

**Supplementary Material**

**Synthesis and characterization of amorphous precipitated silica from alkaline  
dissolution of olivine**

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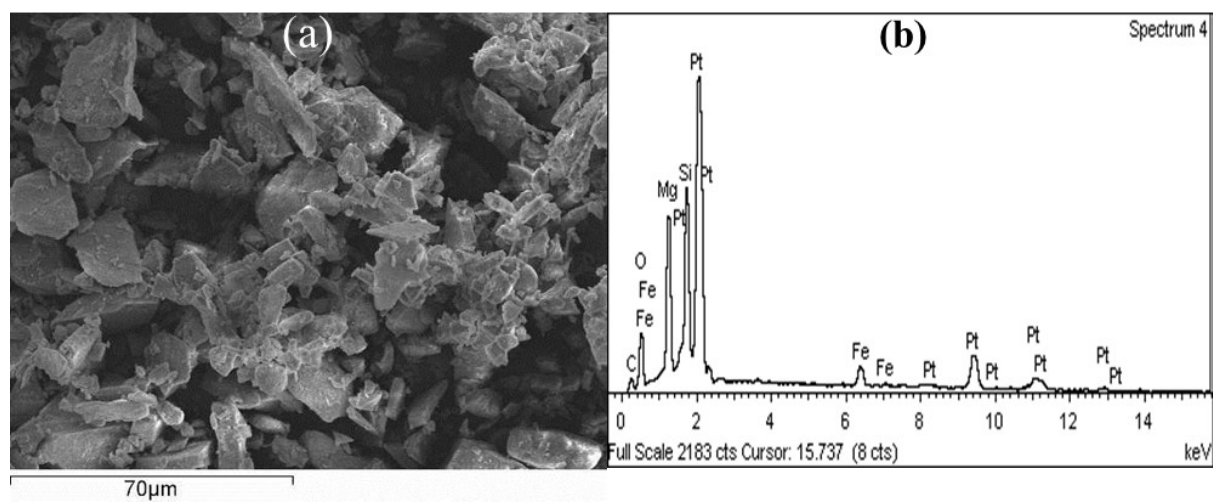
**Table S1.**Chemical analysis of olivine

Serial #	Component	[Wt.%]
1	MgO	40.0
2	Fe <sub>2</sub> O <sub>3</sub>	9.5
3	Al <sub>2</sub> O <sub>3</sub>	0.7
4	CaO	0.4
5	SiO <sub>2</sub>	39.0
6	Loss on ignition	11.4 %

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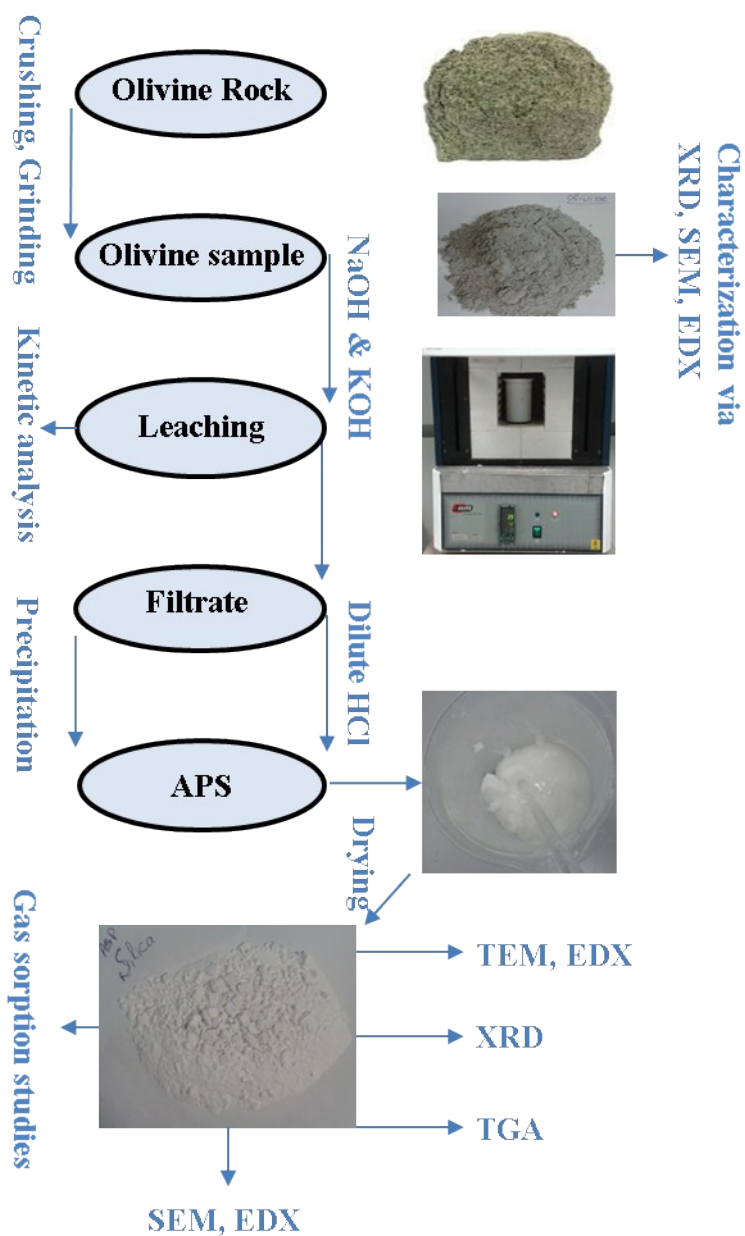


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21 **Fig. S1** Morphological characteristics of ground olivine(a)SEM image (b) EDX spectrum

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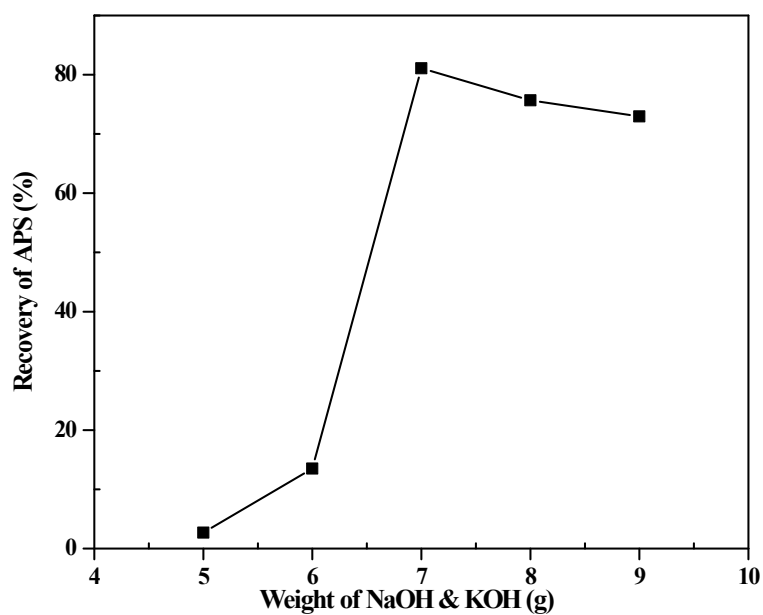
## Schematic of APS synthesis from olivine



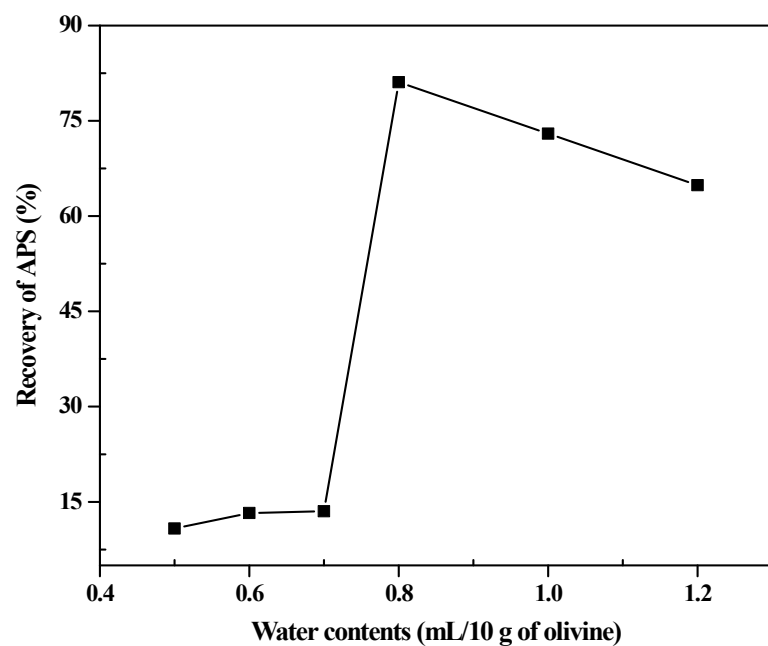
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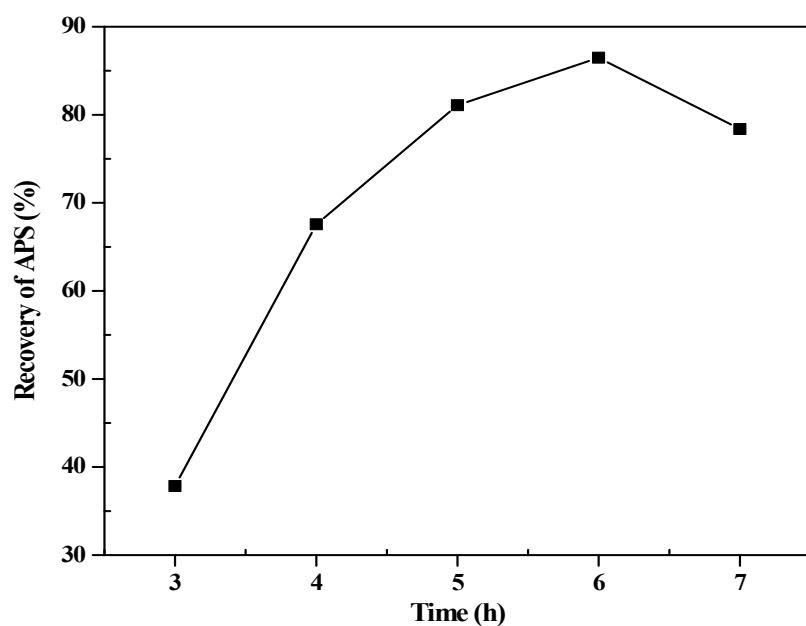
25 **Fig. S2** Schematic of APS synthesis from olivine using alkaline dissolution.



**Fig. S3** Effect of concentration of NaOH and KOH on the leaching of silica from olivine (Temperature 170°C, H<sub>2</sub>O content 0.8 mL, and reaction time 5 h for 10 g olivine).



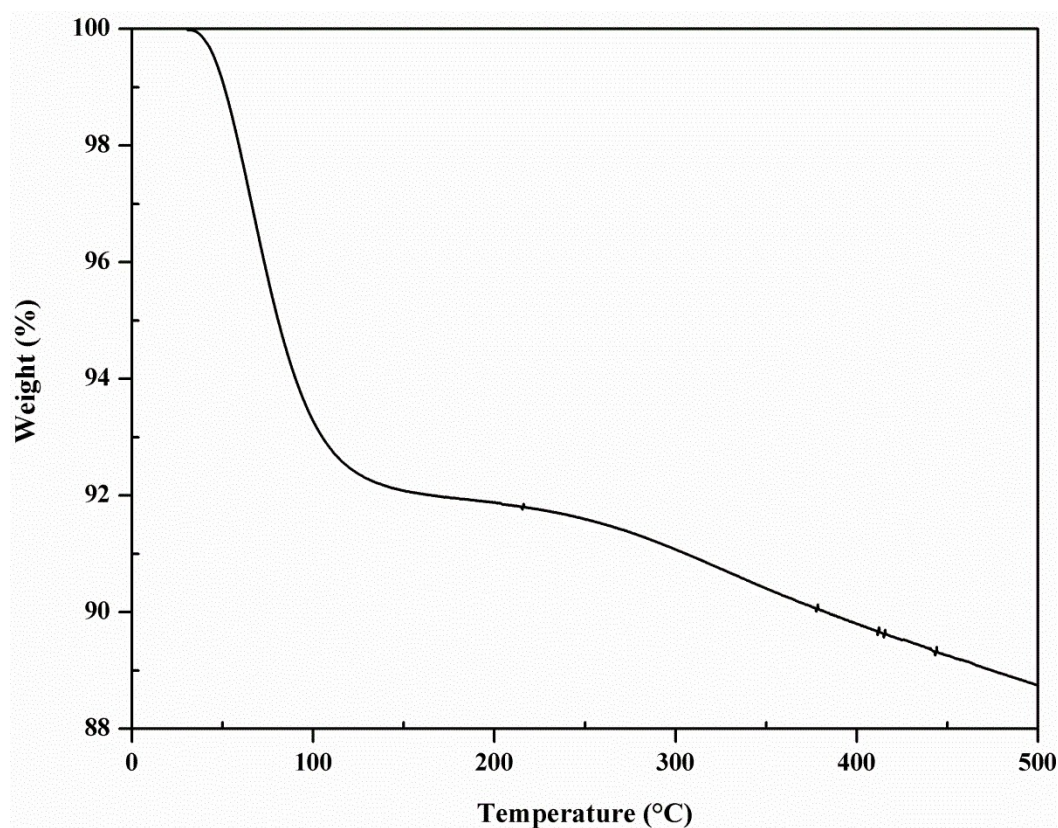
**Fig. S4** Effect of water content on the leaching of silica from olivine (Temperature 170 °C, mass of bases 14 g, reaction time 5 h for 10 g olivine).



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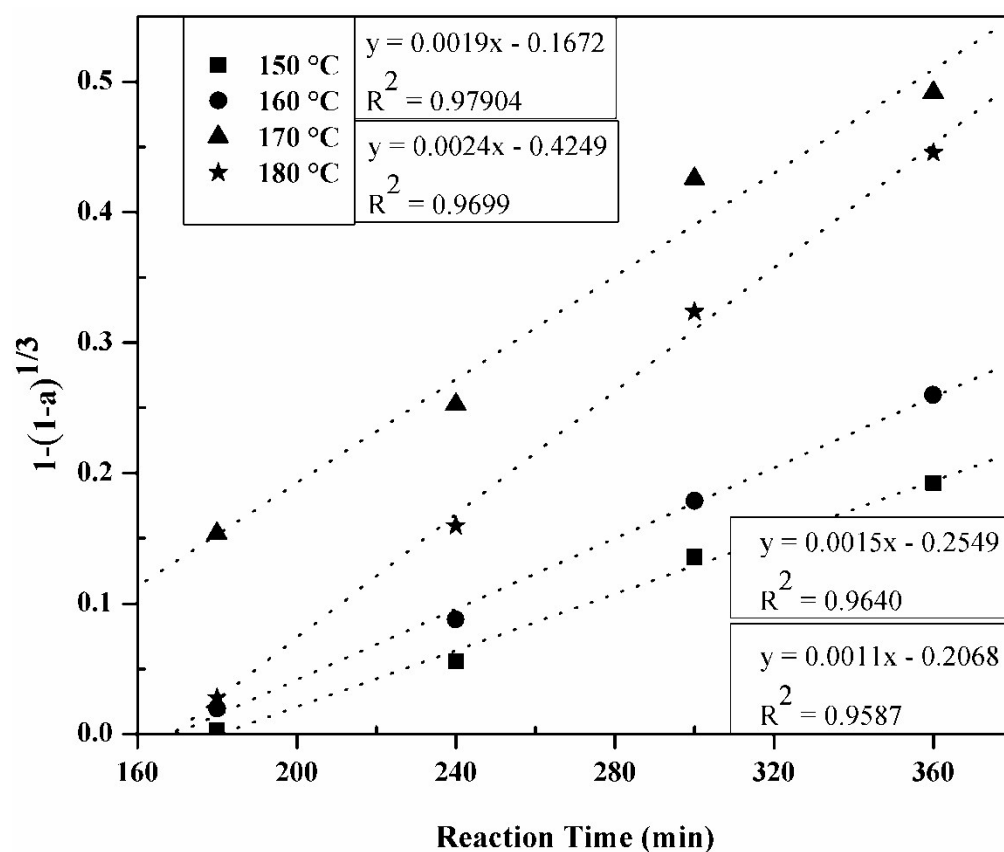
36 **Fig. S5** Effect of reaction time on leaching of APS from olivine (Temperature 170 °C, H<sub>2</sub>O  
 37 content 0.8 mL, mass of bases 14 g for 10 g olivine).



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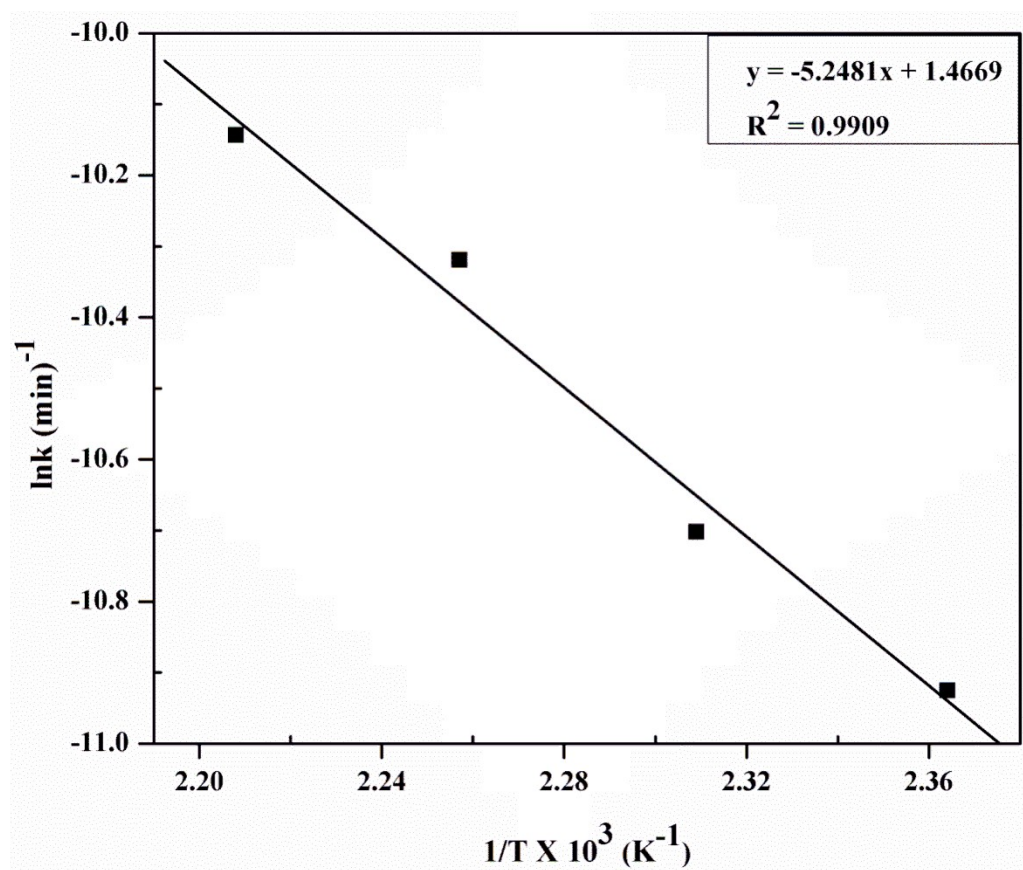
40 **Fig. S6** TGA curve of APS.



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43 **Fig. S7A** plot of  $1 - (1 - \alpha)^{1/3}$  vs. reaction time.



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45 **Fig. S8** Arrhenius plot.