

Electronic Supplementary Information

N-(2-Fluoro-2,2-dinitroethyl)azoles: Novel assembly of diverse explosophoric building blocks for energetic compounds design

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General Information

All the reagents were of analytical grade, purchased from commercial sources, and used as received. Infrared spectra were determined in KBr pellets on a BrukerALPHA spectrometer. The ¹H, ¹⁹F, ¹³C, ¹⁴N and ¹⁵N NMR spectra were recorded on a Bruker AM-300 instrument at 300.13, 282.00, 75.47, 21.68 and 30.42 MHz, respectively. The chemical shift values (δ) are expressed relative to the chemical shift of the [D]solvent or to external standard without correction nitromethane (¹⁴N) and CClF₃ (¹⁹F). Analytical TLC was performed using commercially pre-coated silica gel plates (Silufol UV₂₅₄), and visualization was effected with short-wavelength UV-light. Melting points were determined on Gallenkamp melting point apparatus and they are uncorrected.

CAUTION! All compounds are potentially explosive and should be handled with appropriate precautions.

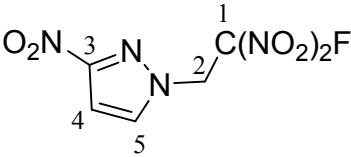
Characterization of Compounds 12-29

General Procedure for the Preparation of K-salts of N-(dinitroethyl)azoles. A mixture of a NH-azole (2.5 mmol) and t-BuOK (5 mmol) in methanol (10 mL) was stirred for 5 min under reflux. 1,1,1-Trinitroethan (2.6 mmol) in methanol (10 mL) was then added dropwise, keeping the flask temperature about 65 °C. After the addition was complete, the reaction mixture was heated under reflux. The reaction was stirred until the starting azole could no longer be detected by thin-layer chromatography, which typically took 1-2 h. The mixture was then cooled to 20 °C, and the yellow-orange solid which separated was collected by filtration and washed with methanol (5 mL) and diethyl ether (3×5 mL), and dried to give crude K-salt **3**. The crude compound was used directly in the next step without any further purification. If necessary, a pure sample could be prepared for analysis by recrystallization; however, the purification is accompanied by large losses.

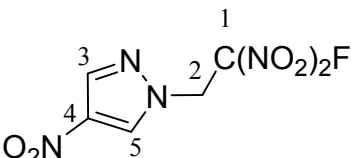
General Procedure for the Preparation of Compounds 12-29.

Selectfluor™ was added with stirring to a slurry of crude K-salt (2 mmol; from the previous stage) in acetonitrile (10 mL) at room temperature. After 1 h of stirring, the reaction mixture was diluted with CH₂Cl₂ (200 mL), washed with brine (2×50 mL), passed through silica gel, and dried over MgSO₄. The solution was evaporated and the residue was purified by recrystallization from a suitable solvent or column chromatography on silica gel to give fluorinated product as a white solid:

1-(2-Fluoro-2,2-dinitroethyl)-3-nitro-1H-pyrazole, **12**

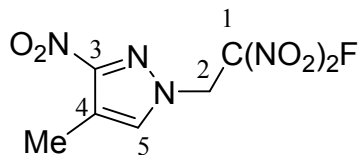
 The compound was prepared followed by **General Procedure**. After recrystallization from CHCl₃ colorless solid **12** was obtained in 68% yield; mp 88-89°C. ¹H NMR (DMSO-*d*₆): 6.18 (d, *J*_{HF} = 15.3 Hz, 2H, NCH₂), 7.17 (s, 1H, CH), 8.16 (s, 1H, CH); ¹³C NMR (DMSO-*d*₆): 52.2 (d, *J*_{CF} = 17.6 Hz, C2), 100.9 (C5), 119.4 (d, *J*_{CF} = 293.9 Hz, C1), 137.2 (C4), 156.9 (C3); ¹⁹F NMR (DMSO-*d*₆): 108.95 (t, 1F, *J* = 14.32 Hz); IR: ν = 3164, 3131, 3035, 2976, 2896, 1618, 1553, 1510, 1458, 1429, 1381, 1355, 1317, 1302, 1220, 1187, 1078, 1061, 1027, 998, 888, 849, 827, 810, 784, 755, 736, 618, 593, 447 cm⁻¹; Anal. calcd. (%) for C₅H₄FN₅O₆ (249,11): C 24.11, H 1.62, N 28.11; found: C 24.31; H 1.74; N 28.20.

1-(2-Fluoro-2,2-dinitroethyl)-4-nitro-1H-pyrazole, **13**

 The compound was prepared followed by **General Procedure**. After recrystallization from CH₂Cl₂ colorless solid **13** was obtained in 68% yield; mp 101-102°C. ¹H NMR (DMSO-*d*₆): 6.11 (d, 2H, *J*_{HF} = 15.1 Hz, NCH₂), 8.45 (s, 1H, CH), 9.01 (s, 1H, CH); ¹³C NMR (DMSO-*d*₆): 51.8 (d, *J*_{CF} = 17.6 Hz, C2), 119.2 (d, *J*_{CF} = 294.0 Hz, C1), 133.2 (C5), 135.9 (C4), 137.6 (C3); ¹⁹F NMR (DMSO-*d*₆): 109.14 (t, 1F, *J* = 14.86 Hz); ¹⁴N NMR (DMSO-*d*₆): -18.85 (bs, NO₂); ¹⁵N NMR (DMSO-*d*₆): -182.2 (NCH₂), -70.8 (C=N-N), -21.5 (C(NO₂) F), -21.0 (C(NO₂)F), -18.0 (NO₂); IR: ν = 3435, 3147, 3129, 3026, 2984, 1624, 1604, 1536, 1510, 1413, 1332, 1309, 1210, 1191, 1139, 1082, 1003, 966, 885, 851, 822, 806, 756, 740, 658, 613, 586,

548, 451 cm^{-1} ; Anal. calcd. (%) for $\text{C}_5\text{H}_4\text{FN}_5\text{O}_6$ (249,11): C 24.11, H 1.62, N 28.11; found C 24.19; H 1.71; N 28.03.

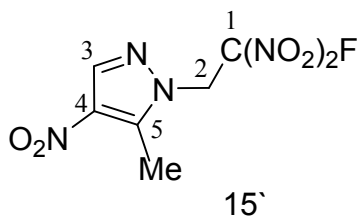
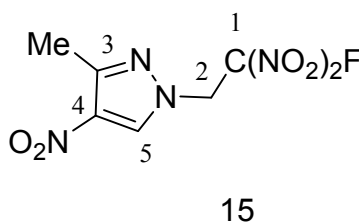
1-(2-Fluoro-2,2-dinitroethyl)-4-methyl-3-nitro-1H-pyrazole, **14**



The compound was prepared followed by **General Procedure**. After recrystallization from CH_2Cl_2 colorless solid **13** was obtained in 32% yield; mp 100-101°C. ^1H NMR ($\text{DMSO}-d_6$): 2.27 (s, 3H, CH_3), 6.09 (d, 2H, $J_{\text{HF}} = 15.3$ Hz, NCH_2), 7.98 (s, 1H, CH); ^{13}C NMR ($\text{DMSO}-d_6$): 9.12 (s, CH_3), 52.1 (d, $J_{\text{CF}} = 17.4$ Hz, C2), 119.3 (d, $J_{\text{CF}} = 293.5$ Hz, C1), 114.9 (C4), 127.3 (C5), 135.8 (C3); ^{19}F NMR ($\text{DMSO}-d_6$): -109.12 (t, 1F, $J = 15.09$ Hz); ^{14}N NMR ($\text{DMSO}-d_6$): -18.70 (bs, NO_2); ^{15}N NMR ($\text{DMSO}-d_6$): -189.0 (NCH_2), -76.7 ($\text{C}=\text{N}-\text{N}$), -21.5 ($\text{C}(\text{NO}_2)\text{F}$), -21.0 ($\text{C}(\text{NO}_2)\text{F}$), -18.1; IR: $\nu = 3421, 3121, 3031, 3002, 2979, 1622, 1591, 1525, 1425, 1377, 1337, 1305, 1220, 1187, 1148, 1076, 1027, 1000, 891, 849, 814, 777, 758, 627, 591, 522$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_6\text{H}_6\text{FN}_5\text{O}_6$ (263,14): C 27.39; H 2.30; N 26.61; found C 27.31; H 2.35; N 26.51.

1-(2-Fluoro-2,2-dinitroethyl)-5-methyl-4-nitro-1H-pyrazole, **15**

1-(2-Fluoro-2,2-dinitroethyl)-3-methyl-4-nitro-1H-pyrazole, **15'**



The mixture of isomers **15** and **15'** is not separated (ratio 1/1). The mixture of compounds **15** and **15'** was prepared followed by **General Procedure**. After recrystallization from CH_2Cl_2 colorless solid **15** and

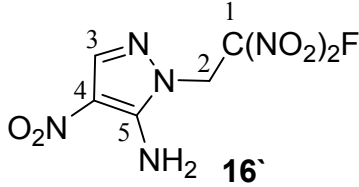
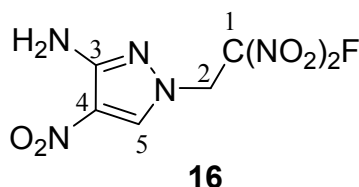
15' was obtained in 65% yield; mp 85-90°C.

NMR data for the mixture of isomers **15** and **15'**:

^1H NMR ($\text{DMSO}-d_6$): 2.39 (s, 3H, CH_3 , **15'**), 2.69 (s, 3H, CH_3 , **15**), 5.93 (d, 2H, $J_{\text{HF}} = 2.1$ Hz, NCH_2 , **15'**), 5.98 (d, 2H, $J_{\text{HF}} = 2.2$ Hz, NCH_2 , **15**), 8.28 (s, 1H, CH , **15'**), 8.85 (s, 1H, CH , **15**); ^{13}C NMR ($\text{DMSO}-d_6$): 10.54 (s, CH_3 , **15'**), 12.87 (s, CH_3 , **15**), 49.52 (d, $J_{\text{CF}} = 17.45$ Hz, C2, **15'**), 51.6 (d, $J_{\text{CF}} = 17.68$ Hz, C2, **15**), 119.24 (d, $J_{\text{CF}} = 294.04$ Hz, C1, **15+15'**), 133.34 (C4, **15**), 133.85 (C4, **15'**), 133.99 (C3, **15'**), 134.46 (C5, **15**), 143.24 (C5, **15'**), 147.07 (C3, **15**); ^{19}F NMR ($\text{DMSO}-d_6$): -109.12 (t, 1F, $J = 14.86$ Hz), -111.02 (t, 1F, $J = 14.75$ Hz); ^{14}N NMR ($\text{DMSO}-d_6$): -23.18 (bs, NO_2); ^{15}N NMR ($\text{DMSO}-d_6$): -182.8 (NCH_2 , **15'**), -189.6 (NCH_2 , **15**), -78.1 ($\text{C}=\text{N}-\text{N}$, **15**), -75.5 ($\text{C}=\text{N}-\text{N}$, **15'**), -21.2 ($\text{C}(\text{NO}_2)\text{F}$), -21.3 ($\text{C}(\text{NO}_2)\text{F}$), -20.8 ($\text{C}(\text{NO}_2)\text{F}$), -20.9 ($\text{C}(\text{NO}_2)\text{F}$), -16.4 (NO_2 , **15**), -16.1 (NO_2 , **15'**).

1-(2-Fluoro-2,2-dinitroethyl)-4-nitro-1H-pyrazol-3-amine, **16**

1-(2-Fluoro-2,2-dinitroethyl)-4-nitro-1H-pyrazol-5-amine, **16'**



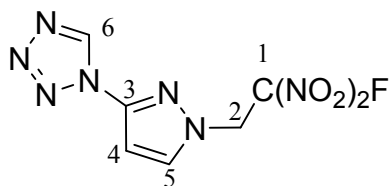
The mixture of compounds **16** and **16'** was prepared followed by **General Procedure**. Solid **16** and **16'** were obtained in 38% yield

(ratio 1/1). The mixture of isomers **16** and **16'** is separated by column chromatography on silica gel (Eluent is CHCl₃, R_f (**16**) = 0.79, R_f (**16'**) = 0.45).

After purified by silica gel chromatography, yellow solid **16** was obtained in 18% yield; mp 167-168°C (CHCl₃). ¹H NMR (DMSO-*d*₆): 5.76 (d, 2H, J_{HF} = 14.40 Hz, NCH₂), 7.91 (s, 2H, NH₂), 8.08 (s, 1H, CH); ¹³C NMR (DMSO-*d*₆): 47.96 (d, J_{CF} = 17.7 Hz, C2), 119.0 (d, J_{CF} = 291.92 Hz, C1), 117.3 (C4), 137.2 (C5), 147.7 (C3); ¹⁹F NMR (DMSO-*d*₆): -110.31 (t, 1F, J = 14.33 Hz); ¹⁴N NMR (DMSO-*d*₆): -20.58 (bs, NO₂); IR: ν = 3467, 3341, 3251, 3146, 3017, 2964, 1641, 1605, 1583, 1506, 1474, 1340, 1318, 1278, 1209, 1082, 848, 807, 763, 700, 666, 592, 547, 498 cm⁻¹; Anal. calcd. (%) for C₅H₅FN₆O₆ (264,13): C 22.74; H 1.91; N 31.82; found C 22.87; H 1.82; N 31.71.

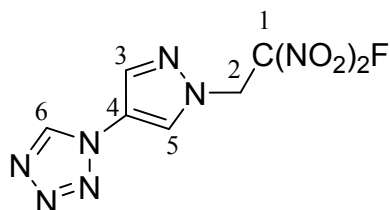
After purified by silica gel chromatography, colorless solid **16'** was obtained in 20% yield; mp 189-190 °C (CHCl₃). ¹H NMR (DMSO-*d*₆): 5.79 (d, 2H, J_{HF} = 14.90 Hz, NCH₂), 6.30 (bs, 2H, NH₂), 8.62 (s, 1H, CH); ¹³C NMR (DMSO-*d*₆): 51.7 (d, J_{CF} = 17.54 Hz, C2), 119.2 (d, J_{CF} = 293.55 Hz, C1), 122.2 (C4), 133.2 (C3), 151.8 (C5); ¹⁹F NMR (DMSO-*d*₆): -109.56 (t, 1F, J = 14.52 Hz); ¹⁴N NMR (DMSO-*d*₆): -22.38 (bs, NO₂); ¹⁵N NMR (DMSO-*d*₆): -322.8 (NH₂), -201.7 (NCH₂), -199.6 (C=N-N), -21.2 (C(NO₂)F), -20.1 (C(NO₂)F), -15.0 (NO₂); IR: ν = 3467, 3341, 3146, 3017, 2964, 1641, 1605, 1583, 1506, 1474, 1415, 1403, 1340, 1318, 1278, 1209, 1082, 1002, 980, 848, 807, 763, 700, 666, 592, 547, 498 cm⁻¹; Anal. calcd. (%) for C₅H₅FN₆O₆ (264,13): C 22.74; H 1.91; N 31.82; found C 22.67; H 2.02; N 31.95.

1-(1-(2-Fluoro-2,2-dinitroethyl)-1H-pyrazol-3-yl)-1H-tetrazole, **17**



The compound was prepared followed by **General Procedure**. After recrystallization from CCl₄ colorless solid **17** was obtained in 55% yield; mp 121°C. ¹H NMR (DMSO-*d*₆): 6.11 (d, 2H, J_{HF} = 15.5 Hz, NCH₂), 6.97 (d, 1H, J = 1.7 Hz, CH), 8.17 (d, 1H, J = 1.7 Hz, CH-Pz), 9.96 (s, 1H, c, CH-Tz); ¹³C NMR (DMSO-*d*₆): 51.8 (d, J_{CF} = 17.6 Hz, C2), 119.4 (d, J_{CF} = 293.2 Hz, C1), 100.6 (C5), 136.4 (C4), 142.4 (C6), 144.3 (C3); ¹⁹F NMR (DMSO-*d*₆): -109.29 (t, 1F, J = 15.37 Hz); ¹⁴N NMR (DMSO-*d*₆): -22.85 (bs, NO₂); IR: ν = 3155, 3121, 3025, 2974, 1617, 1549, 1351, 1304, 1265, 1207, 1122, 1092, 1077, 1018, 962, 888, 849, 810, 787, 729, 658, 626, 588 cm⁻¹; Anal. calcd. (%) for C₆H₅FN₈O₄ (272,16): C 26.48; H 1.85; N 41.17; found C 26.35; H 1.94; N 41.37.

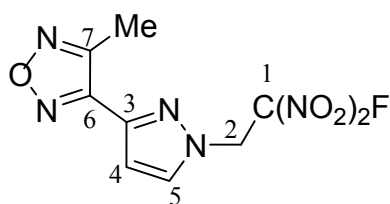
1-(1-(2-Fluoro-2,2-dinitroethyl)-1H-pyrazol-4-yl)-1H-tetrazole, **18**



The compound was prepared followed by **General Procedure**. After recrystallization from CCl₄ colorless solid **18** was obtained in 55% yield; mp 109-110°C. ¹H NMR (DMSO-*d*₆): 6.15 (d, 2H, J_{HF} = 15.2 Hz, NCH₂), 8.27 (s, 1H, CH-Pz), 8.66 (s, 1H, CH-Pz), 9.93 (s, 1H, CH-Tz); ¹³C NMR (DMSO-*d*₆): 51.8 (d, J_{CF} = 17.6 Hz, C2), 119.5 (d, J_{CF} = 292.8 Hz, C1), 118.8 (C4), 127.3 (C5), 135.3 (C3), 143.3 (C6); ¹⁹F NMR (DMSO-*d*₆): -109.67 (t, 1F, J = 15.08 Hz); ¹⁴N NMR (DMSO-*d*₆): -17.38 (bs, NO₂); ¹⁵N NMR (DMSO-*d*₆): -186.6 (NCH₂), -149.7, -72.1, -49.2, -21.1 (C(NO₂)F), -20.6 (C(NO₂)F), -14.5, 13.6; IR: ν = 3139, 3046, 3018, 2971, 1614, 1593, 1482,

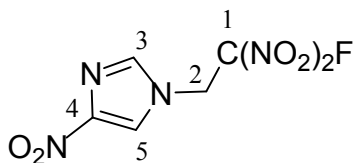
1316, 1298, 1261, 1165, 1098, 1075, 971, 945, 863, 849, 803, 736, 626, 584 cm^{-1} ; Anal. calcd. (%) for $\text{C}_6\text{H}_5\text{FN}_8\text{O}_4$ (272,16): C 26.48; H 1.85; N 41.17; found C 26.57; H 1.74; N 41.26.

3-(1-(2-Fluoro-2,2-dinitroethyl)-1H-pyrazol-3-yl)-4-methyl-1,2,5-oxadiazole, 19



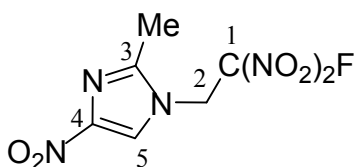
The compound was prepared followed by **General Procedure**. After recrystallization from CCl_4 colorless solid **19** was obtained in 43% yield; mp 132°C . ^1H NMR ($\text{DMSO}-d_6$): 2.47 (s, 3H, CH_3), 6.05 (d, 2H, $J_{\text{HF}} = 15.30$ Hz, NCH_2), 6.94 (d, 1H, CH, $J_{\text{HF}} = 2.1$ Hz), 8.08 (d, 1H, CH, $J_{\text{HF}} = 2.4$ Hz); ^{13}C NMR ($\text{DMSO}-d_6$): 10.5 (s, CH_3), 52.6 (d, $J_{\text{CF}} = 17.63$ Hz, C2), 108.1 (C4), 120.60 (d, $J_{\text{CF}} = 292.97$ Hz, C1), 136.3 (C5), 141.9 (C3), 148.9 (C6), 151.3 (C7); ^{19}F NMR ($\text{DMSO}-d_6$): -110.11 (t, 1F, $J = 15.10$ Hz); ^{14}N NMR ($\text{DMSO}-d_6$): -22.70 (bs, NO_2); ^{15}N NMR ($\text{DMSO}-d_6$): -322.8 (NH_2), -183.1 (NCH_2), -74.6 ($\text{C}=\text{N}-\text{N}$), -20.8 ($\text{C}(\text{NO}_2)\text{F}$), -20.2 ($\text{C}(\text{NO}_2)\text{F}$), 27.7 (N-furazan ring), 28.9 (N-furazan ring); IR: $\nu = 3159, 3138, 3022, 2983, 1619, 1542, 1507, 1429, 1415, 1395, 1364, 1348, 1333, 1312, 1298, 1243, 1215, 1114, 1074, 1040, 1013, 989, 951, 893, 852, 812, 778, 729, 682, 668, 623, 604, 589, 566, 495$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_8\text{H}_7\text{FN}_6\text{O}_5$ (286,18): 33.58; H 2.47; N 29.37; found C 33.63; H 2.39; N 29.46.

1-(2-Fluoro-2,2-dinitroethyl)-4-nitro-1H-imidazole, 21

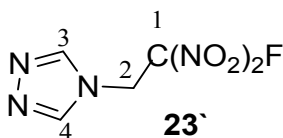
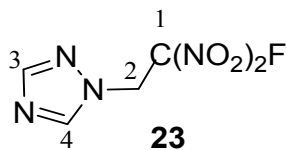


The compound was prepared followed by **General Procedure**. After recrystallization from CH_2Cl_2 colorless solid **21** was obtained in 70% yield; mp $129-130^\circ\text{C}$. ^1H NMR ($\text{DMSO}-d_6$): 5.96 (d, 2H, $J_{\text{HF}} = 16.0$ Hz, NCH_2), 7.94 (s, 1H, CH), 8.38 (s, 1H, CH); ^{13}C NMR ($\text{DMSO}-d_6$): 47.2 (d, $J_{\text{CF}} = 18.05$ Hz, C2), 119.3 (d, $J_{\text{CF}} = 292.59$ Hz, C1), 122.1 (C5), 138.5 (C3), 147.5 (C4); ^{19}F NMR ($\text{DMSO}-d_6$): -107.11 (t, 1F, $J = 15.62$ Hz); ^{14}N NMR ($\text{DMSO}-d_6$): -23.39 (bs, NO_2); ^{15}N NMR ($\text{DMSO}-d_6$): -17.6 (NO_2), -21.1 ($\text{C}(\text{NO}_2)\text{F}$), -21.2 ($\text{C}(\text{NO}_2)\text{F}$), -123.0 ($\text{C}-\text{N}=\text{C}$), -217.3 (NCH_2); IR: $\nu = 3144, 3018, 2961, 1613, 1550, 1521, 1493, 1384, 1348, 1290, 1248, 1216, 1140, 1076, 1042, 983, 900, 852, 826, 809, 752, 709, 670, 602, 566$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_5\text{H}_4\text{FN}_5\text{O}_6$ (249,11): C 24.11; H 1.62; N 28.11; found C 24.23; H 1.71; N 28.02.

1-(2-Fluoro-2,2-dinitroethyl)-2-methyl-4-nitro-1H-imidazole, 22



The compound was prepared followed by **General Procedure**. After recrystallization from CH_2Cl_2 colorless solid **22** was obtained in 36% yield; mp 159°C . ^1H NMR ($\text{DMSO}-d_6$): 2.4 (s, 3H, CH_3), 5.96 (d, 2H, $J_{\text{HF}} = 16.3$ Hz, NCH_2), 8.21 (s, 1H, CH); ^{13}C NMR ($\text{DMSO}-d_6$): 12.49 (d, $J_{\text{CF}} = 2.88$ Hz, CH_3), 46.5 (d, $J_{\text{CF}} = 17.63$ Hz, C2), 119.5 (d, $J_{\text{CF}} = 292.64$ Hz, C1), 122.2 (C5), 146.1 (C3), 146.8 (C4); ^{19}F NMR ($\text{DMSO}-d_6$): -108.25 (t, 1F, $J = 15.47$ Hz); ^{14}N NMR ($\text{DMSO}-d_6$): -19.12 (bs, NO_2); ^{15}N NMR ($\text{DMSO}-d_6$): -219.8 (NCH_2), -126.2 ($\text{C}-\text{N}=\text{C}$), -21.5 ($\text{C}(\text{NO}_2)\text{F}$), -21.0 ($\text{C}(\text{NO}_2)\text{F}$), -17.3 (NO_2); IR: $\nu = 3140, 3041, 2954, 1616, 1513, 1439, 1412, 1366, 1344, 1310, 1279, 1210, 1141, 1072, 1008, 991, 955, 907, 880, 851, 814, 765, 725, 676, 644, 594, 530$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_6\text{H}_6\text{FN}_5\text{O}_6$ (263,14): C 27.39; H 2.30; N 26.61; found C 27.25; H 2.27; N 26.74.

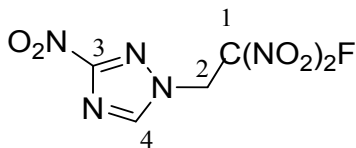
1-(2-Fluoro-2,2-dinitroethyl)-1H-1,2,4-triazole, 23**4-(2-Fluoro-2,2-dinitroethyl)-4H-1,2,4-triazole, 23'**

The mixture of compounds **23** and **23'** was prepared followed by **General Procedure** in 62% yield.

NMR data for the mixture of isomers **23** and **23'** (ratio 10/1):

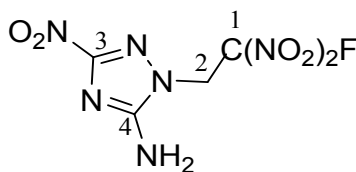
^1H NMR (DMSO- d_6): 5.59 (2H, d, $J_{\text{HF}} = 15.32$ Hz, NCH_2 , **23'**), 6.10 (2H, d, $J_{\text{HF}} = 15.30$ Hz, NCH_2 , **23**), 8.10 (1H, c, CH , **23**), 8.60 (1H, c, CH , **23'**), 8.70 (1H, c, CH , **23**);

After recrystallization from CH_2Cl_2 colorless solid **23** was obtained in 50% yield; mp 83°C. ^1H NMR (DMSO- d_6): 6.10 (d, 2H, $J_{\text{HF}} = 15.30$ Hz, NCH_2), 8.10 (s, 1H, CH), 8.70 (s, 1H, CH); ^{13}C NMR (DMSO- d_6): 49.24 (d, $J_{\text{CF}} = 17.93$ Hz, C2), 119.6 (d, $J_{\text{CF}} = 292.74$ Hz, C1), 147.1 (C5), 153.5 (C3); ^{19}F NMR (DMSO- d_6): -109.50 (t, 1F, $J = 14.89$ Hz); ^{14}N NMR (DMSO- d_6): -23.62 (bs, NO_2); ^{15}N NMR (DMSO- d_6): -188.27 (NCH_2), -123.35, -84.99, -21.20 ($\text{C}(\text{NO}_2)\text{F}$), -20.70 ($\text{C}(\text{NO}_2)\text{F}$); IR: $\nu = 3140, 3041, 2954, 2899, 1616, 1594, 1513, 1439, 1412, 1383, 1366, 1344, 1329, 1310, 1293, 1279, 1210, 1193, 1141, 1072, 1008, 991, 955, 907, 880, 851, 814, 765, 725, 676, 644, 594, 530$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_4\text{H}_4\text{FN}_5\text{O}_4$ (205,11): C 23.42; H 1.97; N 34.15; found C 23.34; H 2.05; N 34.24.

1-(2-Fluoro-2,2-dinitroethyl)-3-nitro-1H-1,2,4-triazole, 24

The compound was prepared followed by **General Procedure**.

After recrystallization from CH_2Cl_2 colorless solid **24** was obtained in 9% yield; mp 217°C; ^1H NMR (DMSO- d_6): 6.30 (d, 2H, $J_{\text{HF}} = 15.10$ Hz, NCH_2), 9.02 (s, 1H, CH); ^{13}C NMR (DMSO- d_6): 50.27 (d, $J_{\text{CF}} = 17.82$ Hz, C2), 119.2 (d, $J_{\text{CF}} = 293.93$ Hz, C1), 150.0 (C4), 163.3 (C3); ^{19}F NMR (DMSO- d_6): -108.41 (t, 1F, $J = 14.76$ Hz); ^{14}N NMR (DMSO- d_6): -24.14 (bs, NO_2); ^{15}N NMR (DMSO- d_6): -174.78 (NCH_2), -131.40, -86.17, -25.90 ($\text{C}(\text{NO}_2)\text{F}$), -22.31 ($\text{C}(\text{NO}_2)\text{F}$), -21.81 (NO_2); IR: $\nu = 3135, 3015, 2970, 2361, 2343, 1608, 1595, 1562, 1510, 1425, 1384, 1466, 1313, 1291, 1216, 1080, 1029, 884, 839, 812, 765, 740, 647, 581$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_4\text{H}_3\text{FN}_6\text{O}_6$ (250,10): C 19.21; H 1.21; N 33.60; found C 19.30; H 1.11; N 33.71.

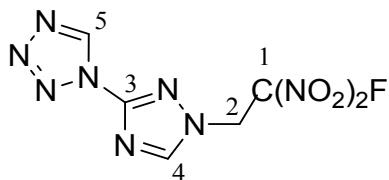
1-(2-Fluoro-2,2-dinitroethyl)-3-nitro-1H-1,2,4-triazol-5-amine, 26

The compound was prepared followed by **General Procedure**.

After recrystallization from $\text{CHCl}_3/\text{MeCN}$ 5/1 colorless solid **26** was obtained in 15% yield; mp 218°C; ^1H NMR (DMSO- d_6): 5.89 (2H, d, $J_{\text{HF}} = 14.80$ Hz, NCH_2), 7.56 (2H, c, NH_2); ^{13}C NMR (DMSO- d_6): 47.4 (d, $J_{\text{CF}} = 17.43$ Hz, C2), 118.7 (d, $J_{\text{CF}} = 293.26$ Hz, C1), 158.0 (C4), 160.4 (C5); ^{19}F NMR (DMSO- d_6): -109.37 (1F, t, $J = 14.00$ Hz); ^{14}N NMR (DMSO- d_6): -23.67 (yIII, c, NO_2); ^{15}N NMR (DMSO- d_6): -319.05 (NH_2), -214.33, -176.33 (NCH_2), -104.37, -23.39 (NO_2), -21.32 ($\text{C}(\text{NO}_2)\text{F}$), -20.81 ($\text{C}(\text{NO}_2)\text{F}$); IR: $\nu = 3415, 3314, 3246, 3153, 1655, 1609, 1533, 1423, 1317, 1271, 1159, 1070$,

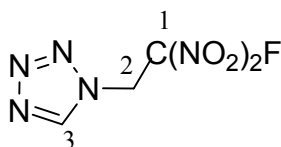
1000, 867, 849, 816, 729, 709, 629 cm^{-1} ; Anal. calcd. (%) for $\text{C}_4\text{H}_4\text{FN}_7\text{O}_6$ (265,12): C 18.12; H 1.52; N 36.98; found C 18.01; H 1.63; N 37.05.

1-(1-(2-Fluoro-2,2-dinitroethyl)-1H-1,2,4-triazol-3-yl)-1H-tetrazole, 27



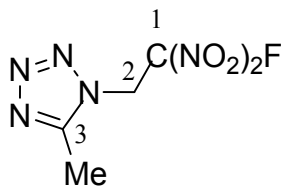
The compound was prepared followed by **General Procedure**. After recrystallization from CHCl_3 colorless solid **27** was obtained in 15% yield; mp 130°C ; ^1H NMR ($\text{DMSO}-d_6$): 6.24 (d, 2H, $J_{\text{HF}} = 14.80$ Hz, NCH_2), 9.01 (s, 1H, CH), 10.08 (s, 1H, CH); ^{13}C NMR ($\text{DMSO}-d_6$): 49.5 (d, $J_{\text{CF}} = 17.84$ Hz, C2), 118.8 (d, $J_{\text{CF}} = 293.92$ Hz, C1), 143.1 (C3), 149.1 (C5), 153.4 (C4); ^{19}F NMR ($\text{DMSO}-d_6$): -108.74 (t, 1F, $J = 14.84$ Hz); ^{14}N NMR ($\text{DMSO}-d_6$): -24.08 (bs, NO_2); ^{15}N NMR ($\text{DMSO}-d_6$): -179.50 (NCH_2), -144.94, -130.16, -103.94, -48.54, -22.02 (C(NO_2) F), -21.51 (C(NO_2) F), -16.64, 16.28; IR: $\nu = 3116, 3034, 1623, 1575, 1503, 1422, 1359, 1303, 1267, 1223, 1203, 1185, 1083, 1029, 980, 865, 848, 814, 771, 735, 655, 643, 611, 500$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_5\text{H}_4\text{FN}_9\text{O}_4$ (273,14): C 21.99; H 1.48; N 46.15; found C 21.90; H 1.59; N 46.27.

1-(2-Fluoro-2,2-dinitroethyl)-1H-tetrazole, 28



The compound was prepared followed by **General Procedure**. After recrystallization from CH_2Cl_2 colorless solid **28** was obtained in 16% yield; mp $86\text{--}89^\circ\text{C}$; ^1H NMR ($\text{DMSO}-d_6$): 6.45 (, 2H, $J_{\text{HF}} = 15.2$ Hz, NCH_2), 9.61 (s, 1H, CH); ^{13}C NMR ($\text{DMSO}-d_6$): 47.6 (d, $J_{\text{CF}} = 18.42$ Hz, C2), 118.6 (d, $J_{\text{CF}} = 293.64$ Hz, C1), 145.7 (C3); ^{19}F NMR ($\text{DMSO}-d_6$): -107.93 (t, 1F, $J = 14.09$ Hz); ^{14}N NMR ($\text{DMSO}-d_6$): -24.23 (bs, NO_2); ^{15}N NMR ($\text{DMSO}-d_6$): -158.60, -47.98, -22.323 (NO_2), -21.81 (NO_2), -12.01, 17.91.; IR: $\nu = 3151, 2989, 2956, 1604, 1479, 1432, 1324, 1296, 1268, 1177, 1103, 1065, 1020, 994, 958, 886, 850, 810, 763, 705, 686, 650, 593$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_3\text{H}_3\text{FN}_6\text{O}_4$ (273,14): C 17.48; H 1.47; N 40.78; found C 17.55; H 1.42; N 40.69.

1-(2-Fluoro-2,2-dinitroethyl)-5-methyl-1H-tetrazole, 29



The compound was prepared followed by **General Procedure**. After recrystallization from CH_2Cl_2 colorless solid **28** was obtained in 18% yield; mp $122\text{--}123^\circ\text{C}$; ^1H NMR ($\text{DMSO}-d_6$): 2.59 (s, 3H, CH_3), 6.26 (d, 2H, $J_{\text{HF}} = 15.1$ Hz, NCH_2); ^{13}C NMR ($\text{DMSO}-d_6$): 8.2 (CH_3), 46.8 (d, $J_{\text{CF}} = 17.63$ Hz, C2), 118.8 (d, $J_{\text{CF}} = 292.64$ Hz, C1), 154.5 (C3); ^{19}F NMR ($\text{DMSO}-d_6$): -109.87 (t, 1F, $J = 14.53$ Hz); ^{15}N NMR ($\text{DMSO}-d_6$): -272.20, -51.08, -22.33 (NO_2), -21.85 (NO_2), 15.74; IR: $\nu = 3014, 2978, 2940, 1602, 1529, 1422, 1371, 1315, 1293, 1253, 1143, 1060, 1006, 898, 850, 814, 744, 670, 571, 499$ cm^{-1} ; Anal. calcd. (%) for $\text{C}_4\text{H}_5\text{FN}_6\text{O}_4$ (220,12): C 21.83; H 2.29; N 38.18; found C 21.85; H 2.21 N 38.08.

X-ray analysis of the involved materials.

X-ray experiments for compounds **12,13,21,28** were carried out using SMART APEX2 CCD diffractometer ($\lambda(\text{Mo-K}\alpha)=0.71073 \text{ \AA}$, graphite monochromator, ω -scans) at 100K. Collected data were processed by the SAINT and SADABS programs incorporated into the APEX2 program package.^[1S] The structures were solved by the direct methods and refined by the full-matrix least-squares procedure against F^2 in anisotropic approximation. The refinement was carried out with the SHELXTL program.^[2S] The details of data collection and crystal structures refinement are summarized in Table 1S along with CCDC numbers which contain the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif.

Table 1S. Crystallographic data for azasydnone derivatives **12,13,21,28**.

	12	13	21	28
formula	$\text{C}_5\text{H}_4\text{N}_5\text{FO}_6$	$\text{C}_5\text{H}_4\text{N}_5\text{FO}_6$	$\text{C}_5\text{H}_4\text{N}_5\text{FO}_6$	$\text{C}_5\text{H}_4\text{N}_5\text{FO}_6$
fw	249.13	249.13	249.13	206.11
crystal system	Triclinic	Monoclinic	Orthorhombic	Orthorhombic
space group	$P-1$	$P2_1$	$Pbca$	$Pna2_1$
$a, \text{\AA}$	7.976(3)	5.4659(12)	6.6055(3)	11.4331(11)
$b, \text{\AA}$	8.218(3)	10.477(2)	11.5186(5)	5.7212(6)
$c, \text{\AA}$	15.344(6)	8.1322(17)	23.4809(10)	11.2434(11)
$\alpha, \text{deg.}$	94.059(8)	90	90	90
$\beta, \text{deg.}$	104.850(7)	94.857(4)	90	90
$\gamma, \text{deg.}$	97.364(8)	90	90	90
$V, \text{\AA}^3$	958.5(6)	464.02(17)	1786.57(14)	735.44(13)
Z	4	4	8	4
$d_{\text{cryst}}, \text{g}\cdot\text{cm}^{-3}$	1.727	1.783	1.852	1.862
$F(000)$	504	252	1008	416
μ, mm^{-1}	0.169	0.174	0.181	0.182
θ range, deg.	2.5 – 28.0	2.5 – 27.15	3.5 – 30.08	3.6 – 28.9
reflections collected	14158	4928	22344	7221
independent reflections/ R_{int}	4628 / 0.0587	2039 / 0.0603	2621 / 0.0166	1930 / 0.0335
Completeness to theta $\theta, \%$	100	100	99.9	99.9
refined parameters	307	154	154	127
$GOF (F^2)$	1.009	0.987	1.039	1.045
reflections with $I > 2\sigma(I)$	2968	1475	2288	1773
$R_1(F)$ ($I > 2\sigma(I)$) ^a	0.0448	0.0431	0.0317	0.0289
$wR_2(F^2)$ (all data) ^b	0.1034	0.0781	0.0878	0.0648
Max diff. peak/hole, $e\cdot\text{\AA}^{-3}$	0.279 / -0.312	0.218 / -0.223	0.393 / -0.257	0.202 / -0.241
CCDC number	1874345	1874346	1874347	1874348

^a $R_1 = \sum |F_o - |F_c|| / \sum (F_o)$; ^b $wR_2 = (\sum [w(F_o^2 - F_c^2)^2] / \sum [w(F_o^2)^2])^{1/2}$

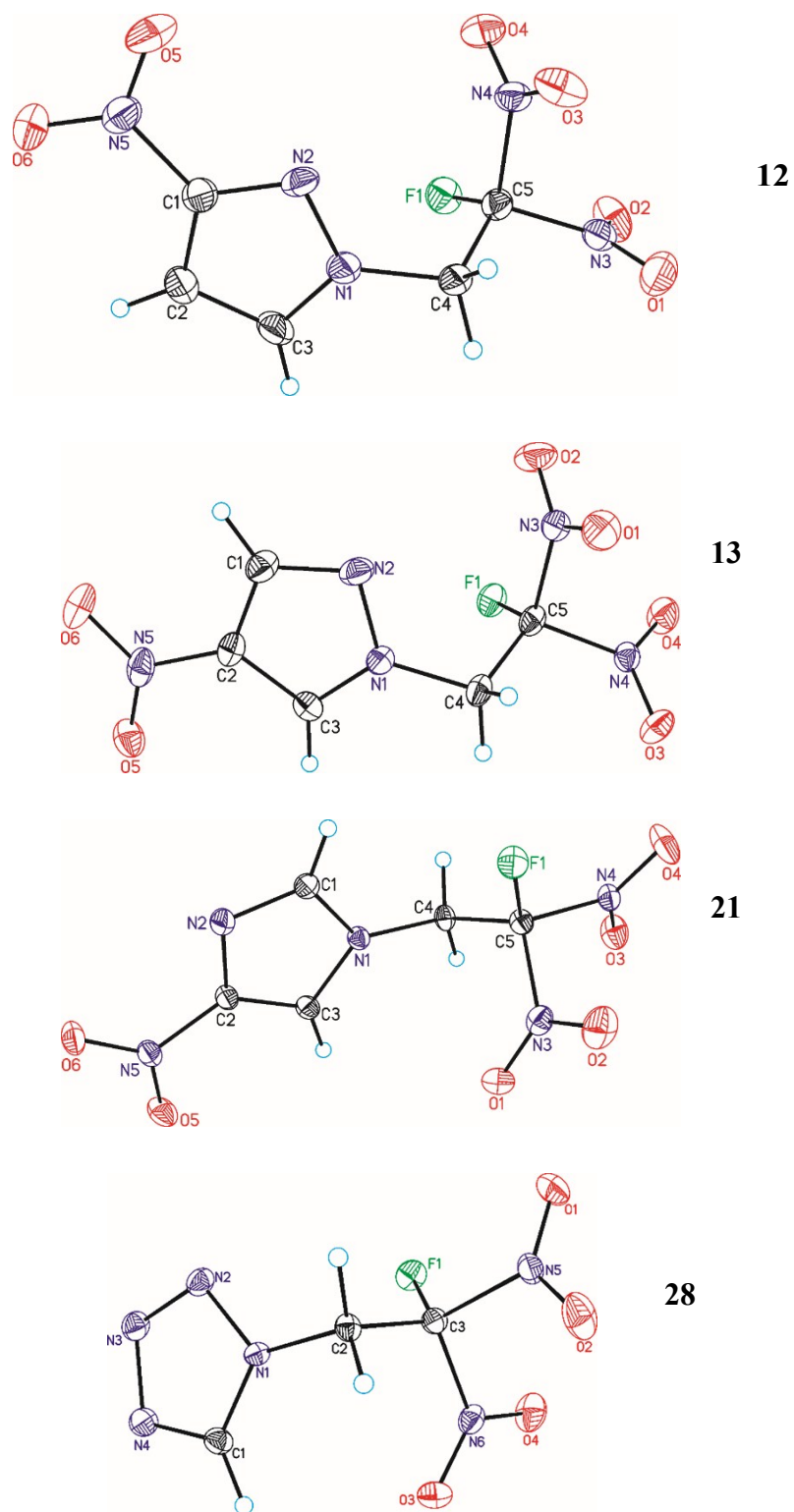


Figure 1S. General view of compounds **12**, **13**, **21**, and **28** with the atom numbering scheme. Displacement ellipsoids are drawn at the 50% probability level

Table 2S. Pair intermolecular interaction energies (kcal/mol) and shortened contacts (Å) of molecule A of compound **12** with its closest environment in the crystal obtained at M052X/6-311G(df,pd) level of approximation.

Entry	Symmetry Code	Atomic pair		Distance	Type of interaction	Molecular pair
1	-1+x,y,z	No close contcts			van der Waals	A...A
2	1+x,y,z	No close contcts			van der Waals	A...A
3	2-x,1-y,-z	F1 O4	O4 F1	2.858 2.858	F...O	A...A
4	2-x,2-y,-z	O5 O5 C4 H4B N2 H4B	C4 H4B O5 O5 H4B N2	3.315 2.531 3.315 2.531 2.511 2.511	C-H...O C-H...N	A...A
5	1+x,y,z	O2 O2	C4' H4'A	3.184 2.398	C-H...O	A...A'
6	1+x,1+y,z	O1	F1'	2.939	F...O	A...A'
7	-x,1-y,-z	No close contcts			van der Waals	A...A'
8	1-x,1-y,-z	O5 O5	C4' H4'A	3.140 2.347	C-H...O	A...A'
9	1-x,1-y,1-z	N3 C4 H4A	O5' O5' O5'	2.914 3.089 2.361	NO ₂ ...NO ₂ C-H...O	A...A'
10	2-x,1-y,1-z	O2 O2 O2 N3	O6' C2' H2' O6'	2.997 3.305 2.524 3.024	NO ₂ ...NO ₂ C-H...O	A...A'

Table 3S. Pair intermolecular interaction energies (kcal/mol) and shortened contacts (Å) of molecule A' of compound **12** with its closest environment in the crystal obtained at M052X/6-311G(df,pd) level of approximation.

Entry	Symmetry Code	Atomic pair		Distance	Type of interaction	Molecular pair
1'	-1+x,y,z	O3'	C3'	3.146	C-H...O	A'...A'
		O3'	H3'	2.483		
2'	1+x,y,z	C3'	O3'	3.146	C-H...O	A'...A'
		H3'	O3'	2.483		
3'	-x,-y,1-z	F1'	O4'	2.910	F...O	A'...A'
		O4'	F1'	2.910		
4'	-x,1-y,1-z	O1'	O5'	2.891	NO ₂ ...NO ₂ C-H...N	A'...A'
		O5'	O1'	2.891		
		N2'	H4'B	2.438		
		H4'B	N2'	2.438		
5'	-1+x,y,z	C4'	O2	3.184	C-H...O	A'...A
		H4'A	O2	2.398		
6'	-1+x,-1+y,z	F1'	O1	2.939	F...O	A'...A
7'	-x,1-y,-z	No	close	contacts	van-der-Waals	A'...A
8'	1-x,1-y,-z	C4'	O5	3.140	C-H...O	A'...A
		H4'A	O5	2.347		
9'	1-x,1-y,1-z	O5'	N3	2.914	NO ₂ ...NO ₂ C-H...O	A'...A
		O5'	C4	3.089		
		O5'	H4A	2.361		
10'	2-x,1-y,1-z	O6'	O2	2.997	NO ₂ ...NO ₂ C-H...O	A'...A
		O6'	N3	3.024		
		C2'	O2	3.305		
		H2'	O2	2.524		

Table 4S. Pair intermolecular interaction energies (kcal/mol) and shortened contacts (Å) of molecule of compound **13** with its closest environment in the crystal obtained at M052X/6-311G(df,pd) level of approximation.

Entry	Symmetry Code	Atomic pair		Distance	Type of interaction
1	-1+x,y,z	O1	N4	2.973	NO ₂ ...NO ₂ C-H...O
		C1	O5	3.170	
		H1	O5	2.464	
2	1+x,y,z	O5	C1	3.170	NO ₂ ...NO ₂ C-H...O
		O5	H1	2.464	
		N4	O1	2.973	
3	x,y,-1+z	No close contacts			van-der-Waals
4	x,y,1+z	No close contacts			van-der-Waals
5	-1-x,-1/2+y,-3-z	O6	N2	3.100	N...NO ₂ C-H...O
		O6	C4	3.268	
		O6	H4A	2.508	
6	-1-x,1/2+y,-3-z	N2	O6	3.100	N...NO ₂ C-H...O
		C4	O6	3.268	
		H4A	O6	2.508	
7	-1-x,-1/2+y,-2-z	No close contacts			van-der-Waals
8	-1-x,1/2+y,-2-z	No close contacts			van-der-Waals
9	-x,-1/2+y,-3-z	O5	C4	3.129	C-H...O
		O5	H4B	2.576	
10	-x,1/2+y,-3-z	C4	O5	3.129	C-H...O
		H4B	O5	2.576	
11	-x,-1/2+y,-2-z	F1	O3	2.806	F...O
12	-x,1/2+y,-2-z	O3	F1	2.806	F...O

Table 5S. Pair intermolecular interaction energies (kcal/mol) and shortened contacts (Å) of molecule of compound **21** with its closest environment in the crystal obtained at M052X/6-311G(df,pd) level of approximation.

Entry	Symmetry Code	Atomic pair		Distance	Type of interaction
1	-1+x,y,z	No close contacts			van-der-Waals
2	1+x,y,z	No close contacts			van-der-Waals
3	-x,-1/2+y,1/2-z	C4 H4B	O5 O5	3.300 2.321	C-H...O
4	-x,1/2+y,1/2-z	O5 O5	C4 H4B	3.300 2.321	C-H...O
5	-1/2+x,1/2-y,-z	F1	O2	2.838	F...O
6	1/2+x,1/2-y,-z	O2	F1	2.838	F...O
7	-x,-y,-z	O4	O4	3.029	NO ₂ ...NO ₂
8	1-x,-y,-z	O2	O4	3.072	NO ₂ ...NO ₂
9	-1/2+x,y,1/2-z	O6 N5 C2	O5 O5 O5	3.000 3.022 3.251	O...π
10	1/2+x,y,1/2-z	O5 O5 O5	O6 N5 C2	3.000 3.022 3.251	O...π
11	-1/2-x,-1/2+y,z	C4 H4A	N2 N2	3.168 2.461	C-H...N
12	-1/2-x,1/2+y,z	N2 N2	C4 H4A	3.168 2.461	C-H...N
13	1/2-x,-1/2+y,z	O1	O6	2.876	NO ₂ ...NO ₂
14	1/2-x,1/2+y,z	O6	O1	2.876	NO ₂ ...NO ₂

Comments

It is seen that predominant contribution into intermolecular bonding for all three compounds is provided by C-H...O(N) and van-der-Waals interactions. In addition, short F...O and NO₂...NO₂ contacts are also observed. Each molecule in the crystal of all three compounds forms nearly the same numbers of intermolecular interactions. However in compound **13** and especially in compound **12** some of the closest neighbours are connected simultaneously by two different types of interactions. As a result, number of closest neighbours of a molecule in the crystals of **12** and **13** are equal to 10 and 12 respectively. In contrary, in compound **21**, intermolecular contacts are distributed over 14 neighbours thereby inducing less constrain for optimal packing.

Table 6S. Tightness of crystal packing as defined by Kitaigorodsky packing coefficient and Δ_{OED} criterion for compounds **12**, **13**, **21**

Crystal packing tightness	compound 12	compound 13	compound 21
Kitaigorodsky packing coefficient	0.703	0.728	0.756
Δ_{OED} criterion	0.151	0.202	0.271

References

- 1S. *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA, **2009**.
- 2S. G. M. Sheldrick, *Acta Crystallogr.* **2008**, *A64*, 112-122.

NMR spectra

