

A series of Ln^{III}_4 clusters: Dy_4 single molecule magnet and Tb_4 multi-responsive luminescent sensor for Fe^{3+} , $\text{CrO}_4^{2-}/\text{Cr}_2\text{O}_7^{2-}$ 4-nitroaniline

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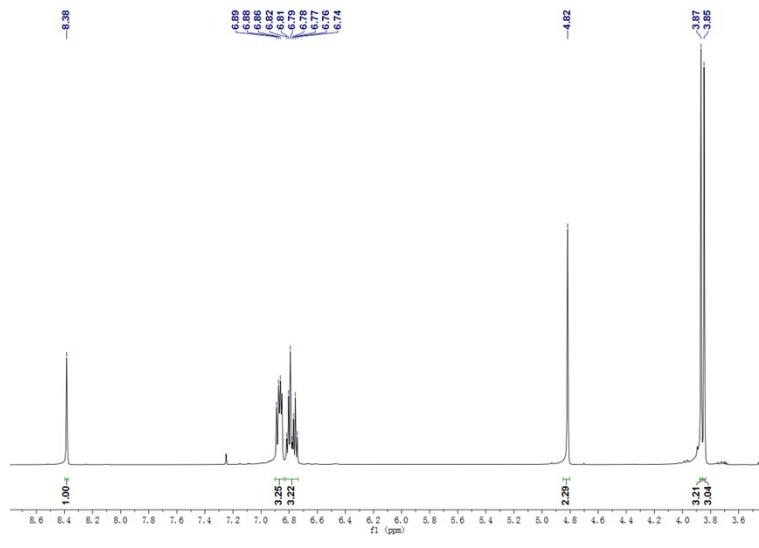


Fig. S1 ^1H NMR spectrum of H_2L .

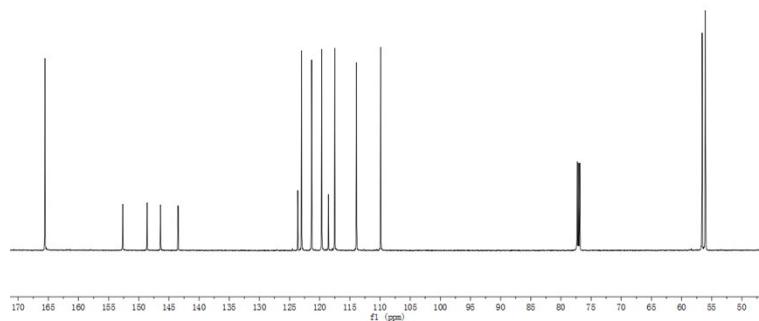


Fig. S2 ^{13}C NMR spectrum of H_2L .

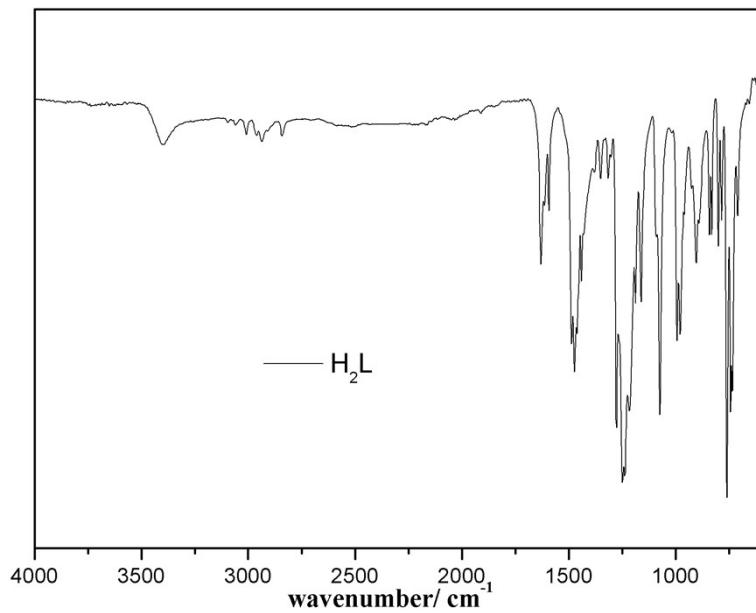


Fig. S3 FT-IR spectrum of H_2L .

Table S1 The selected bond length (\AA) and angle ($^\circ$) for 1-5

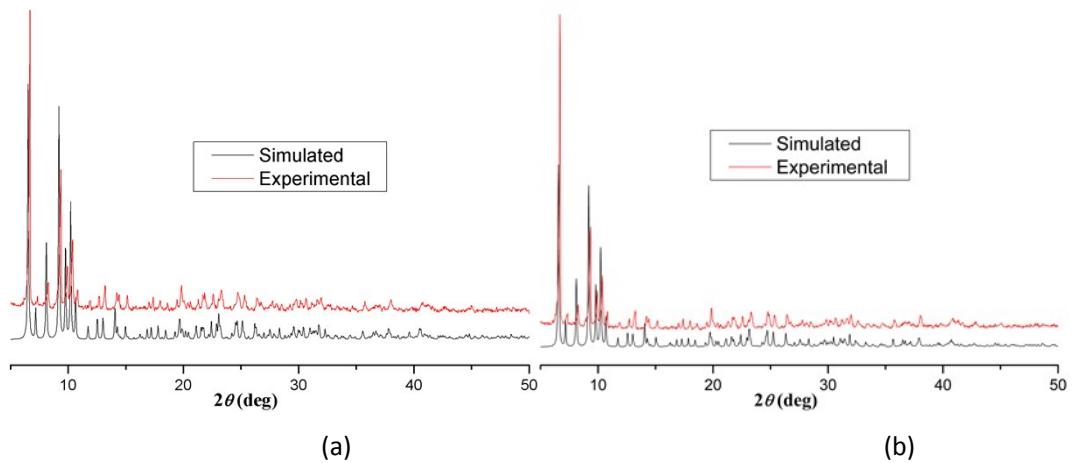
1	Length/\AA	2	Length/\AA	3	Length/\AA
Gd(1)-O(1)	2.310(3)	Tb(1)-O(1)	2.293(2)	Dy(1)-O(1)	2.293(2)
Gd(1)-O(3)	2.202(3)	Tb(1)-O(3)	2.181(2)	Dy(1)-O(3)	2.169(3)
Gd(1)-O(5)	2.451(3)	Tb(1)-O(5)	2.427(2)	Dy(1)-O(5)	2.405(2)
Gd(1)-O(6)	2.444(3)	Tb(1)-O(6)	2.435(2)	Dy(1)-O(6)	2.429(3)
Gd(1)-O(10)	2.368(3)	Tb(1)-O(10)	2.358(2)	Dy(1)-O(10)	2.355(3)
Gd(1)-O(11)	2.548(4)	Tb(1)-O(11)	2.533(3)	Dy(1)-O(11)	2.534(3)
Gd(1)-O(12)	2.478(4)	Tb(1)-O(12)	2.474(3)	Dy(1)-O(12)	2.455(3)
Gd(1)-N(1)	2.509(4)	Tb(1)-N(1)	2.495(3)	Dy(1)-N(1)	2.491(3)
Gd(2)-O(1)	2.332(3)	Tb(2)-O(1)	2.314(2)	Dy(2)-O(1)	2.298(2)
Gd(2)-O(2)	2.586(3)	Tb(2)-O(2)	2.578(2)	Dy(2)-O(2)	2.575(3)
Gd(2)-O(5)	2.391(3)	Tb(2)-O(5)	2.379(2)	Dy(2)-O(5)	2.378(2)
Gd(2)-O(7)	2.305(3)	Tb(2)-O(7)	2.289(2)	Dy(2)-O(7)#1	2.303(2)
Gd(2)-O(7)#1	2.332(3)	Tb(2)-O(7)#1	2.312(2)	Dy(2)-O(7)	2.281(2)
Gd(2)-O(8)#1	2.634(4)	Tb(2)-O(8)#1	2.640(3)	Dy(2)-O(8)#1	2.635(3)
Gd(2)-O(9)	2.280(3)	Tb(2)-O(9)	2.258(2)	Dy(2)-O(9)	2.249(3)
Gd(2)-N(2)	2.479(5)	Tb(2)-N(2)	2.450(3)	Dy(2)-N(2)	2.443(3)
4	Length/\AA	4	Length/\AA	4	Length/\AA
Ho(1)-O(1)	2.276(3)	Ho(1)-O(6)	2.408(3)	Ho(1)-O(12)	2.430(3)
Ho(1)-O(3)	2.167(3)	Ho(1)-O(10)	2.334(3)	Ho(1)-N(1)	2.472(4)
Ho(1)-O(5)	2.393(3)	Ho(1)-O(11)	2.519(3)	Ho(2)-O(1)	2.285(3)
Ho(2)-O(2)	2.568(3)	Ho(2)-O(5)	2.359(3)	Ho(2)-O(7)	2.268(3)
Ho(2)-O(7)#1	2.289(3)	Ho(2)-O(8)#1	2.618(3)	Ho(2)-O(9)	2.234(3)
Ho(2)-N(2)	2.429(4)				
5	Length/\AA	5	Length/\AA	5	Length/\AA
Er(1)-O(1)	2.264(3)	Er(1)-O(10)	2.328(3)	Er(2)-O(1)	2.276(3)
Er(2)-O(7)#1	2.277(3)	Er(1)-O(3)	2.160(3)	Er(1)-O(11)	2.519(3)

Er(2)-O(2)	2.572(3)	Er(2)-O(8)#1	2.614(3)	Er(1)-O(5)	2.389(3)
Er(1)-O(12)	2.417(3)	Er(2)-O(5)	2.344(3)	Er(2)-O(9)	2.224(3)
Er(1)-O(6)	2.400(3)	Er(1)-N(1)	2.457(3)	Er(2)-O(7)	2.262(3)
Er(2)-N(2)	2.424(4)				
1	Angle/°	2	Angle/°	3	Angle/°
O(1)-Gd(1)-O(5)	73.48(11)	O(1)-Tb(1)-O(5)	73.05(8)	O(1)-Dy(1)-O(5)	72.91(8)
O(1)-Gd(1)-O(6)	138.17(12)	O(1)-Tb(1)-O(6)	138.35(8)	O(1)-Dy(1)-O(6)	138.57(9)
O(1)-Gd(1)-O(10)	84.17(12)	O(1)-Tb(1)-O(10)	84.30(9)	O(1)-Dy(1)-O(10)	84.34(9)
O(1)-Gd(1)-O(11)	134.50(12)	O(1)-Tb(1)-O(11)	134.22(9)	O(1)-Dy(1)-O(11)	134.32(9)
O(1)-Gd(1)-O(12)	87.23(12)	O(1)-Tb(1)-O(12)	86.97(9)	O(1)-Dy(1)-O(12)	87.05(9)
O(1)-Gd(1)-N(1)	75.32(13)	O(1)-Tb(1)-N(1)	75.49(9)	O(1)-Dy(1)-N(1)	75.46(10)
O(3)-Gd(1)-O(1)	127.73(12)	O(3)-Tb(1)-O(1)	128.40(9)	O(3)-Dy(1)-O(1)	127.96(10)
O(3)-Gd(1)-O(5)	146.01(12)	O(3)-Tb(1)-O(5)	146.15(9)	O(3)-Dy(1)-O(5)	146.22(9)
O(3)-Gd(1)-O(6)	85.60(12)	O(3)-Tb(1)-O(6)	84.91(9)	O(3)-Dy(1)-O(6)	85.07(9)
O(3)-Gd(1)-O(10)	79.87(13)	O(3)-Tb(1)-O(10)	80.12(9)	O(3)-Dy(1)-O(10)	79.57(9)
O(3)-Gd(1)-O(11)	76.43(13)	O(3)-Tb(1)-O(11)	75.97(9)	O(3)-Dy(1)-O(11)	76.19(10)
O(3)-Gd(1)-O(12)	122.70(13)	O(3)-Tb(1)-O(12)	122.75(9)	O(3)-Dy(1)-O(12)	123.41(10)
O(3)-Gd(1)-N(1)	72.92(13)	O(3)-Tb(1)-N(1)	73.50(9)	O(3)-Dy(1)-N(1)	73.68(10)
O(5)-Gd(1)-O(11)	108.61(12)	O(5)-Tb(1)-O(6)	66.47(8)	O(5)-Dy(1)-O(6)	66.78(8)
O(5)-Gd(1)-O(12)	79.59(12)	O(5)-Tb(1)-O(11)	108.89(9)	O(5)-Dy(1)-O(11)	109.29(9)
O(5)-Gd(1)-N(1)	140.90(12)	O(5)-Tb(1)-O(12)	79.06(9)	O(5)-Dy(1)-O(12)	78.73(9)
O(6)-Gd(1)-O(5)	65.86(11)	O(6)-Tb(1)-N(1)	140.25(9)	O(6)-Dy(1)-N(1)	139.97(9)
O(6)-Gd(1)-O(11)	70.75(12)	O(6)-Tb(1)-O(11)	70.90(9)	O(6)-Dy(1)-O(11)	70.75(9)
O(6)-Gd(1)-O(12)	94.58(13)	O(6)-Tb(1)-O(12)	94.52(9)	O(6)-Dy(1)-O(12)	94.05(10)
O(6)-Gd(1)-N(1)	145.51(13)	O(6)-Tb(1)-N(1)	145.02(9)	O(6)-Dy(1)-N(1)	144.74(10)
O(10)-Gd(1)-O(5)	76.27(12)	O(10)-Tb(1)-O(5)	76.35(8)	O(10)-Dy(1)-O(5)	76.52(9)
O(10)-Gd(1)-O(6)	77.31(12)	O(10)-Tb(1)-O(6)	77.25(9)	O(10)-Dy(1)-O(6)	77.57(9)
O(10)-Gd(1)-O(11)	141.26(12)	O(10)-Tb(1)-O(11)	141.42(9)	O(10)-Dy(1)-O(11)	141.31(9)
O(10)-Gd(1)-O(12)	155.79(13)	O(10)-Tb(1)-O(12)	155.35(9)	O(10)-Dy(1)-O(12)	155.21(9)
O(10)-Gd(1)-N(1)	123.11(13)	O(10)-Tb(1)-N(1)	123.88(9)	O(10)-Dy(1)-N(1)	123.94(10)
O(12)-Gd(1)-O(11)	50.58(12)	O(12)-Tb(1)-O(11)	50.74(9)	O(12)-Dy(1)-O(11)	51.01(9)
O(12)-Gd(1)-N(1)	75.94(13)	O(12)-Tb(1)-N(1)	75.59(9)	O(12)-Dy(1)-N(1)	75.73(10)
N(1)-Gd(1)-O(11)	78.03(13)	N(1)-Tb(1)-O(11)	77.25(9)	N(1)-Dy(1)-O(11)	76.95(10)
O(1)-Gd(2)-O(2)	64.51(11)	O(1)-Tb(2)-O(2)	64.85(8)	O(1)-Dy(2)-O(2)	65.13(9)
O(1)-Gd(2)-O(5)	74.22(11)	O(1)-Tb(2)-O(5)	73.58(8)	O(1)-Dy(2)-O(5)	73.33(8)
O(1)-Gd(2)-O(7)#1	89.14(12)	O(1)-Tb(2)-O(8)#1	76.58(8)	O(1)-Dy(2)-O(7)#1	88.98(9)
O(1)-Gd(2)-O(8)#1	76.69(11)	O(1)-Tb(2)-N(2)	150.82(9)	O(1)-Dy(2)-O(8)#1	76.44(9)
O(1)-Gd(2)-N(2)	151.17(13)	O(2)-Tb(2)-O(8)#1	120.96(8)	O(1)-Dy(2)-N(2)	150.93(10)
O(2)-Gd(2)-O(8)#1	120.64(11)	O(5)-Tb(2)-O(2)	126.99(8)	O(2)-Dy(2)-O(8)#1	121.43(9)
O(5)-Gd(2)-O(2)	127.31(11)	O(5)-Tb(2)-O(8)#1	76.50(8)	O(5)-Dy(2)-O(2)	126.88(9)
O(5)-Gd(2)-O(8)#1	76.77(11)	O(5)-Tb(2)-N(2)	77.27(9)	O(5)-Dy(2)-O(8)#1	76.11(8)
O(5)-Gd(2)-N(2)	77.00(13)	O(7)#1-Tb(2)-O(1)	89.15(8)	O(5)-Dy(2)-N(2)	77.62(10)
O(7)-Gd(2)-O(1)	138.07(12)	O(7)-Tb(2)-O(1)	138.08(8)	O(7)-Dy(2)-O(1)	138.07(9)
O(7)-Gd(2)-O(2)	75.87(12)	O(7)#1-Tb(2)-O(2)	74.22(9)	O(7)#1-Dy(2)-O(2)	74.38(9)

O(7)#1-Gd(2)-O(2)	73.95(12)	O(7)-Tb(2)-O(2)	75.73(9)	O(7)-Dy(2)-O(2)	75.52(9)
O(7)-Gd(2)-O(5)	146.00(12)	O(7)-Tb(2)-O(5)	146.76(8)	O(7)-Dy(2)-O(5)	147.05(9)
O(7)#1-Gd(2)-O(5)	138.09(11)	O(7)#1-Tb(2)-O(5)	137.70(8)	O(7)#1-Dy(2)-O(5)	137.43(9)
O(7)-Gd(2)-O(7)#1	66.35(14)	O(7)-Tb(2)-O(7)#1	66.09(10)	O(7)-Dy(2)-O(7)#1	66.22(10)
O(7)#1-Gd(2)-O(8)#1	61.90(11)	O(7)#1-Tb(2)-O(8)#1	61.85(8)	O(7)#1-Dy(2)-O(8)#1	62.03(8)
O(7)-Gd(2)-O(8)#1	115.29(11)	O(7)-Tb(2)-O(8)#1	115.03(8)	O(7)-Dy(2)-O(8)#1	115.26(9)
O(7)-Gd(2)-N(2)	69.97(14)	O(7)#1-Tb(2)-N(2)	113.93(9)	O(7)-Dy(2)-N(2)	70.37(10)
O(7)#1-Gd(2)-N(2)	113.92(13)	O(7)-Tb(2)-N(2)	70.41(9)	O(7)#1-Dy(2)-N(2)	113.84(10)
O(9)-Gd(2)-O(1)	92.28(12)	O(9)-Tb(2)-O(1)	92.48(9)	O(9)-Dy(2)-O(1)	92.89(9)
O(9)-Gd(2)-O(2)	71.67(12)	O(9)-Tb(2)-O(2)	71.79(9)	O(9)-Dy(2)-O(2)	71.77(9)
O(9)-Gd(2)-O(5)	79.03(12)	O(9)-Tb(2)-O(5)	78.92(9)	O(9)-Dy(2)-O(5)	79.06(9)
O(9)-Gd(2)-O(7)	87.72(12)	O(9)-Tb(2)-O(7)	88.16(9)	O(9)-Dy(2)-O(7)	87.96(9)
O(9)-Gd(2)-O(7)#1	141.02(12)	O(9)-Tb(2)-O(7)#1	141.45(9)	O(9)-Dy(2)-O(7)#1	141.57(9)
O(9)-Gd(2)-O(8)#1	155.34(12)	O(9)-Tb(2)-O(8)#1	155.05(8)	O(9)-Dy(2)-O(8)#1	154.87(9)
O(9)-Gd(2)-N(2)	80.60(14)	O(9)-Tb(2)-N(2)	80.50(10)	O(9)-Dy(2)-N(2)	80.36(11)
N(2)-Gd(2)-O(2)	136.43(13)	N(2)-Tb(2)-O(2)	136.50(9)	N(2)-Dy(2)-O(2)	136.27(10)
N(2)-Gd(2)-O(8)#1	98.39(13)	N(2)-Tb(2)-O(8)#1	98.07(10)	N(2)-Dy(2)-O(8)#1	97.86(10)
4	Angle/°	4	Angle/°	4	Angle/°
O(1)-Ho(1)-O(5)	72.99(9)	O(1)-Ho(1)-O(6)	138.9(10)	O(1)-Ho(1)-O(10)	84.35(10)
O(1)-Ho(1)-O(11)	134.23(10)	O(1)-Ho(1)-O(5)	72.99(9)	O(1)-Ho(1)-O(6)	138.90(10)
O(1)-Ho(1)-O(12)	86.77(11)	O(1)-Ho(1)-N(1)	75.53(11)	O(3)-Ho(1)-O(1)	127.85(11)
O(3)-Ho(1)-O(5)	145.96(11)	O(3)-Ho(1)-O(6)	84.75(11)	O(3)-Ho(1)-O(10)	79.20(11)
O(3)-Ho(1)-O(11)	76.43(11)	O(3)-Ho(1)-O(12)	123.93(11)	O(3)-Ho(1)-N(1)	73.87(11)
O(5)-Ho(1)-O(6)	67.02(9)	O(5)-Ho(1)-O(11)	109.35(10)	O(5)-Ho(1)-O(12)	78.63(10)
O(5)-Ho(1)-N(1)	140.03(11)	O(6)-Ho(1)-O(11)	70.66(10)	O(6)-Ho(1)-O(12)	94.25(11)
O(6)-Ho(1)-N(1)	144.34(11)	O(10)-Ho(1)-O(5)	76.59(10)	O(10)-Ho(1)-O(6)	77.73(11)
O(10)-Ho(1)-O(11)	141.39(11)	O(10)-Ho(1)-O(12)	155.15(11)	O(10)-Ho(1)-N(1)	123.99(11)
O(12)-Ho(1)-O(11)	51.27(11)	O(12)-Ho(1)-N(1)	75.60(12)	N(1)-Ho(1)-O(11)	76.69(11)
O(1)-Ho(2)-O(2)	65.38(10)	O(1)-Ho(2)-O(5)	73.48(10)	O(1)-Ho(2)-O(7)#1	88.39(10)
O(1)-Ho(2)-O(8)#1	76.51(10)	O(1)-Ho(2)-N(2)	150.76(11)	O(2)-Ho(2)-O(8)#1	122.11(10)
O(5)-Ho(2)-O(2)	127.19(10)	O(5)-Ho(2)-O(8)#1	75.50(10)	O(5)-Ho(2)-N(2)	77.31(11)
O(7)-Ho(2)-O(1)	137.84(10)	O(7)-Ho(2)-O(2)	75.12(10)	O(7)#1-Ho(2)-O(2)	74.12(10)
O(7)-Ho(2)-O(5)	147.18(10)	O(7)#1-Ho(2)-O(5)	137.19(10)	O(7)-Ho(2)-O(7)#1	66.46(12)
O(7)#1-Ho(2)-O(8)#1	62.50(10)	O(7)-Ho(2)-O(8)#1	115.54(10)	O(7)-Ho(2)-N(2)	70.75(12)
O(7)#1-Ho(2)-N(2)	114.66(12)	O(9)-Ho(2)-O(1)	92.94(10)	O(9)-Ho(2)-O(2)	71.54(10)
O(9)-Ho(2)-O(7)	87.94(10)	O(9)-Ho(2)-O(7)#1	141.40(10)	O(9)-Ho(2)-O(8)#1	154.52(10)
O(9)-Ho(2)-N(2)	80.19(12)	N(2)-Ho(2)-O(2)	135.99(12)	N(2)-Ho(2)-O(8)#1	97.64(12)
5	Angle/°	5	Angle/°	5	Angle/°
O(1)-Er(1)-O(5)	72.76(9)	O(1)-Er(1)-O(6)	138.85(9)	O(1)-Er(1)-O(10)	84.24(10)

O(1)-Er(1)-O(11)	134.56(10)	O(1)-Er(1)-O(12)	86.88(10)	O(1)-Er(1)-N(1)	75.87(10)
O(3)-Er(1)-O(1)	128.17(10)	O(3)-Er(1)-O(5)	145.47(10)	O(3)-Er(1)-O(6)	84.16(10)
O(3)-Er(1)-O(10)	78.79(10)	O(3)-Er(1)-O(11)	76.26(10)	O(3)-Er(1)-O(12)	124.30(10)
O(3)-Er(1)-N(1)	74.47(11)	O(5)-Er(1)-O(6)	67.24(9)	O(5)-Er(1)-O(11)	109.70(10)
O(5)-Er(1)-O(12)	78.55(10)	O(5)-Er(1)-N(1)	139.93(10)	O(6)-Er(1)-O(11)	70.65(10)
O(6)-Er(1)-O(12)	94.25(11)	O(6)-Er(1)-N(1)	144.04(10)	O(10)-Er(1)-O(5)	76.59(10)
O(10)-Er(1)-O(6)	77.67(10)	O(10)-Er(1)-O(11)	141.18(10)	O(10)-Er(1)-O(12)	155.08(10)
O(10)-Er(1)-N(1)	124.31(11)	O(12)-Er(1)-O(11)	51.57(10)	O(12)-Er(1)-N(1)	75.43(11)
N(1)-Er(1)-O(11)	76.30(11)	O(1)-Er(2)-O(2)	65.43(9)	O(1)-Er(2)-O(5)	73.41(9)
O(1)-Er(2)-O(7)#1	88.60(10)	O(1)-Er(2)-O(8)#1	76.82(9)	O(1)-Er(2)-N(2)	151.00(11)
O(2)-Er(2)-O(8)#1	122.45(9)	O(5)-Er(2)-O(2)	127.27(9)	O(5)-Er(2)-O(8)#1	75.27(9)
O(5)-Er(2)-N(2)	77.63(11)	O(7)-Er(2)-O(1)	137.73(10)	O(7)-Er(2)-O(2)	74.96(10)
O(7)#1-Er(2)-O(2)	74.19(10)	O(7)-Er(2)-O(5)	147.38(10)	O(7)#1-Er(2)-O(5)	137.14(9)
O(7)-Er(2)-O(7)#1	66.24(12)	O(7)#1-Er(2)-O(8)#1	62.70(9)	O(7)-Er(2)-O(8)#1	115.46(9)
O(7)-Er(2)-N(2)	70.62(11)	O(7)#1-Er(2)-N(2)	114.29(11)	O(9)-Er(2)-O(1)	92.77(10)
O(9)-Er(2)-O(2)	71.29(10)	O(9)-Er(2)-O(5)	79.56(10)	O(9)-Er(2)-O(7)	87.92(10)
O(9)-Er(2)-O(7)#1	141.19(10)	O(9)-Er(2)-O(8)#1	154.58(10)	O(9)-Er(2)-N(2)	80.46(11)
N(2)-Er(2)-O(2)	135.78(11)	N(2)-Er(2)-O(8)#1	97.37(11)		

Symmetry transformation: #1 -X, 2-Y, 1-Z for 1, #1 2-X, -Y, 1-Z for 2-5



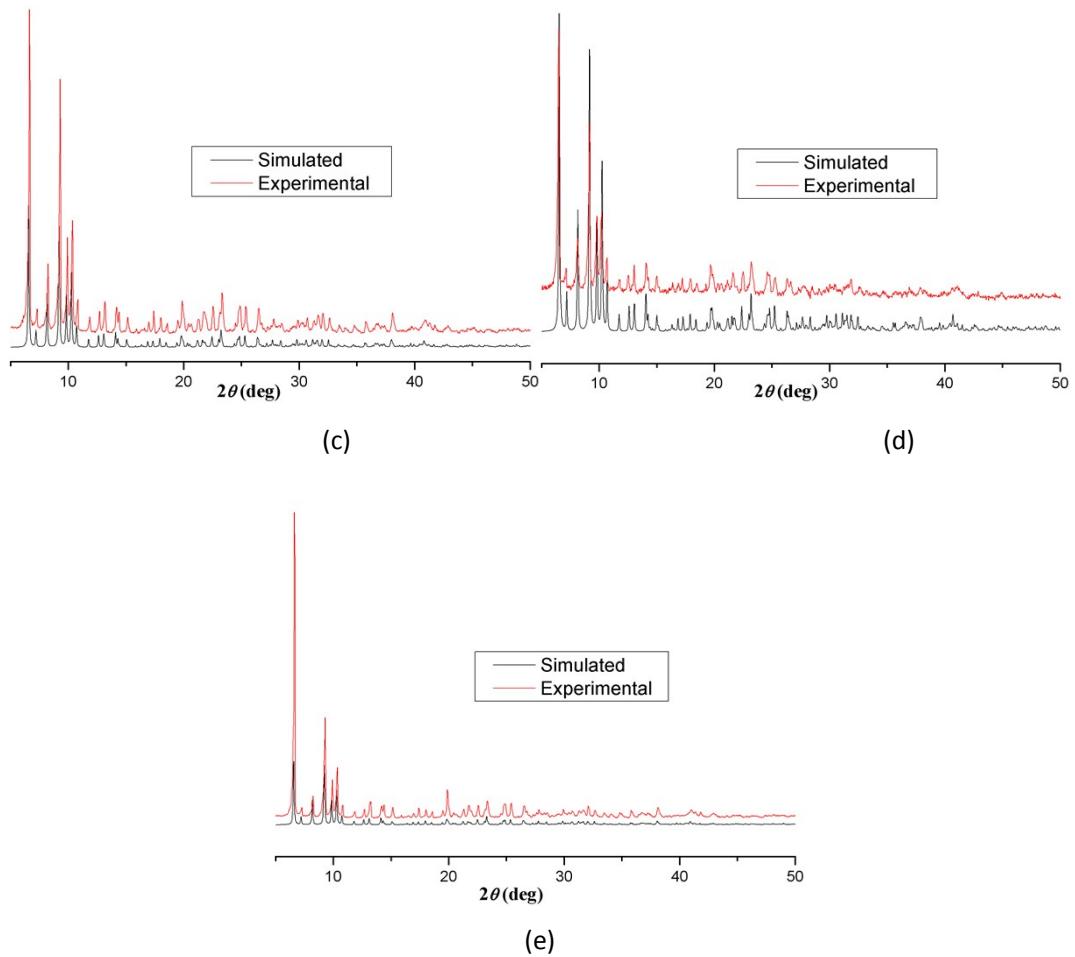
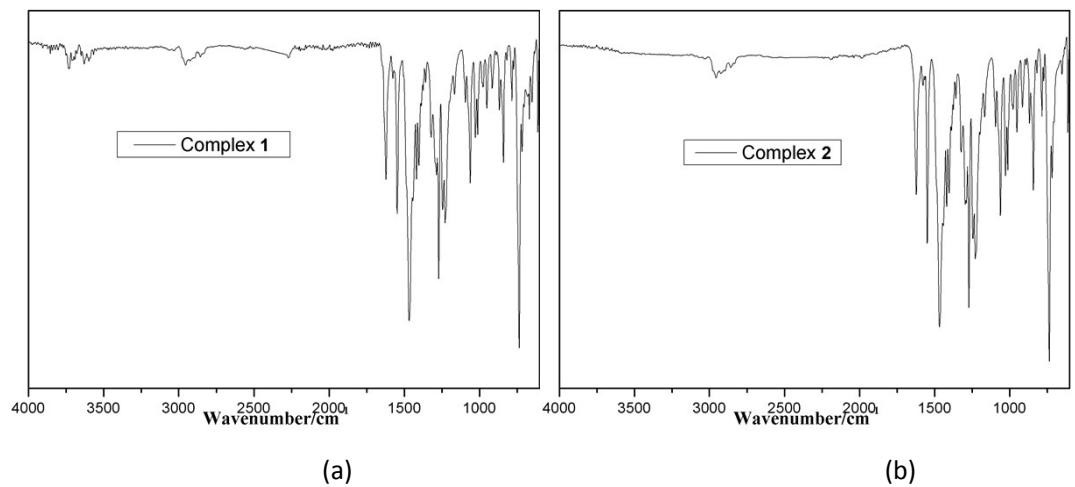


Fig. S4 PXRD patterns of **1(a)**, **2(b)**, **3(c)**, **4(d)**, **5(e)**.



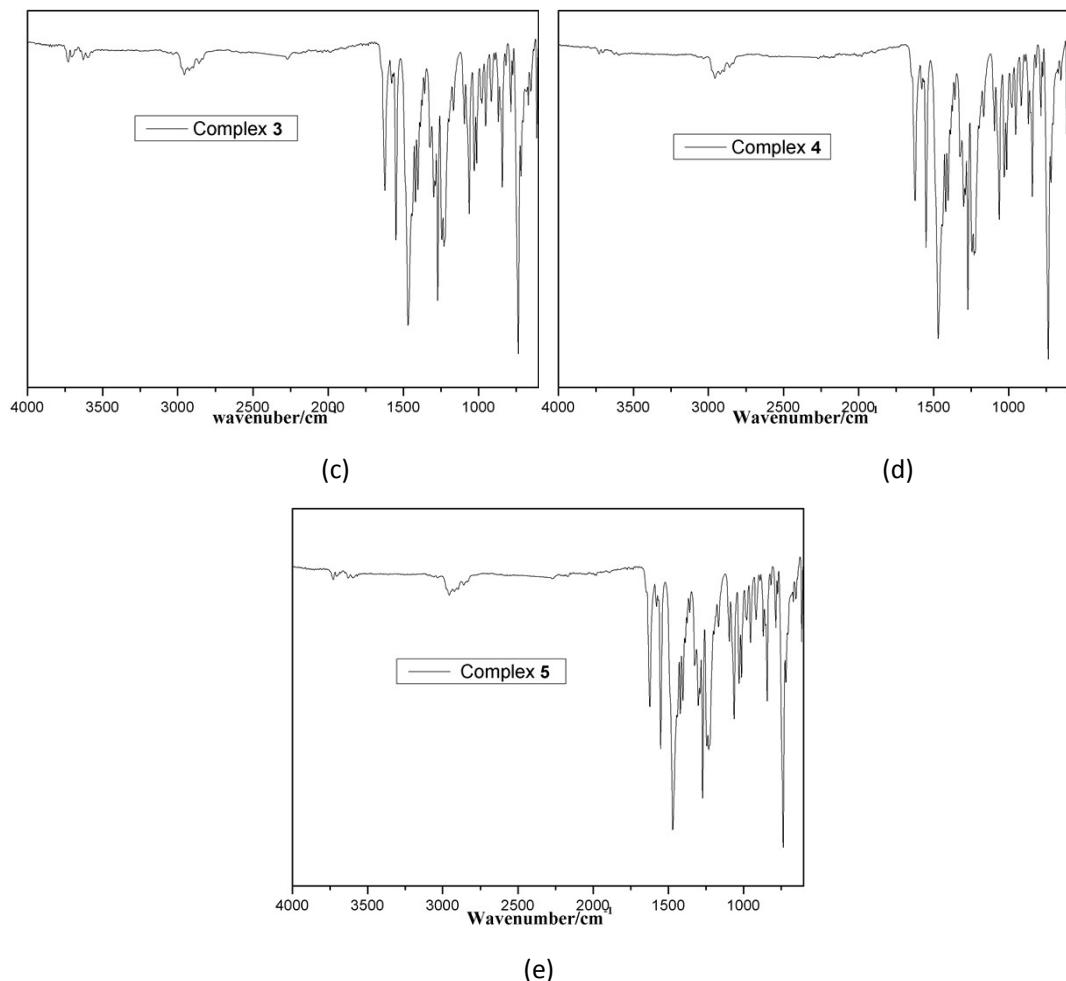


Fig. S5 IR spectra of **1(a)**, **2(b)**, **3(c)**, **4(d)**, **5(e)**.

Table S2. Agreement factor between the coordination polyhedron of the Ln^{III} and the various ideal polyhedral calculated by the SHAPE program

	HBPY-8	CU-8	SAPR-8	TDD-8	JGBF-8	JBTPR-8	BTPR-8	JSD-8	TT-8
Gd1	14.779,	8.898,	1.686,	2.836,	14.348,	2.716,	1.948	5.147	9.501,
Gd2,	14.482,	11.904,	3.123,	2.877,	11.858,	2.745,	2.428	4.685	12.131
Tb1	14.941,	8.967,	1.635,	2.787,	14.521,	2.674,	1.858	5.070	9.572
Tb2	14.587,	11.972,	3.095,	2.913,	11.953,	2.721,	2.435	4.625	12.156
Dy1,	15.029,	9.008,	1.635,	2.709,	14.552,	2.604,	1.787	4.952	9.590
Dy2,	14.735,	12.056,	3.038,	2.904,	12.026,	2.668,	2.416	4.549	12.224
Ho1	15.177,	9.091,	1.567,	2.688,	14.556,	2.534,	1.741,	4.917	9.670
Ho2,	14.872,	12.189,	2.967,	2.924,	11.996,	2.572,	2.313	4.515	12.306
Er1	15.298,	9.127,	1.525,	2.645,	14.588,	2.447,	1.685	4.823	9.686
Er2	14.918	12.242	2.990	2.925	12.031	2.566	2.323	4.478	12.406
1 HBPY-8	Hexagonal bipyramid				D6h				
2 CU-8	Cube				Oh				
3 SAPR-8	Square antiprism				D4d				
4 TDD-8	Triangular dodecahedron				D2d				
5 JGBF-8	Johnson – Gyrobifastigium (J26)				D2d				

6 JBTP-8	Johnson – Biaugmented trigonal prism (J50)	C2v
7 BTPR-8	Biaugmented trigonal prism	C2v
8 JSD-8	Snub disphenoid (J84)	D2d
9 TT-8	Triakis tetrahedron	Td

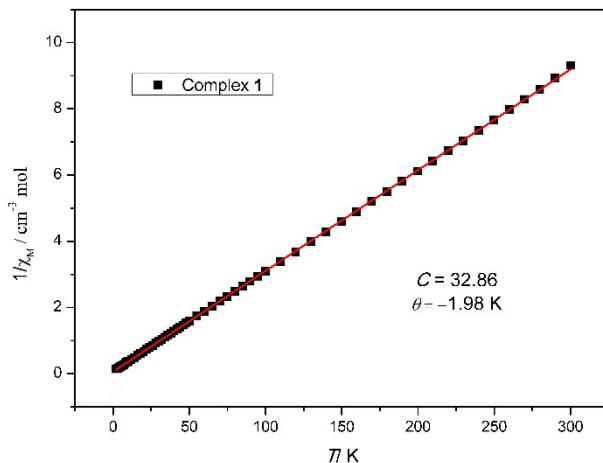


Fig. S6 Plot of $1/\chi_M$ versus T for **1**, the linear fit is the Curie-Weiss law fit at 1 kOe field.

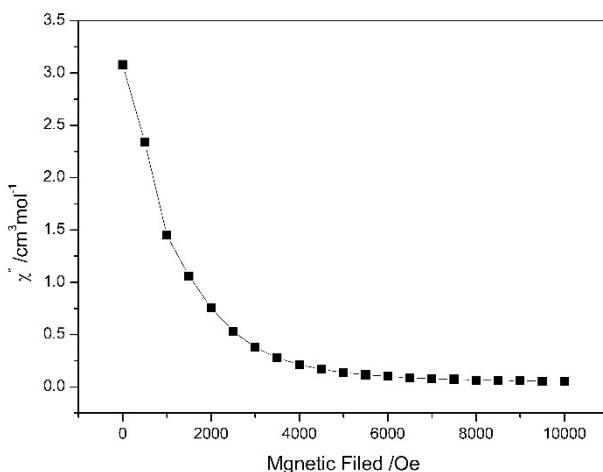


Fig. S7 Temperature dependence of the out-of-phase (χ'') ac susceptibility for **3** under zero dc field at 1000 Hz.

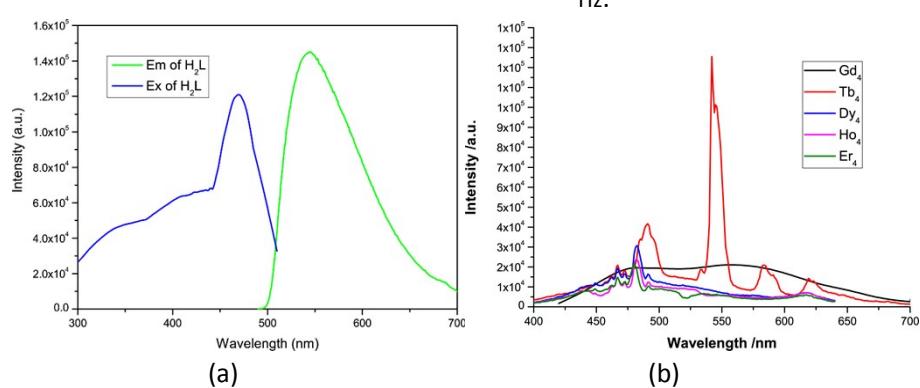


Fig. S8 (a) The excitation and emission spectra of H_2L ligand. (b) Emission spectra of **1-5** in solid state at room temperature.

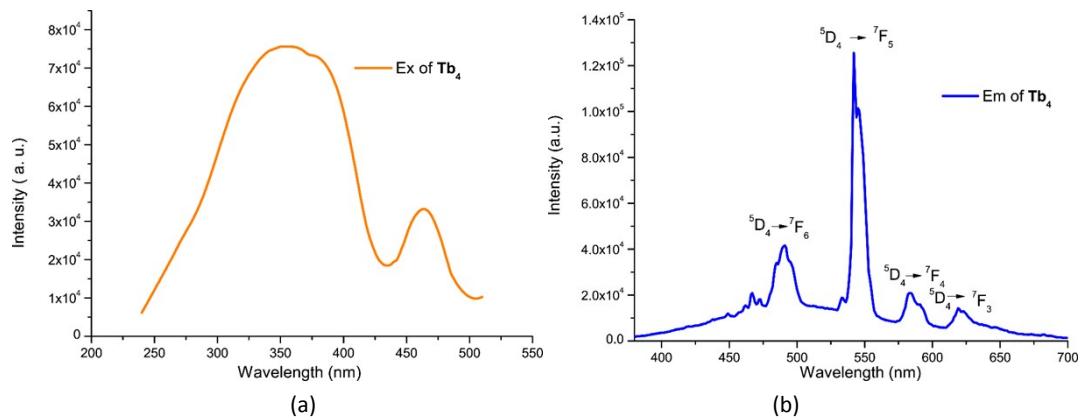


Fig. S9 Excitation spectrum (a) and emission spectrum (b) of **2** in solid state at room temperature.

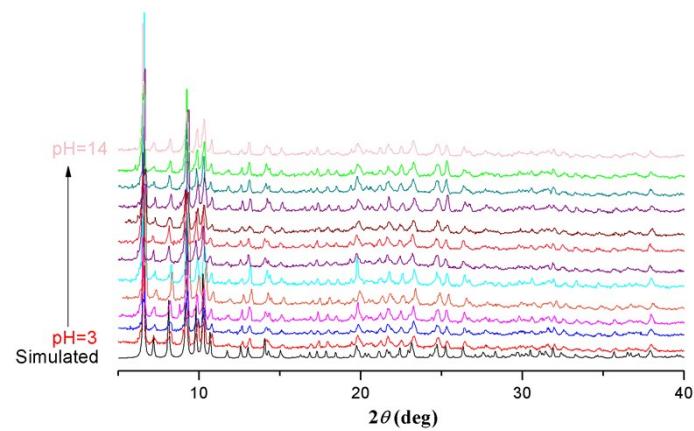


Fig. S10 PXRD patterns for the simulated and experimental samples of **2** soaked in aqueous solutions with pH values in the range of 3-14 for two days.

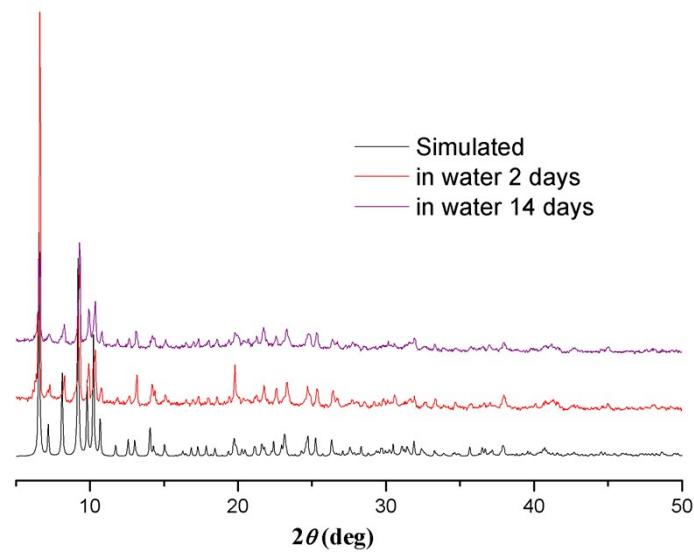


Fig. S11 PXRD patterns for the simulated and experimental samples of **2** soaked in water for 2 days and 14 days.

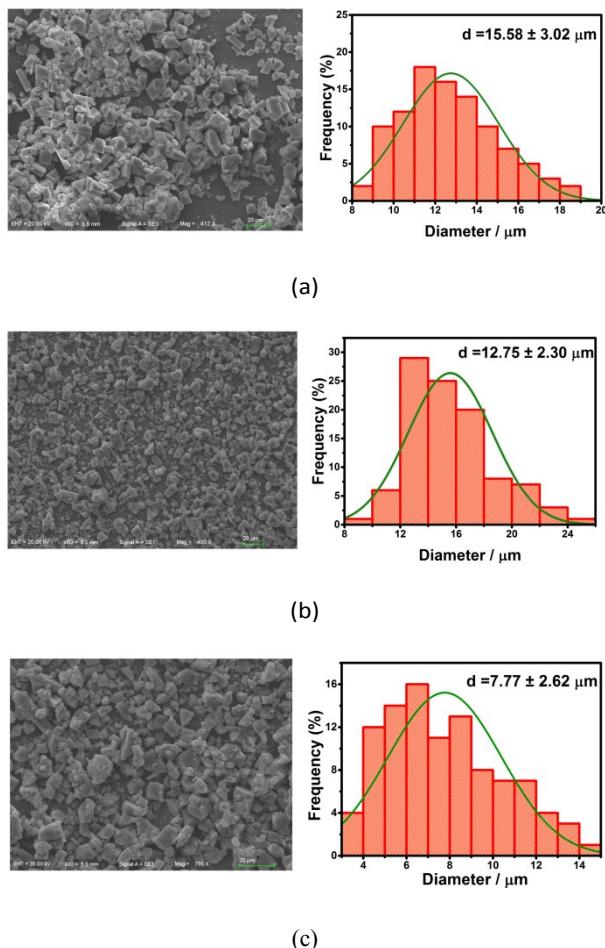


Fig. S12 SEM images (left) and particle size distributions of **2** (right) after being sonicated for 8 (a), 15 (b) and 30 (c) minutes.

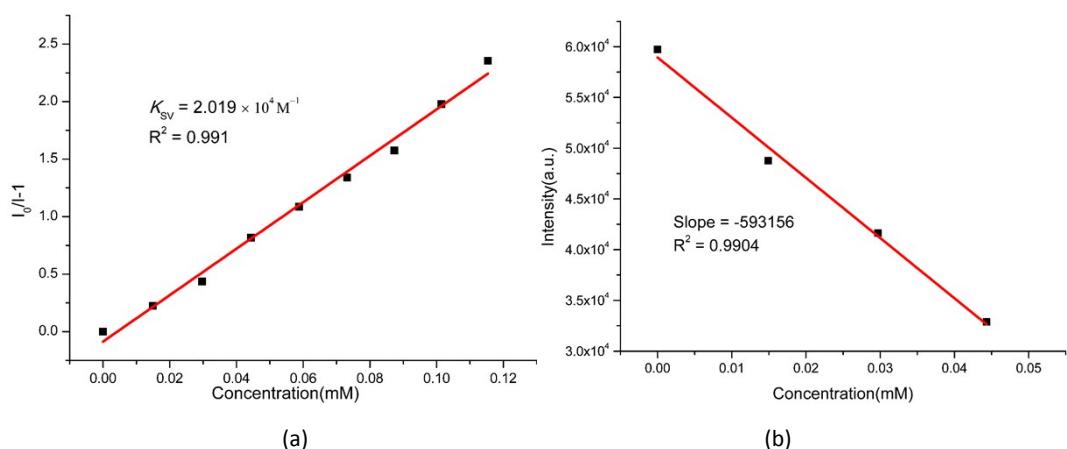


Fig. S13 (a) Stern-Volmer plot of $I_0/I-1$ versus low Fe^{3+} concentration in the aqueous suspension of **2**. (b) Linear region of fluorescence intensity for the suspensions of **2** in water upon incremental addition of Fe^{3+} solutions.

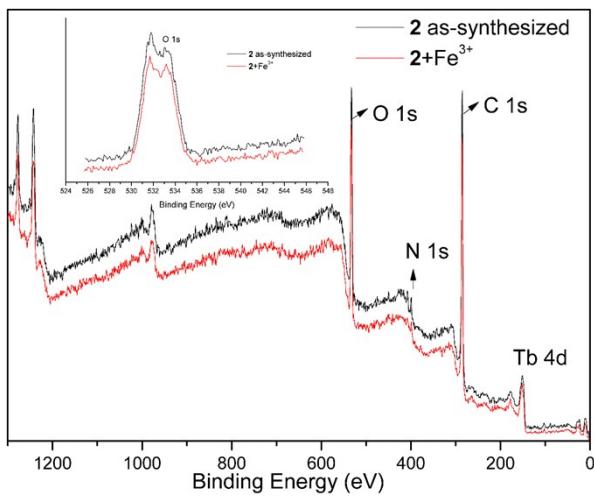


Fig. S14 Comparison of XPS spectra of **2** before (black) and after (red) its immersion in the Fe^{3+} aqueous solution.

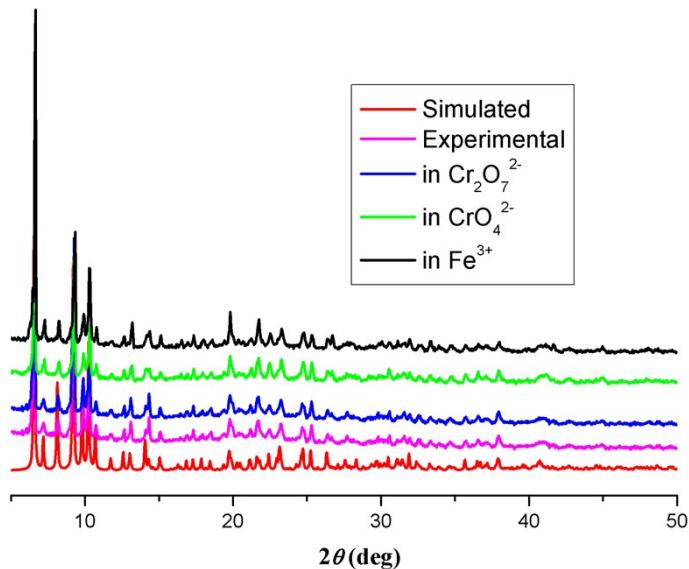


Fig. S15 PXRD patterns for **2** after the detection of Fe^{3+} , CrO_4^{2-} and $\text{Cr}_2\text{O}_7^{2-}$ ions.

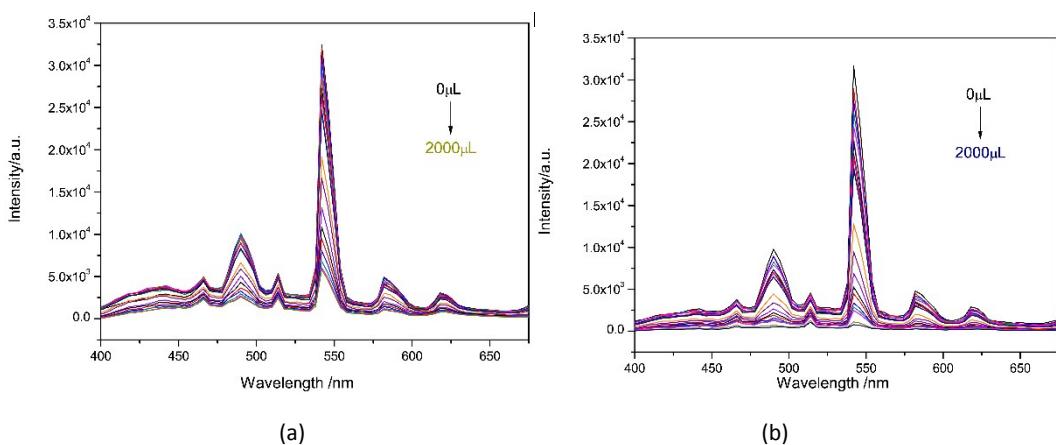


Fig. S16 Luminescent responses of a water suspension of **2** (2 mg/2 mL) towards different concentrations of (a) CrO_4^{2-} ions (2×10^{-3} M, 0 μL -2000 μL) and (b) $\text{Cr}_2\text{O}_7^{2-}$ (2×10^{-3} M, 0 μL -2000 μL).

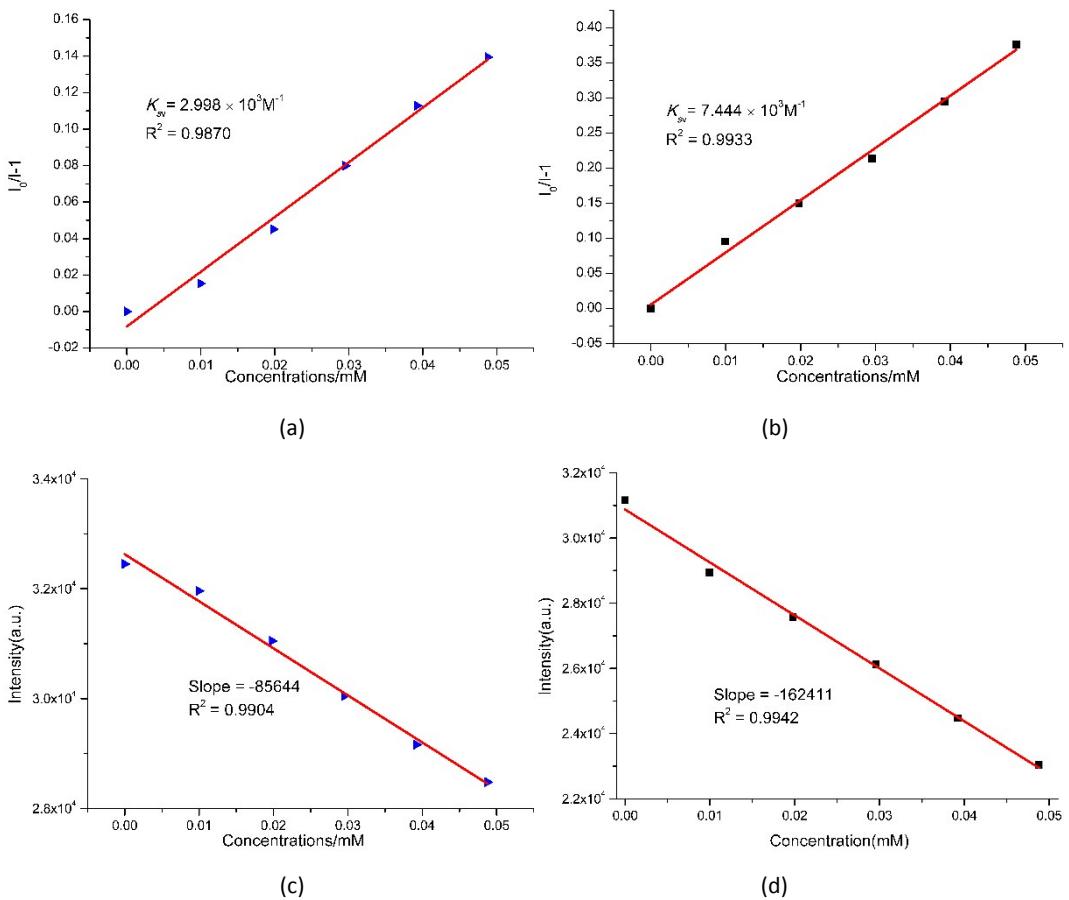


Fig. S17 Stern-Volmer plot of $I_0/I-1$ versus low concentration of CrO_4^{2-} (a) and $\text{Cr}_2\text{O}_7^{2-}$ (b) in the aqueous suspension of **2**, and linear region of fluorescence intensity for the suspensions of **2** in water upon incremental addition of CrO_4^{2-} (c) or $\text{Cr}_2\text{O}_7^{2-}$ (d) solutions.

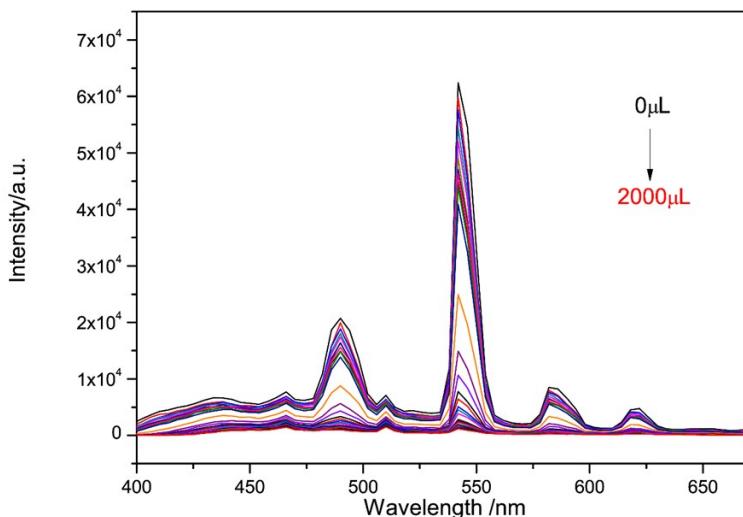


Fig. S18 Luminescence responses of an ethanol suspension of **2** (2 mg/2 mL) towards different concentrations of 4-NA ($1 \times 10^{-3} \text{ M}$, 0 μL -2000 μL).

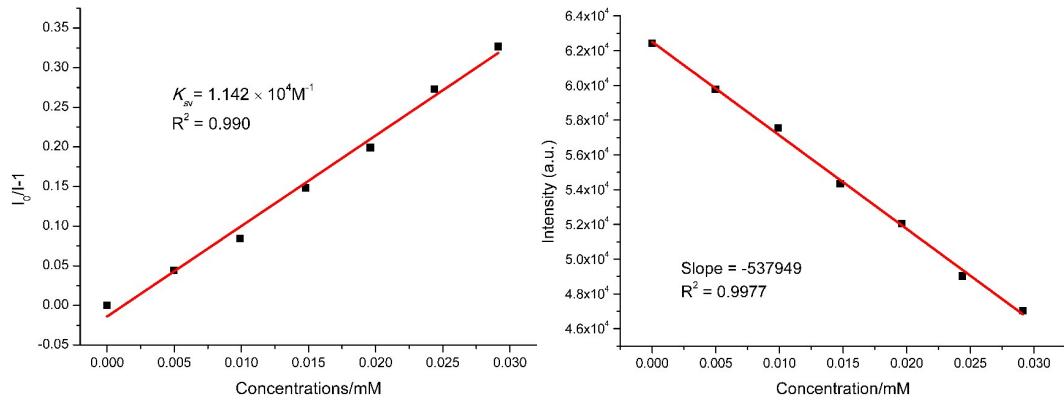


Fig. S19 (a) Stern-Volmer plot of $I_0/I-1$ versus low concentration of 4-NA in the ethanol suspension of **2**. (b) Linear region of fluorescence intensity for the suspensions of **2** in ethanol upon incremental addition of 4-NA solutions.

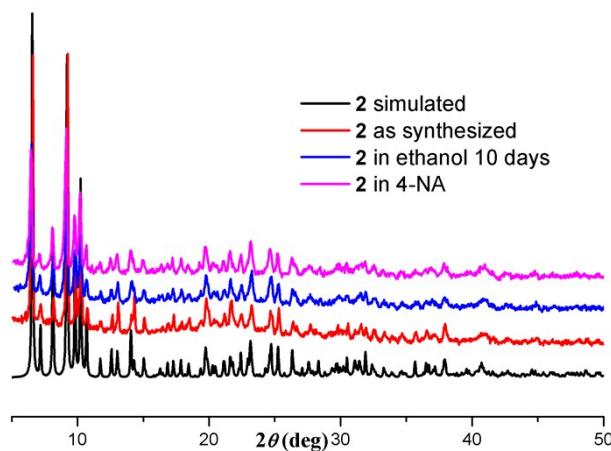


Fig. S20 PXRD patterns for **2** after the detection of 4-NA and being soaked in ethanol for 10 days

Table S3 The comparison of various Ln-complexes fluorescent sensors for Fe^{3+} , CrO_4^{2-} and $Cr_2O_7^{2-}$

Ln based complexes	Analyte	Quenching constant (K_{SV} , M^{-1})	Detection limits	Solvent	Ref
$[Tb_4L_4(NO_3)_2(Piv)_4] \cdot 2CH_3OH$	Fe^{3+}	1.86×10^4	$10\mu M$	water	This work
	CrO_4^{2-}	2.998×10^3	$52\mu M$		
	$Cr_2O_7^{2-}$	7.44×10^3	$27\mu M$		
$\{[Tb_2(Ccbp)_3 \cdot 6H_2O] \cdot 3Cl^- \cdot 4H_2O\}$	Fe^{3+}	1.143×10^5		ethanol	1
$\{[Eu(L1)(BPDC)_{1/2}(NO_3)] \cdot H_3O\}_n$	Fe^{3+}	5.16×10^4		DMF	2
$\{[Tb(L1)(BPDC)_{1/2}(NO_3)] \cdot H_3O\}_n$	Fe^{3+}	4.30×10^4		DMF	2
$\{[Eu(1,5-Nds)_{0.5}(ox)(phen)(H_2O)] \cdot H_2O\}_n$	Fe^{3+}			water	3
$\{[Eu_2(MFDA)_2(HCOO)_2(H_2O)_6]\cdot H_2O\}_n$	Fe^{3+}		$0.33 \mu M$	DMF	4
$\{[(CH_3)_2NH_2] \cdot [Tb(bptc)] \cdot xsolvents$	Fe^{3+}		$18.01 \mu M$	ethanol	5
$[Tb(TBOT)(H_2O)](H_2O)_4(DMF)(NMP)_{0.5}$	Fe^{3+}	5.51×10^3	$130 \mu M$	water	6
	$Cr_2O_7^{2-}$	1.37×10^4	$340 \mu M$		
$[Eu(L2)(HCOO)(H_2O)]_n$	CrO_4^{2-}	1.5374×10^3	$1.2 \mu M$	Water	7
	$Cr_2O_7^{2-}$	2.7626×10^3	$1.0 \mu M$		
$[Tb(L2)(HCOO)(H_2O)]_n$	CrO_4^{2-}	1.3070×10^3	$1.8 \mu M$	Water	7
	$Cr_2O_7^{2-}$	2.1335×10^3	$2.1 \mu M$		

[Eu ₂ (tpbpc) ₄ ·CO ₃ ·4H ₂ O]·DMF·solvent	CrO ₄ ²⁻	4.85×10 ³	0.33 ppm	water	8
	Cr ₂ O ₇ ²⁻	1.04×10 ⁴	1.07 ppm		
{[Eu ₂ (L3) _{1.5} (H ₂ O) ₂ EtOH]·DMF} _n	Fe ³⁺	2.942×10 ³	10 μM	DMF	9
	Cr ₂ O ₇ ²⁻	1.526×10 ³	10 μM		
{[Tb(TATAB)(H ₂ O) ₂]·NMP·H ₂ O} _n	Fe ³⁺	3667	1 μM	water	10
	Cr ₂ O ₇ ²⁻	11106	5 μM		
[Eu(Hpzbc) ₂ (NO ₃)]<·H ₂ O	Cr ₂ O ₇ ²⁻		22 μM	ethanol	11
[Eu(HL4)(H ₂ O) ₂ (NO ₃)]<·NO ₃	Fe ³⁺	4.03×10 ⁴	43μM	water	12
	Cr ₂ O ₇ ²⁻	7.52×10 ⁴	17μM		
[Tb(HL4)(H ₂ O) ₂ (NO ₃)]<·NO ₃	Fe ³⁺	4.54×10 ⁴	16μM	water	12
	Cr ₂ O ₇ ²⁻	4.59 × 10 ⁴	25μM		
[Eu(HPIDC)(m-bdc)·1.5H ₂ O] _n	Cr ₂ O ₇ ²⁻	4.1 × 10 ⁴		water	13
[Tb(HPIDC)(m-bdc)·1.5H ₂ O] _n	Cr ₂ O ₇ ²⁻	6.1 × 10 ³		water	13

Ccbp⁻ = 4-carboxy-1-(4-carboxybenzyl)pyridinium; H₂L1 = 2,5-di(pyridin-4-yl)terephthalic acid, BPDC = biphenyl-4,4'-dicarboxylic acid; 1,5-Nds = 1,5-naphthalenedisulfonate disulfonate; ox = oxalate; phen = 1,10-phenanthroline; m-H₂bdc = 1,3-benzenedicarboxylic acid; H₄bzptc = benzophenone-3,3',4',4'-tetracarboxylic acid; H₃PIDC = 2-(4-pyridyl)-1H-imidazole-4,5-dicarboxylic acid; H₂MFDA = 9,9-dimethyl-fluorene-2,7-dicarboxylic acid; H₄bptc = tetracarboxylic acid; NMP = N-methyl-2-pyrrolidone; H₃TBOT (2,4,6-tris[1-(3-carboxylphenoxy)ylmethyl]mesitylene); H₂L = 5-((2'-cyano-[1,1'-biphenyl]-4-yl)methoxy)isophthalic acid; H₂L3 = 5,5'-(carbonylbis(azanediyl))diisophthalic acid; Htpbpc = 4'-[4,2';6',4'']-terpyridin-4'-yl-biphenyl-4-carboxylic acid; H₃TATAB = 4,4',4''-s-triazine-1,3,5- triyltri-p-aminobenzoate acid; H₂pzbc = 3-(1H-pyrazol-3-yl) benzoic acid; H₃DMPIDC = 2-(3,4-dimethylphenyl)-1H- imidazole-4,5-dicarboxylic acid; H₃TATAB = 4,4',4''-s-triazine-1,3,5-triyltri-m-aminobenzoic acid, H₃PIDC = 2-(4-pyridyl)-1H- imidazole-4,5-dicarboxylic acid; m-H₂bdc = 1,3-benzenedicarboxylic acid; DMF = N,N-dimethylformamide

Table S4 The comparison of various coordination complexes fluorescent sensors for 4-NA

Coordination complex	Quenching constant (K _{SV} , M ⁻¹)	Detection limits	λ _{ex} (nm)	λ _{em} (nm)	Solvent	Ref
[Tb ₄ L ₄ (NO ₃) ₂ (Piv) ₄]·2CH ₃ OH	1.14 × 10 ⁴	8.5μM	360	544	ethanol	This work
[Cd _{1/2} (L5) _{1/3} (bib) _{1/2} (H ₂ O)] _n	6.6 × 10 ⁴			325	DMSO	14
{[Cd _{1/2} (HL5) ₂ (bibp) ₂]<·3H ₂ O} _n	1.1 × 10 ⁴			375	DMSO	14
[Cd ₂ (H ₂ L6) ₂ (H ₂ O) ₅]·5H ₂ O·2DMF	1.81 × 10 ⁴		340	480	isopropanol	15
{[Zn4(μ3-OH) ₂ (L7)(H ₂ O) ₂]<·2DMF} _n			350	400 red-shifted about 40 nm.	DMF	16
[Zn(L8) _{0.5} (1,10-phen)(H ₂ O)]·2H ₂ O	6556		330	456	DMA	17
[Zn(L8) _{0.5} (1,10-phen)(H ₂ O)]·2H ₂ O	3955		318	396	DMA	17

H₃L5 = tris(p-carboxyphenyl)phosphane oxide; bib = 1,4-bis(imidazol-1-yl)benzene bibp = 4,4'-bis(imidazol- 1-yl)biphenyl; H₄L6 = 2,5-bis-(3,5-dicarboxyphenyl)thiopheneamide; H₆L7 = [1,1';4',1'']terphenyl-3,5,2',5',3'',5''-hexacarboxylic acid; H₄L8 = bis-(3,5-dicarboxyphenyl)terephthalamide; DMA = N,N-dimethylacetamide; DMSO = dimethyl sulfoxide

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