

Supplementary Information for

cis-1 Isomers of tethered bismethano [70]fullerene
as electron acceptors in organic photovoltaics

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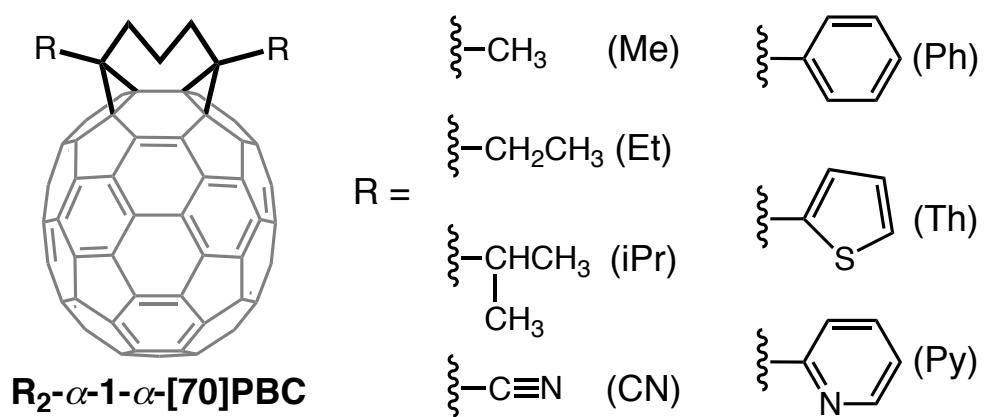


Fig. S1 Structures of R₂-α-1-α-[70]PBC (R = Me, Et, iPr, CN, Ph, Th, Py).

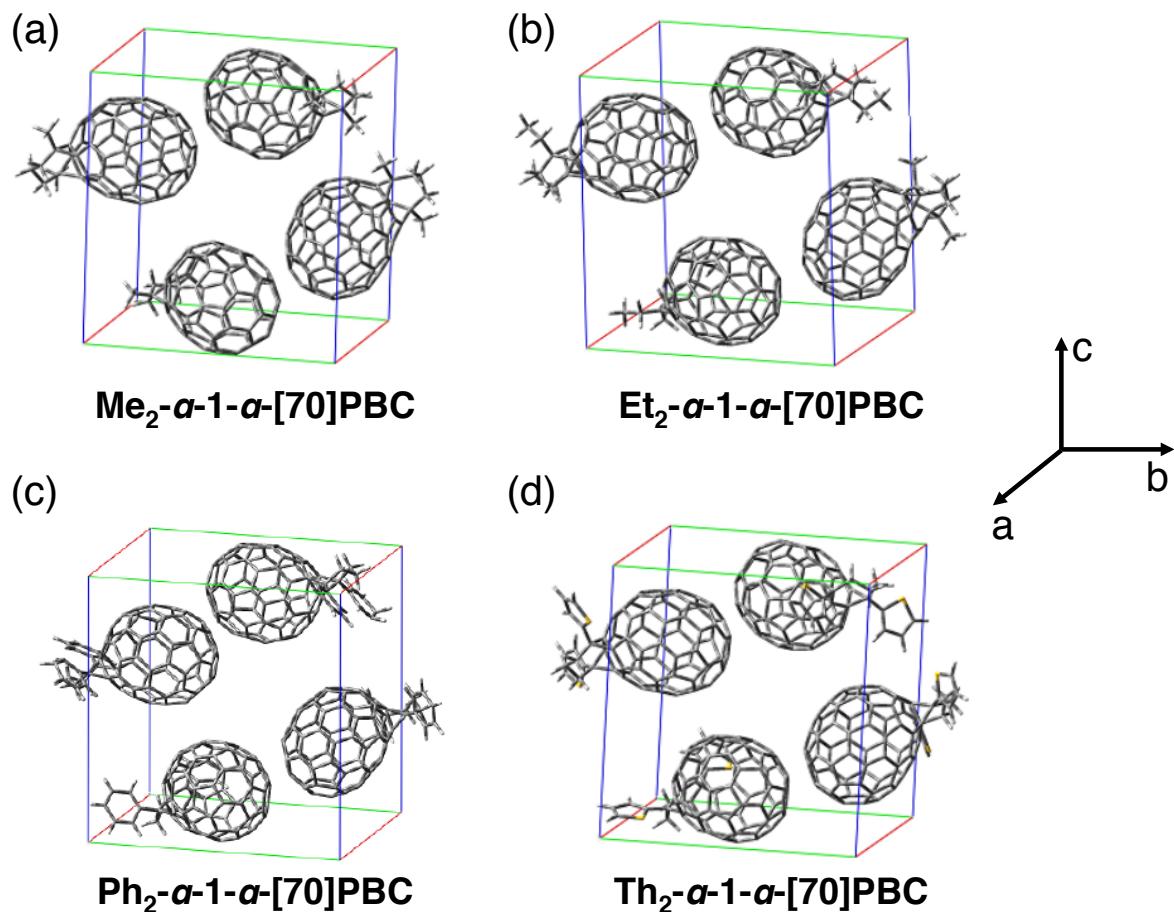


Fig. S2 Representative examples for the structures of $\text{R}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$ solid systems optimized under periodic boundary conditions at PM6-D3 based on MOPAC. (a) $\text{Me}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$, (b) $\text{Et}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$, (c) $\text{Ph}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$, and (d) $\text{Th}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$.

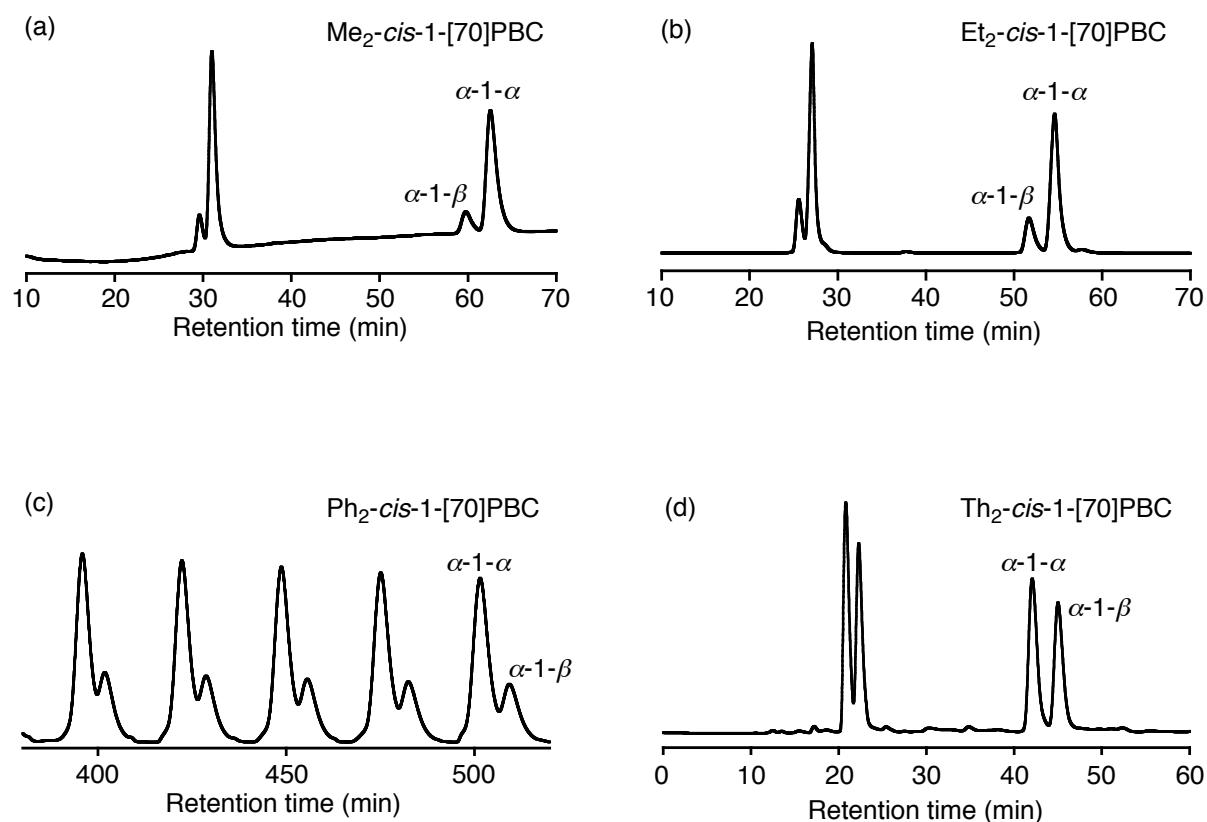


Fig. S3 HPLC profiles with Buckyprep column of (a) $\text{Me}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$, (b) $\text{Et}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$, (c) $\text{Ph}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$, and (d) $\text{Th}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$.

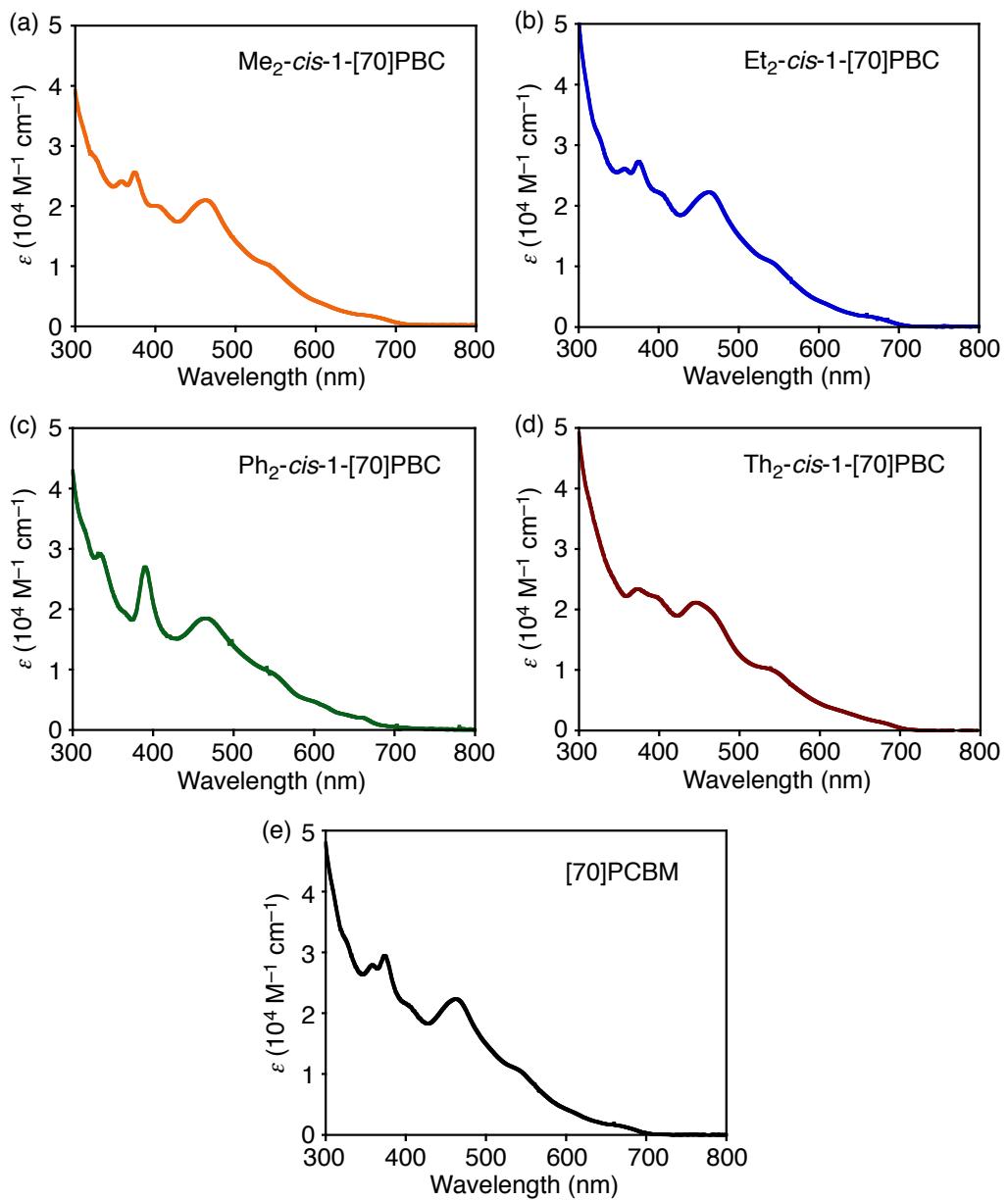


Fig. S4 UV-vis absorption spectra of (a) $\text{Me}_2\text{-cis-1-[70]PBC}$, (b) $\text{Et}_2\text{-cis-1-[70]PBC}$, (c) $\text{Ph}_2\text{-cis-1-[70]PBC}$, (d) $\text{Th}_2\text{-cis-1-[70]PBC}$, and (e) $[70]\text{PCBM}$ in ODCB.

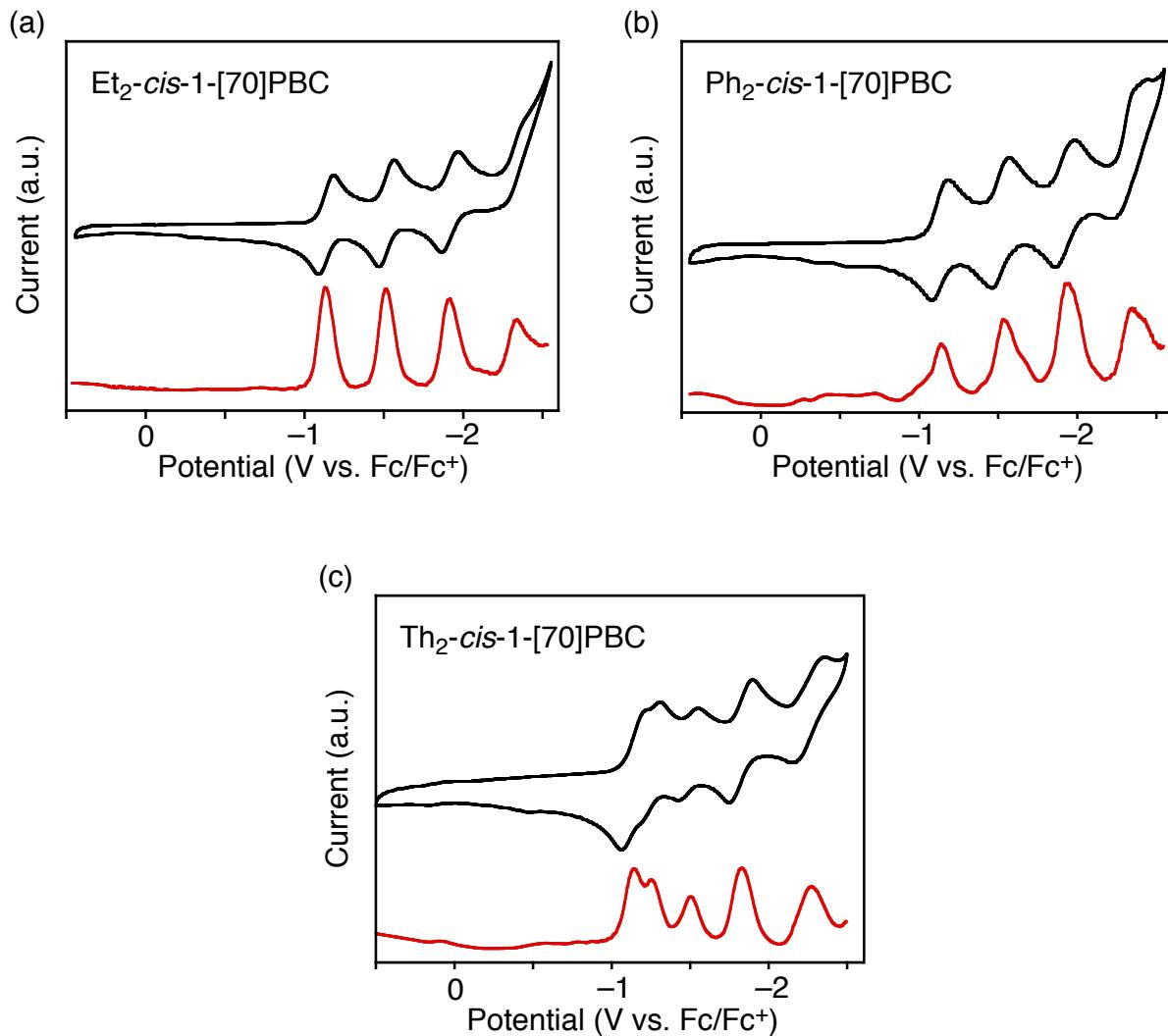


Fig. S5 Cyclic voltammograms (upper) and differential pulse voltammograms (lower) of (a) $\text{Et}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$, (b) $\text{Ph}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$, and (c) $\text{Th}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$ versus a reference electrode of Fc/Fc^+ measured in ODCB/acetonitrile mixture (v/v = 5:1) containing 0.1 M tetrabutylammonium hexafluorophosphate (Bu_4NPF_6). Sweep rate: 0.1 V s⁻¹; reference electrode, Ag/Ag^+ (0.01 M AgNO_3 , 0.09 M Bu_4NPF_6 in acetonitrile).

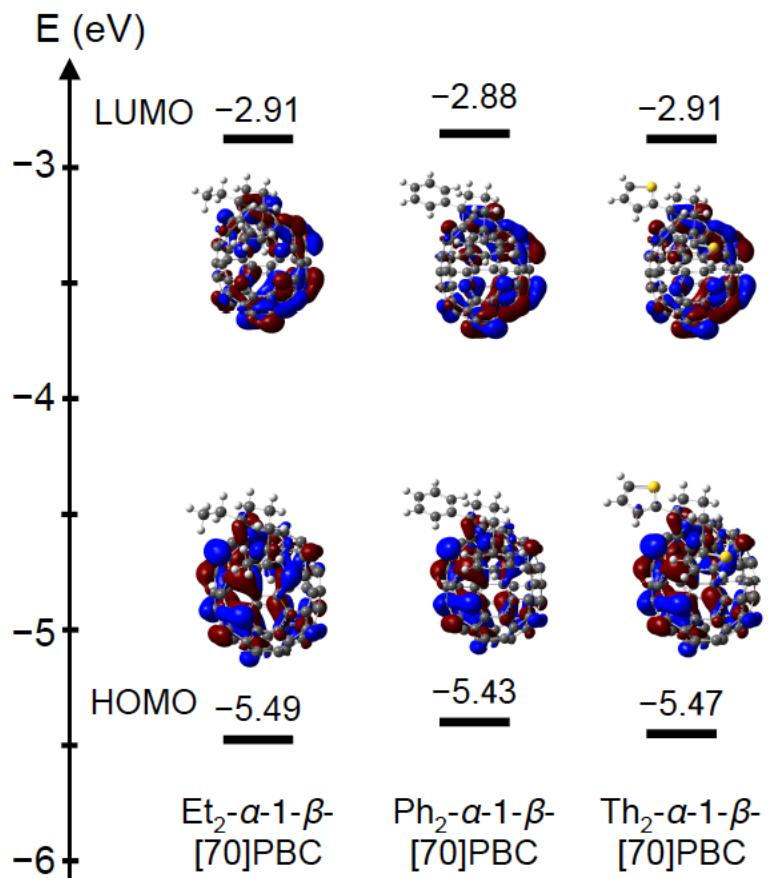


Fig. S6 Optimized geometries, HOMO/LUMO electron density distributions, and energy levels of $\text{Et}_2\text{-}\alpha\text{-}1\text{-}\beta\text{-}[70]\text{PBC}$, $\text{Ph}_2\text{-}\alpha\text{-}1\text{-}\beta\text{-}[70]\text{PBC}$, and $\text{Th}_2\text{-}\alpha\text{-}1\text{-}\beta\text{-}[70]\text{PBC}$ by DFT calculations using RB3LYP/6-31G(d) model.

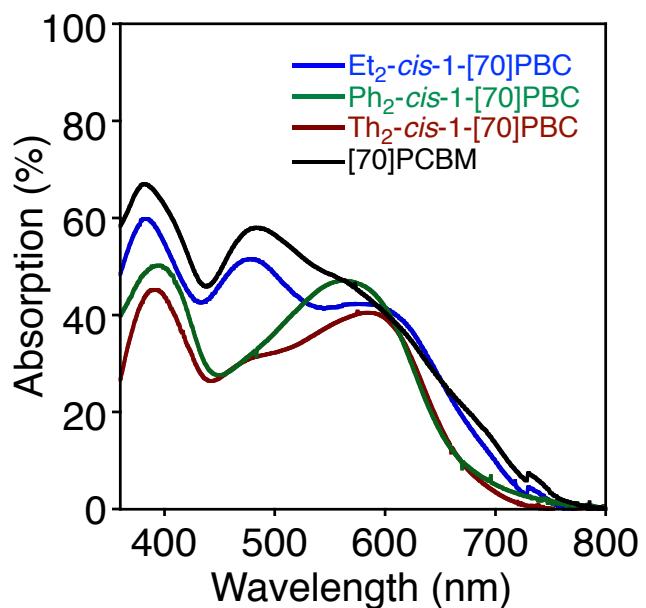


Fig. S7 UV-visible absorption spectra of blend films of PCDTBT:Et₂-*cis*-1-[70]PBC (blue), PCDTBT:Ph₂-*cis*-1-[70]PBC (green), PCDTBT:Th₂-*cis*-1-[70]PBC (brown), and PCDTBT:[70]PCBM (black) on ITO/PEDOT:PSS substrates. Note here that the total absorptions of the ITO/PEDOT:PSS/PCDTBT:fullerene/TiO_x/Al devices are higher than the values in this figure due to the reflection from the Al electrode through the optical spacer, TiO_x.

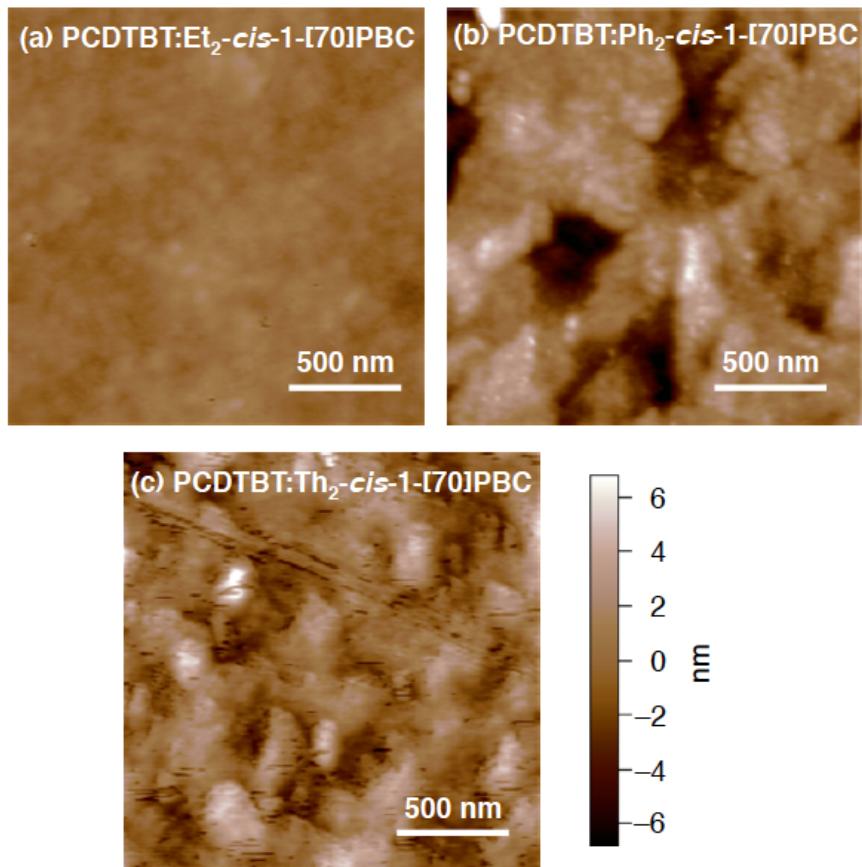


Fig. S8 Tapping-mode atomic force micrographs of ((a) PCDTBT:Et₂-*cis*-1-[70]PBC, (b) PCDTBT:Ph₂-*cis*-1-[70]PBC, and (c) PCDTBT:Th₂-*cis*-1-[70]PBC on ITO/PEDOT:PSS substrates. The color scale represents the height topography, with bright and dark representing the highest and lowest features, respectively. The rms surface roughnesses are (a) 0.59, (b) 2.3, and (c) 1.9 nm, respectively.

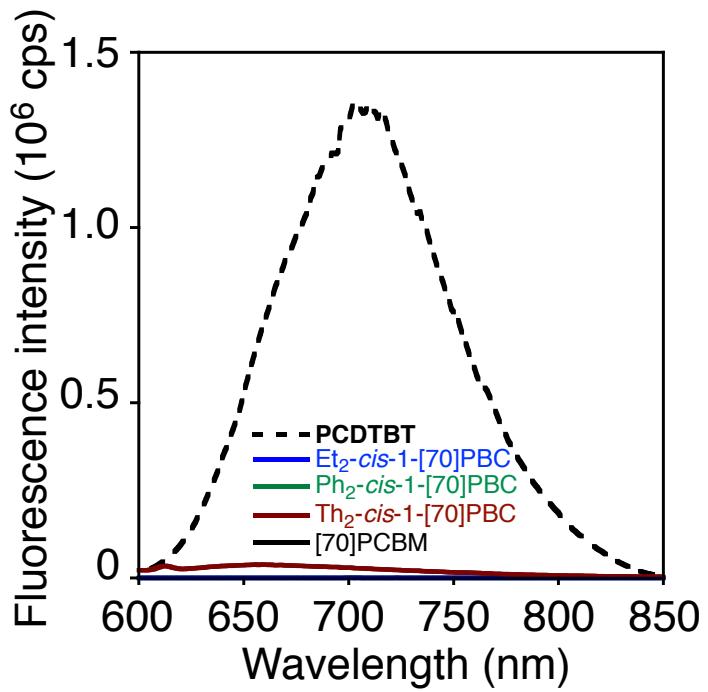


Fig. S9 Photoluminescence spectra of PCDTBT single component film (black dotted line) and blend films of PCDTBT with $\text{Et}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$ (blue solid line), $\text{Ph}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$ (green solid line), $\text{Th}_2\text{-}cis\text{-}1\text{-}[70]\text{PBC}$ (brown solid line), and [70]PCBM (black solid line). The excitation wavelengths are 540 nm. The emission intensities were normalized by that of the PCDTBT film considering the difference in the absorbances of the PCDTBT:fullerene films at the excitation wavelength.

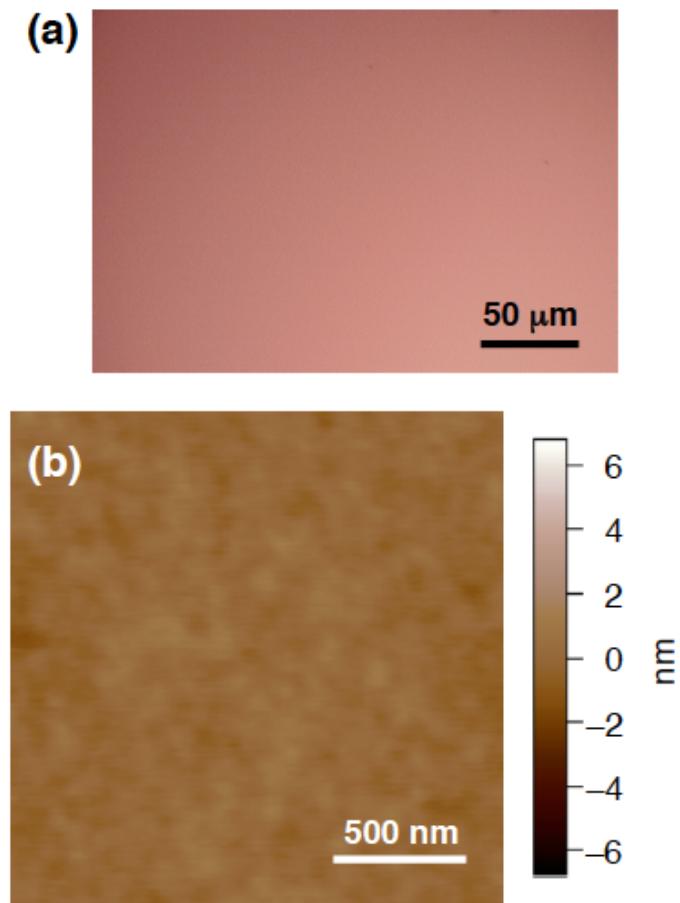


Fig. S10 (a) Optical microscopy image and (b) tapping-mode atomic force micrograph of PCDTBT:[70]PCBM on ITO/PEDOT:PSS substrates. In (b), the color scale represents the height topography, with bright and dark representing the highest and lowest features, respectively. The rms surface roughness is 0.35 nm.

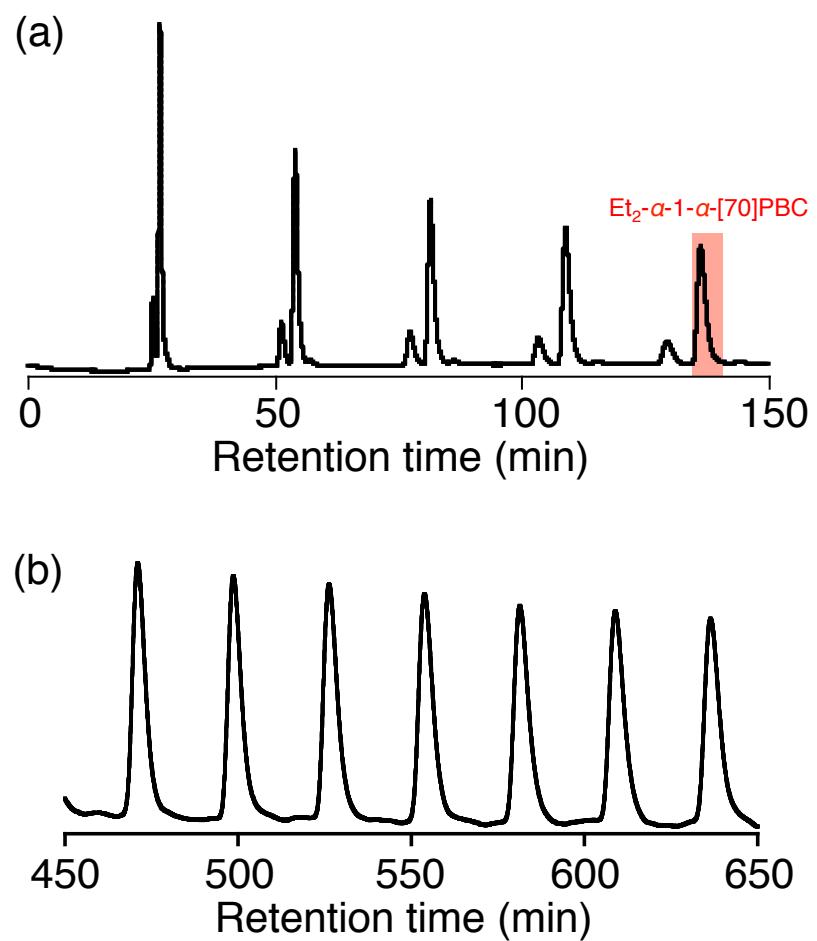


Fig. S11 (a) HPLC profile with 5PBB column of Et₂-*cis*-1-[70]PBC. The fraction marked with red color was collected and assigned to Et₂- α -1- α -[70]PBC. (b) HPLC profile with 5PBB column of Et₂- α -1- α -[70]PBC.

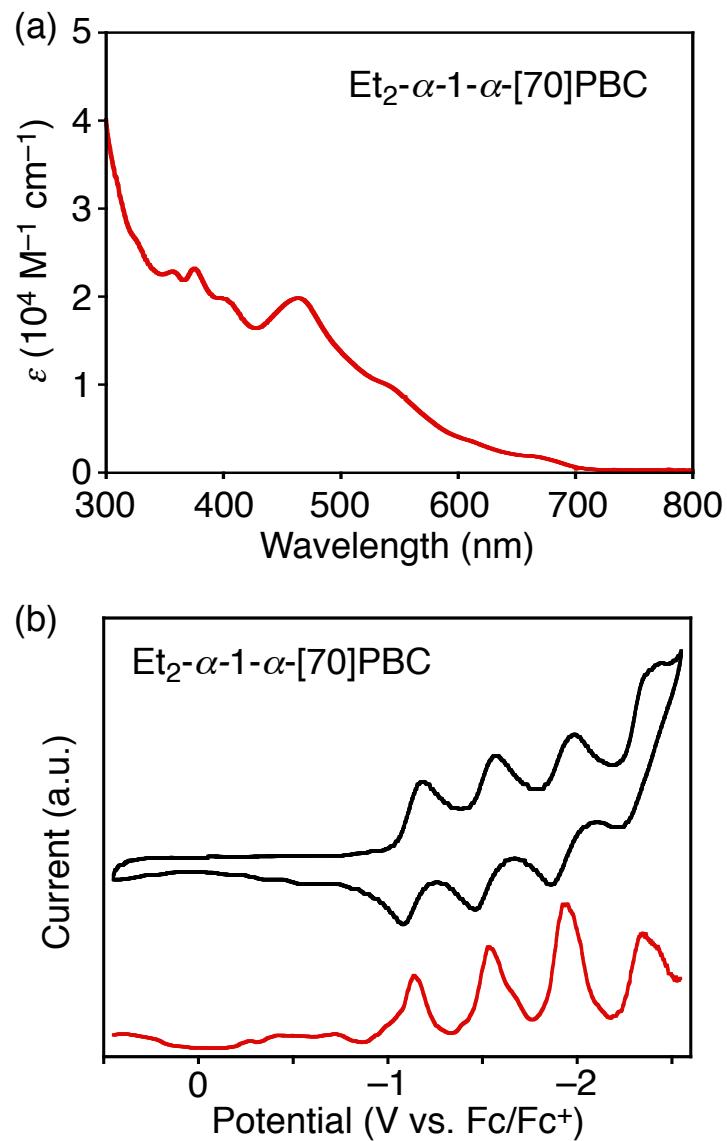


Fig. S12 (a) UV-vis absorption spectrum of $\text{Et}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$ in ODCB. (b) Cyclic voltammogram (upper) and differential pulse voltammogram (lower) of $\text{Et}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$ versus a reference electrode of Fc/Fc^+ measured in ODCB/acetonitrile mixture (v/v = 5:1) containing 0.1 M tetrabutylammonium hexafluorophosphate (Bu_4NPF_6). Sweep rate: 0.1 V s⁻¹; reference electrode, Ag/Ag^+ (0.01 M AgNO_3 , 0.09 M Bu_4NPF_6 in acetonitrile).

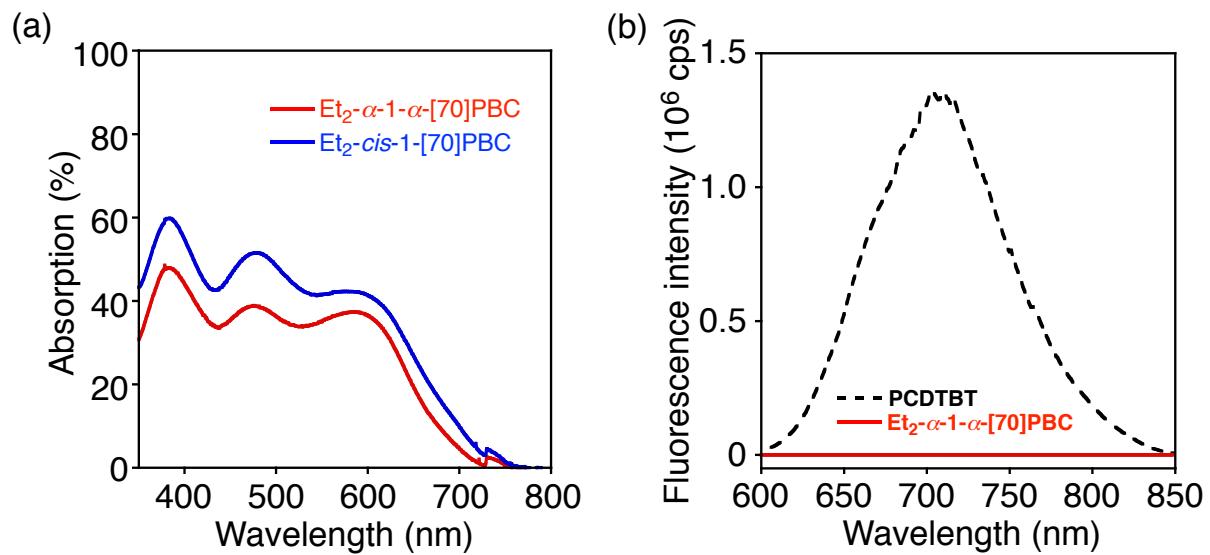


Fig. S13 (a) UV-visible absorption spectra of blend films of PCDTBT: $\text{Et}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$ (red) and PCDTBT: $\text{Et}_2\text{-}c\text{i}\text{s}\text{-}1\text{-}[70]\text{PBC}$ (blue). (b) Photoluminescence spectra of PCDTBT single component film (black dotted) and PCDTBT: $\text{Et}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$ (red). The excitation wavelengths are 540 nm. The emission intensities were normalized by that of the PCDTBT film considering the difference in the absorbances of the PCDTBT: $\text{Et}_2\text{-}\alpha\text{-}1\text{-}\alpha\text{-}[70]\text{PBC}$ film at the excitation wavelength.

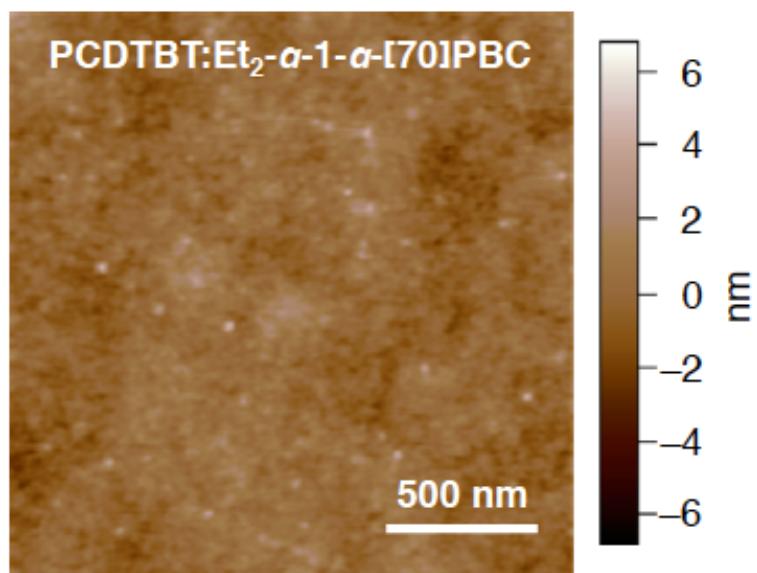


Fig. S14 Tapping-mode atomic force micrographs of PCDTBT:Et₂- α -1- α -[70]PBC on ITO/PEDOT:PSS substrates. The color scale represents the height topography, with bright and dark representing the highest and lowest features, respectively. The rms surface roughness is 0.78 nm.