

Supplementary Material

Silicalite-1 zeolite acidification by Zn modification and its catalytic properties for iso-butane conversion

Guodong Liu,^a Jiaxu Liu,^a Ning He,^a Cuilan Miao,^a Jilei Wang,^a Qin Xin^b and Hongchen Guo^{*a}

^a Department of Catalytic Chemistry and Engineering & State Key Laboratory of Fine Chemicals, Dalian University of Technology, Dalian 116012, P. R. China

^b State Key Laboratory of Catalysis, Dalian Institute of Chemical Physics, Chinese Academy of Sciences, Dalian 116023, P. R. China

* Corresponding author, E-mail: hongchengu@dlut.edu.cn

Table S1. XRF analysis of parent S-1 zeolite.

Elements	Compounds	Composition /wt.%
Si	SiO ₂	99.92
Na	Na ₂ O	0.034
Ti	TiO ₂	0.010
Fe	Fe ₂ O ₃	0.010
Zn	ZnO	0.001
Cl	Cl	0.015

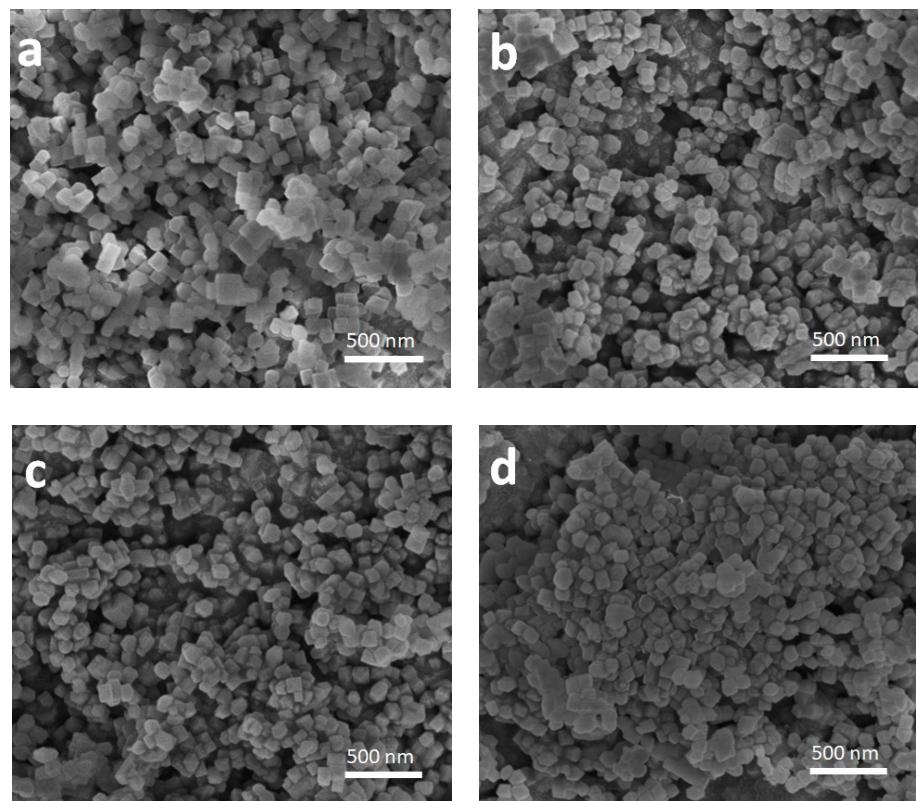


Figure S1. FE-SEM images of $\text{Zn}_x/\text{S-1}$ catalysts with different Zn loading: parent S-1 zeolite (a); $\text{Zn}_{1.0}/\text{S-1}$ (b); $\text{Zn}_{6.0}/\text{S-1}$ (c); $\text{Zn}_{12.0}/\text{S-1}$ (d).

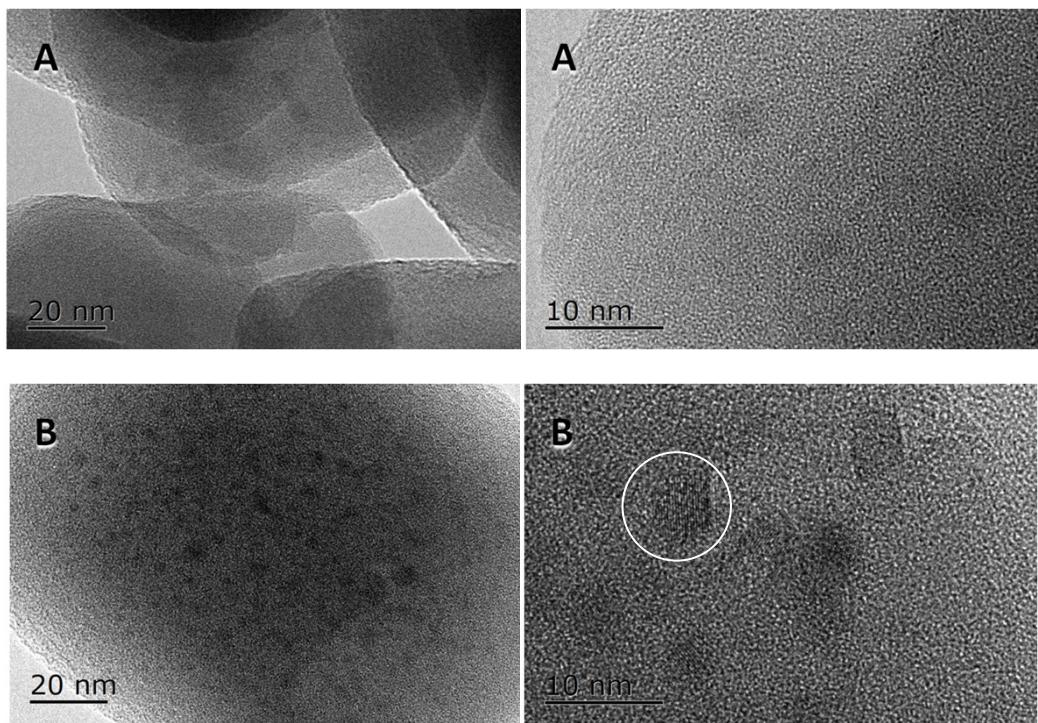


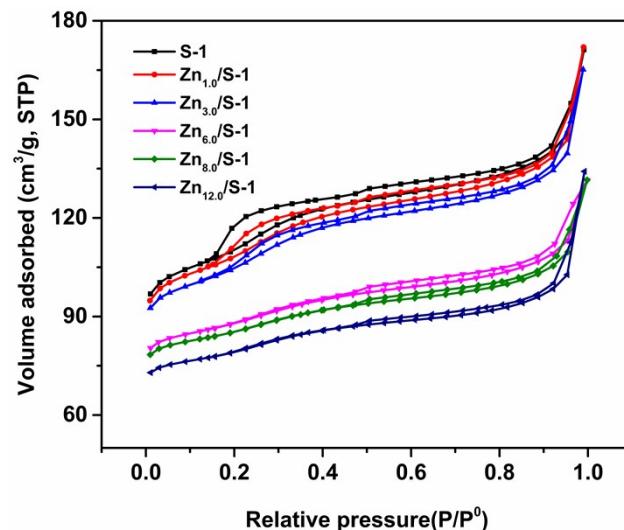
Figure S2. HRTEM images of $\text{Zn}_{6.0}/\text{S-1}$ (A) and $\text{Zn}_{12.0}/\text{S-1}$ (B).

Table S2. The relative crystallinity of $\text{Zn}_x/\text{S}-1$ catalysts

Catalysts	Crystallinity (%)
S-1	100
$\text{Zn}_{1.0}/\text{S}-1$	85.1
$\text{Zn}_{3.0}/\text{S}-1$	80.3
$\text{Zn}_{6.0}/\text{S}-1$	75.2
$\text{Zn}_{8.0}/\text{S}-1$	61.8
$\text{Zn}_{12.0}/\text{S}-1$	53.5

Table S3. The acid amounts of catalysts as determined by $\text{NH}_3\text{-TPD}$

Catalysts	Acid amount (mmol/g)
S-1	0.00367
$\text{Zn}_{1.0}/\text{S}-1$	0.0736
$\text{Zn}_{3.0}/\text{S}-1$	0.134
$\text{Zn}_{6.0}/\text{S}-1$	0.142
$\text{Zn}_{8.0}/\text{S}-1$	0.151
$\text{Zn}_{12.0}/\text{S}-1$	0.137
$\text{Zn}_{6.0}/\text{ZSM-5}$	0.555
HZSM-5	0.676

**Figure S3.** N_2 adsorption-desorption isotherm of $\text{Zn}_x/\text{S}-1$ (x from 0.0 to 12.0).

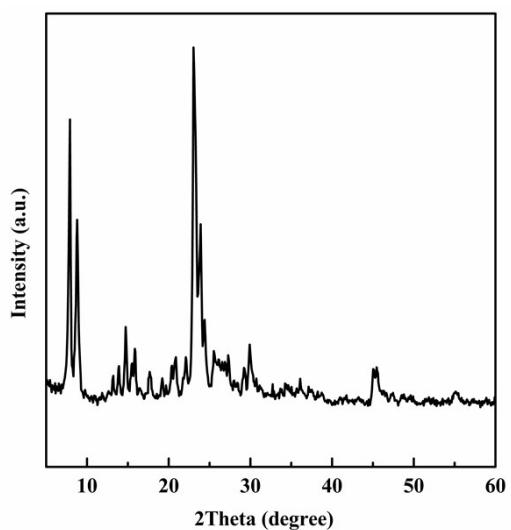


Figure S4. XRD pattern of H-type ZSM-5 zeolite.

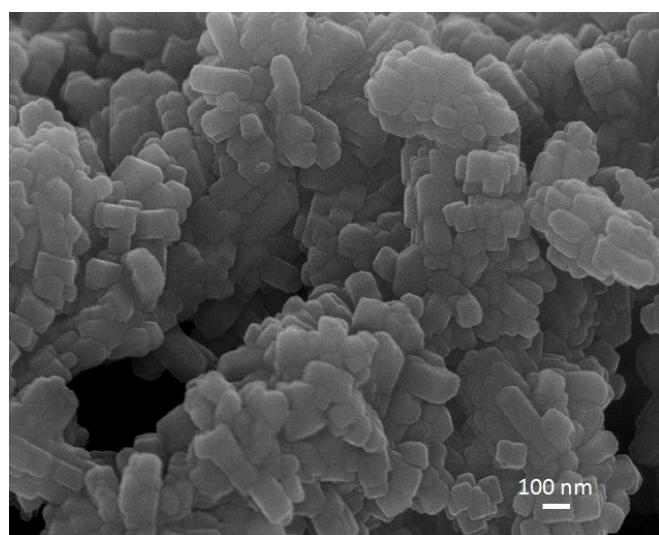


Figure S5. SEM images of H-type ZSM-5 zeolite under the scale bar of 100 nm.