

## Supporting Information for

### **Annealing Temperature Effects on Photoelectrochemical Performance of Bismuth Vanadate Thin Film Photoelectrodes**

*Le Shi<sup>a</sup>, Sifei Zhuo<sup>a</sup>, Mutalifu Abulikemu<sup>a</sup>, Gangaiah Mettela<sup>a</sup>, Thangavelu*

*Palaniselvam<sup>a</sup>, Shahid Rasul<sup>a</sup>, Bo Tang<sup>a</sup>, Buyi Yan<sup>b</sup>, Navid B. Saleh<sup>c</sup> and Peng Wang<sup>\*a</sup>*

<sup>a</sup>L. Shi, Dr. S.F. Zhuo, Dr. M. Abulikemu, Dr. G. Mettela, Dr. T. Palaniselvam, Dr. S. Rasul, Dr. B. Tang, Prof. P. Wang

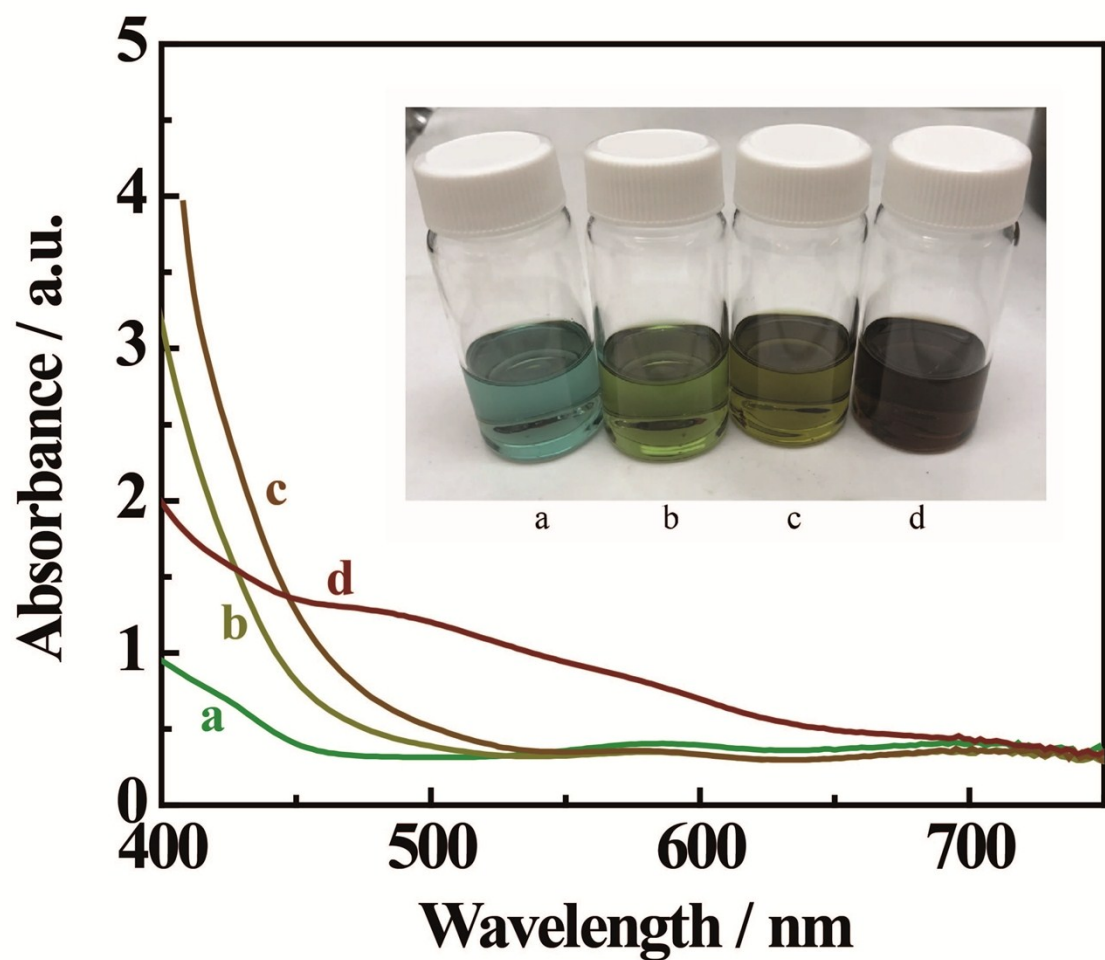
Water Desalination and Reuse Center, Division of Biological and Environmental Sciences and Engineering, King Abdullah University of Science and Technology, Thuwal 23955-6900, Saudi Arabia. E-mail: peng.wang@kaust.edu.sa

<sup>b</sup>B. Y. Yan, Division of Physical Science and Engineering, King Abdullah University of Science and Technology, Thuwal 23955-6900, Saudi Arabia.

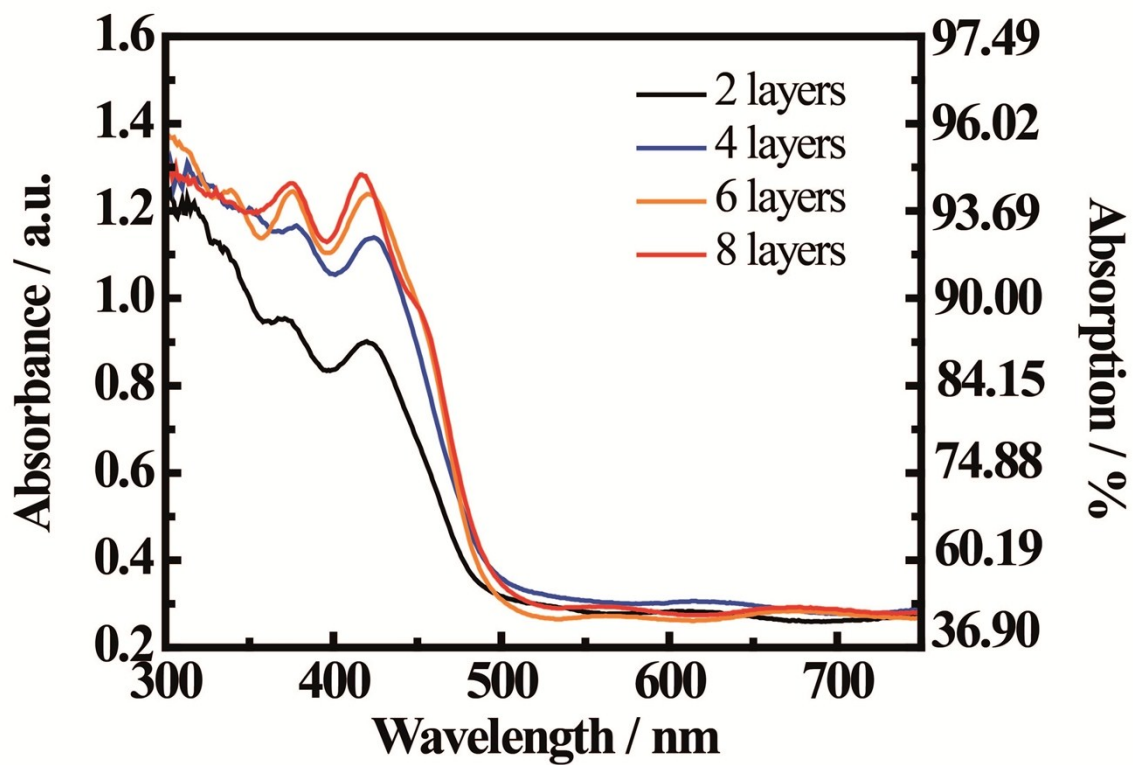
<sup>c</sup>Prof. N. B. Saleh, Department of Civil, Architectural and Environmental Engineering, University of Texas at Austin, Austin, TX 78712, USA.

**Table S1.** Summary of photocurrent densities of pristine BiVO<sub>4</sub> photoelectrodes (in the absence of dopant and cocatalyst) prepared by spin-coating method. Note: backside illumination was used unless otherwise noted.

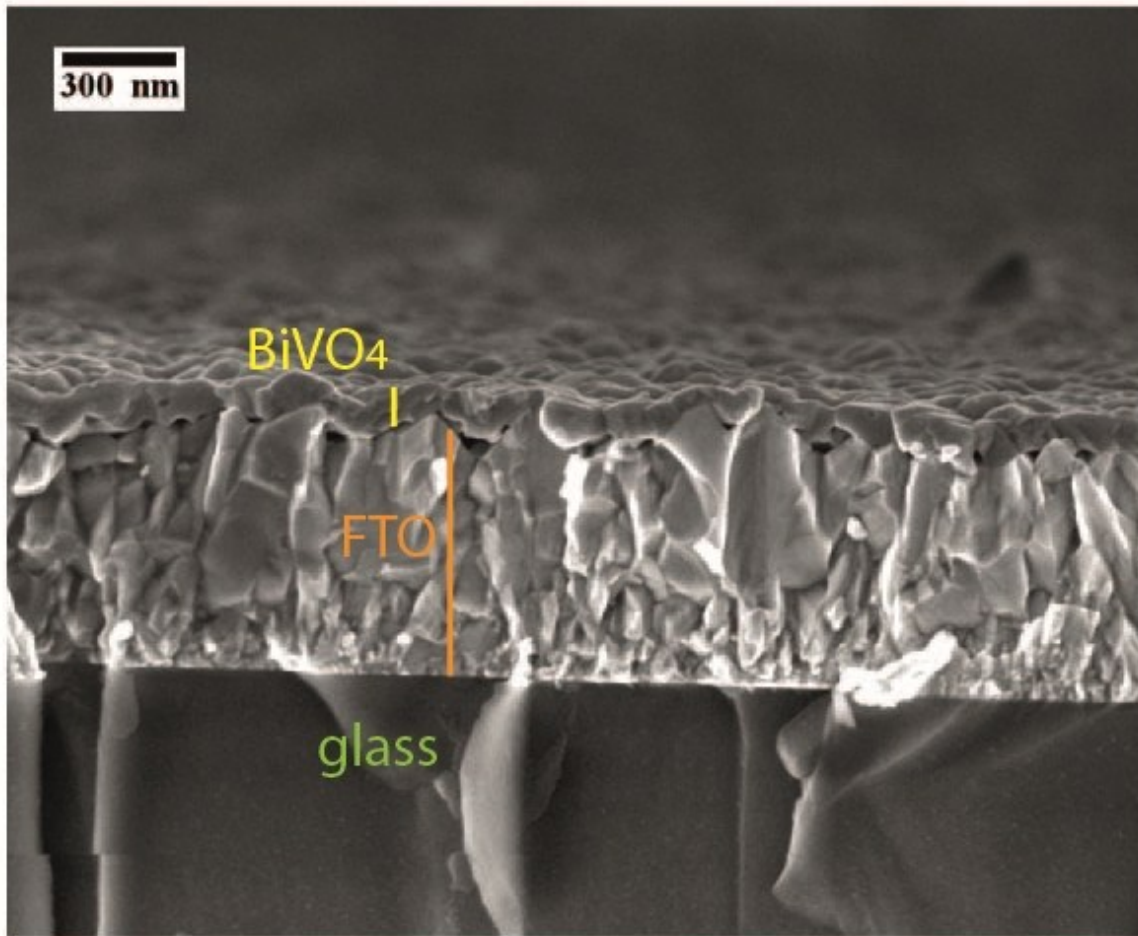
Reference #	Temperature (°C)			Duration time			Film thickness	Light source	Illuminated area (cm <sup>2</sup> )	Photocurrent density (mA/cm <sup>2</sup> at 1.23 V vs. RHE)	Aqueous solution
	dry	Inter-mediate	final	dry (min)	Inter-mediate (min)	final (hour)					
1	150	-	550	5	-	1	5 layers	>420nm	-	0.02 at 1.0 V vs. Ag/AgCl	0.5 M Na <sub>2</sub> SO <sub>4</sub>
2	-	500	-	-	30	-	6 layers	AM 1.5G	0.28	0.2	1.0 M Na <sub>2</sub> SO <sub>4</sub>
3	-	500	500	-	10	2	9 layers	AM 1.5G	0.635	0.25	0.1 M KPi, pH 7.0
4	RT	500	500	-	10	2	8 layers	AM 1.5G	-	0.3	0.1 M KBi, pH 9.2
5	-	450	450	-	20	3	180 nm	AM 1.5G	-	0.76	0.1 M KPi, pH 7.0
6	-	350	460	-	10	2	80 nm	AM 1.5G	-	0.12	0.5 M K <sub>2</sub> SO <sub>4</sub> pH 5.9
7	-	440	470	-	60	5	1 um	AM 1.5G	1.0	0.42	0.5 M Na <sub>2</sub> SO <sub>4</sub> pH 6.6
8	RT	500	-	-	30	-	6 layers	0.9 Wcm <sup>-2</sup>	-	0.17	0.5 M Na <sub>2</sub> SO <sub>4</sub> pH 5.44
9	-	500	500	-	10	8	16 layers	AM 1.5G	-	1.0 at 1.52 V vs. RHE	KPi pH 7
10	-	-	500	-	-	2	7 layers	AM 1.5G	Front side	1.0	0.1 M Na <sub>2</sub> SO <sub>4</sub>
11	-	500	500	-	10	2	4 layers	AM 1.5G	-	0.75	0.1M KPi pH 7
12	-	450	450	-	30	10	15 layers	AM 1.5G	-	0.86	0.1M KPi pH 6.7
This work	100	500	500	2	5	0.6	6 layers	AM 1.5G	1.0 cm	1.23	0.1M KPi pH 7



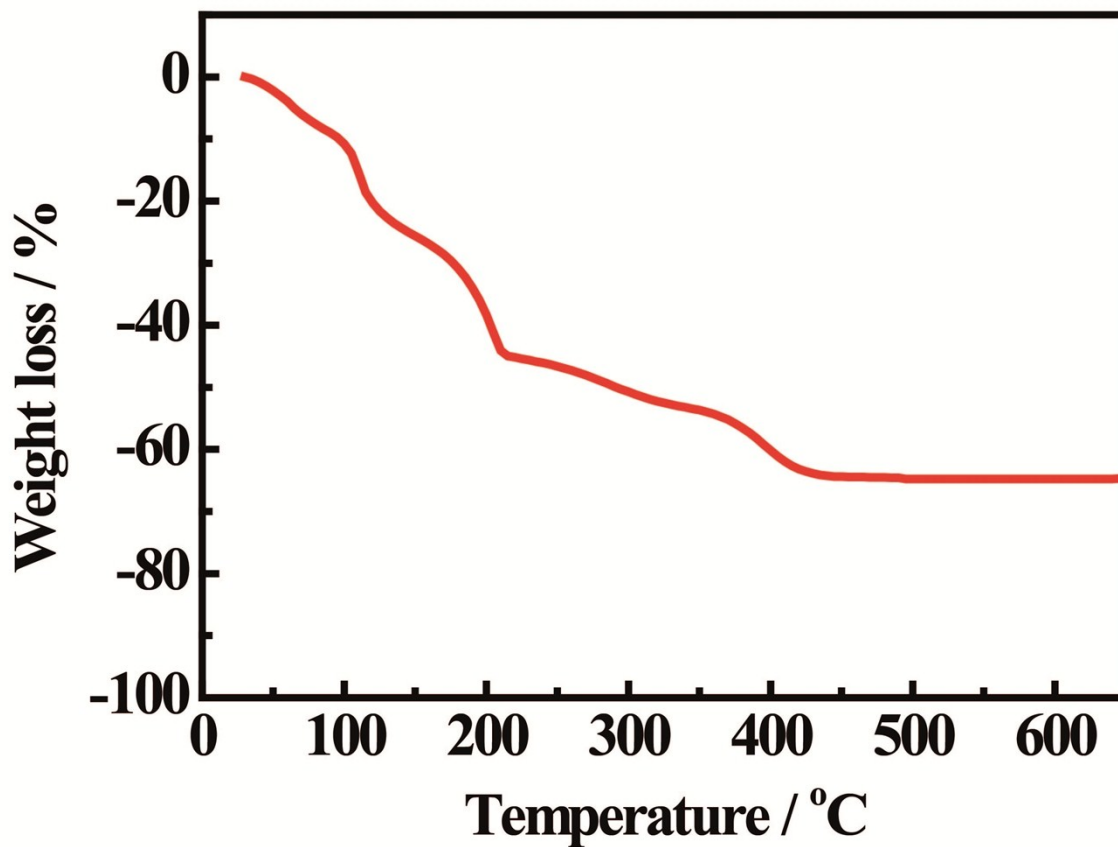
**Figure S1.** UV-vis absorption spectra of BiVO<sub>4</sub> precursor and fresh VO(acac)<sub>2</sub> solution. All the solutions are diluted in acetylacetone (1:10). Inset is the photos of (a) fresh VO(acac)<sub>2</sub> solution (0.22 g of VO(acac)<sub>2</sub> in 10 ml acetylacetone) and BiVO<sub>4</sub> precursor (0.4 g of Bi(NO<sub>3</sub>)<sub>3</sub>·5H<sub>2</sub>O) and 0.22 g of VO(acac)<sub>2</sub> in 10 ml acetylacetone): (b) freshly prepared, (c) after continuously stirring for 10 days, (d) sonication with gentle heating for 10 hours.



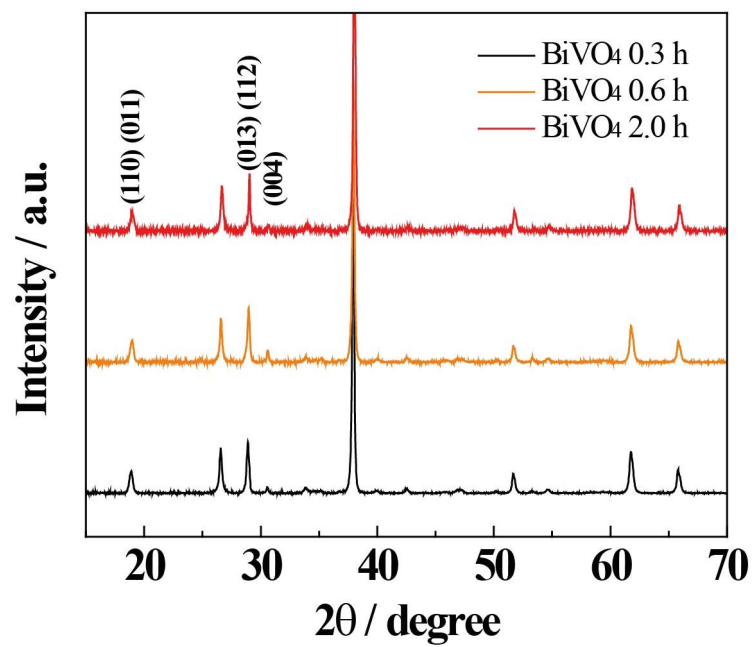
**Figure S2.** UV-vis absorption spectra of BiVO<sub>4</sub> photoelectrode annealed at 500°C with different coating layers.



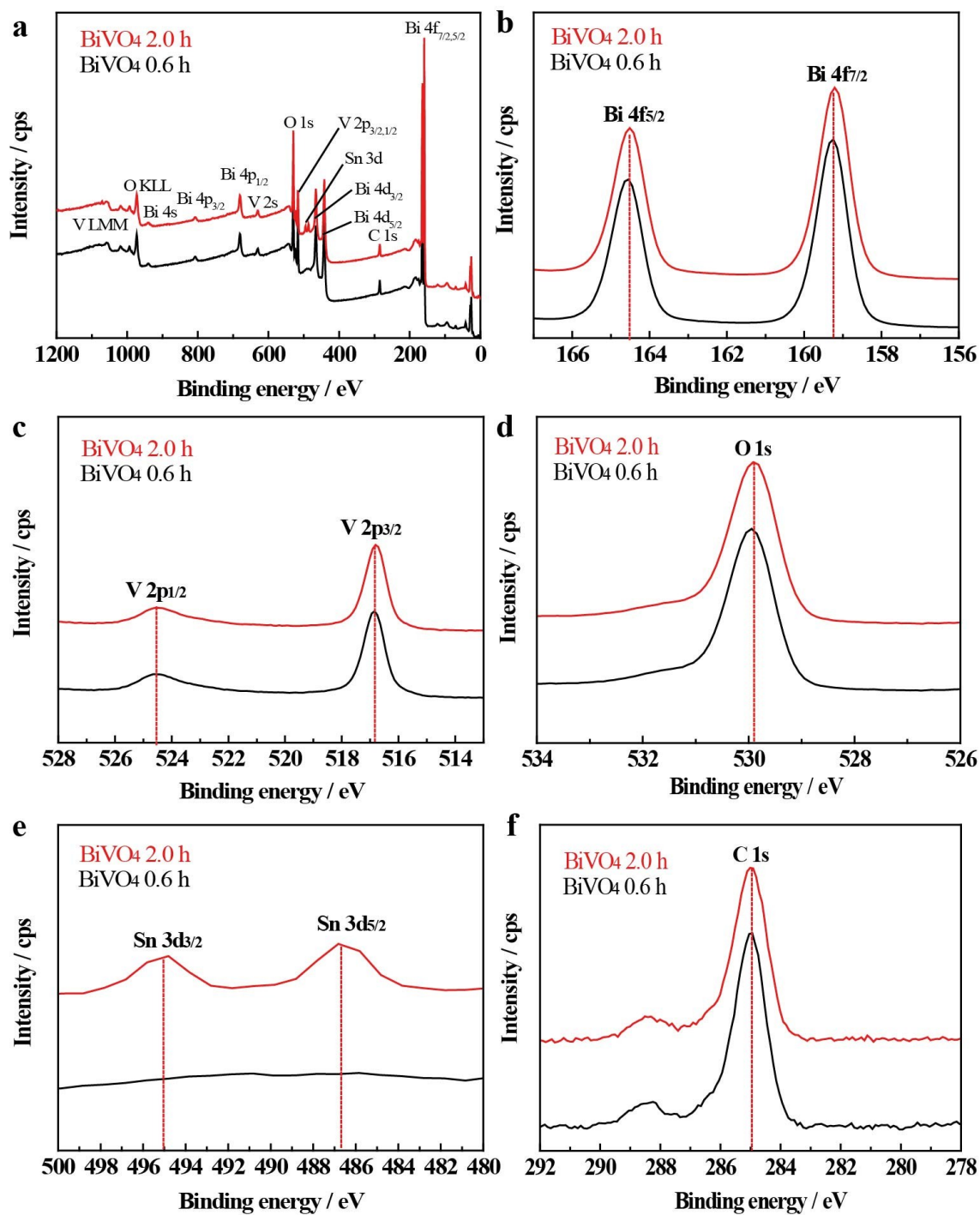
**Figure S3.** SEM image of cross-section of BiVO<sub>4</sub>-500. The top layer of BiVO<sub>4</sub> thin film is ~93 nm thick.



**Figure S4.** Thermal gravimetric analysis (TGA) of BiVO<sub>4</sub> precursor solution. The weight loss was monitored up to 600 °C with a temperature step of 5 °C min<sup>-1</sup>.



**Figure S5.** XRD spectra of BiVO<sub>4</sub>-540 with different annealing duration time.



**Figure S6.** XPS spectra of BiVO<sub>4</sub>-540 with different annealing duration time: (a) full range survey scan of sample BiVO<sub>4</sub>-540, (b) individual spectra of Bi 4f, (c) V 2p spectra (d) O 1s spectra, (e) Sn 3d spectra and (f) C 1s spectra.



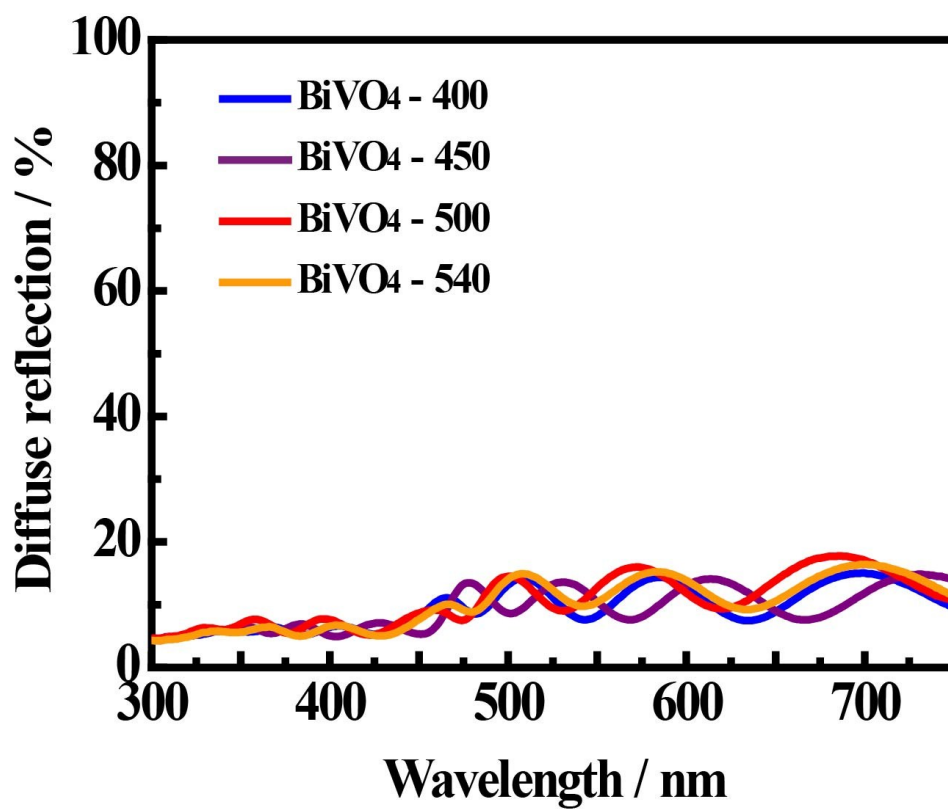
XPS analysis was carried out to gain additional insights (Figure S6). The wide scan spectra of the sample in Figure S6a indicate the presence of element Bi, V and O. The binding energies of Bi 4f<sub>7/2</sub> (159.2 eV), V 2p<sub>3/2</sub> (516.8 eV) and O 1s (529.9 eV) in BiVO<sub>4</sub> are well in accordance with reported BiVO<sub>4</sub> literature.<sup>13</sup> The splitting peaks of Bi 4f<sub>7/2</sub> (159.2 eV) and Bi 4f<sub>5/2</sub> (164.5 eV) are shown in Figure S6b, which are characteristics of Bi<sup>3+</sup>. At the same time, the splitting peaks of V 2p are observed at 524.5 (V 2p<sub>1/2</sub>) and 516.8 eV (V 2p<sub>3/2</sub>), that occur due to the surface V<sup>5+</sup> species (shown in Figure S6c). For comparison, BiVO<sub>4</sub>-500 annealed at 2h is also presented in Figure S6. The additional peaks of Sn 3d<sub>3/2</sub> (495.0 eV) and Sn 3d<sub>5/2</sub> (486.7 eV) are observed, due to the Sn diffusion from FTO substrate during the longer period of annealing. In order to remove the interference from other affects rather than the annealing temperature affect on the BiVO<sub>4</sub> material, the 0.6 h annealing duration is chosen based on such Sn diffusion, which has been reported to have uncertain effects on the photocurrent density.<sup>14-17</sup>

**Table S2.** Grain size information for BiVO<sub>4</sub>-n photoelectrodes; i.e., range, average grain size, and coefficient of variation.

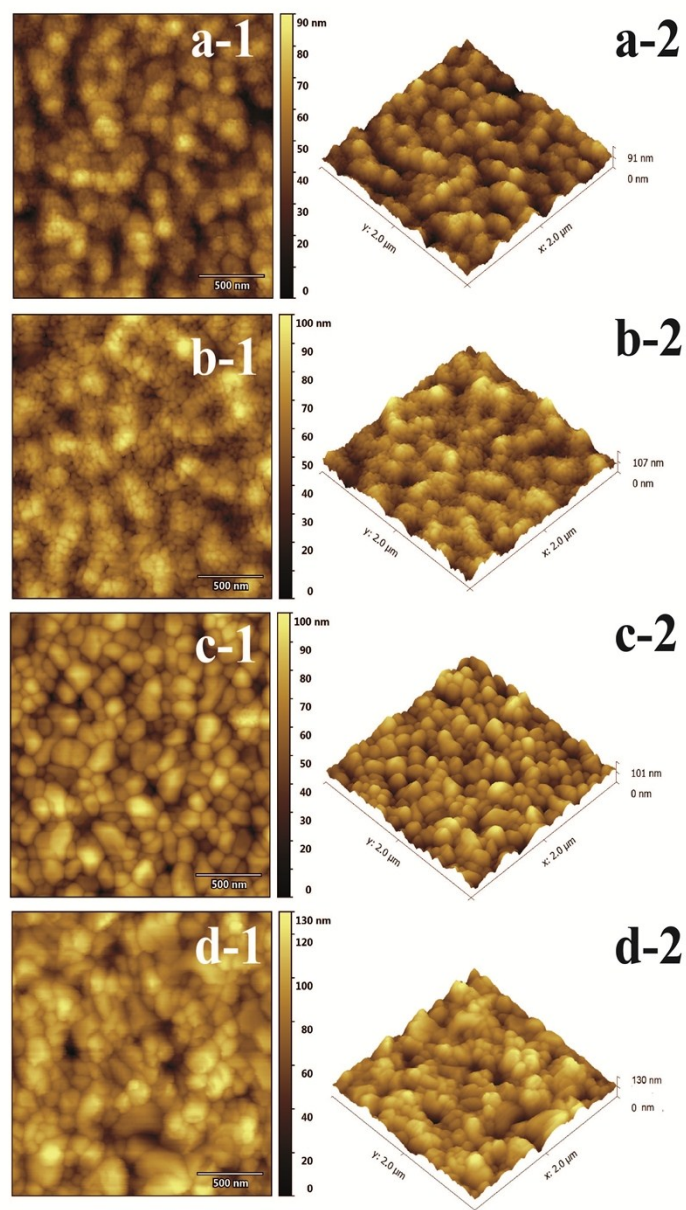
Samples	Size range (nm)	Average size (nm)	Coefficient of variation
BiVO <sub>4</sub> -400	10-50	29	0.24
BiVO <sub>4</sub> -450	10-80	45	0.24
BiVO <sub>4</sub> -500	20-220	107	0.31
BiVO <sub>4</sub> -540	30-360	138	0.50

**Table S3.** XRD crystallite size calculation based on (013) (112) peaks.

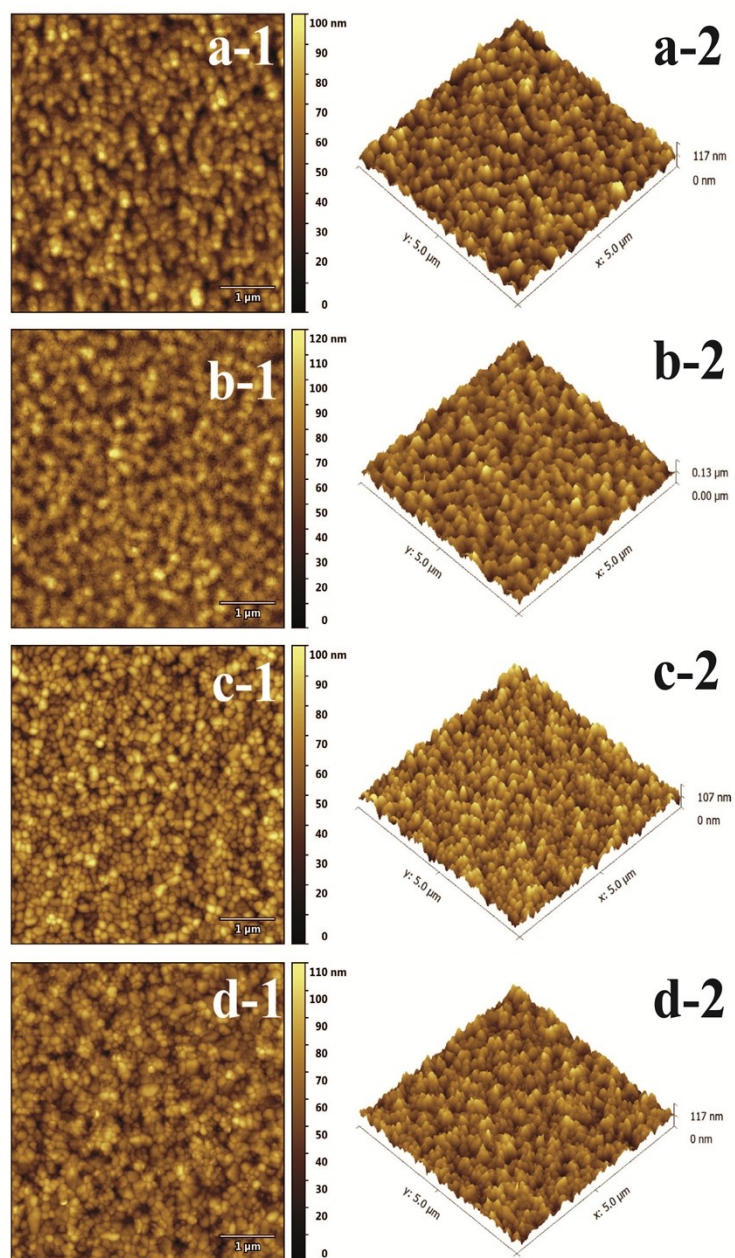
Peak position	Annealing Temperature	Peak position	(FWHM) Peak Width	Crystallite size
(013) (112)	400°C	28.91	0.41	25
	450°C	28.93	0.31	35
	500°C	28.94	0.23	52
	540°C	28.96	0.22	53



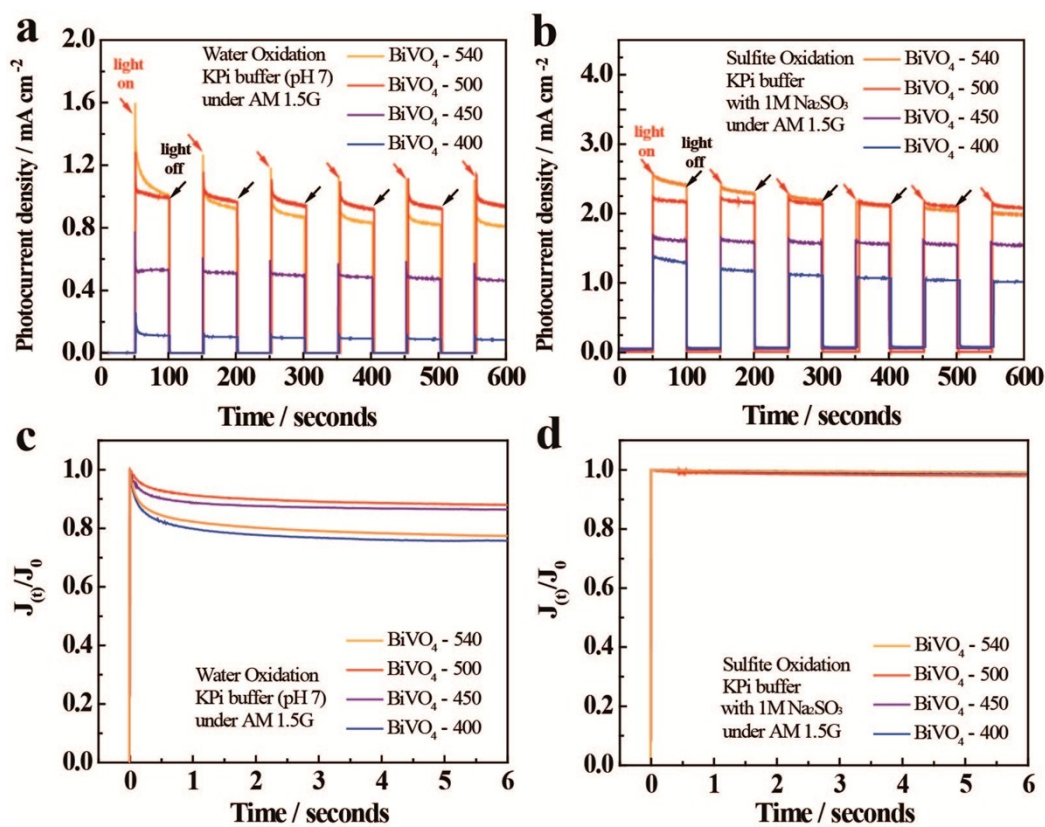
**Figure S7.** UV-Vis diffuse reflection spectra BiVO<sub>4</sub>-400, BiVO<sub>4</sub>-450, BiVO<sub>4</sub>-500 and BiVO<sub>4</sub>-540 photoelectrodes with 6 coating layers.



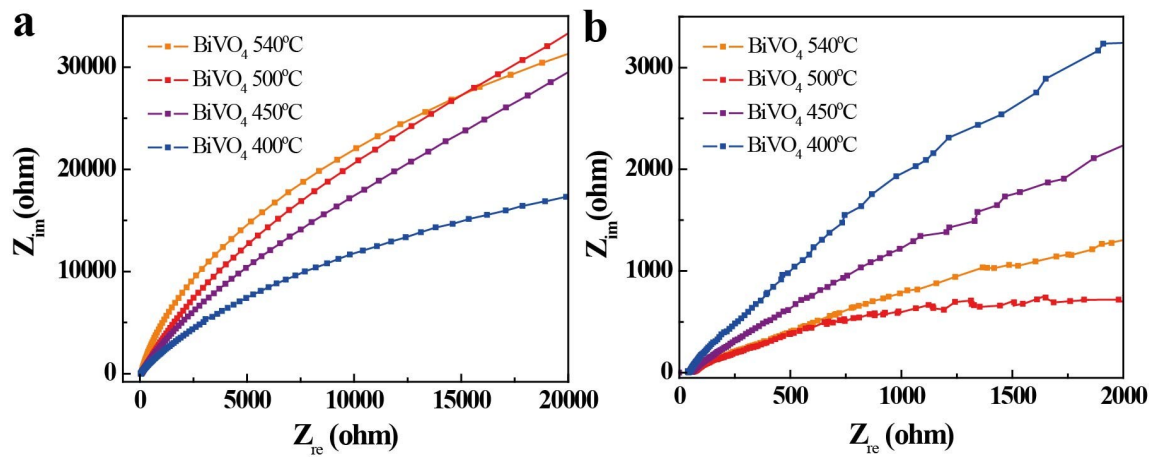
**Figure S8.** Topographic AFM images of (a1-a2) BiVO<sub>4</sub>-400 (Rms = 12.9 nm), (b1-b2) BiVO<sub>4</sub>-450 (Rms = 13.8 nm), (c1-c2) BiVO<sub>4</sub>-500 (Rms = 13.8 nm), and (d1-d2) BiVO<sub>4</sub>-540 (Rms = 13.5 nm). The scanned areas was 2.0 × 2.0 μm in all cases.



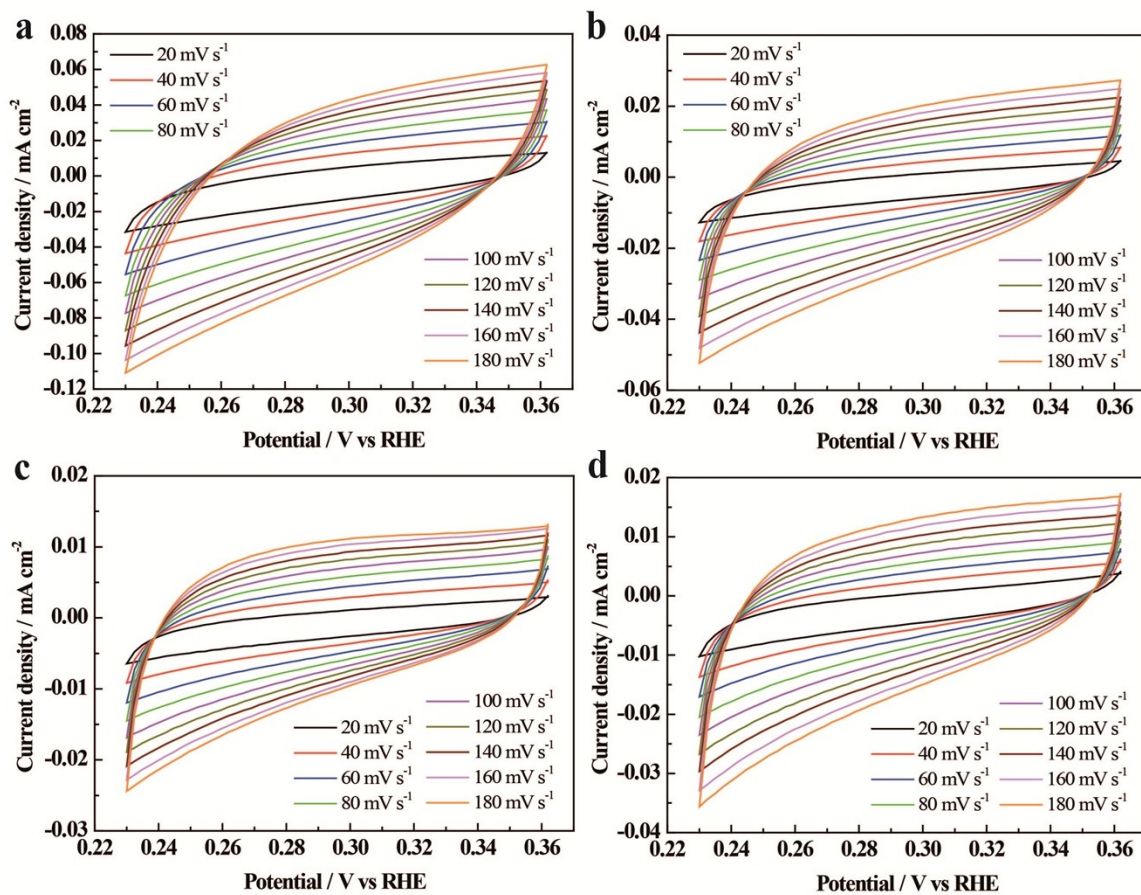
**Figure S9.** Topographic AFM images of the sample (a1-a2)  $\text{BiVO}_4$ -400 (Rms = 13.8 nm), (b1-b2)  $\text{BiVO}_4$ -450 (Rms = 14.5 nm), (c1-c2)  $\text{BiVO}_4$ -500 (Rms = 13.7 nm) and (d1-d2)  $\text{BiVO}_4$ -540 (Rms = 13.6 nm). The scanned area was  $5.0 \times 5.0 \mu\text{m}$  in all.



**Figure S10.** Amperometric I-t curves of BiVO<sub>4</sub>-n samples, measured at 1.23 V vs RHE in 0.1 M KPi buffer (a) and 0.1 M KPi containing 1 M Na<sub>2</sub>SO<sub>3</sub> (b) under AM 1.5G illumination condition with a 50 s light on/off cycles. (c) and (d) are normalized photocurrent transient behaviors of BiVO<sub>4</sub>-n samples, based on the fourth cycle in (a) and (b) respectively. J<sub>0</sub> is the photocurrent peak measured when t equals to 0.

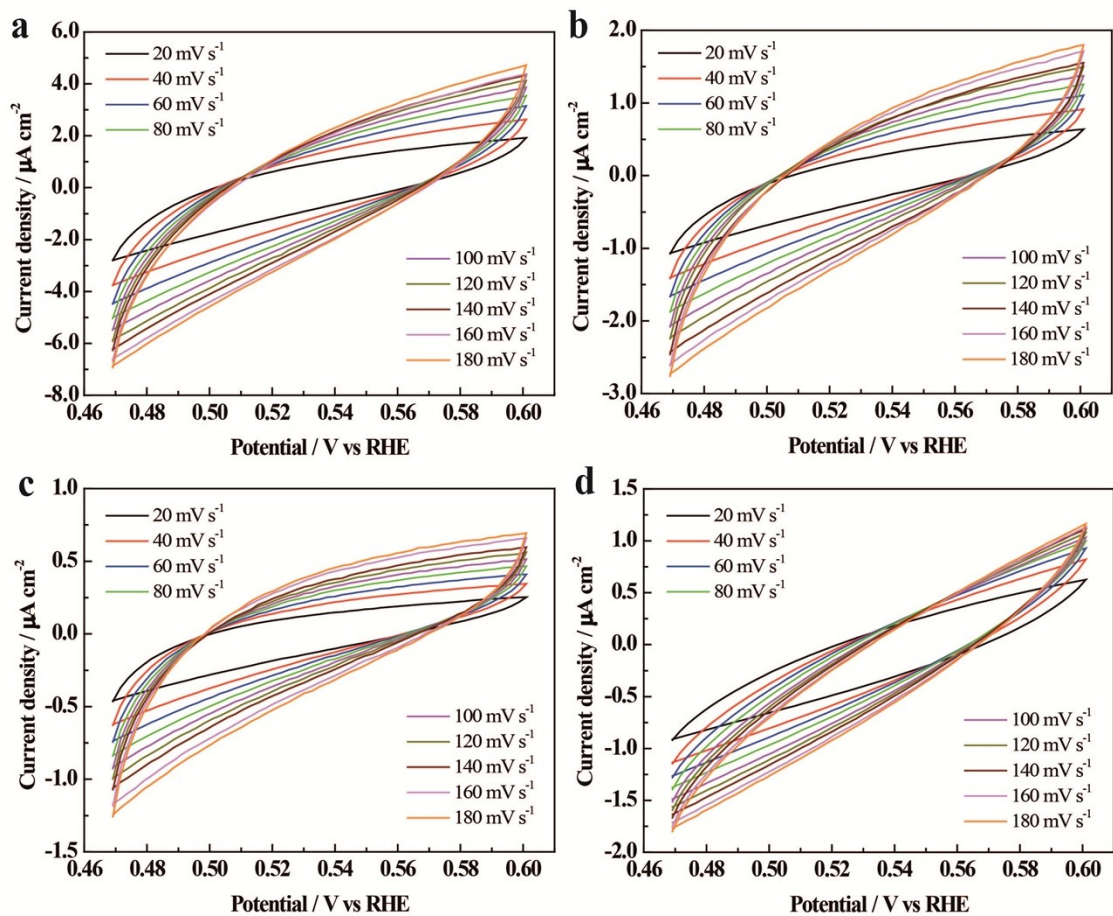


**Figure S11** Nyquist plots ( $Z_{re}$  vs.  $Z_{im}$ ) for (a)  $\text{BiVO}_4$ -n samples under dark condition and (b)  $\text{BiVO}_4$ -n samples under AM 1.5 illumination in 0.1M KPi buffer.

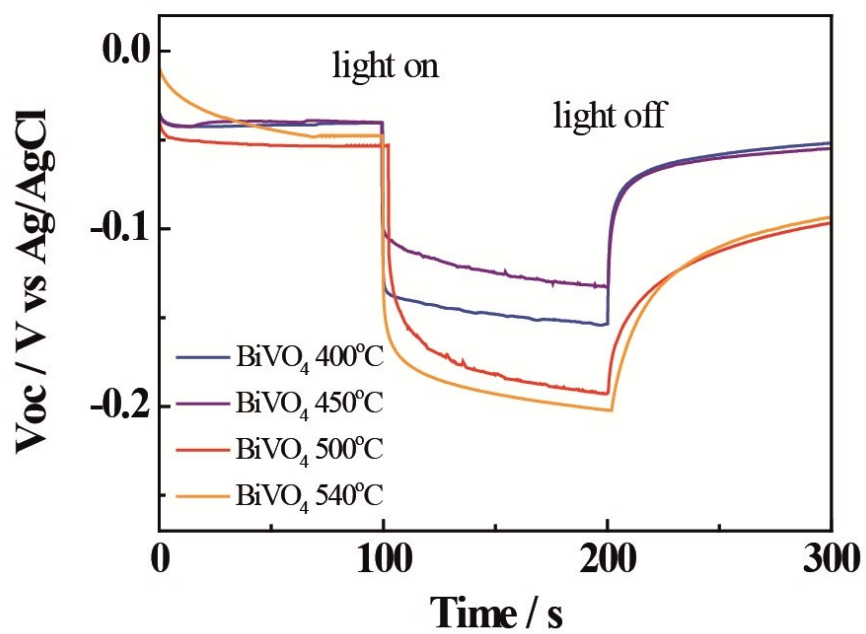


**Figure S12** Cyclic voltammetry (CV) potentiostatic curves of BiVO<sub>4</sub>-400 (a), BiVO<sub>4</sub>-450 (b), BiVO<sub>4</sub>-500 (c) and BiVO<sub>4</sub>-540 (d) measured under light condition, in 0.1 M KPi buffer in the non-Faradaic region with different scan rates from 20 mV s<sup>-1</sup> to 180 mV s<sup>-1</sup>.





**Figure S13** CV curves of BiVO<sub>4</sub>-400 (a), BiVO<sub>4</sub>-450 (b), BiVO<sub>4</sub>-500 (c) and BiVO<sub>4</sub>-540 (d) measured under dark condition, in 0.1 M KPi buffer in the non-Faradaic region with different scan rates from 20  $\text{mV s}^{-1}$  to 180  $\text{mV s}^{-1}$ .



**Figure S14** Time dependence of  $V_{oc}$  curves measured in 0.1 M KPi buffer in the dark and under AM 1.5G illumination.

### Calculation of bulk and surface charge separation efficiency

The photocurrent density ( $J_{PEC}$ , which can be retrieved from LSV) of a sample is determined by the following equation<sup>18</sup>:

$$J_{PEC} = J_{abs} \times \eta_{bulk} \times \eta_{surface}$$

where  $J_{abs}$  is the photocurrent density when absorbed photons into current efficiency (APCE) is 100%,  $\eta_{bulk}$  is the charge separation efficiency in the bulk of sample, and  $\eta_{surface}$  is the charge separation efficiency at the surface of the sample.

Electrolyte buffered with 0.1 M KPi containing 1 M  $Na_2SO_3$  is reasonably assumed to completely suppress the surface recombination of charge carriers without influencing the charge separation in bulk of the sample electrode, which means  $\eta_{surface}$  is 100% in the case with 1 M  $Na_2SO_3$  in the electrolyte.

Therefore,  $\eta_{bulk}$  and  $\eta_{surface}$  can be determined as follows:

$$\eta_{bulk} = J_{sulfite} / J_{abs}$$

$$\eta_{surface} = J_{water} / J_{sulfite}$$

where  $J_{water}$  and  $J_{sulfite}$  are the photocurrent density for PEC water oxidation and sulfite oxidation, respectively. By overlapping light absorption efficiency (Figure 4a in the manuscript) and AM 1.5G spectrum, the  $J_{abs}$  of  $BiVO_4-n$  is calculated to be 6.03 mA  $cm^{-2}$  for  $BiVO_4-400$ , 5.94 mA  $cm^{-2}$  for  $BiVO_4-450$ , 5.98 mA  $cm^{-2}$  for  $BiVO_4-500$  and 5.91 mA  $cm^{-2}$  for  $BiVO_4-540$ . As a result,  $\eta_{bulk}$  and  $\eta_{surface}$  of  $BiVO_4-n$  are calculated and presented in Figure 7b and 7c.

## Reference

1. Chatchai, P.; Murakami, Y.; Kishioka, S.-y.; Nosaka, A. Y.; Nosaka, Y., Efficient photocatalytic activity of water oxidation over WO<sub>3</sub>/BiVO<sub>4</sub> composite under visible light irradiation. *Electrochimica Acta* **2009**, *54* (3), 1147-1152.
2. Kazuhiro, S.; Nini, W.; Yugo, M.; Hitoshi, K.; Nobuko, O.-K.; Hideki, S., Effect of Carbonate Ions on the Photooxidation of Water over Porous BiVO<sub>4</sub> Film Photoelectrode under Visible Light. *Chemistry Letters* **2010**, *39* (1), 17-19.
3. Jeong, H. W.; Jeon, T. H.; Jang, J. S.; Choi, W.; Park, H., Strategic Modification of BiVO<sub>4</sub> for Improving Photoelectrochemical Water Oxidation Performance. *The Journal of Physical Chemistry C* **2013**, *117* (18), 9104-9112.
4. Choi, S. K.; Choi, W.; Park, H., Solar water oxidation using nickel-borate coupled BiVO<sub>4</sub> photoelectrodes. *Physical Chemistry Chemical Physics* **2013**, *15* (17), 6499-6507.
5. Xia, L.; Bai, J.; Li, J.; Zeng, Q.; Li, L.; Zhou, B., High-performance BiVO<sub>4</sub> photoanodes cocatalyzed with an ultrathin  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> layer for photoelectrochemical application. *Applied Catalysis B: Environmental* **2017**, *204* (Supplement C), 127-133.
6. Chemseddine, A.; Ullrich, K.; Mete, T.; Abdi, F. F.; van de Krol, R., Solution-processed multilayered BiVO<sub>4</sub> photoanodes: influence of intermediate heat treatments on the photoactivity. *Journal of Materials Chemistry A* **2016**, *4* (5), 1723-1728.
7. Parmar, K. P. S.; Kang, H. J.; Bist, A.; Dua, P.; Jang, J. S.; Lee, J. S., Photocatalytic and Photoelectrochemical Water Oxidation over Metal-Doped Monoclinic BiVO<sub>4</sub> Photoanodes. *ChemSusChem* **2012**, *5* (10), 1926-1934.
8. Quiñonero, J.; Lana-Villarreal, T.; Gómez, R., Improving the photoactivity of bismuth vanadate thin film photoanodes through doping and surface modification strategies. *Applied Catalysis B: Environmental* **2016**, *194* (Supplement C), 141-149.
9. Li, T.; He, J.; Peña, B.; Berlinguette, C. P., Curing BiVO<sub>4</sub> Photoanodes with Ultraviolet Light Enhances Photoelectrocatalysis. *Angewandte Chemie International Edition* **2016**, *55* (5), 1769-1772.
10. Kalanur, S. S.; Yoo, I.-H.; Park, J.; Seo, H., Insights into the electronic bands of WO<sub>3</sub>/BiVO<sub>4</sub>/TiO<sub>2</sub>, revealing high solar water splitting efficiency. *Journal of Materials Chemistry A* **2017**, *5* (4), 1455-1461.
11. Cheng, B.-Y.; Yang, J.-S.; Cho, H.-W.; Wu, J.-J., Fabrication of an Efficient BiVO<sub>4</sub>-TiO<sub>2</sub> Heterojunction Photoanode for Photoelectrochemical Water Oxidation. *ACS Applied Materials & Interfaces* **2016**, *8* (31), 20032-20039.
12. Ma, Y.; Pendlebury, S. R.; Reynal, A.; Le Formal, F.; Durrant, J. R., Dynamics of photogenerated holes in undoped BiVO<sub>4</sub> photoanodes for solar water oxidation. *Chemical Science* **2014**, *5* (8), 2964-2973.
13. Jia, Q.; Iwashina, K.; Kudo, A., Facile fabrication of an efficient BiVO<sub>4</sub> thin film electrode for water splitting under visible light irradiation. *Proceedings of the National Academy of Sciences* **2012**, *109* (29), 11564-11569.
14. Shinde, P. S.; Annamalai, A.; Kim, Ju H.; Choi, Sun H.; Lee, Jae S.; Jang, Jum S., Exploiting the dynamic Sn diffusion from deformation of FTO to boost the

photocurrent performance of hematite photoanodes. *Solar Energy Materials and Solar Cells* **2015**, *141*, 71-79.

15. Yang, Y.; Zhan, F.; Li, H.; Liu, W.; Yu, S., In situ Sn-doped WO<sub>3</sub> films with enhanced photoelectrochemical performance for reducing CO<sub>2</sub> into formic acid. *Journal of Solid State Electrochemistry* **2017**, *21* (8), 2231-2240.

16. Rachel, M.; Mahfujur, R.; Don, M. J. M.; A., W. C., Activation of Hematite Nanorod Arrays for Photoelectrochemical Water Splitting. *ChemSusChem* **2011**, *4* (4), 474-479.

17. Li, M.; Deng, J.; Pu, A.; Zhang, P.; Zhang, H.; Gao, J.; Hao, Y.; Zhong, J.; Sun, X., Hydrogen-treated hematite nanostructures with low onset potential for highly efficient solar water oxidation. *Journal of Materials Chemistry A* **2014**, *2* (19), 6727-6733.

18. Chang, X.; Wang, T.; Zhang, P.; Zhang, J.; Li, A.; Gong, J., Enhanced Surface Reaction Kinetics and Charge Separation of p-n Heterojunction Co<sub>3</sub>O<sub>4</sub>/BiVO<sub>4</sub> Photoanodes. *Journal of the American Chemical Society* **2015**, *137* (26), 8356-8359.