

## Electronic Supplementary Information

### The Synthesis of Green Fluorescent Carbon Dots for Warm White LEDs

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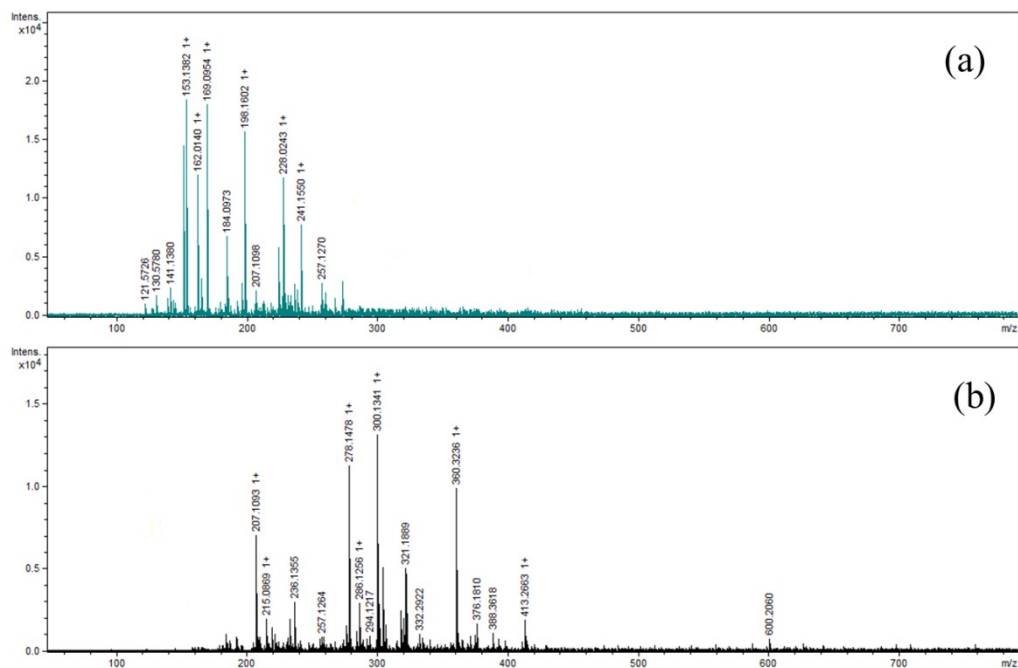


Figure S1 ESI-MS spectra of CDs-b (a) and CDs-s (b) in the region of 0–800 m/z.

Table S1 Elemental composition of CDs-o, CDs-s, CDs-b, CDs-w and CDs-e

	C	H	N	O*
	(wt%)	(wt%)	(wt%)	(wt%)
CDs-o	52.80	5.44	7.76	34.00
CDs-s	55.00	5.60	6.67	32.73
CDs-b	53.24	5.63	7.08	34.05
CDs-w	55.51	4.97	0	39.52
CDs-e	56.52	5.08	0	38.40

\*By difference.

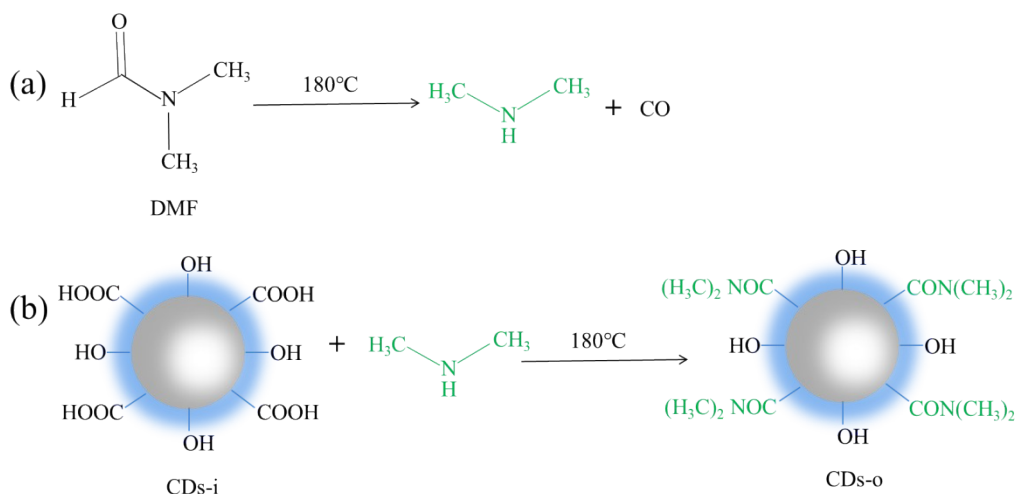


Figure S2 (a) Decomposition of DMF at high temperature. (b) Schematic illustration of the strategy for preparation of CDs-o.

Table S2 The QY of blue emissive CDs prepared by our group

materials	solvent	emission peak (nm)	approach	QY	references
citric acid, KH-792	water	448	Hydrothermal method	57.3%	[1]
citric acid, ethylenediamine	water	444	Hydrothermal method	67.06%	[2]
citric acid, ammonia solution	water	440	Hydrothermal method	35.4%	[3]
citric acid, L-cysteine	water	420	Hydrothermal method	67.41%	[4]
ammonium citrate, phosphates	water	450	Hydrothermal method	62%	[5]

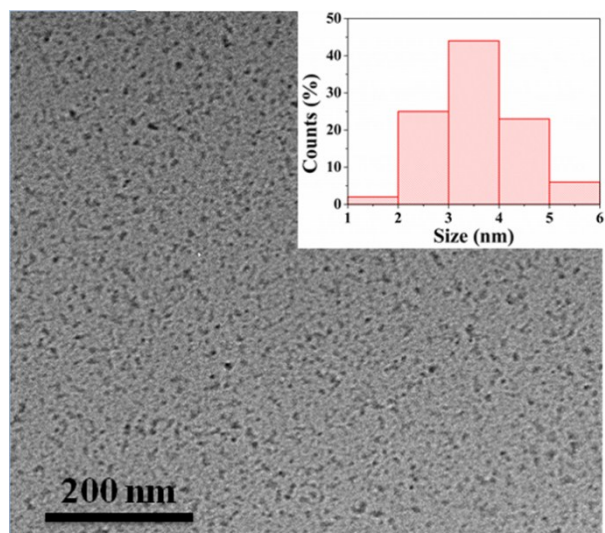


Figure S3 TEM images of CDs-w (inset is the size distribution).

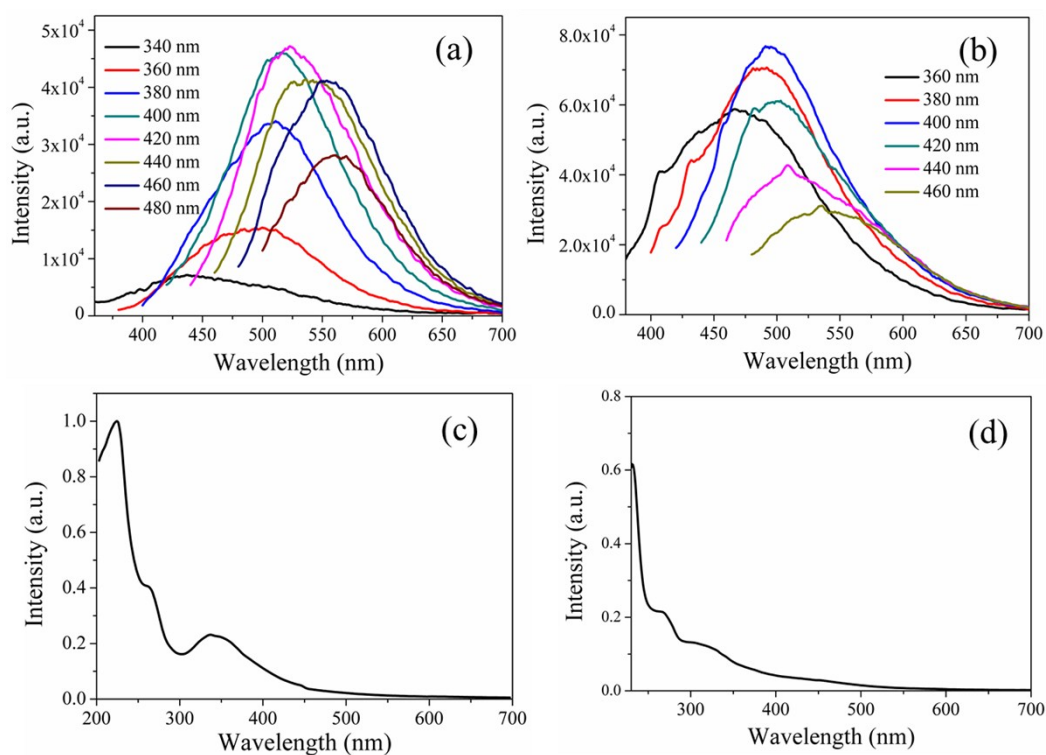


Figure S4 UV-vis absorption spectra, PL spectra at different excitation wavelengths of CDs-we (a, c) and CDs-ee (b, d).

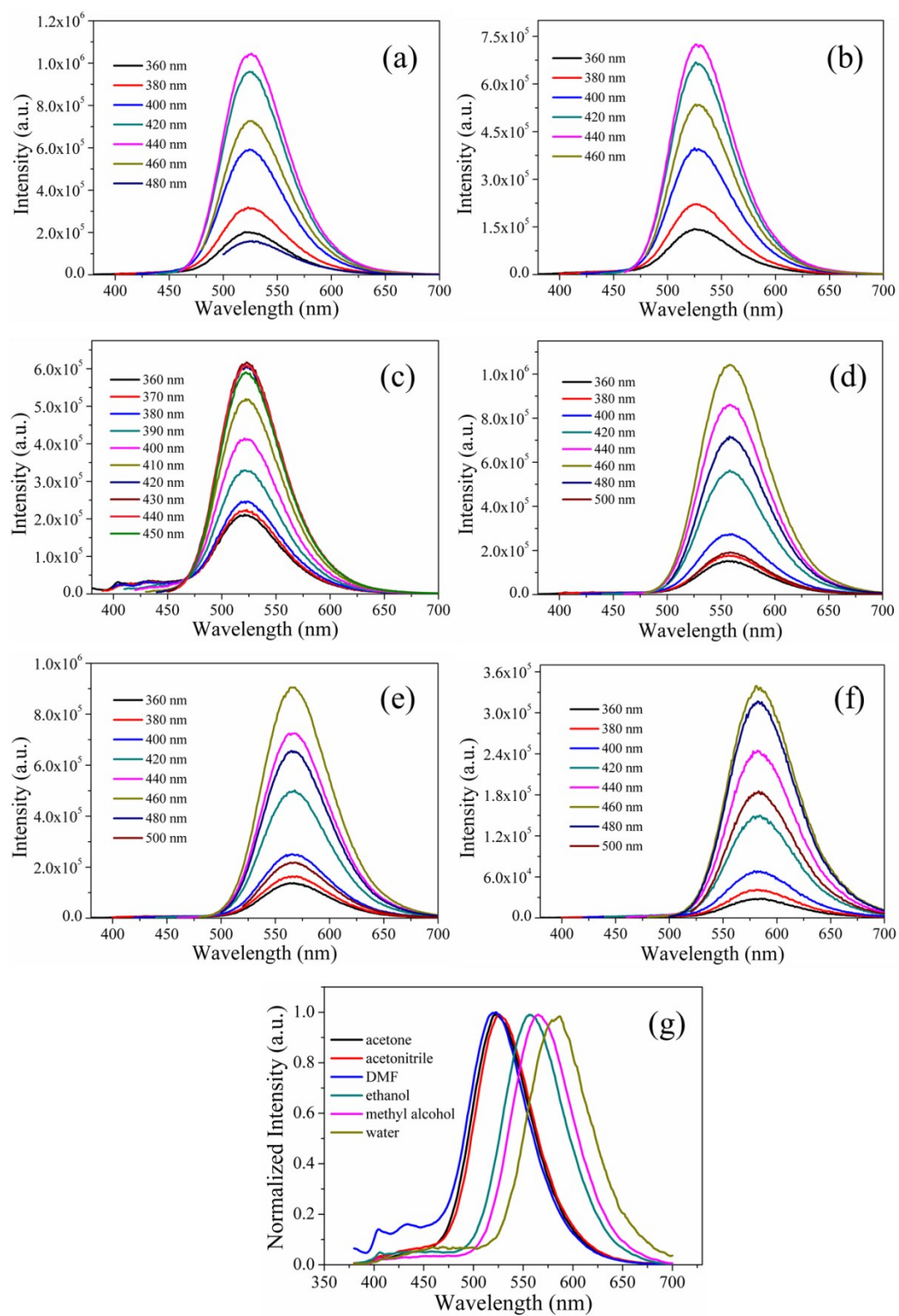


Figure S5 The photoluminescence spectra of CDs-o excited by different excitation wavelengths in solvents of acetone (a), acetonitrile (b), DMF (c), ethanol (d), methyl alcohol (e), water (f), and the photoluminescence spectra of the same CDs in 6 solvents excited by 360 nm light (g).

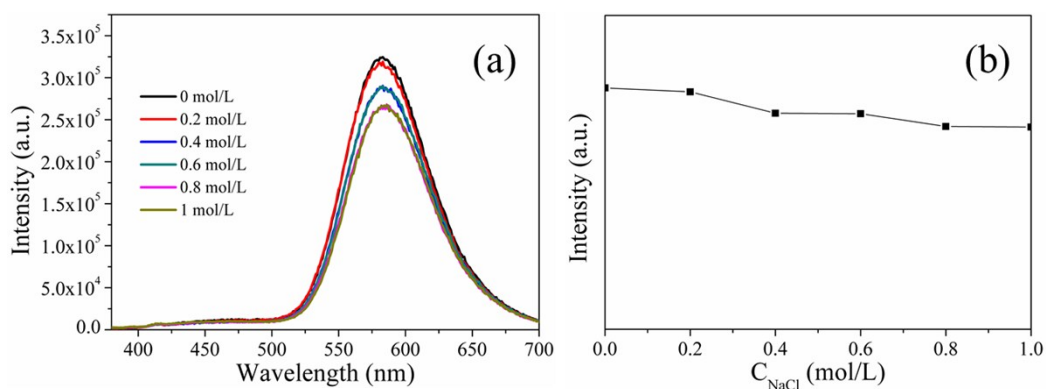


Figure S6 The photoluminescence spectra of CDs-o aqueous solutions (a) and PL intensity (b) (regulated by concentrations of NaCl from 0 to 1 mol/L).

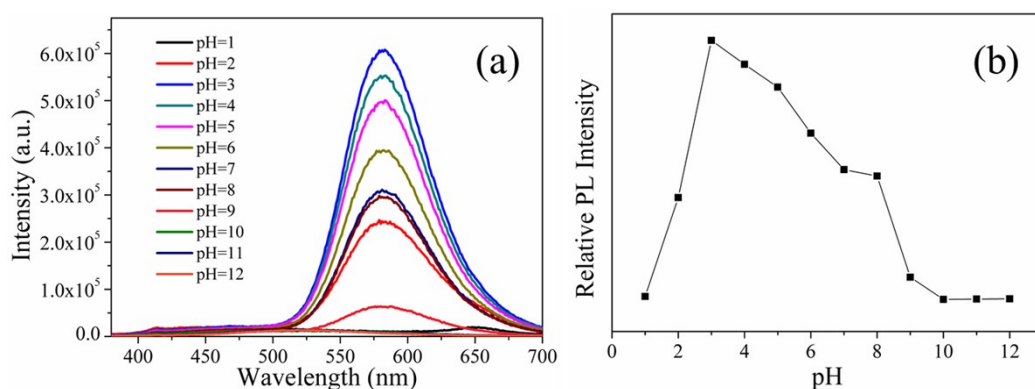


Figure S7 The photoluminescence spectra (a) and relative PL intensity of CDs-o aqueous solution with the increase of pH from 1 to 12.

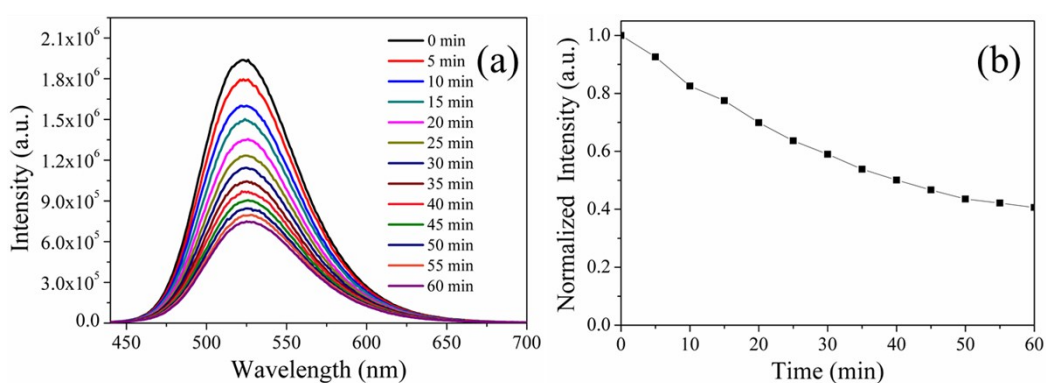


Figure S8 Photoluminescence spectra (a) and variation of PL intensity (b) of CDs-o under UV light during different irradiation time.

## References

- [1] F. Zhang, X. T. Feng, Y. Zhang, L. P. Yan, Y. Z. Yang and X. G. Liu. *Nanoscale*, 2016, **8**, 8618–8632.

- [2] Y. L. Wang, Y. Q. Zhao, Y. Zhang, F. Zhang, X. T. Feng, L. Chen, Y. Z. Yang and X. G. Liu. *RSC Advances*, 2016, **6**, 38761–38768.
- [3] Y. Zhang, P. P. Cui, F. Zhang, X. T. Feng, Y. L. Wang, Y. Z. Yang and X. G. Liu. *Talanta*, 2016, **152**, 288–300.
- [4] Y. L. Wang, Y. Q. Zhao, F. Zhang, L. Chen, Y. Z. Yang and X. G. Liu. *New Journal of Chemistry*, 2016, **40**, 8710–8716.
- [5] F. Zhang, Y. L. Wang, Y. Q. Miao, Y. H. He, Y. Z. Yang and X. G. Liu. *Applied Physics Letters*, 2016, **109**, 083103–083107.