

Supporting Information

Synthesis of silver sulfide nanoparticles and their photodetector applications

Myung Hyun Kang^a, Sung Ho Kim^a, Seunghun Jang^b, Ji Eun Lim^a, Hyunju Chang^b, Ki-Jeong Kong^b, Sung Myung^{*a}, and Joung Kyu Park^{*a}

^aAdvanced Materials Division, Korea Research Institute of Chemical Technology, Daejeon, Korea, E-mail: parkjk@kriect.re.kr

^bCenter for Molecular Modeling and Simulation, Korea Research Institute of Chemical Technology, Daejeon, Korea

#M. H. Kang, S. H. Kim, and S. Jang contributed equally to this work.

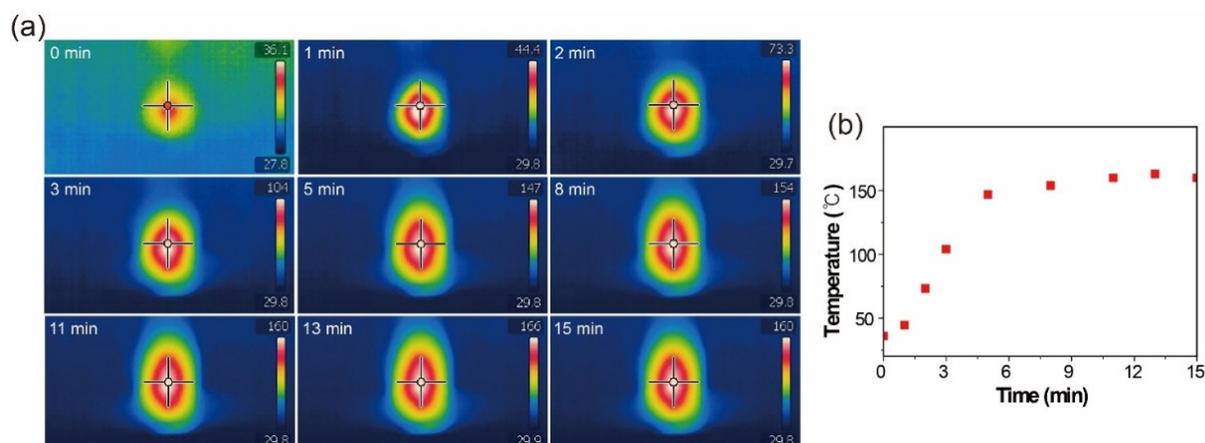


Fig. S1. Sonochemical synthesis of Ag_2S NPs. (a) Thermal images of the reaction vial taken by an infrared camera for the synthesis of Ag_2S NPs. The sonicator in the vial heated the reaction solution up to about $160\text{ }^\circ\text{C}$. The ultrasonic irradiation time usually takes $10 \sim 15$ minutes for the synthesis of Ag_2S NPs. (b) The temperatures of the local hot spot were plotted. The temperature may reach $160\text{ }^\circ\text{C}$ after 5 minutes of sonication.

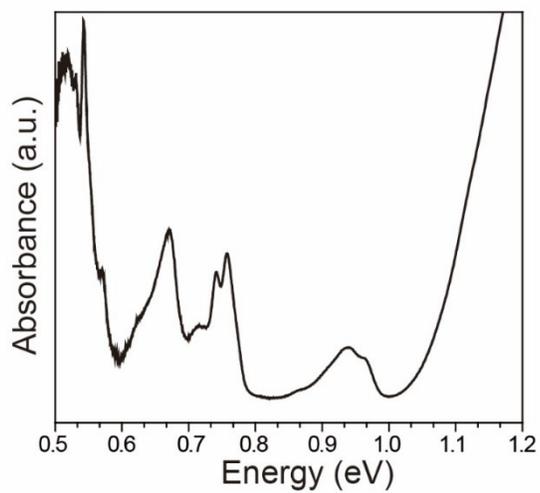


Fig. S2. Absorption spectrum of Ag₂S NPs. 0.01 g of Ag₂S NPs were dispersed in 10 ml Chloroform and measured using a SolidSpec-3700 UV-Vis-NIR spectrophotometer from Shimadzu.

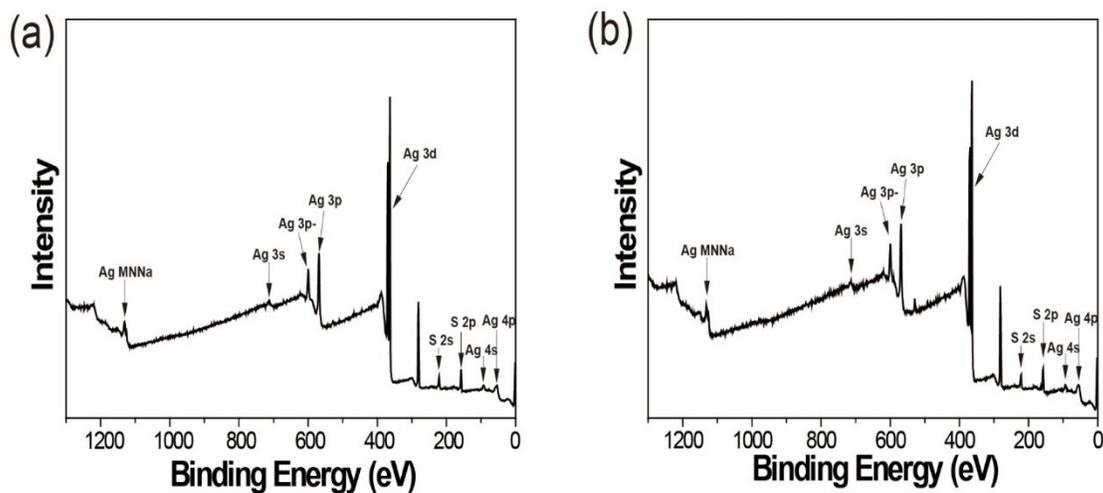


Fig. S3. XPS spectra of (a) Ag_2S and (b) Li-doped Ag_2S NPs.

	Ag (wt%)	Li (wt%)	S (wt%)
Ag_2S	77.9	---	11.7
Li-doped Ag_2S (0.05 mmol)	72.6	0.447	11.5

Fig. S4. Ag-S-Li ratios of the Ag_2S and Li(0.05mmol)-doped Ag_2S NPs by ICP-AES.

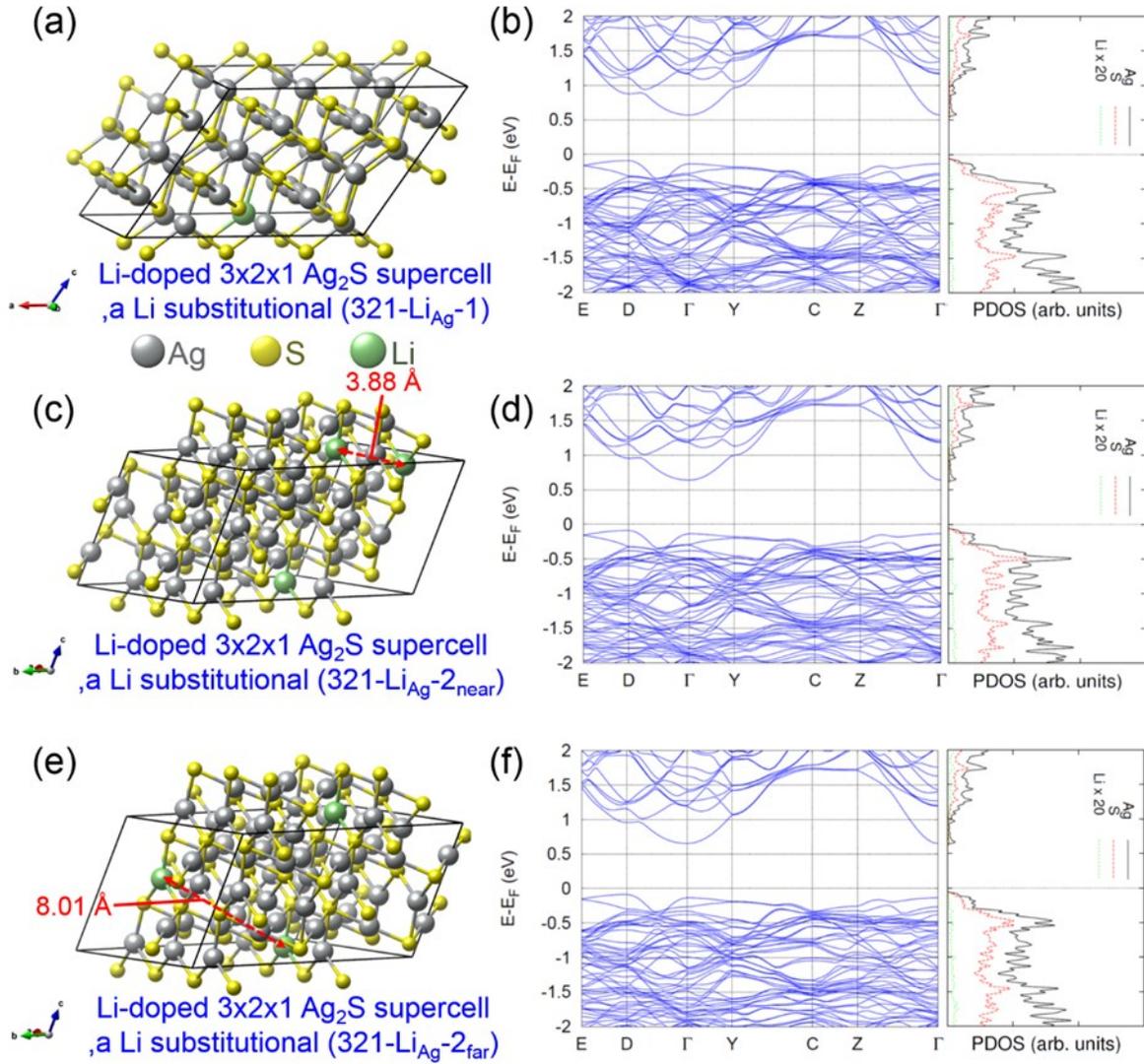


Fig. S5. Calculations for the Li substitutions at the Ag sites in the $3 \times 2 \times 1$ Ag_2S supercell. Calculated optimized structures for the (a) $321\text{-Li}_{\text{Ag}}\text{-1}$, (c) $321\text{-Li}_{\text{Ag}}\text{-2}_{\text{near}}$, and (e) $321\text{-Li}_{\text{Ag}}\text{-2}_{\text{far}}$. The gray, yellow, and green balls represent Ag, S, and Li atoms, respectively. Li-Li distances in $321\text{-Li}_{\text{Ag}}\text{-2}_{\text{near}}$ and $321\text{-Li}_{\text{Ag}}\text{-2}_{\text{far}}$ are 3.88 \AA and 8.01 \AA . Calculated band structures and PDOSs for the (b) $321\text{-Li}_{\text{Ag}}\text{-1}$, (d) $321\text{-Li}_{\text{Ag}}\text{-2}_{\text{near}}$, and (f) $321\text{-Li}_{\text{Ag}}\text{-2}_{\text{far}}$. The Fermi levels of all calculated systems were set to zero.

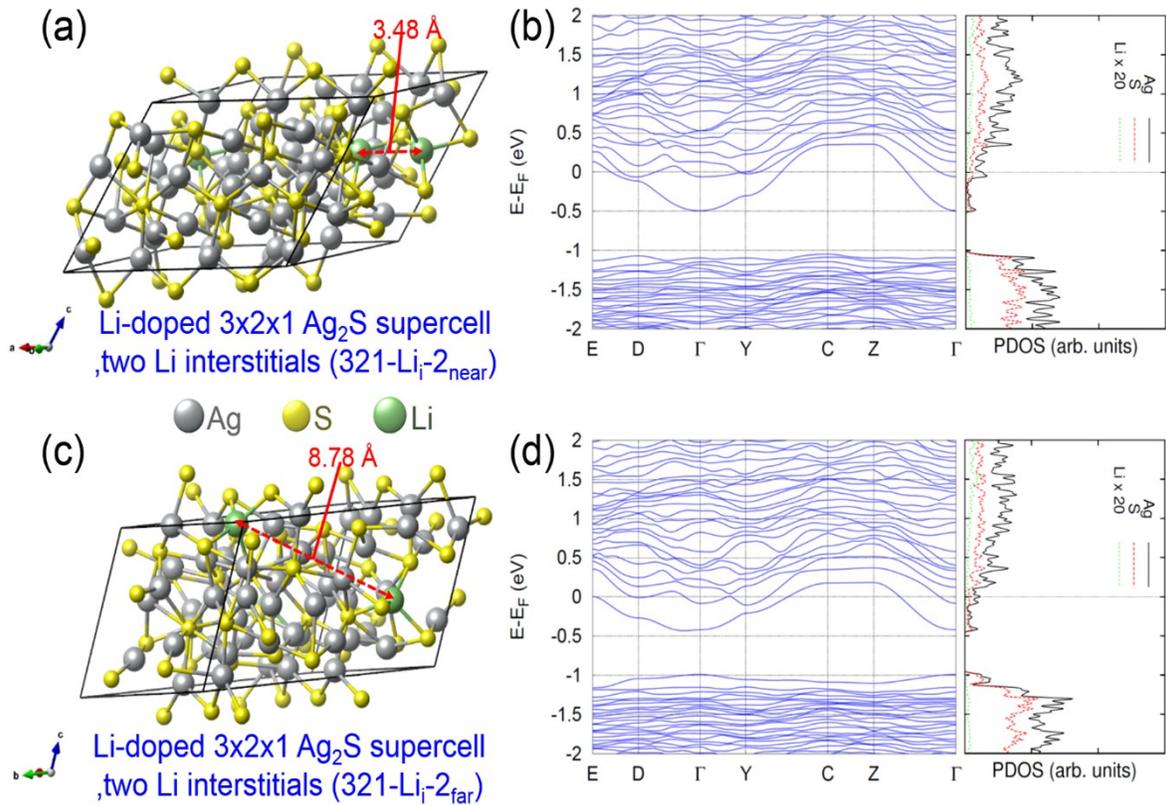


Fig. S6. Calculations for the Li interstitials in the $3 \times 2 \times 1$ Ag_2S supercell. Calculated optimized structures for the (a) $321\text{-Li}_i\text{-2}_{\text{near}}$, and (c) $321\text{-Li}_i\text{-2}_{\text{far}}$. The gray, yellow, and green balls represent Ag, S, and Li atoms, respectively. Li-Li distances in $321\text{-Li}_i\text{-2}_{\text{near}}$, and $321\text{-Li}_i\text{-2}_{\text{far}}$ are 3.48 \AA and 8.78 \AA . Calculated band structures and PDOSs for the (b) $321\text{-Li}_i\text{-2}_{\text{near}}$, and (d) $321\text{-Li}_i\text{-2}_{\text{far}}$. The Fermi levels of all calculated systems were set to zero.

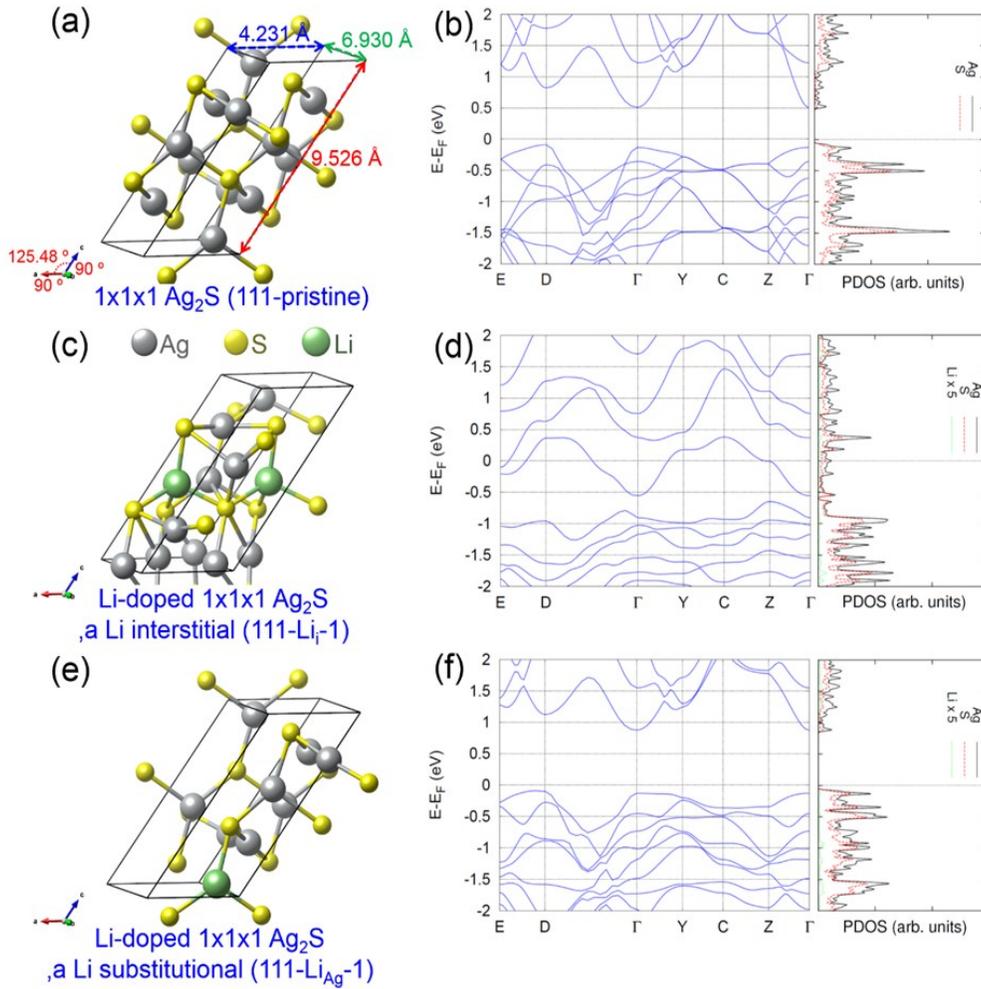


Fig. S7. Calculations for the pristine and Li doped $1 \times 1 \times 1$ Ag_2S . Calculated optimized structures for the (a) 111-pristine, (c) 111- Li_i-1 , and (e) 111- $\text{Li}_{\text{Ag}}-1$. The gray, yellow, and green balls represent Ag, S, and Li atoms, respectively. Calculated band structures and PDOSs for the (b) 111-pristine, (d) 111- Li_i-1 , and (f) 111- $\text{Li}_{\text{Ag}}-1$. The Fermi levels of all calculated systems were set to zero.

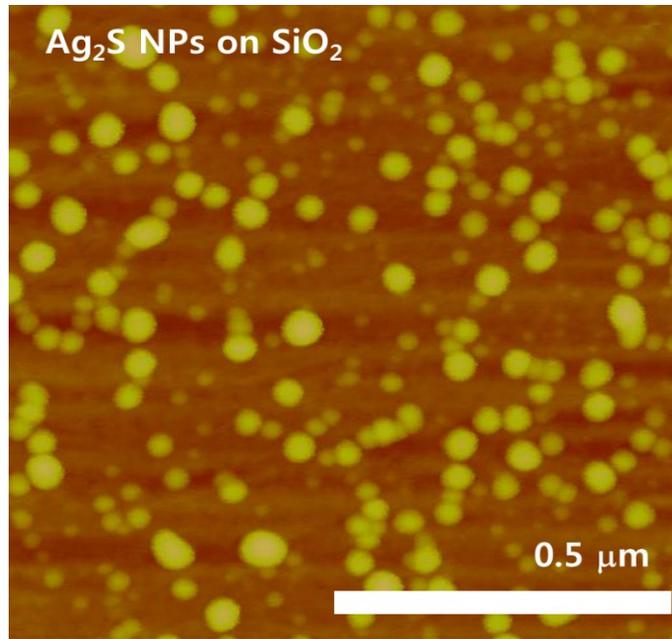


Fig. S8. Atomic force microscopy (AFM) image of Ag₂S NPs density on the SiO₂ substrate.

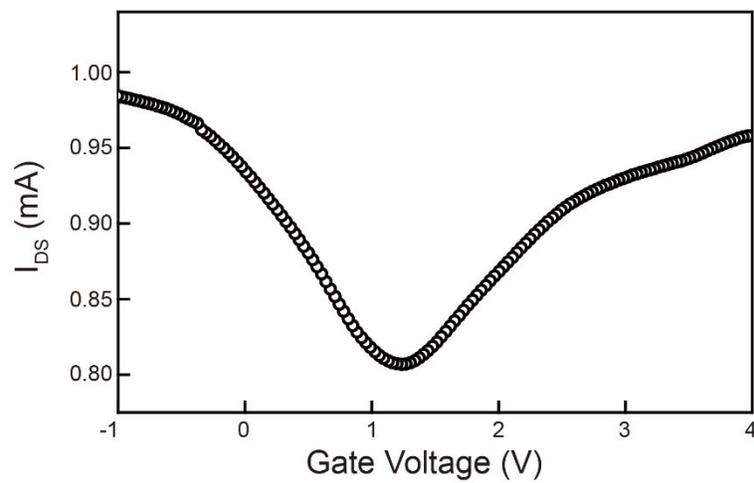


Fig. S9. The transfer curve (I_{DS} - V_G) of pristine graphene devices at $V_{DS} = 0.1$ V.

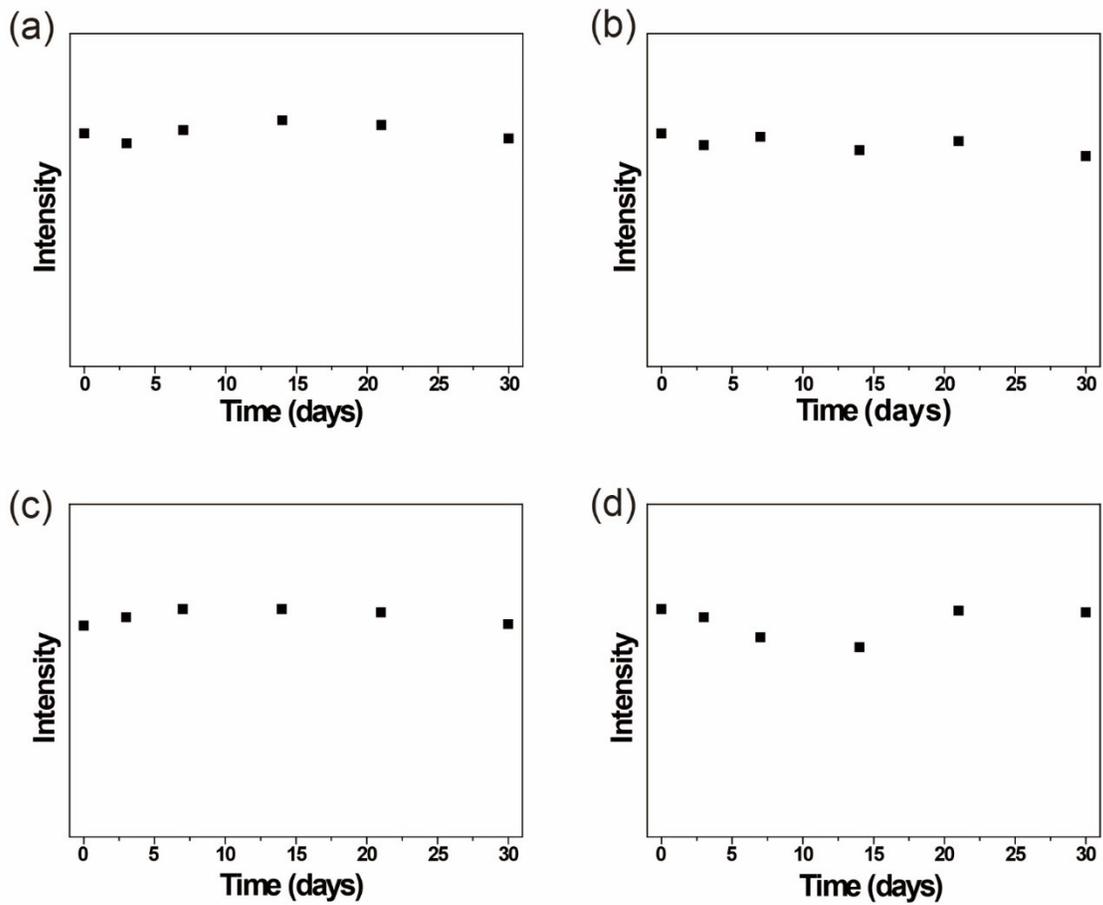


Fig. S10. PL emission intensities of Li-doped Ag₂S NPs (a) UV (365 nm) irradiated for 30 days, (b) RT after 30 days, (c) heat-treated to 70 °C for 30 days, and (d) 150 °C for 30 days.

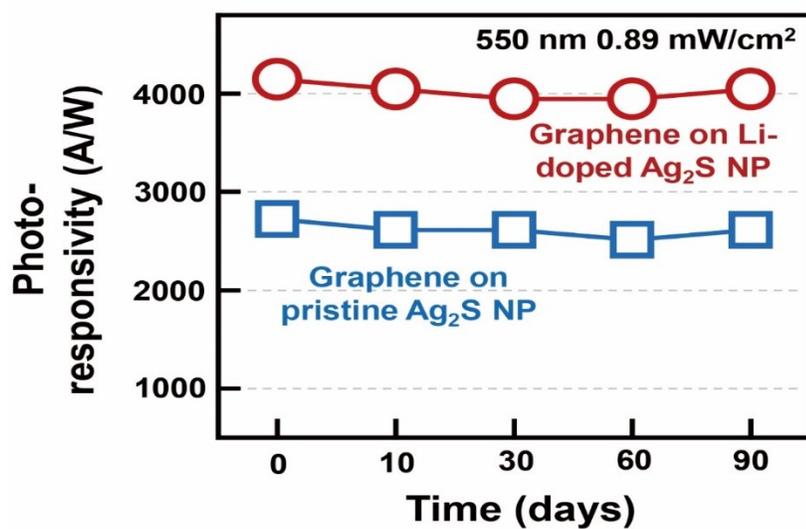


Fig. S11. Long term air-stability test of photodetectors based on graphene and Ag₂S NPs and Li-doped Ag₂S NPs.