Electronic supplementary information

Template-free synthesis of three dimensional porous boron nitride nanosheets for efficient water cleaning

Jie Li^{1,2,*}, Shi He¹, Rui Li^{1,2}, Wei Dai^{1,2}, Junhui Tao^{1,2}, Chuanhui Wang^{1,2}, Junyi Liu², Tian Wu^{1,2,*}, Chengchun Tang³

 ¹ School of Physics and Mechanical & Electronical Engineering, Institute for Functional Materials, Hubei University of Education, Wuhan 430205, P.R. China
² Institute of materials research and engineering, Hubei University of Education, Wuhan 430205, P.R. China
³ School of Materials Science and Engineering, Hebei University of Technology, Tianjin 300130, P. R. China
*Corresponding author. Tel.: +86-27-87943673; fax: +86-27-87943673 E-mail address: lijie@hue.edu.cn (J. Li); twu@whu.edu.cn (T. Wu)



Fig. S1 The photo of the obtained 3D porous BNNSs.

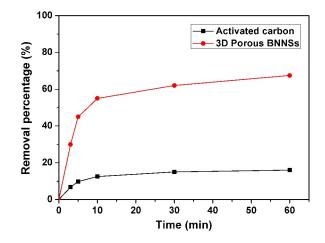


Fig. S2 Effect of contact time on adsorption of MB on the activated carbon and the 3D porous BNNSs reported previously (mass of adsorbent: 50 mg, solution pH: 8, dye concentration: 100 mg L^{-1} , adsorption temperature: 30 ° C).

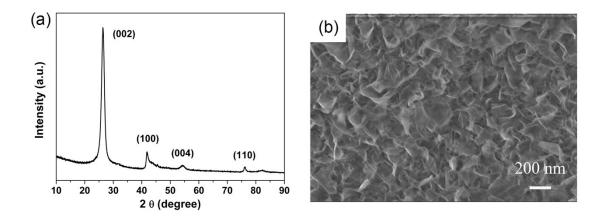


Fig. S3 (a) Powder XRD pattern of the regenerated 3D porous BNNSs. (b) The corresponding SEM image.

Adsorbents	maximum capacity (mg/g)	Ref
porous BNNSs	313	14
BN nanocarpets	272.4	16
Activatec BN	392.2	17
3D BNNSs foam	497	20
BN hollow spheres	191.7	37
3D porous BNNSs	413.3	this work

Table S1. adsorption capabilities of some reported BN materials for organic dyes.

Equation S1.

$$V_{liq} = \frac{P_a V_{ads} V_m}{RT}$$

In which P_a and T are ambient pressure and temperature, respectively, and V_m is the molar volume of the liquid adsorbate.