



RSC Advances

Electronic Supplementary Information (ESI)

In-situ X-ray absorption spectroscopy study of CuO-NiO/CeO₂-ZrO₂ oxides: redox characterization and its effect in catalytic performance for partial oxidation of methane

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Element specific TPR profiles:

Ce L3-edge

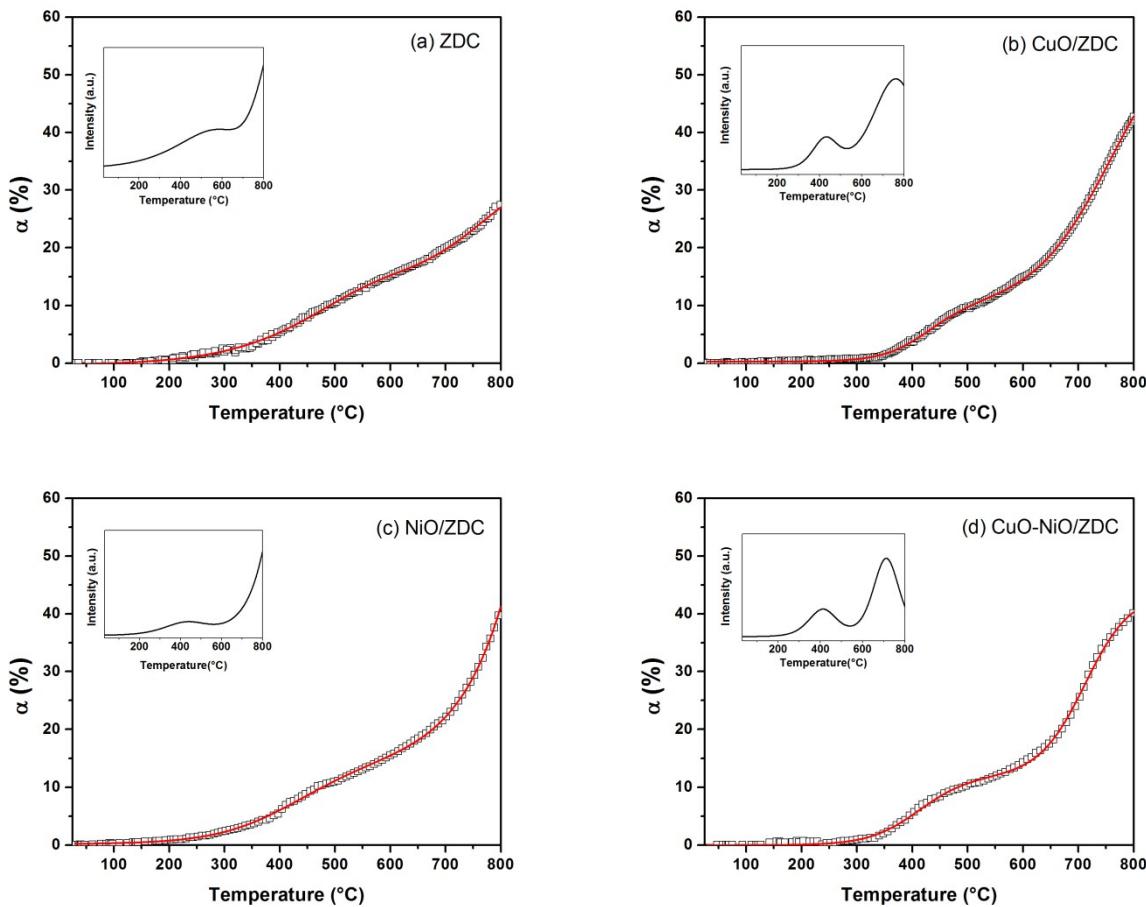


Figure S1: Degree of Ce reduction [$\alpha = \text{Ce}^{3+} / (\text{Ce}^{4+} + \text{Ce}^{3+})$] calculated for samples (a) ZDC, (b) CuO/ZDC, (c) NiO/ZDC and (d) CuO-NiO/ZDC during the course of in-situ XANES TPR experiments in 5 mol % H_2/He atmosphere (square symbols). The red solid line corresponds to the fitting of the reduction profile with two Boltzmann sigmoid functions. In the inset: simulated Ce-TPR profiles obtained from differentiation of the reduction profiles.

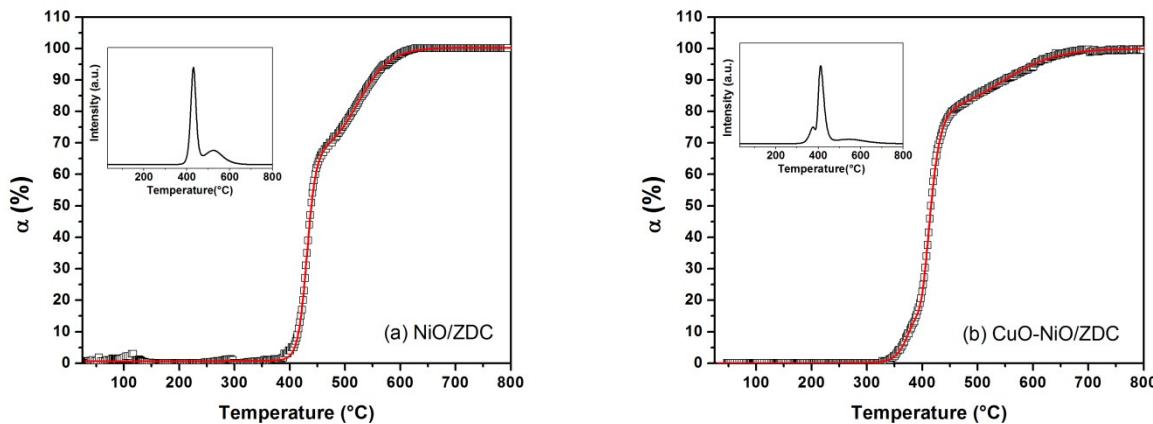
Ni K-edge

Figure S2: Degree of Ni reduction [$\alpha = \text{Ni}^0 / (\text{Ni}^{2+} + \text{Ni}^0)$] calculated for samples (a) NiO/ZDC and (b) CuO-NiO/ZDC during the course of in-situ XANES TPR experiments in 5 mol % H₂/He atmosphere (square symbols). The red solid line corresponds to the fitting of the reduction profile with two step functions in the case of NiO/ZDC and three step functions in the case of CuO-NiO/ZDC. In the inset: simulated Ni-TPR profiles obtained from differentiation of the reduction profiles.

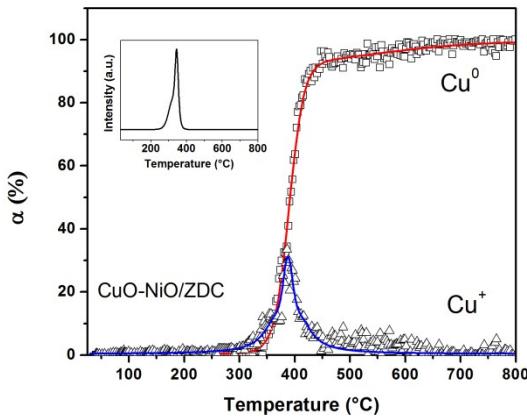
Cu K-edge

Figure S3: Degree of Cu²⁺ reduction [$\alpha_1 = \text{Cu}^0 / (\text{Cu}^{2+} + \text{Cu}^+ + \text{Cu}^0)$] and degree of Cu⁺ reduction [$\alpha_2 = \text{Cu}^+ / (\text{Cu}^{2+} + \text{Cu}^+ + \text{Cu}^0)$] calculated for sample CuO-NiO/ZDC during the course of in-situ XANES TPR experiments in 5 mol % H₂/He atmosphere (square symbols). The red solid line corresponds to the fitting of the Cu²⁺ to Cu⁰ reduction profile with two step functions. The blue solid line corresponds to the fitting of the Cu⁺ to Cu⁰ reduction profile with a pseudo-voigt curve. In the inset: simulated Cu-TPR profile obtained from the addition of both profiles and subsequent differentiation.