Electronic Supplementary Information

"Exploring weak intermolecular interactions in thiocyanate-bonded Zn(II) and Cd(II) complexes with methylimidazole: Crystal structures, Hirshfeld surface analysis and luminescence properties"

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Figure S1, ESI. A view of intermolecular C-H...S hydrogen bonds (dashed lines) forming $R_2^2(14)$ dimer sited at a crystallographic inversion centre of compound (1) solid. Hydrogen atoms not involved in the hydrogen bonding are omitted for the sake of clarity.



Figure S2, ESI. A view of intermolecular N-H...S hydrogen bonds (dashed lines) forming $R_4^3(26)$ ring motifs for compound (2). Hydrogen atoms not involved in the hydrogen bonding are omitted for the sake of clarity.



Figure S3, ESI. A view of intermolecular C-H...S hydrogen bonds (dashed lines) forming $R_4^2(20)$ ring motifs for compound (3). Hydrogen atoms not involved in the hydrogen bonding are omitted for the sake of clarity.



Figure S4, ESI. A view of 3D network showing intermolecular C-H...S hydrogen bonds (dashed lines) for polymeric (4). Hydrogen atoms not involved in the hydrogen bonding are omitted for the sake of clarity.



Figure S5, ESI. TG and DTA curves for the thermal decomposition of the complex [Zn(1-MeIm)₂(SCN)₂] (1) at 5 °C/min in air.



Figure S6, ESI. TG and DTA curves for the thermal decomposition of the complex $[Zn(2-MeIm)_2(SCN)_2]$ (2) at 5 °C/min in air.



Figure S7, ESI. TG and DTA curves for the thermal decomposition of the complex $[Cd(1-MeIm)_4(SCN)_2]$ (3) at 5 °C/min in air.



Figure S8, ESI. TG and DTA curves for the thermal decomposition of the complex $[Cd(2-MeIm)_2(SCN)_2]_n(4)$ at 5 °C/min in air.



Figure S9, ESI. Electronic spectra of the ligands and complexes 1-4 in acetonitrile solutions (10^{-3} M) .



Figure S10, ESI. Decay of the photoluminescence intensity integrated over the whole spectral range of 1-MeIm, 2-MeIm and complexes 1-4.



Table S1, ESI[†]. Hirshfeld contact surfaces $C_{XY}(\%)^*$, proportion of chemical type on the molecular surface Sx (%) and random contacts R_{XY} (%) of the main intermolecular interactions for compounds 1-7.

Contact C _{XY}	1	2	3	4	5	6	7
Н∙ ∙∙Н	23.3	24.5	41.9	21.4	41.2	37.4	43.5
С∙ ••Н	20.7	22.2	22.8	21.2	11.0	10.9	9.7
N· ··H	13.9	12.4	11.1	16.6	7.3	9.7	8.0
S· ··H	33.7	30.1	21.6	31.2	-	-	-
O∙ ••H	-	-	-	-	-	4.6	-
Cl· ··H	-	-	-	-	33.9	-	-
Br∙ ••H	-	-	-	-	-	32.3	26.0
$C \cdot \cdot \cdot C$	-	0.5	1.0	0.0	1.3	1.6	2.0
$C \cdot \cdot \cdot S$	-	3.9	-	4.3	-	-	-
$N \cdot \cdot \cdot S$	-	3.3	-	-	-	-	-
Cl···C	-	-	-	-	1.6	-	-
N· ··C	-	-	-	-	-	-	1.6
Surface Sx							
Н	57.6	56.9	69.7	55.9	67.9	66.3	65.4
С	14.4	14.0	12.7	13.4	8.2	7.9	7.6
Ν	9.0	9.8	6.3	10.2	5.0	6.5	5.5
S	19.0	19.0	11.3	19.3	-	-	-
0	-	-	-	-	-	2.4	-
Cl	-	-	-	-	18.3	-	-
Br	-	-	-	-	-	16.7	17.7
RandomcontactsR _{XY}							
Н∙ ∙∙Н	33.2	32.4	48.5	31.3	46.0	44.0	42.8
С∙ ••Н	16.5	16.0	17.7	15.0	11.0	10.5	9.9
N· ··H	10.4	11.2	8.7	11.4	7.0	8.6	7.2
S· ··H	21.9	21.6	15.7	21.5	-	-	-
O∙ ••H	-	-	-	-	-	3.2	-
Cl· ··H	-	-	-	-	24.8	-	-
Br∙ ••H	-	-	-	-	-	22.1	23.2
$C \cdot \cdot \cdot C$	2.1	2.0	1.6	1.8	0.7	0.6	0.6
C····S	5.5	5.3	-	5.2	-	-	-
$N \cdot \cdot \cdot S$	-	3.7	-	-	-	-	-
Cl···C	-	-	-	-	3.0	-	-
N····C	2.6	-	-	-	-	-	0.8

*Data obtained from CrystalExplorer3.0, including reciprocal contacts.

2-methylimidazole		$[Cd(2-MeIm)_2(SCN)_2]_n$		[Zn(2-MeIm) ₂ (SCN) ₂]		Assignment
FT-IR	Raman	FT-IR	Raman	FT-IR	Raman	
3182 (s)	-	3224 (m)	-	3319 (vs)	-	v NH
3137 (s)	3134 (52)	3147 (s)	3147(18)	3160 (w)	3160 (26)	νCH
2961 (s)	2960 (23)	2954 (vvw)	-	2967 (vvw)	-	va CH ₃
2926 (s)	2925 (53)	2933 (vvw)	2934 (29)	2934 (vw)	2934 (36)	vs CH ₃
-	-	2119 (vs)	2118 (67)	2100 (vs)	2102 (100)	v CN
-	-	2076 (vs)	2077 (100)	2077 (vs)	2077 (51)	v CN
1597 (s)	-	1566 (s)	-	1565 (vs)	-	v C-C + v C-N
1479 (vvw)	1479 (100)	1495 (vw)	1497 (49)	1500 (m)	1502 (100)	$v C-C + \delta a CH_3$
1447 (s)	-	1428 (w)	-	1430 (m)	-	δ a CH ₃
1371 (w)	-	1352 (w)	-	1351 (m)	1353 (16)	δsCH ₃
1303 (m)	1303 (4)	1280 (s)	-	1279 (w)	1279 (17)	
1155 (m)	-	1159 (w)	-	1157 (m)		v ring
1117 (s)	1113 (79)	1098 (w)	-	1140 (s)	1140 (55)	δССН
1048	-	1038 (w)	-	1046 (w)	-	ρCH ₃
995	993 (15)	1012 (w)	-	1023 (vw)	-	$\rho CH_3 + \nu C-N (ring)$
945 (s)	937 (9)	943 (vw)	-	930 (w)	929 (13)	$\rho CH_3 + \delta CCH$
916 (s)	916 (10)	901 (vvw)	-	-	-	δring
-	-	770 (w)	857 (25)	757 (w)	-	v CS
-	-	765 (vvw)	-	747 (w)	-	v CS
683 (s)	682 (39)	672 (m)	674 (12)	666(vw)	684 (9)	ү СН
629	629 (2)	624 (vvw)	-	622 (vw)	620 (6)	γΝΗ
-	-	469 (w)	-	478 (s)	481 (5)	δSCN
-	-	451 (vw)	-	470 (m)	-	δSCN
-	269 (25)	-	283 (13)	-	284 (15)	G CNCN
-	-	-	-	-	254 (9)	v Zn-N(ring)
-	-	-	-	-	212 (17)	v Zn-N (SCN)
-	-	-	208 (25)	-	-	v Cd-N (2Au)
-	-	-	185 (12)	-	-	v Cd-N (NCS)
-	-	-	174 (44)	-	-	v Cd-N (ring)
-	-	-	145 (49)	-	-	v Cd-S (B1g)
-	-	-	112 (94)	-	-	v Cd-N(ring)
-	-	-	107 (96)	-	-	v Cd-N(SCN)
-	-	-	-	-	82 (99)	ρSCN
			75 (99)		64 (99)	τCH ₃

Table S2: IR and Raman bands (in cm⁻¹) for 2-methylimidazole and its Cd(II) and Zn(II) thiocyanate complexes together with their tentative assignment of modes.

1-Methylimidazole		[Cd(1-MeIm) ₄ (SCN) ₂]		[Zn(1-M	[eIm) ₂ (SCN) ₂]	Assignment
FT-IR	Raman	FT-IR	Raman	FT-IR	Raman	
-	3134 (39)	3145 (vw)	3146 (17)	3142	3143 (4)	νСΗ
3127 (vvw)	-	3125(w)	3127 (13)	3120	3124 (3)	νСΗ
3111(vs)	3109 (35)	3111 (m)	3117 (8)	-	-	νCH
2983	-	2984 (vw)	-	-	-	va CH ₃
2952 (m)	2955 (36)	2958 (s)	2959 (10)	2967	2959 (8)	va CH ₃
2812 (w)	2816 (11)	-	2815 (2)	-	-	vs CH ₃
-	-	2083 (vs)	-	2098	2086 (100)	ν C-N
-	-	2066 (s)	2074 (100)	2083	2070 (93)	ν C-N
1669	-	1671 (vw)	-	-	-	v C-N + δa CH ₃
1619	-	1621 (vw)	-	-	-	$v C-C + \delta a CH_3$
1522 (vs)	1519 (15)	1534 (m)	1533 (3)	1545	1541 (21)	δa CH ₃
-	-	1516 (m)	1518 (3)	1525	1522 (30)	δa CH ₃
1422	1422 (9)	1421 (vw)	1419 (4)	1421	1414 (20)	v C-N + δa CH ₃
-	-	1413 (vw)	-	-	-	v C-N + δa CH ₃
1363(m)	-	1369 (vs)	1369 (9)	1372	1370 (50)	$v \operatorname{ring} + \delta a \operatorname{CH}_3$
1329	1328 (20)	1339 (vvw)	1335 (18)	1338	1336 (66)	$v \operatorname{ring} + \delta a \operatorname{CH}_3$
1288	1288 (21)	1282 (m)	1283 (5)	1286	1284 (5)	v ring
1237	1238 (9)	1237 (s)	1239 (2)	1240	1242 (10)	ν C-N + δ HCN
1026 (m)	1032 (33)	1027 (vw)	1030 (2)	1047	1041 (3)	ρCH ₃
-	-	1023 (vw)	-	1025	1024 (66)	ρCH ₃
925 (vs)	-	937 (s)	937 (4)	956	954 (33)	ν C-N + δ CNC
836 (vs)	-	838 (s)	833 (2)	828	837 (77)	үС-Н
-	-	769 (m)	-	768	760 (7)	vC-S
		764 (m)	-	753	-	vC-S
661	-	670 (w)	669 (12)	673	670 (65)	үС-Н
620 (vs)	619 (3)	619 (s)	622 (2)	619	618 (16)	үС-Н
		473 (w)	471 (1)	478	477 (6)	δSCN
-	-	464 (w)	-	473 sh	-	δSCN
-	355 (7)	-	368 (3)	-	366 (21)	δN-CH ₃
-	-	-	-	-	246 (32)	v Zn-N(ring)
			-	-	214 (31)	v Zn-N (SCN)
-	-	-	202 (19)	-	-	vCd-N (SCN)
-	-	-	142 (40)	-	-	vCd-N (ring)
-	-	-	87 (56)	-	82 (58)	ρSCN
-	-	-	65 (33)	-	-	τCH ₃

Table S3: IR and Raman bands (in cm⁻¹) for 1-methylimidazole Zn(II) and Cd(II) thiocyanate complex together with their tentative assignment of modes.

Complex	Steps	Temperature range (°C)	Theoretical mass loss (%)	Observed mass loss (%)
1	1	200-390	47.5	47.0
	2	500-750	28.9	29.0
	Total	-	76.4	76.0
2	1	250-375	47.5	48.0
	2	500-700	28.9	28.0
	Total	-	76.4	76.0
3	1	150-350	58.0	56.4
	2	500-750	19.0	23.6
	Total	-	77.0	80.0
4	1	200-390	42.0	43.0
	2	500-750	25.0	22.0
	Total	-	67.0	65.0

 Table S4: Steps for the thermal decomposition of complexes 1-4 and % of mass loss.