Supplementary Information for

High-pressure synthesis and electrochemical properties of

tetragonal LiMnO₂

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Fig. S1 Cycle performances at 25 °C of the (a) o-LiMnO₂, (b) LMO(5GPa), (c) LMO(8GPa), and (d) LMO(12GPa) samples. Open and closed squares indicate charge and discharge capacities, respectively.



Fig. S2 Charge and discharge curves of the lithium cells with the (a) LMO(5GPa) and (b) LMO(8GPa) samples. The cells were operated at a temperature of 40 °C.

Phase	Space group	Atom	Wyckoff position	g^{*}	x	у	z^*	B^* / Å ²
o-LiMnO ₂	Pmmn	Li1	2b	0.942(1)	1/4	3/4	0.127(2)	1.32
		Mn1	2b	0.058(1)	1/4	3/4	0.127(2)	1.32
		Mn2	2b	0.914(1)	1/4	3/4	0.634(1)	0.55
		Li2	2b	0.086(1)	1/4	3/4	0.634(1)	0.55
		O1	2a	1	1/4	1/4	0.151(1)	0.58
		02	2a	1	1/4	1/4	0.608(1)	0.58
Composition : $\text{Li}_{1.028}\text{Mn}_{0.972}\text{O}_2$ ($\delta = 0.028$), $a_0 = 2.80970(8)$ Å, $b_0 = 4.5766(1)$ Å, and $c_0 = 5.7504(1)$ Å.								
t-LiMnO ₂	$I4_1/amd$	Li3	4 <i>a</i>	0.976(7)	0	3/4	1/8	1.5(7)
		Mn3	4a	0.024(7)	0	3/4	1/8	1.5(7)
		Mn4	4b	0.948(7)	0	1/4	3/8	0.27(6)
		Li4	4b	0.052(7)	0	1/4	3/8	0.26(6)
		O3	8 <i>e</i>	1	0	1/4	0.147(1)	1.1(2)
Composition	: Li _{1.028} Mn _{0.972}	$O_2 \left(\delta = 0.028\right)$), $a_t = 4.1844(1)$	Å, and $c_t = 8.22$	87(7) Å.			

Table S1 Structure parameters of the LMO(5GPa) sample determined by the Rietveld analyses.

Reliable factors : $R_{wp} = 7.646$ %, $R_p = 5.273$ %, and S = 0.939.

 $Mass\ fractions: 86.6\ wt\%\ for\ o-LiMnO_2, 8.6\ wt\%\ for\ t-LiMnO_2, 4.0\ wt\%\ for\ Mn_3O_4, and\ 0.8\ wt\%\ for\ LiMn_2O_4.$

*Constraints : $g(Mn1) = 1 - g(Li1), g(Mn2) = g(Li1) - \delta, g(Li2) = 1 + \delta - g(Li1), z(Mn1) = z(Li1), z(Li2) = z(Mn2), g(Mn3) = g(Li3), g(Mn4) = g(Li3) - \delta, g(Li4) = 1 + \delta - g(Li3), B(Mn3) = B(Li3), and B(Li4) = B(Mn4).$

Table S2 Structure parameters of the LMO(8GPa) sample determined by the Rietveld analyses.

Phase	Space group	Atom	Wyckoff position	g^{*}	x	у	Ζ	B^* / Å ²
t-LiMnO ₂	$I4_1/amd$	Li1	4a	0.991(1)	0	3/4	1/8	1.44(8)
		Mn1	4a	0.009(1)	0	3/4	1/8	1.44(8)
		Mn2	4b	0.977(1)	0	1/4	3/8	0.28(1)
		Li2	4b	0.023(1)	0	1/4	3/8	0.28(1)
		O1	8 <i>e</i>	1	0	1/4	0.141(1)	1.09(2)
Composition	: Li _{1.014} Mn _{0.986}	$O_2 (\delta = 0.014)$), $a_t = 4.18259(2$) Å, and $c_{\rm t} = 8.2$	3443(6) Å.			

Reliable factors : $R_{wp} = 6.076$ %, $R_p = 4.379$ %, and S = 0.6781.

Mass fractions : 96.4 wt% for t-LiMnO₂, 1.2 wt% for Mn_3O_4 , and 2.3 wt% for LiMn₂O₄.

*Constraints : g(Mn1) = 1 - g(Li1), $g(Mn2) = g(Li1) - \delta$, $g(Li2) = 1 + \delta - g(Li1)$, B(Mn1) = B(Li1), and B(Li2) = B(Mn2).



Fig. S3 *Ex situ* Raman spectrum of the charged LMO(12GPa) sample. The inset indicates its charge curve up to 4.8 V. The Raman band at 654 cm⁻¹ is originated from $Li_xMn_2O_4$ spinel phase.



Fig. S4 *Ex situ* Raman spectra of the cycled (a) LMO(5GPa) and (b) LMO(8GPa) samples. Raman spectra were taken at the discharged state indicated by the red arrows in Figs. 5b and 5c in the text.



Fig. S5 Raman spectra for (a) PTFE and (b) AB.