

***Utilization of modified *Dioscorea opposita* Thunb as a novel biosorbent  
for the adsorption of indigo carmine in aqueous solutions***

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**Table S1 Element composition of DOT and DOT@PEI**

Element	DOT		DOT@PEI	
	Weight percentage (%)	Atomic percentage (%)	Weight percentage (%)	Atomic percentage (%)
C	28.309	38.736	45.613	37.061
O	51.329	52.725	51.854	56.122
Mg	0.859	0.581	0.255	0.419
Si	1.000	0.585	0.346	0.657
P	0.902	0.479	0.178	0.374
S	1.294	0.663	0.089	0.193
Cl	0.905	0.419	0.142	0.341
K	8.051	3.384	0.448	1.186
Ca	3.585	1.470	0.635	1.721
Fe	0.265	0.078	0.026	0.099
Zn	3.502	0.880	0.413	1.827

### Figure Captions

**Fig. S1** SEM micrographs of (a) DOT and (b) DOT@PEI

**Fig. S2** EDS spectra of (a) DOT and (b) DOT@PEI

**Fig. S3** FTIR spectra of biosorbents: (a) DOT, (b) DOT@PEI

**Fig. S4** BET isotherm of DOT@PEI

**Fig. S5** The zeta potential of DOT and DOT@PEI (s) at varied pH conditions.

**Fig. S6** Plots of (a) Langmuir isotherm model for the adsorption of Indigo Carmine onto the DOT@PEI at 20°C.

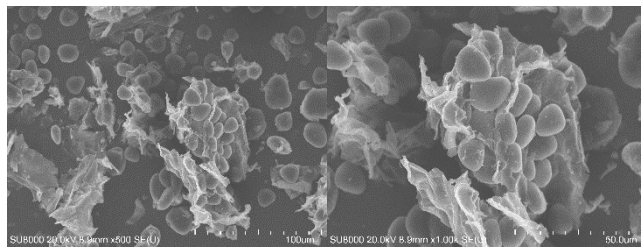
**Fig. S7** Plots of (a) Freundlich isotherm models for the adsorption of Indigo Carmine onto the DOT@PEI at 20°C.

**Fig. S8** Plots of (a) Temkin isotherm models for the adsorption of Indigo Carmine onto the DOT@PEI at 20°C.

**Fig. S9** Pseudo-First-Order Kinetic Model for adsorption of Indigo Carmine onto the DOT@PEI at 20°C.

**Fig. S10** Pseudo-Second-Order Kinetic Model for adsorption of Indigo Carmine onto the DOT@PEI at 20°C.

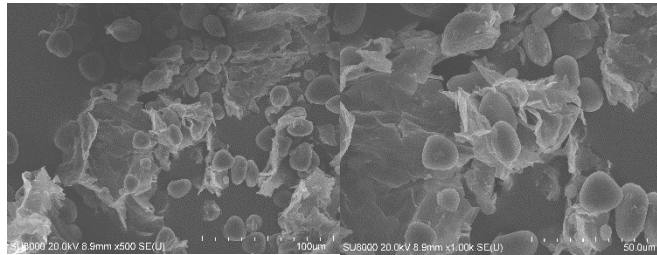
**Fig. S11** Intraparticle diffusion model for adsorption of Indigo Carmine onto the DOT@PEI at 20°C.



500×

1000×

(a) DOT

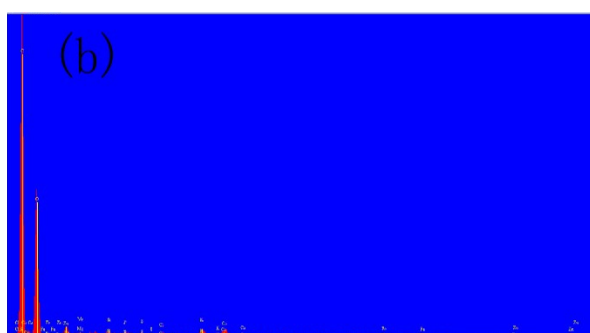
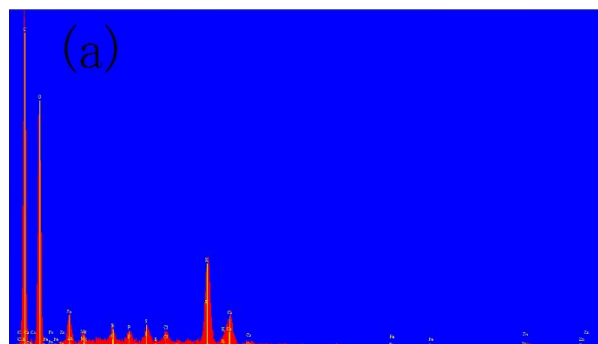


500×

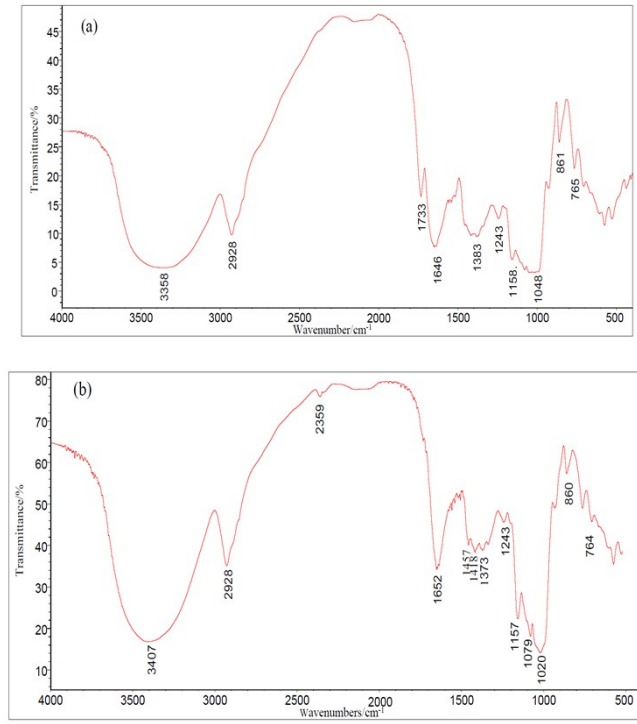
1000×

(b) DOT@PEI

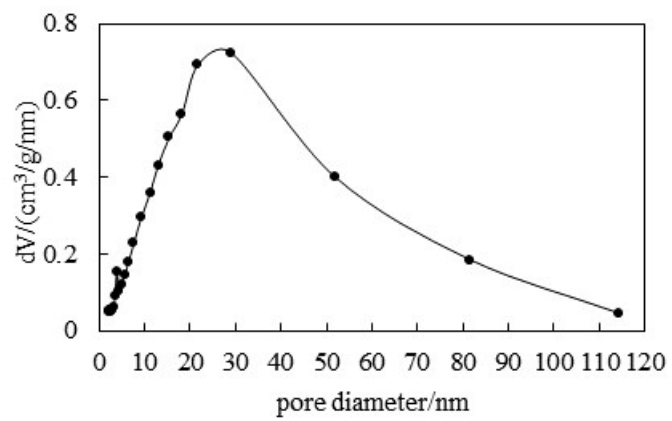
**Fig. S1 SEM micrographs of (a) DOT and (b) DOT@PEI**



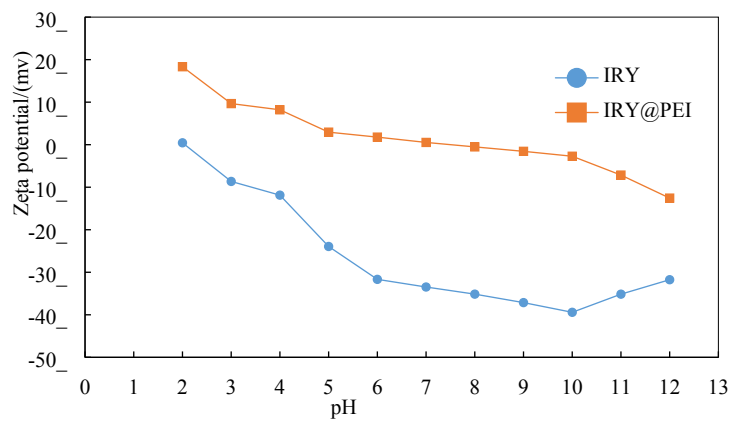
**Fig. S2 EDS spectra of (a) DOT and (b) DOT@PEI**



**Fig. S3 FTIR spectra of biosorbents: (a) DOT, (b) DOT@PEI**

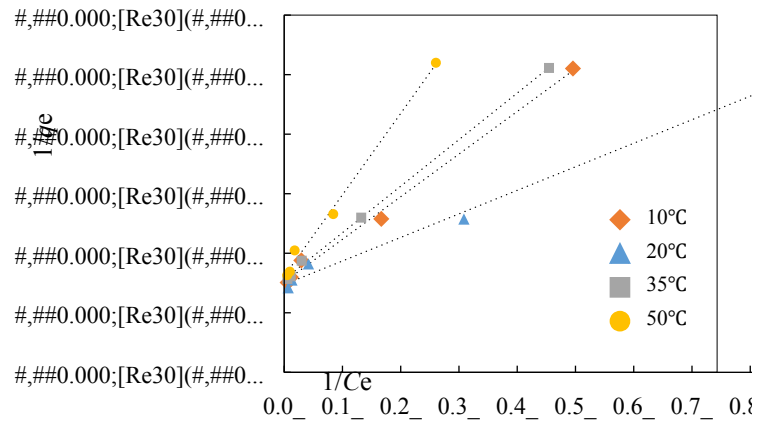


**Fig. S4 BET isotherm of DOT@PEI**

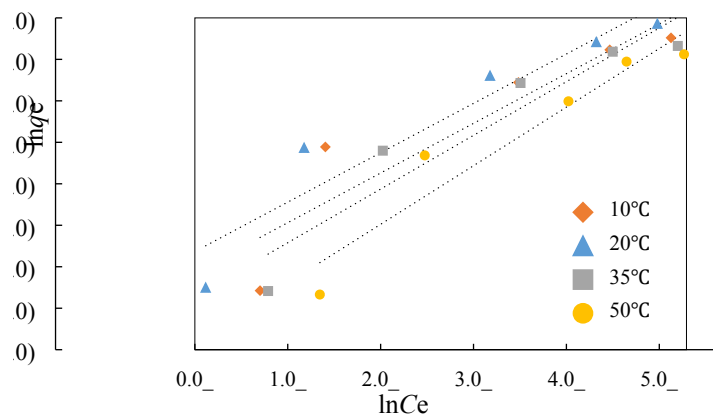


**Fig. S5 The zeta potential of DOT and DOT@PEI (s) at varied pH conditions.**

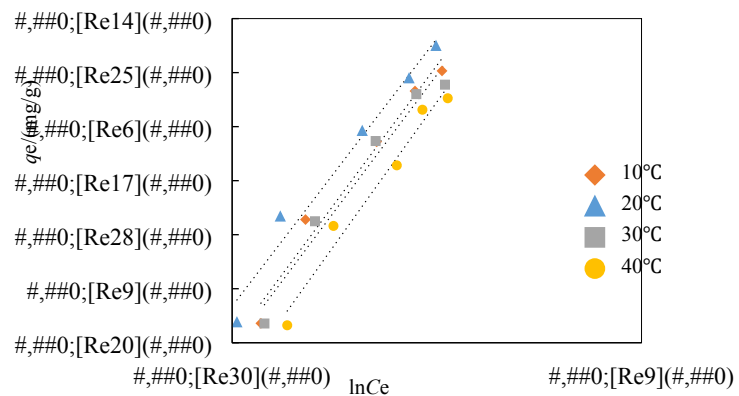




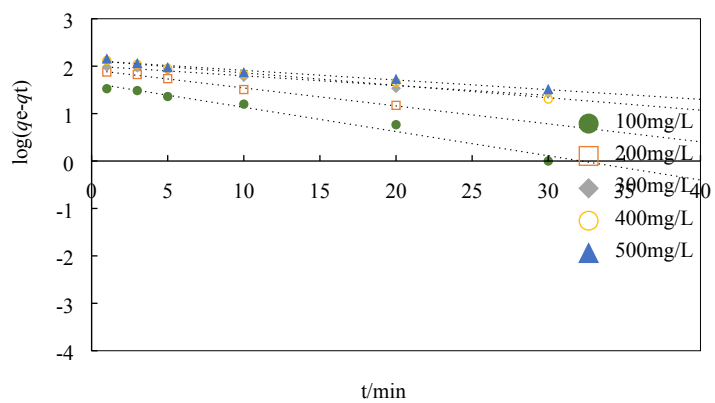
**Fig. S6 Plots of (a) Langmuir isotherm model for the adsorption of Indigo Carmine onto the DOT@PEI at 20°C.**



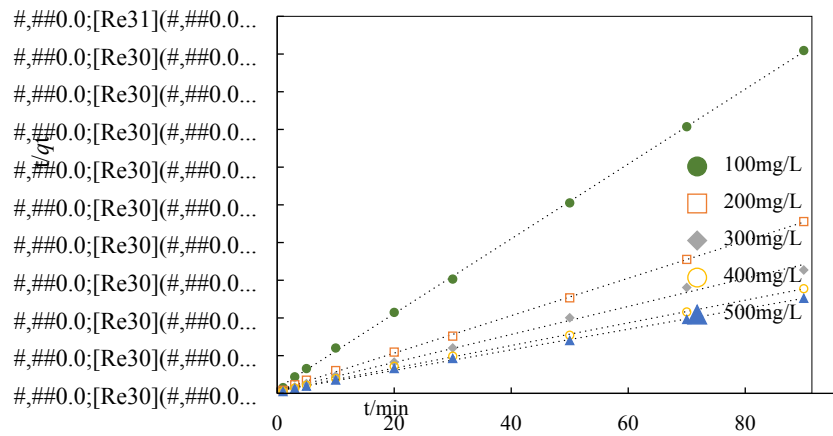
**Fig. S7 Plots of (a) Freundlich isotherm models for the adsorption of Indigo Carmine onto the DOT@PEI at 20°C.**



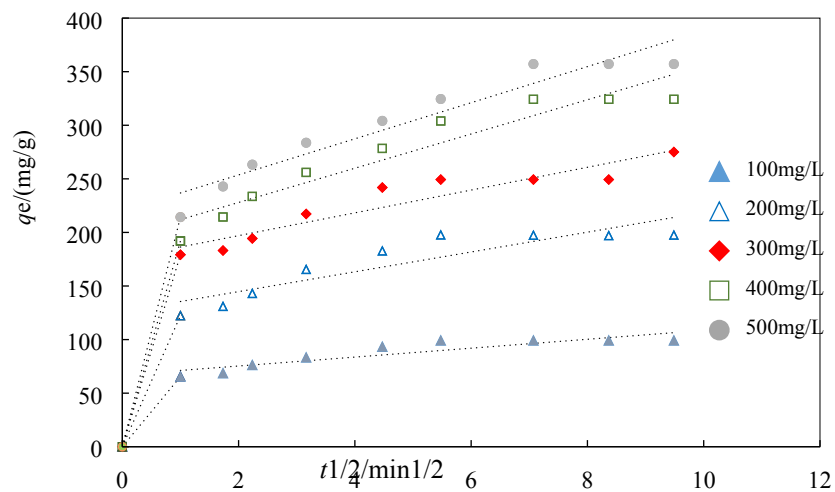
**Fig. S8 Plots of (a) Temkin isotherm models for the adsorption of Indigo Carmine onto the DOT@PEI at 20°C.**



**Fig. S9 Pseudo-First-Order Kinetic Model for adsorption of Indigo Carmine onto the DOT@PEI at 20°C.**



**Fig. S10 Pseudo-Second-Order Kinetic Model for adsorption of Indigo Carmine onto the DOT@PEI at 20°C.**



**Fig. S11** Intraparticle diffusion model for adsorption of Indigo Carmine onto the DOT@PEI at 20°C.