

Electronic Supplementary Information (ESI)

Brønsted acidic ionic liquids for cellulose hydrolysis in aqueous medium: Structural effects on acidity and glucose yield

Shiori Suzuki,[#] Yuko Takeoka, Masahiro Rikukawa and Masahiro Yoshizawa–Fujita*

Department of Materials and Life Sciences, Sophia University, 7-1 Kioi-cho, Chiyoda-ku, Tokyo 102-8554, Japan. E-mail: masahi-f@sophia.ac.jp*

[#] Now at Faculty of Natural System, Institute of Science and Engineering, Kanazawa University, Kakuma-machi, Kanazawa 920-1192, Japan.

The file includes: Supporting Fig. S1–S3 and Table S1–S2

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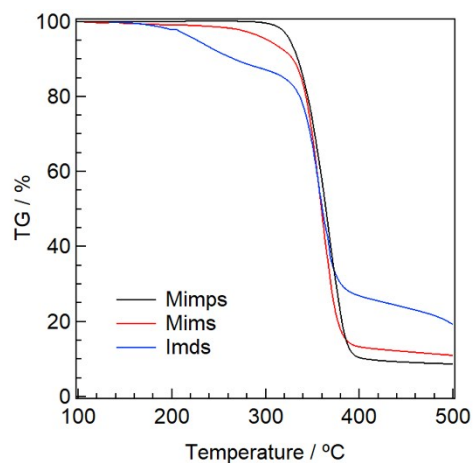


Fig. S1 TG curves of zwitterions.

- Instrument: TG-DTA 7200
(Hitachi High-Technologies, Co. Ltd.)
- Apparatus: open-type
aluminum pan
- Sample loading: 10 mg
- Heating rate: 10 °C min⁻¹
- N₂ flow: 200 mL min⁻¹

Table S1 Thermal property of zwitterions

Zwitterion ^a	$T_{d-5\%}$ ^b / °C
Mimps	326
Mims	302
Imds	226

^a Prior to TG measurement, each zwitterion was dried at 120°C for 1 h under N₂.

^b $T_{d-5\%}$: 5% weight loss temperature.

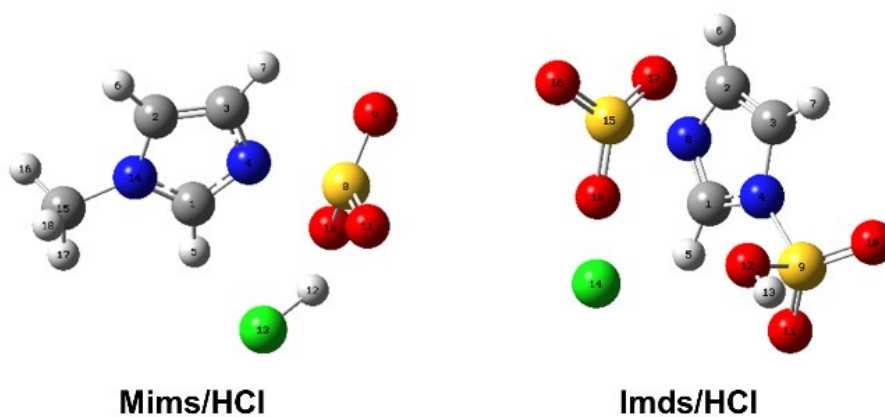


Fig. S2 Optimized molecular structures of BAILs with mono- or disulfuric acid groups using B3LYP/6-311G++(d, p).

Electronic Supplementary Information (ESI)

Table S2 Residual amount of residue after cellulose hydrolysis (160 °C, 15 min) in 1 M BAILs aqueous solution

Zw ^a	HX ^b	Glucose yield / %	Residue ^c / wt.%
Mim ps	H ₂ S O ₄	32.3±2.2	5.6
Bimp s		30.7±2.2	19
Oimp s		36.3±3.8	21

Mim ps	HCl	23.7±0.3	24
Bimp s		26.0±1.5	30
Oimp s		27.7±1.2	26

^a Zw: zwitterion. ^b HX: kinds of acid used to prepare BAILs called as Zw/HX; X is corresponded to anion species. ^c Represented as wt.% to the starting weight of cellulose (10 mg).

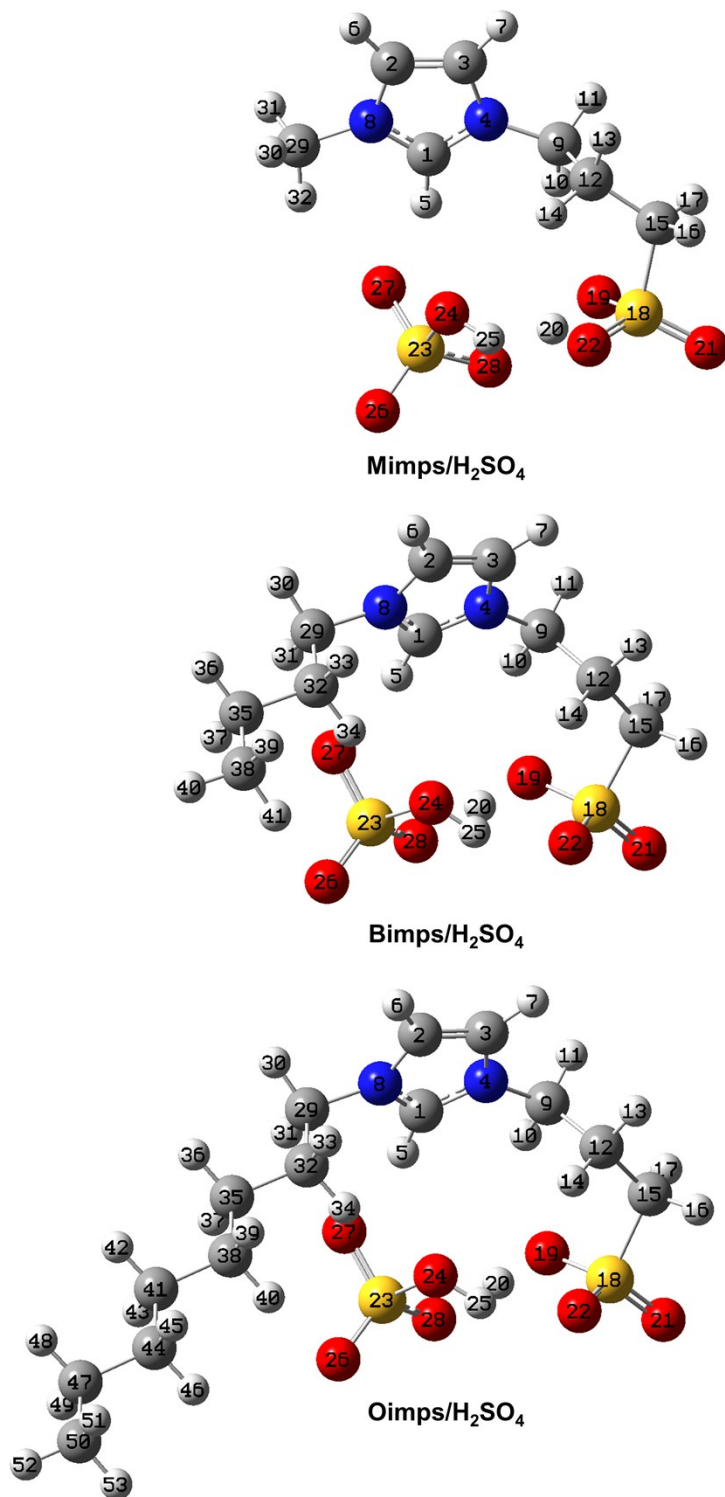


Fig. S3 (1) Optimized molecular structures of BAILs with HSO₄ anion using B3LYP/6-311G++(d, p).

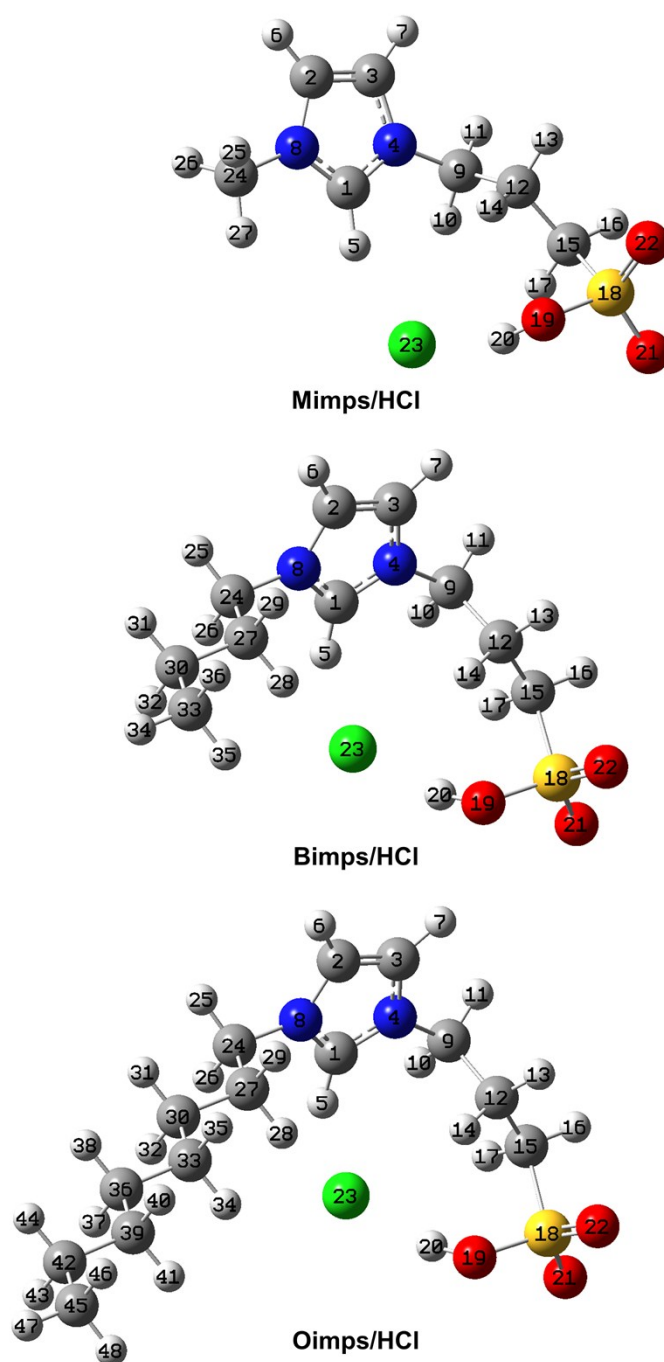


Fig. S3 (2) Optimized molecular structures of BAILs with Cl anion using B3LYP/6-311G++(d, p).

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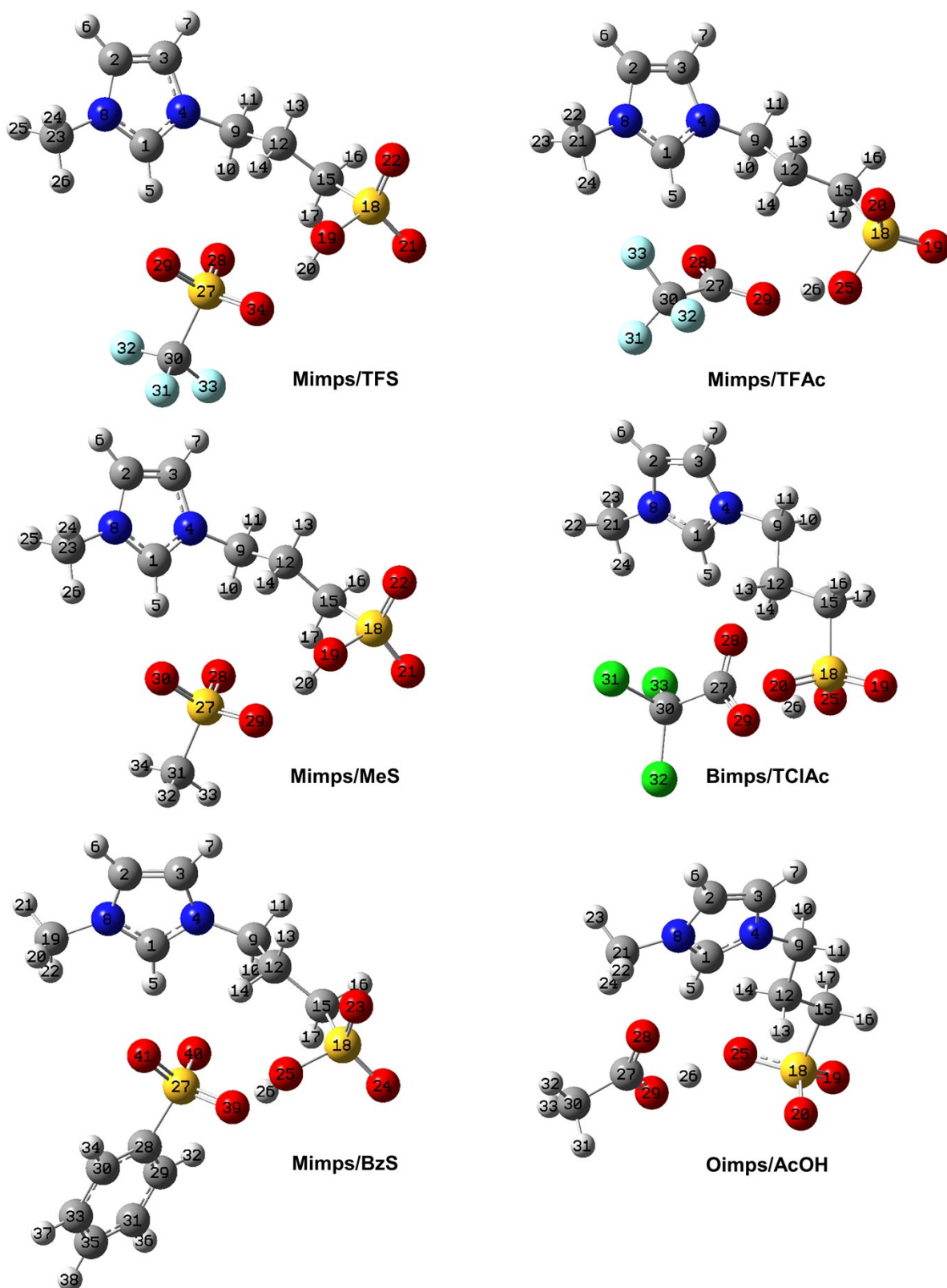
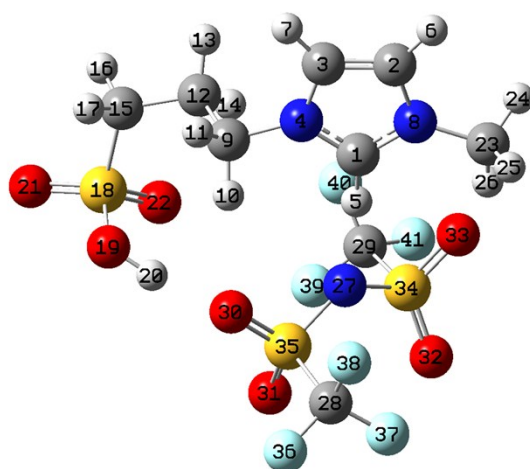
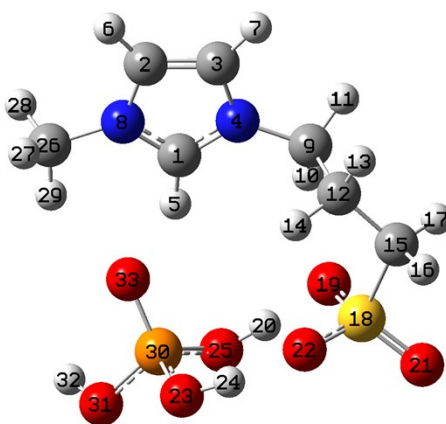


Fig. S3 (3) Optimized molecular structures of BAILs with sulfonic acid anion or carboxylic acid anion using B3LYP/6-311G++(d, p).

Electronic Supplementary Information (ESI)



Mimps/HTFSI



Mimps/H₃PO₄

Fig. S3 (4) Optimized molecular structures of BAILs with TFSI anion or phosphoric acid anion using B3LYP/6-311G++(d, p).