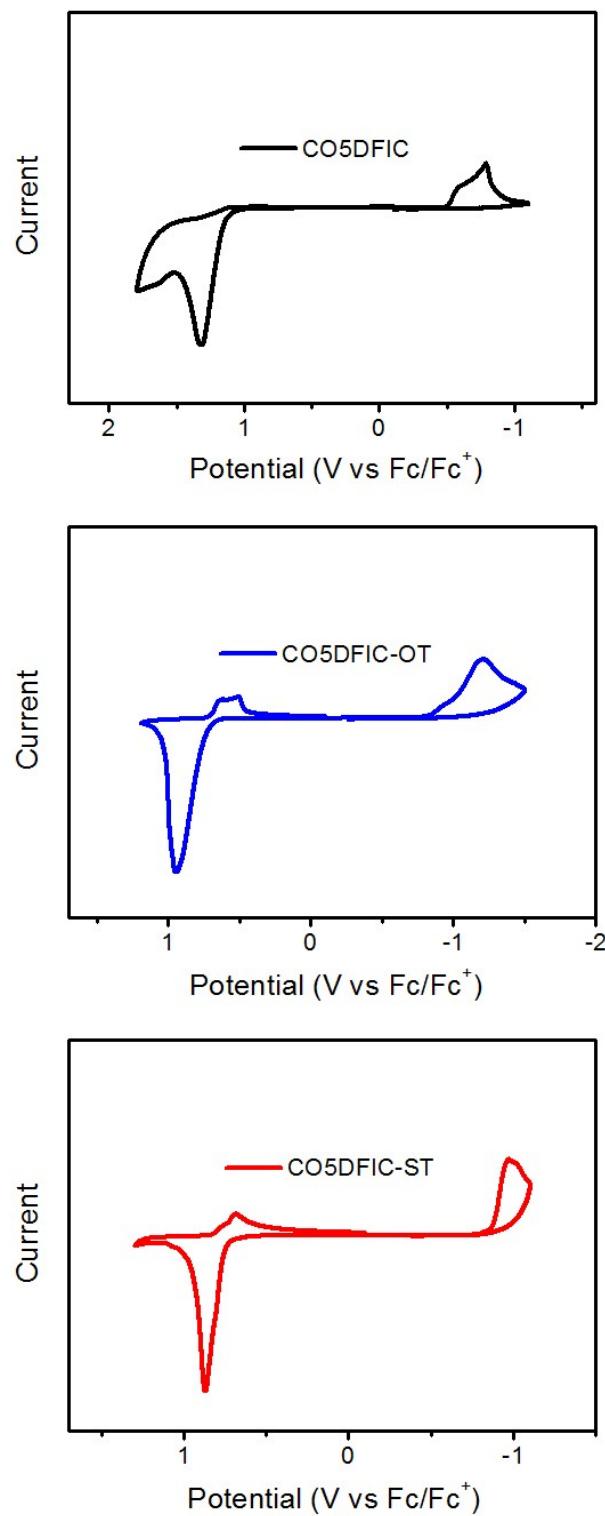
**Fig. S14**  $^{13}\text{C}$  NMR spectrum of **CO5DFIC-ST**.

#### 4. UV-Vis



**Fig S15** Absorption spectra for CO5DFIC, CO5DFIC-OT and CO5DFIC-ST in solution.

## 5. CV



**Fig. S16** Cyclic voltammograms for CO5DFIC, CO5DFIC-OT and CO5DFIC-ST.

## 6. Device fabrication and measurements

### Inverted solar cells

The ZnO precursor solution was prepared according to literature.<sup>[4]</sup> It was spin-coated onto ITO glass (4000 rpm for 30 s). The films were annealed at 200 °C in air for 30 min. ZnO film thickness is ~30 nm. A PTB7-Th:NFA blend in chlorobenzene (CB) with 1-chloronaphthalene (CN) additive was spin-coated onto ZnO layer. MoO<sub>3</sub> (~6 nm) and Ag (~80 nm) was successively evaporated onto the active layer through a shadow mask (pressure ca. 10<sup>-4</sup> Pa). The effective area for the devices is 4 mm<sup>2</sup>. The thicknesses of the active layers were measured by using a KLA Tencor D-120 profilometer. *J-V* curves were measured by using a computerized Keithley 2400 SourceMeter and a Xenon-lamp-based solar simulator (Enli Tech, AM 1.5G, 100 mW/cm<sup>2</sup>). The illumination intensity of solar simulator was determined by using a monocrystalline silicon solar cell (Enli SRC2020, 2cm×2cm) calibrated by NIM. The external quantum efficiency (EQE) spectra were measured by using a QE-R3011 measurement system (Enli Tech).

### Hole-only devices

The structure for hole-only devices is ITO/PEDOT:PSS/PTB7-Th:NFA/MoO<sub>3</sub>/Al. A 30 nm thick PEDOT:PSS layer was made by spin-coating an aqueous dispersion onto ITO glass (4000 rpm for 30 s). PEDOT substrates were dried at 150 °C for 10 min. A PTB7-Th:NFA blend in CB with CN additive was spin-coated onto PEDOT layer. Finally, MoO<sub>3</sub> (~6 nm) and Al (~100 nm) was successively evaporated onto the active layer through a shadow mask (pressure ca. 10<sup>-4</sup> Pa). *J-V* curves were measured by using a computerized Keithley 2400 SourceMeter in the dark.

### Electron-only devices

The structure for electron-only devices is Al/PTB7-Th:NFA/Ca/Al. Al (~80 nm) was evaporated onto a glass substrate. A PTB7-Th:NFA blend in CB with CN additive was spin-coated onto Al. Ca (~5 nm) and Al (~80 nm) were successively evaporated onto the active layer through a shadow mask (pressure ca. 10<sup>-4</sup> Pa). *J-V* curves were measured by using a computerized Keithley 2400 SourceMeter in the dark.

## 7. Optimization of device performance

**Table S1** Optimization of D/A ratio for PTB7-Th:CO5DFIC inverted solar cells.<sup>a</sup>

D/A [w/w]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
1:0.5	0.59	5.25	34.7	1.07 (0.91) <sup>b</sup>
1:1	0.59	12.56	58.0	4.32 (4.01)
1:1.5	0.57	12.32	52.1	3.67 (3.48)
1:2	0.56	12.36	50.2	3.46 (3.28)

<sup>a</sup>Blend solution: 18 mg/mL in CB; spin-coating: 1000 rpm for 60 s.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

**Table S2** Optimization of the active layer thickness for PTB7-Th:CO5DFIC inverted solar cells.<sup>a</sup>

Thickness [nm]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
150	0.57	13.41	43.2	3.31 (3.26) <sup>b</sup>
116	0.59	12.56	58.0	4.32 (4.01)
90	0.60	12.96	64.8	5.03 (4.83)
71	0.60	12.11	65.9	4.82 (4.57)
56	0.60	11.93	64.8	4.65 (4.55)

<sup>a</sup>D/A ratio: 1:1 (w/w); blend solution: 18 mg/mL in CB.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

**Table S3** Optimization of CN content for PTB7-Th:CO5DFIC inverted solar cells.<sup>a</sup>

CN [vol%]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
0	0.60	12.96	64.8	5.03 (4.83) <sup>b</sup>
0.5	0.58	14.13	63.3	5.19 (4.98)
1	0.58	14.33	65.6	5.46 (5.26)
3	0.58	14.19	67.4	5.58 (5.37)
5	0.56	14.35	68.0	5.51 (5.31)
7	0.56	13.80	68.1	5.26 (5.16)

<sup>a</sup>D/A ratio: 1:1 (w/w); blend solution: 18 mg/mL in CB; spin-coating: 1500 rpm for 60 s.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

**Table S4** Optimization of D/A ratio for PTB7-Th:CO5DFIC-OT inverted solar cells.<sup>a</sup>

D/A [w/w]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
1:0.5	0.73	13.92	44.1	4.46 (4.28) <sup>b</sup>
1:1	0.71	17.32	47.9	5.94 (5.55)
1:1.5	0.71	17.49	56.2	6.97 (6.45)
1:2	0.69	15.73	45.2	4.94 (4.71)

<sup>a</sup>Blend solution: 18 mg/mL in CB; spin-coating: 1500 rpm for 60 s.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

**Table S5** Optimization of the active layer thickness for PTB7-Th:CO5DFIC-OT inverted solar cells.<sup>a</sup>

Thickness [nm]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
145	0.71	15.28	49.6	5.37 (5.20) <sup>b</sup>
96	0.71	16.05	53.4	6.10 (5.99)
85	0.71	17.49	56.2	6.97 (6.45)
70	0.71	14.59	56.5	5.88 (5.70)
62	0.72	14.08	56.0	5.65 (5.57)

<sup>a</sup>D/A ratio: 1:1.5 (w/w); blend solution: 18 mg/mL in CB.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

**Table S6** Optimization of CN content for PTB7-Th:CO5DFIC-OT inverted solar cells.<sup>a</sup>

CN [vol%]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
0	0.71	17.49	56.2	6.97 (6.45) <sup>b</sup>
0.5	0.71	17.72	58.0	7.30 (7.11)
1	0.71	18.09	57.9	7.40 (7.18)
3	0.71	17.58	61.0	7.66 (7.37)
5	0.72	16.92	61.2	7.42 (7.14)
7	0.71	16.55	61.6	7.29 (6.98)

<sup>a</sup>D/A ratio: 1:1.5 (w/w); blend solution: 18 mg/mL in CB; spin-coating: 1500 rpm for 60 s.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

**Table S7** Optimization of D/A ratio for PTB7-Th:CO5DFIC-ST inverted solar cells.<sup>a</sup>

D/A [w/w]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
1:0.5	0.75	15.46	43.5	5.02 (4.92) <sup>b</sup>
1:1	0.74	19.23	59.1	8.45 (8.00)
1:1.5	0.75	19.82	59.3	8.78 (8.43)
1:2	0.75	19.21	61.8	8.85 (8.50)
1:2.5	0.74	20.62	59.5	9.07 (8.61)
1:3	0.74	18.53	60.6	8.37 (8.05)

<sup>a</sup>Blend solution: 18 mg/mL in CB with 0.5 vol% CN; spin-coating: 1500 rpm for 60 s.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

**Table S8** Optimization of the active layer thickness for PTB7-Th:CO5DFIC-ST inverted solar cells.<sup>a</sup>

Thickness [nm]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
151	0.73	18.70	49.2	6.74 (6.52) <sup>b</sup>
116	0.74	18.68	60.6	8.36 (8.15)
102	0.74	20.62	59.5	9.07 (8.61)
84	0.74	18.37	62.6	8.47 (8.42)
75	0.73	18.25	62.8	8.42 (8.25)

<sup>a</sup>D/A ratio: 1:2.5 (w/w); blend solution: 18 mg/mL in CB with 0.5 vol% CN.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

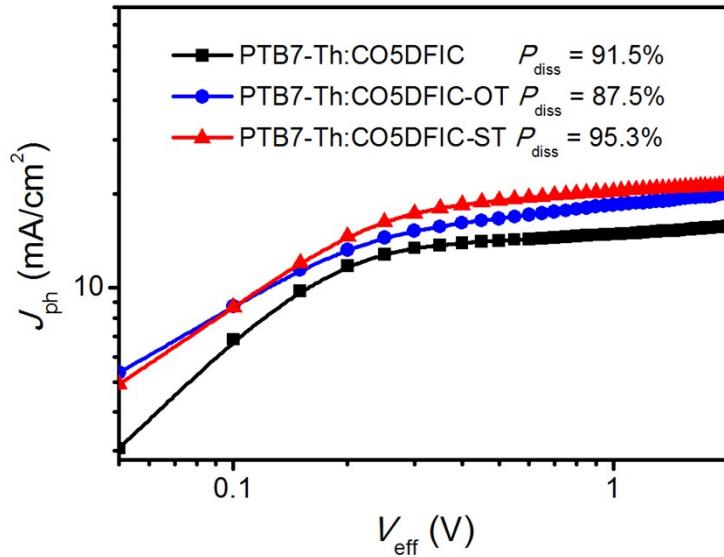
**Table S9** Optimization of CN content for PTB7-Th:CO5DFIC-ST inverted solar cells.<sup>a</sup>

CN [vol%]	$V_{oc}$ [V]	$J_{sc}$ [mA/cm <sup>2</sup> ]	FF [%]	PCE [%]
0	0.74	19.58	59.5	8.58 (8.29) <sup>b</sup>
0.5	0.74	20.62	59.5	9.07 (8.61)
1	0.73	20.79	60.0	9.15 (9.09)
3	0.74	20.71	63.7	9.73 (9.37)
5	0.74	20.53	63.7	9.63 (9.32)
7	0.74	20.12	63.4	9.48 (9.10)

<sup>a</sup>D/A ratio: 1:2.5 (w/w); blend solution: 18 mg/mL in CB; spin-coating: 1500 rpm for 60 s.

<sup>b</sup>Data in parentheses stand for the average PCEs for 10 cells.

## 8. Exciton dissociation probabilities



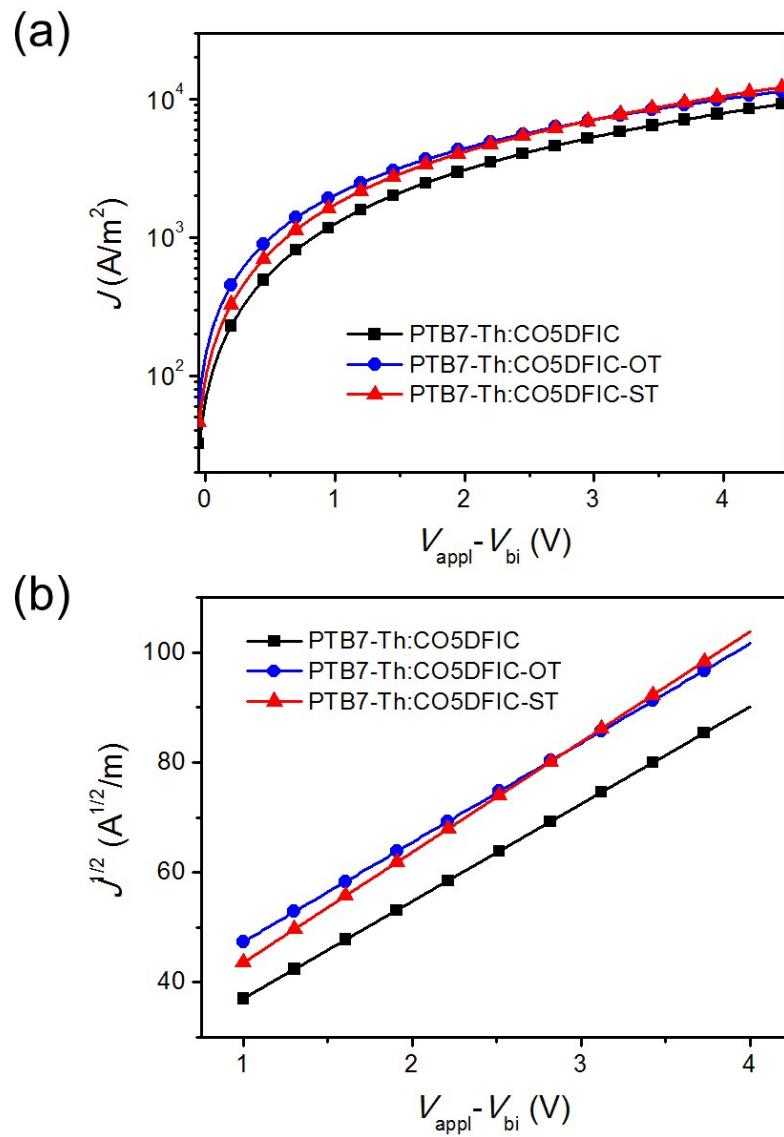
**Fig. S17**  $J_{\text{ph}}$ - $V_{\text{eff}}$  plots.

## 9. SCLC

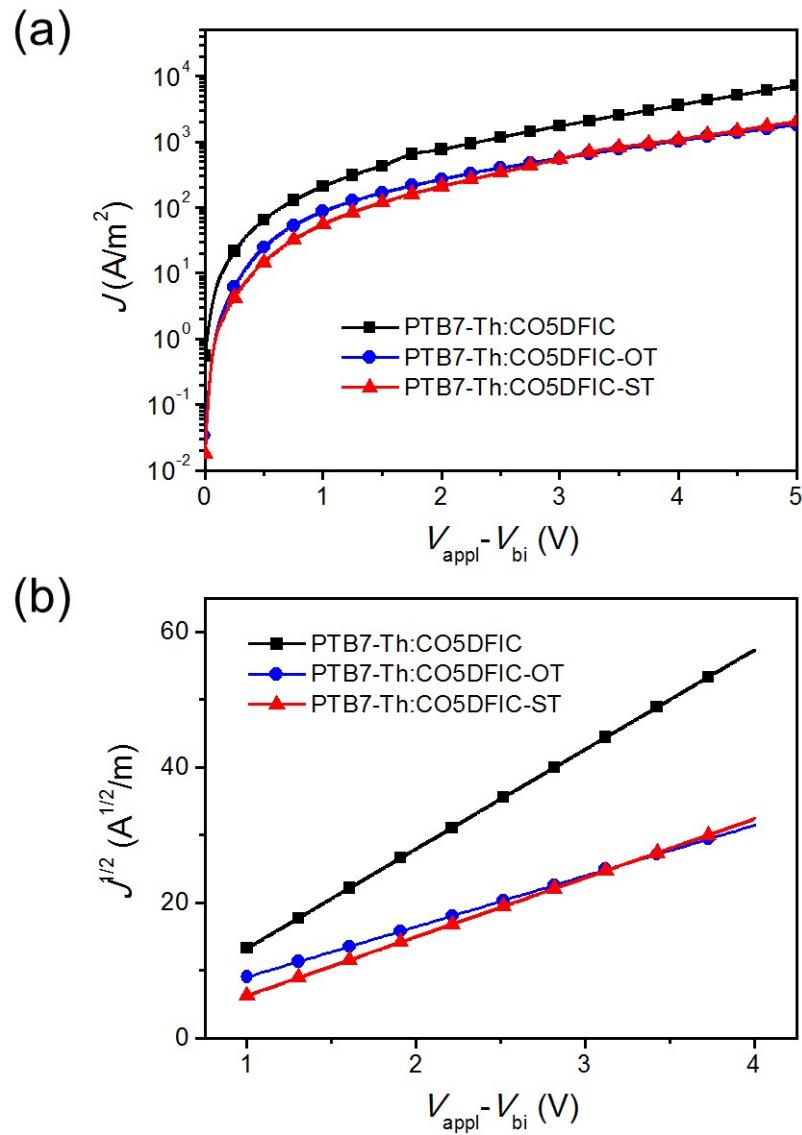
Charge carrier mobility was measured by SCLC method. The mobility was determined by fitting the dark current to the model of a single carrier SCLC, which is described by:

$$J = \frac{9}{8} \varepsilon_0 \varepsilon_r \mu \frac{V^2}{d^3}$$

where  $J$  is the current density,  $\mu$  is the zero-field mobility of holes ( $\mu_h$ ) or electrons ( $\mu_e$ ),  $\varepsilon_0$  is the permittivity of the vacuum,  $\varepsilon_r$  is the relative permittivity of the material,  $d$  is the thickness of the blend film, and  $V$  is the effective voltage ( $V = V_{\text{appl}} - V_{\text{bi}}$ , where  $V_{\text{appl}}$  is the applied voltage, and  $V_{\text{bi}}$  is the built-in potential determined by electrode work function difference). Here,  $V_{\text{bi}} = 0.1$  V for hole-only devices,  $V_{\text{bi}} = 0$  V for electron-only devices.<sup>[5]</sup> The mobility was calculated from the slope of  $J^{1/2}$ - $V$  plots.



**Fig. S18**  $J$ - $V$  curves (a) and corresponding  $J^{1/2}$ - $V$  plots (b) for the hole-only devices (in dark). The thicknesses for PTB7-Th:CO5DFIC, PTB7-Th:CO5DFIC-OT and PTB7-Th:CO5DFIC-ST blend films are 90 nm, 84 nm and 100 nm, respectively.

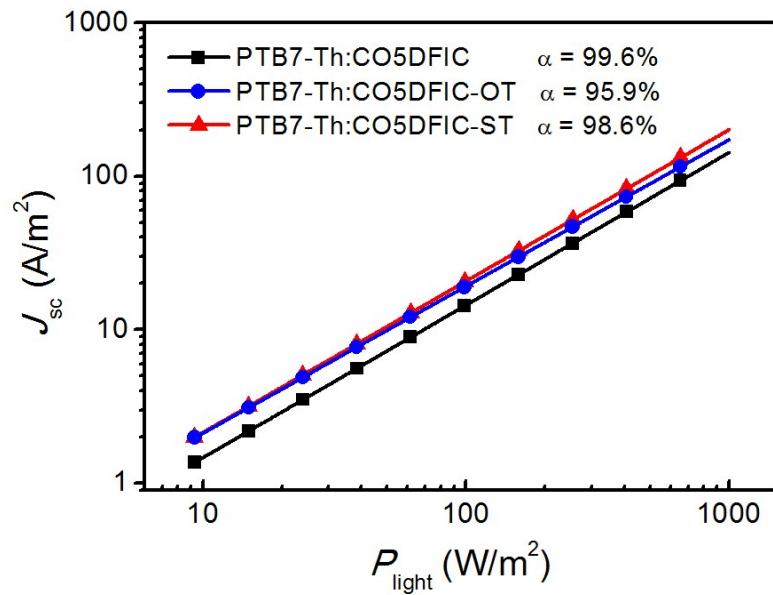


**Fig. S19**  $J$ - $V$  curves (a) and corresponding  $J^{1/2}$ - $V$  plots (b) for the electron-only devices (in dark). The thicknesses for PTB7-Th:CO5DFIC, PTB7-Th:CO5DFIC-OT and PTB7-Th:CO5DFIC-ST blend films are 90 nm, 84 nm and 100 nm, respectively.

**Table S10** Hole and electron mobilities for PTB7-Th:NFA blend films.

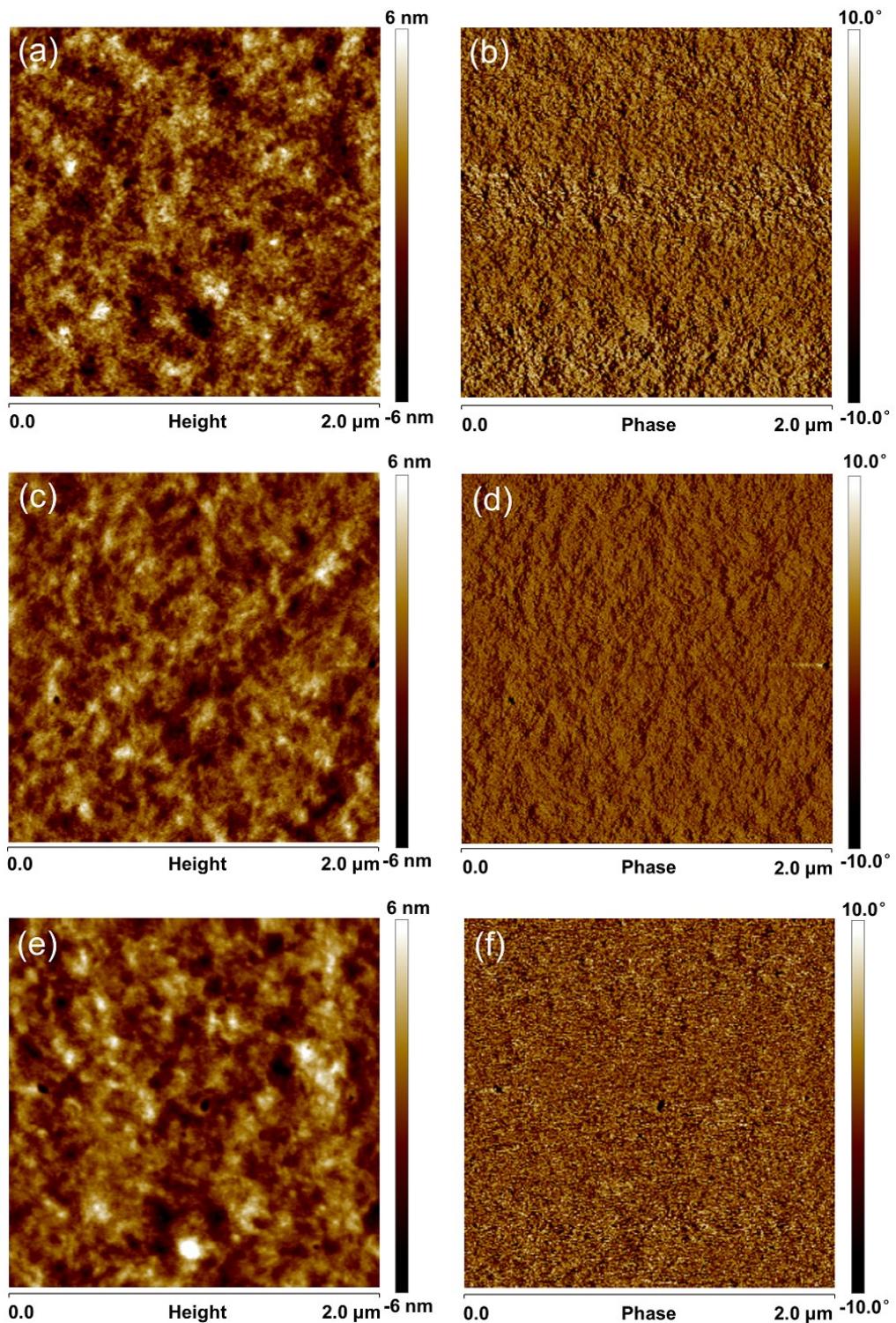
PTB7-Th:NFA	$\mu_h$ [cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> ]	$\mu_e$ [cm <sup>2</sup> V <sup>-1</sup> s <sup>-1</sup> ]	$\mu_h/\mu_e$
PTB7-Th:CO5DFIC	$7.53 \times 10^{-5}$	$4.09 \times 10^{-5}$	1.8
PTB7-Th:CO5DFIC-OT	$7.86 \times 10^{-5}$	$8.60 \times 10^{-6}$	9.1
PTB7-Th:CO5DFIC-ST	$1.32 \times 10^{-4}$	$1.97 \times 10^{-5}$	6.7

## 10. Bimolecular recombination



**Fig. S20**  $J_{\text{sc}}$ - $P_{\text{light}}$  plots.

## 11. AFM



**Fig. S21** AFM height (left) and phase (right) images for the blend films. (a) and (b), PTB7-Th:CO5DFIC film with 3 vol% CN ( $R_{\text{rms}} = 3.67 \text{ nm}$ ); (c) and (d), PTB7-Th:CO5DFIC-OT film with 3 vol% CN ( $R_{\text{rms}} = 1.86 \text{ nm}$ ); (e) and (f), PTB7-Th:CO5DFIC-ST film with 3 vol% CN ( $R_{\text{rms}} = 2.69 \text{ nm}$ ).  $R_{\text{rms}}$ : root-mean-square roughness.

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